

F 30-039-20836

F. Loc. 1550/N;1717/E Elev. 6968 G1 Spd. _____ Comp. _____ TD _____ PB _____
 Casing S. _____ W. _____ Sx. _____ Int. _____ W. _____ Sx. Pr. _____ W. _____ Sx. _____
 Csg. Perf. _____ Prod. Stim. _____

T
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A
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S

I.P. _____ BO D _____ MCF D After _____ Hrs. _____ SICP _____ PSI After _____ Days GOR _____ Grav. _____ 1st Def. _____ s _____	
TOPS	NITD ☒ Well Log
Kirtland	C-103 ☒ Plat ☒ TEST DATA
Fruitland	C-104 Electric Log
Pictured Cliffs	C-122
Cliff House	Dtr Dfa
Menefee	Datr Dac
Point Lookout	
Mancos	
Gallup	
Sanostee	
Greenhorn	
Dakota	
Morrison	
Entrada	
	320

P
C
S

Basin Dak Co. RA S 8 T 27N R 4W U G Oper. EPNG CO

Loc. San Juan 27-4 U No. 95

P. O. Box 1809
Durango, Colorado 81301

April 9, 1975

El Paso Natural Gas Company
P. O. Box 990
Farmington, New Mexico

Gentleman:

Approval to drill your No. 95 San Juan 27-4 Unit Dakota well, located in the SW 1/4 sec. 8, T. 27 N., R. 4 W., Rio Arriba County, New Mexico, on Federal lease Santa Fe 080673, is hereby rescinded and the Permit to Drill returned to you.

This action is occasioned by the filing of an application to drill your No. 125 San Juan 27-4 Unit Pictured Cliffs-Masoverde dual well in the same spot where the No. 95 had been approved. It is our understanding that a new application will be filed to drill the No. 95 at the location previously approved by the Forest Service for the No. 125 well.

Sincerely yours,

(Per. Sig.) J. W. LONG

Jerry W. Long
District Engineer

cc:
Roswell
NMOC (2)✓
U.S.F.S.

JWLong:mbh

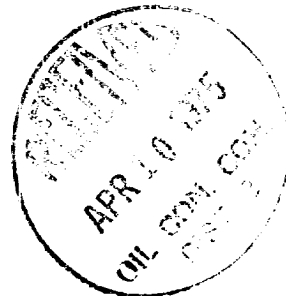


Figure 1. The effect of the concentration of the H_2O_2 solution on the amount of the released H_2O_2 from the H_2O_2 -loaded hydrogel. The amount of the released H_2O_2 was measured by the amount of the released H_2O_2 from the H_2O_2 -loaded hydrogel. The amount of the released H_2O_2 was measured by the amount of the released H_2O_2 from the H_2O_2 -loaded hydrogel.

Figure 1. Schematic representation of the experimental design. The subjects were divided into two groups: the control group and the experimental group. The control group was divided into two subgroups: the control group and the experimental group. The experimental group was divided into two subgroups: the control group and the experimental group. The control group was divided into two subgroups: the control group and the experimental group. The experimental group was divided into two subgroups: the control group and the experimental group.

1. The first step in the process of the investigation is the identification of the problem. This is done by the investigator who is responsible for the study. The investigator must first identify the problem and then determine the scope of the study. The next step is to design the study. This involves determining the methods to be used and the data to be collected. The third step is to collect the data. This is done by the investigator who is responsible for the study. The fourth step is to analyze the data. This is done by the investigator who is responsible for the study. The fifth step is to interpret the results. This is done by the investigator who is responsible for the study. The sixth step is to write the report. This is done by the investigator who is responsible for the study. The seventh step is to present the results. This is done by the investigator who is responsible for the study. The eighth step is to discuss the results. This is done by the investigator who is responsible for the study. The ninth step is to conclude the study. This is done by the investigator who is responsible for the study. The tenth step is to publish the results. This is done by the investigator who is responsible for the study.

1. *Journal of the American Medical Association*, 1997; 277: 1033-1036.

1. *Chlorophyll a* and *Chlorophyll b* were determined by the method of Lichtenthaler and Whistler (1973). The total chlorophyll content was determined by the method of Arar and Cook (1980). The carotenoid content was determined by the method of Lichtenthaler and Whistler (1973). The total phenolic content was determined by the method of Singleton and Rossi (1965). The total flavonoid content was determined by the method of Zhishen et al. (1999). The total protein content was determined by the method of Lowry et al. (1951). The total lipid content was determined by the method of Folch et al. (1957). The total carbohydrate content was determined by the method of Dubois and Gilles (1950). The total ash content was determined by the method of AOAC (1990). The total acid content was determined by the method of AOAC (1990). The total base content was determined by the method of AOAC (1990). The total nitrogen content was determined by the method of Kjeldahl (1900). The total phosphorus content was determined by the method of Molybdenum blue (1900). The total sulfur content was determined by the method of Barium sulfate (1900). The total calcium content was determined by the method of Oxalate (1900). The total magnesium content was determined by the method of Magnesia (1900). The total potassium content was determined by the method of Potassium dichromate (1900). The total sodium content was determined by the method of Sodium chloride (1900). The total iron content was determined by the method of Iron(III) chloride (1900). The total copper content was determined by the method of Copper(II) sulfate (1900). The total zinc content was determined by the method of Zinc sulfate (1900). The total manganese content was determined by the method of Manganese sulfate (1900). The total cobalt content was determined by the method of Cobalt(II) chloride (1900). The total nickel content was determined by the method of Nickel(II) sulfate (1900). The total chromium content was determined by the method of Chromium(III) chloride (1900). The total boron content was determined by the method of Boric acid (1900). The total molybdenum content was determined by the method of Molybdenum trioxide (1900). The total selenium content was determined by the method of Selenium dioxide (1900). The total tellurium content was determined by the method of Telluric acid (1900). The total vanadium content was determined by the method of Vanadyl sulfate (1900). The total niobium content was determined by the method of Niobium pentoxide (1900). The total tantalum content was determined by the method of Tantalum pentoxide (1900). The total hafnium content was determined by the method of Hafnium tetrachloride (1900). The total zirconium content was determined by the method of Zirconium tetrachloride (1900). The total titanium content was determined by the method of Titanium tetrachloride (1900). The total vanadium content was determined by the method of Vanadyl sulfate (1900). The total niobium content was determined by the method of Niobium pentoxide (1900). The total tantalum content was determined by the method of Tantalum pentoxide (1900). The total hafnium content was determined by the method of Hafnium tetrachloride (1900). The total zirconium content was determined by the method of Zirconium tetrachloride (1900). The total titanium content was determined by the method of Titanium tetrachloride (1900).

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