

OIL CONSERVATION COMMISSION  
HOBBS, NEW MEXICO

October 6, 1961

C  
O  
P  
Y  
  
Gulf Oil Corp.  
Box 2167  
Hobbs, New Mexico

Gentlemen:

As you probably know, your W. A. Ramsay "A" Well Nos. 34-K, 36-N and 38-L located in Section 34, T-21-S, R-36-E, which were formerly carried in the South Eunice Pool on the Oil Proration Schedule, are now being carried in the Eumont Pool. This change was made for practical reasons when we created a waterflood area for your Pilot Waterflood on your W. A. Ramsay "A" Lease.

Please report production from these three wells with the other Eumont wells on this lease on Commission Form C-115.

Your prompt attention to this matter will be appreciated.

Yours very truly,

OIL CONSERVATION COMMISSION

Joe D. Ramey  
Proration Manager

JDR/mc

THE POLYMERIZATION OF  
HYDROLYZABLE POLYESTERS

ROBERT H. HARRIS

Department of Chemistry

University of California, San Diego

La Jolla, California 92037

Received May 15, 1968

Revised July 10, 1968

Published August 1, 1968

ABSTRACT

The polymerization of hydrolyzable polyesters

has been studied in the presence of various

metal ions and catalysts. The results

show that the rate of polymerization

is increased by the presence of certain

metal ions, particularly those of the

transition metals. The mechanism of

the reaction is discussed in terms of

the formation of a metal complex with

the ester group. The results are

consistent with the proposed mechanism

for the polymerization of hydrolyzable

polyesters.

Keywords: polymerization; hydrolyzable

polyesters; metal ions; catalysts.

The polymerization of hydrolyzable

polyesters has been studied in the

presence of various metal ions and

catalysts. The results show that the

rate of polymerization is increased

by the presence of certain metal

ions, particularly those of the

transition metals. The mechanism of

the reaction is discussed in terms of

the formation of a metal complex with

the ester group. The results are

consistent with the proposed

mechanism for the polymerization of

hydrolyzable polyesters.

The polymerization of hydrolyzable

polyesters has been studied in the

presence of various metal ions and

catalysts. The results show that the

rate of polymerization is increased

by the presence of certain metal

ions, particularly those of the

transition metals. The mechanism of

the reaction is discussed in terms of

the formation of a metal complex with

the ester group. The results are

consistent with the proposed

mechanism for the polymerization of

hydrolyzable polyesters.

The polymerization of hydrolyzable

polyesters has been studied in the

presence of various metal ions and

catalysts. The results show that the

rate of polymerization is increased

by the presence of certain metal

ions, particularly those of the

transition metals. The mechanism of

the reaction is discussed in terms of

the formation of a metal complex with