NM - 21

GENERAL CORRESPONDENCE

YEAR(S): 1987 - 1969



THIRD FLOOR BROADMOOR BUILDING (505) 393-0505

P.O. Box 1148 Santa Fe, NM 87504-1148

RE: Draft Business Lease No. BL-1186 by and between The State of New Mexico and Petro-Thermo Corporation

Dear Mr. Black:

This is to confirm our telephone conversation yesterday morning with regard to the draft Business Lease referred to above.

Based upon your comments, we have undertaken a diligent search to determine whether the Administrator, Environmental Protection Agency has promulgated final regulations identifying drilling fluids, produced waters and other waste associated with the exploration, development or production of crude oil or natural gas as a hazardous waste.

In this regard, Title 42, United States Code, Section 6921(b)(2) effectively exempts drilling fluids, produced waters and other waste associated with the exploration, development and production of crude oil or natural gas from federal hazardous waste regulations.

Should the Administrator, Environmental Protection Agency choose to take action with regard to this matter, the following steps must be taken before any final regulations listing such materials as hazardous wastes may take effect:

1. A detailed and comprehensive study together with its accompanying report must be submitted to the Committee on Environment and Public Works of the U.S. Senate and the Committee on Interstate and Foreign Commerce of the U.S. House of Representatives [42 U.S.C. Section 4982(n)]; Mr. Nick Black August 5, 1987 Page 2

- 2. Within six (6) months after completion and submission of the study, the Administrator, Environmental Protection Agency must, after public hearing and opportunity to comment, determine whether to promulgate regulations for the classification of drilling fluids, produced waters and other waste associated with the exploration, development and production of crude oil or natural gas as hazardous wastes [42 U.S.C. Section 6921(b)(2)(B)];
- 3. The Administrator, Environmental Protection Agency must then publish his decision in the Federal Register and must transmit that decision to both Houses of Congress together with any proposed regulations he deems appropriate [42 U.S.C. Section 6921(b)(2)(B) and (C)];
- 4. Any such regulations proposed by the Administrator, Environmental Protection Agency shall take effect only when authorized by Act of Congress [42 U.S.C. Section 6921(b)(2)(C)].

In this regard, it is our understanding that the Administrator, Environmental Protection Agency will submit a preliminary report to the committees referred to above not later than August 31, 1987. The detailed and comprehensive study will apparently be submitted as some later date.

By letter dated July 16, 1987, Mr. Rick Lopez requested that Petro-Thermo Corporation reassess its "proposed project giving particular attention to the new E.P.A. criteria regarding surface impoundment of hazardous wastes, ...". This request suggested that drilling fluids, produced waters and other waste associated with the exploration, development or production of crude oil or natural gas would soon be listed as hazardous wastes. We would be most appreciative if you could identify the source of this suggestion so that we might respond to your concerns in a meaningful manner.

In closing, we do not believe that the New Mexico State Land Office is being asked to allow a hazardous waste facility to be constructed on state lands or to assume any accompanying risks. Nevertheless, we welcome the opportunity to review and respond to any additional documents published under the auspices of the Administrator, Environmental Protection Agency which may have been made available to you. Mr. Nick Black August 5, 1987 Page 3

Again, your assistance in this regard will be sincerely appreciated. Very truly yours, John Paul Weber JPW:rm xc: Petro-Thermo Corporation

bxc: David Boyer, Oil Conservation Division P.O. Box 2088 Santa Fe, NM 87501-2088





P.O. BOX 2069 PHONES (505) 393-2417 — 397-3557 HOBBS, NEW MEXICO 88241-2069 July 22, 1987

Mr. Rick Lopez Planning and Development New Mexico State Land Office P.O. Box 2088 Santa Fe, New Mexico 87504-2088

Re: Application For Business Lease Petro-Thermo Corporation Oilfield Waste Disposal Facility

Dear Rick:

In reference to your letter, dated July 16, 1987, Petro-Thermo Corporation wishes to provide the New Mexico State Land Office with the following information:

We have been advised by the NMOCD and our Counsel that the EPA has not issued any regulations which classify the materials to be placed in the Petro-Thermo Corporation proposed disposal facility as hazardous waste. While there apparently have been some studies undertaken regarding this matter, the determination that oilfield liquids are hazardous waste remains at best, hypothetical. Petro-Thermo Corporation has made every effort to comply with the regulations and procedures promulgated by the State Agencies governing the oil and gas industry. Based on the Hydrogeologic Study and the Site Design Plan, the NMOCD has approved the installation of this oilfield waste disposal facility.

In the event a future determination is made designating oilfield waste as hazardous waste, it would have a dramatic detrimental effect on the oil and gas industry. Every existing waste disposal facility, drilling location, production facility, treating plant and petroleum storage facility would be shut down. The economic repercussions for the State of New Mexico would be enormous.

With regard to your request for determining a bond amount, the NMOCD currently provides regulations directed toward bonding requirements for treating plants. The NMOCD recently informed me that oilfield waste disposal sites are included in this category. Rule 312 (A) (6) states: A surety Mr. Rick Lopez Application for Business Lease Page 2

or cash bond in the amount of \$25,000, in a form approved by the Division, conditioned upon compliance with statutes of the State of New Mexico and Rules of the Division and the satisfactory clean-up of site upon cessation of operation in accordance with Part (i) of this Rule. Part (i) states: The Director of the Division may suspend any treating plant permit when it appears that such suspension is necessary to prevent waste, to protect fresh water, or to assure compliance with Division Rules or Orders. In addition, SLO Rule 1.016 (A) and (B) relating to oil and gas leases, requires a bond amount of not less than \$10,000 for each oilfield lease. In the event that a spill breached the boundary of the disposal facility, it is our position that clean-up costs would not exceed \$25,000, and would be a reasonable bond amount.

Finally, it is our general feeling that Petro-Thermo Corporation has negotiated an equitable lease arrangement with the New Mexico State Land Office. To date, our firm has invested in excess of \$40,000 in this disposal project. Based on current market conditions, it will take two to five years to recoup this amount.

I hope that this information will be helpful with regard to expediting the approval of our Application for a Business Lease. If you have any further questions, please do not hesitate to contact me.

Sincerely,

Petro-Thermo Corporation

norm.a.

Robert W. Abbott President

RWA/d1b

PETRO-THERMO CORPORATION

P.O. BOX 2069 PHONES (505) 393-2417 — 397-3557 HOBBS, NEW MEXICO 88241-2069

July 21, 1987

Honorable Bill Humphries Commissioner of Public Lands New Mexico State Land Office P.O. Box 2088 Santa Fe, New Mexico 87504-2088

Re: Application For Business Lease Petro-Thermo Corporation Plata Disposal Facility

Dear Mr. Humphries:

With regard to the above referenced Application for Business Lease, Petro-Thermo Corporation wishes to provide the following information in response to recent BLM comment submitted to the New Mexico State Land Office.

1111 2 2 1987

Reference is made to BLM concern for the two archaeological sites (NMAS 5839 and NMAS 5840). Page 11 of the NMAS Archaeological Clearance Report provided to Mr. Thomas W. Merlan, SHPO, "suggests and recommends clearance for Petro-Thermo Corporation's proposed Tract "B"/Plata Disposal Project. As proposed the core area will embrace the 660' x 660' southwestern quadrant of Tract "B". Recorded archaeological sites NMAS 5839 and NMAS 5840 are well removed from the actual core area of the project." It further suggests that "men and material are to avoid the archaeologically sensitive areas altogether during all phases of work." This can effectively be accomplished by simply staking the two sites and providing quality management by Petro-Thermo Corporation during construction and operation. While it is our position that expensive fencing is unnecessary, as an accommodation to the New Mexico State Land Office, our firm will surround both archaeological sites with barbed wire type fencing.

Reference is made to the BLM suggestion to require disposal pits to be lined. Numerous studies have been undertaken regarding this general area of New Mexico. These studies including those prepared for the BLM, Pollution Control Inc., Sandia National Laboratory, N.M. Bureau of Mines and Mineral Resources and Petro-Thermo Corporation have concluded that this area supports large-scale mining and disposal activities because of the absence of potable waters. In addition, the area encompassing Honorable Bill Humphries Application for Business Lease Page 2

the proposed disposal facility has been previously exempted by NMOCD Order Number R-3221. The pit impoundments as planned, are designed to infiltrate salt water (brine) and water contained in drilling mud and cement through the soil and migrate slowly to the Laguna Plata. The salinity of the seepage will not exceed that of Laguna Plata, inasmuch as produced oilfield fluids are expected to have total dissolved solids concentrations in the range of 25,000 to 75,000 ppm. The total dissolved solids at Laguna Plata is 335,100 ppm. Thus, the seepage will dilute the concentration of the total dissolved solids in Laguna Plata. The total annual rate of evaporation from Laguna Plata is about 5360 acrefeet per year. Under anticipated normal operating conditions, the total rate of Group I waste disposal will be only about 93 acre-feet per year. Therefore, there is ample storage and evaporation potential in Laguna Plata to accommodate the waste seepage. No significant change in the hydrologic regime is expected as a result of the proposed discharge. The BLM has previously approved National Potash Corporation to dispose of "potash slurry" directly into Laguna Plata Lake. The proposed discharge of the Petro-Thermo Corporation facility would be insignificant in comparison to the potash discharge. The BLM suggestion to require lined pits would effectively redefine the disposal facility as a storage facility. Without an infiltration system, each pit impoundment would reach capacity and become useless. The relatively small surface area of each pit would not be conducive to rapid evaporation rates. The pit bottoms could not be serviced and cleaned on a routine basis. The number of pits would grow rapidly requiring the storage facility to expand indefinitely. It is our position that the infiltration method is the only logical solution for long term waste disposal.

Reference is made to the BLM suggestion to require a second berm to be constructed surrounding the entire disposal area. The existing design plan for the disposal facility incorporates a unique gravity system, whereby as each successive pit reaches capacity, a large diameter overflow pipe channels fluids into the adjacent pit (refer to design plan). This method maximizes infiltration time before pit levels begin to rise. In addition, the disposal facility includes a large overflow pit (refer to design plan pit #01) incorporated into the plan to exclusively and effectively provide insurance against unexpected volumes of fluids entering the system, such as large area thunderstorms. A qualified employee will be present at the disposal facility during business hours. While it is our position that a second berm constructed to surround the entire disposal area is unnecessary, as an accommodation to the New Mexico State Land Office, our firm will construct a second berm down slope of the disposal facility.



Honorable Bill Humphries Application for Business Lease Page 3

Reference is made to the BLM concern for a salt mining operation near Laguna Plata Lake. As stated above, the Hydrogeologic Study recommending clearance for this project expects no significant change in the hydrologic regime of Laguna Plata Lake as a result of the proposed discharge. Based on this study, the NMOCD has dismissed concern about this salt mining operation. In addition, the salt mining company has <u>not</u> taken issue against the project. The basis for the BLM concern of this particular issue is unclear.

Reference is made to the BLM suggestion to construct a 6 foot chain link fence to enclose the disposal area. The Hydrogeologic Study recommending clearance for this project points out that this area is very sparsely populated. The dwellings which comprise Halfway, New Mexico are abandoned. Except for Halfway, the only dwelling within two miles of the proposed discharge site is a ranch on the east side of Laguna Plata. In addition, edible vegetation is very scarce in this area. Domestic animals have not been sighted by our firm near the proposed disposal area since the inception of the project. It is our recommendation that fencing for this disposal facility should be similar (barbed wire) to other oilfield locations and tank batteries in remote areas. NMOCD Order Number R-8161-B provides for the proposed facility to "have adequate fencing, gates and cattle guards installed to preclude livestock and unauthorized persons from entering the facility". It is our position that the construction of a 6 foot chain link fence would pose an unnecessary expense.

Reference is made to the BLM suggestion for the pit impoundments to be netted. Similar NMOCD regulated disposal facilities are not required to place netting over the pit impoundments. In addition, netting is not required for drilling location mud pits or production facility emergency pits. The placement of netting over the disposal pits at the Petro-Thermo Corporation facility would be extremely impractical making it virtually impossible to routinely service, skim and clean the pits. For this reason "pit netting" is not a provision of NMOCD Order Number R-8161-B recommending clearance for this project. It is our position that "pit netting" is unnecessary.

The oil and gas industry and the State of New Mexico are partners in the development, conservation and perpetuation of the State's energy resources. Jobs, incomes and taxes generated from this industry are vital to our State's economy. With regard to world energy policies currently being implemented, the management of Petro-Thermo Corporation expects the demand Honorable Bill Humphries Application for Business Lease Page 4

for oil and gas to drastically increase by the early 1990's. Future drilling, secondary and tertiary recovery projects will depend heavily on the existence of facilities such as the one proposed by our firm. Presently there are only two such NMOCD approved sites in Lea County, a major oil and gas producing county. It is important to realize that any unnecessary stipulations or conditions imposed on this proposed disposal facility could conceivably increase the project expenditures beyond current cost effective levels and market conditions.

Petro-Thermo Corporation respectfully requests that the New Mexico State Land Office carefully consider the above information. We hope that it will be helpful with regard to expediting the approval process of our Application for a Business Lease.

Sincerely,

Petro-Thermo Corporation

morm. and

Robert W. Abbott President

RWA/dlb

xc: Mr. William J. LeMay, Director NMOCD

Mr. Dave Boyer, OCD

IN REPLY REFER TO

United States Department of the Interior 1703 (931) 3040 8100 1703 (931) 3040 8100 JUL 13 1987 JUL 10 1987

Mr. William R. Humphries Commissioner State Land Office State of New Mexico P.O. Box 1147 Santa Fe, NM 87504-1147

Dear Commissioner Humphries:

Thank you for the opportunity to comment on your draft Business Lease No. BL-1186 between the State of New Mexico and Petro-Thermo Corporation in Lea County, New Mexico, near Laguna Plata. The Bureau of Land Management (BLM) continues to be concerned about this potential development and its impact on the surrounding BLM administered lands. Our greatest concerns are for the integrity of the Laguna Plata Archaeological District and potential releases of hazardous substances to ground or surface waters. We are also concerned about protecting the visual and wildlife resources of the area.

The Federal land surrounding Laguna Plata has been studied by the BLM for its prehistoric archaeological significance and was found to warrant nomination to the National Register of Historic Places. It is very likely that sites found on this State section will yield important information in conjunction with the overall Laguna Plata Archaeological District. This point of view is also held by the State Historic Preservation Officer, as evidenced by a June 15, 1987, letter to Ms. Zilla Padilla of your staff. In light of the apparent significance of the State land and the documented significance of the surrounding Federal land, development of this area should proceed only with great caution, if at all. The BLM has identified this State section as land desirable for acquisition for purposes of consolidating the archaeological resources.

Your lease describes an area containing 80 acres, yet the completed archaeological survey only covered 40 acres. The entire area should be inventoried. If this discrepancy is as described, we believe this is a critical omission for such a significant area. We suggest your stipulation on cultural resources be worded as follows:

18. Class III Archaeological Survey. Because the surrounding Federal land is being nominated to the National Register of Historic Places as an Historic District, a Class III inventory shall be conducted on the State land which will be directly or indirectly affected by this action. The two archaeological sites (NMAS 5839 and NMAS 5840) identified by the completed inventory are depicted on Map 1 (enclose map showing sites). These two properties shall be fenced and shall be completely avoided by the planned operations.

In the event that either of the sites are adversely affected by the operations, the affected property will be mitigated according to standards approved by the New Mexico State Historic Preservation Officer.

The disposal operation, as planned, is designed to release disposal liquids to the Laguna Plata Lake. Recent presentations by the New Mexico Environmental Improvement Division (EID) (Produced Water and Hydrocarbon Liquids Chemical Constituants of Hydrogeological Concern in New Mexico, 1986, Dennis McQuillan), and the New Mexico Oil Conservation Division (OCD) (Disposal of Produced Water and Oil Field Fluids, 1986, David C. Boyer) have reported that hazardous substances, as defined by the Comprehensive Environmental Response Compensation and Liability Act (CERCLA), typically occur in oil and gas drilling and production waste fluids. We are concerned that some of these substances will be released to the lake through groundwater or surface water routes. CERCLA carries liability for landowners such as the BLM and the State of New Mexico, as well as for operators. We realize there are confining beds beneath the disposal facility and the lake which should prevent deep seepage of wastes. However, lateral seepage above these beds toward the lake, and any attenuation of waste chemicals along this path, has not been defined. The rate volatile compounds will be volatilized, and the rate chemicals will be adsorbed onto soil particles have not been documented. The OCD's determination is based on a professional estimate backed by a safety net of monitoring wells. We have observed the movement of volatile organic compounds over greater distances than 1/4 mile in other areas.

The OCD has taken a position of proceeding with disposal and monitoring to ensure there is no off-site contamination. While monitoring wells are a necessary part of this type of disposal facility operation, it is a small consolation to catch pollutants in groundwater before they get to a more distant location. It only means you may be liable for a small fortune in cleanup costs, rather than a large fortune.

The only possible way to achieve an acceptable level of protection in this situation would be to require lined pits. According to OCD estimates (OCD Case No. 8781, Order No. R-8161, Finding No. 9), this would be feasible. In their analysis, they estimate that the evaporation from Laguna Plata is 60 times the volume of liquid wastes planned for disposal. It would be much more environmentally sound to put the wastes in lined pits and evaporate the liquids directly from the pits, instead of allowing the liquids to seep into the lake before evaporating. This would require a larger surface area, but the operation could be moved since the lake would no longer be needed as a disposal drain. The cost of lining the pits would be small compared to the costs of cleanup or the bond required to ensure cleanup. Another way to catch contaminants before they get very far would be to require samples of the disposal pits on a quarterly basis. This should be required for either lined or unlined pits.

To further minimize the risk, we suggest the following minimum requirements:

o Enclose the entire disposal area with a 6-foot chain link fence with a lockable gate at all entrances. Keep all gates locked during nonoperating hours.

o Have an attendant on duty at all times during operating hours.

o Keep a log of all disposals, including the volume, type, and source of disposed materials, as well as the name and address of the disposer and truck driver.

o Do not allow any disposals of hazardous wastes. Any release of hazardous substances shall be reported as required by CERCLA, Section 102(b).

The OCD refers to these site security measures generally by requiring adequate facilities to preclude unauthorized persons from entering (OCD Case No. 8781 De Novo, Order No. R-8161-B, (1)), but we have found that these items need to be spelled out in detail.

The OCD requirements depend upon monitoring wells to detect contaminant movement through groundwater, but do not require a backup for surface spills. If a pit were breached due to some accident or a large thunderstorm, the contents of the pit probably could not be stopped before they entered the lake. If a second berm was required to surround the entire disposal area, the risk of damage due to a surface spill could be significantly reduced.

The OCD record shows that they have dismissed concern about a sodium mining operation that uses water from the Laguna Plata Lake because, "No party representing the salt mining company appeared at either the examiner hearing or the De Novo hearing to object to the proposed disposal operation." (OCD Case No. 8781 <u>De Novo</u>, Order No. R-8161-A, Finding No. 28). The BLM did raise a concern for this use of the lake water, and we continue to be concerned. We have recently received an expression of continuing interest in this mineral lease operation (Enclosure 1). Please feel free to examine our records if you need documentation of the operation.

We are pleased to see the requirement for a suitable bond to cover restoration costs. The amount of the bond is not specified but should be enough to cover any potential contaminant cleanup costs.

As a multiple use management agency, the BLM is also committed to maintaining the visual and wildlife resources of this area. A requirement to keep facilities low profile and painted sandstone brown will reduce visual impacts. This area does contain threatened and endangered wildlife species. All BLM leases in this area require pits containing hydrocarbons to be netted to prevent wildlife losses.

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We hope you realize that OCD's responsibility is heavily influenced by the definition of fresh water being water with less than 10,000 parts per million total dissolved solids that has a present or reasonably forseeable beneficial use. Release to the environment as defined by CERCLA does not contain this restriction; therefore the lake and shallow groundwater would be considered the environment, regardless of its level of salinity or potential for beneficial use.

The OCD is a regulator in the business of protecting the higher quality water resources, and they have found that the use of these safety net features works well for their purposes. However, as landowner and land manager, the BLM has found that an ounce of prevention can save much more than a pound of cure. As you may be aware, we have had some experience with produced water disposal sites during the past few years, and we have put considerable thought into how we can protect public health and the environment, as well as protect the taxpayer from cleanup liability. We have identified these collective disposal operations as having a much higher risk than single source disposal pits. We feel it would be in the best interest of the State of New Mexico, as well as the BLM, to carefully consider the risks involved before committing to leasing land for this type of facility.

It is our opinion that this facility should have lined pits in order to reduce the risk of contamination to an acceptable level. If you agree with this, we request that you require lined pits and ask Petro-Thermo Corp. to move the proposed facility to other State land in the area to avoid impacts to the Laguna Plata Archaeological District.

Thank you again for your consideration of our concerns. Please notify the lessee that they will need to get a BLM right-of-way permit if access to the facility will be across public land. If you need any more information, please feel free to contact me or my staff.

> Sincerely, /s/LARRY L WOODARD

> > Larry L. Woodard State Director

1 Enclosure:

1 - Letter from Mississippi Chemical Corporation (2 pp) -4



File 1614.4 CRMP (NMO67)

May 14, 1986

hureau of Land Hanagement Arian Hanager Carlsbad Basource Area PO Box 1778 Carlsbad, New Naxico 86220

Dear Strat

After careful review of your draft, Resource Management Plan, we would like to commant on several aspects of the new plan. First, we do not even the need to change the current plan. We teel the current plan provides adequate protection of all resources within the area. In addition, we feel your new plan leans away from the long-standing policy the BLM has on multiple use of federal lands. It is obvious that the mew plan leans heavily towards recreation and away from mining and graning. If a new plan is developed, it should seek to achieve better belence.

If the plan were to be implemented in its current form. Hississippi Chemical has two areas of critical concern. These two areas of concern relate to the proposed special management areas, Maroon Cliffs and Lagma Plata. Our concerns are as follows: ***

Maroon Cliffs;

Heroon Cliffs special management area lies directly east of our current plant alte. The only map showing it's location is your management plan and plant alte. The only map showing it's location is your management plan and points. The only map showing it's location of the Haroon Cliffs area, points. This is of great concern to us since the plan calls for no surface occupancy. We feel you should re-examine the area to insure the your opecial management area doesn't contain land already disturbed by NCC. Further, we feel there should be a buffer around the east side of our plant to allow for future expansion of our tailings pond.

1 would like to remind you of the problems Maroon Cliffs have created for MCC and the BLM in the past. As you probably recall, MCC at one time wonted to build a new mine and mill in the Maroon Cliffs area. Due to burenucratic delays the necessary permits were never obtained. If MCC could have obtained the necessary permits were never obtained. If MCC could have botained the necessary permits, we would have built a new facility from which to mine and mill our one body. With this new facility we would have been able to mine at a higher rate and efficiency.

Box 101 - Carlsbad, New Mexico 88220 Phone (505) 887-5591

Page 1 of 2

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The boundaries for the Maroon Clifts SMA were based on the boundaries established in the tast Eddy-Lea Management Framework Plan for Maroon Cliffs. No legal descriptions or maps of the proposed Cultural Resource SMA's were published in the KMP because 36 CFK part 1215.20 prohibits public disclosure of this information except in very restricted circumstances. The Maroon Cliffs SMA boundary has been amended to exclude any areas presently disturbed by MCC's mining operation. A table of lease numbers that may be affected by the proposed cultural SMA's has been developed and is available for review at the Larisbad Resource Area.

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From this new facility we could have competed with an operation in the Carlabad area and probably would be running today generating royalties for the BLH.

disturbance. To aid you in evaluating the Maroon Cliffs area, please find enclosed a map showing our current facilities and approximate area of surface

Laguna Platat

10-2 In the Laguna Plata area, little mention is made of Williams Brine Co. which is mining salt under an agreement with National Putash or National's disposing of brine in Laguna Plata. We feel the plan should specifically address these two items. We would like the plan to state that it will allow Williams to continue to mine sait from Laguna Plata

and that he has the right to maintain his access. Also, we feel the plan should state National can continue to dispose of brine in Laguna Plate and have surface access to maintain it's pipeline to the lake.

We appreciate this opportunity to comment on this plan. Also, we hope before the plan is finalized, we might be allowed to review it again.

Sincerely.

MISSISSIPPI CHEMICAL CORPORATION

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Ś Tillman Branch General Manager almond Bround

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Enclosure

Page 2 Of 2

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William's Brine Co. has the right to mine salt and maintain access to the operation under the provisions of the existing lease. Also under the existing lease, Mational Potash has the right to dispose of the clear brine in Laguna Plata under existing slipulations. Surface access to the pipeline cannot be denied because the pipeline is on state of New Mexico land and because the right to do so is protected by National's existing lease rights.

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PETRO-THERMO CORPORATION

P.O. BOX 2069 PHONES (505) 393-2417 — 397-3557 HOBBS, NEW MEXICO 88241-2069

June 12, 1987

Honorable Bill Humphries Commissioner of Public Lands New Mexico State Land Office P.O. Box 2088 Santa Fe, New Mexico 87504-2088

Re: Plata Oilfield Waste Disposal

Dear Mr. Humphries:

I appreciated your telephone call June 9 regarding the approval of our Application for a Business Lease.

Per your request at our meeting May 11, we performed and submitted to Rick Lopez a favorable Archaeological Clearance Report recommending installation of the disposal site.

The many delays in this project have seriously impeded the daily business operations of our firm, and the oilfield customers we serve. As I mentioned, Petro-Thermo Corporation has made temporary storage plans until this Business Lease is approved.

In an effort to expedite this matter please do not hesitate to contact me for any further questions you may have.

Sincerely,

Petro-Thermo Corporation

1201 m. al

Robert W. Abbott Manager of Operations

RWA/d1b



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W.R. HUMPHRIES

Commissioner of Public Lands

P.O. BOX \$148 SANTA FE, NEW MEXICO 87504-1148 (505) 827-5713

April 10, 1987

MEMORANDUM

TO: Rick Lopez, Director Planning & Development Division THRU: Art Waskey, General Counsel

FROM: Nick Black, Associate Counsel

SUBJECT: PETRO-THERMO BUSINESS LEASE

Pursuant to our recent conversation, I have reviewed the proposed lease with Petro-Thermo and your file in this matter. Attached are proposed language changes to the standard Business Lease form based upon our conversation and the review. Most of the changes are peculiar to the Petro-Thermo deal.

You will note I have incorporated the provisions of Exhibits "A" and "B" into the body of the lease and have tightened the language somewhat, especially in the area of maintenance of the facility and reclamation at the end of the lease term.

The issue of whether the site is to be a final destination solid waste disposal site or only a temporary holding facility for solid waste was not particularly clear from anything I have reviewed, including some OCD hearing transcripts. Consequently, Section 11 of the lease, regarding maintenance, has been modified to keep the proposed use in line with what I understand Commissioner Humphries' view is regarding the use of State lands for waste disposal. It now requires periodic hauling out of the solid waste material so as to reduce the problem of solid waste accumulation at the end of the lease term.

While the hauling cost may not have been anticipated by Petro-Thermo, I am not sure the State fee estimates based on production water disposal are adequate to support permanent storage of the solid waste. Neither is it clear that reclamation Rick Lopez

is feasible if the waste material is allowed to remain permanently.

We also need to address the need of providing an access easement as the lease site is not contiguous with a public road.

Rather than set out the reasons for the language changes in a lengthy memorandum, it might be more helpful after your preliminary review of the proposed changes that we discuss the matter further. Let me know when you want to discuss this further.

Mark

Nick Black

attachment

2. RENT *** REWRITE section entirely as: ***

"A. Lessee covenants and agrees to pay to Lessor an annual rent in the following amounts in advance during the term of this lease:

Year	1:	\$96	0.0	00		
Year	2:	58	of	Adjusted	Gross	Income
Year	3:	5%	11	n	17	11
Year	4:	78	11	"	**	"
Year	5:	10%	, 11	11	11	11

B. Said percentage rent shall be paid annually, in advance. The first payment of Percentage Rent shall be based upon Lessee's Adjusted Gross Income for the prior year of operations. If, at the end of any Lease Year, the total amount of Percentage Rent paid by the Lessee during said Lease Year exceeds, or is less than, the amount of the Percentage Rent actually due from the Lessee for such year, Landlord either shall credit such excess to the next rent obligation of Lessee or Lessee shall add such deficiency to the payment of the next rent obligation. In the case of the final Lease Year of the Term of this Lease, such excess or deficiency shall be paid to or by the Lessee on or before the 15th day of June next following the last day of the final year of said Lease Term. Gross Income for any Lease Year shall be an amount equal to the gross revenues determined in accordance with generally accepted accounting principles consistently applied, derived by the Lessee from the operation of the subject premises.

Adjusted Gross Income shall mean Gross Income, less an annual operating expense deduction of \$48,000.00.

с. Lessee shall utilize, and cause to be utilized, an accounting system in accordance with good oil and gas production water and related solid waste treatment and disposal practice which will accurately record all Gross Income; and shall keep for at least thirty-six (36) months after expiration of each Lease Year records conforming to such accounting system showing all Gross Income for such Lease Year, including all tax reports, cash register records, sales slips, sales checks, bank deposit records and other supporting data. Within thirty (30) days after the end of each calendar quarter included in the Lease Term, Lessee shall furnish Lessor a statement certified by Lessee of Lessee's Gross Income during such quarter; and within thirty (30) days after the end of each Lease Year, Lessee shall furnish Lessor a statement certified by a Certified Public Accountant of Lessee's Gross Income during the preceding Lease Year. Lessor shall have the right from time to time by its accountants or representatives to audit all statements of Gross Income and in connection with such audits to examine all of Lessee's records (including supporting data and excise and income tax returns) of Gross Income. If any

such audit discloses a deficiency in the payment of Percentage Rent, Lessee shall forthwith pay to Lessor the amount of the deficiency together with Interest of 10% computed from the date when said payment should have been made. If any such audit discloses that the Gross Income by Lessee exceed those reported by more than three (3%) percent, Lessee shall pay the reasonable cost of such audit and examination.

D. If any Rent is received more than five (5) days after it is due, such shall bear Interest of 10% computed from such due date until the date of receipt of payment thereof by Lessor."

4. PERMITTED USE. *** REWRITE section entirely as: ***

"Lessee shall use the land granted in this lease only for the following purpose: treatment and disposal of water produced in conjunction with the production of oil or gas, or both and associated waste hydrocarbons and other related solids obtained in conjunction with the drilling and production of $^{OIC}_{\Lambda}$ and gas into separate unlined pits adjacent to Laguna Plata in the SWMSEMNEM of Section 16, Township 20 South, Range 32 East, NMPM, Lea County, New Mexico, in compliance with Oil Conservation Commission Order No. R-8161, dated February 13, 1986; Order No. R-8161-A, dated May 20, 1986; and Order No. R-8161-B dated October 23, 1986. No other use shall be permitted unless Lessor, in written form, has previously granted his consent in writing.

DRAFT--BLACK

5. IMPROVEMENTS AND RECLAMATION.

*** ADD after

"Necessary appurtenances" the following:

All as more particularly described in the above Division Orders and Lessee's evidence submitted in support thereof, the specifications of which must be approved by Lessor.

*** <u>REPLACE</u> the first clause of the third sentence with the following:

"Upon termination, cancellation or relinquishment of this lease,"

*** ADD the following at the end of that section:

"Upon execution of this Lease, Lessee shall post with the Commissioner a bond from a financial institution acceptable to the Commissioner in a form and an amount acceptable to the Commissioner to secure removal of any improvements and restoration of the land to its condition prior to the placement of Lessee's improvements. In the event Lessee fails to adequately restore the land within six (6) months following termination, cancellation, or relinquishment of the Lease the Commissioner shall proceed to remove any improvements it deems necessary and restore the land to such prior condition and utilize the proceeds of such bond to pay for such removal and restoration. This provision shall survive the termination, cancellation or relinquishment of this Lease.

Lessee shall construct the proposed improvements and commence operation of the facility within one (1) year from execution of this Lease, and continuously operate the facility for the full term of the lease."

7. ASSIGNMENT AND SUBLEASE.

*** ADD after the last sentence:

"Any assignment of this Lease or execution of a sublease of land shall not operate to relieve the Lessee from any obligation created herein."

8. HOLDING OVER. *** REWRITE this provision as follows:

"If, after termination, cancellation or relinquishment, Lessee occupies said land or any improvements remain thereon, or Lessee fails to restore the land to the condition prior to the placement of the improvements, Lessee shall pay as rent therefore at a daily rate of \$_____, until such occupancy ends, the improvements are removed and the land restored as provided herein."

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11. MAINTENANCE. *** INSERT the following at the end of the sentence:

", which shall include, but not be limited to:

A. Performance of regular on-going maintenance operation to prevent and remove the build-up of solids in the water disposal pits that might impede infiltration;

B. Periodic removal of accumulated solid wastes to an approved site off the land leased herein;

C. Protection of wildlife, including notification to Lessor of any wildlife loss and the taking of any steps adequate in the sole view of the Commissioner to prevent future wildlife loss;

D. An operable backhoe shall be maintained at the subject facility in order to repair any breach of the pit dikes; and any spills or breaches of dikes shall be reported to the Lessor immediately;

E. Device adequate to catch spillage at the unloading facility; and

F. Excavation of topsoil for the construction of the pits and stockpiling thereof for future restoration of the site, along with signage adequate to give notice of its future use."

12. RELINQUISHMENT.

*** ADD to the end of the first sentence: ***

", provided such relinquishment shall not be effective until Lessee has fully complied with the provisions of Section 9 herein."

15. COMPLIANCE WITH LAWS.

*** ADD the words:

"orders, judgments" after the word "ordinances" in line (2); and

*** REWRITE line 4 in its entirety to reads as:

... "been or may be enacted, promulgated, filed or entered in all matters and things affecting the..."

CONSULTANTS IN GROUND-WATER HYDROLOGY

SOCORRO, NEW MEXICO

UPDATE ON HYDROGEOLOGIC CONDITIONS NEAR LAGUNA PLATA,

LEA COUNTY, NEW MEXICO,

FOR

PETRO - THERMO CORPORATION

SEPTEMBER

and the second
EEFORE THE
OIL CONSERVATION COMMISSION
Santa No. Mary Merrico
Case No. 8781 Entrol No. 3
Submitted by
Hearing Data

1986

• GROUND-WATER CONTAMINATION • UNSATURATED ZONE INVESTIGATIONS • WATER SUPPLY DEVELOPMENT •



CONSULTANTS IN GROUND-WATER HYDROLOGY

• GROUND-WATER CONTAMINATION • UNSATURATED ZONE INVESTIGATIONS • WATER SUPPLY DEVELOPMENT •

UPDATE ON HYDROGEOLOGIC CONDITIONS NEAR LAGUNA PLATA, LEA COUNTY, NEW MEXICO, FOR PETRO-THERMO CORPORATION

PREPARED BY

DANIEL B. STEPHENS & ASSOCIATES, INC. P. O. BOX 740 SOCORRO, N.M. 87801

SEPTEMBER 1986

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Figure 2. Water Level Elevations

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Figure 3. Map Showing Cross-Section Locations

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DANIEL B. STEPHENS & ASSOCIATES, INC.

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Figure 4. Cross-Section B-B'



Figure 5. Cross-Section E-E'



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Figure 6. Ground Water Flow Towards Clayton Basin

DANIEL B. STEPHENS & ASSOCIATES, INC.

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. . APPENDIX 1 - WELL LOG



Log. No. 5295

Harris Carlos Contractor

NEW MEXICO SCHOOL OF MINES STATE BUREAU OF MINES AND MINERAL RESOURCES SOCORRO, NEW MEXICO

WELL LOG DIVISION

CASING RECORD	ELEVATION 3681	ምም የ
Diam., in Bottom	INITIAL DAILY PROD	UCTION:
	Open	bbls. Oi!
	Open	cu. ft. Gas
	Thg.	bbls, Oil
	Thg.	cu. ft. Gas

COUNTY Eddy
FIELD Wildcat
COMPANY Culbertson & Irwin, Inc.
LEASE State L Well No. 1
LOCATION (34) SW/4
SEC. 2 T. 205 . R. 31E
1650 feet from South line and
330 feet from West line of Section
COMMENCED 5-12-47
COMPLETED 6-12-47
ABANDONED
REMARKS:

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FORMATION	BOTTOM, FEET	FORMATION	BOTTOM, FEET
			· · · · · · · · · · · · · · · · · · ·
Sand	25	Lime and sand	2270
Sandy red shale	75	Lime	2315
Red sand & lime shells	105	Sand	2325
Sandy red shale	115	Lime	2365
Red sand & lime shells	135	Lime and sand	2385
Red sand, water @140-160	165	Lime	2420
Red Rock	280	Lime and sand	2440
Red sand	315	Lime	24.50
Red sand & red shale	420	Lime and sand	2460
-Red rock	550	Lime	2488
Red rock and gypsum	560	Porous lime 1/4 bailer	water pr.hr. 2492 T.D.
Red rock	665	(S.L	.M. 2492)
Hed sand & red shale	745	• -	
_Red rock	786		
Anhydrite	805		
Red sand & anhydrite	840		
Lime and anhydrite	875		
Anhydrite	925		
Lime	950		
Annyarite	970		
Annydrite & red sand	980		
ned sand	1035		
Salt and red sand	1070		
	1175		
Annyarite	1190		
Salt and enhadmite	1275		
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Salt and Potash	1395		داء
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Salt & FOLASH	1550		
Salt and Dotach	1600		
	1065		
Salt & notach	1000		
Salt.	1900		
Anhydrite	1925		
Salt	2025		
Salt & anhydrite	2070		
Anhydrite	2105		
Lime	2160		
Lime and anhydrite	2210		
Brown lime	2220	8	

APPENDIX 2 - WATER CHEMISTRY



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() mg/l	Total non-filter. residue (suspe (00530) Other: Other: Other: N A-H ₂ SO ₄ Nitrate-N + , Ni total (00630)	able onded) itrate-N tal (00610)		mg/l			8.8 21.8 500 - 0 577.0 7.6 5989 5,830	Units mg/I 2 mg/I 2 mg/I mg/I mg/I mg/I	2/18 2/20 2/18 3/13
□ Cnemical oxygen mg/l □ Total Kjeldahl-N mg/l mg/l □ Total organic carbon mg/l □ Other: mg/l mg/l □ Other: □ Other: □ Other: □ Other: □ Other: □ Other: □ Analyst Date Reported Reviewed by □ Laboratory remarks □ 2.6 % 0	Total non-filter: residue (suspe (00530) Other: Other: Other: Other: Other: Other: Nitrate-N + , Nitotal (00630) Ammonia-N to Total Kjeldahl-I	able inded) 		mg/l	CAlcium (00915) Magnesium (00925) Sodium (00930) Potassium (00935) Bicarbonate (00440) Chloride (00940) Chloride (00945) Sulfate (00945) Total filterable residue (dissolved) (70300) Other: F, A-H2 SO4 Nitrate-N + , Nitrate-I dissolved (00631) Ammonia-N dissolved (0531)		8.8 21.8 500,0 577.2 4263 5989 5,430	Units mg/l 2 mg/l _ mg/l _ mg/l _ mg/l _ mg/l _	2/15 2/20 2/18 3/13
Total organic carbon () <t< td=""><td>Total non-filter: residue (suspe (00530) Other: Other: Other: Other: Nitrate-N + , Ni total (00630) Ammonia-N to Total Kjeldahl-l ())</td><td>able onded) </td><td></td><td>mg/l</td><td>Image: Calcium (00915) Image: Calcium (00925) Image: Calcium (00930) Image: Calcium (00940) Image: Calcium (00940) Image: Calcium (00945) Image: Calcium (0094</td><td></td><td>8.8 21.8 500 - 0 577-6 - 263 5989 5,430</td><td>Units mg/l 2 mg/l 2 mg/l mg/l mg/l mg/l mg/l mg/l</td><td>2/18 2/20 2/18 3/13</td></t<>	Total non-filter: residue (suspe (00530) Other: Other: Other: Other: Nitrate-N + , Ni total (00630) Ammonia-N to Total Kjeldahl-l ())	able onded) 		mg/l	Image: Calcium (00915) Image: Calcium (00925) Image: Calcium (00930) Image: Calcium (00940) Image: Calcium (00940) Image: Calcium (00945) Image: Calcium (0094		8.8 21.8 500 - 0 577-6 - 263 5989 5,430	Units mg/l 2 mg/l 2 mg/l mg/l mg/l mg/l mg/l mg/l	2/18 2/20 2/18 3/13
() mg/l Other: Date Reported Reviewed by Other: Analyst Date Reported Reviewed by Laboratory remarks	Total non-filter: residue (suspe (00530) Other: Other: Other: Other: Other: Nitrate-N +, Nitotal (00630) Ammonia-N to Total Kjeldahl-I () Chemical oxyg demand (0034)	able onded) 		umho mg/l mg/l mg/l mg/l	CAlcium (00915) Magnesium (00925) Sodium (00930) Potassium (00935) Bicarbonate (00440) Chloride (00945) Chloride (00945) Sulfate (00945) Total filterable residue (dissolved) (70300) Other: F, A-H2 SO4 Nitrate-N + , Nitrate-I dissolved (00631) Ammonia-N dissolve (00608) Total Kjeldahl-N		8.8 21.8 500 - 0 577.0 5789 5,830	Units mg/l 2 mg/l 2 mg/l mg/l mg/l mg/l mg/l mg/l	2/18 2/20 2/18 3/13
Laboratory remarks	Total non-filter: residue (suspe (00530) Other: Other: Other: Other: Nitrate-N + , Nit total (00630) Ammonia-N to Total Kjeldahl-I () Chemical oxyg demand (0034	able inded) 		umho mg/l mg/l mg/l mg/l	Image: Calcium (00915) Image: Calcium (00925) Image: Calcium (00930) Image: Calcium (00940) Image: Calcium (00940) Image: Calcium (00945) Image: Calcium (0094		8.8 21.8 500 - 0 6 77.6 4 2 6 3 5 989 5,430	Units mg/l 2 mg/l 2 mg/l mg/l mg/l mg/l mg/l mg/l mg/l	2/18 2/20 2/18 3/13
Laboratory remarks	 Total non-filter: residue (suspe (00530) Other: Other: Other: Other: Nitrate-N + , Nit total (00630) Armonia-N to Total Kjeldahl-l () Chemical oxyc demand (0034 Total organic c () Other 	able onded) 		umho mg/l mg/l mg/l mg/l mg/l	Image: Calcium (00915) Image: Calcium (00925) Image: Calcium (00930) Image: Calcium (00940) Image: Calcium (00940) Image: Calcium (00945) Image: Calcium (0094		8.8 21.8 500 - 0 577-0 7.6 - 263 5989 5,430	Units mg/l _ mg/l _ mg/l _ mg/l _ mg/l _ mg/l _ mg/l _ mg/l _ mg/l _ mg/l _	2/18 2/20 2/18 3/13
Laboratory remarks	 Total non-filter: residue (suspe (00530) Other: Other: Other: Other: Nitrate-N + , Ni total (00630) Ammonia-N to Total Kjeldahl-I () Chemical oxyg demand (0034 Total organic c () Other: 	able onded) 		umho mg/l mg/l mg/l mg/l mg/l	Calcium (00915) Magnesium (00925) Sodium (00930) Potassium (00935) Bicarbonate (00440) Chloride (00940) Chloride (00945) Chloride (00945) Chloride (00945) Chloride (00945) Chloride (00945) Chloride (00945) Chloride (00945) Chloride (00945) Chloride (00945) Chloride (00945) Chloride (00945) Chloride (00945) Other: Nitrate-N + , Nitrate-N dissolved (00631) Ammonia-N dissolve (00608) Other: Analyst	/ &	8.8 21.8 50.0 + 0 577.0 5789 5,830	Units mg/l 2 mg/l 2 mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l 	2/14 2/20 2/16 3/13
	 Total non-filter: residue (suspe (00530) Other: Other: Other: Other: Other: Nitrate-N + , Ni total (00630) Ammonia-N to Total Kjeldahl-l () Chemical oxygedemand (0034) Total organic c () Other: Other: 	able inded) itrate-N tal (00610) N gen 0) arbon		Imbo mg/l mg/l mg/l mg/l mg/l	Image: Calcium (00915) Image: Calcium (00925) Image: Calcium (00930) Image: Calcium (00940) Image: Calcium (00940) Image: Calcium (00945) Image: Calcium (0094		8.8 21.8 500 - 0 26 77.2 7.6 4 2 6 3 5 989 5, 430 5, 430 5, 430	Units mg/l 2 mg/l 2 mg/l mg/l mg/l mg/l mg/l mg/l mg/l 	2/18 2/20 2/18 3/13
	 Total non-filter: residue (suspe (00530) Other: Other: Other: Other: Nitrate-N +, Ni total (00630) Ammonia-N to Total Kjeldahl-I () Chemical oxyg demand (0034 Total organic c () Other: Other: Other: 	able onded) 		umho mg/l mg/l mg/l mg/l mg/l	Image: Calcium (00915) Image: Calcium (00925) Image: Calcium (00930) Image: Calcium (00935) Image: Calcium (00935) Image: Calcium (00940) Image: Calcium (00940) Image: Calcium (00945) Image: Calcium (00603) Image: Calcium (0060	 	8.8 21.8 50.0 - 0 5989 5,830 5,830 5,830	Units mg/l _ mg/l _	2/18 2/20 2/18 3/13 3/13
	 Total non-filter: residue (suspe (00530) Other: Other: Other: Other: Nitrate-N + , Ni total (00630) Ammonia-N to Total Kjeldahl-I () Chemical oxyg demand (0034 Total organic c () Other: Other: Other: 	able inded) 		Imbo mg/l mg/l mg/l mg/l mg/l	Image: Calcium (00915) Image: Calcium (00925) Image: Calcium (00930) Image: Calcium (00940) Image: Calcium (00608) Image: Calcium (0060		8.8 21.8 50.0 + 0 5 77.0 5 789 5 430 5 430	Units mg/l 2 mg/l 2 mg/l mg/l mg/l mg/l mg/l mg/l mg/l 	2/15 2/20 2/16 3/13
	 Total non-filter: residue (suspe (00530) Other: Other: Other: Other: Other: Nitrate-N + , Ni total (00630) Armonia-N to Total Kjeldahl-I () Chemical oxyc demand (0034 Total organic c () Other: Other: Other: Other: Other: 	able inded)		Imho mg/l mg/l mg/l mg/l mg/l			8.8 21.8 50 - 0 6 7.6 7.6 7.6 7.6 7.6 7.6 7.6 7.6 7.6 7	Units mg/l 2 mg/l 2 mg/l mg/l mg/l mg/l mg/l mg/l Review	2/18 2/20 2/18 3/13

726 (12/84)

1 1 [·] • • • • •	Albuquerque, NA	A 87106 - (505) 841-2555 NN	, and	d NITROGEN ANALYSIS	
ECEIVED 2	10 86 1	AB WC 566 USER 5930	о 🗆 59600 🕅 отне	ER: 82235	
LIDZIO4	-		util seen		
Hection TIME	1	ATION			
illected by Person//	Agency RANC		Sample Fre	m South See M	NE
	ENVIDONMEN.			AAR 14 of NE 14 So	A
	NM OIL CON State Land	SERVATION DIVISION Office Bldg, PO Box 208	8	7205, R32En	1
)	Santa Fe,	NM 87501	·	ansayo	
Attn:	David_Bo	yer		<u>_</u>	
			Star	tion/	
			Ow	ner	
Bailed		Water level	Discharge	Sample type	
H (00400)		Conductivity (Uncorrected)	Water Temp, (00010)	Conductivity at 25°C (00094)	
. ,		μmho		°C	μmh
eld comments	Ivate	, sepac forme	Amale pond/	puddle at base	
		array cut. in.	ter reging the	mthin Sond Time lens	'et-
	TREATMEN	T Check proper boxes			
No, of samples	/ XNI	F: Whole sample (Non-filtered) F: Filtered in 0.45 µme	field with A: 2 ml	H ₂ SO ₄ /L added	
NA: No ac	id added 🛛 (Other-specify:			
NALYTICAL F	RESULTS from	Units Date analyze	dif. NA	Units Date ana	lyzed
Conductivity (Corrected)		Calcium (00915)	776.0 mall -2-10	
25°C (00095)		μmho	- Magnesium (00925)	/507mg/l	
Total non-filter	abia		-2 Sodium (00930)	mg/l	
residue (suspe	anded)		Dr. Potassium (00935)	allert moll i	
100530			Bicerbonete (00440)	42218 mall 2/16	
(00530)		mg/i	Bicarbonate (00440)	423,8 mg/1 2/16 15276 mg/1 2/2	0
(00530)] Other:] Other:		mg/i	Bicarbonate (00440) Chloride (00940) Sulfate (00945)	$\begin{array}{c c} & $$ $$ $$ $$ $$ $$ $$ $$ $$ $$ $$ $$ $$	0
(00530) Other: Other: Other:		mg/l	Bicarbonate (00440) Chloride (00940) Sulfate (00945) Total filterable residue (dissolved) (70300)	$\begin{array}{c c} & 4 & 2.2.6 \\ \hline 15 & 2.76 \\ \hline 15 & 2.76 \\ \hline 165 & 7 \\ \hline 1/65 & 7 \\ \hline 3642 & 8 \\ \hline 3642 & 8 \\ \hline 1 \\ 1 \\$	<i>o</i>
(00530) Other: Other: Other:		mg/i	Bicarbonate (00440) Chloride (00940) Sulfate (00945) Total filterable residue (dissolved) (70300)	$\begin{array}{c c} 4 & 2.2.6 \\ \hline 15 & 2.76 \\ \hline 165 & mg/l \\ \hline 1/65 & mg/l \\ \hline 2/16 \\ \hline 3642 & mg/l \\ \hline 3642 & mg/l \\ \hline 3/12 \\ \hline \end{array}$	0 5
(00530)] Other:] Other:] Other: F, A-H₂SO	iirate.N	mg/i	Bicarbonate (00440) Chloride (00940) Sulfate (00945) Chloride (00945) Chloride (iterable residue (dissolved) (70300) Chlor: F. A-H, SQA	$\begin{array}{c c} 4 & 2.368 \\ \hline 15 & 2.76 \\ \hline 165 & 7 \\ \hline 1/65 & 7 \\ \hline 3642 \\ \hline \\ $	8
(UUS30)] Other:] Other:] Other: F , A -H ₂ SO ₄] Nitrate-N + , Ni total (00630)	itrate-N	mg/i	Bicarbonate (00440) Chloride (00940) Sulfate (00945) Chloride (00945) Chloride (iterable residue (dissolved) (70300) Chler: F, A-H ₂ SO ₄	$\begin{array}{c c} 4 & 2.86 \\ \hline 15 & 2.76 \\ \hline 165 & 7 \\ \hline 1/65 & 7 \\ \hline 3642\% \\ \hline mg/l \\ \hline 3/12 \\ \hline 3/12 \\ \hline \end{array}$	5
(00530)] Other:] Other:] Other: IF, A-H ₂ SO ₄] Nitrate-N + , Nit total (00630)] Ammonia-N to	itrate-N	mg/i	Bicarbonate (00440) Chloride (00940) Sulfate (00945) Total filterable residue (dissolved) (70300) Other: F. A-H ₂ SO ₄ Nitrate-N +, Nitrate-N dissolved (00631)	$\begin{array}{c c} 4 & 2.2.68 \\ \hline 15 & 2.76 \\ \hline 165 & 7 \\ \hline 1/65 & 7 \\ \hline 3642 & 8 \\ \hline 110 \\ \hline 11$	8
(UUS30)] Other:] Other:] Other: IF, A-H₂SO IF, A-H₂SO IF, A-	itrate-N tal (00610)	mg/i	Bicarbonate (00440) Chloride (00940) Suifate (00945) Total filterable residue (dissolved) (70300) Other: F, A-H ₂ SO ₄ Nitrate-N +, Nitrate-N dissolved (00631) Ammonia-N dissolved	$\begin{array}{c c} & 4 & 2.3.68 \\ \hline & 15 & 2.76 \\ \hline & 15 & 2.76 \\ \hline & mg/l \\ \hline & 2.1/8 \\ \hline & mg/l \\ \hline & 2.1/8 \\ \hline & mg/l \\ \hline & \\ \hline \\ \hline$	3
(00530)] Other:] Other:] Other:] Other: [, A-H ₂ SO ₄] Nitrate-N +, Ni total (00630)] Ammonia-N to] Total Kjeldahi-I ()] Chemical oxyg	itrate-N tital (00610)	mg/i	Bicarbonate (00440) Chloride (00940) Sulfate (00945) Total filterable residue (dissolved) (70300) Other: F, A-H ₂ SO ₄ Nitrate-N +, Nitrate-N dissolved (00631) Armonia-N dissolved (00608) Total Kie/dabLN	<u>42288</u> <u>15276</u> <u>1657</u> <u>1/657</u> <u>36428</u> <u>mg/1</u> <u>3/12</u> <u>mg/1</u> <u>mg/1</u> <u>mg/1</u>	8
(UUS30)] Other:] Other:] Other:] Other: [K, A-H ₂ SO ₄] Nitrate-N +, Ni total (00630)] Ammonia-N to] Total Kjeldahi-I ()] Chemical oxyg demand (0032)] Total organic c	itrate-N	mg/i	Bicarbonate (00440) Chloride (00940) Sulfate (00945) Total filterable residue (dissolved) (70300) Other: Image: state in the stat	<u>4228</u> <u>15276</u> <u>1657</u> <u>1/657</u> <u>36428</u> <u>mg/1</u> <u>36428</u> <u>mg/1</u> <u>mg/1</u> <u>mg/1</u> <u>mg/1</u>	<i>o</i>
(UUS30) Other: Other: Other: Nitrate-N+, Ni total (00630) Ammonia-N to Total Kjeldahi-I () Chemical oxyg demand (0034 Total organic c ()	itrate-N mai (00610) N gen i0) arbon	mg/i	Bicarbonate (00440) Chloride (00940) Sulfate (00945) Total filterable residue (dissolved) (70300) Other: F. A-H ₂ SO ₄ Nitrate-N +, Nitrate-N dissolved (00631) Armonia-N dissolved (00608) Total Kjeldahl-N () Other:	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	<i>e</i>
(UUS30) Other: Other: Other: Nitrate-N + Ni total (00630) Ammonia-N to Total Kjeldani-I () Chemical oxyg demand (0034 Total organic c () Other:	itrate-N 	mg/i	Bicarbonate (00440) Chloride (00940) Sulfate (00945) Total filterable residue (dissolved) (70300) Other: F. A-H ₂ SO ₄ Anrmonia-N dissolved (00631) Anrmonia-N dissolved (00631) Total Kjeldahl-N () Other:	<u>42288</u> mg/l <u>2/16</u> <u>15276</u> mg/l <u>2/18</u> <u>1/657</u> mg/l <u>2/18</u> <u>36428</u> mg/l <u>3/12</u> <u>mg/l</u> <u></u> <u>mg/l</u> <u></u> <u>mg/l</u> <u></u> <u>mg/l</u> <u></u>	3
(UUS30) Other: Other: Other: IF, A-H ₂ SO ₄ Nitrate-N +, Ni total (00630) Ammonia-N to Total Kjeldahi-I () Chemical oxyg demand (0034 Total organic c () Other: Other:	itrate-N	mg/i	Bicarbonate (00440) Chloride (00940) Sulfate (00945) Total filterable residue (dissolved) (70300) Other: F, A-H ₂ SO ₄ Nitrate-N +, Nitrate-N dissolved (00631) Ammonia-N dissolved (00608) Total Kjeldahl-N (Other:	<u>4 2 2 8 mg/l 2/16</u> <u>15 2 7 6 mg/l 2/18</u> <u>1/ 65 7 mg/l 2/18</u> <u>36 4 2 8 mg/l 3/1 2</u> <u>mg/l 3/1 2</u> <u>mg/l 1</u> <u>mg/l 1</u> <u>mg/</u>	3
(UUS30)] Other:] Other:] Other:] Other:] Nitrate-N +, Ni total (00630)] Ammonia-N to] Total Kjeldahi-I ()] Chemical oxyg demand (0034)] Total organic c ()] Other:] Other:	itrate-N tal (00610) N gen io) sarbon ks	mg/i	Bicarbonate (00440) Chloride (00940) Suifate (00945) Total filterable residue (dissolved) (70300) Other: F, A-H ₂ SO ₄ Nitrate-N + , Nitrate-N dissolved (00631) Ammonia-N dissolved (00608) Total Kjeidahl-N (Other:	<u>4 2 2 8 mg/l 2/16</u> <u>15 2 7 6 mg/l 2/18</u> <u>1/ 65 7 mg/l 2/18</u> <u>36 4 2 8 mg/l 3/1 2</u> <u>mg/l 3/1 2</u> <u>mg/l 1</u> <u>mg/l 1</u> <u>mg/l 1</u> <u>mg/l 1</u> <u>mg/l 2</u> <u>mg/l 2</u> <u>mg/</u>	<i>o</i>
(UUSJU)] Other:] Other:] Other:] Other:] Nitrate-N +, Ni total (00630)] Ammonia-N to] Total Kjeldahl-I ()] Chemical oxyg demand (0034)] Total organic c ()] Other:] Other:] Other:	itrate-N tal (00610) N gen i0) sarbon ks	mg/i	Bicarbonate (00440) Chloride (00940) Suifate (00945) Total filterable residue (dissolved) (70300) Other: F, A-H ₂ SO ₄ Nitrate-N +, Nitrate-N dissolved (00631) Ammonia-N dissolved (00608) Total Kjeldahl-N () Other:	<u>4228</u> <u>15276</u> <u>mg/l</u> <u>2//5</u> <u>mg/l</u> <u>2//2</u> <u>36428</u> <u>mg/l</u> <u>36428</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l}</u>	<i>o</i>
(UUS30)] Other:] Other:] Other:] Other: [F, A-H ₂ SO ₄] Nitrate-N + , Ni total (00630)] Armmonia-N to] Total Kjeldahl-I ()] Chemical oxyg demand (0034] Total organic c ()] Other:] Other:] Other:	itrate-N tal (00610) N gen i0) sarbon ks	mg/i	Bicarbonate (00440) Chloride (00940) Sulfate (00945) Chloride (00945) Chloride (00945) Chloride (00945) Chloricable residue (dissolved) (70300) Other: F. A-H ₂ SO ₄ Chloride (00631) Ammonia-N dissolved (00608) Chloride (00608	<u>4228</u> <u>15276</u> <u>mg/l</u> <u>2//5</u> <u>mg/l</u> <u>2//2</u> <u>36428</u> <u>mg/l</u> <u>36428</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>mg/l</u> <u>32486</u> <u>Reviewed by</u> <u>32486</u> <u>Contern</u>	3

Lab Number: HM1141 Date Submitted: 6/20/86 By: Easp

Sample Code: Jaguna Plata Seep Date Analyzed: 6 30 86 ashly Λ Reviewed By:_ Date Reported

Element	ICAP VALUE (MG/L)	<u>AA VALUE (MG/L)</u>	
Aluminum	D.4	· ·	
Barium	0.2	······	
Berylium	20.1		
Boron	2.1		
Cadmium	20.1		
Calcium	890.		
Chromium			
Cobalt	201		
Copper	201	*	
Iron	0.3		
Lead	40.1		
Magnesium	<u>870.</u>		
Manganese	0,31		
Molybdenum	40.1		
Nickel	<0.1	,	
Silicon	4.5		
Silver	<0.1		
Strontium	16		
Tin			
Vanadium	20.1		
Zinc	. ~0.1		
Arsenic		0.045	
Selenium			
Mercury		~0.0005	

ENVIRONMENTAL BUREAU SEND NM OIL CONSERVATION DIVISION FINAL State Land Office Bldg, P0 Box 2088 REPORT Santa Fe, NM 87504-2088 Attn:David Boyer Phone: 827-5312 SAMPLING CONDITIONS Bailed □ Pump Water level Discharge PH (00400) Conductivity (Uncorrected) Water Temp. (00010) Field comments Field comments SAMPLE FIELD TREATMENT Check proper boxes No. of samples / @NF: Whole sample □ F: Filtered in field with submitted / @NF: Whole sample □ F: Filtered in field with Conductivity (Non-filtered) □ F: Sall conc. HNO ₃ added ANALYTICAL RESULTS from SAMPLES No. of added □ Other-specify: □ A: Sml conc. HNO ₃ added ANALYTICAL RESULTS from SAMPLES Nagesium (00915) □ Conductivity (Corrected) □ Calcium (00915) □ Sodium (00930)	alion/ Hicode Amer Sample type Conductivity at 25°C (00094) µmho
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SEND FINAL REPORT TO Attn:	ENVIRONMENT NM OIL CONS State Land Santa Fe, M David_Boy	AL BUREAU SERVATION DIV Office B1dg, NM 87504-208 Ver	/ISION , PO Box 208 B	8	Station/ well code	<i>JNA</i>	
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FOR OCD USE -- Date Owner Notified Phone or Letter?

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STATE OF NEW MEXICO		
MEMORANDUM (OF MEETING OR CON	IVERSATION
Telephone Personal Time		Date
	830	9/16/86
Originating Party		Other Parties
R. ANDERSON -OCT) DAL	WILLIAMS
Subject	WIL	LIAMS BRIDE SUC - 885-6514
MINING OF SALT FROM	LAGUNA P	LATA - HOW + WHERE
USED?		
Discussion	· · · · · · · · · · · · · · · · · · ·	
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, 5 12 Na, 304 4 1. 181 K	<u> </u>	
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Conclusions or Agreements		
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OIL CONSERVATION DIVISION Telephone Personal	Time	Date alu lui
Originating Part		Other Parties
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		BRIAN Collier - EPNG
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Signed ZCAnden

STATE OF NEW MEXICO OIL CONSERVATION DIVISION	G OR CONVERSATION
Telephone Personal Time	Date 9/16/86
Originating Party	Other Parties
R. ANDERSON - OCD	EVERETT SEWELL
Subject	CULLIGAN - 885-4044
EFFECT OF HYDROCARBON	IS ON THEIR RESINS
<u>THEY RECEIVE</u> <u>SALT</u> <u>E</u> <u>FORM, REPESOLVE</u> , <u>D</u> , <u>A</u> <u>PASS IT THROUGH THREE</u> <u>USED ONLY</u> <u>IN</u> <u>INDUSTRIAL</u> <u>POOR</u> <u>QUALITY</u> <u>IE ANY</u> <u>IN THE DRY SALT</u> THEY <u>N</u> <u>CONCENTRATIONS</u> <u>AND</u> <u>THEIR</u> <u>VOLITIZE</u> THEM. <u>DISOLVED</u> <u>LOW</u> <u>CONCENTRATIONS</u> <u>WI</u> <u>ADVERSE AFFECT</u> <u>ON</u> <u>TH</u>	ROM WILLIAMS IN SOLID LLOW IT TO SETTLE AND FILTERS. THIS SOL'N IS APPLICATION, DUE TO ITS NYDROCARBONS ARE PRESENT OULD BE IN NERY LOW PRODESS WOULD PROBABLY HYDROCARBONS IN VERY LL PROBABLY HAVE NO EIR SYSTEMS
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STATE OF NEW MEXICO OIL CONSERVATION DIVISION	ORANDUM OF MEETIN	IG OR CONV	ZERSATION	· · · ·
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Originating Par	<u>·ty</u>		Other Parties	
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Conclusions or Agreements				
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Signed	~ 1
K.	anda

PETRO-THERMO CORPORATION

P.O. BOX 2069 PHONES (505) 393-2417 — 397-3557 HOBBS, NEW MEXICO 88241-2069

August 18, 1986

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RECEIVED

AUG 1 9 1986

OIL CONSERVATION DIVISION

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Mr. R. L. Stamets, Director Oil Conservation Division Energy and Minerals Department State of New Mexico Post Office Box 2088 Santa Fe, New Mexico 87501-2088

Re: Rescinding of Authorization for Temporary Disposal Pit

Dear Mr. Stamets:

Reference is made to your August 12, 1986 letter rescinding temporary authority for the disposal use of the earthen pit located at the Blinebry-Drinkard Salt Water Disposal System, Well No. A-22, located in Section 22, Township 22 South, Range 37 East.

As you are aware, Petro-Thermo Corporation originally made application for a hearing before the Oil Conservation Commission to consider it's request for an exception to Order No. R-3221 November 19, 1985. In response to comprehensive testimony presented at two hearings, the Commission entered Order Numbers R-8161 and R-8161-A. In view of these Orders, it is apparent that the Oil Conservation Commission fully recognizes the importance of additional approved oilfield related waste disposal sites.

Presently, because of competitive reasons, Petro-Thermo Corporation has been effectively cut off from disposing at Parabo Inc. operated by Unichem International (Rowland Trucking) and Pollution Control, Inc. (General Petroleum) operated by Mr. Larry Squires. In addition, the New Mexico State Land Office has not yet acted on our December 6, 1985 Application for a Business Lease covering the Eastern one-half Section 16, Township 20 South, Range 32 East, N.M.P.M., Lea County, New Mexico.

In an attempt to comply with your request to remove all solid and liquid waste from our temporary disposal pit, I am forwarding today a proposal to Mr. Jim Baca requesting the New Mexico State Land Office expedite approval of our application for a Business Lease so that we may begin August 18, 1986 R.L. Stamets Director, OCD Rescinding Authorization Petro-Thermo Corp.

construction on the first series of earthen pits at Plata Disposal in accordance with the Disposal Site Plan previously submitted.

Petro-Thermo Corporation solicits the help of the Commission in overcoming the aforementioned obstacles to facilitate resolving this matter expeditiously. In this regard it is our desire to cooperate fully.

Sincerely,

Petro-Thermo Corporation

nAm. all #

Robert W. Abbott Vice President

RWA/aj

xc: Mr. Jim Baca Mr. John Weber Mr. Jerry Sexton Mr. Ernest Padilla Page 2



ENERGY AND MINERALS DEPARTMENT

OIL CONSERVATION DIVISION



TONEY ANAYA GOVERNOR

August 12, 1986

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87501-2088 (505) 827-5800

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CERTIFIED MAIL RETURN RECEIPT REQUESTED

Mr. Robert W. Abbott Petro-Thermo Corporation P. O. Box 2069 Hobbs, New Mexico 88241-2069

RE: AUTHORIZATION FOR TEMPORARY DISPOSAL IN PIT IN 22-T22S-R37E

Dear Mr. Abbott:

Authorization for the temporary disposal use of the earthen pit located at the Blinebry-Drinkard SWD System Well No. A-22 located in 22-T22S-R37E is hereby rescinded.

This pit was authorized for the temporary disposal of brine, fresh water-based drilling mud, and waste cement only. You were notified that authorization would be rescinded if there was evidence that the pit overflowed or if Water Quality Control Commission or Oil Conservation Division rules or regulations were violated.

OCD personnel have inspected the pit and obtained documentation that the pit overflowed in violation of the authorization. Disposal or storage of any fluids or solids into this pit will immediately cease. All solids and liquids will be removed to an OCD-approved disposal location within two (2) weeks, or by August 25, 1986.

Sincerely,

Jujon

Jor R. L. STAMETS Director

RLS:JB:dp

cc: OCD - Hobbs

STATE OF NEW MEXICO ENERGY AND MINERALS DEPARTMENT OIL CONSERVATION DIVISION **INBBS DISTRICT OFFICE** August 11, 1986 POST OFFICE BOX 1980 TONEY ANAYA HOBBS, NEW MEXICO 88240 GOVERNOR aug 1 2 1986 (505) 393-6161 **CONSERVATION DIVISION** SANTA FE Mr. R. L. Stamets Director Oil Conservation Division P.O. Box 2088 Santa Fe, New Mexico 87504-2088 SUBJECT: Agua, Inc. Clean out pit at A-22 SWD

Dear Mr. Stamets:

Over a period of approximately two weeks, Agua has had a number of vent boot run-overs on their Eunice Blinebry/Drinkard SWD system. They are working on the system, trying to clear up the problem. Water and oil which is being picked up at runovers are being hauled into cleanout pit. Out field inspector did witness vacuum truck dumping into pit, also the pit had been run over from excessive dumping.

On 8-6-86, Eddie Seay made inspection of area, found pools of salt water still on location from runovers, pit has approximately 2 feet of water, mud, and oil in it. We were not aware of what permission was granted for the use of pit, but would recommend that Agua be restricted from using such pit.

Pictures and Inspectors trip reports will accompany this letter.

Please advise us as to your decision.

Very truly yours

OIL CONSERVATION DIVISION

Jerry Sexton

Supervisor, District I

JS:bp

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H	U (E)	С	4		5 other - Caprock area Gre Caprock Queen are Mr. Caudill. Fou Tr. 44-G 34-13-3 this time. Leak covering 15 x 50 Also on General O oil. Company man	at Western and General Op a - checking leaks turned nd small leak on Great We 4. Approx. 10 to 15 bbls at Tr. 26 #10, leak fixed area. No major leaks wer perating Tract 37 #3 leak was notified.	perating d in by Rancher estern Rock Queen s - Not leaking at d. Approx. 5 - 10 bls re found as reported. c with approx. 2 bls
Н	U	S	4		l other - Eunice area Aqu Emergency cleanou location. Sample Santa Fe for acti	a SWD A-22 t pit had been used and m taken and pictures taken on.	run over onto n - will refer to
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ENERGY AND MINERALS DEPARTMENT

STATE OF NEW MEXICO

TONEY ANAYA GOVERNOR POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87501-2088 (505) 827-5800

June 20, 1986

Mr. Thomas Kellahin Kellahin & Kellahin Attorneys at Law P. O. Box 2265 Santa Fe, N.M. 87504-2265

Dear Mr. Kellahin:

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This letter will confirm the Commission's action on June 19, 1986, relative to the request for rehearing in Case No. 8781 De Novo filed by Snyder Ranches, et al., June 9, 1986.

This Case will be reopened on August 7, 1986, and additional testimony will be accepted in all or part of the Grounds for Rehearing set out in your application dated June 9, 1986, as set out below:

- 5. Additional evidence will be heard relative to the issues raised in this paragraph.
- 6. Additional evidence wil be heard relative to the issues raised in this paragraph.
- 7. The Commission will accept briefs by August 4, 1986, on the general issues raised in this paragraph and will accept testimony relative to the allegation that "migration of contaminated waste water will destroy the grazing grasses and vegetation under the ownership and control of Snyder Ranches, Inc." Based upon the briefs filed, the Commission may or may not choose to accept testimony as to any other issue raised in this paragraph.

The rehearing will be limited to the above-described issues only. Snyder Ranches et al., as applicant for the rehearing, will be expected to present their case first. This case will be scheduled first on the docket, and in keeping with current practice, the Commission would hope to make our decision immediately at the conclusion of the rehearing.

Exhibite

Page 2

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If there are any questions on this matter, please do not hesitate to call me.

Sincerely, aline R. L. STAMETS Director

RLS:dp

cc: Jerry Sexton Fran Cherry, BLM

	New Mexico Hea SCIENTIFIC LAE 700 Camino de S Albuquerque, NM	Ith and Enversional Enversional Enversion Items SORATORY Stores SION Salud NE 1/87106 — (505) 841-2	Department N 2555	859 999	GE. ar		VATER CH OGEN AN	EMISTR' ALYSIS	Y
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Collected by - Person/A	AILEY/SO	EAY /OCD		SPR1,		SW	ARROY) 01 R 0-	> F
SEND FINAL REPORT TO ► Attn:	ENVIRONMEN NM OIL CONS State Land Santa Fe, I David Bo;	TAL BUREAU SERVATION DIV Office B1dg NM 87504-208 yer	VISION , PO Box 2088 8	3		ation/			
Phor	ne: 827-58	312			we Or	nil code wner	<u> </u>		
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SAMPLE FIELD	TREATMEN	T — Check prope	er boxes				. <u> </u>		
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Lab Number:	HM1141
Date Submitted:	6/20/86
By: Carp	

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Sample Code: Jaguna Plata Seep Date Analyzed: 6 30 86 Reviewed By: Jin Ashly Date Reported: 8/11/86

Element	ICAP VALUE (MG/L)	AA VALUE (MG/L)	
Aluminum	D.4		
Barium	0.2		
Berylium	20.1		
Boron	2.1		
Cadmium	20	1	
Calcium	890.		
Chromium	20.1		
Cobalt	401		
Copper	40.1		
Iron	0.3		
Lead	201		
Magnesium	870.	·	
Manganese	0,31		
Molybdenum	40.1		
Nickel	<0.1		
Silicon	4.5		
Silver	<0.1	·	
Strontium	16	<u> </u>	
Tin			
Vanadium	20,1	····	
Zinc	~0.1	· · · · · · · · · · · · · · · · · · ·	
Arsenic		0.095	
Selenium			
Mercury		<0.0005	

ENVIRONMENT	New México Heal SCIENTIFIC LAB 700 Camino de S Albuquerque, NM	Ith and Environment ORATORY DIVISION alud NE 1 87106 (505)	Department N 2555 NN	GE		VATER CH OGEN AN	IEMIST	'RY S
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Jason Kellahin W. Thomas Kellahin Karen Aubrey

KELLAHIN and KELLAHIN Attorneys at Law El Patio - 117 North Guadalupe Post Office Box 2265 Santa Fe, New Mexico 87504-2265

Telephone 982-4285 Area Code 505

June 9, 1986

HAND-DELIVERED

RECEIVED

Mr. Richard L. Stamets Oil Conservation Commission Post Office Box 2088 Santa Fe, New Mexico 87504

Application for Rehearing Re: Case No. 8781 DeNovo Order No. R-8161-A

Dear Mr. Stamets:

On behalf of Pollution Control Inc. and Snyder Ranches, Inc. please find enclosed an Application for Rehearing of the referenced case.

Very tr<u>ul</u>y your W. Thomas Kellahin

WTK:mh Enclosure

John P. Weber, Esg., (w/enc.) cc: Maddox, Renfrow & Saunders Post Office Box 5370 Hobbs, New Mexico 88214

> Mr. Larry Squires (w/enc.) Pollution Control Post Office Box 1060 Lovington, New Mexico 88260

Mr. Joe Ramey (w/enc.) Post Office Box 6016 Hobbs, New Mexico 88241

J. W. Neal, Esq., (w/enc.) Post Office Box 278 Hobbs, New Mexico 88241

Tim Kelly (w/enc.) Geohydrology Associates 4015 Carlisle, NE, Suite A Albuquergue, New Mexico 87107

JUN 9 1986

OIL COMSERVATION DIVISION

BEFORE THE OIL CONSERVATION COMMISSION NEW MEXICO DEPARTMENT OF ENERGY AND MINERALS

IN THE MATTER OF THE HEARING CALLED BY THE OIL CONSERVATION COMMISSION OF NEW MEXICO FOR THE PURPOSE OF CONSIDERING:

> CASE NO. 8781 <u>DeNovo</u> ORDER NO. R-8161-A

APPLICATION OF PETRO-THERMO CORPORATION FOR AN EXCEPTION TO DIVISION ORDER NO. R-3221, AS AMENDED, AND FOR AUTHORIZATION TO DISPOSE OF ASSOCIATED WASTE HYDROCARBONS AND OTHER SOLIDS, OBTAINED IN CONJUNCTION WITH THE DRILLING AND PRODUCTION OF OIL AND GAS INTO UNLINED PITS, LEA COUNTY, NEW MEXICO.

APPLICATION FOR REHEARING

COMES NOW POLLUTION CONTROL INC. and SNYDER RANCHES, INC., and pursuant to the provisions of Section 70-2-25 N.M.S.A., 1978, apply to the Oil Conservation Commission of New Mexico for Rehearing of the above captioned case and order, and in support thereof state:

STATEMENT OF FACTS:

On April 9, 1986, the New Mexico Oil Conservation Commission ("Commission") held a hearing on the application of Petro-Thermo Corporation for a permit to use the SW/4 SE/4 NE/4 of Section 16, T20S, R32E, NMPM, Lea County, New Mexico for the commercial disposal of waste material from oil and gas field operations, including produced salt water and solid wastes.

The disposal facility is to be located on State of New Mexico lands under the management and control of the Commissioner of Public Lands. At the time of the hearing, Petro-Thermo Corporation had not obtained a business lease from the Commissioner of Public Lands of New Mexico to use the surface for this purpose.

The application of Petro-Thermo Corporation was opposed at the Commission hearing by Snyder Ranches, Inc., which is the owner of federal grazing leases adjacent to the applicant's proposed facility and is an interested party affected by this application. In addition, the application was opposed by Pollution Control Inc. which has an approved surface disposal facility and is also an interested party affected by this application.

On May 20, 1986, the Commission entered Order R-8161-A which approved the application-of Petro-Thermo Corporation.

Within twenty days of the date of that order, Pollution Control Inc. and Snyder Ranches, Inc., have filed this Application for Rehearing.

- 2 -

GROUNDS FOR REHEARING:

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1. THE COMMISSION HAS FAILED TO MAKE AN ESSENTIAL JURISDICTIONAL FINDING CONCERNING THE PROTECTION OF CORRELATIVE RIGHTS.

Order R-8161-A fails to set forth the fundamental factual findings raised at the hearing on how, if at all, the approval of this application will protect the rights of Snyder Ranches, Inc. It was undisputed in the evidence that the contaminated waste water would migrate off the proposed Petro-Thermo Site. The Commission has made no finding that correlative rights will be protected. See Sims v. 186 Mechem, 72 N.M. (1963) and Faskin v. Oil Conservation Commission, 87 N.M. 292, 532 P2d 588 (1975).

2. LACK OF PROPERTY INTEREST IN APPLICANT AT THE TIME OF HEARING DENIES COMMISSION JURISDICTION TO ENTER ORDER.

Commission Order Finding (19) is erroneous. Petro-Thermo failed to establish a property interest in this case. That failure compels the Commission to deny the application in accordance with Division Rule 1203. Petro-Thermo has no lease, no ownership and no permission to utilize the proposed surface for this facility. The rights to this tract are vested in the Commissioner of Public Lands and in the absence of

- 3 -

his prior approval, Petro-Thermo cannot bring a case before the Commission.

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Under the definition section of the Division Rules and Regulations, an "Owner" is defined as the "person who has the right to drill into and to produce from any pool and to appropriate the production either for himself or for himself and another." An "Operator" is defined as a person "who, duly authorized, is in charge of the development of a lease or the operation of a producing property." Petro-Thermo Corporation under the Division's definitions is neither an owner or an operator.

3. COMMISSION HAS FAILED TO MAKE ESSENTIAL FINDINGS CONCERNING BENEFICIAL USE ("NEED") FOR THIS FACILITY AND HAS COMMITTED ERROR IN DENYING OPPONENTS AN OPPORTUNITY TO PRESENT EVIDENCE OF LACK OF NEED FOR THIS FACILITY.

The Division's Rules and Regulations define surface waste as "... the unnecessary or excessive surface loss or destruction without beneficial use, however caused..."

The Commission has committed reversable error in precluding or ignoring evidence of "need" for this facility during the hearing held on April 10, 1986. The extent to which the surface can be "wasted" is directly linked to the guestion of need. For example, if all existing facilities in the area do

- 4 -

not have the capacity to handle the volumes Petro-Thermo proposed for this facility, then the use of the surface would be reasonable and waste of the surface would not occur. Conversely, in the absence of proof of need, any use of the eight acre tract would be unreasonable and therefore constitute surface waste. The Commission has erroneously precluded evidence on an essential element of proof.

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4. THE COMMISSION ORDER LACKS SUFFICIENT FINDINGS OF ULTIMATE FACTS TO SUPPORT ITS APPROVAL OF THE DISPOSAL RATE OF 30,000 BARRELS PER DAY.

Petro-Thermo Corporation's testimony was that they anticipated to dispose of only 2250 barrels of produced salt water a day. (Petro-Thermo Exhibit 10, page 15).

There is not the faintest clue in any of the findings explaining the Commission's reasoning in approving 30,000 barrels a day disposal volumes when the applicant only anticipates needing 2,250 barrels a day. The Commission's Order on this issue violates the standards set forth in <u>Fasken v. Oil Conservation</u> Commission, 87 N.M. 292, 532 P2d 588 (1975).

5. THE FINDINGS OF ORDER R-8161-A ARE NOT SUPPORTED BY SUBSTANTIAL EVIDENCE.

Applicant has failed its burden to prove that

- 5 -

the contaminated discharge water can be safely deposited into the facility without adversely affecting fresh water.

The evidence at the hearing was that if the seepage from the impoundments at the proposed waste facility migrated off-site towards Laguna Plata, the discharged water could migrate out the west side of the Plata into Nash Draw and on to the Pecos River. The conclusion from all of the hydrologic evidence is that, from current data, none of the experts know where and at what rate the discharged water will migrate.

The Commission violates Section 70-2-12B(15) by the approval of this application.

6. THE COMMISSION'S DECRETORY PARAGRAPH NO. (2) OF ORDER R-8161-A DENIES SNYDER RANCHES AND POLLUTION CONTROL INC. PROCEDURAL DUE PROCESS.

The Commission has only required that Petro-Thermo submit a revised plan acceptable to the Director of the Oil Conservation Division for the installation and sampling of monitoring wells. Such an order provision fails to afford Pollution Control Inc. and Snyder Ranches, Inc. with an opportunity to appear and contest the proposed monitoring system. This provision effectively removes the opponents from the essential process of participating in determining

- 6 -

the method by which this monitoring system, yet to be proposed, is supposed to protect their correlative rights.

Further, previously approved monitoring systems agreed to by Petro-Thermo and the Division, as set forth in Division letter dated February 18, 1986, were contested at the Commission hearing by Pollution Control and Snyder Ranches hydrologist and the Commission has failed to make appropriate findings.

7. BY APPROVING THE DESIGN OF Α DISPOSAL FACILITY THAT DOES NOT PROHIBIT THE MIGRATION OF THE DISCHARGED WASTE WATER BEYOND THE BOUNDARIES OF THAT FACILITY THE COMMISSION HAS EXCEEDED ITS STATUTORY AUTHORITY AND JURISDICTION AND THE COMMISSION ORDER R-8161-A IS VOID.

The Commission has failed to require adequate means to prevent the contaminated waste water from migrating off of the facility and onto the property of Snyder Ranches, Inc.

It is undisputed that the produced waste water that is to be disposed of in the unlined surface pits at the proposed Petro-Thermo facility will leak through the bottom and sides of the pits and migrate beyond the boundaries of the proposed facility. In fact, the applicant's entire design and plan for the facility is based upon that principal. The migration of contaminated waste water will destroy the grazing

- 7 -

grasses and vegetation under the ownership and control of Snyder Ranches, Inc.

The Commission has granted to Petro-Thermo a disposal permit authorizing the disposal for profit of waste water which will migrate beyond the boundaries of that facility. Such action constitutes underground trespass, exceeds the jurisdiction of the Commission and its statutory authority. The order is void.

8. THE COMMISSION IS PRE-EMPTED BY FEDERAL LAW FROM ENTERING AN ORDER THAT AFFECTS THE MANAGEMENT OF THE BUREAU OF LAND MANAGEMENT AND CONTROL OF ADJOINING FEDERAL LANDS.

The Commission Order R-8161-A recites at length the concerns and statements expressed at the hearing by Bureau of Land Management District Director but then arbitrarily ignores those concerns and enters an order that adversely affects the right of the Bureau of Land Management to effectively manage and control the adjoining federal lands.

The Commission has attempted to exercise judgment and control over federal lands which exceeds the jurisdiction of the Commission.

9. THE COMMISSION ORDER FAILS TO MAKE ADEQUATE FINDINGS CONCERNING WASTE.

The New Mexico Supreme Court in Continental Oil

- 8 -
<u>Co. v. Oil Conservation Commission</u>, 70 N.M. 310 (1962) and in <u>Sims v. Mechem</u>, 72 N.M. 186 (1963) requires the Commission to make findings that are sufficiently extensive to show not only the jurisdiction but the basis of the Commission's order.

Order R-8161-A fails to make adequate findings concerning how the approval of this application will prevent waste.

- 10. THE FOLLOWING COMMISSION FINDINGS ARE ARBITRARY, CAPRICIOUS, CONTRARY TO LAW, AND ARE NOT SUPPORTED BY SUBSTANTIAL EVIDENCE.
 - 1. Finding (15), (a) through (k).
- Finding (19), after the phrase "said Section 16."
- 3. Finding (22).

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- 4. Finding (23).
- 5. Finding (24).
- 6. Finding (25).
- 7. Finding (26).
- 8. Finding (29).
- 9. Finding (31).
- 10. Finding (32).
- ll. Finding (33).
- 12. Finding (34).
- 13. Finding (35).
- 14. Finding (37).
- 15. Finding (38).

-9-

WHEREFORE, Pollution Control Inc. and Snyder Ranches, Inc. respectfully request that the Commission grant a Rehearing in the above styled case and that after rehearing, the Commission vacate and set aside its Order R-8161-A and enter its Order denying the application of Petro-Thermo Corporation in this matter.

Respectfully submitted,

By:

W. Thomas Kellahin, Esq. Kellahin & Kellahin Post Office Box 2265 Santa Fe, New Mexico 87501

and

J. W. Neal, Esq. 116 N. Turner Hobbs, New Mexico 88240

Rehearing request Snyder Ranches et al Case 8781 De Novo R-8161-A (1) Correlative Rights dette detined in 70-2-33 H. Relate to production diama pool. Nothing to do with surface disposed Issues in this case. Relates to protection of fresh water only. Agree AST (2) Lack of standing by Petro Thermo is alleged because PV trod 170 property interest in Section 16, Findings No 19 and No 6 address this issue. I think we are right. Whet does July Think? FT is not as a will obser or operator but would a appear to be a person with a property interest. Petro Theam was correct Tobring case because OCS (3) Applicant suggests the tunecessery use of the surface for the PT decility is <u>surface</u> woste as defined in the law and OCD rules, The low 70-2-3 B and Rule O.1 refer to surface waste of oil and/or gas, only not the surface of the ground. Agree ASB

Appliciant suggests BCC, erred in not Hpplicunt suggests WCC erred in not bearing testimony as to need for facility. Nothing I am aware of in the Oil& Gross Act requires or calls for us to consider need. To do so might well be (probably is) outside our statution authority. Agree Agris (4) Applicant states that Ware in subficient findings in the order to show reasoning for approving 30,000 bornels per day capacity when regular expected Volume is 2,250 bpd. Sthire is only one dividing as to This volume (Finding No 15 g). The record is good but if Snyder found a sympethetic judge This might be a problem. Transcript pas-sould designed for 30,000 mild date (5) Applicant allelyes the OCC did not have sufficient evidence relative to The geopydrology of the area to make findings that all fluid mignetion would be toward and the the lagune and the t such we thids could not move out of the Lagune to the libert.

This allegation gives me more trouble Than any of the rest. I feel PT's Case was weakined because then were unable to put their hydrologist back on the stand to answer The issues raised by Snyder's hydrogist, While appropriate technical Staff believe X PT's cose is right I worry about The adequay of the record in This regard. See commention net page, IA (6) At the haring, Snycher's hydrologist atocked the sampling and monitor will program. I do not believe they asked to be a party to any future program to be approved by the Director, much the There could be a problem here in thet only one finding (NO37) discusses The monitoring the plan. If the des sympathetic judge could agree with Snyder on this. On the other hand, it the case is strong for approval of the system, the monitoring plan is extraneous and might not be considered worthy of overturning The order for. Agree ACT&

(7) Applicant 50ggosts OCC cannot opprove dispisal operations when Aluids will arguste From the property. Applicant also offers That the Huid will migrate to their property and destroy gross and Vegetation. The statues (70-2-12 B 15) only requires The statues (70-2-12 13) only requires us to require to produced total and used water to protect fresh water. If does not repaire as to see that it is not have the property on which it is disposed. This is a worrysome area though where a sympothetic judge could make bod low. There was tratingny that the operation could have grass but none I remember source the the grass harmed would be on Snyder's property. No water hairm Snyler's grafs since will moste towards lake. (8) Applicant argues OCC is premempted from affecting adjoining federal lands. This is not an argument made by The BLM witness, I believe our findings 26 Thru. 31 address BLM's concerns

(9) Applicant suggests The orden is involid in that it was not no t This address was to Sufficiently. is not a woste or correlative rights case but a drish water protection cose.

(10) This is a catchell complaint which seems to be addressed elsewhen in the above.

If there is any real question whether this order would be upheld by a judge, want it to PT's advantage to have a rehearing so that the case could be made antight" it seems logical that altho PT would have extra expense, they would at least be assured of a permet. Without a releasing, they could love everething The orland got all nequires reasonable protection of presh water supplies , Such reasonable protection is afforded when The preponderance of evidence shows that it is unlikely that sresh water will De imported to badwersele by the discharge both ground water (a seep and surface Water (Lequine Plata) currently exceed protectable quality (10, Domg/L) and Donor need further protestion. The argument that discharge to the subsurface will reach Nagh brow on to The pecos Rever the fallecoust and this Should cause the permit to be denied is equally Unrasonable. The amount of time to reach the surer (if indeed it does) app acologic time most intervals, and vo

(both absolute on & relative to other natural and monmadedischarges) in will be minusule. Therefore, The OCS did not et unresonally m'qrantinp the permit. Specifie comments on Kelly's Dight points; and a deguacy of record of hearing = D The thickness of the allurial cover at the site won'The known until l'etrotherm drilles. A rehearing won't help in that area if they con I get land office permission to drill. A rehearing will allow discussion and entry into The record of a contours map of redbed elevations mean the site, This can be obtained from our well records. 2). As mentioned above, a map would help show this that reduced toug surfaces slope towards Laguna Plats. Also, mopping of the location and stuke and dip the outerops could be presented, 3 Mapping from our records would provide this information also.

(4) Same of (3) Comment on 2 3 & 4, there is considerable discussion in the record as to whether fluids will migrate to Laquine Plate of to West, of to the Lequine than to the coep. Both killed agree that either metals interpretation is possible, It does not matter since the issue is presh water tutable prests water is present between Laquna Plato and Nath Asaco to the west. Even Kelly agreet that the water down by Loving to before discharge to the Recot is In The 200,000 ppm TS5 nange. (5) Since The pondo and topographially on a hill, any ground wates / mounding, and to poor maintenance will be oriented in the direction of the Loquina, and not Quires' leasedaheato the East, Based on Pollection Control's experience, The area impacted by mounding of separe was loaves, which would be within the

6) Keller agreed that evaporation in Leanna Plata is adequate for any seepage that neaches it. Since ponderaporation is northe primary dispotal method This point does not need further Liscussion. > Aronge of disposal fluid analys is adequate in my opinion. The springs cited in the report are 7500-9000 ppm chlorike Inot TOS, Since Then, Petrotherm complex a toring and found it to be 196,000 TAS The earlier springs were sampled m' 1969). This evidence would be entered into the pecord and demolify Kellyt. contention that their is significant difference in m' Wates Quality between Laguna's Cotana APlata B) My analysis thous 225,000, but no matter. Kelly admits that even the no. -potagy playas have. The values in excess of 150,000ppr

In summary, The record could stand on its own if the person reading or reviewing Tunderston training clearly the relationship between distances, Times, volumes, concentrations, etc., of The locations and fluids involved. Since & That may not accup, it is better to have a rehearing and have itall spelledout, and add the new mormation.

PROBLEMS WHICH HAVE NOT BEEN ADEQUATELY ADDRESSED BY PETRO-THERMO REPORT.

and the state of the second of

- 1. The thickness of the alluvial cover is unknown at the proposed site. 20 Within Section 16, the thickness ranges from 0 to 130 feet, but it is completely unknown at the proposed site itself.
- 2. The upper surface of the redbeds is an erosional surface of considerable relief. There is no evidence presented by the report which confirms that the redbed surface slopes directly toward Laguna Plata.
- 3. The report does not disprove work by Reed (1969) which indicates a bedrock channel which would result in a westward migration of ground-water from Laguna Plata (illustration).
- 4. The report, Figure 3, shows that the 3,440-foot contour is closed, thus indicating that Laguna Plata is a closed depression. Data on the map shows no justification for closing the contour. The intrepretation shown in Figure 3 is not supported by work by Hunter (1985) or by Geohydrology Assoc., (1979). (illustrations)
- 5. No evidence is presented in the report which substantiates that the disposal ponds will function properly. In fact the very nature of drilling mud is to cause plugging of natural porosity in sediments.
- Evaporation of fluids should be calculated for surface area of the disposal ponds and NOT for Laguna Plata.
- 7. The report does not contain any chemical analyses of water samples from the fluid which will be disposed. The TDS range is reported to be 25,000 to 75,000 ppm but springs at Laguna Plata have less than 9,000 ppm.
- 8. The concentration of 335,100 ppm reported in report for Laguna Plata is a concentrated brine resulting from emporation on the lake floor or is a residual concentration from potash discharge by Kerr-McGee

ENGINEERING AND DESIGN OF PLATA DISPOSAL FACILITY SECTION 16, TOWNSHIP 20 SOUTH, RANGE 32 EAST LEA COUNTY, NEW MEXICO

> PETRO-THERMO CORPORATION HOBBS, NEW MEXICO

> > PREPARED FOR:

NEW MEXICO OIL CONSERVATION DIVISION CASE NO. 8781

APRIL 9, 1986

BEFORE THE OIL CONSERVATION COMMISSION South Fr, New Merico
Case No. 8781 Emilia No. 8
Submitted by
Hearing Date





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	and leases to either a Federal agency or a State agency, z the use of this form and the number of copies to be issued by, or may be obtained from, the local Federal s. and/or State laws and regulations. All attachments dance with Federal requirements. Consult local State	, so state in item 22, and in any attachments. , so state in item 22, and in item 24 show the producing arate report (page) on this form, adequately identified, puting and the location of the cementing tool. for items 22 and 21 above.) $0Sx_*(Circ 300 \ sx to pit_*(VOC 24 \ hrs_*Circ 8)$	W/176 BW. Closed DV W/2000/ , circ 200 sx	68. GROLOGIC MARKERS	TOP	MKAS. DEPTH TRUE VERT. DEPTH	ustler 880 (&2633) ansill 2525 (&988)	irst Bone Springs [7018] (-4105)	iecond II II. I 92951(-57821)	olfcamp Limestone 108501 (-73371)	trawn Lime 11668' (-8155') . .toka Sand 12070' (-3557')	Inrrow Limestone 12313 -8800' Iorrow Sand 12728 -9215'						
INSTRUCTIONS	ete and correct well completion report and log on all types of lands i tws and regulations. Any necessary special instructions concerning without procedures and practices, either are shown below or will be 4, and 33, below regarding separate reports for separate completions mitted, copies of all currently available logs (drillers, geologists, su 1 be attached hereto, to the extent reguired by applicable Federal locations on Federal or Indian land should be described in accord	(where not otherwise shown) for depth measurements given in oth production from more than one interval zone (multiple completion), (if any) for only the interval reported in item 33. Submit a sepa showing the additional data pertinent to such interval. eds for this well should show the details of any multiple stage ceme form for each interval to be separately produced. (See instruction " $\approx Csg-Cont$ 1d: Opered DV (≈ 28001 m/ 800	hrs thru JV. Znd: 900 sx trinn Class C cht. Pumped plug down cmt.	TS THEREADE; CORED INTERVALS; AND ALL PRILL-STEM TESTS, INCLUDING	DESCRIPTION, CONTENTS, ETC.	See DST Summaries attached										NSE 102		U 5. GOVERNMENTS PRINTING OF 166 : 1963 O 0031836
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COM Lico or 8040 DUEST FOR PERMISSION STATION STATISTICS FORMATION RECORD NEW MEXICO OIL CONSERVICTION COOMISSION , MOR Santa-No. Now Moxto 15 5 10cm 15 Calechi 15 110140 25 Red Sand 40 50 10 Sand & Gravel 50 .50 200 Red Bed ORO AN INSand IBSION. 200 240 1 CR ME 240 120 360 Red Bed & Sand 360 400 40 Sandy Shale RET ASE 400 500 BTADILY 26 MI JINAAnh'y dr'I t'e aniwold 865 910 AREA 640 ACRES LOCATE WILL CORRECTLY 910 920 10 Shale & Salt 1. 1. 1. 1. 920 Sam Weiner 1080 Annydrite & Potasi Jarge (and a start Straid and a start Straid 1110 0g2294 2332 2294 501 aoul62 Lime (Show of Gashat 2338) 2332 2394 🐇 2394 2405 2444 2467 2482 2495 2520 Hard, Gray, Lime At OE Yell 2530.21 Despammos spilling 2559. Soft Lime - Hole filled woth Oil 2559. TOTAL DEPTH TOTAL DEPTH 2556 DI I 516 34 Elevation above sea level at top of casing she (marries , marries and The information given is to be kept confidential until The trai spirit in stations OIL BANDS OR ZONES to BENGE IN BEINDE STO HOMAS INO Burthe Conservation Loppanias have not tone nd rest 61302 M No. 1, from_____2495____ w unis barran in 1 lled wir at that ur A 1587 2559 1.1. Month and The other and No. 2. from _ 2556 No. 6, Iton in to third and the devirent of the No. 3. from. THE AND THE BUT AVANTRATION THE MAN THE tophists invest measuration Gerri lever to a Ca Include data on rate of water inflow and elevation to which water road in hole. reet 50 feet from ton 300 No. 1, from 280 de's No. 2. from. No. 3. from leef No. 4. from CASING RECORD 1.3 T & FILLED FROM PURPOR PERFORATED, AO GRIB PREESADS TROMAW TOOT RUY ASIR TYNOMA MART $\mathbf{0T}$ MO <u>.a.</u> 936 8 8-1/8", 32 .T.P. 2367

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	NEW N Well Loc	EXICO OIL C	ONSERVATION	COMMISSION	FORM C-128 Revised 5/1/57
	SEE INSTRUCT	TONS FOR COMPT	ECTION A	UN THE REVERSE SIDE	<u>ــــــــــــــــــــــــــــــــــــ</u>
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If the answer to qu	estion two is "no," li	st all the owners a	nd their respective in	terests below:	· · · · · · · · · · · · · · · · · · ·
wner			Land Descri	ption	
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	S E (CTION B		CER	TIFICATION
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(See Instructions and Spaces for Additional Data on Reverse Side)

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R. OEL N. M. P. M., HALLWAY Field,	<u>_</u> 68		_County.
Well is <u>990</u> feet south of the North line and <u>2310</u> feet west of	of the East line of	Section 16	<u>1200</u>
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CASING RECORD

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FORMATION RECORD

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		TIE ST 17	BEFORE THE OIL CONSERVATION COMMISS
		Stor por the	Saulto Fo, New Moxico
	S		Case No. 8781 Emilibit No. 1
		A A A	Hearing Date
			and and a second se
	STATE CORPORATI	ON COMMISSION	I OF NEW MEXICO
•		•	
Centific	cate of Publi	ic Convenie	ence and Necessity
Docket No		-	No21449
A Certificat	e of Public Convenience and	l Necessity is hereby gr	anted
1	PETRO-TH	IERMO CORPORAT	ION
whose office or pla	ce of business is $at - P \cdot O$.	Box 2069, Ho	bbs, New Mexico
to operate a Comn	non Motor CarrierFr	eight	Service over the route and l
the schedules here	tofore approved by this Con	nmission, or which may	y hereafter be approved by this Commission
said route being as	foliows:		
ат то		ATED DETNEEN	
11.7	NOFURIATION OF W	ATER DETWEEN	AV COUNTIES NEW
	TDING, MURA, SAN	MIGUEL AND QU	AT COUNTIES, NEW
ME	VICU, UVER IRREGU	LAR RUUIES, U	NDER NUN-SCHEDULED
SEI	(VICE.		

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This Certif	icate to remain in effect fro	om and after date her	reof, subject to applicable provisions of th
New Mexico Motor issued purs Witness th	Carrier Act, and Rules, Reg suant to Order da e signature and seal of the	ulations and Requirem ted July 8, 1 Commission at Santa	ents prescribed thereunder., and is 981. 1 Fe, New Mexico, this <u>8th</u> day (
July	, 1981		
	MAAAA		
	()	State	Corporation Commission
	THE AL)	of New Mexico
3.24		By A	1. 11
		2)	u fline
		1	ne fline Chairman
ATTEST:			Mult. Ellett

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AUGUST 27, 1981

PAGE 2

CERTIFICATE OF PUBLIC CONVENIENCE AND NECESSITY NO. 21449

PETRO-THERMO CORPORATION

HOBBS, NEW MEXICO

ENDORSEMENT

With all necessary documents on file, and with the records of this Commission as authority, Certificate of Public Convenience and Necessity No. 21449 is hereby endorsed to include intrastate-- operating rights as follows:

Transportation of oilfield equipment and supplies (as defined in Order issued by the State Corporation Commission the 17th day of September 1947), between points and places in Lea and Eddy Counties, New Mexico, over irregular routes, under nonscheduled service.

Transportation of houses, between points and places in Lea and Eddy Counties, New Mexico, over irregular routes, under nonscheduled service.

Transportation of water and crude oil to and from wildcat wells to common carrier loading points and refineries, and residue oil from tank batteries, in Chaves, Lea, Eddy, Curry and Roosevelt Counties in New Mexico; over irregular routes and under non-scheduled service.

Transportation of all liquids in bulk, in tank trucks, used in, or in connection with, the discovery, development, production, refining, manufacture, processing, and storage of natural gas, and petroleum and their products and by-products, providing that all hauls of refined petroleum products be restricted to movements to oil well locations and the resultant salvage thereof from oil well locations, with points of origin and destination in each instance being within Chaves, Lea, Eddy, Curry and Roosevelt Counties, New Mexico only, with no diversion of shipment in transit to storage, over irregular routes, under nonscheduled service.

EFFECTIVE THIS 27TH DAY OF AUGUST, 1981.

NEW MEXICO STATE CORPORATION COMMISSION ERNA. Chairman ELLIO Α. Commissioner

JIMMIE W. GLENN, Commissioner

ATTEST: SMITH. Director

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Page 3

February 10, 1982



CERTIFICATE OF PUBLIC CONVENIENCE AND NECESSITY NO. 21449

PETRO-THERMO CORPORATION

<u>E N D O R S E M E N T</u>

Pursuant to the Commission's Order dated January 18, 1982, Certificate of Public Convenience and Necessity No. 21449 is hereby endorsed to include operations as follows:

Transportation of fresh water, brine water (salt water) residue oil from tank batteries,oil base drilling mud and waterbase drilling mud in liquid form only, acid CO2 liquid, drilling mud containing barite, hydraulic fracturing fluids, and basic sediments between points and places in Lincoln, Guadalupe and DeBaca Counties, New Mexico.

Effective this 10th day of February, 1982.

ATTEST:

NEW MEXICO STATE CORPORATION COMMISSION no Eric P. Serna, Chairman

Petro-**T**hermo corporation

P.O. BOX 2069 PHONES (505) 393-2417 — 397-3557 HOBBS, NEW MEXICO 88241-2069



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LAGUNA PLATA AREA MAP	1
WILLIAMS SINK USGS MAP	2
WATER SAMPLE OF LAGUNA PLATA	3
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PLATA DISPOSAL DESIGN	6
PIT AND TANK CHART	7

PETRO-THERMO CORPORATION



APRIL 9, 1986

NEW MEXICO COUNTIES IN WHICH PETRO-THERMO CORPORATION HAS HAULING AUTHORITY

UNION	
MORA	
HARDING	
SAN MIGUEL	
GUADALUPE	
QUAY	
DE BACA	
ROOSEVELT	
CURRY	
LINCOLN	,
CHAVES	
LEA	
EDDY	

BEFORE THE OIL CONSERVATION COMMISSION Santa Fe, New Mexico
Case No. 8781 Exhibit No. Z
Submitted by
Hearing Date

50 YEARS



STATE OF NEW MEXICO ENERGY AND MINERALS DEPARTMENT OIL CONSERVATION DIVISION



TONEY ANAYA GOVERNOR

March 21, 1986

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE. NEW MEXICO 87501 (505) 827-5800

Mr. Robert W. Abbott Petro-Thermo Corporation P. O. Box 2069 Hobbs, N.M. 88241-2069

Re: Request for Extension of Time for Temporary Use of Pit

Dear Mr. Abbott:

This letter is to confirm a telephone conversation today between Jamie Bailey, OCD Field Representative, and John Weber regarding an extension of the permit for temporary disposal of brine, fresh water-based drilling mud, and waste cement in the emergency overflow pit located at the Blinebry-Drinkard SWD System Well No. A-22 in 22-T22S-R37E. Your latest temporary permit expired March 20, 1986.

Due to time restrictions associated with Case No. 8781 which was brought to hearing on December 18, 1985, and will be heard <u>De</u> <u>Novo</u> before the Commission on April 9, 1986, you are hereby granted an extension until April 10, 1986, for the specified temporary use of the pit.

An inspection of the area will be made on April 3, by OCD personnel, and they are authorized to order immediate closure of the pit if conditions warrant.

Authorization for the temporary use of this pit may be rescinded at anytime if there is evidence that the pit has overflowed or if Water Quality Control Commission or Oil Conservation Division rules or regulations have been violated.

Sincerely R. L. STAMETS

R. L. STAMETS Director

RLS:dp

cc: Jerry Sexton David Boyer Michael Stogner

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BEFORE THE	1118	L.M.	1170	ا ا المنطقة الموطقة
OIL CONSERVATION CO Schia Fe, New Mex	MræS ico	SIGNUL!	V LU	
Case No. 8781 Exhibit No	<u>, 3</u>	184 31 11	0163.	
Submitted by				
Hearing Dare				
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PETRO-THERMO CORPORATION

P.O. BOX 2069 PHONES (505) 393-2417 — 397-3557 HOBBS, NEW MEXICO 88241-2069



	· · ·	
NUMBER OF WELLS	· · · ·	
AS OF 1-1-85	· .	

LEA COUNTY	<u>01L</u> 13897	<u>GAS</u> 1410	OIL AND GAS 15307
EDDY COUNTY	5850	1390	7240
CHAVES COUNTY	1443	762	2205
	21190	3562	24752

HOBBS POOL

BLINEBRY	23
BLINEBRY, EAST	: 1
CHANNEL SAN ANDRES	0
DRINKARD	22
PADDOCK	3
G-SA	479
SAN ANDRES, EAST	20

BEFORE THE OIL CONSERVATION COMMISSION Santa Fo, New Mexico
Case No. 8781 Intituit No. 4
Submitted by
Hearing Date



December 16, 1985

State of New Mexico Energy and Minerals Department Oil Conservation Division P. O. Box 2088 Santa Fe, New Mexico 87504-2088

RE: Application for Solids and Produced Water Disposal Site

Gentlemen:

We have been advised that Petro-Thermo Corporation has applied to the Oil Conservation Division for authorization to operate a disposal site for oil field related solids and liquids in Section 16, Township 20 South, Range 32 East.

We understand that Petro-Thermo Corporation will undertake the disposal of oil field related solids and liquids in a manner that will afford reasonable protection against contamination of fresh water supplies.

It is in the public interest, convenience, and necessity to grant Petro-Thermo Corporation's application. Presently, there are only two approved waste disposal sites of this kind serving the entire southeastern New Mexico area. There is a need for additional approved waste disposal sites convenient to our currently operated oil and gas properties. Petro-Thermo Corporation's proposal will meet this need.

We also believe that the waste of valuable energy resources can be avoided by approving Petro-Thermo Corporation's application. Properties which would, from the standpoint of the production of crude oil, be otherwise uneconcmical, could be more efficiently operated given a conveniently located waste disposal site.

Please consider our endorsement of this proposal as a positive step forward in the continued development of New Mexico's important energy industry.

Sincerely yours,

Jus F. Junten, President Natural Resources Inginaering, Inv.

BEFORE THE
OIL CONSERVATION COMMISSION
Santa Fe, New Mexico
Care No. 8781 Exhibit No. 5
Submitted ev
Hearing Date



Amoco Production Company

Post Office Box 68 Hobbs, New Mexico 88240

L. R. Smith District Manager

December 16, 1985

File: SGH-4211-594.5

Re: Petro-Thermo Corporation Proposed Disposal Site

New Mexico Oil Conservation Division P. O. Box 2088 Santa Fe, NM 87504-2088

As a major producer of oil and gas in southeastern New Mexico, Amoco Production Company feels that the approval of the solids and produced water disposal site in Section 16, T-20-S, R-32-E in Lea County would benefit the oil and gas industry in this area of the state.

The site proposed by Petro-Thermo Corporation is located in an area which has been granted an exception under NMOCD Order R-3221-B to the no pit rule. Also, the proposed site is centrally located between Hobbs and Carlsbad and would provide convenient access from producing operations in both Lea and Eddy Counties.

JR Smit

LBG/sh APRD01-H

BEFORE THE OIL CONSERVATION COMMISSION Senia Fo, New Moxico
Case No. 8781 Tria No. 6
Submitted wy
Hearing Date

P. O. DRAWER "D" MONUMENT, NEW MEXICO 88265

December 9, 1985

State of New Mexico Energy and Minerals Department Oil Conservation Division P.O. Box 2088 Santa Fe, New Mexico 87504-2088

RE: Application for approved Waste Disposal Site

As a member of the oil and gas producing community in Lea County, we recognize the importance of adequate waste disposal facilities to the future economic development of our industry in Southeast New Mexico. With only two approved waste disposal sites in the area, the need exists for additional environmentally safe and NMOCD approved facilities both to inhance comp etition in the disposal segment of the industry and to provide additional waste disposal capacity. Future drilling, secondary and tertiary recovery projects will depend on the existence of such facilities.

EN TRADA HEES CORPORAT.

The Petro-Thermo Corporation proposal for a waste disposal site in the NE4 of Sec. 16, T-20S, R-32E, Lea County, will help meet the disposal requirements of the oil and gas industry in Southeast New Mexico. Your prompt consideration of their application will be appreciated.

Yours very truy, Sw.

S.W. Small District Superintendent

XC: Petro-Thermo Corp. P.O. Box 2069 Hobbs, New Mexico 88240 ATTN: Mr. R.W. Abbott

BEFORE THE OIL CONSERVATION COMMISSION Santa Fe, New Mexico
Case No. 8781 Entitlet No. 7
Submitted by
Hearing Date





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PETRO-**T**HERMO CORPORATION

P.O. BOX 2069 PHONES (505) 393-2417 — 397-3557 HOBBS, NEW MEXICO 88241-2069



WATER ANALYSIS

SAMPLE: LAGUNA PLATA LAKE WATER

LOCATION: APPROXIMATELY 2500' FNL, 1800' FEL SECTION 10, TOWNSHIP 20 SOUTH, RANGE 32 EAST LEA COUNTY, NEW MEXICO

DATE: DECEMBER 11, 1985

SPECIFIC GRAVITY AT 60 ⁰ F	1.2205
P ^H	7.34
CALCIUM	940 MG/L
MAGNESIUM	3,317 MG/L
SODIUM	124,644 MG/L
BICARBONATE	71 MG/L
CARBONATE AS C _A CO ₃	16,000 MG/L
HYDROXIDE	NOT RUN
SULFATE	10,124 MG/L
CHLORIDES	196,012 MG/L
IRON	.25 MG/L
BARIUM	NOT RUN
MANGANESE	NOT RUN

TOTAL DISSOLVED SOLIDS

335,108 MG/L

PETRO-**T**HERMO CORPORATION

P.O. BOX 2069 PHONES (505) 393-2417 — 397-3557 HOBBS, NEW MEXICO 88241-2069



WATER ANALYSIS

SAMPLE: SPRING DISCHARGE

LOCATION: APPROXIMATELY 1220' FNL, 1320' FEL SECTION 16, TOWNSHIP 20 SOUTH, RANGE 32 EAST LEA COUNTY, NEW MEXICO

DATE: MARCH 27, 1986

SPECIFIC GRAVITY AT 60 ⁰ F	1.03	35
P ^H	8.21	1
CALCIUM	801	MG/L
MAGNESIUM 1	,633	MG/L
SODIUM 15	,594	MG/L
BICARBONATE :	170	MG/L
CARBONATE	30	MG/L
HYDROXIDE	0	MG/L
SULFATE 16	,375	MG/L
CHLORIDES 18	,000	MG/L
IRON	.3	MG/L
BARIUM	0	MG/L
MANGANESE	Not	RUN

TOTAL DISSOLVED SOLIDS

52,605 MG/L



|. |





PETRO-THERMO CORPORATION

P.O. BOX 2069 PHONES (505) 393-2417 — 397-3557 HOBBS, NEW MEXICO 88241-2069



PLATA DISPOSAL

PIT AND TANK CHART

PIT OR TANK NUMBER	CAPACITY (BBLS)	LENGTH (FT)	WIDTH (FT)	DEPTH (FT)	BOTTOM ELEVATION (FT)
W1	7480	100	60	10	3496
W2	6411	100	60	9	3490
W3 ·	4274	100	60	7	3487
W4	4274	100	60	7	3484
W 5	4274	100	60	7	3481
W6	5236	70	60	10	3492
W7	5343	100	60	8	3488
W8	4274	100	60	7	3486
W9	4274	100	60	7	3483
W10	4274	100	60	7	3481
W11	5343	100	60	8	3492
W12	5343	100	60	8	3488
W13	5343	100	60	8	3487
W14	4274	100	60	7	3483
W15	4274	100	60	7	3481
W16	2778	65	60	7	3479
	77469				
S1	3117	100	25	10	3495
S 2	3117	100	25	10	3491
S 3	2671	100	25	g	3489
S4	2226	100	25	8	3487
S 5	2226	100	. 25	8	3485
S6 /	2226	100	25	8	3483
S7	2226	100	25	: 8	3481
58	<u>2226</u> 20035	100	25	8	3480
01	4274	. 100	60	7	3478
02	<u>4274</u> 8548	100	60	7	3477
Т1	750				3509
Т2	750				3509
Τ3	750				3509
Т4	750				3509
Т5	1000				3509
т6	$\frac{1000}{5000}$				3509







STATE OF NEW MEXICO ENERGY AND MINERALS DEPARTMENT OIL CONSERVATION DIVISION



GOVERNOR

May 6, 1986

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87501 (505) 827-5800

MEMORANDUM

TO: R. L. STAMETS, DIRECTOR

FROM: JAMI BAILEY, FIELD REPRESENTATIVE II AVB 59 JB DAVE BOYER, ENVIRONMENTAL BUREAU CHIEF AVB

SUBJECT: PETRO-THERMO APPLICATION FOR EXEMPTION TO ORDER NO. R-3221 AND DISPOSAL PERMIT FOR LAGUNA PLATA - STEPHENS' REBUTTAL TO KELLY'S OBJECTIONS

At the OCC hearing on April 9 and 10, Tim Kelly submitted a list of problems (attached) which he said were not adequately addressed in the Petro-Thermo report. Our comments on Kelly's eight objections and Stephens' replies are as follows:

1. Stephens' report stated "The thickness of the alluvium ranges from 15 to 130 feet in the northeast quarter of Township 20 South, Range 32 East, Section 16, based on drillers' logs. Exposures in arroyos just north of the site suggest that the alluvial cover may be less than 10 feet thick beneath the site." Kelly objected that the thickness ranges from 0 to 130 feet, and that it is completely unknown at the proposed site itself. Stephens' rebuttal was that alluvium-bedrock contacts were observed in outcrops on the north side of the property at the west end of Laguna Plata, and the estimated 10-20 feet of alluvium beneath the site were based on extrapolations.

Bailey Comment: A drillers log for the well drilled in the Petro-Thermo 1/4 1/4 1/4 section had 100 feet of caliche underlain by 30 feet of sand. The drillers log for a well northwest of the site indicates 119 feet of sand and caliche underlain by 30 feet of water sand. (See Attachment 1) In my opinion, these logs indicate the possibility of (a) a redbed erosional channel, (b) a synclinal trough, or (c) block faulting associated with formation of the laguna. If a channel is indicated, it is justification for Reed's report (1969) that may have interpreted a bedrock channel which could result in northwestern fluid movement along the top of the the redbeds. However, redbeds outcropping west of the laguna show a southeast dip, possibly indicating closure of a synclinal trough. A trough was mentioned in Ed Reed's testimony in Case 4047 on March 19, 1969, in which he described the general area as structurally occupied by a broad synclinal trough reflected at the top of the Triassic, with the lowest part in the area of Laguna Plata. Laguna Gatuna and Laguna Tonto are in the central portion of the regional synclinal area.

Page 2

If block faulting is responsible for the increase in alluvial cover, as is the opinion of Paul Kautz, District I Geologist, the fault block dip of the redbeds could be away from the laguna in the disposal site area. (For this possibility, see my next comment.)

<u>Boyer Comment</u>: I agree with Stephens on the point, but note that two logs in the report show 130 feet of alluvium. However, 14 other logs show 12-50 feet, a figure I believe to be more reasonable. The two wells having thick alluvium show an equivalent reduction in redbed thickness and were drilled by the same company and driller 30 days apart in 1940. To settle the point, the company should be required to submit logs and results from the monitor well drilling to OCD before operation.

2. Stephens' report cites work by Nicholson and Clebsch (1961) in which drill holes in Laguna Plata encountered red beds at 20 to 41 feet below the surface, or at an elevation between 3410-3389. Kelly objects that "there is no evidence presented by the report which confirms that the redbed surface slopes directly toward Laguna Plata." Stephens replied that "the surface slope of the redbeds logically must be toward Laguna Plata."

<u>Bailey Comment</u>: The drillers log of the well directly north of the site indicates a redbed elevation of 3445. The well in the proximity of the disposal site shows redbeds at 3366, indicating a southwest dip in that limited area, or a fault. Redbed elevation in the laguna is approximately 3410, indicating a gentle northeast dip into the laguna. I agree with Stephens.

Boyer Comment: The dip or surface slope of the redbeds is not as important as the water level gradient in determining direction of fluid movement. If there is water naturally occurring above the redbed surface at the site, seepage fluid will mix with it and flow down the hydraulic (potentiometric) gradient. If no water is present, seepage will follow the structural surface of the redbeds until water is reached, mix, and flow down the potentiometric gradient. In this case the potentiometric gradient is towards Laguna Plata as shown by Figure 3 of Stephens' report (Exhibit 9).

3. Kelly objected that Stephens report does not disprove Reed's work that indicates a bedrock channel which would result in westward movement of ground water from Laguna Plata. Stephens responded that there is no evidence for a bedrock channel noted in field reconnaissance work, and if one did exist, it could convey ground water toward Laguna Plata.

Bailey Comment: Reed testified in Case 4047 on 3/19/69 to the following: "Q: And you say there is a possibility that the water from the west side of that lake might flow west? A: (Reed) I just leave that as an open question. Q: You don't have information on it? A: No, Sir." Earlier, Reed had testified "It would be my judgment that it would be acceptable, and that no damage would occur to the quality of ground water presently existing in the area to use Laguna Gatuna and Laguna Plata for salt water disposal"

Boyer Comment: I agree with Stephens.

Page 3

4. Kelly objected to closure of the 3440 water level contour line, which indicates Laguna Plata is a closed depression. Stephens responded with water level, lake, and spring elevations in the area.

Bailey Comment: Stephens' response indicated that he had sufficient control to indicate closure of the 3440 contour line.

Boyer Comment: The newest USGS topographic quadrangle (Williams Sink, 1:24,000, provisional, 1985) shows the lake elevation at less than 3430'. Two drill holes near the center of the lake are shown at elevations of 3431 and 3429 feet. The seep investigated by me on February 4, 1986 is approximately 750 feet north and west of the site. Drainage of the seep was to the north and east to Laguna Plata. Therefore, I agree with Stephens.

5. Kelly objected that the disposal ponds will plug with drilling mud and will not function properly. Stephens responded the ponds may require maintenance and that seepage is the preferred means of disposal.

Bailey and Boyer Comment: Seepage of fluids is preferred in order to tie up the heavy metals and degrade dissolved hydrocarbons which will be present in the waste fluid. Ponds that accept drilling fluid will require regular maintenance, and/or decanting of waters into adjacent ponds.

6. Kelly objected to the method of calculating fluid evaporation, stating that it should be calculated for the disposal ponds and not for the laguna. Stephens responded that there was ample data in the report to make the calculations, but that evaporation is not the primary means of disposal.

Bailey Comment: This is a moot point; Laguna Plata has 1,241 acres within the lowest closing contour.

Boyer Comment: I agree with Stephens. Also, slightly more evaporation would occur at the site than in Laguna Plata due to less salt being present in the disposal ponds.

7. Kelly objected that no chemical analysis of disposed fluids were presented and that springs at Laguna Plata have a TDS of 9,000 ppm. Stephens responded with figures that show that the TDS for the expected waste water was not less than the TDS of the springs.

Bailey and Boyer Comment: Kelly obviously made a mistake.

8. Kelly objected that the reported Laguna Plata TDS was due to concentration of brine or of potash discharges. Stephens responded that saline lakes were widespread in the area and that there were others that had not received waste waters from any source. The potash wastes have not had a significant impact on the composition of the evaporate minerals.

<u>Bailey Comment</u>: It is highly unlikely that given the size of the laguna and the infiltration method of disposal, that a waste disposal operation could have a significant detrimental impact on the salt mining operation in Laguna Plata. Boyer Comment: I agree with Stephens. Also, due to the highly saline concentration of the water, heavy metals may already be in solution at relatively high concentrations.

Some additional remarks by Stephens were made in response to Kelly's contention that there are significant differences between Laguna Gatuna and Laguna Plata. The single comment is as follows:

Boyer Comment: I agree with Stephens' responses (except for the obvious error in his reporting chloride for TDS in my seep sample), and add that the required drilling at the site will determine the thickness of alluvium, depth to the redbeds, presence of water, and resolve most site geologic issues.

The following are our conclusions in this matter:

Bailey and Boyer:

The central issue of the hearings on Case 8781 was to determine if Petro-Thermo should be granted an exception to Order No. R-3221 and a permit to dispose of waste hydrocarbons and other solids associated with oil and gas drilling and production. Given the evidence of no protectable fresh ground water at the site or in the lake, and further evidence that water flow is towards Laguna Plata and away from any potable water that could be affected by such discharges, an exception to order No. R-3221 should be permitted. Further, if there is a naturally occurring discharge from laguna Plata to the west, the effect that the proposed operation would have on such a discharge is miniscule since existing volumes of salt and concentrated brine far overwhelm any proposed Petro-Thermo contribution. The method of disposal of fluids into ponds which will allow infiltration of these fluids into the subsurface should prevent hydrocarbon and heavy metal contamination (beyond any existing levels) in the area of the salt mining operation. Damage to the surface at the facility site will no doubt occur, but the Petro-Thermo operation, as outlined in their reports, should not have an adverse effect on any fresh waters that may be located in the area, and should not impair any existing or future use of the waters in Laguna Plata.

JB:DB:dp

Attachments







DANIEL B. STEPHENS & ASSOCIATES, INC. CONSULTANTS IN GROUND-WATER HYDROLOGY

• GROUND-WATER CONTAMINATION • UNSATURATED ZONE INVESTIGATIONS • WATER SUPPLY DEVELOPMENT •

April 22, 1986

Mr. R.L. Stamets Director OCD Energy & Minerals Dept. P.O. Box 2088 State Land Office Bldg. Santa Fe, New Mexico 87501



Dear Mr. Stamets:

Reference is made to your direction at the close of the hearing no. 8781 that any additional submissions by Petro-Thermo Corp. be made no later than April 24, 1986. This is to forward agreements regarding the testimony provided by Mr. Tim Kelly for Pollution Control, Inc. and Snyder Ranches.

Pollution Control Inc. contends that there are eight "Problems Which Have Not Been Adequately Addressed by Petro-Thermo Report". This is attached (Attachment 1) and my responses to each follow:

1. The 10-20 feet of alluvial cover thickness beneath the site is estimated from extrapolations of the alluvium - bedrock contacts easily observed in outcrops on the north side of the property at the west end of Laguna Plata. This depth is at the low end of the range of values reported in well logs shown for section 16 on page 6 of my hydrogeologic report to Petro-Thermo in December 1985.

2. There are obvious outcrops of red beds elevations above the water level of Laguna Plata in arroyos and along the shore just north of the site. As indicated in my report on page 5, there is approximately 60 feet of difference in elevation of the red bed surface between the site and Laguna Plata. The surface slope of the red beds logically must be toward Laguna Plata.

3. My hydrogeologic report indicates the presence of red beds along the west and southwest portions of Laguna Plata just north and west of the proposed site. Small Springs and seeps were observed to exist near the red bed-alluvium contact. No evidence for a bedrock channel in the vicinity of the site was noted in field reconnaissance work. If a channel exists at the site, it is possible that it conveys ground water <u>toward</u> Laguna Plata.

In reference to the water level contour map Figure 3 on 4. page 13 of my report, there is adequate justification for closing The work by Hunter (1985) and Geohydrology the 3440 contour. Assoc. (1979) is in error in this locale, partly because they did not include lake and spring elevations. The free water elevation of Laguna Plata is about 3431 feet (msl), (not 3440 as shown on Figure 3). A shallow well less than 2 miles west of the site indicates a water level of 3440 ft. The elevations of Laguna Toston southwest of the site is approximately 3476 feet. There are also springs and seeps which have been noted on the west end of Laguna Plata; in fact, one of these was sampled by Mr. Dave Boyer of NMOCD. Clearly, shallow water level data indicates an east and northeast component of flow near the site which, when combined with other data, provides ample hydrogeologic evidence to close the 3440 water level contour. It may be expected that a ground water flow divide exists somewhere west of Laguna Plata which isolates flow to Nash Draw.

5. The disposal ponds may require maintenance to function properly. The soils appear to be sandy and have the potential to allow for adequate seepage if a clogging layer is not present. However, seepage is a preferred means of disposal, in that the slow travel time of flow in the soil and shallow aquifer allow natural processes to filter and degrade hydrocarbons in the seepage before they enter Laguna Plata. It is not an uncommon practice to pipe discharge directly to playa lakes as a means of disposal.

6. There is ample data in the report which will allow one to easily calculate evaporation from the disposal ponds. Evaporation is not intended to be relied upon as a means of waste disposal.

7. Springs at Laguna Plata do <u>not</u> have TDS concentrations less than 9,000 ppm. Springs at the east end of Laguna Plata have <u>chloride</u> concentrations which range from 7446 to 8864 mg/l and sulfate concentrations which are approximately 12,000 mg/l. Thus, TDS is at least 20,000 mg/l.

At my suggestion, a seep at the east end of Laguna Plata (20.32.11.323) was sampled by Mr. Jim Thornton of Petro Thermo Inc. At this seep, TDS was 196,443 mg/l and chloride was 74,000 mg/l (Attachment 2A, 2B). The spring at the west end of Laguna Plata sampled by Mr. Dave Boyer had a TDS of 36,428 mg/l (Attachment 3). Therefore, the TDS range expected for the waste water is not less than that at springs.

8. The occurrence of saline lakes in depressions of the land surface overlying salt and anhydrite formations is widespread in eastern New Mexico. There are numerous other saline lakes in the region which have not received waste water from any known source in the past. Certainly, potash mining wastes have contributed to the mineralization of Laguna Plata. The fact that a commercial salt operation exists at Laguna Plata suggests that discharge from potash mining has not had a significant impact on



DANIEL B. STEPHENS & ASSOCIATES, INC.

the composition of the evaporite minerals. The existing concentration of salt in the lake is not likely to be significantly affected by the proposed disposal operation.

Pollution Control Inc., in their exhibit 6 contends that there are significant differences when one makes a "Comparison of Gehydrologic Conditions at Laguna Gatuna with Laguna Plata" (Attachment 4). My comments in response now follow:

1. Natural Water Quality - The <u>TDS</u>, not chloride, concentration of waste water is expected to range from 25,000 to 75,000 ppm (see page 15 of my hydrogeologic report). The spring which is closest to the site, sampled by Dave Boyer (Attachment 3), has a chloride concentration of 36,428 mg/l. This falls within the range of values reported for Laguna Gatuna.

2. Ground-water Flow - There are seeps and springs at the west side of Laguna Plata which have been noted in my hydrogeologic report and inspected by NMOCD. The water level in Laguna Plata is at the lowest elevation of the lakes and shallow water levels in wells which are located in the area. Shallow water level date to the west of Laguna Plata do not support a mechanism of discharge to the west.

3. Distribution of Triassic Rocks - Red beds have been observed in outcrop in arroyos at the site. It is in this area in particular where the occurrence of red beds is relevant to the waste disposal operation.

I hope this communication clarifies the questions posed by Pollution Control Inc..

Please do not hesitate to call me if I can be of further assistance.

Yours very sincerely,

Daniel B. Stephens, PhD President

DBS/mt

attachments

cc: J. Weber J. Thornton







NICHEM INTERNATIONAL

707 NORTH LEECH P.O.BOX 1499

HOSES, NEW MEXICO 88240

COMPANY : AGUA INC. DATE : 04/18/86 FIELD,LEASEWWELL : LAGUNA PLATA SPRING WATER SAMPLING FOINT: DATE SAMPLED : 04/14/86

SPECIFIC GRAVITY = 1.133 TOTAL DISSOLVED SOLIDS = 196443 PH = 8.01

		ME/L	MG/L	
CATIONS				-
CALCIUM MAGNESIUM BODIUM	(CA)+2 (MG)+2 (NA),CALC.	37.3 298. 2852.	787. 3625. 65379.	•
ANIONS	,			• • • • •
BICARBENATE CARBONATE HYOREXIDE BULFATE	(HCO3)-1 (CO3)-2 (OH)-1 (SO4)-2	7.4 0 0 1082.	451. 0 0 52000	•

SULFATE	•	(S04)-2	1082.	
CHLORIDES		(CL)-1	2100	
-	•			

DISSCLVED GASES

CARBON DIOXIDE Hydrogen Sulfide Cxygen	(C82) (H2S) (02)	· · · · ·	NOT NOT NOT	RUN RUN RUN		
IRON(TOTAL) BARIUM	(FE) (BA)+2		0		 3.6 .28	

MANGANESE	 (MN)	NOT RUN	•	•	
		4			
		•			

ICNIC STRENGTH (MOLAL) =4.176

SCALING INDEX	TEMP
CARBONATE INDEX Calcium Carbonate scaling	30C 86F 1.55 LIKELY
CALCIER IL FATE INDEX Calcier Sulfate Scaling	28.3 Lixely

10011 STRENGTH IS TOO HIGH FOR CARBONATE METHOD

SCIENTIFIC LAN 700 Camino de S Albuquerque, Ni	BORATORY DIVISION Salud NE M 87106 — (505 - 2555 MM	GEI a	NERAL WATER CHEMISTRY	TACIHM ON
DATE RECEIVED 2/0/86 N Soliton DATE 06/02/04 Collection TIME	AB WC 566 USER 59300 CODE 59300 SITE INFORM->	о □ 59600 XX от иТи Seep - b	HER: 82235 (Petro Thers aguna PLata	m)
Collected by - Person/Agency Roy		<u>Cample Fra</u>	AMW 1/4 of NE 1/4 Se	NE CH6
ENVIRONMEN END NM OIL CON INAL State Land PORT Santa Fe, Attn: <u>David Bo</u>	TAL BUREAU SERVATION DIVISION Office Bidg, PO Box 208 NM 87501 yer	8	TOD S, R 30En Grooyon	
	Water level	Discharge	Sample type	
PH (00400)	Conductivity (Uncorrected)	Water Temp. (00010)	•C	µmho
SAMPLE FIELD TREATMEN No. of samples submitted / XN	T — Check proper boxes F: Whole sample (Non-filtered) Γ: Filtered in 0.45 μme	field with I A: 2 m	ni H₂SO₄/L added	
NALYTICAL RESULTS from	n SAMPLES			l
Conductivity (Corrected) 25°C (00095)	Units Date analyze	d F. NA Calcium (00915) Magnésium (00925)	<u>776.0</u> mg/l <u>2-10</u> <u>1507</u> mg/l <u>u</u>	iyzed
Total non-filterable residue (suspended) (00530) Other: Other:	mg/l	Sodium (00930) Potassium (00935) Bicarbonate (00440) Chloride (00940) Sulfate (00945)	$\begin{array}{c c} \hline 11 \ 28 \ 02 \ mg/l	0
□ Other:		 (dissolved) (70300) Other: 	<u>36428</u> mg/1 <u>3/12</u>	<u>s</u>
Nitrate-N + , Nitrate-N total (00630) Ammonia-N total (00610)	mg/l mg/l	F, A-H ₂ SO ₄	ma/i	
	mg/l	 Ammonia-N dissolved (00608) Total Kjeldahl-N () 	mg/l	
()) Other: Other:	mg/l	Analyst	Date Reported Reviewed by 3 24 86 Caller	1
Laboratory remarks		· · · · · · · · · · · · · · · · · · ·		

PROBLEMS WHICH HAVE NOT BEEN ADEQUATELY ADDRESSED BY PETRO-THERMO REPORT

ATTACHMENTI

- 1. The thickness of the alluvial cover is unknown at the proposed site. zo or Within Section 16, the thickness ranges from 0 to 130 feet, but it is Arrow (
- The upper surface of the redbeds is an erosional surface of considerable relief. There is no evidence presented by the report which confirms that the redbed surface slopes directly toward Laguna Plata.
- 3. The report does not disprove work by Reed (1969) which indicates a bedrock channel which would result in a westward migration of ground-water from Laguna Plata (illustration).
- 4. The report, Figure 3, shows that the 3,440-foot contour is closed, thus indicating that Laguna Plata is a closed depression. Data on the map shows no justification for closing the contour. The intrepretation shown in Figure 3 is not supported by work by Hunter (1985) or by Geohydrology Assoc., (1979). (illustrations)
- 5. No evidence is presented in the report which substantiates that the disposal ponds will function properly. In fact the very nature of drilling mud is to cause plugging of natural porosity in sediments.
- 6. Evaporation of fluids should be calculated for surface area of the disposal ponds and NOT for Laguna Plata.
- 7. The report does not contain any chemical analyses of water samples from the fluid which will be disposed. The TDS range is reported to be 25,000 to 75,000 ppm but springs at Laguna Plata have less than 9,000 ppm.
- 8. The concentration of 335,100 ppm reported in report for Laguna Plata is a concentrated brine resulting from emporation on the lake floor or is a residual concentration from potash discharge by Kerr-McGee



50 YEARS

STATE OF NEW MEXICO ENERGY AND MINERALS DEPARTMENT OIL CONSERVATION DIVISION



GOVERNOR

March 21, 1986

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE. NEW MEXICO 87501 (505) 827-5800

Mr. Robert W. Abbott Petro-Thermo Corporation P. O. Box 2069 Hobbs, N.M. 88241-2069

> Re: Request for Extension of Time for Temporary Use of Pit

Dear Mr. Abbott:

This letter is to confirm a telephone conversation today between Jamie Bailey, OCD Field Representative, and John Weber regarding an extension of the permit for temporary disposal of brine, fresh water-based drilling mud, and waste cement in the emergency overflow pit located at the Blinebry-Drinkard SWD System Well No. A-22 in 22-T22S-R37E. Your latest temporary permit expired March 20, 1986.

Due to time restrictions associated with Case No. 8781 which was brought to hearing on December 18, 1985, and will be heard <u>De Novo</u> before the Commission on April 9, 1986, you are hereby granted an extension until April 10, 1986, for the specified temporary use of the pit.

An inspection of the area will be made on April 3, by OCD personnel, and they are authorized to order immediate closure of the pit if conditions warrant.

Authorization for the temporary use of this pit may be rescinded at anytime if there is evidence that the pit has overflowed or if Water Quality Control Commission or Oil Conservation Division rules or regulations have been violated.

Sincerely, R. L. STAMETS

Director

RLS:dp

cc: Jerry Sexton David Boyer Michael Stogner

50 YEARS

STATE OF NEW MEXICO ENERGY AND MINERALS DEPARTMENT OIL CONSERVATION DIVISION





GOVERNOR

March 21, 1986

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87501 (505) 827-5800

Mr. Robert W. Abbott Petro-Thermo Corporation P. O. Box 2069 Hobbs, N.M. 88241-2069

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R. L. STAMETS Director

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cc: Jerry Sexton David Boyer Michael Stogner



STATE OF NEW MEXICO ENERGY AND MINERALS DEPARTMENT DIL CONSERVATION DIVISION



GOVERNOR

February 18, 1986

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE. NEW MEXICO 87501 (505) 827-5800

I.

Mr. John Weber Maddox, Renfrow & Saunders Attorneys at Law P.O. Box 5370 Hobbs, NM 88241

Dear Mr. Weber:

In accordance with Paragraph 2 of Division Order R-8161, a plan for the installation and sampling of monitor wells at the proposed Laguna Plata Petro-Therm site has been discussed by Environmental Bureau Chief David Boyer, Petro-Therm Engineer James Thornton, and consultant hydrologist Dr. Daniel Stephens. Agreement has been reached that three shallow monitor wells will be installed prior to operation, inspected monthly for fluids, and sampled every six months if fluids are detected. The particulars of well location, completion and type of sampling are provided below:

- 1) Two monitoring wells shall be located at a distance no greater than 200 feet north of the north boundary of the 8.264 acre area within Tract B as shown on the attached plat map. These two wells shall be located at distances of approximately 70 and 200 feet east of the west boundary line of Tract B. The third well shall be installed within Tract B to the north of the first two wells at a location to be agreed to after further surface inspection of topographic and geologic features.
- 2) Monitoring wells shall be drilled through the alluvium with the base completed in the first clay, claystone or shale in the redbeds. The wells shall be constructed of 4-inch diameter PVC pipe which is slotted or perforated from a distance of 4 feet beneath the surface to total depth, and shall be adequately gravel packed or otherwise completed to allow fluids to enter the well for sampling, but to prevent silting. The wells shall have the upper four feet cemented to prevent surface fluid entry.
- 3) The wells shall be checked monthly for fluids and the results reported monthly to the Division's office in Santa Fe.

4) Upon detection of fluids in any of the monitoring wells, sampling of these fluids shall take place and be repeated at six-month intervals. Samples shall be analyzed for heavy metals and purgeable aromatic hydrocarbons as listed on the attached sheet. A copy of the results shall be submitted to the Division office in Santa Fe for review as to the nature and threat to human health, if any, of allowing such seepage movement to continue towards Laguna Plata. This review will take into consideration the fact that Laguna Plata is not, and does not have the potential to be, a drinking water source.

The plan described above will satisfy the requirements of Paragraph 2 of the above order. As provided for in the order, the Director of the Division may by administrative order rescind the authorization and/or require additional conditions be met if it is determined that such rescission or additional conditions would serve to protect fresh water supplies from contamination, assure the protection of human health or livestock, and the prevention of waste.

If you have any questions on the monitoring and sampling aspects of this order, please contact Mr. David Boyer at the above address or at 827-5812.

R. L. STAMETS Director

RLS/DB/dp

cc: David Boyer, OCD Santa Fe Jerry Sexton, OCD Hobbs Fran Cherry, BLM Carlsbad Daniel Stephens, Socorro



PETRO - THERM ANALYSIS OF WATER SAMPLES

Water samples from the monitoring wells shall be analyzed for the following dissolved hydrocarbons (BTX):

Benzene o-xylene Ethylbenzene m-xylene Toluene p-xylene

The suggested method is EPA Method 602 which is a purgeable aromatic scan and costs less than the use of a gas chromatograph/mass sprectrometer. Minimum detection limit should be 10 ppb (or 0.01 mg/l). The standard sample is 40 ml collected in a glass vial with a teflon septum seal. No air should be trapped between the water and the seal.

Water samples should be analyzed using an inductively coupled argon plasma scan (ICAP) with a minimum detection limit of 100 ppb (0.1 mg/l). One scan provides concentrations for the following elements:

Aluminum Barium Beryluim Boron Cadmuim Calcium Chronimum Cobalt Copper Iron Lead Magnesium Manganese Molybdenum Nickel Silicon Silver Strontium Tin Vanadium Zinc

In addition samples shall be analyzed for arsenic, and mercury using atomic adsorption methods. Minimum detection levels should be 10 ppb (0.01 mg/l) for arsenic and 1 ppb (0.001 mg/l) for mercury. A single one quart plastic container should be sufficient for all of the heavy metal analyses. Samples should be preserved with 5 ml of concentrated nitric acid.

The use of scans will provide much information on contaminants but is very much less time consuming and expensive than individual analyses. Your consultant can provide you with the names of several laboratories that will provide these services at a reasonable cost. The laboratory selected should also provide further information on sampling and preservation procedures. Contact the OCD or your consultant for the desired method of sampling to prevent false results from being obtained.



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- 3		18 De	cember 1985
4		EXAMI	NER HEARING
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7	IN THE M	ATTER OF:	
8		The application of	Patro-Thermo Cor- Cler
9		peration for an exc Order No. R-3221 an	ception to Division 8781 nd for authoriza-
10	v	tion to dispose of hydrocarbons and of	associated waste ther solids obtianed
11		in conjunction with production of oil a	n the drilling and and gas into a dispo-
12		sal site on the sur New Mexico.	cface, Lea County,
13			
- 14	BEFORE:	Michael E. Stogner	, Examiner
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16	•	(BDBNCODT)	DE UELETIG
17		IRANSCRI	T OF BEARING
18			
19		APPEA	ARANCES
20	For the D)ivision:	Jeff Taylor Attorney at Law
	· ·		Legal Counsel to the Division : Energy and Minerals Dept.
21			Santa Fe, New Mexico 87501
<u>77</u>	For Petro	-Thermo:	Ernest L. Padilla Attorney at Law
23			PADILLA & SNYDER P. O. Box 2523
24	A CAR		Santa Fe, New Mexico 87501
25			John Paul Weber
	JAI	127 585 AD	MADBOX; RENFROW & SAUNDERS P. O. Box 5370
i			NUDDS, NEW MEXICO 88241

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	· 1·	APPEA	RANCES	
	2			
	3	For Snyder Ranches & Pollution Control,	W. Thomas Kellahin Attorney at Law	
	4	Inc.:	KELLAHIN & KELLAHIN P. O. Box 2265	ł
	5		Santa Fe, New Mexic	o 87501
	6			
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5 1 2 MR. STOGNER: This hearing will 3 come to order. We'll now call Case Number 8781. 4 MR. TAYLOR: The application of 5 Petro-Thermo Corporation for an exception to Division Order 6 No. R-3221 and for authorization to dispose of associated 7 waste hydrocarbons and other solids obtained in conjunction 8 with the drilling and production of oil and gas into a dis-9 posal site on the surface, Lea County, New Mexico. 10 MR. STOGNER: We'll now call 11 for appearances. 12 MR. WEBER: Sir, my name is 13 John Paul Weber. I'm with the law firm of Maddox, Renfrow, 14 and Saunders in Hobbs, New Mexico. 15 I appear here today with Mr. 16 Ernest L. Padilla, law firm of Padilla and Snyder, of Santa 17 Fe, on behalf of the applicant, Petro-Thermo Corporation and 18 its Aqua Division. 19 MR. STOGNER: Thank you, Mr. 20 Any other appearances? Weber. 21 KELLAHIN: Yes, Mr. Exami-MR. 22 ner. I'm Tom Kellahin of Santa Fe, New Mexico. I'm appear-23 ing on behalf of Snyder Ranches. Snyder Ranches has grazing 24 leases in the area of the application. 25 In addition, I'm appearing on

6 1 behalf of Pollution Control, Inc., which is the current Oil 2 Conservation Division permitted disposer of produced salt 3 water into Laguna Plata, and so our -- Pollution Control's 4 interest immediately offsets Mr. Abbett's acreage in Section 5 We are adjoining neighbors but we are appearing to see 16. 6 about the operations here. 7 MR. ABBOTT: Is he putting on 8 testimony or introducing himself? 9 MR. KELLAHIN: Those are the 10 parties I represent here today, Mr. Stogner. 11 MR. STOGNER: Thank you, Mr. 12 Kellahin. 13 Are there any other appear-14 ances? 15 Are there any parties who wish 16 to make any statements at this time? 17 Would you please stand up, 18 identify yourself? 19 MS. WOOD: My name is Nancy 20 Wood. I work with the State of New Mexico Historic Preser-21 vation Division. I'm an archaeologist. 22 MR. ABBOTT: Who do you repre 23 sent? 24 MS. WOOD: The Historic Preser-25 vation Division.

7 1 MR. ABBOTT: And what part is 2 that of --3 MR. STOGNER: Please continue. MS. WOOD: Okay. It's part of 5 New Mexico Historic Preservation Division, Cultural Affairs. 6 Our concern is with the protec-7 tion of archaeological sites in the wicinity. There's a 8 number of known important archaeological sites around the 9 vicinity of Laguna Plata. 10 Approximately fourteen of them 11 were determined eligible for the Natural Register of Histor-12 ic Places in approximately 1975. 13 I checked the State archaeolo-14 gical records. We don't know of any known sites that are on 15 this specific area that's proposed for development but given 16 the proximity to the lake margin there is a reasonably good 17 probability that there may be sites (not clearly audible) 18 particularly (not clearly audible). 19 The only way to tell if there 20 are some is if an archaeological survey is done. 21 The other thing that I just 22 point out is that under -- under the State wanted to Cul-23 tural Properties Act it is a violation of the Act to injure 24 or destroy archaeological sites on State lands. 25 The other information that Ι

1 wanted to point out is that it's my understanding that the 2 Bureau of Land Management has some interest in preserving 3 (not understood) Laguna Plata Area as an archaeological the Δ reserve. 5 There would be something more 6 to take up with the Bureau of Land Management, but that's my 7 understanding, and one of our -- one of our responsibilities 8 under the law is to inform other State agencies and local 9 governments of possible effects that their actions may have 10 on (not audible clearly) resources. 11 Our usual recommendation for 12 something like this is an archaeological survey should be 13 done to try to protect these resources. 14 Thank you, Ms. STOGNER: MR. 15 wood. Is that everything you have? 16 MS. WOOD: Yeah, it is. Thank 17 you. 18 MR. STOGNER: Are there any 19 other statements? 20 Ms. Wood, is your MR. TAYLOR: 21 agency recommending that a cultural survey be made? Is that 22 a recommendation or are you just saying that's something you 23 sometimes do? 24 MS. WOOD: Okay, it is a fairly 25 standard recommendation that we do make. There is no re-

8

9 1 quirement under State law that surveys be done but it is a 2 recommendation that I would make in order to protect ar-3 chaeological ---4 MR. TAYLOR: Are you making it, 5 then? Is that what you're saying? 6 MS. WOOD: Yes, I am making 7 that recommendation. 8 MR. TAYLOR: And you're with 9 the Division of Cultural Affairs of the Historical Preserva-10 tion -- what's the rest of the name of it? 11 WOOD: Just Historic Pre-MS. 12 servation Division. 13 MR. TAYLOR: Historic Preserva-14 tion Department? 15 MS. WOOD: Division. 16 MR. TAYLOR: Division of the 17 Division of Cultural Affairs? 18 MS. WOOD: Office. 19 MR. TAYLOR: The Office of Cul-20 tural Affairs. 21 MS. WOOD: Yeah. 22 MR. TAYLOR: I have to get all 23 these divisions and offices straight. 24 Okay, thank you. We'll make 25 that a part of the record.

10 1 MR. STOGNER: Okay, are there 2 any other statements at this time before we get started? 3 There being none, will all wit-4 nesses at this time please stand and be sworn? 5 6 (Witnesses sworn.) 7 8 STOGNER: MR. Weber, you Mr. 9 may proceed. 10 MR. WEBER: Sir, I would like 11 to make a brief opening statement. 12 MR. STOGNER: Okay. 13 MR. WEBER: Petro-Thermo Cor-14 poration comes here today to seek an exception to Division 15 Order R-3221, the general "no pit" order which was entered 16 on the 1st of May, 1967, by the Oil Conservation. 17 There has been a real need in 18 southeastern New Mexico for additional approved sites for 19 disposal of oilfield related liquids and solids. This need 20 has been communicated to Petro-Thermo Corporation and they 21 have very actively searched the area to find a suitable 22 site, a site that would not contaminate any existing fresh 23 water supplies. 24 We feel, and we'd like to re-25 view the regulatory history relating to this particular area

11 1 and request that you take administrative notice of the var-2 ious orders that have been entered with regard to this area 3 in order to understand why we feel that this is the best 4 possible site in southeastern New Mexico for this sort of 5 disposal facility. 6 By order dated July 25, 1968, 7 that was Order No. R-3221-B in Case Number 3806, the Oil 8 Conservation Commission exempted certain areas of Lea Coun-9 ty, New Mexico, from the prohibition against disposal of 10 production water in unlined surface pits. 11 Among the areas exempted was 12 Range -- correction, Section 16, Township 20 South, Range 32 13 East. 14 The Commission at that time 15 thought that the purpose of Order No. R-3221 would not be 16 (not understood) by its enforcement in this area. 17 I would ask you to note that 18 the proposed disposal site is located within the bounds of 19 Section 16. 20 By letter dated April 16, 1969, 21 the Oil Conservation Commission again considered this parti-22 cular area. They considered it in Case Number 4047, result-23 ing in Order R-3725. 24 By this order the Oil Conserva-25 tion Commission specifically permitted the disposal of pro-

12 1 duction water in an actual salt lake known as Laguna Plata 2 in Lea County, New Mexico. The Commission found that the 3 utilization of Laguna Plata for the disposal of production 4 water would not constitute a hazard to existing fresh water 5 supplies in the area. 6 Examiner, That, Mr. is the : 7 history . We feel that there is a definite legislative 8 need, a need which we will show through the testimony of our 9 witnesses. 10 We will show that we have 11 developed detailed engineering plans to eliminate any 12 possibility of contamination of existing fresh water 13 supplies. 14 We have very carefully examined 15 the geology the topography of the area, and have reached the 16 conclusion that once again the disposal would not constitute 17 a hazard to existing fresh water supplies. 18 Sir, at this point I would like 19 to call as our first witness Mr. Abbott, the President of 20 Petro-Thermo Corporation. 21 MR. STOGNER: Before we do 22 that, Mr. Weber, let me make sure I've got this clear. 23 You referred to Case Number 24 3806 and 4047, I believe? 25 MR. WEBER: That is correct.

13 Ŧ I will take MR. STOGNER: 2 administrative notice of both of those cases. 3 Please continue. 4 5 WILLIAM G. ABBOTT, 6 being called as a witness and being duly sworn upon his 7 oath, testified as follows, to-wit: 8 9 DIRECT EXAMINATION 10 BY MR. WEBER: 11 0 Sir, would you please state your full 12 name? 13 Yeah. My name is William G. Abbott and I A 14 live in Hobbs, New Mexico. I'm ---15 Sir, what is your -- sir, what is your Q 16 profession? 17 I'm a petroleum engineer. Α 18 0 From what institution did you receive 19 your undergraduate degree, sir? 20 I received by degree from the University 21 of Texas in January of 1948. 22 Q Sir, what was your specialty or area of 23 concentration? 24 Actually my degree is in mechanical A 25 engineering but I specialized in petroleum engineering.

1 Sir, are you a member of any professional 0 2 societies or organizations? 3 Yes. I belong to the Society of Profes-Α 4 sional Engineers, Society of Petroleum Engineers, and API, 5 American Petroleum Institute. 6 Sir, do you possess any licenses as an 0 7 engineer? 8 I'm licensed as a professional en-Yes. A 9 gineer in the State of New Mexico; also the State of Texas. 10 Sir, how long have you been a resident of 0 11 Hobbs, New Mexico? 12 I've lived in Hobbs, New Mexico since 13 1951. 14 Sir, could you please explain your work 15 history? 16 Yes. I was transferred to Hobbs. New 17 Mexico, in 1951 with Amerada Petroleum Corporation, and I 18 stayed with Amerada Petroleum Corporation until the middle 19 of 1957 when I went to work as Manager of Rice Engineering 20 in Hobbs, New Mexico. 21 Sir, what sorts of things Ò does Rice 22 Engineering do? 23 Well, Rice Engineering specializes in 24 salt water disposal and in 1967 I formed my own corporation, 25 Agua, Incorporated, to specialize in the salt water dispo-

15 1 sal. 2 All right, sir. Sir, have you ever had 0 3 an opportunity to testify before the Oil Conservation 4 Commission? 5 Yes, sir. 6 In what capacity did you so testify? Q. 7 With Amerada I was District Engineer and 8 testified and then with Rice Engineering as Division Mana-9 ger, and then with Agua as President. 10 Sir, were your qualifications accepted by 11 the Oil Conservation Commission? 12 Yes, they were. A 13 MR. WEBER: Sir, I tender this 14 witness as an engineer. 15 MR. STOGNER: Are there any 16 objections? 17 MR. KELLAHIN: No, sir. 18 MR. STOGNER: Thank you. There 19 being none, Mr. Abbott is so qualified. 20 Mr. Abbott, would you please describe in 21 general terms the history of Petro-Thermo Corporation? 22 Yes. We formed Petro-Thermo Corporation A 23 in 1970. We had a problem in -- during that time of 24 disposing of oilfield waste, namely oil and BS, tank 25 bottoms, and we formed Petro-Thermo to clean tanks and then

1 later on we got authority to -- from the Corporation Commis-2 sion to be permitted to haul oilfield waste. 3 Sir, is Petro-Thermo Corporation a common 0 4 motor carrier operating under a Certificate of Public 5 Convenience and Necessity issued by the State Corporation 6 Commission? 7 A · Yes, sir, they are. 8 And do you also have authorization to 0 9 move produced water from the Oil Conservation Commission? 10 Yes. A 11 Sir, in what counties and if you could 0 12 please use the maps which are posted on the wall as exhi-13 bits. 14 I think Exhibit One shows our counties of 15 authority cross hatched. You'll notice the full east side 16 of New Mexico, starting up here in Union County. We have a 17 trucking yard at Clayton, New Mexico, up in Union County, 18 and then we have all these counties cross hatched and we 19 have another terminal and main office at Hobbs, New Mexico. 20 Sir, do you also have a division of Q 21 Petro-Thermo Corporation called Agua, do you not? 22 Yes, sir, A 23 And what does Agua do and where does it Q 24 do it? 25 Oh, well, we operate disposal systems. A

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17 1 been hired by the oil companies to dispose of We've their 2 water in the various systems and Agua operates those sys-3 the largest being down at Eunice, the Blinebry-Drintems, 4 kard salt water disposal system, with approximately 550 5 wells. 6 Sir, that particular system is covered by 7 a temporary permit, is it not, sir? 8 A Well, we have a temporary permit in that 9 area issued by the Oil Conservation Division to dispose of 10 I can't remember when it runs out. It was a solids. 11 temporary, about a sixty day permit, so we need to proceed 12 and to get a permanent place for the disposal of solids or 13 waste. 14 Sir. would you please describe the 15 reclaiming operations that are undertaken by Agua, 16 Incorporated? 17 A Yes. We -- we operate a reclaiming 18 plant. It's called our Goodwin Reclaiming Plant where we 19 treat the tank bottoms to try to get a merchantable oil to 20 sell as pipeline oil. 21 We also sell from -- from that spot 22 drilling fluid additives. That's low. low grade 23 hydrocarbons that are used in the drilling industry when 24 they drill a well to add to the drilling mud. 25 Sir, do the reclaiming operations that Ø

18 1 you undertake, or that Agua, Incorporated, undertakes result 2 in the conservation of valuable energy resources? 3 A Yes. We produce a merchantable product 4 and sell it, it's taxed, and we think it does prevent waste 5 in the oilfield. 6 0 Sir, you've indicated that Petro-Thermo 7 Corporation in its trucking operation serves Lea, Eddy, and 8 Chaves Counties. 9 Will you please tell us the approximate 10 number of oil wells in those counties? 11 Yes, as you can see from this Exhibit A 12 Two, all these wells, most of them are in Lea County, coming 13 from Hobbs and covering the whole of Lea County, Lea County 14 is approximately 80 miles long and 40 miles wide and in the 15 county there's over 15,000 oil and gas wells. 16 In Eddy County, they're growing. There's 17 approximately 7200, over 7200 wells. 18 And in Chaves County there's about 2200 19 wells. 20 And if you'll notice, this proposed waste 21 disposal site is right in the middle of all the wells. It 22 can be reached by main highways and is a very good place to 23 haul to. 24 Sir, could you, using Exhibit Number Two 0 25 and the following exhibit, Exhibit Number Three, describe

what is meant by the Hobbs Pool?

A Yes. The Hobbs Pool is a pool right surrounding and in the City of Hobbs. About one third of the
wells are in the city limits of Hobbs. There are over -there are about 480 wells right now in the Hobbs GrayburgSan Andres Pools.

7 Those -- that pool is unitized, the north 8 end being operated by Shell Oil Company and the south end 9 unitized and operated by Amoco.

10 And I -- I can see that in the future,
11 within -- if the price of oil stays up. that this waterflood
12 will expand into CO2 flood and also they'll probably double
13 the number of wells drilled in the Hobbs' Pool.

14 Q So these are what are called tertiary 15 means of recovery?

16 A Yes, that's right. Of course, when they 17 drill these wells in this area especially, all the cuttings 18 and all the mud has to be hauled out; they can't be left in 19 the pits, be covered over; they have to be hauled away and 20 that's why a solids waste area is needed.

21 Q So where are they hauled away to?
22 A Well, right now there's only two author23 ized places that I know of. One is an area down north and
24 east of Eunice called Parabo. Parabo is limited somewhat
25 because of the geology and the hydrology. It will probably

19

20 be shut down this -- this winter because the water level 2 gets too high. When that happens they just have to shut it 3 down. 4 The other site is out at the Laguna 5 Gatuna. 6 Q Who operates that site, sir? 7 A Laguna Gatuna operates -- is operated by 8 what do you call it -- I guess the Pollution Control, yeah, 9 Incorporated. 10 Q Are there any limitations placed on the 11 amount of waste which can be disposed of in Laguna Gatuna? 12 Well, I think their order reads 30,000 A 13 barrels day, or something like that. I don't know if they 14 could ever approach that capacity. I just don't have the 15 knowledge. 16 Q Sir, have you received ---17 I can point it out here. A This is Laguna 18 Gatuna, this area. 19 MR. STOGNER: What exhibit are 20 you referring to? A. 46 21 MR. WEBER: Sir, that was Exhi-22 bit Four. 23 A Four. 24 MR. STOGNER: And on Exhibit 25 Four what did you point to?

21 1 A Laguna Gatuna. 2 MR. STOGNER: It's on the far 3 east side --4 A Yeah, that's where Pollution --5 MR. STOGNER: -- of the map. 6 A -- Control operates (not understood) 7 site, or disposal site, and on the same Exhibit Four it 8 shows Laguna Plata. 9 STOGNER: Thank you, MR. Mr. 10 Abbott. 11 · · · O Sir, have you received from any other 12 operators of oil and gas wells, or producers, any indication 13 of for additional disposal facilities need in 👘 the 14 southeastern portion of New Mexico? 15 Yes, we've received copies of letter A 16 submitted to the Oil Conservation Division. I don't know 17 the total number of them, since we haven't received copies 18 of all of them. 19 Sir, I show you three letters which have 20 been marked Five -- Exhibits Five, Six, and Seven, and ask 21 you if you can recognize those? 22 Yes. This first one is a letter A from 23 Amerada Hess Corporation; the second is a letter from Amoco 24 Production Company; and the third one is Natural Resources 25 Engineering, Inc...

22 Mr. Abbott, would you take Exhibit Number Q 2 Five and read that letter into the record? 3 This is the ---MR. **KELLAHIN:** I'm going to 5 object, Mr. Examiner, to a letter being read into the 6 record as part of the evidence in this case. 7 I think the typical way the 8 Commission handles communications from nonparticipants in a 9 hearing is to place them in your case file and read them as 10 a practical matter, but we object to those things being 11 introduced into evidence without a proper foundation, and 12 none has thus far been made. 13 MR. WEBER: Mr. Examiner. I 14 feel that a proper foundation has in fact been laid. These 15 are letters from oil and gas operators in Hobbs, Lea County, 16 New Mexico area. They have all, true, been sent directly 17 to the Oil Conservation Division; however, carbon copies of 18 those letters have been sent to Mr. Abbott in his capacity 19 as President of Petro-Thermo Corporation. 20 MR. **KELLAHIN:** The letters 21 speak for themselves, Mr. Examiner, there's no need to read 22 them into the record to highlight Mr. Abbott's testimony. 23 It's inappropriate. 24 WEBER: Perhaps I can use MR. 25 them in a different way.

Q In very general terms, what factors have been indicated to you as the important factors with regard to the need for an additional disposal site?

Well, the need, of course, is generated 5 by the volume that will have to be disposed of and also, in 6 the last two years, as the Conservation Division realizes, 7 there have been some very serious salt water flows. In 8 fact, we've been engaged to haul concentrated brine and Red-9 beds, and so on, from wells that are flowing out of control, 10 and that -- that has to be disposed of right away because 11 it's so -- so corrosive and so full of chlorides.

In fact on one -- one occasion, this well was flowing at such a high rate that there were over forth transports hauling 24 hours a day to keep up with the flow, and to my knowledge, I think the well is still flowing, but I think it's under control now, they're disposal downhole.

There is another area just recently, just south, I think it was mostly east of Lovington, same way, a drilling well, and salt water started flowing out of that well and that had to be hauled off.

Now this, this is very difficult because it isn't clean salt water. I mean it isn't water that you (not understood) a disposal well. It's water laden with Redbed solids and so on, and you couldn't use it to be disposed in a disposal system; it would plug up the wells.

24 So sites are needed such as this Laguna 2 Plata site that we propose. 3 Sir, faced with the foreseen need in the 4 oil and gas community, what steps did you take in determin-5 ing the location of your proposed disposal site? 6 Well, first of all, we -- we needed a Ά 7 centrally located area, one that we could see from our view 8 it wouldn't pollute any water, and it would cause the least 9 disturbance to -- to people, and that's why we selected this 10 Laguna Plata. 11 0 Is this a very populated area? 12 No, it's very, very sparsely populated. A 13 What consideration did you give to the 14 existing road map? Will the road map support the movement 15 of transport trucks? 16 λ Yes. I think the roads coming into this 17 area, you have a 176 State highway coming up from the Eunice 18 area, which is very, very important because there are a lot 19 of wells in the Eunice area. 20 Then up from the north there's a road 21 that ties into the area of Maljamar, which is being water-22 flooded at the present time and if one of those wells got 23 away they'd have to take it some place and this would be 24 centrally located. 25 It's also located close to the Eddy Coun-

25 1 ty wells. 2 And, sir, is there not an existing 0 3 caliche road which leads almost up to the point of the dis-4 posal site? 5 There was a dry hole drilled by TXO Å Yes. 6 and they built a caliche road within just a short distance, 7 probably 400-500 feet from this proposed site of ours. 8 Sir, to your knowledge has Laguna Plata 9 been previously approved for the disposal of waste? 10 A That's what I understond, yes, sir. 11 Q Do you know if it's now being so used? 12 No, sir. Å. 13 Sir, for what capacity did you direct the 14 design for the disposal facility? 15 We designed it for 30,000 barrels a day 16 of water. I don't know if we'll ever reach that, but it's 17 over-designed as far as the pits that we -- and the tanks. 18 0 Sir, what is your actually immediate 19 needs? 20 I think probably around 2200 barrels a A 21 day of produced water. 22 Sir, how quickly could you put into place 23 a disposal facility once the necessary approvals have been 24 granted? 25 Ά I think we could do it in thirty to sixty

26 1 days. 2 Sir, what sophisticated design features 0 3 have you directed to be incorporated in your plans to avoid 4 the possibility of contamination of existing fresh water 5 supplies? 6 Well, I think our engineer probably could A 7 go into more detail, and the hydrologist, but on the Exhibit 8 Five is our layout of the pits. The pits on the left on 9 this Exhibit Five are the water pits, and also on Exhibit 10 Nine we have a model that shows those pits. 11 MR. STOGNER: Excuse me, I show 12 Exhibit Five as being a letter. 13 Is that Exhibit Eight you're 14 referring to, I believe? 15 MR. WEBER: That is correct, 16 Mr. Examiner. 17 A Yeah, Exhibit Eight, excuse me. 18 MR. STOGNER: Thank you. 19 Exhibit Nine is a model made from the to-20 pographic map of the area and as you can visualize, the 21 tankage will be up here on this pad. The entrance to this 22 area will be right up here. 23 MR. STOGNER: Now, are you re-24 ferring to the extreme southwest portion of --25 Yes.

27 1 MR. STOGNER: -- Exhibit Number 2 Eight, is that right? 3 A That's right. This -- this Exhibit Nine 4 is a 600 by 600 foot plat and it's located in the northeast, 5 it's in the east half of the northeast quarter of Section 6 16, 20, 30. 7 0 2. 8 32, yeah, and if you'll notice on this 9 model, we will locate the gunbarrels and oil storage tanks 10 up on the pad. The water will go on into these pits and 11 that will be described in detail from -- with our engineer. 12 The solids are roughly on the righthand 13 side, the pits on the righthand side. That has been de-14 signed so that the solid pits can be cleaned out by dozers 15 or backhoes. 16 STOGNER: Before we leave MR. 17 Exhibits Eight and Nine, these contours that you show on 18 both of these, are those one-foot intervals? 19 MR. THORNTON: Yes. Yes. 20 MR. STOGNER: Who is telling me 21 that? Who are you? 22 A Mr. Jim Thornton. 23 Our engineer. A 24 STOGNER: So this is some-MR. 25 what exaggerated.

28 ۱ A Yes. 2 MR. STOGNER: I don't remember 3 it being that steep out there. 4 Thank you, Mr. Abbott. You may 5 continue. 6 A That's all right. 7 Mr. Abbott, do you presently have under Q 8 consideration by the State Land Office an application for a 9 business lease regarding this property? 10 A Yes. I understand we have applied for a 11 business lease. 12 Sir, in your opinion will approval of the 0 13 proposed disposal site satisfy existing need in the oil and 14 gas industry? 15 Yes, I think that site is needed, A 16 especially since Parabo is -- I don't think it will survive 17 the winter and they'll -- they'll -- everybody needs another 18 spot. 19 Given the fact that you have an oil 0 20 reclaiming operation, will the use of this particular 21 facility serve the needs of conservation of valuable natural 22 resources? 23 We've found that when produced Δ Yes. 24 water is hauled to a tank and separated properly, there will 25 -- we will recover some waste oil.

29 1 MR. WEBER: Mr. Examiner, 2 have no further questions of this witness. 3 Thank you, MR. STOGNER: Mr. 4 Weber. 5 Mr. Kellahin, your witness. 6 KELLAHIN: MR. Thank you, Mr. 7 Stogner. 8 9 CROSS EXAMINATION 10 BY MR. KELLAHIN: 11 Mr. Abbott, you've described for us your 0 12 coprorate structure in terms of Agua, Inc. and Petro-Thermo, 13 and you described those in terms of "we". Are there other 14 principals besides you in either of those companies? 15 A Yes, it's incorporated. There are other 16 stockholders. 17 Are you the principal managing executive 0 18 for both of those companies? 19 Yes. A 20 Who are the other major principals, Mr. 21 Abbott, that would participate with you in making decisions 22 about the construction and location of this type of facil-23 ity? 24 Well, we have a management team working A Sec. 25 for Agua and Petro-Thermo: Myself, Jim Thornton, our engin-

30 1 eer, and my two sons, Bob and Jim. 2 Is the proposed use of this facility one 0 3 that is confined to allowing you and your trucking operation 4 to have a facility to dispose of these solids and oilfield 5 waste or do you propose to make this a public facility for 6 the industry? 7 No, we will make it a public facility for Α 8 the industry. This is the oil industry. 9 I understand. Q 10 Right. A 11 Q Other truckers and haulers and --12 A Yes. 13 Q -- other disposers on some financial 14 basis --15 Yes, sir. A 16 -- with your facility? Exactly what Q 17 substances would be disposed of in the various pits as 18 indicated on Exhibit Number Eight? 19 Well, I -- the details I'll leave to my 20 engineer, but roughly, the pits, as pointed out on the left 21 of our exhibit, will be water pits, produced water pits. 22 The pits on the righthand side of the 23 exhibit would be the solids pits; that is, cement, drilling 24 muds, and that sort of thing. 25 Do you propose to utilize the facility

31 1 for all of the produced water, tank bottoms, oil and BS that 2 you now accumulate, do you propose to take all those sub-3 stances and run it through this unit up to whatever the max-4 imum requested is? 5 Actually, we'll probably haul it off and A 6 haul any -- any merchantable oil to our treating plant. 7 Besides these pits we'll have tanks. 8 We'll have two 1500 barrel gunbarrels or -- and then an oil 9 storage tank for the BS and oil on location. 10 Are you familiar with the -- on Exhibit 11 Number Four, Mr. Abbott, you've shown us the approximate lo-12 cation of your facility to the south and west of Laguna Pla-13 ta. 14 Yes, sir. 15 And you've identified for us the area to 16 the east at Laguna Gatuna where Pollution Control operates a 17 facility, a disposal facility. 18 1 **A** Yes, sir. 19 All right, are you familiar with the Pol-0 20 lution Control facility at Laguna Gatuna? 21 No, I've never been out there. My engin-22 eers and other management have. 23 Are you aware, or have you been informed 24 of how your operation compares or differs with the facility 25 that's in place at Laguna Gatuna operated by Pollution Con-

32 ١ trol? 2 No, I'm not familiar at all with it. Α 3 Do you currently have trucking disposal 0 4 commitments from Amerada Hess, Amoco, and this Natural Re-5 sources, Inc.? 6 A We disposed -- we've worked for all three 7 of the companies, yes, sir, but we don't have any planned, 8 steady work. We're just available for their hauling. 9 Currently, without this proposed facil-10 ity, Mr. Abbott, what are you doing as a hauler with these 11 solids and oilfield waste? 12 Well, our -- right now our disposal water 13 we have a disposal well in -- at our treating plant and we 14 dispose of the produced water there. 15 We also haul in tank bottoms and any 16 any BS, basic sediments, and use our -- our treating plant 17 at -- that we -- at our Goodwin Treating Plant, but as far 18 as the muds, Redbeds, and so on, we just have a temporary 19 permit for a pit down at Eunice, which we're disposing in. 20 Do you propose to divert proposed water 21 from any of your existing disposal facilities and move them 22 to the Laguna Plata proposed site? 23 No, sir, just emergencies, and I don't 24 see any in the future. 25 Have you or your company utilized in the Q

1 past the Pollution Control disposal facilities at Laguna Ga-2 tuna? 3 Yes, sir, we've -- we've used them. 0 Do you know whether or not that existing 5 facility continues to have the capacity to meet the need 6 that you propose to fill with your facility? 7 No, sir, I don't know. A 8 0 You don't know that? 9 No. 10 With regards to your facility, and I've 0 11 lost the section, I think it was Section 16? 12 A Yes, sir. 13 All right, within Section 16, and we're 14 looking at the northeast quarter and then again on Exhibit A 15 it's approximately the east half of the northeast quarter, 16 that 80-acre tract there ---17 Yes, sir. A 18 All right, when we look at that site, 19 whose ownership is the surface subject to? 20 That's the State of New Mexico. A 21 Ő Okay. You've talked about going through 22 permitting procedures to acquire necessary permits to uti-23 lize this facility, Mr. Abbott. Apart from the Oil Conser-24 vation Division approval, what other permits or authorities 25 are you aware are required of you before you commence opera-

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34 1 tions? 2 I don't know. There's probably others A 3 but I am not aware of them. 4 0 You've mentioned to us that you have 5 filed for a business lease from the State of New Mexico to 6 utilize the surface? 7 Yes, sir. A 8 Q I assume you're aware that that's a re-9 quirement for the site? 10 Yes, sir. A ... 11 Are there any permits that you are re-0 12 quired from the Corporation Commission, to your knowledge? 13 No, sir. A 14 Your trucking permits from them are not Q 15 affected by --16 No, sir. A 17 Q -- this operation? 18 Not that I know of. Α 19 All right. And I believe I understood Q 20 you to tell us that at this point the business permit from 21 the State of New Mexico has not been issued to you? 22 We've applied. A 23 In what name, sir, have you applied? 0 Do 24 you recall? 25 No.

35 1 Do you currently have any knowledge as to Q 2 what the surface of this tract is being used for now by the 3 State of New Mexico? It's just a grazing lease, as far as Ά I 5 know. 6 Do you know, sir, who the current grazing Q 7 lessee is for that tract? 8 Yes. We -- I think we -- one of our man-9 agement team contacted the present grazing lessee. 10 0 Do you know the name of the current graz-11 ing lessee? 12 No, I don't remember his name. A 13 With regards to the construction of the Ó 14 facility, as shown on Exhibit Eight, Mr. Abbott, where is 15 that, those pits in relation to the high water mark for La-16 guna Plata? 17 I think our engineer will show in his A 18 testimony. 19 Q Do you know of your own knowledge approx-20 imately where that might be? 21 Yeah, it's probably the north one-sixth 22 of the north tract, which we call Tract A; 40-acre Tract A. 23 0 Is the plan for construction of the fac-24 ility one in which you propose to confine the produced wa-25 ters and the discharges substances within the area of the pits

36 1 shown in Exhibit Eight and Nine? 2 Yes, sir, at the present time. Α 3 Are you seeking authority to dispose of 0 4 salt water into Laguna Plata? 5 Α No, sir. 6 You told us, Mr. Abbott, that the facil-7 ity would be available for general use. 8 A : Yes, sir. 9 Have you come up with some charges for 10 the use of that facility at this point? 11 No, we -- we haven't made any firm char-A 12 ges. 13 Q What is charged to you by the facility 14 that you said was virtually full? 15 At Parabo? A 16 Parabo. 17 Yeah. 18 That's it. Q 19 I think the solids charge there is \$1.00 A 20 a barrel. 21 Do the econmics of the cost to you as 0 a 22 hauler or to other haulers, to your knowledge, do those 23 costs bear in the decision you've made about the siting of 24 this facility? 25 A Yes. We're familiar with other areas in

1 and so on. The oil industry needs a place to -- to Texas. 2 put these solids and I think this -- this -- the location of 3 our proposed site is -- is -- will help the whole oil indus-4 try. 5 How long have you been in this type of Q 6 business, Mr. Abbott? 7 I've been in the disposal end of it since A 8 probably 1948 when I was working with Amerada. I worked on 9 my first salt water disposal system down in the Gulf Coast 10 with Amerada and then subsequently worked on some with Amer-11 ada in Lea County. 12 And in conducting your salt water dispo-13 sal operations have you utilized for the waste disposal, the 14 salt waste disposal, have you utilized any other sites other 15 than Pollution Control or the Parabo sites? 16 Yes, this temporary site down at Eunice. 17 That's the only two sites that I know of. 18 And is the temporary site at Eunice the 19 one that you're currently utilizing to hold the solids that 20 are generated from your business? 21 No, not from our business; generated by A 22 our hauling business from other operators. 23 So your business now for disposing of the 0 24 solids, you're utilizing Pollution Control and the Parabo 25 sites at present?

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38 1 No, we were. A 2 Oh. 0 3 But we're not now. A 4 You've indicated to us that the Parabo 0 5 site, and you think it has perhaps reachd its limit in terms 6 of having water volumes that are too high --7 Yes. A 8 Does that facility still have the capa-9 city to take the solid wastes? 10 I don't known about the solid wastes. 11 You've indicated to us that the Pollution 12 Control site at Laguna Gatuna, that you're not aware that 13 that's full? 14 No, sir, I'm not aware of that. A 15 MR. KELLAHIN: May I have a 16 minute? 17 MR. STOGNER: I'm ready whenever 18 you're ready. 19 MR. KBLLAHIN: Thank you, sir. 20 Are you familiar enough with the opera-21 at the site, Mr. Abbott, that you could take tions me 22 through in a general way how the site is to be utilized by a 23 trucker that brings salt water to you and tank bottoms? Can 24 you give me a general idea of how it runs through the sys-25 tem?

39 1 A Yes, sir, I believe so in a general way. 2 The -- this is shown on Exhibit Eight. 3 The produced water pits are shown here on the left. The en-4 trance to our -- our facility is right here. This would be 5 the entrance and the trucks would come in here and unload at 6 the gunbarrels. They'd unload into two 1500-barrel -- or 7 750-barrel gunbarrels, excuse me, and the water would flow 8 out of the syphons from these gunbarrels, there would be two 9 unloading places, would flow into this pit. 10 MR. STOGNER: The pit marked W-11 12 12 Yes. 13 MR. STOGNER: Okay. 14 This pit is designed so that the connec-Ά 15 tion between W-1 and W-2 would be a large conduit below the 16 waterline and the water would flow into W-2. 17 We've designed it so that the water would 18 have to zigzag through these pits, which will slow the velo-19 city down in the pits and the solids would drop out and also 20 any oil that was trapped with the water could be removed. 21 Of course, most of the oil will come off 22 right at the gunbarrels. 23 Our engineer can go into more detail but 24 that's just generally -- i 25 Just a follow-up question, 0 Mr. Abbott,

40 . 1 once the water hits the W-1 and W-2, you build up, I assume, 2 some solid sediments in those bottoms, you periodically 3 clean them, and what do you do with the stuff after you 4 clean them? 5 We'll probably get some build-up. A I 6 We haven't -- we haven't -- I don't know how don't know. 7 much build-up we're going to have, really. 8 ê e Thank you, Mr. Abbott. 0 9 MR. STOGNER: Mr. Weber, any 10 redirect? 11 MR. WEBER: Yes, sir, I have a 12 few questions. 13 14 REDIRECT EXAMINATION 15 BY MR. WEBER: 16 Just to bring you to a Q point of 17 reference, Mr. Kellahin in his questions asked whether or 18 not Petro-Thermo Corporation and its trucking operations. 19 dealt with the three companies whose letters you have exa-20 mined here today. 21 Yes, sir. A 22 Are these, to your knowledge, all the 0 23 letters that have been sent? 24 No, there may be more that they sent di-А 25 rectly and failed to send us a copy. I have no knowledge.
1 Q Sir, if you could give us an estimate of 2 the number of companies which have trucking operations car-3 rying oil and gas field related waste?

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A Well, in the immediate area of southeastern New Mexico there are probably forty different trucking
companies. I have't added them up. They run all the way
from Jal in the south up to Crossroads on the north and also
to the west in the Carlsbad area and to the north at Maljamar, or west, northwest.

10 Q Is it usual that these transportation 11 companies would be wedded to one particular producer of oil 12 and gas for all their hauling of wastes?

13 A No, I don't believe so. From my know14 ledge of the trucking industry in Lea County, most of it's
15 divided up and it's according to the best service in that
16 they're the ones who get the business.

17 Q You talked about the very small number of
18 approved waste disposal sites in southeastern New Mexico.

Assume for a moment that your temporary exception for the pit in Eunice expires. Assume for a moment that Parabo is no longer able to accept waste, and assume for one reason or another Laguna Gatuna is not available to you or to other of the 39, the 40 or so, truckers in the area.

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What alternatives do they then have?

42 1 The only alternative is to come to A the 2 Conservation Division and ask for approval of a site. 3 Are there any additional sites, say, 0 in 4 Texas or in other areas? 5 Δ I believe there are some sites in Texas 6 but you'd have to have trucking authority to truck the waste 7 to Texas, and we don't. We have some authority in Texas but 8 not extensively. 9 What ramifications if all those things 0 10 occur would that have on the oil and gas industry in south-11 east New Mexico? 12 A Well, it would be -- it would be disas-13 trous. They'd have to shut-in some wells. 14 And what would happen to the 40 truckers? Q 15 Well, they'd go broke. Ά 16 MR. WEBER: I have no further 17 questions. 18 MR. STOGNER: Thank you, Mr. 19 Weber. 20 Kellahin, any more cross Mr. 21 examination? 22 Are there any other questions 23 of Mr. Abbott? 24 I'm going to waive cross exam-25 ining Mr. Abbott at this time. I reserve the right to re-

43 1 call him at a future time. 2 MR. WEBER: Mr. Examiner, at 3 this time I'd like to call Mr. James Thornton as our second 4 witness. 5 6 JAMES D. THORNTON, 7 being called as a witness and being duly sworn upon his 8 oath, testified as follows, to-wit: 9 10 DIRECT EXAMINATION 11 BY MR. WEBER: 12 Q Sir, would you please state your full 13 name? 14 James Douglas Thornton. A 15 And where do you reside? Q 16 A Hobbs, New Mexico. 17 0 Mr. Thornton, by whom are you employed? 18 Α Agua Division of Petro-Thermo. 19 0 And in what capacity are you employed? 20 I'm an engineer. A 21 How long have you been employed as an en-Q 22 gineer with Petro-Thermo? 23 Seven months. A 24 0 What are your general duties and respon-25 sibilities at Petro-Thermo?

44 1 A Oh, the operation and engineering and de-2 sign of several salt water disposal systems that Agua oper-3 ates. 4 Mr. Thornton, where did you receive your 0 5 undergraduate degree? 6 A Texas A & M. 7 Q And what degree did you receive and when 8 did you receive it? 9 A Petroleum engineering degree in December, 10 1984. 11 And petroleum engineering was your spe-12 cialty or area of concentration? 13 Yes, sir. Ä 14 Q Are you a member of any professional so-15 cieties or organizations? 16 A · Yes, sir, I'm a Junior Member of the So-17 ciety of Petroleum Engineers. 18 WEBER: Sir, at this point MR. 19 we would ask that Mr. Thornton be qualified as an engineer. 20 MR. STOGNER: Are there any ob-21 jections? 22 MR. KELLAHIN: No objection. 23 MR. STOGNER: Mr. Thornton is 24 so qualified. 25 Thornton, as part of your general Q Mr.

45 1 duties and responsibilities, were you responsible for devel-2 oping the engineering plans for the proposed disposal facil-3 ity at Laguna Plata? Α Yes, I was. 5 0 Are those engineering plans, are they 6 shown by an exhibit which is on the wall? 7 Yes, they are. It's Exhibit Number А 8 Eight. 9 Q. Was that exhibit prepared by you or under 10 your supervision? 11 A Yes, it was prepared by myself. 12 Did you also prepare a model? Q 13 A Yes, I did, a topographic model of the 14 area. 15 And is that the model which has been num-16 bered as Exhibit Nine? 17 Yes, sir, it is. 18 In developing your engineering plans what 0 19 sources of information did you rely upon? 20 Well, first my task was to find a spot or 21 two for disposal so we had to search the area and we came up 22 with an exempted section and I did use material, books that 23 have been previously written on the subject, such as the 24 Groundwater Contamination Report in a book by the Environ-25 mental Protection Agency on brine disposal treatment prac-

46 1 tices relating to the oil production industry. 2 Did you actually get out on the ground of 0 3 the proposed site before preparing your plans? 4 Yes, I did. A 5 Did you have an opportunity to visit Ó 6 other salt water solids disposal sites? 7 **A** ⁶ Yes, I did. 8 Please describe the site that you have 0 9 selected. 10 The site is located in Section 16. It's λ 11 the southeast guarter of the northeast guarter. It's 600 12 foot by 600 foot starting on the southwest side. The very 13 southwest corner has a pad where trucks can enter and exit 14 freely. There are tank batteries for use to separate and 15 five water disposal pits, four solids pits, and an overflow 16 or emergency pit. 17 Now, the topography of the site that you 0 18 selected is rather unusual. Please describe the topography. 19 Well, it's downward sloping hill caused 20 by a sink, a sink in the area. 21 And to what feature does it slope to-0 22 wards? 23 Laguna Plata. A 24 And what is Laguna Plata. Q 25 It's a large salt water lake. A

47 1 And why did you design the pits in such a Q 2 manner that each successor pit was on a lower level? 3 So that movement between pits was much Α 4 easier, easily accomplished. 5 What was your primary consideration in 0 6 designing this facility? 7 The evaporation rate of the fluids in La-8 guna Plata. 9 You spoke of the evaporation rate in La-0 10 guna Plata. Could you please tell us what that rate is? 11 A Mr. Dan Stephens, our hydrologist, knows 12 more about that. I did read his report, however, and it was 13 4.4 feet per year. 14 What types and quantity waste materials 0 15 did you design this site for? 16 Would you repeat that? Ά 17 What types and quantity of waste mater-0 18 ials --19 Okay, well, we ---A 20 -- did you design the proposed site for? 0 21 The quantities, the types and quantities **A** 100 - 50 - 50 22 were production water. There is oil associated with produc-23 tion water, and also drilling fluids and cement. 24 The -- we set a limit of 30,000 barrels 25 per day. It's only 20 percent or less of what the evapora-

48 1 tion of the lake will handle. 2 The guantities of each were 26,500 bar-3 rels would be water. 2500 barrels would be oil, and 1100 4 barrels would be the actual solids. 5 And how much of this 30,000 barrel capa-Q 6 city do you anticipate actually using? 7. Only 2250 barrels per day is our expected A 8 initial rate. 9 What is the actual maximum capacity, Q as-10 suming that you have steady stream of traffic up and down 11 State Road 176? 12 That would be 30,000, approximately A 13 30,000 barrels per day, that's nothing else could possibly 14 get into those pits, the number of unloadin lines. 15 Mr. Thornton, could you please step up to 16 the map, and beginning with the unloading line, describe the 17 flow of solids and liquid wastes? 18 Okay. When trucks enter this pad they 19 come up to either -- there are two sets of two unloading 20 lines. One is for solids, meaning drilling fluid and ce-21 ment, and the other is for production water. 22 The truck comes in here, hooks up to one 23 of the two unloading lines which is connected to a 700-bar-24 rel gunbarrel. There are two of these. 25 What is the purpose and function of the

49 1 gunbarrel tank? 2 It is to separate the oil from the water Α 3 before entering the tank. 4 - Q How does that accomplish that? 5 A By gravity. The gunbarrels have a water 6 leg attached to each one. 7 And to where do the usuable hydrocarbons 8 go? 9 Into the center tank, marked P-2, which A 10 holds 1000 barrels. 11 What will be done with the oil that comes 0 12 from that tank? 13 The oil will be taken from one of our A 14 trucks to our reclamation plant, the Goodwin Treating Plant, 15 and we will re-treat this oil and recover some pipeline oil 16 from that. 17 Now how is the tank battery connected to Õ 18 the disposal pits? 19 The two gunbarrels are connected to the 20 disposal pits through a water leg. 21 What's the purpose of setting up the five 0 22 disposal pits that you have on succeeding lower levels? 23 **A** . The purpose is to, number one, each pit 24 is lower than the other, meaning that fluid can be 25 transferred and controlled much easier.

50 1 The pipe, the conduit right here, which 2 will be 12, at least 12 inches in diameter, are staggered 3 such that the maximum amount of retention time or settling 4 time can be accomplished. 5 Q That's why the conduits are offset? 6 Yes, sir. A 7 Now have you calculated the 8 designed the size and calculated the capacity of each of the 9 salt water disposal tanks? 10 I'll go ahead and -- the pit W-1, Ά Yes. 11 or water disposal pit, the first one holds 7480 barrels; 12 that is at a level 3-foot from the top of the pit. It will 13 actually hold 10,686. 14 The second pit will hold 7480. 15 The third pit will hold 6411. 16 Number four will hold 5343 barrels. 17 Fifth pit, 4274 barrels. 18 And total capacity of these water pits 19 will be 30,988 barrels with -- go ahead. 20 0 you indicated that you Now, -- Mr. 21 Thornton, I understand that you have established a 3-foot 22 leeway between the top of the pit and the maximum water 23 level. 24 Why did you do so? 25 Α To protect against any spills that might

51 1 occur from either extremely several 35-year rain in six 2 hours or any other operational problem, such as maybe sedi-3 ment that's cemented solid that are settling out into the 4 pit from clogging the conduit. 5 Now, your salt water disposal pits are 0 6 established or connected, rather, to an overflow pit. What 7 is the purpose and function of that pit? 8 A Well, to further protect the -- any over-9 flowing of the pit due to the same problems mentioned be-10 fore. 11 Another over-design? 0 12 A Right. 13 What is the size of the overflow pit? 0 14 A It's 100 foot by 60 foot by 3 foot deep. 15 And what is its capacity in barrels? 0 16 It is 3206 barrels. A 17 How many solids pits have you included in 0 18 your engineering plans? 19 Four solids pits. 20 0 Can you please tell me the purpose and 21 function of the solids pits? 22 Α The solids pits are designed to handle 23 any mud or cement that the trucks may haul in. The truck 24 merely hooks up the line, either line, and disposes into a 25 pit, into these pits on both sides because of build-up, un-

1 equal build-up if we just had one going, say. 2 Q What is the size of each of the proposed 3 solids pits? A It's 100 foot by 24 foot or 25 foot. 5 They are designed that size so that it is easily accessible 6 to backhoes or dozers and such to clean them out. 7 What is the capacity of each in barrels? 8 A Solid disposal pit number one is 3117 9 barrels, and again I've got a maximum. That's 3 foot from 10 the top of the dike that will be built around each one of 11 these pits. It's 4,452, actually. 12 Solid pit two is 3,117 barrels; solids 13 pit three is 2,671; and solids pit four is 2,226. These 14 give a total capacity of 11,131 barrels. 15 0 Would you please explain the conduit sys-16 tem that you have designed which links the solids pits? 17 A As I said before, the -- each unloading 18 line is connected to one side of the pit so as to equalize 19 the solids over the whole pit, because solids are viscous 20 and tend to build up on one side, but we went ahead and got 21 6-inch conduit running through the pits on each side with 22 valves on each one of the pits -- I mean one of the inlet 23 lines to the pits. 24 The way it will operate is three of these 25 valves will be closed; we'll use one pit at a time. When

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53 that is built up, we'll close that pit off and use the other 1 one, and we'll continue down here until you get down here to 2 the fourth pit and we need the first pit by the time we get 3 to this one. This one should have dried out and we will be Δ able to clean it out and then re-use it again. 5 0 Are the solids pits also connected to the 6 overflow pit? 7 Yes, sir, they're connected. The over-A 8 flow pit has a series of 6-inch lines connected to each pit 9 so the level of all pits will never exceed the height of the 10 dike. 11 Is this another intentional --12 Yes, this is another over-design again. 13 What other improvements to this 600 foot 14 by 600 foot area have you included in your design? 15 Two roads coming into the site. The area 16 will be fenced with a 4-strand barbed wire fence around the 17 whole 600-foot area and the oading pad. 18 What impact, if any, will all your over-19 20 design features have upon any discharge into Laguna Plata? It will keep any solids from entering 21 into Laguna Plata. 22 These pits will contain everything 23 they're designed to contain. It will operate so that it is over-designed for safety. 24 25 Why have you over-designed it?

54 1 Α To provide safety for groundwater conta-2 mination. We don't want to --3 I have no further MR. WEBER: 4 questions. 5 MR. STOGNER: Kellahin, Mr. 6 your witness. 7 MR. KELLAHIN: Thank you. 8 9 CROSS EXAMINATION 10 BY MR. THORNTON: 11 0 Mr. Thornton, you said earlier in your 12 direct examination that you had had an opportunity to visit 13 other sites in coming up with a design for this facility? 14 Yes, I did. Α 15 0 What sites have you visited in making 16 your study to determine what type of design for this site? 17 A Pollution Control, Laguna Gatuna. 18 You have been at that facility? Q 19 Yes, sir. A 20 When did you examine that site, Q Mr. 21 Thornton? 22 A There were a couple times. There was one 23 just recently with Mr. Stephens. 24 0 How long have you worked on this particu-25 lar design prospect for Agua, Inc.?

55 Approximately three months. A 2 Can you describe for us in what ways, 0 if 3 at all, your proposed facility here differs from the one at 4 Pollution Control? 5 Yes. The -- number one, we are using the 6 topography of the area. 7 Number two, our pits, each of our pits 8 are -- will not go, say, above the level of the dike. It 9 will never be equal to the level of the dike. 10 This three-foot freeboard that you're 11 talking about ---12 Right. Right. Ά 13 -- is the difference. When you 0 said 14 awhile ago that the evaporation rate given to you by the hy-15 drologist was some number and the volumes used in this dis-16 posal facility were approximately 20 percent of the evapora-17 tion rate --18 Right. A 19 0 -- in what context are you saying that? 20 Have you calculated or has someone calculated the evapora-21 tion rate for the surface of the pits? 22 No, of the lake itself. A 23 0 Oh, I see, okay. When trucks come into 24 the facility, do you propose that waste products run through 25 your system will be in trucks that are equipped to either

56 1 pump or discharge fluids through the tanks? 2 Right. Α 3 Do you receive or do you propose to re-4 ceive materials that will come in dump trucks? 5 This is -- this is A No, we will not. 6 strictly oilfield waste. We do not -- we do not want any 7 other solids at our disposal site. 8 Drill cuttings, I understand, sometimes 0 9 come in dump trucks, that kind of thing, are you designed to 10 handle that kind of disposal? 11 I wasn't aware of that and the design can 12 be -- I mean it can be altered but I don't -- I don't see 13 any need to because there is -- I had never heard of such a 14 thing. 15 All right. When -- when we have the ca-Q 16 pacity of the four solid pits and I think you gave us a num-17 ber of a little over 11,000 barrels, is that the --18 1100 barrels. Oh, I'm sorry. Yes, sir, A 19 you're right. 20 The capacity of the four solids pits? 21 A Right. 22 11,000 barrels? Have you estimated how 0 23 it will take you to fill up those pits before you have long 24 to clean them out? 25 That's depending -- that's largely depen-A

57 1 ding on how much is disposed in there and that is not a de-2 finite quantity. 3 Once a solid pit becomes full or has 0 to 4 be cleaned out, what is your plan for the disposal or the 5 storage of those solids? 6 They will just be placed on the pit -- on A 7 the pad. 8 You would take the solids after all 11-9 quids have evaporated out of these solids pits, those solids 10 that are left remaining after evaporation? 11 A Yes, the clays from the drilling mud and 12 the cement itself. 13 You'd clean the pits and then take that 14 solid material and place it on the pad. 15 Right. A 16 All right. Are any of the pits lined? 0 17 No, they are not. A 18 You've described for us a few ways in 19 which your facility was different than Pollution Control. 20 Are you aware of any other material ways that your facility 21 is different than the one at Pollution Control? 22 Yes. We do not separate our oil in the A 23 pits. We separate it in the tank. 24 In that first ank there? Q 25 Right, the first two.

58 1 Q All right. Are there any other differ-2 ences between your design and the facility at Pollution Con-3 trol? 4 A They discharge into the -- into Laguna --5 Gatuna. 0 6 A -- Gatuna directly. We do not. Ours is 7 an indirect method. 8 Your proposal then would be that the Q 9 water in the pits, it's your intent to have that water re-10 main confined to the pit. 11 It will seep towards Laguna Plata. A 12 But in terms of a direct discharge into Q 13 Laguna Plata, you haven't designed that nor do you propose 14 to do that? 15 No, we do not. We don't see a need for A 16 it. 17 All right. 0 18 KELLAHIN: I wonder if we MR. 19 might take a few minute break? 20 STOGNER: Now would be a MR. 21 good time to take about a ten to fifteen minute break. 22 23 (Thereupon a recess was taken.) 24 25 MR. STOGNER: The hearing will

59 1 come to order. 2 Kellahin, I believe you Mr. 3 were ready to cross examine? Thornton, just a few more questions 4 Mr. 0 5 about the operations at the proposed site. 6 Uh-huh. A 7 Do you propose to fence in the facility Q 8 in any way to keep livestock and --9 A Yes, sir. 10 Off the property? 0 Yes, sir. These are steel posts I've got 11 up here and between them we'll have four strands of barbed 12 13 wire. 14 The entrance is -- both entrances, or the 15 entrance and the exit will be -- will have a cattleguard so 16 no cattle or livestock and enter into that area. 17 How will you handle the day to day opera-Ö 18 tions in terms of manning the facility? Will this be staf-19 fed 24-hours a day or you going to open and close it during 20 particular hours? What is the proposed plan? 21 Any time the site is open there will a 22 person on the site looking over the disposal area. 23 You propose to have it manned, then, 0 SO 24 that when the compound is open there will a person in charge 25 to direct the proper utilization of the facility by

60 1 truckers? 2 Yes, we will. A 3 Have you done any kind of analysis of the 0 4 chemicals or substances that will be removed from the solid 5 waste pits and placed on the pad? Have you made any studies 6 of those or analyses of those types of materials? 7 A No, sir, I have not, but it is -- they're 8 probably the same composition as what is going into 9 Pollution Control solids pits. 10 Do you propose to put these on the pad in 11 such a way that they will remain confined either with some 12 kind of liner underneath or that they'll be covered so that 13 the salts and whatever else they are may not blow away or 14 dissipate into the adjoining properties? 15 No, we hadn't planned on putting anything 16 there. We did not anticipate it blowing away. 17 All right. Thank you. 18 MR. KELLAHIN: I have nothing 19 further. 20 MR. STOGNER: Thank you. Mr. 21 Weber, any redirect? 22 MR. WEBER: Yes, sir, just a 23 few questions. 24 25

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1	REDIRECT EXAMINATION
2	BY MR. WEBER:
3	Q You mentioned in discussing the
4	difference between your proposed design and that presently
5	employed by Pollution Control, certain differences.
6	Among those differences was the use of
7	topography. What did you mean by using topography?
8	A The area is down has a downward slope
9	towards Laguna Plata. Any any water disposal or solid
10	disposal, whatever, it will run downhill to the into the
n	pit.
12	Ω Could you show us the flow using your
13	model, which has been marked as Exhibit Nine?
14	A The flow will run downward, and the water
15	pits, as I said, they were staggered, and it will be
16	filtered through the the underlying sand layer that runs
17	toward Laguna Plata.
18	Q What is the benefit of that filtering
19	process?
20	A It cleans up all the any suspended
21	particles that might be in the production water.
22	Q Are you indicating, then, that the
23	eventual discharge into Laguna Plata will be cleaner than
24	had you put a direct line?
25	A Yes, it will be much cleaner.

62 1 Q You talked about conduits. Do the con-2 duits in your proposed design differ at all from the con-3 duits presently in use by Pollution Control? 4 Yes. They're larger. Α 5 Q What benefit, if any, does that provide? 6 Any problems with lines plugging due to A 7 any build-up of suspended solids, as I mentioned to you be-8 fore. 9 Is the possibility of plugging of conduit 10 a real possibility? 11 No, it's a remote possibility. A 12 Q Have you developed any plans to deal with 13 the remote possibility should it occur? 14 The, number one, the design does Ά Yes. 15 incorporate the overflow pit and if we do have a build-up of 16 solids in the bottoms of the pits, we can divert the water 17 such that we can clean each pit separately. 18 Now with regard to cleaning of pits, 19 there was some question with regard to the placement of 20 solid materials on the pad that you will have constructed. 21 If any difficulty develops are your plans 22 flexible enough to provide contingencies to handle that? 23 I don't follow. 24 Is this a preliminary plan that you're 25 presenting or is it a final ---

63 1 A Yes, this is a --2 -- plan? Q 3 This is a preliminary plan. A If we do 4 need more pits we will dig more pits to insure that we can 5 handle what we are getting. 6 And if you do need to employ other Q 7 methods with regards to solids disposal, you can accomplish 8 that as well? 9 Yes. Ά 10 MR. WEBER: I have no further 11 questions. 12 MR. STOGNER: Mr. Kellahin? 13 MR. KELLAHIN: No, sir. 14 15 CROSS EXAMINATION 16 BY MR. STOGNER: 17 Q Mr. Thornton, just a few basic questions 18 here. 19 Let's go back to Exhibit Number Eight and 20 your solids -- I'm sorry, the line into your solids pits you 21 said was 6-inch conduit, is that right? 22 Yes, sir. I'm sorry, that was 8-inch. 23 When you come in there an empty cement 24 through these 8-inch lines, what would -- what will Petro-25 Thermo use to wash those lines out with?

64 Α Gravity. 2 Q How much slope does these 8-inch lines 3 have? 4 Well, up on the pad the -- if you'll give Å 5 me just a minute here I can give you height. 6 Q Well, I guess what I'm getting at, you 7 don't think that cement is going to set up in those lines? 8 No, if it does, we can put another line. A 9 Okay. Q ' 10 A There, and there will be somebody there 11 at all times to make sure that nothing happens. 12 Okay. 0 Let's go over to your water pits. 13 The conduit, the 12-inch diameter pipe between the two 14 that's staggered, where are those actually set within the 15 pit? 16 Three feet below the level, A below the 17 dike. 18 Okay. When W-1, or Pit No. 1, when it 0 19 fills up with solids, you propose to take those solids out 20 and then spread it around the pad, is that correct? 21 Α. Yes, sir, that was S-1. 22 Q I'm talking about W-1. 23 W-1 is a water disposal pit. It has pro-24 duction water only going to it. The whole west -- the west 25 pits there are water disposal pits. The east ones are so-

65 1 lids disposal pits, solids disposal pits with an overflow 2 pit. 3 So you don't anticipate W-l filling up 0 4 with silt. 5 A No. The line that is connected between 6 those two pits is located a food below the top of the dike. 7 The water level will never reach that high to interconnect 8 into each pit, and if it does, the water would flow down to 9 overflow pit, the overflow pit. 10 MR. STOGNER: I have no 11 further questions of Mr. Thornton. 12 Are there any other questions 13 of this witness? 14 MR. WEBER: I have none. 15 MR. STOGNER: If not, he may be 16 excused at this time but we may bring him back to answer 17 some more questions. 18 Thank you. 19 Sir, we'd like to WEBER: MR. 20 call as our next witness, Mr. Dan Stephens. 21 22 DANIEL BRUCE STEPHENS, 23 being called as a witness and being duly sworn upon his 24 oath, testified as follows, to-wit: 25

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2	DIRECT EXAMINATION
3	BY MR. WEBER:
4	Q Mr. Stephens, will you please state your
5	full name?
6	A My name is Daniel Bruce Stephens.
7	Q And where do you reside, Mr. Stephens?
8	A In Socorro, New Mexico.
9	Q And are you the principal of Daniel B.
10	Stephens and Associates, Consultants in Groundwater Hydrolo-
11 -	dà.
12	A That's correct.
13	Q How long have you had this consulting?
14	A I've been doing consulting in New Mexico
15	for about six years.
16	Q Mr. Stephens, from what institution did
17	you receive your undergraduate degree?
18	A I went to Penn State University.
19	Q And what degree did you receive and when
Z 0	did you receive it?
21	A Bachelor of Science and a degree in geo-
22	logical science in 1971.
Z3	Q Were you singled out for any honors?
24	A I graduated with honors and I was given
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67 1 an award as the outstanding senior in the College of Earth 2 and Mineral Science. 3 From what institution did you recieve 0 4 your graduate degree? 5 Α A Masters degree in hydrology at Stanford 6 in 1974 and a PhD in hydrology at the University of Arizona 7 in 1979. 8 At those institutions were you singled 0 9 out for any particular honors? 10 A No. 11 Are you a member of any professional or-0 12 ganizations? 13 A Yes. I'm a member of the American Geo-14 physical Union, the American Association of Groundwater 15 Scientists, Soil Science Society of America, Sigma Xi. 16 Ó What is Sigma Xi? . . . 17 It's a scientific honorary society. А 18 Have you been published in any scientific 19 or technical journals? 20 Yes. I've published in the Water Resour-Α 21 ces Research, the American Society of Civil Engineers, Gen-22 eral Hydraulics, Groundwater Journal, and General Hydrology, 23 a number of things. 24 Have you delivered any papers at any 25 scientific or technical meetings?

68 1 Yes, commonly to several per year. Α 2 Q So then you estimate you delivered some-3 thing in excess of a dozen papers? 4 Α At least. 5 Q Have you been employed as a consultant to 6 any state agencies? 7 I'm a consultant to the State A of 8 Colorado's Department of Health and the State of New Mexico 9 Environmental Improvement Division. 10 If you were to say you had a single 0 11 specialty, what would that specialty be? 12 Primarily in problems of seepage through 13 materials towards -- as it moves towards the water table. 14 Q What practical experience have you had 15 regard to investigating problems of seepage with in the 16 State of New Mexico? 17 Well, we had, in addition to some consul-A 18 ting to the State Environmental -- State of New Mexico's En-19 vironmental Improvement Division on uranium mill tailings, 20 we've done a field investigation and laboratory study of 21 seepage from an impoundment on the Ogallala area in the Clo-22 vis vicinity. 23 Have you ever had the opportunity to pre-0 24 pare a complete hydrogeologic -- hydrologic study? 25 Yes.

69 1 MR. WEBER: Mr. Examiner, I 2 would at this point offer Mr. Stephens as an expert hydrolo-3 gist. 4 MR. STOGNER: Any objections? 5 MR. KELLAHIN: Dr. Stephens is 6 so qualified. 7 0 Stephens, have you had the opportun-Dr. 8 ity to study the hydrology of those tracts of land proposed 9 by Petro-Thermo Corporaiton of use as a waste disposal site? 10 Yes. A : 11 Have you had an opportunity to review Q 12 those engineering plans prepared by Mr. Jim Thornton for 13 Petro-Thermo? 14 A Yes. 15 Have you had an opportunity to actually Q 16 visit the site of the proposed waste disposal facility? 17 Yes, I did. A 18 0 Have you prepared a report with regard to 19 your findings? 20 Yes. A 21 Q I show you now what has been marked as 22 Exhibit Number Ten in Case Number 8781 and ask you if you 23 can recognize that? 24 A Yes, that's the report prepared by my-25 self.

Q Please describe in very general terms the
 hydrologic conditions in the vicinity of the proposed
 disposal site.

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4 The water-bearing units that are of Α interest are usually above the Permian section, which is at 5 a depth of 800 or so feet, and within the interval, the 6 first 800 feet, there are notable occurrences of water 7 in the redbeds formation, which comprises maybe 750 or more 8 feet of that -- that interval, particularly in sandstone 9 layers within the Chinle, which is the upper member, 10 and also in the Santa Rosa sandstone in the lower portion of the 11 Triassic Redbeds. 12

There's also an alluvial cover that may be variable thickness, perhaps at the site 20, 10 to 20 feet, maybe, and in places around the site there seems to be groundwater which occurs in the alluvium but is very discontinuous because of the irregular nature of the redbeds and the low rates of natural infiltration.

19 At the site there doesn't seem to be any 20 significant amount of water in the alluvial section. 21 Primarily water that occurs at the site is expected to be in 22 the Triassic redbeds.

23 Q Does the site lead towards Laguna Plata
24 and if you could, please explain what Laguna Plata is.

Laguna Plata is a body of water which is

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71 1 located in the deepest portion of a collapse feature that 2 created a number of other lakes in the area and it's regional sink for groundwater discharge at shallow depths 3 4 and at greater depths in the Triassic section. 5 So at the site groundwater flows towards 6 the -- towards Laguna Plata. On all sides of Laguna Plata 7 there's convergent groundwater. 8 Now you indicated that Laguna Plata is a Q 9 We have in the very near vicinity Laguna Garegional sink. 10 tuna, Laguna Toston. What can you tell me about the rela-11 tive elevations of these in relation to Laguna Plata? 12 Laguna Plata is the lowest water surface 13 of those several bodies you mentioned. It serves 35 14 regional collection point for groundwater where most of the 15 discharge from the system occurs. 16 Is both Laguna Plat and the proposed dis-17 posal site within the collapse feature? 18 Yes, that's correct. 19 Where are the Triassic redbeds in rela-0 20 tion to Laguna Plata and the proposed waste disposal plant? 21 The redbeds occur, the interface between A. 22 the redbeds and the alluvial material at the site is, oh, 23 perhaps 20 to 30 feet below land surface. 24 In Laguna Plata the drilling that was 25 done in the lake encountered the redbeds at, maybe, 30 feet below the surface of the redbeds, so there's in a sense an
offset in the redbeds, possibly due to some deformation by
faulting or actual folding into the collapse structure.

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Q Did you have an opportunity to walk up and down arroyos and determine the general depth of the redbeds underneath the alluvial sands and what, if anything unusual did you discover?

A In walking through some of the arroyos it
9 appears that the section that's visible there from land sur10 face is a thin veneer of dune sand. It might be a foot to
11 several foot thick.

12 Then there's a light red to tan sandstone 13 with some green sandstone layers in it that appears to me to 14 be a sandy member of the Chinle and that might be maybe 20 15 feet thick, and then there's a very noticeable contrast of 16 the shale, a dark brown redbed shale that is very prominent 17 and one of the features that struck me when I was there was 18 that there seemed to be a line of seeps, a diffuse zone of 19 discharge of groundwater that occurred just above this red-20 dish horizon, which was very -- in a clay stone you'd call 21 it a shale, and appeared to me as though it marked the -- it 22 was an impeding horizon naturally, under natural conditions 23 that caused discharge to leave at this contact, and we 24 traced it for maybe 100 yards laterally to the north from 25 the site.

73 1 In practical terms what does this mean? Q 2 That any infiltration that falls from A 3 the permeable sand and surfacial deposits percolates down, 4 perhaps it goes to 20, 30 feet, depending on particular lo-5 cations at the site, and then moves laterally down dip, or 6 down this interface to the north towards Laguna Plata and 7 discharges. 8 The practical significance that struck me 9 is that if this is occuring under natural conditions, chan-10 ces are that this is a very good barrier to downward perco-11 lation due to seepage from these pits. 12 Aren't Triassic redbeds generally consid-13 ered to be virtually impermeable? 14 In the vertical direction that's a very 15 often assumed condition. 16 Ö. Do the Triassic redbeds act as a barrier 17 to any seepage into sand stringers which may be found below 18 them? 19 That's my opinion, yes. A 20 What is the direction of the major flow 0 21 of surface and subsurface water from the disposal site? 22 To the north. 23 To the north in the general direction of n 24 Laguna Plata. 25 To the north towards Laguna Plata. A

74 Q 👘 Tell me a little bit about Laguna Plata. 1 What is it? 2 A It's a salt lake; a point of groundwater 3 discharge; there are springs surrounding it, probably upward 4 5 moving water. At the very shallow depths it comes into the adjacent areas and there's a high concentration of salt 6 7 rocks occuring there at the present time. What do you mean by a high concentration Q 8 9 of salt? Could you give us a numer which would indicate how concentrated the salt is or ---10 11 Ά A chemical analysis recently done that was reported to me gives the total dissolved solids concen-12 13 tration of 335,000 milligrams, or parts per million chlor-14 ides. That would be about 192,000 parts per million. 15 Is that total dissolved solids number Q 16 significant? 17 It's much more concentrated, than A 💈 sea 18 water by an order of magnitude. 19 About ten times as concentrated as Q sea 20 water? 21 A Yes. 22 Is there any leakage of water from Laguna 0 23 Plata into adjoining formations? 24 Not that I have any evidence for. Ά There 25 is a vertical component of the hydraulic gradient that one

1 could infer because of the pressures in underlying forma-2 tions are less than the potential level of the lake, but 3 there hasn't been demonstrated any significant amount of 4 leakage based on water chemistry data.

⁵ Q Based upon your study and inspection,
⁶ have you been able to formulate an opinion as to any reason⁷ ably foreseeable beneficial use of Laguna Plata?

8 A From my own -- my own point of view, I
9 don't see any change in the current pattern of use of Laguna
10 Plata.

Have you had an opportunity to measure Have you had an opportunity to measure the surface area of Laguna Plata and to calculate the evaporation which would occur on it?

14 The surface area of Laguna Plata A 18 15 approxiantely two square miles. Based on studies that were 16 done for the Bureau of Land Management in the potash dis-17 trict, they found that the evaporation rate from brine lakes 18 is approximately 4.4 feet per year; over that two square 19 miles gives an annual evaporation rate of about 5630 acre 20 feet per year.

21 Petro-Thermo's You proposed have 22 engineering plat. Have you been able to take their 23 discharge of some, the maximum discharge of some 30,000 bar-24 rels a day of liquid and reduce that to acre feet so as to 25 compare it with the evaporation rate?

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1 A It would be about 1500 acre feet per year 2 at the maximum.

3 In terms based upon the assumption that 0 all the materials, waste in the proposed disposal facility would have flowed directly into Laguna Plata, would they be evaporated?

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Essentially the material that would seep A 8 in has a potential to evaporate. The normal operating con-9 dition, though, is much, much less than the 30,000 barrels a 10 I believe the number is 2250 barrels a day, which if day. 11 we look at the waste that would come from the salt water 12 ponds, if all of that seeped into the lake, that, and assum-13 ing none evaporated, all of that went into Laguna Plata, it 14 would be about 93 acre feet per year, or maybe less than 2 15 percent of the total inflow to Laguna Plata would be -- that 16 would be the increase that would flow to Laguna Plata as a 17 result of seepage from this operation, assuming no evapora-18 tion took place in the pits.

19 What would be the practical effect of 20 this discharge of brine water into Laguna Plata?

21 I think there would be no measurable con-22 sequence. From the practical standpoint, I don't think 23 there would be consequence. From a mass balance standpoint 24 there has to be a small increase in the stage, maybe on the 25 order of tenths of feet and of course as the stage rose and

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77 1 the surface area expanded slightly, we would be looking at 2 even more rate of evaporation on a larger water surface. 3 Based upon your study and inspection, 0 4 have you been able to form an opinion regarding the effects 5 on any discharge -- or on Laguna Plata from any discharge? 6 That was basically negligible? 7 That's correct. 8 Based upon your study and inspection have Q 9 you been able to make any determinations as to the presence 10 of fresh water at shallow depths in the vicinity of the pro-11 posed disposal? 12 The nearest well we were able to located 13 is up, up gradient about 2-1/2 miles near Halfway. Its 14 quality is marginal for drinking. The little bar at Halfway 15 is abandoned. 16 There's a windmill to the east approxi-17 mately three miles distant from the site, which is used for 18 stock watering, or had been used for stock watering; appar-19 ently abandoned now, and it does have water in thew alluvium 20 that seems to be good quality. 21 0 Now you mentioned that these locations 22 were up gradient from the proposed disposal site and I sup-23 pose up gradient from Laguna Plata. What difference does 24 that make? 25 seepage from Laguna any Plata

78 1 would be likely to move down gradient and there's no way 2 possible for that seepage to contaminate wells that far up 3 gradient. 4 Based upon your study and your inspection Ő 5 have you formed an opinion as to whether the discharge water 6 and solids could move to subsurface in such a manner as to 7 commingle in the reasonably foreseeable future with an un-8 contaminated water supply? 9 I don't foresee that as a probability. 10 MR. WEBER: I have no further 11 questions. 12 MR. STOGNER: Mr. Kellahin, 13 your witness. 14 MR. KELLAHIN: Thank you, Mr. 15 Stogner. 16 17 CROSS EXAMINATION 18 BY MR. KELLAHIN: 19 Stephenson, I'd like you to help me Dr. 0 20 understand the relationship in this site specific area be-21 tween the potential for evaporation versus the infiltration 22 rate of the liquids into the ground. 23 Let's assume some fact situations and 24 then you tell me what will be the effect on this particular 25 operation.

The first assumption I'd like you to make is let's assume that the infiltration rate of the fluids has been totally impaired. Let's assume that may have occurred with solids becoming deposited on the bottom of the ponds so that infiltration is minimal, if at all.

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6 Should that occur what is the capacity of 7 the proposed plan to by means of evaporation handle a cer-8 tain volume of disposed liquids? My question is, without 9 infiltration and using your evaporation rates from Laguna 10 Plata, can you estimate for me what would be the capacity of 11 the facility if they to rely solely on evaporation?

12 A Approximately, approximately three, three
13 to four acre feet per year.

14 Q I need some help. How many barrels of
15 oil are we talking about in relation to an acre foot?
16 A 7580, well, it's about 20 -17 MR. ABBOTT: 7758.

A It's about 20 --

18

19

25

Q

MR. ABBOTT: 7758.

20 MR. KELLAHIN: Somebody give me 21 a number that everybody likes.

A I think there's about -- I don't know, about 22 or so thousand barrels per day in an acre foot per year.

All right. Have you taken into consider-

80 1 ation the effect the evaporation rate -- well, let me start 2 over. 3 Can you tell me or have you studied the 4 effect that the oil skim or the oil slick on the surface of 5 the ponds will have in terms of its effect on the evapora-6 tion rate? 7 It's my understanding that one of Α the 8 features of the pond is to recover any floating hydrocarbons 9 that has been bypassed in the gunbarrels and that will be 10 skimmed off. 11 If there were a cover of -- a veneer of 12 oil slick on top, it would depress the rate of evaporation 13 but as far as this site's concerned in order for it to oper-14 ate it's got to be viewed essentially as an infiltration 15 gallery rather than an evaporation system. Evaporation in 16 my calculation is a negligible amount. This primarily 17 should be viewed as an infiltration system which uses the 18 soil as a filter, allows the retention in the settling bases 19 and recovery of any floating product. 20 Assuming the operator on a regular basis Q 21 attempts to skim the oil but we still have a small viscosity 22 of oil on the surface, and assuming lack of infiltration, 23 what effect does that have on your evaporation calculation 24 of Laguna Plata using the salt brine evaporation calcula-25 tion?

1 81 2 Α The 4.4 feet per year? 3 Yes, sir, what effect will that oil skim 0 4 on the evaporation have? Is there a way to estimate for us 5 the impact? 6 Α I can't do it here, I think one could 7 look in the literature for some available studies. I be-8 lieve there are some but I just don't have that one with me. 9 All right, now let's talk about the sys-10 tem the way you anticipate it to work, not as evaporation 11 ponds --12 Α Right. 13 -- but as a mechanism whereby we have 0 14 high rates of infiltration and we have the fluids then 15 migrating down gradient into the Laguna Plata. 16 Have you calculated or estimated the 17 rates of infiltration of the produced water in the absence 18 of having the infiltration impaired by substances collecting 19 on the bottoms of the ponds. Let's disregard that kind of 20 problem for the moment. 21 Okay. Let me rephrase your question so I 22 can 23 0 Okay. 24 -- understand it. Are you asking me what А 25 infiltration rate would be if there were no impeding the

82 1 layer? 2 Yes, sir, the impeding layer being some-0 3 thing not physically there now. It would be oil or mud or 4 some --5 We haven't done any field tests. That 6 would be required in order to calculate the permeability of 7 the subsoil. 8 It appears to me, based on preliminary 9 calculations and inspection of the site that the soil has a 10 potential permeability to allow that much water to infil-11 I think the point about build-up of solids in the trate. 12 bottom of the pond is an important one and really will re-13 quire, I think, as part of the plan a regular maintenance of 14 the pit to allow infiltration to occur. 15 Let me interrupt you a moment, Doctor. Q. . . 16 You talked about a volume. What volume are we talking 17 about? Is this still the 30,000 barrels a day? 18 No. No, this would be the 2250 barrels a 19 day --20 All right. Q 21 -- calculation. The other -- the other 22 could be handled -- if the hydraulic conductivity of the 23 formation underneath it is about 10 to the minus 3 centi-24 meters a second, 1000 feet per year, I believe, it could 25 handle it. I don't think it's quite that much but I -- we

1 haven't done any field tests to determine it.

Yes.

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On the other hand if it's 10 to the minus A centimeters a second, you know, with the 2250 barrel a day rate, I think that's about what it is, without a clogging layer.

Q If we're using the 2250 barrels a day as the anticipated use for the facility, you've indicated that maintenance of the ponds in terms of scrapping the pond bottoms in order to maximize the ability of the fluids to infiltrate might be a prudent thing to do.

12 Q Can you give us -- can you give us some 13 estimation as a hydrologist of how frequent that maintenance 14 operation ought to be? How are we going to judge when we 15 ought to start cleaning the bottoms of the ponds?

16 That's a question I haven't thought about 17 but it is an ongoing problem when people are doing artifi-18 cial recharge along rivers, say, in Orange County, Califor-19 nia, where they do this thing to get water into aquifers for 20 re-use. It's an ongoing operation. I think one may want to 21 -- a period of a couple of weeks, perhaps, maybe on that or-22 der, have one pit out of commission for drying and disking 23 or removing some of the sludge or sediments, suspended 24 materials, bypassing it, bypassing that particular pit and 25 putting another one in use.

84 1 That's a guess. I've not scoured the 2 literature for the experience that's available concerning 3 this. MR. KELLAHIN: May I have a mo-5 ment, Mr. Stogner? 6 I'm through with Mr. Stephen-7 son, thank you. 8 MR. STOGNER: Thank you, Mr. 9 Kellahin. 10 Mr. Weber, anything --11 MR. WEBER: Yes, sir, I have a 12 few --13 MR. STOGNER: -- further? 14 WEBER: .-- further ques-MR. 15 tions for Dr. Stephens. 16 17 REDIRECT EXAMINATION 18 BY MR. WEBER: 19 Dr. Stephens, you were asked just a few 20 moments ago to assume that the infiltration rate of fluids 21 on the bottom of a disposal pit was totally impaired, was 22 zero. What would the practical effect of that be? 23 Well, a couple of things would have to A 24 happen. Bither the pits would fill up; in order to prevent 25 any spillage over the sides you'd cut back operations. That would be easy to do when the pits were not full and taking on any more water. They could be operationally shut down until they could be improved in terms of their infiltration characteristics.

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⁵ Q So if this assumption actually occurred,
⁶ a total -- the infiltration rate of the fluids would be to⁷ tally impaired, that would be something entirely obvious to
⁸ the operators, I guess you could say.

A I would think so. If it spilled over the
sides the whole system would be eroded and there wouldn't be
any -- wouldn't be any other recovery of hydrocarbon on the
water surface, so its not to their advantage to have the
water flowing over land into Laguna Plata.

14 Q And if the infiltration rate was such 15 that there would be no flow, how difficult a job woud it be 16 to improved the infiltration rate?

17 Well, it's a common -- it's a common 18 problem in artificial recharge studies but it's -- it's one 19 that -- there's plenty of artificial recharge experience 20 that can be relied upon to reclaim the permeability of the 21 formation. It's an engineering problem that can be overcome 22 with either disking and drying, actual removal of the some 23 of the materials with, perhaps, a backhoe to get down to the 24 basic formation.

MR. WEBER: I have no further

	86
1	questions at this time.
2	MR. STOGNER: Mr. Kellahin?
3	MR. KELLAHIN: No, thank you.
4	
5	CROSS EXAMINATION
6	BY MR. STOGNER:
7	Q Dr. Stephens, I'd like to refer ot Figure
8	3 of your Exhibit Number Ten, which is your report, and this
9	Figure 3 is Water Level Evaluations and Depths to Water, and
10	in the center of your plat you show Laguna Plata and over to
11	the east you have several circles and sqwiggley lines, what
12	you show as being springs.
13	What is that water makeup and where is
14	the water coming from that feeds those springs?
15	A The makeup of the water?
16	Q Yes, sir.
17	A Do you mean its chemical composition?
18	Q Chemical composition, salinity.
19	A Table 3, Page 8, tabulates the chloride
20	concentrations at those springs. They're in the most of
21	them are on the order of 8000 parts per million chloride.
22	Q Okay, what feeds these springs? Where is
23	this water coming from?
24	A It appears to me that there's a couple of
25	possible sources. One, it might be a mixture of different

1 We have dune material, limitless dune material in sources. 2 the north side of these lakes which is very good infiltra-3 I think there's some water which tion characteristics. 4 comes into Laguna Plata from the north and there's also some 5 that comes from the east, possibly in drainage out of Laguna 6 Gatuna, in that vicinity, and mixing and so there's possibly 7 two, two sources, one would be areal infiltration through 8 the surfacial deposits which are permeable, and also lateral 9 flow from adjacent areas, which could be in the vicinity of 10 Laguna Gatuna nearest to these (not audible because of 11 coughing.) 12 This is contained in your report and 0 13 needless to say, I haven't had a chance to look at it yet. 14 Let's talk about the redbeds for a little 15 bit between Laguna Plata and back toward the proposed dispo-16 sal site. 17 Over the disposal site itself how far 18 down before you hit redbeds? 19 My estimate is about 25 to 30 feet. 20 That's for the clay, what you would call the redbed, that 21 marcon clay layer that I described which was a barrier to 22 seepage. That appears to be about 25 to 30 feet down.

23 Q And the thickness of the redbed in this 24 general area?

25

Approximately 800, 750, 800 feet. Total

88 1 redbeds, including Dewey Lake, Santa Rosa, Chinle. 2 · O Okay, you mentioned that there is some 3 sandstone deposits within the redbed, is that right? 4 Within the Triassic generaly, that's cor-5 rect, and the Santa Rosa, of course, is a sandstone. 6 Q Where is the Santa Rosa in respect to the 7 redbed in here? 8 The depth is probably several hundred Å 9 feet below the site. 10 Okay, within the redbed itself is there 0 11 any deposits of sandstone that are water bearing? 12 Yes. 13 There is. Are those deposits inter-chan-14 neled with other sandstone deposits or are they layered, 15 separated? 16 They're layered. It's my feeling that 17 it's the latter; not too much interfingering with Santa 18 Rosa. 19 Q . Do you know what the depth would be from 20 the top of the Redbed to your first sandstone layer that has 21 water in it? 22 I really haven't had a study of that A 23 stratigraphy. There's a lot of drilling reports, which 24 you'll find contained in the report and it's very difficult 25 to use these types of drillers logs to correlate one, what

89 1 might be called the water sand and one layer to another. Ι 2 don't think there's that sufficient geologic control based 3 on drillers logs to make a stratigraphic horizon, but you'll see that it does occur in the first couple hundred feet, 400 5 feet, there are occurrences of water in the -- in these 6 units. 7 Regionally, however, the quality of water 8 in the Triassic is poor. 9 What do you mean poor? 0 10 High in chlorides; above drinking water 11 standards. 12 So I can understand it, let's talk about 13 Laguna Plata itself. 14 Is the -- is the redbeds present within 15 the bottom of the lake or -- I guess they aren't. 16 Is there any percolation downward from 17 Laguna Plat at, say, a maximum water level, at a flood 18 stage, say? 19 It would be very local, as it were, some 20 going into the bank at the stage the Laguna Plata rose dur-21 ing a (not understood), there would be some lateral movement 22 into the adjacent soil, but when the level fell it would 23 move back in, the bank storage type in effect. 24 There's, in the subsurface contours on 25 the deeper zones within the redbeds which show convergence

	90
1	of flow, even at great depth, that's been described in the
2	report by Nicholson and Clebsch, 1961 on the hydrogeology in
3	the area, and their map shows that the (not understood) met-
4	ric surface slopes towards Laguna Plata and if there were
5	any downward leakage, it would presumably be still contained
6	within this zone of convergent flow. But if there were a
7	lot of downward discharge under current conditions, the sal-
8	inity of those zones that were receiving this discharge
9	would be very, very high, Laguna Plata being a saline lake.
10	Q Again, just so I can understand, the
11	water that accumulates in Laguna Plata is all evaporated;
12	none of it migrates out, except, like you were saying, at a
13	high flood stage it could go back and theoretically those
14	waters would then
15	A Come back in or themselves evaporate from
16	the soil directly.
17	Q You referred to a Nichols and Clebsch?
18	A Nicholson and Clebsch.
19	Q Nicholson and Clebsch, I'm sorry.
20	A C-L-E-B-S-C-H.
21	Q Are those in your references?
22	A Yes, sir.
23	Q You stated that Laguna Plata was the low-
Z4	est of the lakes within this area. What is the elevation of
25	the lake bed?

91 1 I recall looking at a benchmark out in Α 2 the lake during the field survey. I believe it was 3431 3 feet. 4 And how much lower is that than Laguna 0 5 Gatuna to the east? 6 Probably 60 feet. A 7 And Laguna Tonto up in the farther east Q 8 and north? 9 I'd have to look at the map. I don't Α 10 know offhand. Looks like on the order of 100 feet. 11 Are you referring to a regular USGS map 12 13 That's correct. The Laguna Gatuna 7-1/2 A 14 Minute Quadrangle. 15 Okay, thank you. Okay, how about Laguna Q Second 16 Toston, which is down south? 17 Tuston or Toston? 18 It appears to be about 60 feet, also. A 19 MR. STOGNER: I have no further 20 questions of this witness. 21 Are there any questions of the 22 witness? 23 All right, Mr. Weber. 24 MR. WEBER: If I could ask just 25 one further question.

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2	REDIRECT EXAMINATION
3	BY MR. WEBER:
4	Do the error Writersteinstein automatic
5	the calt lakes which wenter have the calt
6	che salt lakes which you've described, Laguna Gatuna, Laguna
	Plata, and Laguna Tonto, Laguna Toston?
- 7	A That's correct.
8	Q Thank you.
9	MR. WEBER: Nothing further.
10	MR. STOGNER: Are there any
11	other questions of Dr. Stephens?
12	If not, he may be excused.
13	MR. WEBER: Mr. Examiner, T
. 14	would like at this time to move the admission of Exhibits
15	One through Ten.
16	
17	Chiections?
18	OD Jec CTOUR 1
10	MR. KELLAHIN: I don't recall
17	which letter was marked what exhibit number. Our only
20	objection went to that exhibit. We have no objection to the
21	others.
22	MR. STOGNER: I believe the
23	Exhibits were Five, Six, and Seven, the letters from Amoco,
24	Amerada, and another
25	MR. KELLAHIN: Natural

	93
١	Resources?
2	MR. STOGNER: Are those the
3	letters that you refer to that
4	MR. KELLAHIN: Yes, sir.
5	MR. STOGNER: Mr. Kellahin, I'm
6	going to overrule your objection and allow those into evi-
7	dence.
8	MR. WRBRR: In this regard I
9	would request that the copy of the letter from Bravo Energy
10	Incorporated, which was received by the Oil Conservation
11	Division on 12-9. 1985 also be included in this file
12	MR. STOCNED: Okay the record
13	will so show we did receive that being we the och re-
14	ceived that on December 9th. 1985, and it was made part of
15	the case file at that time.
16	Nr Woher da wou have anothing
17	further?
18	
19	ther
20	
21	MR. STOGNER: Mr. Kellahin?
22	MR. KELLAHIN: Thank you, Mr.
23	Examiner. I propose not to put on any direct evidence on
24	benair or my client at this point and we are prepared to
25	make a prief closing statement. We would like an oppor-
	tunity to submit a proposed order in this case but I have no

.

94 1 witnesses to call at this time. 2 MR. Thank you, STOGNER: Mr. 3 Kellahin. 4 I believe if there is nothing 5 else, then we are ready for closing statements. 6 MR. KELLAHIN: We seem to have 7 a number of people here today. There perhaps is somebody 8 here that wants to make a statement other than me and Mr. 9 Weber. 10 Are there other parties here? 11 MR. STOGNER: Thank you for 12 correcting me. Since there is a large contingency here, 13 would anybody like to stand and make any kind of a state-14 ment? We'll start from this end of the room and work 15 around. 16 Please stand and identify your-17 self. 18 MR. BILL TOM: Bill Tom from 19 Andrews, Texas. We are the present lessee on the grazing 20 lease concerned and we have not relinquished the grazing 21 lease at the present time. We are against the proposal be-22 cause of our ranching interest at this time. 23 MR. STOGNER: Okay, what was 24 your name again, please? 25 MR. TOM: Bill Tom.

95 1 MR. STOGNER: Tom, Mr. the 2 grazing lease that you're referring to is a grazing lease 3 from the State? 4 MR. TOM: This is correct. 5 MR. TAYLOR: Is our understand 6 ing correct that your permission is needed to change the 7 grazing lease to a business type lease? 8 MR. TOM: This is correct as I 9 understand it. 10 MR. STOGNER: Per the rules and 11 regulations of the State Land Office. 12 Any further statements? 13 There being none by any par-14 ties, we're ready for closing statements. 15 Mr. Kellahin, you may go first. 16 Mr. Weber, you may go last. 17 MR. **KELLAHIN:** Mr. Stogner, 18 I'll be quite brief. 19 We would, first of all, request 20 that you grant us a week or ten days in which to submit to 21 you a proposed order that we would believe appropriate for 22 entry in this case. 23 The reason I suggest that is 24 Dr. Stephens' report, his report that I have not read. I 25 like to review that and try to understand it before would

we attempt to suggest to you how the Division might want to enter an order for this particular site. My client, as the testimony, has shown, operates Laguna Gatuna and Pollution Control,

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Inc.. There have been some comparisons made between this proposed site and the Pollution Control site. I'd like to examine what we've heard today in terms of what we do on our facility and see if I can't suggest an order to you that protects our interest.

Mr. Abbott has indicated, and I Mr. Abbott has indicated, and I believe Mr. Weber told you in his beginning comments, that there had been an order approving the use of Laguna Plata for salt water disposal. He referenced Order No. R-3725 and that is an order that was issued back in '69 to Mr. Larry Squires, who is the principal of Pollution Control.

Because of the close proximity 17 of this project to our approval of the use of Laguna Plata 18 for salt water disposal it's important for us to recommend 19 to you an order that minimizes the impact that this opera-20 tion may have on our potential use of Laguna Plata.

As you can see from this Order
3725, this order only allows the use by Larry Squires of Laguna Plata for salt water disposal and does not allow him to
use it for oilfield waste and solids.

This order has been subse-

97 1 quently amended in small part by providing for the use of Laguna Plata -- Laguna Gatuna in this order as the site 2 for 3 the use of the solid waste. We want to try to suggest to 5 you a proposed order that accommodates Mr. Abbott as best we 6 7 project may have on our interest in the area. 8 In addition, Mr. Squires is the 9 Manager of Snyder Ranches, which is the surface owner not of 10 this particular site but of the adjoining property. 11 We would appreciate the cour-12 tesy that you could extend to us to give us seven or ten 13 days to give you an order so you could deal in specific sub-14 stances about the nuts and bolts of the order itself. 15 In closing, the only point I 16 see that gives me some concern in the presentation Mr. Weber 17 has made, is that there may be fundamental jurisdictional 18 defect in the application at this point. The Commission 19 rules, as you know, require under 1203 that the initiation 20 of a hearing can be done by the Division, the attorney gen-21 eral, any operator or producer, or anyone having a property 22 interest may institute proceedings for a hearing. 23 I think it's apparent -- it's 24 implicit upon your exercising of authority in this case to 25 make a decision about the jurisdiction. The testimony Was

1 that Mr. Abbott has applied for a business lease from the 2 State of New Mexico to utilize the surface. You have to re-3 solve whether or not the filing of an application would vest 4 Mr. Abbott's company with a sufficient property interest by 5 which he could be an applicant today.

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I would like an opportunity to search some of the other cases that we've put on here before the Commission before I give you what my opinion of the law is, but I want to raise that as an issue because my recollection is in the past the Commission has required that the applicant obtain a business lease from the State Land Com-12 missioner before proceeding with his application.

Now, I say that with some qual-14 ification because I think it's incumbent upon myself and Mr. 15 Weber to determine if in fact that does constitute a fatal 16 flaw in the application. It might be possible to process 17 the application here, contingent upon approval by the Land 18 Commissioner, so I'm suggesting there may be other solutions 19 but I raise that issue as one that is still before you in 20 this case. It is that Mr. Abbott's companies don't own the 21 surface. We know from Mr. Tom, Toms, that this ranch owns 22 the grazing, and it may cause you the same concern it causes 23 me that the party here lacks sufficient standing before the 24 Commission to bring the application.

> Having said all those things,

	99
1	now, we would like to have an opportunity to submit to you a
2	draft order.
3	MR. STOGNER: Thank you, Mr.
4	Kellahin.
5	Mr. Weber?
6	MR. WEBER: Yes, sir. Petro-
7.	Thermo Corporation would also request the opportunity to
8	present to you a proposed order within a ten day period and
9	we would press for a speedy disposition of this particular
10	matter in view of the fact that its temporary permit to dis-
11	pose of solids near Eunice, New Mexico, will expire on or
12	about the 19th of January of next year.
13	A quick decision might well
14	minimize any adverse impact from the closure of one of the
15	three available solids disposal sites in southeastern New
16	Mexico.
17	I would like to now turn to
18	that interesting jurisdictional question raised by Mr. Kel-
19	lahín.
20	This is really not a new or a
21	novel argument. Mr. Kellahin raised it on the 23rd of Sep-
22	tember, 1981, in the matter of Loco Hills Water Disposal
Z 3	Company. Loco Hills also wanted an exception to Order Num-
24	ber R-3221.
25	Loco Hills was also in the same

posture as Petro-Thermo Corporation. It had made but had
not received final approval from the State Land Office with
regard to its business lease. That was Case Number 7329.
Mr. R. L. Stamets was the Examiner.

5 There, as here, the application 6 had been made but had not reached its final approval. We 7 would argue that here, as there, the jurisdictional objec-8 tion should be dismissed. It's my understanding in that 9 particular case Mr. Stamets that there was a sufficient pro-10 perty interest even though final approval had not been re-11 ceived.

Understanding full well the position of the present grazing lessee, it is our contention that it is not necessary to obtain a relinquishment of his grazing lease; that in accordance with the rules and regulations of the State Land Office it would be possible to grant that lease in the absence of a relinquishment.

18 Examiner, we have tried to Mr. 19 show through our presentation today that a legitimate need 20 exists in southeastern New Mexico for additional approved 21 disposal sites. There are a limited number of disposal 22 sites now and it's quite possible that an emergency could 23 really have a significant impact not only on trucking opera-24 tions but on oil and gas production within that section.

25

Mindful of the need to devoid

1 -- to avoid any possible of discharge into a reasonably 2 foreseeable fresh water source, we have been very careful to 3 select the site, a site which limits any possible -- possi-4 bility of contamination of adjacent fresh water sources. We 5 have selected a site which the Oil Conservation Commission 6 has exempted from the operation of its Order 3221, exempted 7 in the sense that it permitted the disposal of production 8 water in unlined surface pits. We have selected an area by the 10 Laguna Plata which -- into which the disposal of production 11 water would not constitute a hazard to fresh water supplies 12 in the area. 13 We feel that we have met the 14 burden demonstrating the absence of the possibility of con-15 tamination, met the burden of showing that an exception to 16 Order No. R-3221 should be granted to Petro-Thermo. 17 That's it. 18 MR. STOGNER: Thank you, Mr. 19 Weber. 20 Is there anything further in 21 Case Number 8781? 22 There being none -- before I 23 take this case under advisement, I've been doing some figur-24 If I allow seven days or a week it would be on Christing. 25 mas, and if I allow ten days it would be on the 28th, which



103 1 2 CERTIFICATE 3 4 I, SALLY BOYD, C.S.R.. DO HEREBY CERTIFY that the foregoing Transcript of Hearing before the 5 Oil Conservation Division (Commission) was reported by 6 me; that the said transcript is a full, true, and correct record 7 of the hearing, prepared by me to the best of my ability. 8 9 10 11-12 Javeng le 13 <u>25</u>2 14 15 16 I do hereby certify that the foregoing is 17 a complete record of the proceedings in the Examiner hearing of Case No. 878/ 18 heard by me on 18 Merember 19 85 19 allus Examiner Oil Conservation Division 20 21 22 23 24 25

STATE OF NEW MEXICO

ENERGY AND MINERALS DEPARTMENT DIL CONSERVATION DIVISION



50 YEARS

GOVERNOR

January 7, 1986

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87501 (505) 827-5800

Mr. Robert W. Abbott Petro-Thermo Corporation P. O. Box 2069 Hobbs, New Mexico 88241-2069

> Re: Request for Extension of Time for Temporary Use of Pit

Dear Mr. Abbott:

We have received your request dated December 23, 1985 for an extension of your 90-day permit for temporary disposal of brine, fresh water-based drilling mud, and waste cement in the emergency overflow pit located at the Blinebry-Drinkard SWD System Well No. A-22 in 22-T22S-R37E. Your current temporary permit expires January 19, 1986.

Due to time restrictions associated with Case No. 8781 which was brought to hearing on December 18, 1985, you are hereby granted a 60-day extension for the specified temporary use of the pit. The extension carries the following restrictions:

- 1) The pit will not be enlarged or allowed to overflow or breach.
- 2) By March 20, 1986, solid and liquid waste will be removed from the pit and disposed of in an approved manner.
- 3) There will be no further extensions of time for the specified temporary use of the unlined pit. Tanks or a lined pit may be authorized for continued disposal of brine, fresh water-based drilling mud, or waste cement at that location.

Authorization for this temporary use of the pit may be rescinded if there is evidence that the pit has overflowed or if Water Quality Control Commission or Oil Conservation Division rules or regulations have been violated.

Sincerely

R. L. STAMETS Director

RLS/JB/dp

cc: Jerry Sexton David Boyer Michael Stogner

PETRO-THERMO CORPORATION

P

P.O. BOX 2069 PHONES (505) 393-2417 — 397-3557 HOBBS, NEW MEXICO 88241-2069

December 23, 1985

Mr. R. L. Stamets Oil Conservation Division Energy and Minerals Department P. O. Box 2088 Santa Fe, New Mexico 87504-2088

Dear Mr. Stamets:

On October 21, 1985, a 90-day permit for temporary use of the emergency overflow pit located at the Blinebry-Drinkard SWD System Well No. A-22 was granted to Petro-Thermo Corporation.

As you are aware, Petro-Thermo Corporation has applied for authorization to dispose of produced water, associated waste hydrocarbons and other solids at a site in the E/2 NE/4 of Section 16, Township 20 South, Range 32 East. A hearing to consider this application was held on December 18, 1985 before Examiner Michael E. Stogner.

At the conclusion of the hearing, Petro-Thermo Corporation respectfully requested that an Order be issued prior to the expiration of its temporary permit.

Nevertheless, our attorneys have advised us that there is a possibility of a hearing \underline{de} <u>novo</u>. Given this circumstance, it is possible that a final decision will not be reached prior to January 19, 1986.

In view of these developments, Petro-Thermo Corporation again requests that it be authorized temporary use of the permit for one full year. Please contact our office if you require further information.

Yours Sincerely,

Petro Thermo Corporation

IN M. al By: Robert W. Abbott

Vice-President

WGA/sfp

HYDROGEOLOGIC CONDITIONS NEAR LAGUNA PLATA, NEW MEXICO, RELEVANT TO THE APPLICATION TO THE OIL CONSERVATION DIVISION TO DISPOSE OILFIELD WASTE BY PETRO-THERMO CORPORATION

Prepared for:

Petro-Thermo Corporation P.O. Box 1978 Hobbs, New Mexico

By:

.

Daniel B. Stephens and Associates 600 Neel Avenue Socorro, New Mexico

December 1985

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APPENDIX

Appendix 1. Well Logs

SUMMARY

Petro-Thermo Corporation is proposing to discharge oil field wastes at a site adjacent to Laguna Plata in southwestern Lea County. On the basis of available hydrologic data, an exemption to Oil Conservation Commission Order No. 3221 is requested.

At the site, ground water occurs at shallow depths in redbeds and possibly in alluvium. The direction of flow is northward toward Laguna Plata, a salt lake located within a collapse structure. Springs indicate that ground water discharges to Laguna Plata. The thickness of the very low-permeable redbeds beneath the site is about 750 feet.

After separation in a gunbarrel, brine and oilfield fluids will be diverted to unlined pits where additional free oil will be skimmed for recovery. Much of the waste water will seep into the subsurface and migrate toward Laguna Plata. The concentration of dissolved solids in the waste water is expected to be less than that of the native water in Laguna Plata. The average evaporation of Laguna Plata is more than 60 times the estimated average sustained rate of fluid waste disposal. Thus, seepage from the disposal operation will evaporate from Laguna Plata.

Daniel B. Stephens and Associates

PETRO-THERMO CORP. Waste Disposal at Laguna Plata

INTRODUCTION

This report was prepared at the request of Mr. Robert W. Abbott, Vice President of AGUA Inc., a division of Petro-Thermo Corporation. Petro-Thermo Corporation is applying to the NM Oil Conservation Division for a permit to discharge ground water from proposed oilfield waste disposal ponds to be constructed near Laguna Plata, New Mexico.

The purpose of this report is to evaluate present hydrogeologic conditions in this vicinity of the proposed waste disposal site. The scope of work of this report includes a brief review of available literature, survey of existing well records, compilation of existing chemical analyses of water samples, and a field reconnaissance of the site.

SITE DESCRIPTION

The proposed oilfield waste ponds comprise approximately 4 acres located in the SW 1/4 of the SE 1/4 of the NE 1/4 of Section 16, Township 20 South, Range 32 East, Lea County, New Mexico. This site is approximately 2.5 miles northwest of Halfway, New Mexico, which is about 37 miles west of Hobbs, NM on US Highway 180 (Figure 1). The site is about 0.15 miles south of Laguna Plata, a natural salt lake.

The land surface topography at the site slopes to the northeast with a gradient of approximately 230 feet per mile, toward Laguna Plata. Vegetation at the site is very sparse, consisting mostly of grasses and mesquite.

Mean annual precipitation in the area is about 9 inches per year, much of which falls in the summer months during intense thunderstorms. Average annual temperature for the nearby towns of Maljamar and Pearl is approximately 61 degrees Fahrenheit.

The average rate of evaporation from open bodies of fresh water is about 6.1 feet per year (Hunter, 1985); these rates are enhanced during the spring when the winds are strongest. A study in the potash mining district to the southwest of the site found that the evaporation rate from a brine pond ranged from about 10.9 feet per year in the summer to about 2.9 feet per year in the winter (Geohydrology Associates, Inc, 1979). In a previous study in the same area, the average evaporation rate from a brine lake was determined to be about 4.4 feet per year (Geohydrology Associates, Inc., 1978). The evaporation rate from brine is less than that for fresh water, owing to the lower vapor pressure of the brine. The annual rate of evaporation from Laguna Plata is

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PETRO-THERMO CORP. Waste Disposal at Laguna Plata



Figure 1 - Location Map

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PETRO-THERMO CORP. Waste Disposal at Laguna Plata

approximately 5630 acre-feet per year (3490 gallons per minute), based on a lake surface area of 2 square miles (1280 acres) shown on topographic maps and the 4.4 feet per year estimate of evaporation rate.

The area is very sparsely populated. The dwellings which comprise Halfway, New Mexico are abandoned. Except for Halfway, the only dwelling within two miles of the proposed discharge site is a ranch on the east side of Laguna Plata.

HYDROGEOLOGIC CONDITIONS

Geology

The site lies within the Permian basin, a subsurface structural feature, which has been a target of oil and gas exploration. The rocks within the basin include Precambrian to Recent age strata. The units which have hydrologic significance are of Triassic age and younger, inasmuch as no potable water is known to occur in older rocks anywhere in the basin.

The Paleozoic section which overlies the Precambrian basement is reported to be as much as 16,800 feet thick on the west side of Lea Co. (Nicholson and Clebsch, 1961). The geologic units in this section include mostly limestone and dolomite, however evaporite deposits of Permian age, such as salt and anhydrite, occur in the upper parts of the section. The youngest Paleozoic unit beneath the site is the Rustler formation, chiefly anhydrite with salt and "redbeds". Drill logs in T20S.R32E.Sec 16. (Appendix 1) indicate the depth to the top of the Permian section is approximately 800 feet below land surface.

The Dewey Lake red-beds, a Triassic or Permian age siltstone, shale and sandstone overlies the Rustler formation. Its thickness may range from 40 to 400 feet (Nicholson and Clebsch, 1961).

The Dockum group, which overlies the Dewey Lake formation, includes the Santa Rosa sandstone in the lower part of the section and the Chinle shale in the upper part. These two units comprise the "Triassic redbeds". The Santa Rosa sandstone is reported to include some shale, and the unit ranges in thickness from 140 to 300 feet (Nicholson and Clebsch, 1961). The Chinle formation consists of claystone and fine-grained sandstone. Gypsum is reported to be a common secondary mineral associated with the redbeds. At the site, the thickness of the Santa Rosa and Chinle is difficult to determine from drillers logs, owing to the interbedded nature of the shale and sandstone which occur within each formation. However, the available logs (Appendix 1)

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show that the combined thickness of redbeds is about 750 feet, consisting mostly of shale and sandstone.

The regional dip of the Triassic redbeds is approximately one to two degrees to the east or southeast. The Santa Rosa formation outcrops south of the Laguna Plata. A shale, assumed to be the Chinle, outcrops just north of the Laguna Plata, and the redbed surface is exposed in arroyos at the southwest end of the Laguna Plata at an estimated elevation of 3460 feet, about 30 feet above the lake surface. Drill holes in Laguna Plata indicate that the redbed surface is 20 to 41 feet below the surface of Laguna Plata (Nicholson and Clebsch, 1961). The surface of the redbeds is depressed in the vicinity of Laguna Plata, as well as near Laguna Gatuna, Laguna Tonto, and Laguna Toston. This structural feature is attributed to localized dissolution of the underlying Permian halite and anhydrite, with subsequent collapse of the overlying redbeds into the depression. Thus, the dip of the redbeds is locally reversed near the collapse depressions. The dip of the redbeds is northeast at the site.

In many parts of the region the Ogallala formation overlies the Triassic units. However, in the vicinity of the site this formation has been removed by erosion. Quaternary alluvium was deposited in the topographic depressions where the Ogallala was removed (Nicholson and Clebsch, 1961). The alluvium consists of poorly-sorted, unconsolidated sand with some clay. In places caliche occurs within the alluvium; the escarpment at the south edge of the site is underlain by a thick caliche layer. The thickness of the alluvium ranges from 15 to 130 feet in the northeast quarter of T2OS.R32E.Sec.16 (Table 1), based on drillers logs. Exposures in arroyos just north of the site suggest that the alluvial cover may be less than 10 feet thick beneath the site. There is also a thin veneer of dune sand and small stabilized dunes at the site.

Principal Water-Bearing Units

Potable ground water is reported to occur in Triassic and younger units in parts of southern Lea County. However, there are only scattered occurrences of potable ground water in areas surrounding the proposed site of waste disposal (Tables 2 and 3). Regionally, the Santa Rosa sandstone is the principal waterbearing unit. Ground water may also occur in sandstone layers within the Chinle. Well yields are typically very low, owing to the low permeability of the formation. Nicholson and Clebsch (1961) report that well 20.32.18.233 which tapped the Santa Rosa had a specific capacity of 0.2 gallons per minute per foot of drawdown. Some of the well logs in Appendix 1 show the occur-

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Well Number	Thickness of Alluvium (ft)	Thickness of Redbeds (ft)
20.32.6.22	40	826
20.32.8.44	12	828
20.32.12.44	25	1020
20.32.16.33	40	835
.16.144	15	860
.16.124	44	808
.16.411	30	835
.16.243	130	- 700
.16.213	130	710
.16.341	50	813
.16.233	50	815
.16.134	20	850
.16.31	35	840
.16.32	45	828
.16.244	15	765
20.32.18.32	35	760

TABLE 1. Thickness of Geologic Units Determined from Well Logs

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Location			Depth	Altitude	WT Depth	Date of
No.	Owner	Aquifer	of Well	of Well	Below LS	Measurement
19.32.34.42322	Halfway Water Co.	TRS	575	3559.0	247.38	12/14/76
19.33.17.11224		QAL	131	3650.0	116.84	2/05/84
19.33.26.42221	Mark Smith	QAL	100	3608.0	92.97	1/29/81
10.32.01.314114	W. N. Snyder	QAL	30	3452.0	89.2	3/24/54
20.32.17.13	*****	QAL	90	3449.0	9.0	2/28/79
20.32.18.233	Freeport Sulfur	TRS	400	3452.0	89.2	3/24/54
20.32.22.33		TRC	160	3513.0	30.0	2/28/79
20.32.23.33132		QAL		3541.0	39.83	2/19/81
.23.43312	B. Stanford	TRC	78	3551.0	36.78	2/19/81
.24.3333	T. Bingham	QAL	65	3555.0	37.69	2/19/81
.27.14332	J. Frey	QAL	25	3539.0	23.32	2/19/81
.27.32322	T. Bingham	QAL		3530.0	15.33	2/19/81
20.32.30.142	*****	QAL		3505.0	9.94	6/11/54
20.32.31.13		TRC	240	3550.0	135.12	3/15/79
.36.21442	B. Smith	QAL	50	3581.0	43.88	9/18/72
.36.22311	B. Smith	QAL	65	3586.0	45.82	2/19/81
20.33.04.43211		QAL	58	3556.0	33.19	3/19/68
.05.34321	Pan Amer. Petr Co	TRS	680	3552.0	277.52	2/19/81
.15.22143		TRS		3582.0	335.10	4/20/55
.18.12322		TRS		3521.0	245.58	7/25/72
.20.22224		OAL.	52	3536.0	35.0	2/19/81
21.32.6.11	I. A. Allred	QAL	55	3597.0	46.21	3/10/81

TABLE 2. Well Inventory Data

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			Chloride		
		Geologic	Concentration*	Date	
Well 1	Number	Formation	(ppm)	Sampled	Use
19.32	.08.22411	TRS	16	3/13/85	Stock
19.33	.18.133223	TRS	312	2/15/83	OWD
	.26.42221	QAL	326	1/16/78	Stock
	.26.42221	QAL	306	10/08/76	Stock
20.31	.13.412433	QAL	635	12/22/48	Stock
	.13.414	Williams Lake	110,750	2/27/84	Lake
	.13.414411	. QAL	6,660	2/27/84	
	.16.234441	TRS	785	12/22/48	Stock
	.16.24331	TRC	673	11/30/65	Stock
	.16.421111	TRC	355	11/30/65	Stock
20.32	•	Laguna Plata	196,012	12/13/85	Lake
20.32	.17.13	QAL	172,828	12/22/78	
	.22.33	TRC	5,136	12/19/78	
	.23.43312	TRC	362	2/69	Comm-
					ercial
	.24.333	QAL	85	2/69	Wind-
					mill
	.24.333	QAL	42	9/11/72	Stock
	.36.21442	QAL	290	9/18/72	Stock
20.33	•	Laguna Gatuna	158,000	2/69	
20.33	.04.43211	QAL	12,978	10/24/68	Stock
	.21.22224	QAL	3,518		
20.31	.01.13143	QAL	57	8/18/72	Domes- tic
Sprin	g #1	SE end of	8,864	2/12/69	
-	-	Laguna Plata			
Sprin	g #2	E end of	7,446	2/12/69	
-	-	Laguna Plata			
Sprin	g #3	E end of	7,446	2/12/69	
		Laguna Plata			
Sprin	g #4	E end of	7,978	2/18/69	
		Laguna Plata			
Sprin	g #5	S end of	163,105	2/18/69	
		Laguna Gatuna			
Sampl	e #6	Gatuna, in	72,333	2/18/69	
		draw North of			
		Highway			
Sampl	e #7	Gatuna, NW end	27,657	2/18/69	
		at oil well,			
		NW OF Well in	•		
		ravine			

TABLE 3. Chloride Concentrations (Source: N.M. State Engineer's Office, Roswell, NM, and Geohydrology Associates, Inc., 1979, 1984)

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8

TABLE 3 - continued

Well Number	Geologic Formation	Chloride Concentration* (ppm)	Date Sampled	Use
Sample #8	Gatuna, NW ex at oil well b of well in ra	nd 10,992 NE avine	2/18/69	
			•	

* Recommended drinking water standard is 250 ppm.

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rence of ground water within sandstones of the redbeds. Where it is saturated, the alluvium also may yield water to However, the areal extent of the saturated portions is wells. limited as a result of the irregular nature of the redbed surface. Thus ground water in the alluvium near the site is not sufficient in volume to comprise a laterally extensive aquifer which has potential for development, except locally for domestic and stock watering uses. At the north end of the site and along the southwest side of Laguna Plata, there are a few seeps which occur at the contact between a dense red shale within the Chinle and an overlying sandstone member. This shale horizon apparently serves as a barrier to water which infiltrates the sandy surfi-There is no evidence of an alluvial aquifer cial deposits. beneath the site, based on field reconnaissance. Any significant water-bearing unit beneath the site is expected to occur in the Triassic redbeds.

The depth to the water table is about 37 feet near Halfway and about 22 feet at the ranch one mile east of Laguna Plata. The depth to water decreases toward Laguna Plata. Topographic maps show that there are numerous springs on the east side of Laguna Plata which mark the intersection of the water table with the land surface. These springs also mark the locations of points of groundwater discharge to the Laguna Plata. This discharge presumably originates, in part, from seepage from Laguna Gatuna, which is about 60 feet higher in elevation. There are few available data on the chemical quality of ground water (Tables 3 and 4). No wells are known to produce potable ground water within approximately three miles of the site. A well in the alluvium (20.32.1.322) at the ranch northeast of Laguna Plata produces water which is not potable. East of Halfway, an alluvial well (20.32.18.32) yields potable water having chloride concentrations of 42 ppm (parts per million) (Table 2). This well is reported to be used to water stock. In the Triassic redbeds the chemical quality of ground water in wells is also variable, ranging from 21 to 785 ppm (Table 4 and Figure 2). Well 20.32.23.433, completed in the Chinle at Halfway, has a chloride concentration of 200 ppm.

Groundwater Movement

Based on available water level elevation data, shallow ground water in alluvium and upper redbed formations flows toward Laguna Plata (Figure 3). The springs also suggest that ground water moves toward this topographically low area. In the deeper Triassic units, ground water also moves toward the area containing the salt lakes (Nicholson and Clebsch, 1961). There is a vertical component of hydraulic gradient downward from the

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TABLE 4. Chemical Analyses of Selected Wells Near The Site. Values in parts per million except pH and E (micromhos). (Source: Nicholson and Clebsch, 1961.)

Well #	Date	(ft)	\$10 ₂	Ca	Mg	Na+K	HC03	S04	<u>c1</u>	F	NO 3	TDS	<u>E.C.</u>	рH
<u>Laguna Plata</u>												•		
	12/13/85	0		940	3,317	124,644	71	10,124	192,012			335,108		7.34
Alluvium														
230.32.1.322	7/1/54													*
Triassic Redb	oeds													
19,32.8,224	12/9/58		19	10	. 13	131	306	74	21,	1.2	6.4	426	682 ·	8.0
19.34.9.114	12/9/58	33	41	430	65	675	189	1,680	560	0.3	139	3,680	4,660	7.1
20.32.23.433	12/13/85	78		51.	3 48.6	123	292	54	200			770		7.94
21.33.2.231	9/4/58	1150			+		336	95	20				3,370	



Figure 2 - Chloride Concentrations

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Figure 3 - Water Level Elevations and Depths to Water

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shallow aquifer and Laguna Plata toward the deeper Triassic water-bearing units. If there were downward movement across the low-permeable shales, the quality of groundwater in the waterbearing Triassic sandstones would be poor, owing to the high salinity of Laguna Plata.

The proposed waste disposal site is situated within about 0.15 miles of the south shore of Laguna Plata. Seepage from the impoundments is expected to infiltrate through the underlying dune veneer, alluvium, and shallow sandstone toward the water table. The depth to the water table beneath the site is expected to be approximately 20 to 30 feet. Shallow ground water which may occur at present beneath the site may be perched on the Chinle shale layer observed in arroyos in the field reconnaissance. This layer would cause a ground water mound to develop beneath the waste pits and divert seepage northward and down-dip toward Laguna Plata or to an arroyo draining toward the lake.

The time for seepage to reach the Laguna Plata is difficult to estimate, owing to the absence of aquifer properties and water level data. However, the rate of ground water movement is likely to be on the order of not more than 100 feet per year; thus, the time for seepage to reach Laguna Plata would be on the order of 8 years. The shallow depth to the water table and the relatively permeable nature of the surficial materials will result in relatively rapid transport of seepage through the vadose zone.

There are no water users downgradient from the disposal site. After the seepage reaches the Laguna Plata, practically all of the seepage will evaporate. A minor amount of seepage may move downward toward the lower Triassic water-bearing units. However, regional ground water flow in these units also converges toward the salt lakes. There are no known sources of potable groundwater in sediments underlying the Triassic redbeds at Laguna Plata.

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SUMMARY OF WASTE DISPOSAL PLAN

The disposal plan includes wastes which fall into three general groups. Group I includes brine, salt water, and water contained in drilling mud and cement. Group II wastes include oil and basic sediment (low quality oil which separates from gun barrels). Group III includes solid wastes. The estimated maximum possible volumes of wastes from these three groups are 26,500, 2250, and 1100 bpd (barrels per day), respectively, for a total of 30,000 bpd. However, on a sustained basis, under normal operating conditions, the total rate of waste disposal for all three groups is anticipated to be only about 2250 bpd (106 acre-feet per year) from all three waste types.

The wastes will be separated mechanically in a gunbarrel upon arrival at the site. The liquids from the separation, Group I, will be diverted to a series of five shallow ponds, 60 x 100 feet each. Their depths will range from about 7 to 10 feet. Oil which was not separated in the gunbarrel will be skimmed from the surface of the ponds and pumped to tanks. There is an additional pit downstream of the Group I and II waste pits to contain unexpected overflow.

A significant portion of seepage from the Group I pits will infiltrate the soil and migrate to the Laguna Plata. The salinity of the seepage is not likely to exceed that of Laguna Plata, inasmuch as produced oilfield fluids are expected to have total dissolved solids concentrations in the range of 25,000 to 75,000 ppm. The total dissolved solids at Laguna Plata is 335,100 ppm (Table 3). Thus, the seepage will dilute the concentration of the total dissolved solids in Laguna Plata. The total annual rate of evaporation from Laguna Plata is about 5360 acre-feet per year. Under anticipated normal operating conditions, the total rate of Group I waste disposal will be only about 93 acre-feet per year. Therefore, there is ample storage and evaporation potential in Laguna Plata to accommodate the waste seepage. No significant change in the hydrologic regime is expected as a result of the proposed discharge.

Group II wastes will be stored in tanks and removed from the site for reprocessing and recovery.

Group III solid wastes will be spread into a series of four shallow pits, each about 24 x 100 feet and ranging in depth from 6 to 8 feet. The solids will be alternately dried during the filling of the pits. The dried materials will be excavated and spread on a caliche caprock pad for long-term storage.

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APPENDIX 1 - Well Logs



NEW MEXICO SCHOOL OF MINES STATE BUREAU OF MINES AND MINERAL RESOURCES

SOCORRO, NEW MEXICO

NEW MEXICO WELL LOG DIVISION

Casing Record: 811

1132'

New Elexico Lea County Argo Royalty Co. Burner Sell #1 Sec. 6 T.205 R.32E 750 N. L. - 990 E. L. Elevation: **8513** Commenced: 1-31-35 Completed: 3-12-35

I. P. Abandoned

Formation	Bottom	Formation	Bottom
Red sand	40	- Gray line	2611
Red hods	100	Hard gray line	2639
Red gend	115	Brown line	2653
Red heds	125	Broken gray lime	2664
Red sand & shale	190	Gray sand	2679
Red heds	· 235	Gray sand	2700
R sand	280	Shale breaks	2704
R. Leter sund	290	Hard gray lime	2718
Red sandy shale	335	Lime and shale breaks	2727
Later sand	340	Hard brown sandy lime	2754
Red sendy shale	350	Lhite lime	2785
Red sandy shale	385	Sand	2805
Red rock & red beds	730	TD	2-725
hed rock and red beds	866		•
Anhydrite	890	Rote: Last report T. D.	. 2810
Anhvdvite	945	Lime fishing bit.	
Gr. shale	955		
Anhvdrite	1005		
Salt	1040	-	
Salt and anhy.	1065		
Shale, red	1075		
Anhy.	1090		
Gr. lime	1105		
Anhydrite	1112		
Brown shale	1117		
Reā beds	1128		
Salt and anhy.	1130		
Shale	1175		
Salt	1190		
Shale	1245	·	
Salt	1255		
Skip in Log			
Salt	1625		
Salt	1870		
Salt and anhy.	1880		
Salt	2330		
Anhy.	2510		
Br. Line	2534		

1644

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NEW MEXICO SCHOOL OF MINES

STATE	BUREAU	OF	MINES	AND	MINERAL	RESOURCES	C

SOCORRO, NEW MEXICO

NEW MEXICO WELL LOG DIVISION

CASING REC	ORD	FIEVATION 74 Q4	(1&9)	200
Diam., in. E	lottom	INITIAL DAILY PROD	UCTION:	FELI
10-3/11	1331	Open		bbls. Oil
고 아프 이가 또 요 준 배	1155	Open	•	cu. ft. Gas
04	1100	Tbg.		bbls. Oil
•		Tbg.		cu. ft. Gas

COUNTY Lea
FELD Wildcat
COMPANY Argo Royalty Company
LEASE Burner permit No. 2 Well
LOCATION (%) C SE SE
sec. 8 . t. 205 . r. 32E
660 feet from South line and
660 feet from east line of Section
COMMENCED 6-13-35
COMPLETED 7-23-35
ABANDONED
REMARKS: D&A

FORMATION BOTTO	M, FEET	FORMATION	BOTTOM, FEET
Rock	12	Salt	2060
R _{ed} rock	195	Salt and potesh	2080
Water sand	205	Salt	2130
Red rock. sand. wtr.	230	Salt and notash	2170
Red rock	235		2215
Red sand and shale, 10 BWPH	280	Anhydrite and salt	2245
Red bed	315	Salt	2330
Red bed and red rock	350	Anhydrite and salt	2345
Red rock and sand	390	Anhvārite	2385
Red send and rock	420	Lime	2417
Red bed	515	Lime and red sand	2426
Hard sand	545	Hard lime	2517
R _{ed} rock	590	Hard lime and blue shale	2527
Hard sand	625	Broken lime and sand	2540
R _e d rock	670	Lime and red sand	2549
Sand and red shale	710	Red lime and red sand	2588
Red rock and red shale	750	Lime	2610
Red rock and gyp	795	Broken lime and blue shall	le 2618
Red rock	880	Red and white lime, hard	2624
Anhydrite	915	Hard white lime	2645
Red rock and salt	935	xxxxxXXXXXXXXXXXXXXX	37.
Anhydrite	955	Blue shale and lime	2676
Blue shale	973	Hard white lime and green	1
Anhydrite and lime	995	shale breaks	2685
Anhydrite	1025	Lime	2798
S _a lt	1065	Sand T. D.	2803
Anhydrite and potash	1070		
Potash	1075		
Brown shale	1080		
Anhydrite and lime	1085		
Anhydrite	1118		
Blue shale	1123		
Redrock	1142		
Salt	1155		
Anhydrite	1160		
Blue and red shale	1180	1	
Red rock and salt	1200		
Red rock, salt and annyorite	1205	· · ·	
Palt and red rock	TS30		
Delt and potash	1330		
Salt and red rock	1360	•	
Hara anhydrite	1370	1	
Salt and potash	1550	•	

Log. No. 128

NEW MEXICO SCHOOL OF MINES STATE BUREAU OF MINES AND MINERAL RESOURCES SOCOBRO, NEW MEXICO

CASING 1	RECORD	- ELEVATION		FEET		
Diam., in	Bottom	INITIAL DAILY	:			
81	1117	Open Open Tbg. Tbg.	Ρ.	డ	A.	hbls. Oil cu. ft. Gas bbls. Oil cu. ft. Gas

COUNTY LEA
FIELD Halfway
COMPANY Brewer Drilling Co.
LEASE MONTOE Well No. 1
LOCATION (1/4) SE SE
sec. 12 т. 20 S , в. 32 E
660 feet from South line and
660 feet from East line of Section
COMMENCED 6-8-43
$_{\text{COMPLETED}}$ 7–16–43
ABANDONED
REMARKS:

FORMATION	BOTTOM, FEET	FORMATION	BOTTOM, FEE
T.ime	10	Salt and anhydrite	
Red sand	25		2010
Red bed	250	Annyarite	2620
Red shale	300	Salt and potash	2650
Red rock	3/0	Anhydrite	2690
ed mock	115	Lime	2705
led fock	41)	Pink lime	2720
	425	White lime	2770
	440	Lime	2815
	405	Brown lime	2830
led bed	475	Grav lime	2845
band	487	Grav shale	2865
Red rock	500	Lime	2870
Red bed	515	Red hed	2875
Red sand - water	535	Red shale and lime shalls	2885
Red rock	665	Limo	2000
Shale, red	730		2025
Red rock	1045		2725
Anhydrite	1121	Shale and gypsum	2733
Red rock and shale	1135	Line and red snale	2945
Anhydrite and red rock3	1150	Lime and shale breaks	2960
Shale	1180	Lime	2992
Inhydrite	1210	Sandy lime	3022
59]+	1220	Lime	3055
hudrite	1205	Lime, showing oil and gas	3056
Solt and cholo	1205	Sandy lime	3120
all and shale	1200	Water sand	3126
	1320		-
ray 11me	1340		
led snale	1350	Red beds 10 to 1045 ft	
balt	1380		
nnydrite	1386		•
Salt and shale	1510		
Salt	1715		
nhydrite	1725		
hhydrite and salt	1740		
Salt	1805		-1
Salt and potash	2330		/
Salt and anhydrite	2410		`
Salt and potash	2485		
hvdrite	2520		

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NEW MEXICO SCHOOL OF MINES STATE BUREAU OF MINES AND MINERAL RESOURCES SOCOBRO, NEW MEXICO

C

CASING R	ECORD	ELEVATION INITIAL DAILY PRODUCTION		FEET
Diam., in	Bottom			TON:
	4 " 459 " 940	Open	D&A	bbls, Oil
10 3/4"		Open		cu. ft. Gas
8 5/8"		Thg.		bols. Oil
		Tbg.		cu. ft. Gas

COUNTY	Le	a				
FIELD	Hal	fway				
COMPAN	y Argo	Oil	Cor	D •		
LEASE	Tex	ss-S	tate	пУц	Well No.	2
LOCATIC	N (14)	SW S	W		•	
SEC.	L6	т.	205	, R	· 32E	
660	feet	rom S	outh	line :	and	
660	feet i	rom W	est	line	of Section	
COMMEN	CED	6-1	1-41			
COMPLE	TED	7-1	2-41			
AEANDO	NED	D&A				
REMARI	KS:					

FORMATION	BOTTOM, FEET	FORMATION	BOTTOM, FEET
Cellar	8	Anhydrite	1398
Caliche	40	Slat and Potesh	1520
RedBed and Sand	70	Salt	2105
Sandy shale	80	Anhydrite	2120
Bed Rock	120	Selt	2182
Sandy shale	125	Anhydrite	2205
Red Rock	155	Slat	2270
Red Shale	182	Salt and Potash	2290
Sand. Red	220	Salt	2292
Sandy Shale, Red	250	Anhydrite	2332
Red Rock	295	Lime-medium	2370
Sand	305	Red Rock-Soft	2372
Red Rock	355	Lime-hard-gray	2468
Sandy Shale, Red	385	Lime-Meduim-bronw	2505
Sand, Red	405	Lime-gray-hard	2520
Shale, Red	430	Line-Sandy-Red-Medium	2525
Red Rock	470	Line-hard-gray-show oil at 2530	2535
Shale, Red	505	Shale-soft-red	2 545
Red Rock	545	Lime-hard-gray	2577
Shale, Red	640	Lime, brown	2581
Red Rock	745	Lime,gray,hard	2590
Shale, Red	810	Lime & sand-gray, meduim	2600
Red Shale	875	Lime, brown & gray-sôft	2609
Anhydrite	900	Lime, gray-hard	2646
Red Rock	915	Lime, gray-medium	2666
Selt	930	Lime, pay-soft gray	2674
Anhydrite-hard	1020	Sand-gray-soft	2676
Salt	107 0	Lime, gray-medium	2694
Anhydrite	1130	Sand, gray medium T.D.	2696
Red Rock	1140		
Salt	1170		
Anhydrite	1190	·	
Salt and Potash	1250		
Anhydrite	1265		
Salt and Shale	1290		
Salt and Potash	1380		1

NEW MEXICO SCHOOL OF MINES STATE BUREAU OF MINES AND MINERAL RESOURCES SOCGARO, NEW MEXICO

WELL LOG DIVISION

CASING	RECORD	ELEVATION 3510	FFFT
Diam in	Bottom	INITIAL DAILY PRODUCTION	ON:
10-3/4 8-5/8	450 946	Open 25 per hr. Open	bbls. Oi! cu. ft. Gas
7.	2613	Tbg.	bbls. Oil
		Tbg.	cu. ft, Gas

COUNTY	Lea			
FIELD	Halfway			
COMPANY	Argo Oi	l Corp.	•	
LEASE St	tete		w	ell No. 1
LOCATION	N (¥4)S/2 🗄	e/2 NV.,	4	
SEC.	16 т.	2 05	, R.	32E
2310	feet from	North	line and	i
1980	feet from	West	line of a	Section
COMMEN	CED 1-7-4	10		
COMPLET	ED 2 4	1 0		
ABANDO	VED			
REMARKS	s.			

FORMATION	BOTTOM, FEET	FORMATION		BOTTO	M, FEET
Caliche	15	Anhydrite, hard			2195
Sandy shale, red	75	Salt			2300
Red rock	145	Anhydrite, hard			2325
Red shale	215	Lime, hard, gray			2366
Send	225	Red rock			2369 9
Red rock	235	Limer, hard gray			2426
Red shale	245	Lime, hard, brown			2450
Water send	260	Sandy lime			2478
Red shale	285	Lime, hard, brown			2 486
Red rock, sandy	330	Lime, hard, gray			2497
Red sahle, sandy	400	Line, hard, brown			2 507
Red rock	455	Lime, hard, gray			2519
Red shale	610	Lime, hard, grav			2546
Red rock	630	Lime, hard, gray			2583
Red shale, hard	685	Soft sand, Show GAS			2625
Red rock	755	Grav lime, hard			2625
Red shale	800	Lime, soft, OTL		T.D.	2627
Red rock	845				
_Red shale	875				
Anhydrite	910				
Red shale	920				
Anhydrite	1020				
Salt	1070				
Potash, hard, red	1080				
Red rock	1090				
Anhydrite	1125				,
ked rock	1140				
Salt	1160			·	
Anhydrite	1177				
Salt	190				
Salt and red rock	1235			4	
Salt	1275				
Salt and potash	1325				
Selt	1380				
Anhydrite	1395	•	1.		
Salt	1500				
Anhydrite	1517		•		
Salt	1990				
Annyarite	2002	· ·			
Salt	2017				
Ann ydrite	2030		•		
Salt	2175				

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Log. No. 4555

NEW MEXICO SCHOOL OF MINES STATE BUREAU OF MINES AND MINERAL RESOURCES SOCORRO, NEW MEXICO

WELL LOG DIVISION

CASING 1	RECORD	ELEVATION 3489 INITIAL DAILY PRODUCTION		FEET
Diam., in	Bottom			X:
10-3/4 8-5/8 7"	431 920 2547	Open Open Tbg. Tbg.	400	bbls. Oi! cu. ft. Gas bbls. Oi} cu. ft. Gas

Salt and potash

COUNTY Lea '			
FIELD Halfway	•		
COMPANT Argo	Oil Corp.	•	
LEASE State		Wel	1 No. 3
LOCATION (%) N	ENW		
sec. 16	т. 205	, R. 3	2E
990 feet fr	om North	line and	
2310 feet fr	om J est	line of Se	ection
COMMENCED 2-	9-41		
COMPLETED 3-1	5-41		
ABANDONED			
REMARKS:			

FORMATION	BOTTOM, FEET	FORMATION	BOTTOM, FEET
Cellar	8	Anhydrite	1990
Caliche	15	Salt and potash	2172
Red sand	44	Anhydrite	2193
Red shale	100	Salt	2220
Red rock	183	Salt and potash	2265
Send 5 BWPH @ 205'	214	Salt, white	2295
Red shale	220	Anhydrite	2330
Red sandy shale	3 00	Hard gray lime	2340
Red rock 12 BWPH	335	Lime	2355
Red sandy shale 18 BWPH	375	Grav lime	2366
Red sandy shale	435	Red shale	2370
Red rock	520	Gray lime	2382
Red shale	565	Blue shale	2385
Red rock	640	Gray lime	2387
Red shale	685	Hard grav lime	2397
Red rock	· 740	Grav lime	2423
Red rock and shale	7 80	Hard grav lime	2430
Red rock	810	Grav lime	2450
Red shale	852	Broken lime	2478
Anharite	875	Lime	2485
Red shale	8 80	Red lime	2494
Anhydrite	95 5	Grav lime	2547
Anhydrite, white	985	Brown lime	2550
Salt, white	1050	Gray lime	2584
Anhydrite	1098	Broken brown lime	2593
Red and blue shale	1105	Grav lime	2630
Red shale and salt	1115	Hard gray lime	2635
Salt and red shale	1135	Broken lime	2650
White anhydrite	1150	Lime	2 6 ට
Salt and red shale	1155	Total depth	2681
Salt and red rock	125 5		
Salt and potash	1315		
Selt and anhydrite	1355		
Salt and potash	1475	•	
Anhydrite	1490		
Salt	1510		
Saltand potash	1560		1
Salt	1575		
Salt and potash	1805		V
Salt	1850		

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Log. No. 4528

NEW MEXICO SCHOOL OF MINES STATE BUREAU OF MINES AND MINERAL RESOURCES SOCOBRO, NEW MEXICO

CASING	RECORD	ELEVATION 3511 FEET
Diam., in	Bottom	INITIAL DAILY PRODUCTION:
10-3/4	456	Open In 8 hr. 85 bblbbls. Oil
8-5/8	932	Open cu. ft. Gas
7	2477	Tbg. bbls. Oil
		Thg. cu. ft. Gas

COUNTY Lea
FIELD Halfway
COMPANY Argo Oil Corp.
LEASETEXES-State "B" Well No. 3
LOCATION (%) NW NW SE
SEC. 16 T. 20S , R. 32E
2310 feet from South line and
2310 feet from East line of Section
COMMENCED 1-3-41
COMPLETED 2-2-41
ABANDONED
REMARKS:

FORMATION . BOTT	rom, feet	FORMATION	BOTTOM, FEET
Caliche	30	Anhydrite	1400
Red shale	45	Salt and potash	1425
Gravel	65	Salt	1485
Red rock	105	Salt and potash	1540
Red rock sandy 1/2 bailer w.p.h.	125	Salt	1630
Red rock	18 0	Salt and potash	1700
Sand	190	Anhydrite	1710
Red rock	200	Selt and potash	1820
Water send	230	Salt .	1870
Red rock sandy	280	Salt and potash	1995
Sandy shale	345	Anhydrite and potash	2010
Red rock	365	Salt	2085
Send with water	370	Anhydrote	2135
Red rock	390	Salt	21 70
Sand with 25 bailers water per hr	395	Anhvdrite	21,90
Red rock	459	SaltBase of salt	2285
Shele	485	Anhydrite	2322
Red rock	6 00	Anhydrite	2322
Shhed shale	615	Lime. grav	2445
Red rock	745	Lime. broken. sendy	2462
Shale	795	Lime, brown	2497
Red rock	865	Lime, grav	2516
Anhydrite	895	Shale, grav	2519
Shale, red	905	Lime, broken sandy	2525
Anhydrite	939	Lime, brown	2541
Salt	945	Lime, grav	2597
Anhydrite	1012	Lime, sendy - hole full OIL	T.D.2604
Selt	1060		
Anhydrite	1125		
Salt and red rock	1155		
Anhydrite	1170	·	
Salt and red rock	1175		
Anhydrite, salt and potash	1230		
Red rock and salt	1275		
Salt and potash	1365		

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Log No. 4483

NEW MEXICO SCHOOL OF MINES STATE BUREAU OF MINES AND MINERAL RESOURCES SOCORRO, NEW MEXICO

CASING	RECORD	ELEVATION 3	496	FRET
Diam., in	Bottom	INITIAL DAILY	PRODUCTIO	N:
12	420	Open	Dry	bbls. Oil
10 .	930	Open		cu. ft. Gas
8	2400	Tbg.		bbls. Oil
		Tbg.		cu. ft. Gas

COUNTY Lea
FIELD Halfway
COMPANNOrth Shore Corp.
LEASE TEXAS-State "B" well No.1
LOCATION (4) SW SF ME
SEC. 16 T. 20-5 .R. 32-E
2310 feet from North line and
990 feet from East line of Section
COMMENCED 10-7-40
COMPLETED 11-6-40
ABANDONED
REMARKS:

	FORMATION .	BOTTOM,	FEET	FORMATION	BOTTOM, FEET
	caliche		100		
	Sand		130		
-	Red bed		400		
	Sand and shale		770		
	sand and red bed		830		
	Anhydite		870		
	Anhy, sand and salt		892		
	Anhvdrite		900	· .	
	T.ime		925		
	Anhydrite		982		
	Salt		1035		
	Anhy. salt and sand		1065		
	T.ime		1100		
	Anhy. salt and sand		1175		
	salt		1270		
	Anhy. salt sand and potash		1355		
	Salt		1445		
	Anhy. salt and potash		1460		
	Salt		1575		
	Anhy. salt and potash		1725		
	Salt		1785		
	Anhy. salt and potash		1820		
	Salt		1860		
ļ	salt anhy. and potash		2020		
	Salt		2180		
	Anhydrite		2210		
	Salt		2330		
	Anhydrite		2371		
	Line		2410		
	Lime and sand	2420	275Ø		
	Lime		2 485		
	Limeand sand		2535		
	Sand		2585		\ \
	Sand and lime		2648)
	Lime sand and benite		2700		- -
	Lime		2705		
	Limeand sand		2715		
	Lime	T.D.	2728		

Log No. 4459

NEW MEXICO SCHOOL OF MINES STATE BUREAU OF MINES AND MINERAL RESOURCES SOCOBRO, NEW MEXICO

CASING RECORD		ELEVATION 348	I FEET
Diam., in	Bottom	INITIAL DAILY PR	ODUCTION:
10"	420	Open	480 bbls, Oil
8-5/8	901 * 9"	Open	cu. ft. Gas
5글	2494	Tbg.	bbls. Oil
		Tbg.	cu, ft. Gas

COUNT	r Lea						
FIELD	Halfway	y					
COMPA	NNort	n Shi	ore (lorp	ora	tion	
LEASE	Texas-	-Sta	te "/	T.n	We	ll No.	1
LOCATI	ION (4)	SW N	ii ne				
SEC.	16	Т.	20-9	3	, R.	32-1	3
990	feet fro	m No:	rth	line	e and		
2310	feet fro	om Ea	st	lin	e of S	ection	
COMME	NCED S	ept.	1, 1	1940			
COMPL	ETED S	ept.	28,	194	0		
ABAND	ONED						
REMAR	KS:						

FURMATION ,			FORMATION		BOTTOM, FEET
Sand & Calv.		110			
Water Sand		130			
Red Beds		250			
Sand & red shale		800			
Sand		840			
Anhydrite		875			
Sand & salt		890			
Anhydrite		910			
Anhydrite & lime		925			
Anhydrite		975			
Salt		1040			
Anhvdrite Shale & salt		1070			
Lime		1100			
Potash		1110			
Sand & Anhydrite		1120			
Anhydrite	ŧ	1140			
Sand & anhydrite		1170			
Anhydrite & salt		1200			
Salt		1275			
Salt & Potash		1330			
Anhydrite		1345			
Salt & Potash		1470			
Salt		1540			
Salt & Potash		1765			
Anhydrite		1785			
Salt & Potash		1930	i		
Anhydrite Potash & Salt		1980			
Salt		2035			
Anhydrite		2065			
Salt & Anhydrite		2130			
Anhydrite		2160			
Salt		2245			
Anhydrite		22QA			
Lime & Anhydrite		2310	•		
Brown Lime & sand		2342			
Brown lime & grev		2388			
Sand & lime		2430		1	
Sand Brown & grev Lime		2467		i i	
Sand		2490		V	
Lime	ToDo	2505			

stencil made V Log No. 4407

NEW MEXICO SCHOOL OF MINES STATE BUREAU OF MINES AND MINERAL RESOURCES SOCOBRO, NEW MEXICO

CASING	RECORD	ELEVATION 3511	FEET
Diam., in	Bottom	INITIAL DAILY PRODUC	TION:
10-3/4	450	Open	bbls. Oil
8-5/8	883	Open	cu. ft. Gas
7	2606	Tbg.	bbls. Oil
		Tbg.	cu. ft. Gas

COUNTY	r Lea
FIELD	Halfway
COMPAN	Argo Cil Corporation
LEASE	Texas-State "B" well No. 2
LOCATI	ON (%) 2_ S.,
SEC.	16 t. 20 , r. 32
<u>8</u> 90	feet from South line and
1650, сомме	teet from Nest line of Section NCED 3-12-40
COMPLI	ETED 7-8-40
ABAND	ONED
REMAR	ks: Swabbed only water.

FORMATION	BOTTOM, FEET	FORMATION	BOTTOM, FEET
Caliche	õ	Salt	1845
Anhydrite & Red Rock	50	Salt & Potash	1920
Red Shale	115	Salt d Fotash	2010
Red Hock	205	Anhydrite Salt & Potash	2050
Red Sand	215	Salt & Fotash	2120
Red Rock	255	Salt & Anhydrite Shells	2150
Red Sand - water	280	Anhdarite	2210
Rod Shale	325	Salt & Potash	2270
Red Hock	390	Salt	.2275
Sand - water	400	Anhvárite	2325
Red Rock	410	Lime	2422
Rei Shele	490	Lie	2460
Red Rock	550	Li e sicd Sandy	2476
Sandy Shale	560	Lime	2491
hed Rock	570	Red Line	2512
Red noch & Shale	586	Sand-Shoy dead 011 2518	2519
Red Shale	600	Line	2521
Red noch	920	Broken Sandy Line	2533
Red Shale	263	Lime	2579
Anhyārite	S93	Red Sandy Line	2536
Red Shale	877	Line - show oil 2629	2640
Anhydrite	915	Line & Bentonite	2662
Anhydrite & Salt Shalks	935	Lime - show oil 2685 T.D	. 2683
Anhydrite	970		•
Gray Anhydrite	1020		
Salt	1080		
Anhydrite (gray)	1135		
Anhydrite & Potash	1145		
Anhvdrite & Salt	1170		
Anhydrite	1175		
hed Rock Broken	1225		
Salt - Fotash	2295		
Salt a and Shal-	1510	•	
Salt	1390	1	
Anhydrite	1400		
Salt-potesh w/blue shells	1455	\checkmark	
Anhvarite, salt & rotash	1525		
Anhvarite Salt & Shells	1585		
Solt & Fotseh	1200		

/ stencil made LOS No. 4403

NEW MEXICO SCHOOL OF MINES STATE BUREAU OF MINES AND MINERAL RESOURCES SOCOBRO, NEW MEXICO

CASING	RECORD	ELEVATION	FEET
Diam., in	Bottom	INITIAL DAILY PH	RODUCTION:
8-5/8	926	Open	360 bbls. Oil
7	2367	Open	cu. ft. Gas
-		Tbg.	bbls. Oil
		Tbg.	cu. ft. Gas

county Léa
TELD Halfway
COMPANY Sam Weiner
EASE Wentz State well No. 1
OCATION (1/4) S., S. TE
ес. 16 т. 20 , в. 32
310 feet from horth line and
310 feet from Last line of Section
COMMENCED $5 - 30 - 40$
COMPLETED 7-11-40
BANDONED
REMARKS:

FORMATION	BOTTOM, FEET	FORMATION	BOTTOM, FEET
Caliche	15		
Red Sand	40		
Sand & Gravel	50		
Red Bed	200		
Sand	240		
Red Bed & Sand	360		
Sandy Shale	400		
Red Bed	500		
Shale & Gyp	530		
Red Rock & Red Bed	265		
Likyärite (910		
Shale & Jalt	920		
Salt & Anlydrite	1030		
Anhydrite / Potash	1110		
Salt &hydrite	2294		
Anhydrite	2532		
Line (Show of Gas at 2338)	2394		
Brown Line	2405		
Î.i.:C	2444		· ·
Grey Sandy Line	2467		
Brown Line	2492		
Brown & Gray Line	2495		
Lime - Show of Oil	2520		
Broken Line	2530		
Mard Gray Line	2556		
Soft Line - Lole filled with	0il 2559	T.J.	

NEW MEXICO SCHOOL OF MINES STATE BUREAU OF MINES AND MINERAL RESOURCES SOCOBBO, NEW MEXICO

CASING RECORD		ELEVATION	3510	DEEC
Diam., in	Bottom	INITIAL DAILY	PRODUCTION	FEEI
L0-3/4"	450	Open	460	bbls. Oil
8-5/8"	950	Open		cu. ft. Gas
711 .	2026	Tbg.		bbls. Oil
		Tbg.		cu. ft. Gas

COUNTY Lea
FIELD Halfway
COMPANYArgo Cil Corporation
LEASE State Well No. 2
LOCATION (%) SHING NW
SEC. 16 T. 20-S.R. 32-E
2310 feet from North line and
4290 feet from East line of Section
COMMENCED 2-2-40
COMPLETED 3-17-40
ABANDONED
REMARKS:

FORMATION	BOTTOM, FEET	FORMATION	B	OTTOM, FEET
Caliche	20	Anhydrite		2315
Shale Soft Red	115	Anhydrite & Brown	1 Lime Hard	2333
Red Rock	170	Lime Hard Gray		2385
Sandy Shale Red	330	Shale Soft Blae		2388
Red rock	365	Lime Hard Gray		2433
Sand	385	Lime Hard Brown		2449
Sandy Red Shale	425	Lime Hard Gray		2475
Shale Red	47 5	Lime HardBrown		2502
ned nock	525	Lime Hard Grav		2509
Red Shale	580	Lime Hard Brown		2545
Sandy Shale Red	710	Lime Hard Grav		2589
hed nock	725	Sand		2599
anhydrite	73 5	Lime Hard Brown		2610
Sandy Shale Red	805	Lime Hard Grav		2677
Red Rock	870	Lime Hard Brown		269279
Anhydrite	910	Lime Soft Grav		2682
Shale Soft Red	915	Line Hard Grav		2688
Anhyàrite & Shale 💦 🔪	93 5	Line Ned Gray	ΠD	2680161
Anhydrite Hard	1013	aruy		2009 0
Salt	1075			
Potash	1080			
Shale Blue Soft	1085			
Anhyarite Eard	1125			
Red Rock	1135			
Salt	1160			
Anhvarite Hard	1175			
Salt & Potash	1260			
Salt & Red Rock	1315			
Salt	1375	•		
Anhvdrite Hard	1390			
Salt	1500			
Anhydrite Hard	1520			
Salt	1900			
Salt & Potash	1940	•		
Selt	2000			
Anhydri te	2010			
Selt	2030			
Anbydrite	2000 2032	•	1	
Solt & Potesh	2005	• • • • • • • • • • • • • • • • • • •	J	
Salt -	2070 20175	· ·		
	6110 6110			
Annyarite	2205			
Dalt	2500			

LOF NO4018

NEW MEXICO SCHOOL OF MINES STATE BUREAU OF MINES AND MINERAL RESOURCES SOCORBO, NEW MEXICO

CASING I	RECORD	ELEVATION 3510	FEET
Diam., in	Bottom	INITIAL DAILY PRO	DUCTION:
10-3/4	420	Open	240 bbls. Oil
8-5/8	931	Open	cu. ft. Gas
7"		Tbg.	bbls. Oil
-		Tbg.	cu. ft. Gas

COUNTY Lea
FIELHAlfway
COMPANY Argo Oil Coporation
LEASE TEXAS-State Well No.1
LOCATION (%)NW SW
SEC.16 T. 20-S, R-32-E
1980 feet from Sth line and
660 feet from West line of Section
commenced10/17/39
COMPLETED 11/14/39
ABANDONED
REMARKS: Texas Co. farm-out

FORMATION	BOTTOM, FEET	FORMATION	BOTTOM, FEET
Caliche	35		
Red Rock Red Shale & Sand	165		
Red Shale & Sandy	4 54		
Red Rock and Shale	875		
Anhyd. & Red Rock	1025	-	
Salt Potash & Red Rock	1095		
Anhyd. & Salt	1185		
Red Rock & Salt	1230		
Salt & Anhyd.	2330	-	
Lime Harā Gray	2450		
Lime Med Gray	2475	•	
Lime Hard Gray	2527		
Soft Lime	2610		
Lime	2657	TD	

Log. No. 3924

Viller Charles

NEW MEXICO SCHOOL OF MINES STATE BUREAU OF MINES AND MINERAL RESOURCES SOCOBRO, NEW MEXICO

WELL LOG DIVISION

CASING RE	CORD	ELEVATION		3512) FFET
Diam., in	Bottom	INITIAL D.	AILY	PRO		TION:
10 3/4"	421'	Open				hbls, Oi!
8 5/8"	10001	Open	P,	යි	Α.	cu. ft. Gas
7"	2590 (Tbg.				bbls. Oil
		Thg.				cu. ft. Gas

COUNTY Lea
FIELD Halfway
COMPANY Argo Oil Corporation *
LEASE TERAS-State "B" Wen No. 1
LOCATION (%) NE SW
SEC. 16 T. 205 . R. 32E
1980 feet from SOUth line and
1980 feet from West line of Section
COMMENCED 9-13-39 - Deepened 5-25-4
COMPLETED 2-21-40 - Deepened 8-19-1
ABANDONED *Formerly West Lea Oil Co
REMARKS: *Formerly F. M. Farley.
**Formerly Tex-State No. 1-

FORMATION	BOTTOM, FEET	FORMATION		BOTTOM, FEET
Lime	10	Anhydrite		2320
Caliche	25	Lime brown		2760
Yellow sandy shale	45	Lime cray		2700
Red shale	85	Lime, brown		2500
Red shale and sand	190	Shale and shella		2472
Lime	200	Lime, red	·	2410
Red shale and sand	215	Shale and shells		2525
Lime, sand and shale	265	Lime		2929
Red sand; water	270	Sandy line		27(7)
Red mud	290	Grav lime		2070
Red sand	325	Cray lime		2620
d bed	355	Lime		2019
ned shale	420	Sandy lime sharp		2722
Red rock	421	Lime hard gray		2765
Red rock and shells	455	Lime and synsum		2780
Red shale and shells	530	Lime		2/00
Lime and red rock	670	Sand, soft		3036
Red rock and shells	760	Lime: HFW		376z m n
Red shale and shells	873			
Anhydrite	910			
Red rock and salt shells	920			
Anhydrite	938			
Anhydrite and lime	1020			
Anhydrite and salt	1040			
Salt	1085			
Red shale	1090			
Anhydrite	1105			
Lime	1135			
Red rock	1140			
Red rock, salt and shells	1250		\backslash	
Anhydrite, salt and red ro				
Salt	1390			
Anhyarite	1405	•	•	
Salt	1460			
Salt and shells	1525			
	2000			
Sait and annyarite	2030			, · · · ·
ALV, WRITE	2090			
Annyurite, sait and potasn	2105			
Dall and Shells	2100			
Annyarite	2200			
Salt and shells	2223			

Log No. 1129

NEW MEXICO SCHOOL OF MINES STATE BUREAU OF MINES AND MINERAL RESOURCES SOCORRO, NEW MEXICO

WELL LOG DIVISION

CASING RECORD		17° I. F	EVATION 3460	TTT
Diam., in.	Bottom	INI	TIAL DAILY PROD	UCTION:
151	45	51	Open	bbls. Oil
12	93	51	Open	cu. ft. Gas
8 <u>1</u>	245	91	Tbg.	bbls. Oil
6 5/8	321	.3	Tbg.	cu. ft. Gas

COUNTY Lea		
FIELD		
COMPANY	stern Drilli	ng Co.
LEASE Stat	te	Well No.
LOCATION (%)	SE NE NE	
SEC. 16	t. 205, r.	- 32 E
	feet from	line and
,	feet from	line of Section
COMMENCED	1-29-31	
COMPLETED	9-10-31	
ABANDONED		
REMARKS:		

Abandoned and Plugged.

FORMATION	BOTTOM, FEET	FORMATION BOT	BOTTOM, FEET	
Light gyp	15	Anhydrite	2210	
Red beds	160	Salt and anhydrite	2260	
water	165	Salt	2300	
Red rock	230	Salt and anhydrite	2315	
Water	235	Anhvārite	2360	
Red rock	425	Grav sandy lime	2375	
Water sand	430	Grav hard lime S.G. 2390	2385	
Red rock	455	Grav lime	2405	
Red beds	505	Eroken lime with red and	~	
Red rock	535	brown shale	2410	
Red rock and gyp	615	Anhydrite	2425	
Red rock	720	Grav lime	2435	
Hed rock and red beds	755	Grav lime and anhydrite	2445	
ned rock	780	Grav lime	2448	
nhvdrite	925	STAR	2459	
Red beas	930	Grav lime	2500	
Salt	935	Grav lime and blue shale	2505	
Annvörite	940	Lime and shale	2515	
Salt	950	Red and green shale	2520	
enhvdrite	9 80	Hard lime and shale	2530	
White selt	985	Lime shells and brown shele	2545	
Anhvdrite and salt	1020	Grav lime	2555	
anhydrite	1035	Shelly line and shale	2575	
Anhydrite and lime shells	1050	Grav sand S. dead 0 at 2575-80	2580	
Red mud	1060	Herd white lime	2600	
Salt and anhydrite	1125	white lime	2625	
red shale	1135	Brown sandy lime and blue sha	ale	
Salt and anhydrite	1265	Brown bond, Fino did bido on	2645	
balt	1285	.hite lime	2655	
Anhvāri te	1295	White line and bentonite	2705	
Selt	1325	white lime	2740	
Salt and anhydrite	1390	Grav line 400' sulphur water		
-nhvdrite and slat	1460	in $1\frac{1}{2}$ hours from 2745!	2750	
Salt and anhydrite	1705	White water sand and lime	2760	
Salt	1855	.hite lime	2780	
Salt and anhydrite	1975	Dark lime increase water	2790	
Salt	1995	Grav lime	2810	
Inbydrite eir nocket	2000	Dark gray lime	2820	
sel+	2105	Grav line	2930	
nhudrite end selt	9145	Slue sendy lime	2845	
WINATTRE' CHA SCTA	9170	Erev candy lime	2845	
Darv Thite colt and contrarito	9195	Cher line	2990	
"UTCE SETC ENG EUNYOFICE	0010	GIRY TIME	200V	
JIEC	STA0			

Log No. 1129

NEW MEXICO SCHOOL OF MINES STATE BUREAU OF MINES AND MINERAL RESOURCES SOCOBRO, NEW MEXICO

WELL LOG DIVISION

	ELEVATION	FEET
Diam., in. Bottom	INITIAL DAILY PRODUCTION	N :
	- Open	bbls. Oil
· ·	Open	cu. ft. Gas
	Tbg.	bbls. Oil
	The	cu. ft. Gas

COUNT	ry Lea		
FIELD			
COMP	ANT .e	stern Drilli	ng Co.
LEASE	2	State	Well No. 1
LOCAT	ION (%)		
SEC.	16	т. ZO, к.	32
		feet from	line and
		feet from	line of Section
COMMI	ENCED		
COMPL	ETED		
ABANI	DONED		
REMAN	RKS:		

FORMATION	BOTTOM, FEET	FORMATION	BOTTOM, FEET
White lime	2895	White lime	3945
Gray lime	2910	Hard white lime	3955
White lime	2925'	White lime	3 98 3
Gray lime	2975	white lime	4013
"hite lime	2990	SLM TD	4005
hard white lime	3010	<u>.</u>	
"hite lime	3020		
Gray sandy lime	3035		
Gray lime	3045	·	
hite lime	3120		
Linite lime	3166		
Gray lime	3175		
white lime	3184		
Gray lime	3192		
White lime	3200		
SLM	320 8		
Gray lime	3213		
Lhite lime	3225		
Cray lime	3248		
Lhite lime	3265		
white lime	3275		
Anhydrite	3282		
Anhydrite and white lime	330 3		
Gray and white lime	3324		
Brown lime	3411		
Brown and hard gray lime	3431		
hite lime	3474		
Gray lime	3505		
white sandy lime salt water	3520		
white lime	3638		
Thite lime 1730' water	3705		
White sandy lime increase w	ater 3718		
hite line and sand	3723		
hite sandy lime	3747		
Gray sandy lime	3760	•	
white lime	3769		
Gray sandy lime and bentoni	te 3776	_	
Gray lime	3857		
Gray lime	3865		
Gray sandy lime	3889		
Gray sand	3900		
Gray sandy lime	3905		
Grav lime	3922		

NEW MEXICO SCHOOL OF MINES

STATE BUREAU OF MINES AND MINERAL RESOURCES SOCORRO, NEW MEXICO

NEW MEXICO WELL LOG DIVISION

CASING RECORD		FLEVATION	34	65	SW	~ ▼ ▼▼▼▼▼
Diam., in.	Bottom	INITIAL DAI	LY F	ROD	UCTION:	OX FEEL
24	18	Open	_		•	bbls. Oil
$15\frac{1}{2}$	434	Open	P.	Æ	А,	cu. ft. Gas
125	798	Tbg.		•.		bbls. Oil
8.	2540	Tbg.				cu. ft. Gas

COUNTY Lea FIELD Sant Later COMPANY The Texas Company LEASE Humphreys No. 1 Well LOCATION (%) NE NE SW 32 E , T. 20 5 , R. SEC. 18 feet from S 2310 line and 2310 W feet from line of Section 4-26-29 COMMENCED **4** 7-11-29 COMPLETED ABANDONED

REMARKS:

	FORMATION BOTT	OM, FEET	FORMATION BOTTO	M, FEET
	Gypsum and caliche	35	White salt and brown shale	2455
	Red sandy shale	130	Brown and gray anhydrite	2470
	Red sand	210	White and gray anhydrite	2525
	Red Sandy shale	795	Tan and brown dolomite	2575
-	White and gray anhydrite	820	Grav sand, little lime	2585
	Red shale and salt	845	Gray sand and green sandy shall	le
	Gray and white anhydrite	865	dray band and groom bandy bho.	2595
	Brown dolomite	875	Buff lime and hentonitic shale	A
	Brown dolomite. little anhydr	ite	Pari rime and perioditore shart	2615
		885	Light and huff lime	2630
	Grav and white anhydrite	945	Light and buff lime and green	2000
	Pure white salt	1005	Digne and buil lime and green	2640
	Salt. anhydrite and gray shal	e1015	Light and buff lime and block	0FUG
	Salty. red shale	1025	Digno and puri lime and place	2660
	Pink salt	1035	Brown lime some red and green	0002
	Grav and white anhydrite	1055	BIOWH LIME, BOME IEU AHU BIEE	
	Brown dolomite	1075	Sandy Shale	2070
	galty red chale	1085	Gray and tan 11me	2090
	Dink colt	1105	Red and green sand	ູລາບບ
	Fink ond white enhydrite	1115	Red and green sand, little red	1
	Pink and white annyulite	1125	snale, some gray lime	2710
	Link Ralt, come red chole	1135	white lime and dolomite, some	
	Pink salt; some led shale	1165	red lime and red sandy sha	ale
	Red shary salu Diak and white calt	1105		2770
	Pink and white sait	1305	White and red sandy shale	2795
	Pink Ball	TOOD	Red sandy lime, little white	
	Pink and white sait, some gra	, 1330	lime	2816
		1350	White and pink dolomite	2840
	White Balt	1360	Grayish white dolomite	2850
	Salt and red and gray shale	1000	White dolomite, trace red shall	Le
	Pink sait	1500		2885
	Hed shale and salt	1090	White and red dolomite and red	1
	Pink sait	1(40	sand	2905
	Pink salt, little red poly-	1050	White dolomite	2910
	halite	1820	Red and white dolomite	2919
	White salt	18.40	White dolomite	2960
	Brown dolomite and annydrite	1900	White dolomite and bulish gray	Ϋ́
	Pink salt and polyhalite	2140	bentonite shale	2985
	White salt	2200	White and red dolomite and blu	lish
	Pink salt	2300	gray bentonitic shale	2995
	White salt; little anhydrite	2330	Red sandy dolomite, white dole	omite
	White salt; trace of brown	L.	and green bentonitic shall	-3000
<i>.</i> `.	shale	2425	Green bentonitic shale red a	
	White salt	2445	dolomite and little 100 80	anuy
			automite and little white	

ろ

NEW MEXICO SCHOOL OF MINES

1-

E BUREAU OF MINES AND MINERAL RESOURCES SOCORRO, NEW MEXICO			COUNTY Lea Field			
NE		O WELL LO	DG DIVISION	COMPANY T LEASE HUI LOCATION (%)	he Texas C mphreys NE NE S	ompany No. 1 well
CASING	RECORD Bottom	ELEVATION INITIAL DAILY	3475 Approx FEET PRODUCTION:	sec. 18 3210 3210	. T. 20 S feet from S feet from	, R. 32 E line and W line of Section
24	18	Open	bbls. Oil	COMMENCED	4-26-29	
155	434	Open	cu. ft. Gas	COMPLETED	7-11-29	
125	798	Tbg.	bbls. Oil	ABANDONED		
81	2540	Tbg.	cu. ft. Gas	REMARKS:		

FORMATION	BOTTOM, FEET	FORMATION	BOTTOM, FEET
White dolomite, little gree	en		
bentontic shale	3020		
White dolomite and green			
bentonitic shale	3028		
Green bentonitic shale, tra	ace		
white sand	3031		
White dolomite, some green			
bentonite and little sa	and 3036	· .	
White dolomite	3100		•
White dolomite, porous and			
showing dead oil, hole	full		
of sulphur water. Plug	ged		
and abandoned 7-10-28	3105		







STATE OF NEW MEXICO ENERGY AND MINERALS DEPARTMENT OIL CONSERVATION DIVISION



50 YEARS

GOVERNOR

October 21, 1985

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87501 (505) 827-5800

Petro-Thermo Corporation P.O. Box 2069 Hobbs, NM 88241-2069

Attention: Robert W. Abbott

Re: Permit for Temporary Use Disposal Pit

Dear Mr. Abbott:

A 90-day permit for temporary use of the emergency overflow pit located at the Blinebry-Drinkard SWD System Well No. A-22 in Section 22, Township 22 South, Range 37 East is hereby granted. This permit authorizes the temporary disposal of brine and fresh water-based drilling mud and waste cement into the existing pit with the following provisions:

- 1) The pit will not be enlarged or allowed to overflow or breach.
- 2) Within 90 days, or by January 19, 1986, solid waste will be removed from the pit and disposed of in an approved manner.

Authorization for this temporary use of the pit may be rescinded if there is evidence that the pit has overflowed or if Water Quality Control Commission or Oil Conservation Division rules or regulations have been violated.

Sincerely

S. Frence Salar St.

R. L. STAMETS Director

RLS/JB/dp

cc: Jerry Sexton David Boyer

STATE OF NEW MEXICO OIL CONSERVATION DIVISION	ANDUM OF MEETING	G OR CONV	ERSATION
Telephone Personal	Time 10:30	AM	Date 10/21/85
Originating Party			Other Parties
Jomi Bailey		Robt.	abbet - VP Petro - Thermo
Subject		393	-2417
Demporary user of	easthan p	4020	P.205. 37 E you due poon
- of dulling much + w	acto, camen	Ø. .	
Discussion	Charlo Attach	- 10/0/	25 Ages 7 : 12 - 12 - 2
(Inlined put is sussantly sme	roence overt	z-19/0 Duspet	Agua Hou Suld Well A-22. Of is
appropriately 15' × 30' × 8	· Jucks an	e cure	nthe being wachen out into pit.
Expected volume for to	myorary co.	e, io 30	2 bbls /week of dulling mud
and waste cement			
		·····	
	<u></u>		
	<u> </u>		······································
· · · · · · · · · · · · · · · · · · ·		· · · · · · · · · · · · · · · · · · ·	
<u>Conclusions or Agreements</u>			
			· · · · · · · · · · · · · · · · · · ·
Distribution 7 cle	Sig	gned	Smi Bile
		7	~ 7
•			-



P.O. BOX 2069 PHONES (505) 393-2417 — 397-3557 HOBBS, NEW MEXICO 88241-2069

October 9, 1985

CONSERVATION DIVISION

Mr. R.L. Stamets, Director New Mexico Oil Conservation Div. Post Office Box 2088 Santa Fe, New Mexico 87501

Dear Mr. Stamets:

Mr. Phil Baca has advised Petro-Thermo Corporation to seek your approval for the temporary use of an earthpit located in Section 22, Township 22, Range 37. Therefore, Petro-Thermo requests that the NMOCD provide a temporary one year permit to use this open pit for the purpose of washing out our transport trucks that have carried drilling muds and waste cement. No hydrocarbons (BS) will be put into the pit and all produced waters will be hauled to an approved salt water disposal well.

Currently Petro-Thermo is seeking approval for a large solids disposal site located near Laguna Plata in Section 16, Township 20 South and Range 32 East. Since this is a previously exempted area, approval should not take long after the hearing. We would appreciate any assistance you might give in acquiring proper approval for this project.

After such approval is given, Petro-Thermo will clean out the temporary pit and will wash out all trucks at the new solids disposal site.

Please contact our office if there is any further information you may require.

Yours Sincerely,

Petro-Thermo Corporation

A.M.d.

Robert W. Abbott Vice President

RWA/aj
AGUA, INC. POST OFFICE BOX 1978 HOBBS, NEW MEXICO 88241

> NG. BR

TELEPHONE: 505 393-6188

September 30, 1985

State of New Mexico Oil Conservation Division P.O. Box 1980 Hobbs, New Mexico 88240

Attn: Jerry Sexton

Dear Mr. Sexton:

This letter is in reference to your letter dated September 23, 1985 regarding repair of disposal wells, supplying requested information, and the use of unauthorized pits.

CATO SWD WELL P-5

Tubing was inspected and bad joints replaced June 4, 1985. A bottom hole pressure test was then run August 28, 1985. The results of this test indicated a pressure of 294 PSI should exist at the surface. A pressure of 320 PSI was reached when oil was pumped down the annulus June 21, 1985 and July 10, 1985 but dropped rapidly within an hour to 170 PSI. After several days the pressure on the casing reached 0 PSI. Because this well only injects an average of 4700 bbls/month, AGUA must give the operators a chance to decide if a casing repair is economically feasible. Therefore AGUA will send out an authority for expenditure form to all Parties of the Cato SWD System to seek approval for the suspected casing leak. Approximately 30 days will be required from October 1, 1985 to seek approval. An additional 14 days will be required to actually repair the suspected casing leak.

GOODWIN SWD WELL E-31

The annulus and tubing were bled down July 18, 1985. A tubing pressure of 1400 PSI and a casing pressure of 1300 PSI resulted. This indicated either obstructions or oil in the tubing. Therefore we have been trying to rid the tubing of both oil and/or obstructions by periodically bleeding down the tubing. Approximately October 21, 1985 the wellbore will be cleaned out and the annulus loaded with oil. Page 2 OCD Letter Dated September 23, 1985 AGUA; Repair of SWDS's

BLINEBRY-DRINKARD WELL NO. N-18

A bottom hole pressure test was performed August 15, 1985. Results of this test indicated no tubing or casing leaks existed. Mechanical integrity is excellent. Since the OCD judgement pertaining to the integrity of the well proved wrong, AGUA has had to proceed with extreme caution with requests to repair other wells or provide expensive information.

BLINEBRY-DRINKARD SWD WELL A-22

On July 1, 1985 a lightening storm caused an electrical outage at AGUA's well. Electricity is used to pump the water down the wellbore. When the electricity went out, the pump could no longer inject the produced water. When the tank level reached the top, several boards along the bottom of the redwood tank cracked. During this period, SWD Well No. N-18 or SWD Well No. H-35 could not be tested further until the repair was finished July 14, 1985.

BLINEBRY-DRINKARD SWD WELL NO. H-35

The annulus was bled off and reloaded with oil on June 11, 1985. The greatest differential recorded was 150 PSI. A small amount of oil was seen in the tubing and therefore repairs are scheduled to start September 26, 1985. The well will be cleaned out with a special chemical to determine if obstructions inside the tubing caused an inaccurate pressure differential. If the 300 PSI pressure differential is not achieved, the well will be bled down so that a pulling unit can be put on the well. At least 30 days will be required to bleed the pressure down. The water coming to the injection well as well as the water coming from bleeding the well down will be diverted to SWD Well No. A-22. The tubing will then be pulled and tested for leaks. Bad tubing will be replaced. At least 15 days will be required after the well is bled down to repair the tubing. Page 3 OCD Letter Dated September 23, 1985 AGUA; Repair of SWDS's

EMERGENCY PITS

The emergency pits at H-35 are currently being emptied. Recent problems with tank repairs and the electric motor being burnt out at A-22 have caused water to run into the pits at A-22 and H-35. The pit at C-2 has had water in it only after a rain storm. No production water of any kind has ever been in the pit at C-2. Any tracks found at the emergency pits at H-35 and A-22 were caused by trucks pulling water from the pits and not unloading into the pits.

AGUA suggests that the OCD contact Don Johnson of Conoco to clear up the matter of trucks dumping at the Goodwin SWD System.

BOOT RUNOVERS

Ninety percent of all boot runovers have been caused by excessive amounts of solid matter, basic sediment, and oil being delivered into AGUA's Systems. Faulty dump valves also contribute to adding excessive amounts of gas to the System causing boot runovers. AGUA's contractual agreement with the Parties of each System states "only water reasonably free from solid matter, basic sediment and oil shall be delivered to the System." Therefore AGUA feels responsible for only a small percentage of boot spills. Since most of our roustabout time is spent with oil and basic sediment clean up, correction of other problems becomes more difficult.

AGUA, Inc. is presently working on plans for a large solids disposal site. Our management has contacted Mr. Phil Baca, Environmental Engineer with the OCD in Santa Fe and the State Land Office on obtaining a suitable site and plans for this disposal.

Yours very truly,

AGUA, Inc.

W.G. Abbott Manager

JDT/dlb

xc: R.L. Stamets - OCD Director Jeff Taylor - OCD Attorney

ENERGY AND MINERALS DEPARTMENT DIL CONSERVATION DIVISION

TONEY ANAYA GOVERNOR

November 30, 1983

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87531 (505) 327-5300

Mr. Charles S. Dahlen Bureau of Land Management Box 1778 Carlsbad, New Mexico 88220

Dear Mr. Dahlen:

The Oil Conservation Division would not object to Kerr-McGee disposing of tailings into Laguna Plata. Under Order No. R-3725 the Oil Conservation Commission approved brine disposal into Laguna Plata and Laguna Gatuna.

Laguna Gatuna is presently being utilized by Pollution Control as a disposal site for oilfield wastes and should not be considered for utilization by Kerr-McGee.

Since state lands are involved, I will forward your letter of November 2, 1983, to the State Land Office.

Yours very truly,

JOE D. RAMEY Director

JDR/fd

1	STATE OF NEW MEXICO ENERGY AND MINERALS DEPARTMENT
2	OIL CONSERVATION DIVISION STATE LAND OFFICE BLDG.
3	SANTA FE, NEW MEXICO
4	8 August 1984
E	EXAMINER HEARING
3	
6	
7	
8	IN THE MATTER OF:
9	Application of Pollution Control, CASE
10	Inc. for amendment to Division 8292 Order No. R3725, Lea County, New Mexico.
11	
10	
1 <i>4</i> .	
13	BEFORE: Richard L. Stamets, Examiner
14	
15	TRANSCRIPT OF HEARING
10	
16	
17	
18	APPEARANCES
19	
20	For the Oil Conservation W. Perry Pearce
21	Division: Attorney at Law Oil Conservation Commission
22	State Land Office Bldg. Santa Fe, New Mexico 87501
23	For the Applicant: W. Thomas Kellahin Attorney at Law
24	KELLAHIN & KELLAHIN P. O. Box 2265
25	Santa Fe, New Mexico 87501

INDEX TIM KELLY Direct Examination by Mr. Kellahin Cross Examination by Mr. Stamets Questions by David Boyer EXHIBITS PC Exhibit One, Report **17**[′]

3 1 2 STAMETS: The hearing will MR. 3 please come to order. 4 We'll call next Case 8292. 5 MR. PEARCE: That case is on 6 the application of Pollution Control, Inc. for amendment to 7 Division Order No. R-3725, Lea County, New Mexico. 8 MR. KELLAHIN: If the Examiner please, I'm Tom Kellahin of Santa Fe, New Mexico, appearing 9 on behalf of the applicant and I have one witness to be 10 sworn. 11 MR. PEARCE: Are there other 12 appearances in this matter? 13 14 (Witness sworn.) 15 16 TIM KELLY, being called as a witness and being duly sworn upon his 17 oath, testified as follows, to-wit: 18 19 DIRECT EXAMINATION 20 BY MR. KELLAHIN: 21 0 Mr. Kelly, for purposes of the record 22 would you please state your name and occupation? 23 Α My name is Tim Kelly. I'm from Albuquer-24 que and I'm a consulting hydrologist. Q Mr. Kelly, as a hydrologist, have 25 you

1	4
2	previously testified before the New Mexico Oil Conservation
3	Division and had your qualifications as a hydrologist ac-
4	cepted and made a matter of record?
5	A Yes, they have been.
5	Q And have you prepared, pursuant to your
0	employment by Pollution Control, a hydrologic assessment of
7	the area involved in this application?
8	A Yes, I have.
9	MR. KELLAHIN: We tender Mr.
10	Kelly as an expert hydrologist.
11	MR. STAMETS: He is considered
12	qualified.
13	Q Mr. Kelly, let me refer to your package
14	of exhibits, which we have simply marked as Exhibit Number
15	One, and ask you to turn to page 28 of that report, and as
16	an introduction for the Examiner, would you describe for us
10	Control operations in the Laguna Catura area?
17	A Yes sir In February of 1969 Pollution
18	Control retained the services of Ed L. Reed of Midland.
: 19	Texas, to prepare an assessment of the area referred to as
20	the salt lakes in western Lea County. It included Laguna
21	Gatuna, Laguna Plata, and Laguna Tonto. And an application
22	was made at that time. I believe it's Case Number 4047; was
23	heard on March 19th, 1969, in which Pollution Control re-
24	quested the use of Laguna Gatuna and Laguna Plata and Laguna
25	Tonto as a site for disposal of oilfield brine.

1	. 5
2	The application was approved for use at
3	of disposal of oilfield brine in Laguna Gatuna and Laguna
4	Plata, and subsequently, Pollution Control began operations
-	at Laguna Gatuna, which is shown in detail on page 28 of Ex-
5	hibit One.
6	This shows in the north half of Section
7	18 of Township 20 South, Range 32 East, the present site of
8	their operations, which have which they have had in oper-
9	ation since 1969.
10	They have also proposed a new site on the
11	southeast site of Laguna Gatuna, which is shown in Section
10	17.
14	Q All right, sir, if you'll turn to the
13	first appendix following page 36 in the Exhibit Number One,
14	is that the Commission Order R-3725 that you've made refer-
15	ence to that's approved the current operations of Pollution
16	Control at Laguna Gatuna?
17	A Yes, it is.
18	Q All right, sir. Would you outline for us
19	generally, Mr. Kelly, what Pollution Control proposes to do
20	at its site in the southwest quarter of Section 17?
20	A They propose to use this site in addition
21	to their existing site for disposal of oilfield brine and
22	waste products from the oil industry at this site.
23	I might mention that the site has been
24	selected because of the lease which they presently have and
25	also its proximity to Highway 62/180, which makes it some-

. . .

2 what more accessible to trucks.

1

5

3 Q What was the purpose of having Pollution
4 Control retain you as a hydrologist to study this area?
What were you looking to study, Mr. Kelly?

6

A The plans by Pollution Control were to add the additional site in Section 17 and at the same time to update the hydrologic assessment of the area, since their operation had been continuing for fifteen years, to determine if there had been any adverse effects from their previous operations and what the effect of the new site might be on the hydrologic system.

12 Q In going about studying for that goal, what information did you review and what studies did you un-13 dertake?

14 A The first thing we did was review the
15 Reed study in detail and the Reed study consisted primarily
16 of one illustration or exhibit, which was used in 1969, and
17 that is included in our report as a plate.

18 Q All right, let's look at that for a minute. Let's unfold one of those and look at it.

20 Let me try and understand what this is.
20 This represents Mr. Reed's work as consulting hydrologist
21 and is the basis upon which the 1969 order was entered ap22 proving Pollution Control's use of Laguna Gatuna for a dis23 posal site?

sir, and this, then, was the

That's correct.

All right,

Α

0

25

7 1 basis where you started your review of this property. 2 Right. We first of all reviewed the tes-А 3 timony from the hearing and then reviewed the map. We then made a literature and file search 5 of available data, of which a considerable amount had been 6 collected in the past, both from work which we had done or 7 the Bureau of Land Management in that area, and the WIPP site studies, which are nearby, and then we made an on-site 8 evaluation in which we actually went into the field, updated 9 the geologic map as best we could. We looked at the water 10 quality information and the water levels which Reed had 11 measured, as well as interviewed Mr. Snyder -- excuse me, 12 Mr. --13 Squires. Q 14 Mr. Squires with Snyder Ranches, and to Α 15 determine what the history of the water use in that area was. We also talked to some additional ranchers. 16 On the basis of this we prepared our re-17 port which is submitted here. 18 All right, sir, on Exhibit -- page 34 of Q 19 Exhibit Number One, is that a tabulation of the reference 20 material and other studies that you reviewed and included in 21 your analysis of this area? 22 Α Yes, it is. 23 0 Has a --24 I might -- I might mention that these re-А ferences are the ones which are specifically referenced in 25

8 1 These are not necessarily all the ones we evaour report. 2 luated. 3 As an expert hydrologist, Mr. Kelly, do 0 4 you believe that you had an adequate data base from which to 5 reach certain conclusions with regards to the continued 6 suitability of Laguna Gatuna as a disposal facility? 7 Yes, I do. А 8 Before we go into detail on the 0 facts surrounding your conclusions, Mr. Kelly, I think it might be 9 helpful if we simply turn to page 30 of your report and have 10 you give us a general synopsis from page 30 and 33 of the 11 six major conclusions you have reached based upon your 12 study, and then we'll go back and talk about each one of 13 those items. 14 All right. Laguna Gatuna is a natural А 15 ground water discharge point. It is the site where the fa-16 cility is now in operation. The information in that area indicates that the ground water is naturally discharging in-17 to Laguna Gatuna so that the flow is to the lake rather than 18 away from it. 19 The same thing is true of Laguna Plata, 20 which is also shown on this plate several miles to the 21 northwest. 22 second concludion we made was The that 23 natural discharge from springs at Laguna Gatuna and Laguna 24. Plata is much more highly mineralized than the water that is being produced from wells in the area or from the water 25

1 which is being disposed of by Pollution Control, Incorpor-2 ated. So the natural water is worse than what is being put 3 in there from the oilfield sources. 1 All right, sir. 0 5 Α Our third conclusion was that the site of 6 Laguna Gatuna is suitable for the discharge of as much as 7 30,000 barrels brine per day. This was what the original application was for. 8. The fourth conclusion was that after fif-9 teen years of operation by Pollution Control there appears 10 to be no adverse impacts on the hydrologic system in that 11 area. 12 Our fourth is that the solid wastes which 13 have been disposed of at Laguna Gatuna have not in any way 14 been detrimental to the hydrologic system, and our final 15 conclusion was that the facility which is proposed in the 16 southwest corner of Section 17 would not adversely impact the hydrologic conditions, although we see no reason to in-17 crease the original allocation which was granted of 30,000 18 barrels per day combined from the two facilities. 19 0 All right, sir. Let's go back, then, Mr. 20 Kelly, and follow your report using the order that you have 21 placed them on the table of contents page, and have you 22 first of all discuss for us in a general way the geology of 23 the project area and focus in on the availability of any 24 fresh water aquifers in the area. Α The significant structural control, Nash 25

10 1 Draw to the west, which is a result of the solution of 2 brines from the Rustler formation and the top of the Salado 3 formation, which has resulted in the collapse of Nash Draw and, in my opinion, Laguna Plata and Laguna Gatuna and 5 Laguna Tonto are all extensions of Nash Draw. They simply 6 are not physiographically or topographically joined. 7 Q All right, let's go to page three of the 8 package of Exhibits and have you use that as a plat from 9 which you can reference the geology. Α All right. The site itself is at Laguna 10 Gatuna, which is shown in Township 20 South, Range 33 East, 11 and about seven miles east of the Lea/Eddy County line. 12 Nash Draw is formed along the west edge 13 of Lea County and -- but primarily in Eddy County, so that 14 it is just off the margin of the map to the left. 15 These sites, then, are just to the north-16 east of Nash Draw, and the WIPP site, where there's been a 17 considerable amount of drilling and testing performed. The beds, then, in this area dip to the 18 east beneath Eddy County and are controlled to a large ex-19 tent by the Delaware Basin. 20 To the north and east on the plat is а 21 line that says Mescalero Ridge. What is that? 22 Α Mescalero Ridge is the west and the . 23 southwest boundary of the Ogallala formation. That is 24 has a bearing on this particular project because the origi-25 nal ranchers in the vicinity of the salt lakes had a very

11 1 difficult time finding water for stock and domestic pur-2 poses. Most of the water was brackish. 3 When the potash mines and the refineries 4 for the potash industry went into Nash Draw, as a source of 5 water they piped water from the high plains or north of Mes-6 calero Ridge to the Nash Draw area. 7 The pipelines, as a trade off by the 8 ranchers, were then tapped by ranchers to provide water for their use, primarily in this salt lake area. 9 So that many of the wells which were ori-10 ginally shown on the Ed Reed map have fallen into disrepair 11 because of the better quality and more dependable supply 12 which is obtained from the pipeline. 13 So he was able to measure some water 14 levels but most of these wells are no longer in use simply 15 because the water quality is much poorer than is available. 16 All right, sir, we'll come back in a 0 17 minute to those wells that are still in use in the area, but let me have you go to page four of the Exhibit Number One 18 and have you give us the -- cite specific geologic features 19 at --20 А All right. 21 Q -- Laguna Gatuna. 22 Α Figure 2 on page 4 shows a cross section 23 of Laguna Gatuna. The lowermost formation are the Dewey 24 Lake Redbeds, which are shown by the horizontal lines. The Dockum Group forms the bedrock in that area beneath the lake 25

1	12
2	itself, and then there is a think veneer of alluvial and
3	playa deposits, both on the upper ridges and also in the
3	base of the playa itself.
4	There is an intermittent lake in the
5	playa and the fault zones indicated on both sides of this
6	lake, or playa, are in my opinion the avenues through which
7	ground water from the Rustler formation is moving upward and
8	being discharged as springs along the boundaries of the
9	playa itself.
10	Any discharge from Pollution Control fa-
11	cilities, which are diagrammatically shown on the left, come
12	down into the lake itself from the northwest corner and from
12	the left.
15	The new facility is illustrated by that
14	tank and would also empty into the playa itself.
15	The
16	Q As a hydrologist, do you see any adverse
17	consequences of significance to the fact that the point of
18	discharge for Pollution Control as at the higher ground
19	areas adjacent to the laguna itself, rather than down in the
20	laguna?
21	A The any water which is held up on the
22	boundaries is confined in surface impoundments and may, in
23	ract, enter to some extent into the very thin alluvium, but
24	contained in the boundaries of the place itself without the
24	detting out into the middle of the lake
25	geteing out into the midule of the lake.

13 1 Does it make any hydrologic difference 0 2 whether or not the discharge is up at the points you've de-3 picted on the schematic rather than down at the lake level? 4 No, it doesn't. Α 5 All right. Let's go back, then, Mr. Kel-0 6 ly, and look at the Reed plat and have you identify for us 7 any wells that Mr. Reed studied that continue to be used. To my knowledge none of the wells which 8 Α Reed evaluated are still in use. 9 There are two which we were able to 10 measure the water level in; however, they were not in a suf-11 ficient state of repair to actually pump a water sample from 12 them, so we were able to measure the water level but not the 13 -- but not collect a sample. 14 These two, one is located in the north-15 west corner of Section 25, which is southwest of Laguna Gatuna, and this shows a water level -- an elevation of 3555 16 and water level of 3516, or 38 feet, 38.6 feet below land 17 surface. When we measured that the water level was less 18 than a foot below the level that Reed measured, so the water 19 level, the natural water table in that particular well had 20 declined less than a foot in the fifteen years since Reed 21 did his work. 22 0 What significance do you make of that 23 fact? 24 That there has certainly been no effect А from water contributed to Laguna Gatuna and I would attri-25

14 1 bute it simply to a gradual decline in the water level with 2 time. 3 Conversely, if the water level had been 0 ·4 increased? 5 The water level should have risen; would А 6 have had to have come from some source, either much more 7 precipitation or some source such as water being emptied into Laguna Gatuna or some other source. 8 Is that well at a location hydrologically 0 9 where it would be down gradient from water disposed of in 10 Laguna Gatuna? 11 No, it's up gradient. It's about, well, А 12 let me see, the water level in that well is about 21 feet 13 higher than Laguna Gatuna but if the water in Laguna Gatuna 14 had risen significantly it should have affected the regional 15 ground water flow. There could have been some deline, but I 16 would not have expected much, so in fact both of these wells that we were able to remeasure have a higher water level 17 than the base of Laguna Gatuna. All of the rest of the 18 wells were in disrepair. 19 0 You made reference to Nash Draw and to 20 the potash operations. Is there a plat that shows the loca-21 tion of that area? 22 The illustration on page 25, Figure А 23 Three. 24 Well, let's make sure everybody's got Q that. 25

15 1 All right, sir, let's discuss this plat. 2 This shows in the very southeast corner А 3 of the map the topographic contours show a significant de-4 That is the northernmost edge of Nash Draw pression there. 5 and it shows the proximity of Nash Draw to Laguna Tulston 6 (sic), Laguna Plata, and Laguna Gatuna. 7 The rest of the draw is off to the left side; however, these water table contours show a regional 8 flow of ground water from the 3525 foot contour towards to 9 the west and northwest so that on the north and west side of 10 Laguna Plata the water table is as much as a hundred feet 11 below that to the east side of the project area. 12 All right, would you summarize for us 0 13 findings and conclusions with regards to the your ground 14 water movement? 15 A Yes, sir. We prepared this contour map 16 based on the data which Reed had generated which we were able to measure and water levels which have been produced 17 since the Reed study, and this shows a regional ground water 18 flow essentially from east to west with local variations 19 around Laguna Plata and also Nash Draw, where the 3425 foot 20 contour makes a large swing back to the southeast. 21 The reason that we did this was it shows 22 a more regional ground water flow, whereas Reed simply drew 23 arrows showing what he supposed to be directions of ground 24 water flow, but by working with a regional area we were able 25 to see the large pictures, whereas Reed was looking at very

16 1 minor changes in a small area and therefore I felt that the 2 regional pictures would supplement the work that Reed had 3 done. 4 All right, sir. Q 5 Α So that there is no conflict from what we 6 have done with what Reed did. We simply expanded his, as 7 shown in Figure 3. 8 All right, sir, let's go on and have you 0 summarize your findings with regards to the water quality 9 data. 10 А The water quality which Reed evaluated 11 indicates that the oilfield brine in the area is less highly 12 mineralized than the natural discharge in Laguna Gatuna and 13 Laguna Plata. 14 We have the information from Pollution 15 Control and the data which they provided us, and we found no 16 contradiction in this data. The conclusion being, then, that the highly mineralized water being discharged into La-17 guna Gatuna and Laguna Plata has to originate from some 18 deeper source, presumably either the Rustler or more logic-19 ally from the so-called Brine aquifer on top of the Salado 20 formation, and the regional gradients are such that it would 21 move up along joints and fault zones which would be asso-22 ciated with Laguna Plata and Laguna Gatuna. 23 Let's get sites specific now, Mr. Kelly, 0 24 and have you give us your opinion as to the suitability of 25 Laguna Gatuna, both in the northwest corner of the laguna

and the southeast corner of the laguna as sites for the disposal of produced salt water brines and other waste products.

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Our conclusion was that Laguna Gatuna А is 5 an excellent site for the purposes with which Pollution Con-6 trol is using it. The work by Reed was accurate. Due to 7 highway construction in that area there were more exposures of the Dockum Group, which is shown on page 28, Figure 4, as 8 This substantiated our conclusions that the alluvial TR. 9 material is extremely thin in that area and the amount of 10 brine which has been disposed of by Pollution Control in the 11 past fifteen years has never resulted in a permanent pool of 12 Laguna Gatuna. With its surface area of 383 acres it is 13 adequate to evaporate all of the brine which is being dis-14 posed of in the lake by Pollution Control.

15 Q Let's go to page 29 and have you describe
16 for us the evaporation studies that were conducted.

17 A We conducted some evaporation studies in 18 the Nash Draw area, which is just a few miles to the west, and we concluded that the evaporation rate, the summer evaporation rate, from a brine surface in that area was approximately 6.69 gallons per minute, or roughly 229 barrels of brine per acre per day.

22 On the other hand, the winter evaporation
23 loss was approximately 13 barrels of brine per acre per day.
24 With the minimum surface area of Laguna Gatuna, there is the
25 evaporation potential of 87,700 barrels per day during the

summer and about 5000 barrels per day during the winter. This is well within the annual disposal range of Pollution Control and clearly these evaporation calculations have shown that they are adequate to take care of the amount of brine being discharged by Pollution Control. Q Let's go now, Mr. Kelly, to pages 31 and 32, which are the discharge rates recently used at Laguna

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Gatuna.

10 A Right. This is information which I believe has been submitted to the Oil Conservation Division, but they simply show the monthly disposal rate for 1983 and 12 1984, both as a graph and then on page 32 in the cumulative 13 totals for the individual months.

And I might mention that the original application and grant was for 30,000 barrels per day, whereas
if you'll look at the monthly totals on Table 3, page 32, it
is considerably less than that, and I would assume that the
discharges at the present time, in fact I believe that Mr.
Foster told me that the highest discharge rate by Pollution
Control occurred in the early eighties but are not much less
than what you see here on -- on Table 3.

21 Q All right, sir, in addition to the conclusions that you've made on pages 31 and 33, I'd like to direct you back now to the Division Order of April of '69, and go through some of the findings that were made back in '69, and have you conclude for us whether you still concur

19 1 or disagree with any of those findings. 2 Let's start with -- do you have the or-3 der? 4 Yes, I do. You're starting on page one Α 5 of that order? 6 Yes, sir, on Finding No. 3 they make re-0 7 ference to the areawide Order R-3229, which prohibits the - 8 disposal of produced salt water brines in unlined pits. It 9 then goes on --All right, there is -- as near as А Right. 10 we have been able to determine there is no potable water in 11 By potable water I'm using the definition that this area. 12 the EID uses of 1000 parts per million. 13 That is also the State Engineer's defini-0 14 tion on --15 А Oh, yes. 16 -- Finding No. 4, page 2 of the order? Q Yes. 17 А 0 All right, sir. 18 Kelly said MR. STAMETS: Mr. 19 1000 and Finding 4 is 10,000. 20 That's -- okay. The difference А there, 21 one, the State Engineer uses 10,000 as a definition of fresh 22 water, whereas I'm using the definition of potable water, 23 that is water suitable for human consumption. 24 So your standard is even higher than the 0 State Engineer's standard --25

1	20
2	A Yes.
3	Q for water to be protected?
4	A Right.
-	Q All right, and using your higher
5	standard
6	A We can find no evidence that there is any
7	water in the area which could be considered potable, other
8	than at one time there was a well at what was then called
9	Midway. It was a bar and service station located on Reed's
10	map in the south half of Section 23, and approximately two
11	and a half miles southwest of Laguna Gatuna.
12	This shows a chloride of 362 parts per
	million. This water was potable, was used in the operation,
13	but the facility has been destroyed and the wells are aban-
14	doned.
15	Q All right.
16	Q That was the only fresh water we were
17	able to find, and this was verified by other ranchers in the
18	area who are still in operation, who haul water or take it
19	from the pipeline.
20	Q Let's go back to the State Engineer's
)1	standard of 10,000 parts per million.
27	Do you find any water in this area that
22	is of that quality or greater?
23	A There is a lot of water in the area
24	that's greater than 10,000.
25	Q I've got this backwards. I meant 10,000

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21 1 or less. 2 There is very little water in the area Α 3 that's 10,000 or less. 4 Most of the water, and certainly the 5 water from the springs, exceeds this -- this amount; water 6 which is naturally discharged into the lake itself. 7 For example, at Laguna Gatuna you can see 8 lake sample identified by Reed which had chlorides of а 9 158,000 parts per million and sulfates of 125,000. All right, let me make sure I'm clear. 0 10 Are there any waters in the area containing 10,000 parts per 11 million or less of total dissolved solids which have a pre-12 sent or reasonably foreseeable beneficial use that might be 13 impaired by the discharge of water in Laguna Gatuna as the 14 applicant proposes to do? 15 No, sir. А 16 All right. Let's go down to Finding No. 0 think you've concluded for us that this water is not 17 7. Ι fresh water in the lagunas? 18 That's correct. Α 19 All right, sir, and Finding No. 8? 0 20 The -- I conclude with this finding that Α 21 the underlying Redbeds are virtually impermeable and the --22 any seepage which would get into, or which would be -- any 23 water which would be impounded in the lakes would not seep 24 into the underlying formation. 25 All right, sir, and Finding No. 9? Q

22 1 А These -- the synclinal structure does 2 exist and that the flow of surface and subsurface water into 3 the boundaries is towards those lakes. 4 0 All right, sir, so you concur and believe 5 the Finding No. 9 is supported by substantial evidence? 6 А Yes, I do. 7 Q Let's go to No. 10. Α I also agree with this finding, that 8 there is no leakage from Laguna Plat and Laguna Gatuna, 9 simply because, first of all, the hydrologic gradient indi-10 cates that it toward the lakes rather than away, but also, 11 the evaporation surface at the bottom of each of these lakes 12 is great enough to evaporate any natural or artificially 13 discharged brine into those lakes. 14 0 Finding No. 11 is directed towards Laguna 15 Tonto, which is not the subject of our application here. Α That's correct. 16 Let's go to Finding No. 12 with regards 0 17 to utilization of Laguna Gatuna. Do you -- do you concur 18 with that finding? 19 Yes, I do. It does not constitute a ha-А 20 zard to fresh water supplies that may exist in the area. 21 I believe that most of these other find-22 ings pertain to the -- to Laguna Tonto to a large extent. 23 0 Yes, sir, I agree with you. I think that you have covered the essential findings in the prior order 24 that would apply to the current application. 25

23 1 In conclusion, then, Mr. Kelly, do you 2 believe the continued use of Laguna Gatuna as a disposal 3 site for as much as 30,000 barrels of brine per day is still 4 a suitable disposal site? 5 Yes, I do. Α 6 And do you see any adverse consequences 7 of changing or adding to the point of disposal by adding the southwest quarter of Section 17 to the disposal operation? 8 А No, sir. 9 Based upon your studies and knowledge of 0 10 the area, Mr. Kelly, do you see any adverse consequences of 11 the fifteen years, or so, operation by Pollution Control in 12 this Laguna Gatuna as a disposal facility? 13 No, we saw no evidence at all. A 14 And do you see any adverse consequences 0 15 hyrologically to the continued use of Laguna Gatuna as a re-16 pository for solid oilfield waste products --Α No, sir. 17 -- drilling cutting and drilling muds? Q 18 No, sir. In my opinion it's probably one Α 19 of the most suitable sites in the area. 20 Was Exhibit Number One prepared by you or 0 21 compiled under your direction and supervision? 22 Α Yes, it was. 23 Q All right, sir. 24 MR. KELLAHIN: Mr. Examiner, that concludes our examination of Mr. Kelly. 25 We have con-

24 1 cluded our examination of Mr. Kelly by discussion of oil 2 well solid waste products. That is paragraph 3 of our ap-3 plication. It is also specifically addressed in the current 4 -- now I've lost my place. 5 I'll admit I can't pick it out 6 real quickly, Mr. Stamets, but the application in this case 7 seeks to have a finding addressing the use of this disposal 8 facility for -- as a repository for these oilfield waste products, including the drill cuttings and drilling muds. 9 practical matter, this As а 10 site has been used for very many years for that purpose. 11 Mr. Kelly has demonstrated that he sees no adverse conse-12 quences from continuing that to occur and we would request 13 that a specific finding and approval for that part of the 14 operation be included in the order. 15 We move the introduction of Ex-16 hibit Number One. MR. STAMETS: Exhibit Number 17 One will be admitted. 18 19 CROSS EXAMINATION 20 BY MR. STAMETS: 21 Mr. Kelly, is it your opinion that with 0 22 the 30,000 barrels of water per day disposal limitation that 23 no water can move out of the area of Laguna Gatuna? 24 Yes, sir, it is. The summer evaporation А 25 rate would certainly more than cover that.

25 1 The winter evaporation rate would not, 2 but the hydrologic conditions are such that even if a pond-3 ing occurred during the winter, it would be evaporated dur-4 ing the summer. 5 So it is my opinion that that would be 6 the case. 7 Are there other MR. STAMETS: 8 quesitons of the witness? MR. BOYER: Yes. 9 QUESTIONS BY MR. DAVID BOYER: 10 My name is David Boyer. I'm a staff hy-Q 11 drogeologist with the Oil Conservation Division. I have a 12 few questions of Mr. Kelly. 13 Mr. Kelly, am I correct in understanding 14 you agreed with the finding of No. 11 on that 1969 order, 15 that the evidence indicates that there may be some leakage 16 of water into -- to the southeast and therefore southwestward toward Lagune Gatuna? Did I understand you correctly 17 on that? 18 Α Not in -- not in relation to Laguna 19 We did not discuss Laguna Tonto in the original Tonto. 20 findings. 21 Laguna Tonto was excluded from use by 22 Pollution Control. 23 Q So you did not -- you did not investigate 24 that particular --25 А No.

26 1 All right, I was -- getting -- thing. 0 2 back to Figure 3 on page 25, you showed the hydrologic con-3 tours and it would show a couple of things. First off, that this -- it is my under-5 that the water table contour map was prepared by standing 6 you for inclusion in this report. 7 That's correct. Α 8 0 Okay. It shows that, according to the contours, that you could have movement northwesterly out of 9 Laguna Gatuna towards the northwest if the hydrologic flow 10 lines are followed. 11 Is it a possibility also that you might 12 a closed contour around Laguna Gatuna that would move have 13 material into the laguna instead of to the northwest? 14 Yes, sir, there is. Ά 15 That was not investigated, though, 0 and 16 you don't have sufficient information? No, there's not sufficient information. 17 Α These are 25 foot contours and certainly with additional 18 drilling information we might be able to verify that, but I 19 might mention that the water quality in Laguna Plata is gen-20 erally worse than that in Laguna Gatuna, so I, if it did 21 move to the northwest, I would assume that Laguna Plata 22 would become the discharge point. 23 All right. On the -- on the map prepared 0 24 by Reed, you went back and determined that the well in the northwest one-quarter of Section 25 to the southwest of La-25

27 1 guna Gatuna was able to be measured, is that correct? 2 Α Yes. 3 Did you -- did you attempt to get a con-0 4 ductivity measurement to that well at all? 5 No, the well had a windmill on it but the Α 6 windmill was not operative, so we could not get a sample 7 from it with the sucker rods. There was not enough room to sample it. 8 Nor was there enough room to get a con-0 9 ductivity probe down -- down inside it at all, losing it or 10 possibly getting a conductivity measurement? 11 А Well, Mr. Boyer, our conductivity measure 12 has probe about six inches long and we wouldn't have а 13 reached the 16 feet, but I presume a downhole conductivity 14 meter could have been used, yes, sir. 15 And one additional question, the well 16 that is shown in the northwest one-quarter of Section 21, that shows that there was water that was probably greater 17 than 1000 pps, but certainly less than 10,000 in the Reed 18 map. 19 That was unavailable for any type of 20 measurement or water level or anything like that? 21 No, I thought that was the one I referred А 22 to as the other measurement. I could be wrong. Let me --23 If you will refer to page 23, the fourth 24 listing from the top in Table 2 identifies a windmill at location 20-33-21, 111, with a surface elevation of 25 3536.

28 1 That is the well in question here. 2 The water level on January 25th of 1984 3 was 35.42 feet below land surface. 4 When Reed measured the water level it was 5 36.6 feet. 6 And that well is also inoperable and you Q 7 were unable to get a water level -- I mean a water sample? 8 А Right. Right. MR. BOYER: That's the extent 9 of my questions. 10 MR. STAMETS: Are there any 11 other questions of this witness? He may be excused. 12 there anything further in Is 13 this case? 14 The case will be taken under 15 advisement. 16 (Hearing concluded.) 17 18 19 20 21 22 23 24 25

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CERTIFICATE I, SALLY W. BOYD, C.S.R., DO HEREBY CERTIFY that the foregoing Transcript of Hearing before the Oil Con-servation Division was reported by me; that the said tran-script is a full, true, and correct record of the hearing, prepared by me to the best of my ability. Soony W. Boyd CSR I do hereby certify that the form. a correcte e corport of the proc the Excusion hearing heard xaminer Oil Conservation Division



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	Sodium*(Nort)= (call)); Bicarbonate*((HCO);)); Casbonate*(CO);;-); Casbonate*(CO);-); Sulphate*(CO);-); Sulphate*(SO);-); Chioride*(C1-) Chioride*(C1-) Dissolved*Solids:on*Evap::at*103?=105?; C.	1/80, 70⊁ 1/6, 605 NO∓ NOT 54.13 186.12	411545 10172 FOUND FOUND 2600 6600
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	Sodiume(Nad) Bicarbonates(HCO);)+ Carbonates(CO); ;-) Carbonates(CO); ;-) Nydroxider(OH-) Sulphate (SO); -) Chloride (C1-)	1/800,70) 1/6,60 NOT 54.13 186.12 76.15 16.60	411545 10112 FOUND FOUND 2600 6600 3808 830
	Sodium*(Nad*)= (calib); Bicarbonate*(HCO);)) Carbonate*(CO:;); Carbonate*(CO:-); Chloride (C1-) Chloride (C1-) Dissolved: Solids:on: Evap: at: 103*: 105*: C. Hardness: as: Co. CO: Carbonate-Hardness: as: CaCO; (temporary); Nan-Carbonate-Hordness: as: CaCO; (temporary); Nan-Carbonate-Hordness: as: CaCO; (temporary);	180:.70) 16.60 NOT NOT 54.13 186.12 76.15 16.60 59.55	41154 1012 FOUND FOUND 2600 6600 3808 3808 830 2978
	Sodiume(Nad) Bicaibonatek(HEO):)+ Casbonatek(HEO):)+ Casbonatek(CO):;) Hydroxider(OH-) Sulphate (SO):) Chloride (C1-) Chloride (C1-) 6::99Pacci@ 68t Dissolved: Solids::on: Evap:: at: 103::: 105:: C. Hardness::as: Co. CO): Carbonate: Hardness::as: CaCO:; (temporary): Non-Carbonate: Hardness::as: CaCO:; (temporary): Carbonate: Hardness::as: CaCO:;	1/80:.70) 1/6:.60 NOT 54.13 186.12 	411545 1012 FOUND FOUND 2600 6600 3808 830 2978 830
	Sodiume(Nad) Bicarbonate(HCO);)+ Carbonate(CO); -) Carbonate(CO); -) Sulphate (SO); -) Chioride (C1-)	1/80:.70) 1/6:.60 NOT 54.13 186.12 	411545 1012 FOUND FOUND 2600 6600 6600 3808 830 2978 830
	Sodiums(Nad)) Bicarbonotes(HCO);)+ Carbonotes(CO); (-): Carbonotes(CO); (-): Sulphate. (SO); -): Chioride (CI -)	1/6, 60 NOT NOT 54.13 186.12 76.15 16.60 59.55 16.60 1.010 EXHIBIT ∷O. 7	41154 1012 FOUND FOUND 2600 6600 3808 830 2978 830 2978 830
	Sodiums(Nat-)- (calk)) Bicasbonates(#ICO3-)- Cabonates(#ICO3-)- Sulphate (SO)- Chioride (C1-)	1/6, 60 NOT NOT 54,13 186,12 76,15 76,60 59,55 16,60 1.010 EXHIBIT NO, 7 SALT LAZE PO	41154 1012 FOUND FOUND 2600 6600 3808 830 2978 830 2978 830

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	lease Bass #2	Samolina Date	9/19/68
	Type of Sompla <u>Wellhead</u>		
	WATER ANALYSIS	18-20:3	3
	IONIC: FORM	_me/1 *	mg/[*
	Calium (Ca++)	51.90	1040
	Mognesium (Mg++)	32.24	4072
	2000ma(INOBER)	177.10	4072
		W. Law L	2
	Bicarbonate (HCO.)	13.40	817
	Bitarbonote (HCO,); Carbonate (CO, ; -);	13.40 NOT	81.7 FOUND
	Bičarbonate (HCO,) Cárbonate (CO, -) Hydroxide (OH=)	13.40 NOT NOT	81.7 FOUND FOUND
	Bitarbonate (H2O ₃ ,); Cárbonate (CO, -) Hydroxide (OH=) Sulphate (SO, -)	13.40 NOT NOT 64.54	81.7 FOUND FOUND 3100
	Bičarbonote (HCO,) Corbonate (CO, -) Hydroxiděr (OH-) Sulphate (SO, -) Chloride (C1 -)	13.40 NOT NOT 64.54 183.30	817 FOUND FOUND 3100 6500
	Bitarbonate (H2O,); Cárbonate (CO, ; -) Hydroxide (OH-) Sulphate (SO, -) Chloride (C1-)	13.40 NOT NOT 64.54 183.30	81.7 FOUND FOUND 3100 6500
	Bičarbonote (HCO,) Corbonate (CO, -) Hydroxiděr (OH-) Sulphate (SO, -) Chloride (C1 -)	13.40 NOT NOT 64.54 183.30	81.7 FOUND FOUND 3100 6500
	Bisarbonote (HCO,) Cărbonate (CO, -) Hydroxide (OH=) Sulphate (SO, -) Chloride (C1 -)	13.40 NOT NOT 64.54 183.30	817 FOUND FOUND 3100 6500
	Bicarbonate (HCO,) Carbonate (CO, -) Carbonate (CO, -) Hydroxider(OH-) Sulphate (SO, -) Chloride (C1 -) Chloride (C1 -) Chloride (C1 -) Dissolved Solids: on Evap: at 103° - 105° C	13.40 NOT NOT 64.54 183.30	81.7 FOUND FOUND 3100 6500
	Bisarbonote (HCO,) Carbonate (CO, -) Hydroxide (OH-) Sulphate (SO, -) Chloride (C1 -) 6: 9: ph/c @.68: °F Dissolved Solids on Evap: at 103° - 105° C Hardness as Ca CO,	13.40 NOT NOT 64.54 183.30 84.14	81.7 FOUND FOUND 3100 6500 4207
	Bisarbonote (HCO,) Carbonate (CO, -) Hydroxide (OH-) Sulphate (SO, -) Chloride (C1-)	13.40 NOT NOT 64.54 183.30 84.14 13.40	81.7 FOUND FOUND 3100 6500 4207 670
	Bicarbonote (HCO,) Cárbonate (CO, -) Hydroxider (OH =) Sulphate (SO, -) Chloride (C1 -) Chloride (C1 -)	13.40 NOT NOT 64.54 183.30 84.14 13.40 70.74	81.7 FOUND FOUND 3100 6500 4207 670 3537
	Bičarbonate (HEO,) Cárbonate (CO, -) Hydroxide (OH-) Sulphate (SO, -) Chloride (C1 -) Chloride (C1 -)	13.40 NOT NOT 64.54 183.30 84.14 13.40 70.74 13.40	81.7 FOUND 3100 6500 4207 670 3537 670
	Bisarbonote (HCO,) Carbonate (CO, -) Hydroxide (OH=) Sulphate (SO, -) Chloride (C1-)	13.40 NOT NOT 64.54 183.30 84.14 13.40 70.74 13.40 1.010	81.7 FOUND 3100 6500 4207 670 3537 670
	Bičarbonote (HCO,) Cárbonate (CO) Hydroxiděr (OH =) Sulphate (SO, -) Chloride (C1 -)	13.40 NOT NOT 64.54 183.30 84.14 13.40 70.74 13.40 1.010	81.7 FOUND FOUND 3100 6500 4207 670 3537 670
	Bicarbonate (HCO,) Cárbonate (CO, -) Hydroxide (OH -) Sulphate (SO, -) Chloride (C1 -) Choride Solids: on Evap: at 103° - 105° C Hardness: as Ca CO, Carbonate Hardness: as CaCO, (lemporary) Non-Carbonate Hardness: as CaCO, (permanent) Alkalinity as CaCO, Specific Gravity c 68° F	13.40 NOT NOT 64.54 183.30 84.14 13.40 70.74 13.40 1.010 EXIIIBIT NO. 7B	81.7 FOUND 3100 6500 4207 670 3537 670
	Bičarbonate (HCO,) Cárbonate (CO, -) Hydroxide (OH-) Sulphate (SO, -) Chloride (C1 -) Chloride (C1 -)	13.40 NOT NOT 64.54 183.30 84.14 13.40 70.74 13.40 1.010 EXNIBIT NO. 7B SALT LAKE POOL	81.7 FOUND 3100 6500 4207 670 3537 670
	Bičarbonate (HCO,) Cárbonate (CO -, -) Hydroxide (OH =) Sulphate (SO, -) Chloride (CI -)	13.40 NOT NOT 64.54 183.30 84.14 13.40 70.74 13.40 1.010 EXHIBIT NO. 7E SALT LAXE POOL County, Hew Mes	81.7 FOUND 3100 6500 4207 670 3537 670
	Bičarbonate (HCO,) Cárbonate (CO ; -) Hydroxide (OH =) Sulphate (SO, -) Chloride (CI -)	13.40 NOT NOT 64.54 183.30 84.14 13.40 70.74 13.40 1.010 EXNIBIT NO. 7B SALT LAXE POOL County, New Mos OIL WELL	81.7 FOUND FOUND 3100 6500 4207 670 3537 670
	Biearbonote (MCO,) Cárbonote (CO, ; -) Hydroxide (OH-) Sulphote (SO, -) Chloride (C1-)	13.40 NOT NOT 64.54 183.30 84.14 13.40 70.74 13.40 1.010 EXNIBIT NO. 7B SALT LAXE POOL County, New Mes OIL WELL WATER ASLANCE	81.7 FOUND 3100 6500 4207 670 3537 670

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<i></i>	-		Bass #3		10/24/68
1		Leose	0035 0	Sompting U	ole
		Type of Sample	· · ·		
			WATER ANALYSIS		
		IONIC FORM		me/1 *	mg/1 •
	Calcium (Ca++)			65.77	1318
	Magnesium (Mg++)		· .	53.13	646
	Sodium (Na+)		(cal.)	728.10	16.739
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	میں جو بی اور بر میں اور	· · · · · · · · · · · · · · · · · · ·			
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	- Bicarbonate (HCO,)			10 57	1104
	Bicarbonate (HCO,) Carbonate (CO)			19.57 NOT	1194
	Bicarbonate (HCO ₃) Carbonate (CO 3 -) Hydroxide (OH -)			19.57 NOT	1194 FOUND
	Bicarbonate (HCO ₃) Carbonate (CO 3 -) Hydroxide (OH -) Sulphate (SO -)			19.57 NOT NOT	1194 FOUND
	Bicarbonate (HCO ₃) Carbonate (CO ₃ -) Hydroxide (OH-) Sulphate (SO) Chloride (C1-)			19.57 NOT NOT 115.80	1194FOUND556226_2025
	Bicarbonate (HCO ₃) Carbonate (CO ₃ -) Hydroxide (OH-) Sulphate (SO) Chloride (C1-)			19.57 NOT NOT 115.80 711.63	1194 FOUND FOUND 5562 25,235
	Bicarbonate (HCO ₃) Carbonate (CO ₃ -) Hydroxide (OH-) Sulphate (SO) Chloride (C1-)			19.57 NOT NOT 115.80 711.63	1194 FOUND FOUND 5562 25,235
	Bicarbonate (HCO ₃) Carbonate (CO ₃ -) Hydroxide (OH-) Sulphate (SO) Chloride (C1-)			19.57 NOT NOT 115.80 711.63	1194 FOUND FOUND 5562 25,235
	Bicarbonate (HCO ₃) Carbonate (CO ₃ -) Hydroxide (OH-) Sulphate (SO) Chloride (C1-)			19.57 NOT NOT 115.80 711.63	1194 FOUND FOUND 5562 25,235
	Bicarbonate (HCO ₃) Carbonate (CO 3 -) Hydroxide (OH -) Sulphate (SO) Chloride (C1 -) Chloride (C1 -)			19.57 NOT NOT 115.80 711.63	1194 FOUND FOUND 5562 25,235
	Bicarbonate (HCO ₃) Carbonate (CO 3 -) Hydroxide (OH -) Sulphate (SO) Chloride (C1 -) Chloride (C1 -)	03° - 105° C		19.57 NOT NOT 115.80 711.63	1194 FOUND FOUND 5562 25,235
	Bicarbonate (HCO ₃) Carbonate (CO ₃ -) Hydroxide (OH-) Sulphate (SO) Chloride (C1-) 6: 7* ph·c @ 68: ^o F Dissolved Solids on Evap. at 11 Hardness as Ca CO,	03° - 105° C		19.57 NOT NOT 115.80 711.63	1194 FOUND 5562 25,235 5945
	Bicarbonate (HCO ₃) Carbonate (CO ₃ -) Hydroxide (OH-) Sulphate (SO) Chloride (C1-) 6.7° ph·c @ 68.°F Dissolved Solids on Evap. at 11 Hardness as Ca CO ₃ Carbonate Hardness as CaCO ₃	03° - 105° C 5 (lemporary)		19.57 NOT NOT 115.80 711.63 118.90 19.57	1194 FOUND 5562 25,235 5945 979
	Bicarbonate (HCO ₃) Carbonate (CO ₃ -) Hydroxide (OH-) Sulphate (SO) Chloride (C1-) 6.7° ph c @ 68.°F Dissolved Solids on Evap. at 1 Hardness as Co CO ₃ Carbonale Hardness as CoCO ₃ Non-Carbonate Hardness as C	03° - 105° C 5 (lemporary) 2aCO3 (pormaneni)		19.57 NOT NOT 115.80 711.63 118.90 19.57 99.33	1194 FOUND FOUND 5562 25,235 5945 979 4967
	Bicarbonate (HCO ₃) Carbonate (CO ₃ -) Hydroxide (OH-) Sulphate (SO) Chloride (C1-)	03° - 105° C 5 (temporary) CaCO3 (permanent)		19.57 NOT NOT 115.80 711.63 118.90 19.57 99.33 19.57	1194 FOUND 5562 25,235 5945 979 4967 979
	Bicarbonate (HCO ₃) Carbonate (CO ₃ -) Hydroxide (OH-) Sulphate (SO) Chloride (C1-)	03° - 105° C 5 (temporary) CaCO3 (pormanent)		19.57 NOT NOT 115.80 711.63 118.90 19.57 99.33 19.57 1.030	1194 FOUND 5562 25,235 5945 979 4967 979
	Bicarbonate (HCO ₃) Carbonate (CO ₃ -) Hydroxide (OH-) Sulphate (SO) Chloride (C1-)	03° - 105° C 5 (temporary) CaCO ₃ (permanent)		19.57 NOT NOT 115.80 711.63 118.90 19.57 99.33 19.57 1.030	1194 FOUND 5562 25,235 5945 979 4967 979
	Bicarbonate (HCO ₃) Carbonate (CO ₃ -) Hydroxide (OH-) Sulphate (SO) Chloride (C1-)	03° - 105° C 5 (temporary) CaCO ₃ (permanent)		19.57 NOT NOT 115.80 711.63 118.90 19.57 99.33 19.57 1,030	1194 FOUND 5562 25,235 5945 979 4967 979
	Bicarbonate (HCO ₃) Carbonate (CO ₃ -) Hydroxide (OH-) Sulphate (SO) Chloride (C1-)	03° - 105° C (temporary) CaCO, (permanent) s per Liter		19.57 NOT NOT 115.80 711.63 118.90 19.57 99.33 19.57 1.030 EXHIBIT NO. 70	1194 FOUND 5562 25,235 5945 979 4967 979
	Bicarbonate (HCO ₃) Carbonate (CO ₃) Carbonate (CO ₃) Hydroxide (OH-) Sulphate (SO) Chloride (C1-) Chloride (C1-) Dissolved Solids on Evap. at 11 Hardness as Ca CO ₃ Carbonate Hardness as CaCO ₃ Non-Carbonate Hardness as CaCO ₃ Non-Carbonate Hardness as CaCO ₃ Specific Gravity c 68° F • mg/l= milligram • me/l = milligram	03° - 105° C 5 (temporary) CaCO, (permanent) ss per Liter raients per Liter		19.57 NOT NOT 115.80 711.63 118.90 19.57 99.33 19.57 1.030 EXHIBIT NO. 70 SAL1_LAKE POOL	1194 FOUND 5562 25,235 5945 979 4967 979
	Bicarbonate (HCO ₃) Carbonate (CO ₃) Carbonate (CO ₃) Hydroxide (OH-) Sulphate (SO) Chloride (C1-) Chloride (C1-) Dissolved Solids on Evap. at 11 Hardness as Ca CO ₃ Carbonate Hardness as CaCO ₃ Non-Carbonate Hardness as CaCO ₃ Non-Carbonate Hardness as CaCO ₃ Specific Gravity c 68° F • mg/l = milligram • me/l = milligram	03° - 105° C (temporary) CaCO, (permanent) as per Liter rolents per Liter		19.57 NOT NOT 115.80 711.63 118.90 19.57 99.33 19.57 1.030 EXHIBIT NO. 70 SALT LAKE POOL County, Rey Me	1194 FOUND 5562 25,235 5945 979 4967 979
	Bicarbonate (HCO ₃) Carbonate (CO ₃) Carbonate (CO ₃) Hydroxide (OH-) Sulphate (SO ₃ -) Chloride (C1-) Chloride (C1-) Dissolved Solids on Evap. at 1 Hardness as Ca CO ₃ Carbonate Hardness as CaCO ₃ Non-Carbonate Hardness as Ca Alkalinity as CaCO ₃ Specific Gravity c 68° F • mg/l = milligram • me/l = milligram	03° - 105° C 5 (temporary) CaCO, (permanent) ss per Liter rolents per Liter		19.57 NOT NOT 115.80 711.63 115.80 711.63 115.80 711.63 115.70 19.57 99.33 19.57 99.33 19.57 1.030 EXHIBIT NO. 70 SALI LAKE POOL BE County, Rev Me	1194 FOUND 5562 25,235 5945 979 4967 979
	Bicarbonate (HCO ₃) Carbonate (CO ₃) Carbonate (CO ₃) Hydroxide (OH-) Sulphate (SO) Chloride (C1-)	03° - 105° C 5 (temporary) CaCO, (permanent) ss per Liter rolents per Liter		19.57 NOT NOT 115.80 711.63 115.80 711.63 115.80 711.63 115.90 19.57 99.33 19.57 1.030 EXHIBIT NO. 70 SALT LAKE POOL BA County, Rev Me OIL WELL	1194 FOUND 5562 25,235 5945 979 4967 979

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ting and the second second		type of Sumple-	//////////////////////////////////////	· · · · · ·	
		۷ 	VATER ANALYSIS UNI	<u>to 1-20</u>	· - 3 <u>-</u> 3
		IONIC.FORM		me/l *	mg/t *
	Colcium (Ca++)				
				45.91	920
	Mognesium (Mg++)		/ 1. X:	<u>45.91</u> <u>32.24</u>	920 392
	Magnesium (Mg++) Sodium (Na+)		(cal.)	45.91 32.24 184.81	920 392 4249
	Magnesium (Mg++) Sodium (Na+)		(cal.)	45.91 32.24 184.81	920 392 4249
	Magnesium (Mg++) Sodium (Na+)		(cal.)	45.91 32.24 184.81	920 392 4249
	Bicarbonate (HCO,)		(cal.)	45.91 32.24 184.81	920 392 4249
	Bicarbonate (HCO ₂) Carbonate (CO ₂ -)		(cal.)	45.91 32.24 184.81 7.40 NOT	920 392 4249 451 FOUND
	Bicarbonate (HCO,) Carbonate (CO ,) Hydroxide (OH -)		(cal.)	45.91 32.24 184.81 7.40 NOT NOT	920 392 4249 451 FOUND FOUND
	Bicarbonate (HCO,) Carbonate (CO , -) Hydroxide (SO, -)		(cal.)	45.91 32.24 184.81 7.40 NOT NOT 66.62	920 392 4249 451 FOUND FOUND 3200
	Bicarbonate (HCO ₂) Garbonate (HCO ₂) Carbonate (CO ₃) Cylphate (SO ₄ -) Chloride (C1 -)		(cal.)	45.91 32.24 184.81 7.40 NOT NOT 66.62 188.94	920 392 4249 451 FOUND FOUND 3200 6700
	Bicarbonate (HCO,) Garbonate (HCO,) Carbonate (CO ,) Carbonate (CO , -) Hydroxide (OH -) Sulphate (SO, -) Chloride (C1 -)		(cal.)	45.91 32.24 184.81 7.40 NOT NOT 66.62 188.94	920 392 4249 4249 451 FOUND FOUND 3200 6700
	Magnesium (Mg++) Sodium (Na+) Bicarbanate (HCO,) Carbanate (CO , -) Hydroxide (OH -) Sulphate (SO, -) Chloride (C1 -)		(cal.)	45.91 32.24 184.81 7.40 NOT 66.62 188.94	920 392 4249 451 FOUND FOUND 3200 6700
	Magnesium (Mg++) Sodium (Na+) Bicarbonate (HCO,) Carbonate (CO 7 -) Hydroxide (OH -) Sulphate (SO -) Chloride (C1 -)		(cal.)	45.91 32.24 184.81 7.40 NOT NOT 66.62 188.94	920 392 4249 451 FOUND FOUND 3200 6700
	Magnesium (Mg ++) Sodium (Na +) Bicarbonate (HCO,) Carbonate (CO 7 -) Hydroxide (OH -) Sulphate (SO, -) Chloride (C1 -) 78ph c @ 68? °F		(cal)	45.91 32.24 184.81 7.40 NOT NOT 66.62 188.94	920 392 4249 451 FOUND FOUND 3200 6700
	Magnesium (Mg ++) Sodium (Na +) Bicorbonate (HCO,) Carbonate (CO 7 -) Hydroxide (OH -) Sulphate (SO, -) Chloride (C1 -) 78ph c @ 68% °F: Dissolved Solids on Evap. at 10)3°: 105° C	(cal.)	45.91 32.24 184.81 7.40 NOT NOT 66.62 188.94	920 392 4249 451 FOUND FOUND 3200 6700
	Magnesium (Mg++) Sodium (Na+) Bicarbonate (HCO ₂) Carbonate (CO 7 -) Hydroxide (OH -) Sulphate (SO -) Chloride (CI -) Z. Bph c @ 68? °F: Dissolved Solids on Evap. at 10 Hardness as Ca CO ₃)3°- 105° C	(cal.)	45.91 32.24 184.81 7.40 NOT 66.62 188.94 78.15	920 392 4249 451 FOUND FOUND 3200 6700 3200 6700
	Magnesium (Mg++) Sodium (Na+) Bicarbonate (HCO,) Carbonate (CO T -) Hydroxide (OH -) Sulphate (SO, -) Chloride (C1 -) Z. Sph c @ 68? °F: Dissolved Solids on Evap. at 10 Hardness as Ca CO, Carbonate Hardness as CaCO,)3°- 105° C (temporary)	(cal.)	45.91 32.24 184.81 7.40 NOT 66.62 188.94 78.15 7.40	920 392 4249 451 FOUND FOUND 3200 6700 3908 370
	Magnesium (Mg ++) Sodium (Na +) Bicarbonate (HCO,) Carbonate (CO 7 -) Hydroxide (OH -) Sulphate (SO, -) Chloride (C1 -) 7.8ph c @ 68? °F Dissolved Solids on Evap. at 10 Hordness as Ca CO, Carbonale Hardness as CaCO, Non-Carbonate Hardness as Ca)3°- 105° ⊂ (lemporary) (aCO3 (permaneni)	(cal)	45.91 32.24 184.81 7.40 NOT 66.62 188.94 78.15 78.15 7.40 70.75	920 392 4249 451 FOUND FOUND 3200 6700 3200 3700 370 3538
	Magnesium (Mg ++) Sodium (Na +) Bicarbonate (HCO,) Carbonate (CO 7 -) Hydroxide (OH -) Sulphate (SO, -) Chloride (C1 -) Z. 8ph c @ 68; °F: Dissolved Solids on Evap. at 10 Hardness as Ca CO, Carbonate Hardness as CaCO, Non-Carbonate Hardness as CaCO, Alkalinity as CaCO,)3°- 105° C (temporary) (aCO, (permanent)	(cal)	45.91 32.24 184.81 7.40 NOT NOT 66.62 188.94 78.15 78.15 7.40 70.75 7.40	920 392 4249 4249 451 FOUND FOUND 3200 6700 6700 3200 6700 3200 6700 3200 6700 3200 6700 3200 6700 3200 370 3538 370
	Aggnesium (Mg ++) Sodium (Na +) Bicarbonate (HCO,) Carbonate (CO 7 -) Hydroxide (OH -) Sulphate (SO, -) Chloride (C1 -) Dissolved Solids on Evap. at 10 Hardness as Ca CO, Carbonate Hardness as CaCO, Non-Carbonate Hardness as CaCO, Specific Grovity c 68° F)3°- 105° C (temporary) .aCO ₃ (permanent)	(cal.)	45.91 32.24 184.81 7.40 NOT 66.62 188.94 78.15 7.40 70.75 7.40 1.010	920 392 4249 451 FOUND FOUND 3200 6700 3700 3538 370 3538
	And an end of the second se)3°- 105° C (temporary) (aCO, (permanent)	(cal.)	45.91 32.24 184.81 7.40 NOT 66.62 188.94 78.15 7.40 70.75 7.40 1.010	920 392 4249 451 FOUND FOUND 3200 6700 3700 3700 3538 370

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	LeaseBroaks T-7-4	Sampling Da	ite9/19/68
	The Wallbard	•	
	Туреког затрів <u>тусті неча.</u>		<u> </u>
	WATER ANALYSIS K	· :1-20.3	<u>ک</u>
		me/l	mg/l •
and sharts always	Calcium: (Ca++)	36.93	740
	Magnesium (Mgs++)	27.63	336
	Scotum (Na.+)	169.67	3901
	e fan de ferste fan de ferste en ferste ferste ferste ferste skriver en de ferste ferste ferste generale. En ste		· · · · · · · · · · · · · · · · · · ·
			1
	Bicarbonate*(HCO3)	21.00	1 281
	Carbonate (COM-)		L FOUND
		NOI	FOUND
	Hydroxide (OH -)	NOT	FOUND
	Hydroxide (OH -) Sulphote (SO , -)	NOT NOT 46.85	FOUND FOUND 2250
	Hydroxide (OH-) Sulphote (SO) Chloride (C))	NOT NOT 46.85 166.38	FOUND FOUND 2250 5700
	Hydroxide (OH -) Sulphote (SO, -) Chloride (C) -)	NOI NOT 46.85 166.38	FOUND FOUND 2250 5900
	Hydroxide (OH -) Sulphote (SO, -) Chlorido (C) -)	NOI NOT 46.85 166.38	FOUND FOUND 2250 5700
	Hydroxide (OH -) Sulphote (SO, -) Chlorido (C) -)	NOI NOT 46.85 166.38	FOUND FOUND 2250 5700
	Hydroxide (OH -) Sulphote (SO , -) Chlorido. (C) -) 65.6624 c@r688	NOI NOT 46.85 166.38	FOUND FOUND 2250 5900
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	Hydroxide (OH -) Sulphate (SO , -) Chloride (C) -) 6:.6: • • @*68: * 5 Dissolved Solids: on Evap. at 103* - 105* C Hordness: as: Co CO;	NOI NOT 46.85 166.38	FOUND FOUND 2250 5700
	Hydroxide (OH -) Sulphote (SO, -) Chloride (C) -) 6:.60 - 00 - 00 - 00 - 00 - 00 - 00 - 00 -	NOI NOT 46.85 166.38 64.56 21.00	FOUND FOUND 2250 5700 3228 1050
	Hydroxide (OH -) Sulphote (SO, -) Chloride (C) -) Significative C@1685 Dissolved Solids: on Evap. at 103° - 105° C Hardness: as: Ca CO, Carbonate Hordness: as CaCO; (temporary) Nan-Carbonate Hardness: as CaCO; (permanent)	NOI NOT 46.85 166.38 64.56 21.00 43.56	FOUND FOUND 2250 5700 3228 1050 2178
	Hydroxide (OH -) Sulphote (SO, -) Chloride (C) -) 6:.6cth+c@r68:** Dissolved Solids: on Evap: at 103** 105* C Hardness: as Ca CO, Carbonate Hardness: as CaCO; (temporary) Non-Carbonate Hardness: as CaCO; (permanent) Alkalinity as CaCO;	NOI NOT 46.85 166.38 64.56 21.00 43.56 21.00	FOUND FOUND 2250 5700 3228 1050 2178 1050
	Hydroxide (OH -) Sulphote (SO , -) Chloride (C1 -)	NOI NOT 46.85 166.38 64.56 21.00 43.56 21.00 1.005	FOUND FOUND 2250 5700 3228 1050 2178 1050
	Hydroxide (OH -) Sulphote (SO, -) Chloride (C) -)	NOI NOT 46.85 166.38 64.56 21.00 43.56 21.00 1.005	FOUND FOUND 2250 5900 3228 1050 2178 1050
	Hydroxide (OH -) Sulphate (SO , -) Chloride (C) -)	NOI NOT 46.85 166.38 64.56 21.00 43.56 21.00 1.005	FOUND FOUND 2250 5900 3228 1050 2178 1050
	Hydroxide (OH -) Sulphate (SO, -) Chloride. (C) -)	NOI NOT 46.85 166.38 64.56 21.00 43.56 21.00 1.005	FOUND FOUND 2250 5700 3228 1050 2178 1050
	Hydroxide (OH -) Sulphate (SO, -) Chloride. (C) -) Chloride. (C) -) Signature (@f68 ^{2, * p}) Dissolved Solids: on Evap: at 103 [*] - 105 ⁶ C Hardness: as Ca CO, Carbonate Hordness: as CaCO ₃ (temporary) Non-Carbonate Hardness: as CaCO ₃ (temporary) Non-Carb	NOI NOT 46.85 166.38 64.56 21.00 43.56 21.00 1.005 EXHIBIT 1:0. 7E	FOUND FOUND 2250 5700 3228 1050 2178 1050
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	Hydroxide (OH -) Sulphote (SO, -) Chloride. (C1 -) 6:	NOI NOT 46.85 166.38 64.56 21.00 43.56 21.00 1.005 EXHIEIT NO. 7E SALT LACE POOL County, New Me	FOUND FOUND 2250 5700 3228 1050 2178 1050 2178
	Hydroxide (OH -) Sulphote (SO, -) Chloride. (C) -) Signological Solids: on-Evap: at 103° - 105° C Hardness: as Ca CO, Carbonate: Hardness: as CaCO; (temporary) Nan-Carbonate: Hardness: as CaCO; (permanent) Alkalinity as CaCO, Specific Gravity c 68° F * mg/l = milligrams per Liter * me/l = milliequivalents per Liter Lee	NOI NOT 46.85 166.38 64.56 21.00 43.56 21.00 1.005 EXHIBIT NO. 7E SALT LACE POOL County, New Ma OTL WILL	FOUND FOUND 2250 5700 3228 1050 2178 1050 2178

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	Field	<u> </u>	
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	Leose <u> </u>	Sampling Date	9/19/08
	Type of SampleWellhead		
and the second second second second second second second second second second second second second second second			
	WATER ANALYSIS MUT	N 1. Z . S	
	IONICFORM	me//	mg/l-•
	Calcium (Ca++)	34:93	700
	Mognesium: (Mg++)	27.63	336
	Sodium (No+)	1:63.29	3754
<u>,</u>		and a state of the	
			and a second second second second second second second second second second second second second second second
	Bicarbonate (HCO,)	26.59	1622
i a tig	Carbonate (COT, -)	NOT	FOUND
	Hydroxide (OH -)	NOT	FOUND
č – t	Sulphate (SO)	38.52	1.850
	Chloride (C1-)	160.74	5700
l.			and the factor of the
			te a tradition de la constante de la constante de la constante de la constante de la constante de la constante La constante de la constante de
	6, 7 ^{, 19, 10} @'63 ^{EC.}		
	Dissolved Solids on Evopt at 103 ^{2*} 105 ^g C	10000000000000000000000000000000000000	
	Hordness as Ca CO	62.56	3128
é i	Carbonote Hardness as CaCO, (lemporary)	26.56	1330
6 <u>.</u>	Non-Carbonate Hardness as, CaCO's (permanent);	36.00	1800.
	Alkolinity os CaCO,	26.56	». 1330
1	Specific Gravity c 68° F	1.005	
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an an the second s	• mg/l= milligrams per Liter	EXHIBIN NO. 7F	. * .
t t	• me/l = milliequivalents: per Liter	SITE TIXE DOOL	
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		- Sources - Jook PROXI	
1		<u>OIL WELL</u>	·
		MATER ANALYSIS	
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		TELEPHONE: HO	BBS 393-6215
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	UNITED CHEMICAL	CORPOR	CATION
NE-JUNE		P. O. B	SX 1499
	HOBBS, NEW ME	XICO 88240	
Company.	Minerals, Incorporated	· · · · · · · · · · · · · · · · · · ·	
Pield		$(X_{i})^{-1}$	0/10/48
Leas	Salt Lake	Sampling Date	
Type of S	ample		
		718 20	~~
	WATER ANALYSIS	<u>1/10 / U-</u>	ma/list
IONIC FOI		0.00	200
		1667.17	20, 272
Sodium (Na+)	(cal.)	5395.75	124,048
			1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1
Bicerbonste (HCO.)		1'4-80'	-902
Dical Bondre (nico))			
Corbonate (CO 3 -)		NOT	FOUND
Corbonde (CO 3 -) Hydroxide (QH-)		NOT	FOUND FOUND
Carbonate (CO 3 -) Hydroxide (OH -) Sulphate (SO 4 -)		NOT 2602.50: 4455.60	FOUND FOUND 125,000
Carbonate (CO 3 -) Hydroxide (QH-) Sulphate (SO -) Chloride (C1-)		NOT NOT 2602,50: 4455,60	FOUND: FOUND: 125,000 158,000
Corbonde (CO 3 -) Hydroxide (OH-) Sulphote (SO 1 -) Chloride (C1-)		NOT NOT 2602.50 4455.60	FOUND FOUND 125,000 158,000
Corbonate (CO 3 -) Hydroxide (OH-) Sulphote (SO 4 -). Chloride (C1-)		NOT NOT 2602.50: 4455.60	FOUND FOUND 125,000 158,000
Carbonate (CO 3 -) Hydroxide (OH-) Sulphate (SO 4 -) Chloride (C1-) 7.7 ph c@ 68: °F		NOT NOT 2602.50: 4455.60	FOUND: FOUND: 125,000 158,000
Carbonate (CO 3 -) Hydroxide (QH-) Sulphate (SO -) Chlaride (C1-) Chlaride (C1-) 7.7 ph c @ 68: °F' Dissolieed Solids; on Evap. at: 103°- 105° C'		NOT NOT 2602.50: 4455.60	FOUND: FOUND: 125,000 158,000
Carbonate (CO 3 -) Hydroxide (OH -) Sulphate (SO 4 -) Chloride (C1 -) Chloride (C1 -) 7.7 ph c @ 68: °F' Dissolived Solids; on Evap. at 103°- 105° C Hardness as Co CO3		NOT 2602.50 4455.60 1677.15	FOUND FOUND 125,000 158,000 83,858
Carbonate (CO ₃ -) Hydroxide (OH-) Sulphate (SO ₄ -). Chloride (C1-) 7.7 ph c @ 68: °F Dissolved Solids; an Evap. at 103°- 105° C Hardness as Ca CO ₃ Carbonate Hardness as CaCO ₃ (temporary)		NOT 2602.50: 4455.60 1677.15 14.80	FOUND: FOUND: 125,000 158,000 83,858 83,858 740
Carbonate (CO 3 -) Hydroxide (QH-) Sulphate (SO -) Chloride (C1-) Chloride (C1-) 7.7 ph c @ 68: °F Dissolwed Solids; on Evap. at 103°- 105° C Hardness as Ca CO3 Carbonate Hardness as CaCO3 (temporary) Non-Carbonate Hardness as CaCO3 (temporary)	n1)	NOT 2602.50: 4455.60 1677.15 14.80 1662.35 14.80	FOUND: FOUND: 125,000 158,000 158,000 83,858 83,858 740 83,118
7. 7 ph c @ 68 °F Dissolved (CO ₃ -) Chloride (CI-) Chloride (CI-) 7. 7 ph c @ 68 °F Dissolved Solids; on Evap. at 103° - 105° C Hardness as Ca CO ₃ Carbonate Hardness as CaCO ₃ (temporary) Non-Carbonate Hardness as CaCO ₃ (temporary) Non-Carbonate Hardness as CaCO ₃ (temporary)	nt)	NOT 2602.50 4455.60 1677.15 14.80 1662.35 14.80 1.250+	FOUND: FOUND: 125,000 158,000 158,000 83,858 740 83,118 740
Carbonate (CO 3 -) Hydroxide (OH-) Sulphate (SO 4 -) Chloride (C1-) 7. 7 ph c @ 68: °F Dissolved Solids; an Evap. at 103°- 105° C Hardness as Ca CO ₃ Carbonate Hardness as CaCO ₃ (temporary) Non-Carbonate Hardness as CaCO ₃ (temporary) Non-Carbonate Hardness as CaCO ₃ (termana Alkolinity as CaCO ₃	nt)	NOT 2602.50 4455.60 1677.15 14.80 1662.35 14.80 1.250+	FOUND: FOUND: 125,000 158,000 83,858 83,858 740 83,118 740
Carbonate (CO 3 -) Hydroxide (QH-) Sulphate (SO 4 -) Chloride (C1-) 7.7 ph c @ 68: °F Dissolved Solidsron-Evap. at 103°- 105° C Hordness as Co CO 3 Carbonate Hardness as CaCO3 (temporary) Non-Carbonate Hardness as CaCO3 (temporary)	nt)	NOT 2602.50: 4455.60 1677.15 14.80 162.35 14.80 1.250+ EXILIBIT 1.0. B	FOUND FOUND 125,000 158,000 83,858 83,858 740 83,118 740
Carbonate (CO 3 -) Hydroxide (OH-) Sulphate (SO 4 -) Chloride (C1-) 7.7 ph c @ 68: °F Dissolved Solids; an Evap. at 103°- 105° C Hardness as Ca CO 3 Carbonate Hardness as CaCO (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hard	nt)	NOT 2602.50 4455.60 1677.15 14.80 1662.35 14.80 1.662.35 14.80 1.250+ EXHIBIT NO. 81 3AL1 LAKE POO	FOUND: FOUND: 125,000 158,000 158,000 83,858 740 83,118 740 83,118
Carbonate (CO 3 -) Hydroxide (QH-) Sulphate (SO -) Chloride (CI -) Chloride (CI -) 7.7 ph c @ 68: °F Dissolved Solidsron Evap. at 103°- 105° C Hardness as Co CO, Carbonate Hardness as CaCO; (temporary) Non-Carbonate Hardness as CaCO; (temporary) Sole (temporary) Sole (temporary) Sole (temporary) Hardness (temporary) Sole (temporary) Hardness (temporary) Hardness (temporary) Hardness (temporary) Hardness (temporary) Hardness (temporary) Hardness (temporary) Hardness (temporary) Hardness (temporary) Hardness (temporary) Hardness (temporary) Hardness (temporary) Hardness (temporary) Hardness (temporary) Hardness (tempo	nt)	NOT NOT 2602.50: 4455.60 1677.15 14.80 1662.35 14.80 1.250+ EXHIBIT NO. BI SALT LAKE POO SALT LAKE POO SALT LAKE POO SALT LAKE POO	FOUND FOUND 125,000 158,000 83,858 740 83,118 740 83,118 740
Carbonate (CO 3 -) Hydroxide (QH-) Sulphate (SO 4 -) Chloride (C1-) 7.7 ph c @ 68 °F Dissolved Solids; on Evap. at 103°- 105° C Hardness as Ca CO 3 Carbonate Hardness as CaCO (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hardness (temporary) Non-Carbonate Hard	nı)	NOT 2602.50 4455.60 1677.15 14.80 1662.35 14.80 1.250+ EXHIBIT 10.81 3AL1_LAKE POO To County, Heat SUPPACE LAKE	FOUND FOUND 125,000 158,000 158,000 83,858 740 83,118 740 83,118

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	UNITED CHE	MICAL CORPO	PRATION
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	CompanyMinerals	Incorporated	·
	Salt Isla	. Etald	
	FieldSalf Lake	<u>Field</u>	
	LeaseSalt Sprin	1 gs Sampling D	lote 10/24/68
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	type or Sample	······································	· · · ·
	WATER ANALYSIS	SIJ NE 19-20	・ ジェー
	IONIC FORM	me/l	mg/l • · ·
Calcium (Ca++1)		27.59	553
Magnesium (Mg;++)		586.62	7133
Södiums(Nost)	(cal.)	2389, 29	54,930
Bicdibonales(HGOA)		4.80	292
Carbonater(COL)		7.60	228
₩₩ ₩₩		NOT	FOUND
Sulphale#(SO#)		1532_14	73,590
Chloride: (CI)		1458.96	51,736
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8.4, ph c @ 68	i de la companya de la companya de la companya de la companya de la companya de la companya de la companya de Esta de la companya d		
Dissolved-Solids opsEvap, at 1	03%- 105° C		
Hardhess as Gor CO		614.21	
Carbonate Hardness as CaCO	5.(Itimporary)	12.40	620
Non-Carbongte Hardness: ast	CaCO; (permanent)	100	30,091
Albalinity as CoCO3	<u> </u>	12.40	620
Specific Gravity c 68° F		1.115	
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	Lease	Wail #3_ Wells	Sampling	Date 10/24/68
	Lease	Wail #3Walls	Sampling	Dote_10/24/68_
	Lease Type of Sample	Wall #3Wells	Sompling	Dote_10/24/68. Nell #24
	Lease Type of Sample	Wolf #3 Wells	Sampling <u>Water b</u> 4-20-33	Doie_10/24/68. <u>Ve// [#]24</u>
	Lease Type of Sample IONIC FORM	Wolf #3 Wells	Sampling <u>Nater 1</u> <u>4-20-32</u> mc/1*	Dore10/24/68. Nell #24
Colcium (Co++;)	Lease Type of Sample IONIC FORM	Wolf #3 Wells	Sampling Water 1 4-20-32 mc/1 21.61	Dore10/24/68_ <u>well #24</u> <u>mg/l</u>
Colcium (Ca++) Magnesium (Mg++)	Lease Type of Sample IONIC FORM	Wolf #3 Wells	Sampling <u>Water</u> <u>Yater</u> <u>Mater</u>	Date_10/24/68_ <u>Nell #24</u> ing/1 433 1488
Colcium (Ca++) Magnesium (Mg++) Sodium (Na+)	Lease Type of Sample IONIC FORM	Wolf #3 Wells water analysis //// SE (col.)	Sampling <u>Water</u> <u>4->0-30</u> me/l <u>21.61</u> 122.37 <u>561.94</u>	Dote_10/24/68. <u>Nell ≠24</u> <u>ing/1</u> 433 1488 12,919
Colcium (Ca++;) Magnesium (Mg++;) Sodium (Na+;	Lease Type of Sample IONIC FORM	Wolf #3 Wells water analysis //// 55 (cal.)	Sampling <u>Water</u> <u>Y->0-35</u> <u>mc/1</u> <u>21.61</u> 122.37 <u>561.94</u>	Dore10/24/68. //e// #24 mg/l 433 1488 12,919
Calcium (Ca++) Magnesium (Mg++) Sodium (Na+)	Lease Type of Sample IONIC.FORM	Wolf #3 Wells water analysis //// 55 (col.)	Sampling <u>Water</u> <u>4-20-32</u> me/l <u>21.61</u> 122.37 <u>561.94</u>	Date_10/24/68_ <u>AVe// #24</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u> <u>attack</u>
Colcium (Ca++.) Magnesium (Mg++.) Sodium (Na+) Bicarbonate (HCO3.)	Lease Type of Sample IONIC FORM	Wolf #3 Wells water analysis //// 55 (col.)	Sampling Water 1 4->0-35 mc/1 21.61 122.37 561.94 4.60	Dore10/24/68. //e// #24 mg/l 433 1488 12,919 286
Colcium (Ca++) Magnesium (Mg++) Sodium (Na+) Bicarbonate (HCO ₃) Carbonate (CO ₋₃ -)	Lease Type of Sample IONIC FORM	<u>Wolf #3</u> <u>Wells</u> water analysis <i>//// SE</i> (col.)	Sampling <u>Water 1</u> <u>4->0-35</u> <u>me/1</u> <u>21.61</u> 122.37 <u>561.94</u> <u>4.60</u> 0.80	Date_10/24/68. $\sqrt{e/1} \neq 24$ 433 1488 12,919 286 24
Colcium (Ca++) Magnesium (Mg++) Sadium (Na+) Bicarbonate (HCO ₃) Carbonate (CO ₋₃ -) Hydroxide (OH-)	Lease Type of Sample IONIC.FORM	Wolf #3_Wells WATER ANALYSIS //// 55 (cal.)	Sampling <u>Water</u> <u>Y->0-35</u> <u>me/l</u> <u>21.61</u> 122.37 <u>561.94</u> <u>4.60</u> 0.80 NOT	Dore10/24/68. <u>Nell ≠24</u> <u>mg/1</u> 433 1488 <u>12,919</u> 286. 24 FOUND
Calcium (Ca++) Magnesium (Mg++) Sodium (Na+) Bicarbonate (HCO ₃) Carbonate (CO ₋₃ -) Hydroxide (OH-) Sulphate (SO ₄ -)	Lease Type of Sample IONIC FORM	Wolf #3_Wells WATER ANALYSIS //// 55 (cal.)	Sampling <u>Water</u> <u>Y->0-32</u> <u>mc/1</u> <u>21.61</u> <u>122.37</u> <u>561.94</u> <u>4.60</u> 0.80 <u>NOT</u> <u>334.54</u>	Dore10/24/68. <u>well #24</u> <u>mg/1 *</u> 433 1488 12,919 286 24 FOUND 16,068
Colcium (Ca++.) Magnesium (Mg++.) Sodium (Na+) Bicarbonate (HCO ₃ .) Carbonate (CO ₋₃) Hydroxide (OH -) Sulphate (SO ₄) Chloride (C1-)	Lease Type of Sample IONIC FORM	Wolf #3 Wells water analysis //// 55 (col.)	Sompling <u>Water</u> <u>Y->0-35</u> <u>mc/1</u> <u>21.61</u> <u>122.37</u> <u>561.94</u> <u>4.60</u> 0.80 <u>NOT</u> <u>334.54</u> <u>365.98</u>	Dore10/24/68. //e// #/24 433 1488 12,919 286. 24 FOUNB 16,068 12,978
Calcium (Ca++) Magnesium (Mg++) Sodium (Na+) Bicarbonate (HCO ₃) Carbonate (CO -3) -) Hydraxide (OH -) Sulphate (SO) Chloride (C1 -)	Lease Type of Sample IONIC FORM	Wolf #3 Wells WATER ANALYSIS //// 55 (cal.)	Sampling <u>Water 1</u> <u>4->0-32</u> <u>me/1</u> <u>21.61</u> <u>122.37</u> <u>561.94</u> <u>4.60</u> 0.80 <u>NOT</u> <u>334.54</u> <u>365.98</u>	Dore_10/24/68. <u>Nell ≠ 24</u> <u>ing/1</u> 433 1468 12,919 286 24 FOUND 16,068 12,978
Colcium (Ca++) Magnesium (Mg++) Sodium (Na+) Bicarbonate (HCO,) Carbonate (CO,) Carbonate (CO, -) Hydroxide (OH -) Sulphate (SO, -) Chloride (C1 -)	Lease Type of Sample IONIC FORM	<u>Woif #3</u> <u>Wclls</u> water analysis <i>//// SE</i> (col.)	Sampling <u>Water</u> <u>4->0-35</u> <u>me/l</u> <u>21.61</u> 122.37 <u>561.94</u> <u>4.60</u> 0.80 NOT <u>334.54</u> <u>365.98</u>	Dore10/24/68_ <u>Nell ≠24</u> 433 1488 12,919 286: 24 FOUNB 16,068 12,978
Calcium (Ca++) Magnesium (Mg++) Sodium (Na+) Bicarbonate (HCO ₃) Carbonate (CO -3 -) Hydroxide (OH -) Sulphate (SO) Chloride (C1 -) Chloride (C1 -) Chloride (C1 -)	Lease Type of Sample IONIC.FORM	<u>Woi</u> <u></u> Woi <u></u> <u></u> Water Analysis,///, <i>SE</i> (cal.)	Sampling <u>Water 1</u> <u>4->0-35</u> <u>me/1</u> <u>21.61</u> <u>122.37</u> <u>561.94</u> <u>4.60</u> 0.80 <u>NOT</u> <u>334.54</u> <u>365.98</u>	Dore_10/24/68. <u>Nell ≠24</u> <u>ing/1</u> 433 1488 12,919 286 24 FOUND 16,068 12,978
Colcium (Ca++) Magnesium (Mg++) Sodium (Na+) Bicarbonate (HCO ₃) Carbonate (CO -3 -) Hydroxide (OH -) Sulphate (SO) Chloride (C1 -) Chloride (C1 -) Chloride (C1 -) Chloride (C1 -)	Lease Type of Sample IONIC FORM	<u>Woil #3</u> <u>Wells</u> water analysis//// <i>SE</i> (cal.)	Sampling <u>Water</u> <u>Y->0-35</u> mc/l 21.61 122.37 561.94 4.60 0.80 NOT 334.54 365.98	Dore_10/24/68. <u>ve// #24</u> <u>ing/l</u> 433 1488 12,919 286 24 FOUNB 16,068 12,978
Calcium (Ca++) Magnesium (Mg++) Sodium (Na+) Bicarbonate (HCO ₃) Carbonate (CO ₋₃ -) Hydroxide (OH-) Sulphate (SO ₄ -) Chloride (C1-) Chloride (C1-)	Lease Type of Sample IONIC FORM	<u>Woif #3</u> <u>Wells</u> water analysis//// <i>SE</i> (col.)	Sompling <u>Water</u> <u>Y->0-35</u> <u>mc/1</u> <u>21.61</u> <u>122.37</u> <u>561.94</u> <u>4.60</u> 0.80 <u>NOT</u> <u>334.54</u> <u>365.98</u> <u>143.98</u>	Dore10/24/68. //e// #24 mg/l 433 1488 12,919 286. 24 FOUNB 16,068 12,978 7199
Colcium (Ca++) Magnesium (Mg++) Sodium (Na+) Bicarbonate (HCO,) Carbonate (CO,) Carbonate (CO, -) Hydroxide (OH -) Sulphate (SO, -) Chloride (C1 -) Chloride (C1 -) E,3:ph.c.@:68: °F Dissolved Solids on Evop. at 10 Hardness as Ca.CO, Carbonate Hardness as CaCO,	Lease Type of Sample IONIC FORM 000000000000000000000000000000000000	<u>Woif #3</u> <u>Wells</u> water analysis <i>//// SE</i> (col.)	Sampling Water 1 4->0.35 mc/1 .21.61 122.37 .561.94 4.60 0.80 NOT .334.54 .365.98 	Dore10/24/68. //e// #24 1433 1488 12,919 286. 24 FOUND 16,068 12,978 7199 270
Colcium (Ca++) Magnesium (Mg++) Sodium (Na+) Bicarbonate (HCO,) Carbonate (CO, -) Carbonate (CO, -) Hydraxide (OH -) Sulphate (SO, -) Chloride (C1 -)	Lease Type of Sample IONIC FORM 03°- 105° C s (temporary) CaCO, (permanent)	<u>Woil #3</u> <u>Wclls</u> water analysis (col.)	Sompling <u>Water 1</u> <u>4->0-35</u> <u>me/1</u> <u>21.61</u> <u>122.37</u> <u>561.94</u> <u>4.60</u> 0.80 <u>NOT</u> <u>334.54</u> <u>365.98</u> <u>143.98</u> <u>5.40</u> <u>138.58</u>	Dore_10/24/68. <u>Nell</u> ≠ 24 mg/l 433 1488 12,919 286 24 FOUND 16,068 12,978 7199 270 6929
Colcium (Ca++) Magnesium (Mg++) Sodium (Na+) Bicarbonate (HCO ₃) Carbonate (CO ₋₃ -) Hydroxide (OH-) Sulphate (SO ₄ -) Chloride (C1-) Chloride (C1-)	Lease Type of Sample IONIC.FORM 0010 000 00	<u>Woil #3</u> <u>Wells</u> water analysis//// <i>SE</i> (cal.)	Sampling 4 - 20 - 32 mc/l 21.61 122.37 561.94 4.60 0.80 NOT 334.54 365.98 143.98 5.40 138.58 5.40	Dore10/24/68. <u>Nell</u> <u><u><u></u><u></u><u></u><u></u><u></u><u></u><u></u><u></u><u></u><u></u><u></u><u></u><u></u><u></u><u></u><u></u><u></u></u></u>

• me/l = millicquivalents per Liter

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<u>SALT LAKE POOL</u> Lea County, New Mexico

> <u>THREE WILLS</u> MATLE ALALISIS

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	TELEPHONE:	HOBBS 393-6215
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UNITED CHEMICA	L CORPO	DRATION
(KE-TONE) OF NEW	MEXICO	
601 NORTH LEECH	P. O.	BOX 1499
HOBBS, NEW A	MEXICO 88240	
CompanyMinerals, Incorporated		
FieldSalt Lake	· .	· · ·
Bass #1 w/a far 11/2/		11/4/68
Leose Dass #1 V/c-1 CF V/C/1	Sampling D	olo1/ 4/ 00/
Type of Sample/&	205-33	E
WATER ANALYSIS	later We	11 #26
	me/l	ing/l*
Calcium (Co++)	59.33	1189
Magnesium (Mg++)	105.68	1235
Sodium (No.+) (calculated)	508.85	11.698
Iron		212
Bicarbonate (HCO',)	0.20	12
Carbonate (CO _{3, -})	Not	Found
Hydroxide (OH-)	Not	Found
	81.09	3895
	592.57	21,013
5.9¢h c 68 °F		
Dissolved Solids on Evap. at 103° - 105° C		
Hardness as Ca CO3	165.01	8251
	0.20	10
Carbonate Hardness as CaCOs (temporary)		8241
Carbonate Hardness as CaCO3 (temporary) Non-Carbonate Hardness as CaCO3 (permanent)	164.81	
Carbonate Hardness as CaCO; (temporary) Non-Carbonate Hardness as CaCO; (permanent) Alkalinity as CaCO;	164.81 0.20	10
Carbonate Hardness as CaCO; (temporary) Non-Carbonate Hardness as CaCO; (permanent) Alkalinity as CaCO; Specific Gravity c 68° F 1.025	0.20	10
Carbonate Hardness as CaCO; (temporary) Non-Carbonate Hardness as CaCO; (permanent) Alkalinity as CaCO; Specific Gravity c 68° F 1.025	164.81 0.20	10
Carbonate Hardness as CaCO; (temporary) Non-Carbonate Hardness as CaCO; (permanent) Alkolinity as CaCO; Specific Gravity c 68° F 1.025	164.81 0.20	
Carbonate Hardness as CaCO; (temporary) Non-Carbonate Hardness as CaCO; (permanent) Alkalinity as CaCO; Specific Gravity c 68° F 1.025 • mg/l= milligrams per Liter • mg/l= milligrams per Liter	EX1	10 HIBIT NO. 8 F
Carbonate Hardness as CaCO; (temporary) Non-Carbonate Hardness as CaCO; (permanent) Alkalinity as CaCO; Specific Gravity c 68° F 1.025 · mg/l = milligrams per Liter · me/l = milligrams per Liter	EXI	10 HIBIT NO. 8 F LT LAKE POOL
Carbonate Hardness as CaCO; (temporary) Non-Carbonate Hardness as CaCO; (permanent) Alkalinity as CaCO; Specific Gravity c 68° F 1.025 • mg/l= milligrams per Liter • mg/l= milligrams per Liter • mg/l= milligrams per Liter Calcium Carbonate Scaling Index - Negative at 860 Calcium Sulfate Scaling Index - Negative	EXI DF Lea Con	HIBIT NO. 8 F

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KE-TONE	ON NORTH LEECH	EMICAL OFNEWME	CORPOR x100 P. O. BO	ATION x 1499
		OBBS, NEW MEX		3
· · ·	5	W-		
	Lease Type of Sample		Sampling Date_	
	WATER ANALYSI	5		
	IONIC FORM		me/l *	mg/l *
 Colcium (Ca++)			66.87	1340
/agnesium (Mg++)			18.67	227
odium (Na+)	(calculated)		229.14	5268
Iron				13
				,, , , , , , , , , , , , , , , , ,
Bicarbonate (HCO3)			Not F	ound
Carbonate (CO 3 -)			0.60	18
Hydroxide (OH-)			0.20	3
Sulphate (SO , -)			39.10	1878
Chloride (C1-)			274.78	9744
· · · · · · · · · · · · · · · · · · ·				
9.3hc 68 °F				:
Dissolved Solids on Evap. at 1	03°-105° C			······
Hardness as Ca. CO3			85.58	4277
Carbonate Hardness as CaCO	a (temporary)		0.80	40
Non-Carbonate Hardness as (CaCO3 (permanent)		84.74	4237
Alkalinity as CaCO3			0.80	40
			····	and the second second second second second second second second second second second second second second second

* mg/I = milligrams per Liter

* me/l = milliequivalents per Liter

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	INITED CHEM	ICAL CORPO	ORATION
	(¢ KE-TONE) ORTHED CHEEN		ORALION
1. t	601 NORTH LEECH	. P. C	. BOX 1499
	HOBBS	, NEW MEXICO 88240	- 4
			34
	Minerals, I	ncorporated	
	Company	·····	
. :	FieldSalt_Lake_E	ield	
	Binaham We	ell Samalias (Data 10/24/68
	LeoseBingham We	ellSampling (Date 10/24/68
	LeaseBingham We Type of SampleNV NIC VID	ellSampling [_2/-20.33	Date 10/24/68
	LeoseBingham We Type of Sample/V/V////////////////////////////	ellSompling (Sompling (SomeSome Worten - Mer.!	Dote 10/24/68
SH-SOM	Leose Bingham We Type of Sample Mir Mir Mir Mir Mir Mir Mir Mir Mir Mir	ellSompling (21-20-33 W/5+2- M/2 ! me/l	Dote <u>10/24/68</u>
	Lease Bingham We Type af Sample Wir Mir Mir Mir Mir Mir Mir Mir Mir Mir M	ellSampling (Date <u>10/24/68</u>
****	LeoseBingham We Type of Sample//// /// /// /// WATER ANALYSIS IONIC FORM Calcium (Ca++)	ellSampling (Dote <u>10/24/68</u> // <i>4 3 4</i> mg/l-* <u>398</u> 209
	LeoseBingham We Type of Sample///////////////////////////////	ell	Dote 10/24/68
	LeoseBingham We Type of Sample	ellSompling (2/ - 20 - 3 - 3 W-5 + 5 M-2 - 1 19.86 17.19 84.21	Dote <u>10/24/68</u> // <i>H B</i> ↓ <u>398</u> 209 ↓ 1936 ↓
	LeoseBingham We Type of SampleWWILL VU WATER ANALYSIS IONIC FORM Calcium (Ca++) Magnesium (Mg++) Sodium (Na+)(cal.)	ellSompling (21-20-33 WS+2 W2 / 19.86 17.19 84.21	Dote <u>10/24/68</u> <u>mg/l</u> <u>398</u> <u>209</u> <u>1938</u> ∟
	LeoseBingham We Type of SampleWW Mic Mid WATER ANALYSIS IONIC FORM Calcium (Ca++) Magnesium (Mg++) Sodium (Na+) (cal.)	ell Sompling (21-20.33 Weter Met me/1 19.86 17.19 84.21	Dote <u>10/24/68</u> mg/l- <u>398</u> 209 <u>1936</u>
	LeoseBingham We Type of Sample	ell Sompling (21-20.33 Weter Med me/1 19.86 17.19 84.21	Dote <u>10/24/68</u> / # <u>8</u> <u>1</u> mg/l- <u>398</u> 209 <u>-</u> <u>1936</u> <u>-</u>
	LeoseBingham We Type of Sample	ell Sompling (21-20.33 W5+2- M24 me/l 19.86 17.19 84.21 	Dote 10/24/68
	LeoseBingham We Type of SampleWATER ANALYSIS WATER ANALYSIS IONIC FORM Calcium (Ca++) Magnesium (Mg++) Sodium (Na++) (cal.) Bicarbonate (HCO ₃) Carbonate (HCO ₃)	ellSompling (2/ - > 0 - 3 -3 W-5 +	Dote 10/24/68
	LeoseBingham We Type of Sample	ellSompling (2/ - > 0 · 3 3 W5+5- W2 / 19.86 17.19 84.21 3.21 NOT NOT	Dote 10/24/68
	LeoseBingham We Type of Sample	ell Sompling (21-20-33 WS+2- W2 / 19.86 17.19 84.21 3.21 NOT NOT 18.84 27	Dote 10/24/68
	LeoseBingham We Type of SampleWW MUL WU WATER ANALYSIS IONIC FORM Calcium (Ca++) Magnesium (Mg++) Sodium (Na+) (cal.) Bicarbonate (HCO,) Carbonate (HCO,) Carbonate (CO, -) Hydroxide (OH-) Sulphate (SO) Chloride (C1-)	Sompling C 21 - 20 · 3 3 Ws+s- me/l 19.86 17.19 84.21 3.21 NOT 18.84 99.21	Dote 10/24/68 mg/l- 398 209 1936 - 196 FOUND FOUND 905 3518
	LeoseBingham We Type of Sample	Sompling C 21 - 20 · 3 3 Wster me/l 19.86 17.19 84.21 3.21 NOT 18.84 99.21	Dote 10/24/68 mg/l- 398 209 1938 - 196 FOUND FOUND 905 3518 -
	LeoseBingham We Type of Sample	Sompling C 2/ - 20.33 Wster me/l 19.86 17.19 84.21 3.21 NOT 18.84 99.21	Dote 10/24/68
	LeoseBingham We Type of Sample	€ Sompling (2/-20.33 W=+=M= / 	Dote 10/24/68
	Leose Bingham We Type of Sample MM MM MM MM WATER ANALYSIS IONIC FORM Calcium (Ca++) Magnesium (Mg++) Sodium (Na++) (cal.) Bicarbonate (HCO,) (cal.) Bicarbonate (HCO,) Carbonale (CO) Hydroxide (OH -) Sulphate (SO) Chloride (C1 -)	ell Sompling (2/-20.33 W=+=- M=4 me/l 19.86 17.19 84.21 	Dote 10/24/68
	Leose Bingham We Type of Sample MM MM MM MM WATER ANALYSIS IONIC FORM Calcium (Ca++) Magnesium (Mg ++) Sodium (Na+) (cal) Bicarbonate (HCO,) Carbonate (CO) Hydroxide (OH -) Sulphate (SO) Chloride (C1 -) Massel	Ell Sompling (2/-20-33 W=+=- W= / 19.86 17.19 84.21 3.21 NOT 18.84 99.21	Dote 10/24/68
	LeoseBingham We Type of Sample	Ell Sompling (2) - 20 - 3 - 2 WE + - WE / 19.86 17.19 84.21 	Dote 10/24/68
	Leose Bingham We Type of Sample NV NV NV NV WATER ANALYSIS IONIC FORM: Calcium (Ca++) Magnesium (Mg++) Sodium (Na,+) (cal.) Bicarbonale (HCO;) Carbonale (OO; -) Hydroxide (OH -) Sulphate (SO) Chloride (C1-) Zosolved Solids on Evap: at 103° - 105° C Hardness as Ca CO; Carbonate Hardness as CoCO; (temporary)	Ell Sompling (2/ - 20 - 3 -3 WE + 2 - ME / 19.86 17.19 84.21 	Dote 10/24/68 mg/l- 398 209 1936 FOUND FOUND FOUND 905 3518 1853 161
	Leose Bingham We Type of Sample NV Inc. yd. WATER ANALYSIS WATER ANALYSIS Calcium (Ca++) Wagnesium (Mg++) Sodium (Na++) (cal.) Bicarbonale (HCO;) (cal.) Bicarbonale (OF,) (cal.) Sulphate (SO, -) (cal.) Chloride (C1-) (clium (Ca++)) Non-Carbonale Hardness as CaCO; (temporary) Non-Carbonale Hardness as CaCO; (permanent)	€ Sompling (2/ - > 0 · 3 3 W=+=M= / 	Dote 10/24/68 mg/l- 398 209 1936 FOUND FOUND FOUND 905 3518 1853 161 1692 209
	Lease Bingham We Type of Sample Mix mix mit WATER ANALYSIS WATER ANALYSIS Calcium (Ca++) Magnesium (Mg++) Sodium (Na+) (cal.) Bicarbonale (HCO3) (cal.) Bicarbonale (CO.2) Hydroxide (OH-) Sulphate (SO) Chloride (C1-) Z. 55hret @:68b.°E* Dissolved Solids on Evap: at 103 ^{5*} 105° C Hardness as Ca CO; Carbonate Hardness as CaCO, (temporary) Non-Carbonate Hardness as CaCO, (permanent) Alkalinity as CaCO,	Ell Sompling (2) - 20 - 3 - 3 WE + 2 - ME / 19.86 17.19 84.21 	Dote 10/24/68
	Lease Bingham We Type of Sample MM MM MM MM MM WATER ANALYSIS IONIC FORM Calcium (Ca++) Magnesium (Mg++) Sodium (Na,+) (cal.) Bicarbonale (HCO,) (cal.) Bicarbonale (OG, -) (cal.) Hydroxide (OH-) Sulphate (SO, -) Chloride (C1-) (cl.) Z. 5phrct: @*68,*E* Dissolved Solids on Evap. at 103* 105* C Hardness as Ca CO, (temporary) Non-Carbonute Hardness as CaCO, (temporary) Non-Carbonute Hardness as CaCO, (permanent) Alkalinity as CaCO, Specific Gravity c 68* F	Ell Sompling (2) - 20.33 WEFEF MEN 19.86 17.19 84.21 0 3.21 NOT 18.84 99.21 3.7.05 3.21 33.84 3.21 1.005	Dote 10/24/68
	Leose Bingham We Type of Sample NV NV NV NV WATER ANALYSIS WATER ANALYSIS Calcium (Ca++) Magnesium (Mg++) Sodium (Na+) (cal.) Bicarbonate (HCO,) (cal.) Bicarbonate (CO) (cal.) Hydroxide (OH -) Sulphate (SO) Chloride (C1 -) Chloride (C1 -) Z. 55hr et @: 68; °E* Dissolved Solids on Evap: at 103° 105° C Hardness as Ca CO; Carbonate Hardness as CaCO, (temporary) Non - Carbonate Hardness as CaCO, (permanent) Alkalinity as CaCO, Specific Gravity c 68° F Second Carbonate (Solids on Evap)	Sompling (2/ - 20.33 Wster me/l 19.86 17.19 84.21 3.21 NOT 18.84 99.21 33.21 13.84 37.05 3.21 13.84 13.21 13.21 1.005	Dote 10/24/68
	Leose Bingham Wet Type of Sample NV NV NV NV WATER ANALYSIS Calcium (Ca++) Magnesium (Mg ++) Sodium (Na+) (cal.) Bicarbanate (HCO,) Carbanate (CO) Hydroxide (OH -) Sulphate (SO) Chloride (C1 -) Zi Sahre: @:68: "F" Dissolved Solids on Evap: at 103" - 105" C Hardness as Ca CO, Carbanate Hardness as CaCO, (temporary) Non - Carbanate Hardness as CaCO, (permanent) Alkalinity as CaCO, Specific Gravity : 68" F	Ell Sompling (2/-20.33 Watter Mail Mathematical me/l 19.86 17.19 84.21 3.21 NOT 18.84 99.21 37.05 3.21 33.84 33.21 1.005 EXHIBIT ::0. 2 2	Dote 10/24/68
	Leose Bingham Wet Type of Sample NV NV NV NV WATER ANALYSIS WATER ANALYSIS Calcium (Ca++) Magnesium (Mg ++) Sodium (Na+) (cal.) Bicarbonale (HCO;) (cal.) Bicarbonale (CO; -) (cal.) Hydroxide (OH-) Sulphate (SO -) Chloride (SO -) Chloride (C1-) Zi 58hre: @:68; "F" Dissolved Solids on Evap: at 103° · 105° C Hardness as Ca CO; Carbonate Hardness as CaCO, (temporary) Non-Carbonate Hardness as CaCO, (permanent) Alkalinity as CaCO; Specific Gravity : 68° F mg/l = milligrams per Liter	Ell Sompling (2/-20.33 Ws+s- M24 mc/l 19.86 17.19 84.21 3.21 NOT NOT 18.84 99.21 37.05 3.21 33.84 3.21 1.005 EXHIBIT ::0. 3 2.45	Dote 10/24/68

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SOUTHWESTERN LABORATORIES

FORT WORTH DALLAS HOUSTON MIDLAND BEAUMONT TEXARKANA

CONSULTING. ANALYTICAL CHEMISTS AND TESTING ENGINEERS

	Mi	dland,	Texas	2-25-69	i .	File No(<u>c-1902</u>	-R1	
Report of te	ests on	Water							
То		Mr. Ed L.	Reed]	Date Rec	rd. 2-18	8-69
Received fr	om	Mr. Ed L.	Reed						
Identificatio	on Marks	As Shown							
								Mg/L	Mg/L
<u>Lab. No.</u>	<u>Sample</u>	Descriptio	on				<u>Ch</u>	loride	Sulfate
C-5151	No. 1-	A, Soil san	nple, N	end of I	onto				
	(1:1	extract)				****	- 4	8931	37698
C-5152	No. 2,	Spring, SI	V Gatuna	a, S of H	lighway –		- 16	3,105	24594
C-5153	No. 3,	South side	e of Gat	tuna			- 6	6660	29728
C-5154	No. 4,	Gatuna, i	n draw l	N of High	way		- 7	2333	24273
C-5155	No. 5,	Gatuna, N	V end a	t oil wel	.1, NW 0	f well			1.1.1.1.1
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C-5156	No. 6,	Gatuna, N	N end, 1	VE of oil	. well,	ravine			· ·
	flow	ing South	ه : هه رُهه _و بين , تهه : هه (هه ، ه				- 1	.0992	13771
C-5157	No. 7,	Spring No	. 4, P1a	ata			- .	7978	12643
C-5158	No. 1,	Salt crys	tals, To	onto (Moi	.st):			,	
	Chlo	ride (C1)	4.20% 1	by weight					-
	Sulf	ate (SO_{ℓ})	29.23% 1	by weight					
	No. 1-	A = - NO Si	ulfide (or Sulfit	.e detec	ted.			
	NO. To			or Burrre					
		· .							
								. #7# 1.***	
		na ing taong siya.						a na ana ao amin'ny faritr'o dia mampika dia Na mampika dia ma	مرجو بيد وراجر المركز والمهرمين
	1997 - 1997 1997 - 1997 - 1998 -					ار بالمرجعة - موجوعة معرفة الإنتقالية	and a second		
				경험은 1억원 관계 (1억원 - 1월 14일 (1억원) 1억원 - 1월 14일 (1억원) 1억원		a and the Lark		مورد در محمد المحمد المحمد المحمد المحمد	e mana da ser a la composición de la composición de la composición de la composición de la composición de la c
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				SC	UTHWEST	ERN LAI	BORATO	RIES	
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Lab. No.					Vank	H/	Ka I		
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Our letters and repo and reports apply or	orts are for the nly to the sampl	exclusive use of the c les tested and are not i	lients to whom lecessarily indic	they are addresse ative of the qualit	d. The use of our ies of identical	namea must re or similar prod	eccive our pructs.	rior written ap	proval. Our letters

FORM NO. 130-B

THWESTERN LABORATORIES

FORT WORTH DALLAS HOUSTON MIDLAND BEAUMONT TEXARKANA

CONSULTING. ANALYTICAL CHEMISTS AND TESTING ENGINEERS

	Midland,	_ Texas	2-13-59	File No	C-1902-R1
Report of tests on	Water				
То	Mr. Ed L	. Reed			Date Rec'd. 2-12-69
Received from	Mr. Ed I	. Reed			
Identification Marks	Lea Coun Reed Ha	ty, New 1	Mexico, Larry r. from tap.	Squires, 2 wells.	sampled by Joe WL 42.5'.

Mg/L

Chloride ----- 362

Sulfate ----- 309

Conductivity ----- 1861 Micromhos/cm @ 25° C.

Copies: 3cc Mr. Ed L. Reed

SOUTHWESTERN LABORATORIES

Barton ack

Lab. No. C-5121

Our letters and reports are for the exclusive use of the clients to whom they are addressed. The use of our names must receive our prior written approval. Our letters and reports apply only to the samples tested and are not necessarily indicative of the qualities of identical or similar products.

FORM NO. 130-8

SOUTHWESTERN LABORATORIE

FORT WORTH DALLAS HOUSTON MIDLAND BEAUMONT TEXARKANA

CONSULTING, ANALYTICAL CHEMISTS AND TESTING ENGINEERS

	Midland,	_ Texas	2-13-69	File No.	<u>C-1902-R</u>]	-
Report of tests on	Water					
То	Mr. Ed L	. Reed			Date Rec'd.	2-12-69
Received from	Mr. Ed L	. Reed				
Identification Mark	s Lea Coun Reed, #1 WL 39	ty, New 6, Sec. .58'.	Mexico, Larr 25-T20G, R32	y Squires, E, from wa	sampled b ste trough	by Joe 1,

Mg/L

Chloride ----- 85

Sulfate ----- 82

Conductivity ----- 837 Micromhos/cm @ 25° C.

Copies: 3cc Mr. Ed L. Reed

SOUTHWESTERN LABORATORIES

Jac he H Barton

Lab. No. C-5120

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SOUTHWESTERN LABORATORIES

FORT WORTH DALLAS HOUSTON MIDLAND BEAUMONT TEXARKANA

CONSULTING. ANALYTICAL CHEMISTS AND TESTING ENGINEERS

	Midland,	Texas	2-13-69	File No. C-1902-R1	
Report of tests on	Water				
То	Mr. Ed L.	. Reed		Date Rec'd. 2-12-6	9
Received from	Mr. Ed L	Reed			
Identification Mark	s Lea Count	:y, New	Mexico, Larry	Squires, sampled by Joe	: -

Mg/L

Chloride ----- 8864

Sulfate -----11930

Conductivity ----- 10,000 / Micromhos/cm @ 25° C.

Copies: 3cc Mr. Ed L. Reed

SOUTHWESTERN LABORATORIES

Jack H

Lab. No. C-5122

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FORM NO. 130-B

SOUTHWESTERN LABORATORIES

FORT WORTH DALLAS HOUSTON MIDLAND BEAUMONT TEXARKANA

CONSULTING, ANALYTICAL CHEMISTS AND TESTING ENGINEERS

	Midland,	Texas	2-13-69	File No	<u>2-1902</u> -R1	
Report of tests on	Water					
То	Mr. Ed L	. Reed			Date Rec'd. 2-	12 - 69
Received from	Mr. Ed L	. Reed				
Identification Mark	s Lea Coun Reed, Sp	ty, New ring #2,	Mexico, Larr due East of	y Squires, Laguna Pla	sampled by ata at head	Joe water

Mg/L

Chloride ----- 7446

Sulfate ----- 12743

Conductivity ----- 10,000 / Micromhos/cm @ 25° C.

Copies: 3cc Mr. Ed L. Reed

Jack H Barlon

Lab. No. C-5123

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FORM NO. 130-B

THWESTERN LABORATO

FORT WORTH DALLAS HOUSTON MIDLAND BEAUMONT TEXARKANA

S

CONSULTING, ANALYTICAL CHEMISTS AND TESTING ENGINEERS

1	Midland,	Texas	2-13-69	File No	<u>C-1902</u> -R1	
Report of tests on	Water					
То	Mr. Ed I	L. Reed			Date Rec'd.	2-12-69
Received from	Mr. Ed 1	L. Reed				
Identification Marks	Lea Cour	hty, New #3 just	Mexico, Larr North of #2.	y Squires, 200 ft	by Joe R at head w	eed, ater.

Mg/L

Chloride ----- 7446

Sulfate -----11755

Conductivity -----10,000 / Micromhos/cm @ 25° C.

Copies: 3cc Mr. Ed L. Reed

SOUTHWESTERN LABORATORIES

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FORM NO. 130-8

Lab. No. C-5124