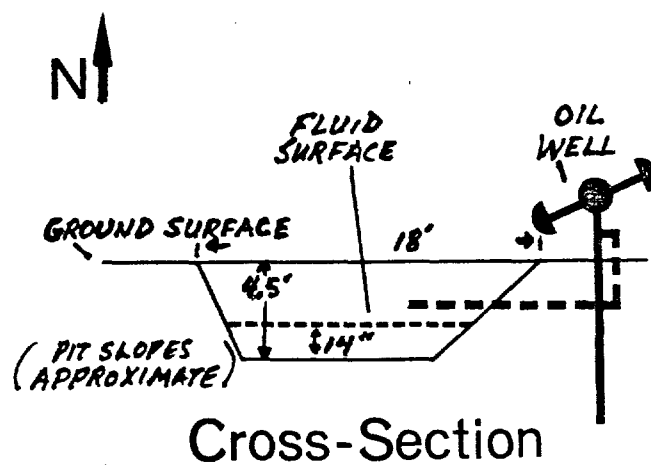
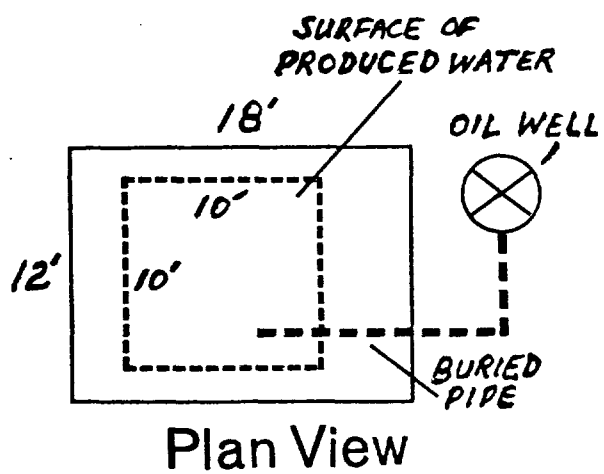


EXHIBIT ZAMAN-3

DIMENSIONS OF PRODUCED WATER DISPOSAL PIT  
Duncan Oil Field  
T29N, R16W, Sec. 6, 50' FSL, 50' FWL, San Juan County, N.M.



March 27, 1985

M E M O R A N D U M

TO : Akhtar Zaman, Director  
Minerals Department

FROM : Masud-Uz-Zaman, Director  
Water Management Department

SUBJECT: Duncan Oil Field Well No. 6-11

We are in the process of conducting Duncan Oil Field's groundwater impact study for the hogback area. Well No. 6-11 has been identified in the field. We would appreciate it if you could provide us a complete well record including electric logs for this well.

  
Masud-Uz-Zaman

UNITED STATES  
DEPARTMENT OF THE INTERIOR  
GEOLOGICAL SURVEY

## APPLICATION FOR PERMIT TO DRILL, DEEPEN, OR PLUG BACK

## 1a. TYPE OF WORK

DRILL ☒DEEPEN ☐PLUG BACK ☐

## b. TYPE OF WELL

OIL  
WELL ☒GAS  
WELL ☐

OTHER

SINGLE  
ZONE ☐MULTIPLE  
ZONE ☐

## 2. NAME OF OPERATOR

Walter Duncan

## 3. ADDRESS OF OPERATOR

Box 234, Farmington, NM 87401

## 4. LOCATION OF WELL (Report location clearly and in accordance with any State requirements.\*)

At surface

50' FSL - 50' FWL

At proposed prod. zone

## 14. DISTANCE IN MILES AND DIRECTION FROM NEAREST TOWN OR POST OFFICE\*

8 miles east-southeast of Shiprock, NM

## 15. DISTANCE FROM PROPOSED\*

LOCATION TO NEAREST  
PROPERTY OR LEASE LINE, FT.  
(Also to nearest drlg. unit line, if any)

50'

## 16. NO. OF ACRES IN LEASE

640

17. NO. OF ACRES ASSIGNED  
TO THIS WELL

2-1/2

18. DISTANCE FROM PROPOSED LOCATION\*  
TO NEAREST WELL, DRILLING COMPLETED,  
OR APPLIED FOR, ON THIS LEASE, FT.

560'

## 19. PROPOSED DEPTH

690'

## 20. ROTARY OR CABLE TOOLS

Cable 0-20' Rotary 20-690'

## 21. ELEVATIONS (Show whether DF, RT, GR, etc.)

4984' GR

## 22. APPROX. DATE WORK WILL START\*

9-20-75

## 23.

## PROPOSED CASING AND CEMENTING PROGRAM

SIZE OF HOLE	SIZE OF CASING	WEIGHT PER FOOT	SETTING DEPTH	QUANTITY OF CEMENT
	7"	20#	20'	5 sx
6-1/4"	4-1/2"	9.5#	675'	75 sx

1. Drive 7" csg thru boulders to approx 20'
2. Drill 6-1/4" hole to top of Dakota formation approx 675'
3. Set 4-1/2" csg cemented w/75 sx
4. Drill 3-7/8" hole to approx 690' w/air to test Dakota formation
5. If commercial oil production is indicated 2-3/8" tbg will be run and the well completed naturally in the open hole.

RECEIVED

SEP 18 1975

U. S. GEOLOGICAL SURVEY  
FARMINGTON, NM

IN ABOVE SPACE DESCRIBE PROPOSED PROGRAM: If proposal is to deepen or plug back, give data on present productive zone and proposed new productive zone. If proposal is to drill or deepen directionally, give pertinent data on subsurface locations and measured and true vertical depths. Give blowout preventer program, if any.

## 24.

SIGNED

Jim L. Jacobs

TITLE

Agent

DATE

9-8-75

(This space for Federal or State office use)

PERMIT NO.

APPROVAL DATE

APPROVED BY

TITLE

DATE

CONDITIONS OF APPROVAL, IF ANY:

\*See Instructions On Reverse Side

**NEW MEXICO OIL CONSERVATION COMMISSION  
WELL LOCATION AND ACREAGE DEDICATION PLAT**

Form O-107  
Superseded by O-128  
Effective 1-1-67

All distances must be from the outer boundaries of the Section

Owner <b>Walter Duncan</b>		Lease <b>North Hogback</b>		Well No. <b>6-11</b>
Unit Letter <b>M</b>	Section <b>6</b>	Township <b>29 North</b>	Range <b>16 West</b>	County <b>San Juan</b>
Actual Footage Location of Wells:				
<b>50</b> feet from the <b>South</b> line and <b>50</b> feet from the <b>West</b> line				
Ground Level Elev. <b>4984</b>	Producing Formation <b>Dakota</b>	Pool <b>Slickrock Dakota</b>	Dedicated Acreage <b>Proration Unit: 40</b>	

1. Outline the acreage dedicated to the subject well by colored pencil or hachure marks on the plat below.
2. If more than one lease is dedicated to the well, outline each and identify the ownership thereof (both as to working interest and royalty).
3. If more than one lease of different ownership is dedicated to the well, have the interests of all owners been consolidated by communitization, unitization, force-pooling, etc?

☐ Yes    ☐ No    If answer is "yes," type of consolidation \_\_\_\_\_

If answer is "no," list the owners and tract descriptions which have actually been consolidated. (Use reverse side of this form if necessary.) \_\_\_\_\_

No allowable will be assigned to the well until all interests have been consolidated (by communitization, unitization, forced-pooling, or otherwise) or until a non-standard unit, eliminating such interests, has been approved by the Commission.

SEC. 6			

50'

50'

Proposed Location

<b>CERTIFICATION</b>	
I hereby certify that the information contained herein is true and complete to the best of my knowledge and belief.	
Name <b>Jim L. Jacobs</b>	
Position <b>Agent</b>	
Company <b>Walter Duncan</b>	
Date <b>9-8-75</b>	
I hereby certify that the well location shown on this plat was plotted from field notes of actual surveys made by me or under my supervision, and that the same is true and correct to the best of my knowledge and belief.	
<b>Sept. 2, 1975</b>	
<div style="display: flex; justify-content: space-between;"> <div>             Date Surveyed <b>Sept. 2, 1975</b> </div> <div>             Surveyor <b>Frederick H. Reed</b> </div> </div>	
<div style="display: flex; justify-content: space-between;"> <div>             Registered Professional Engineer and/or Land Surveyor </div> <div> </div> </div>	

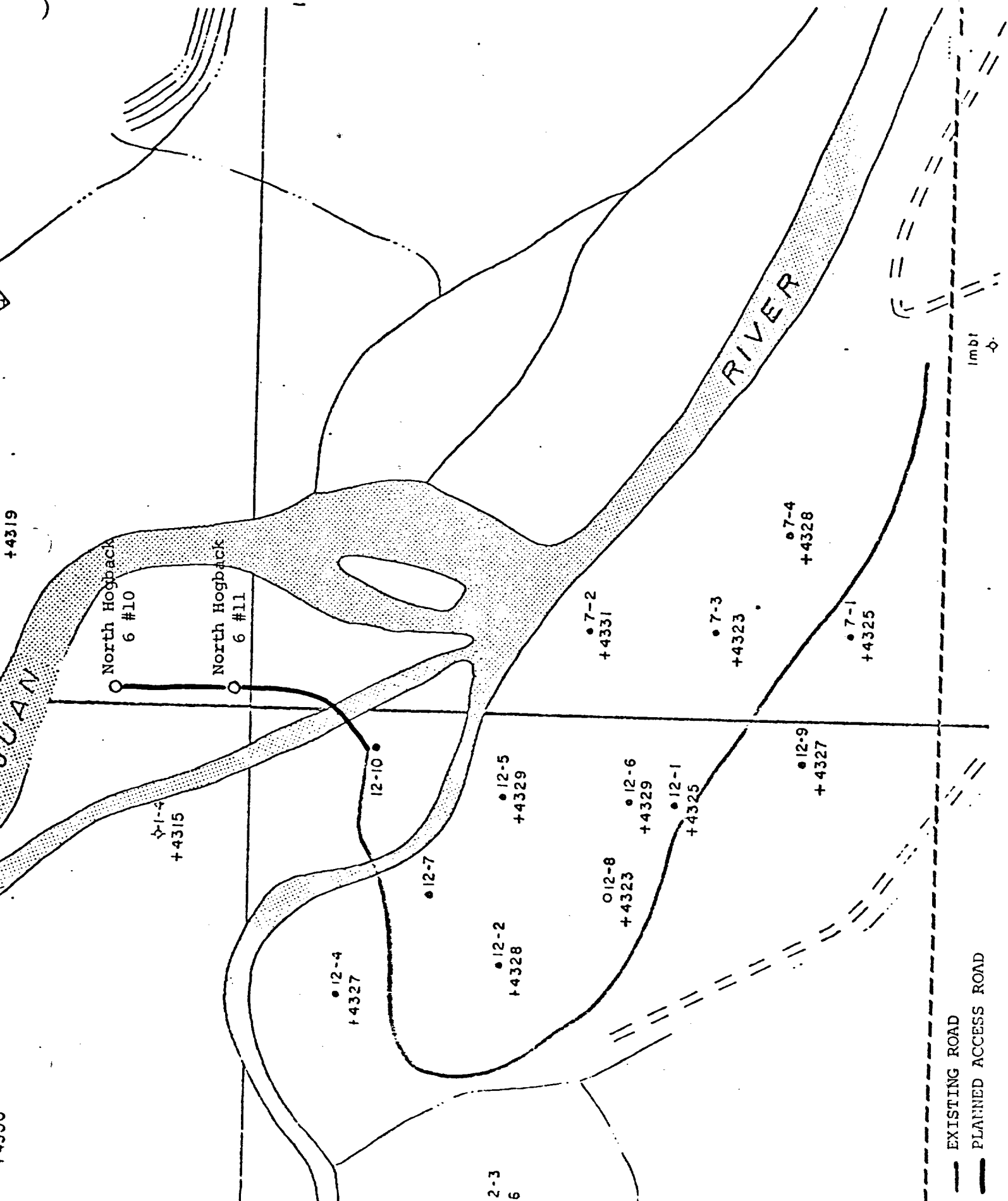
DEVELOPMENT PLAN

Walter Duncan

North Hogback 6 #11

1. Existing roads shown on attached plat.
2. Planned access from existing road approximately 600 ' to new location.
3. Location of well: 50' FSL - 50' FWL  
Sec 6, T29N, R16W  
San Juan County, NM
4. No additional well(s) planned on this lease at this time.
5. Existing oil tank will be used.
6. Water will be secured from San Juan River.
7. Waste materials will be buried on location or in reserve pit.
8. No permanent camp is planned; trailer house will be used on location while drilling.
9. Do not plan to build airstrip.
10. See attached plat for layout.
11. Pits will be filled, location cleaned up, and area not in use for producing well will be reseeded using broadcast method of planting.

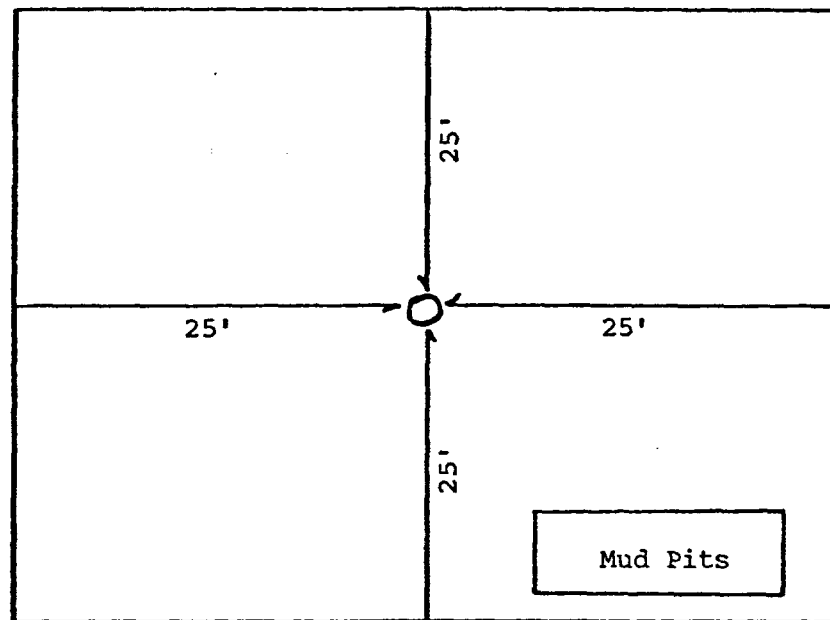
RECEIVED  
SEP 12 1955  
U. S. GEOLOGICAL SURVEY  
FERTILIZATION DIV.



WALTER DUNCAN

North Hogback 6 #11

Proposed Location Layout



Reserve Pit

N  
↑

U.S. GEOLOGICAL SURVEY  
WATER RESOURCES DIVISION  
FEB 27 1976  
U. S. GEOLOGICAL SURVEY  
ROSWELL, NEW MEXICO

FILE NO.

COMPANY WALTER DUNCAN 6-29N-16W

WELL North Hogback 6 No. 11

FIELD Slickrock Dakota

COUNTY San Juan STATE New Mexico

LOCATION: 50° FSL & 50° FWL

Other Services

sec 6 TWP 29N RGE 16W None

Permanent Datum Ground Level Elev. 4984

Log Measured from GL 0 Ft. Above Permanent Datum

Drilling Measured from GL Elevations: KB ----- DF ----- GL 4984

RECEIVED

FEB 26 1976

Date 1-29-76

Run No. One

Depth—Driller 680

Depth—logger 658

Bottom Logged Interval 657

Top Logged Interval 30

Casing—Driller 7 @ 27

Casing—Logger 30

Bit Size 6 1/4

Type Fluid in Hole Water

Density and Viscosity

pH and Fluid Loss

Source of Sample

Rm @ Meas. Temp.

Rmf @ Meas. Temp.

Rmc @ Meas. Temp.

Source of Rmf and Rmc

Rm @ BHT

Time Since Circ.

Max. Rec. Temp. Deg. F.

Equip. No. and location

Recorded By

Witnessed By

Jacobs

FOLD HERE

Changes in Mud Type or Additional Samples

Scale Changes

Date	Sample No.	Type Log	Depth	Scale Up Hole	Scale Do
Depth-Driller					
Type Fluid in Hole					
Dens.	Visc.				
pH	Fluid Loss	cc	cc		
Source of Sample					
Rm @ Meas. Temp.	@ °F	@ °F	Run No.	Tool Type	Tool Position
Rmf @ Meas. Temp.	@ °F	@ °F	One	EL-2"	Free
Rmc @ Meas. Temp.	@ °F	@ °F			
Source Rmf Rmc					
Rm @ BHT	@ °F	@ °F			
Rmf @ BHT	@ °F	@ °F			
Rmc @ BHT	@ °F	@ °F			

Equipment Data

Remarks:

SPONTANEOUS POTENTIAL

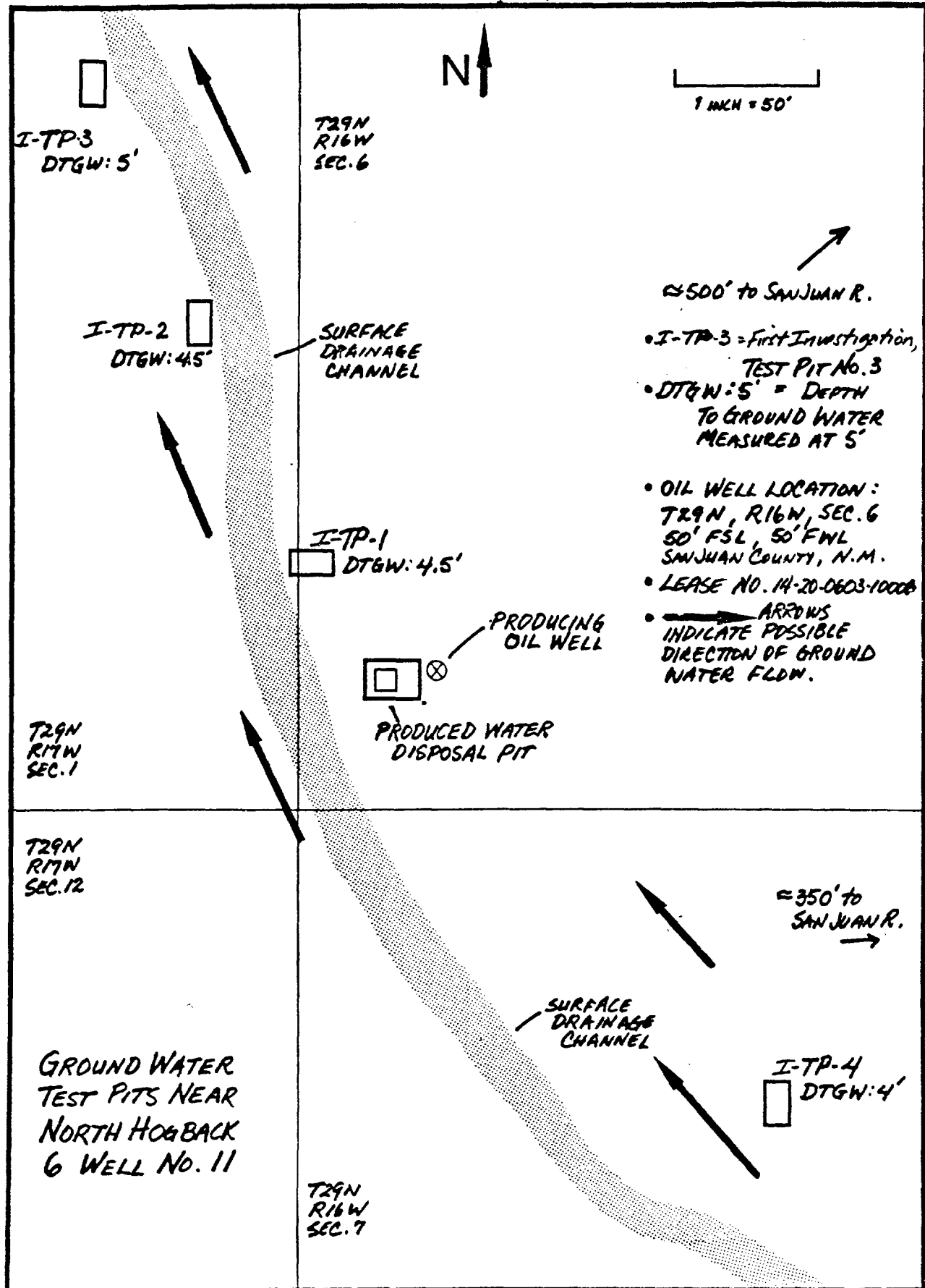
DEPTHS

RESISTIVITY

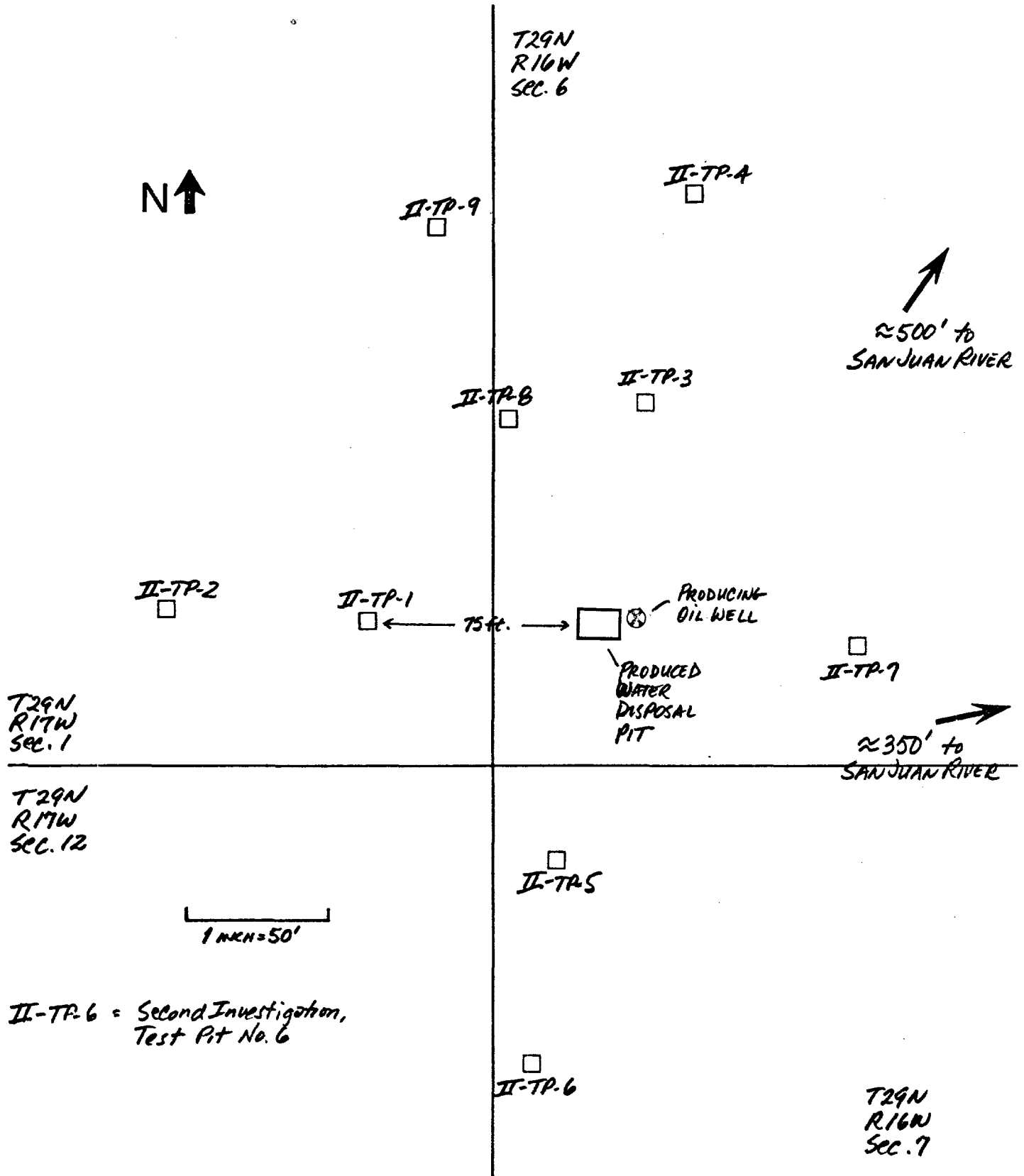
RESISTIVITY



SITE LOCATION MAP  
DUNCAN OIL FIELD HYDROLOGIC INVESTIGATION  
February 25, 1985



SITE LOCATION MAP  
DUNCAN OIL FIELD HYDROLOGIC INVESTIGATION  
March 18, 1985



LITHOLOGIC FEATURES OF TEST PITS  
DUNCAN OIL FIELD STUDY AREA  
February 25, 1985  
(dimensions are in length, width and depth, respectively)

---

Test Pit No.	Description of Observed Soil Conditions and Ground Water
1	Medium-to-coarse-grained sands with gravel and cobbles present; dark black oily stained soil at 3.5 feet below surface; water entered pit at 4.5 feet below surface; after <u>30 minutes</u> , pit bottom had filled with 6 inches of water; water was bluish-and greenish-black with obvious oily film on top and strong hydrocarbon smell.
2	1 foot of topsoil; 3 feet of gravelly medium-to-coarse-grained sand with trace of silt; soils saturated with black oily substance at 4 feet below surface; water entered pit at 4.5 feet below surface; after 20 minutes, water was brownish black and cloudy with noticeable oily sheen and strong hydrocarbon odor.
3	Sand and gravel observed to bottom of pit; dark gray to black staining of pit walls at 4 feet below surface; water entered pit at 5 feet below ground surface, initially from east side and within minutes also from south side; after 40 minutes, water was light brown and cloudy with obvious sandy sediment; thin oily sheen observed on top of water; discernable, but weak hydrocarbon odor noted.
4	2.5 feet of topsoil, silty fine sand to sandy silt (soils observed to be much more siltier than those of Test Pits 1, 2 or 3); roots and fibers plentiful in soil; gravelly medium-to-coarse-grained sands with gravel, cobbles and boulders from 2.5 to 5 feet below ground surface; moisture observed on pit walls at 3 feet below ground surface; water entered pit at 4 feet below ground surface and filled bottom of pit to 9 inches; very slight oily film observed on top of water, but no hydrocarbon or other foreign odors noted.

Lithologic Logs of Test Pits  
Duncan Oil Field  
Hydrologic Investigation, March 18, 1985  
(based on field notes of Masud Zaman)

Test PIT #1

- 0 - 1' Top Soil, Brown silty fine to coarse sand, gravel, boulders, roots, fibers, etc.
- 1 - 3.65' Gray-brown tan gravelly medium coarse SAND, occasional boulder and cobble, trace of silt
- 3.65' - 6.0' Dark-gray to black oily gravelly medium coarse SAND, occasional boulder, cobble

Static Water Level

11:15 AM - thin oily layer on water surface: 5.0'

1:42 PM - Collected Purgeable water sample PIT #1: 4.1'

Gary Eiceman collected water and soil samples

Test PIT #2

- 0.1' Sandy gravel, boulders, cobbles and root fibers, etc.
- 1 - 3.3' Gray brown fine to coarse SAND, trace of silt
- 3 - 6.0' Gray brown medium coarse SAND with fine to medium gravel, black staining at 4.8'

Static Water Level

11:35 AM - 5.0'

1:40 PM - 4.4'

Gray Eiceman collected water and soil samples

Test PIT #3

- 0 - 1.5' Top soil, silty sand, gravel, boulder, cobble, root fibers, etc.
- 1.5' - 4.1' Gray brown fine to coarse SAND, some fine to medium gravel, occasional boulder
- 4.1' - 6.5' Gray oily fine to coarse SAND, gravel, boulders, etc.

Water seeping into the pit from all directions at about 5.0' below the surface.

Static Water Level

12:21 PM - 5.1'

1:45 PM - 4.2'

Gary Eiceman collected water and soil samples

Masud Zaman collected two perge samples and one soil sample

Test PIT #4

0 - 0.5' Top soil, brown silty SAND, gravel, boulder, cobbles, roots, etc.

0.5' - 5' Gray-brown gravelly medium coarse SAND, occasional boulder, roots

Static Water Level

12:33 PM - 3.8'

1:46 PM - 3.3'

Gary Eiceman collected water and soil samples

Test PIT #5

0 - 1.5' Top soil, brown silty sand, gravel, boulder, cobble, roots and fibers

1.5' - 6.0' Gray-brown fine to coarse SAND, some gravel, occasional boulder

Static Water Level

12:40 PM - 5.0'

1:43 PM - 4.1'

Gary Eiceman collected water and soil samples

Test PIT #6

0 - 0.5' Top soil, gray brown silty fine SAND, occasional boulders

0.5' - 3.0' Gravelly tan fine to coarse SAND, some boulders and cobbles

3.0' - 6.0' Gray-brown medium coarse SAND, trace of silt, some gravel

Static Water Level

12:45 PM - 5.8'

1:43 PM - 4.0'

Gary Eiceman collected water and soil samples

Test PIT #7

0 - 1' Top soil, brown sandy SILT with some gravel and boulders, some root/fibers

1 - 5.0' Gray-tan brown gravelly medium coarse SAND, some boulder and cobble

Static Water Level

12:56 PM - 3.10'

1:44 PM - 3.3'

Gary Eiceman collected water and soil samples

\*PRODUCED WATER PIT

PERGE SAMPLES FROM PIT BY MASUD ZAMAN

Gary Eiceman also collected water samples

Test PIT #8

0 - 1' Top soil, brown tan sandy-silty CLAY roots (lot)

1 - 3.5' Gravelly brown medium coarse SAND, some boulder and cobbles

3.5' - 6.0' Dark oily medium coarse sand, some gravel, boulder

Static Water Level

1:20 PM - 4.8'

1:46 PM - 4.2'

Gary Eiceman collected water and soil samples

Test PIT #9

0 - 1.8' Gravelly brown silty fine to coarse SAND, boulder and cobbles, root fibers

4.10' Gray-brown medium coarse SAND, some fine to medium gravel

4.10' - 6.0' Black oily medium coarse SAND, some fine to medium gravel

Static Water Level

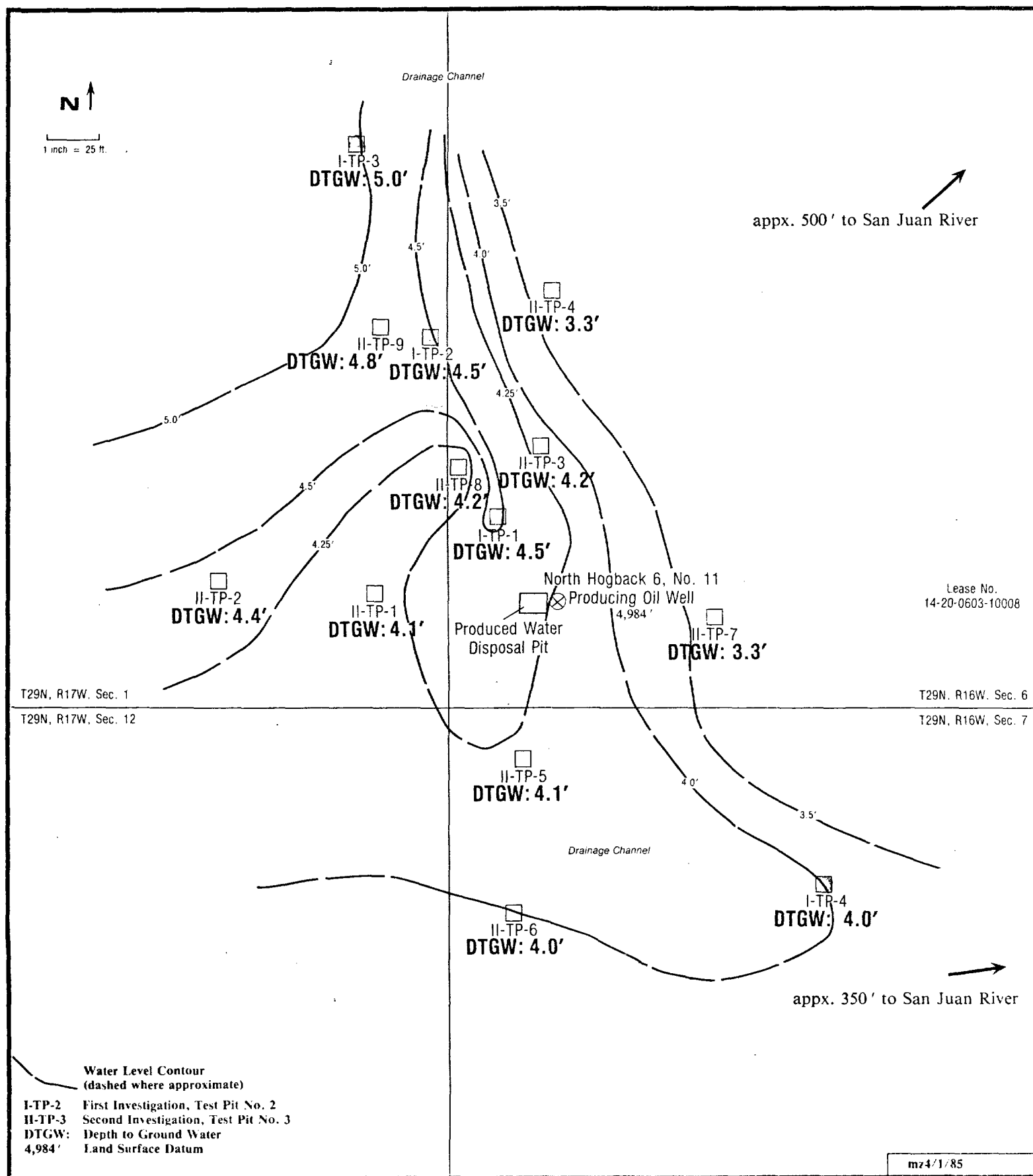
1:36 PM - 5.3'

1:48 PM - 4.8'

Gary Eiceman collected water and soil samples

ALL TEST PITS WERE DUG BY PORTABLE BACK-HOE  
FURNISHED BY NAVAJO DEPARTMENT OF OPERATIONS & MAINTENANCE

# WATER LEVEL MAP DUNCAN OIL FIELD HYDROLOGIC INVESTIGATION February 25 and March 18, 1985 San Juan County, N.M.



EXHIBIT

HYDROLOGIC GRADIENT CALCULATIONS  
Duncan Oil Field Investigations  
(prepared by M. Zaman)

Calculation No. 1

$$\begin{aligned} \frac{\text{Change in Water Table Level}}{\text{Horizontal Length of Study Area}} & \quad \frac{(dh)}{(dl)} = \text{gradient constant} \\ \text{gradient constant} \times 5,280 \text{ feet/mile} & = \text{hydrologic gradient (ft./mile)} \\ \frac{1.5 \text{ feet}}{400 \text{ feet}} \times 5,280 \text{ feet/mile} & = 19.80 \text{ feet/mile} \end{aligned}$$

Calculation No. 2

$$\begin{aligned} \frac{5,280 \text{ feet/mile}}{\text{Horizontal Length of Study Area}} & = n \text{ miles} \\ n \text{ miles} \times dh \text{ (Change in Water Table Level)} & = \text{Hydrologic Gradient} \\ \frac{5,280 \text{ feet/mile}}{400 \text{ feet}} & = 13.2 \text{ miles} \\ 13.2 \text{ miles} \times 1.5 \text{ feet} & = 19.8 \text{ ft./mile} \end{aligned}$$



GROUND WATER SAMPLING DATA  
DUNCAN OIL FIELD STUDY AREA  
February 25, 1985

Sample No.	Pit and Sample Identification	Fluid Sampled	Parameters Sampled	Time
1	Separator fluid from pipe emptying into disposal pit	produced water	purgeable aromatics,	1130
2	Fluid standing in unlined pit next to producing oil well	produced water	purgeable aromatics, general chemistry, metals, nitrates	1215
3	Test Pit No. 1	ground water	purgeable aromatics, general chemistry, metals, nitrates	1340
4	Test Pit No. 2	ground water	purgeable aromatics, general chemistry, metals, nitrates	1355
5	Test Pit No. 3	ground water	purgeable aromatics	1440
6	Test Pit No. 4	ground water	purgeable aromatics, general chemistry, metals, nitrates	1430

In the table above, "general chemistry" refers to dissolved solids and major ions, and "purgeable aromatics" to the benzene family of hydrocarbons, namely benzene, toluene, ethylbenzene and xylene.

mz3/8/85

ANALYTIC RESULTS FOR WATER SAMPLING IN THE DUNCAN OIL FIELD  
 Field Investigation Conducted by M. Zaman et al., February-March 1985  
 Township 29 North, Range 16 West, Section 6, 50 Feet FSL, 50 Feet FWL  
 (all concentrations in milligrams per liter or parts per million)

February 25

	Sample 1	Sample 2	Sample 3	Sample 4	Sample 5	Sample 6
	End of Pipe	Prod'd H2O	I-TP-1	I-TP-2	I-TP-4	I-TP-3
<u>Organics:</u>						
Ethylbenzene	.044	.04	ND	.005	ND	ND
Benzene	.0088	.104	.220	detected	.021	.100
Metaxylene	.4	.341	.009	.170	.004	ND
Orthoxylene	.084	.073	ND	ND	ND	ND
Paraxylene	.08	.071	.003	.008	ND	ND
Phenols (total)	3.8	.37	.50	.57	.55	NR
Toluene	.135	.137	ND	ND	ND	ND
Unidentified	.02	.022	.03	.322	ND	ND
Lab Detection						
Limit	.025	.050	.010	.001	.002	.001
Compounds detected, but not quantified	aliphatics, 3-carbon benzenes	aliphatics, 3-carbon substituted benzenes,	3-carbon substituted benzenes	22 peaks	7 peaks	1 peak
		1- and 2-methyl-naphthylene				
<u>General Chemistry:</u>						
Alkalinity	1,501.	1,221.	1,201.	734.5	152.	NR
Calcium	12.	10.	16.	88.1	176.3	NR
Chloride	204.4	122.5	199.5	63.	12.1	NR
Hardness(Tot.)	20.	20.	56.	108.	216.	NR
Magnesium	1.9	2.4	9.7	4.8	9.7	NR
Nitrate(N-NO3)	1.92	1.95	10.9	120.0	2.5	NR
pH(@ 20 C. lab)	8.25	8.7	7.7	7.6	7.9	NR
Potassium	2.7	1.5	1.5	2.3	.7	NR
Sodium	805.	791.2	696.9	319.7	36.8	NR
Sulfate	570.	223.	367.	67.	15.8	NR
TDS	1,655.5	1,701.5	1,379.6	603.8	234.3	NR
<u>Metals:</u>						
Arsenic	.142	.232	.241	.145	ND	NR
Barium	1.08	1.264	.539	.853	.055	NR
Cadmium	.004	.003	<.001	.003	.001	NR
Chromium	.03	.023	.036	.016	.009	NR
Lead	.045	.045	.044	.148	.019	NR
Mercury	<.002	<.002	<.002	<.002	<.002	NR
Selenium	.077	.1745	.081	.097	.013	NR
Silver	.007	.0099	.005	<.001	<.001	NR

(continued)

(continued)

ANALYTIC RESULTS FOR WATER SAMPLING IN THE DUNCAN OIL FIELD  
Field Investigation Conducted by M. Zaman et al., February-March 1985

March 18\*

	Sample 1 Produced Water	Sample 2 II-TP-1	Sample 3 II-TP-3
<u>Organics:</u>			
Ethylbenzene	.106	ND	ND
Benzene	detected, <50 ppb	ND	ND
Metaxylene	.39	ND	.059
Orthoxylene	.15	ND	ND
Paraxylene	.08	ND	ND
Toluene	.18	ND	ND
Phenols (total)	NR	NR	NR
Lab Detection Limit	.050	.050	.005
Compounds detected, but not quantified	6- to 8-carbon aliphatics @ 10 ppb	10- to 20-carbon aliphatics @ 100-500 ppb	

(Samples taken March 18, 1985 were not analyzed for general chemistry, metals or nitrates.)

---

ND	Not detected
NR	Sample not analyzed
I-TP-2	Feb. 25 investigation, Test Pit No. 2
II-TP-3	March 18 investigation, Test Pit No. 3
ppb	parts per billion

REPORT TO:



Dept of Water Management  
PO Box 308  
Windsorock Az 86515

LABORATORY

LAB NUMBER

DRG-165/H  
2/27/85

85-0165-B

SLD Users Code No.

ALL CONTAINERS WHICH THIS FORM ACCOMPANIES ARE COLLECTIVELY REFERRED TO AS "SAMPLE".

## CERTIFICATE OF FIELD PERSONNEL

Sample Type: Water ☒ Soil ☐ Other ☐Water Supply and/or Code No. PIT #1City & County DURAN OIL FIELD, SAN JUAN COUNTY, N.M.Collected (date & time) (850225/100) By (name) M. ZAMAN, D. PaynepH=       ; Conductivity=        umho/cm at        °C; Chlorine Residual=       Dissolved Oxygen=        mg/l; Alkalinity=       ; Flow Rate=       

Sampling Location, Methods &amp; Remarks (i.e. odors etc.)

*preserved in cyclohexane*

I certify that the statements in this block accurately reflect the results of my field analyses, observations and activities. Signed       I certify that I witnessed these field analyses, observations and activities and concur with the statements in this block. Signed       Method of Shipment to Laboratory       THIS FORM ACCOMPANIES        septum vials with teflon-lined discs identified as:specimen       ; duplicate       ; triplicate       ; blank(s)       ,and        amber glass jug(s) with teflon-lined cap(s) identified as       ,and        other container(s) (describe)        identified as       .

Containers are marked as follows to indicate preservation (circle):

NP: No preservation; sample stored at room temperature (~20°C).

P-ICE: Sample stored in an ice bath.

P-Na<sub>2</sub>O<sub>3</sub>S<sub>2</sub>: Sample preserved with 3 mg Na<sub>2</sub>O<sub>3</sub>S<sub>2</sub>/40 ml and stored at room temperature.

## CERTIFICATE(S) OF SAMPLE RECEIPT

I (we) certify that this sample was transferred from        to       at (location)        on(date & time)        and that the statements in this block are correct.Disposition of Sample       . Seal(s) Intact: Yes ☐ No ☐.Signature(s)       I (we) certify that this sample was transferred from        to       at (location)        on(date & time)        and that the statements in this block are correct.Disposition of Sample       . Seal(s) Intact: Yes ☐ No ☐.Signature(s)

## ANALYSES REQUESTED

LAB. No.: ORG- 165

PLEASE CHECK THE APPROPRIATE BOXES BELOW TO INDICATE THE TYPE OF ANALYTICAL SCREENS REQUIRED. WHENEVER POSSIBLE LIST SPECIFIC COMPOUNDS SUSPECTED OR REQUIRED.

QUALITATIVE	QUANTITATIVE	PURGEABLE SCREENS	QUALITATIVE	QUANTITATIVE	EXTRACTABLE SCREENS
		ALIPHATIC HYDROCARBON SCREEN			ALIPHATIC HYDROCARBONS
	X	AROMATIC HYDROCARBON SCREEN			CHLORINATED HYDROCARBON PESTICIDES
		HALOGENATED HYDROCARBON SCREEN			CHLOROPHENOXY ACID HERBICIDES
		GAS CHROMATOGRAPH/MASS SPECTROMETER			HYDROCARBON FUEL SCREEN
					ORGANOPHOSPHATE PESTICIDES
					POLYCHLORINATED BIPHENYLS (PCB's)
					POLYNUCLEAR AROMATIC HYDROCARBONS
					TRIAZINE HERBICIDES
		SPECIFIC COMPOUNDS			SPECIFIC COMPOUNDS

REMARKS:

## ANALYTICAL RESULTS

COMPOUND	[PPB]	COMPOUND	[PPB]
GC/MS purgeables			
Benzene	88		
Toluene	135		
Ethylbenzene	44		
p-Xylene	80		
m-Xylene	400		
o-Xylene	84		
		* DETECTION LIMIT	25ppb

REMARKS: Aliphatic compounds & 3 carbon substituted benzenes were also detected but not quantitated.

## CERTIFICATE OF ANALYTICAL PERSONNEL

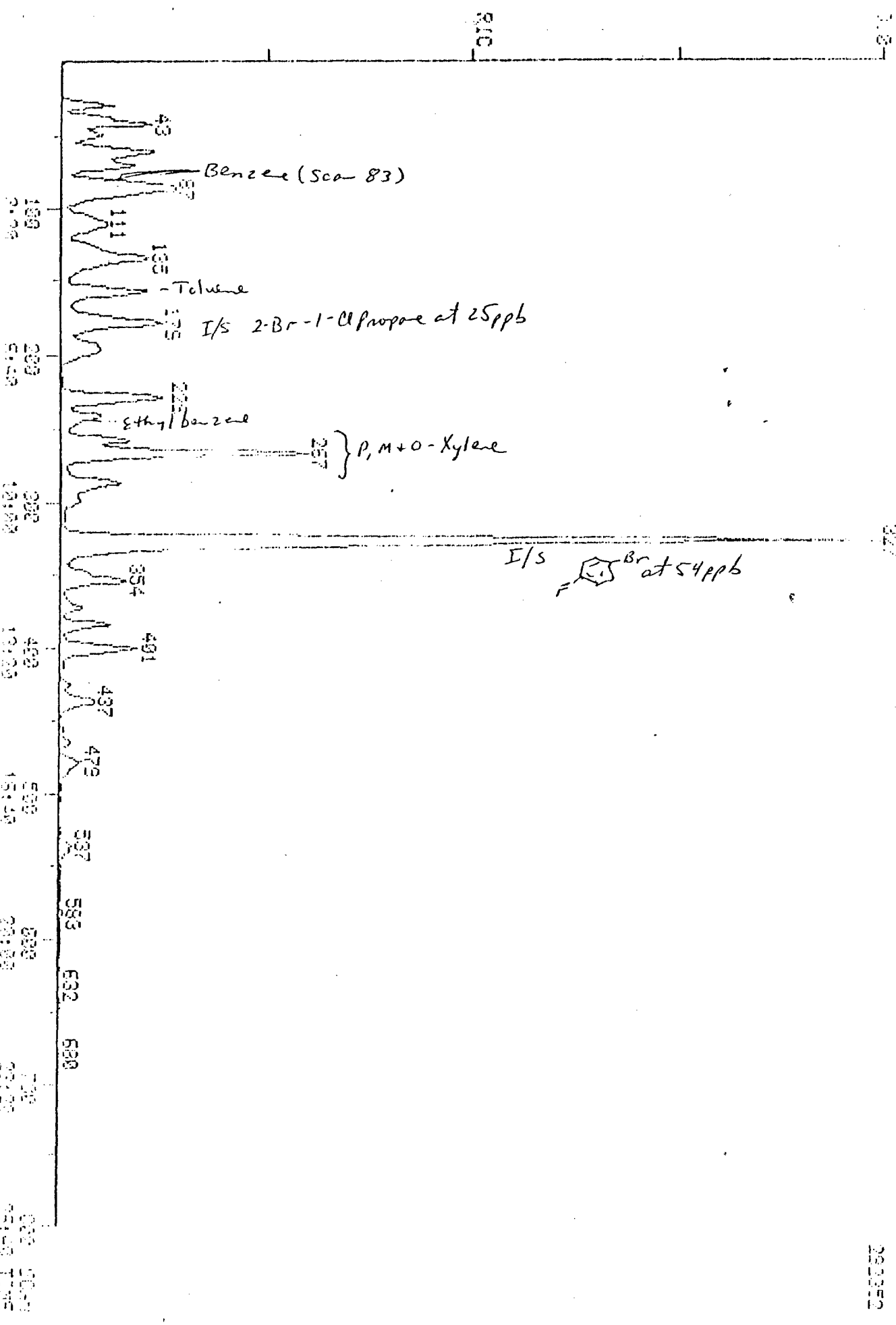
Seal(s) Intact: Yes \_\_\_ NO X. Seal(s) broken by: \_\_\_\_\_ date: \_\_\_\_\_  
 I certify that I followed standard laboratory procedures on handling and analysis of this sample unless otherwise noted and that the statements in this block and the analytical data on this page accurately reflect the analytical results for this sample.  
 Date(s) of analysis: 3/13/85. Analyst's signature: R. Meyer  
 I certify that I have reviewed and concur with the analytical results for this sample and with the statements in this block. Reviewers signature: \_\_\_\_\_

RIC  
 03/13/85 17:22:39  
 SAMPLE: ORG-155  
 RANGE: 0 1, 800 LABEL: N G. 327  
 220:5800 DILUTION

DATA: ORG155 #287  
 CALL: 0811885 #4

SCANS 1 TO 800

2913512



N. H. SCIENTIFIC LABORATORY DIVISION  
QUANTITATION REPORT FILE: DRG165

DATA: DRG165.11  
03/13/85 17:20:00  
SAMPLE: DRG-165 200:5000 DILUTION  
SUBMITTED BY: TDD ANALYST: RFM

AMOUNT=AREA(HGHT) \* REF. AMNT/(REF. AREA(HGHT) \* RESP. FACT)  
RESP. FAC. FROM LIBRARY ENTRY

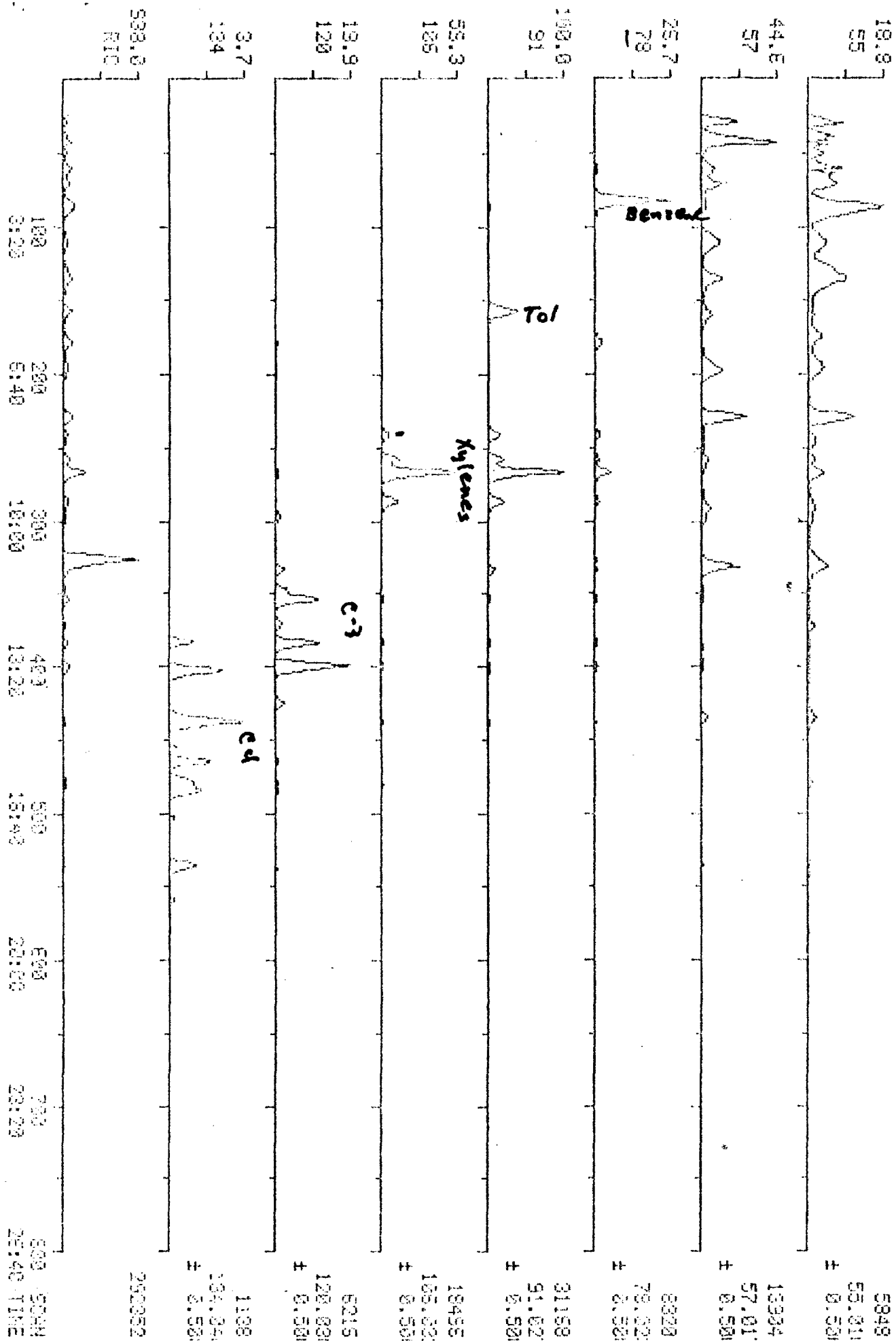
NO NAME  
1 I/S (4-FLUOROBROMOBENZENE)  
2 I/S (2-BROMO-1-CHLOROPROPANE)  
3 BENZENE  
4 1,2-DICHLOROETHANE  
5 TOLUENE  
6 1,2-DIBROMOETHANE  
7 ETHYLBENZENE  
8 P-XYLENE  
9 M-XYLENE  
10 O-XYLENE

NO	M/E	SCAN	TIME	REF	RR1	METH	AREA(HGHT)	AMOUNT	ZTOT	
1	174	327	10:54	1	1.000	A BB	407586.	54.210 UG/L	50.18	<i>25 at 54</i>
2	77	179	5:58	1	0.547	A BB	126628.	20.541 UG/L	19.02	<i>1/3 at 25</i>
3	78	83	2:46	1	0.254	A BB	45424.	3.518 UG/L	3.26	<i>88</i>
4	NOT FOUND									
5	91	157	5:14	1	0.480	A BV	74949.	5.410 UG/L	5.01	<i>135</i>
6	NOT FOUND									
7	91	242	0:04	1	0.740	A BB	27640.	1.774 UG/L	1.64	<i>44</i>
8	91	258	0:36	1	0.787	A BV	32469.	3.187 UG/L	2.95	<i>80</i>
9	91	267	0:54	1	0.817	A VB	167729.	16.036 UG/L	14.85	<i>400</i>
10	91	287	9:34	1	0.878	A BB	36103.	3.345 UG/L	3.10	<i>84</i>

RIC + MASS CHROMATOGRAMS  
 03/13/85 17:20:00  
 SAMPLE: ORG-165 200:5000 DILUTION  
 RANGE: 6 1, 366 LABEL: N 0, 4.0 QUAN: P 0, 1.0 BASE: U 20, 3

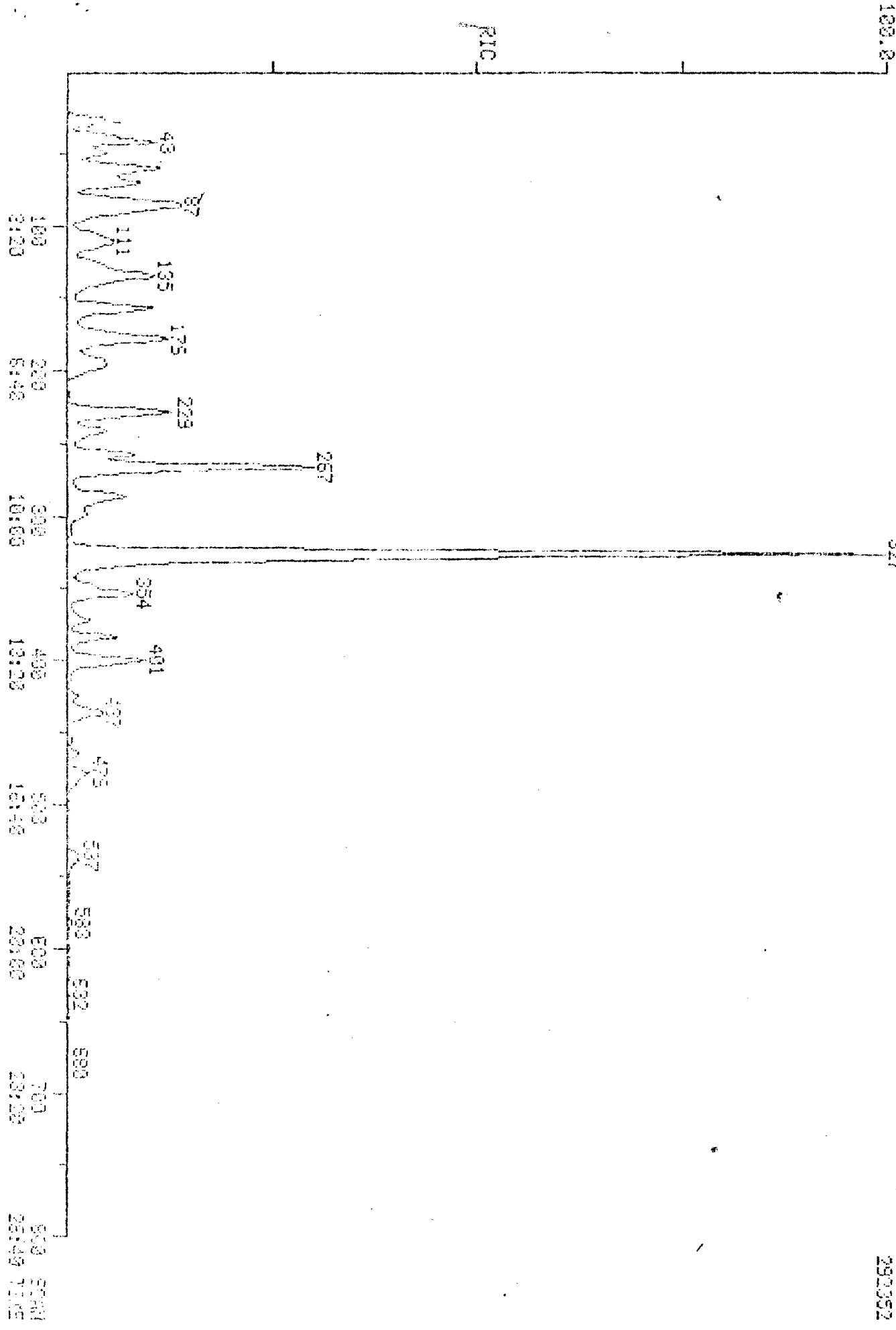
DATA: ORG165 #297  
 CELL: 0311685 #4

SCANS 1 TO 932





RIC  
 03/13/85 17:20:00  
 SAMPLE: CRG-165 200:5000 DILUTION  
 RANGE: 0 1, 800 LABEL: N 0, 4, 8 QUEN: A 0, 1.0 BASE: U 20, 3  
 DATA: CRG165 #287  
 CALL: C011685 #4  
 SCANS 1 TO 800  
 292352



## WATER CHEMICAL ANALYSIS

## NAVAJO TRIBAL UTILITY AUTHORITY

OWNER COPY

SAMPLE NO. 6045SAMPLE LOCATION DUNCAN FIELD PIT #1DATE COLLECTED 2-25-85DATE RECEIVED 2-26-85COLLECTED BY DON PAYEDATE OF FINAL ANALYSIS 3-29-85ADDRESS IHSTECHNICIAN ABMEQ.

TEST	PARAMETER	METHOD	RESULTS	mg/l
✓	ALKALINITY <u>CO<sub>3</sub> 18.03 / HCO<sub>3</sub> 15.73</u>	TITRAMETRIC	<u>1,501.0 mg/l CaCO<sub>3</sub></u>	
✓	CALCIUM <u>.24</u>	TITRAMETRIC OR AA	<u>12.0 mg/l CaCO<sub>3</sub></u>	75-200
✓	CHLORIDE <u>5.76</u>	TITRAMETRIC	<u>204.7 mg/l Cl</u>	250
✓	TOTAL HARDNESS	TITRAMETRIC	<u>20.0 mg/l CaCO<sub>3</sub></u>	500
✓	MAGNESIUM <u>.16</u>	CALCULATED OR AA		
	MANGANESE	SPECTROPHOTOMETRIC OR AA		0.05
	IRON	SPECTROPHOTOMETRIC OR AA		0.3
✓	pH	ELECTRODE	<u>8.25 @ 20C</u>	6.5-8.5
	PHOSPHATE	SPECTROPHOTOMETRIC		
✓	POTASSIUM <u>.07</u>	FLAME PHOTOMETER	<u>2.7 mg/l K</u>	1000-2000
✓	SODIUM <u>35.0</u>	FLAME PHOTOMETER	<u>805.0 mg/l Na</u>	
✓	SULFATE <u>11.97</u>	TITRAMETRIC <u>turbidity</u>	<u>570.0 mg/l SO<sub>4</sub></u>	250
✓	TOTAL DISSOLVED SOLIDS	ELECTRODE	<u>1,655.5 mg/l CaCO<sub>3</sub></u>	500
	TURBIDITY	NEPHELOMETER		
	FLUORIDE	ELECTRODE		1.4

SAMPLE LOCATION \_\_\_\_\_DATE RECEIVED \_\_\_\_\_COLLECTED BY \_\_\_\_\_DATE OUT 3-29-85ADDRESS \_\_\_\_\_TECHNICIAN FH

TEST	PARAMETER	METHOD	RESULTS	MCL
X	ARSENIC	ATOMIC ABSORPTION	<u>.1436</u>	0.05
X	BARIUM	ATOMIC ABSORPTION	<u>1.08</u>	1.0
X	CADMIUM	ATOMIC ABSORPTION	<u>.004</u>	0.01
X	CHROMIUM	ATOMIC ABSORPTION	<u>.0300</u>	0.05
	IRON	ATOMIC ABSORPTION		N/A
X	LEAD <u>.2</u>	ATOMIC ABSORPTION	<u>.0447</u>	0.05
	MANGANESE	ATOMIC ABSORPTION		N/A
	MERCURY	FLAMELESS ATOMIC ABSORPTION	<u>2.002</u>	0.002
X	SELENIUM	ATOMIC ABSORPTION	<u>.0765</u>	0.01
X	SILVER	ATOMIC ABSORPTION	<u>.0069</u>	0.05
	NITRATE (AsN)	CADMIUM REDUCTION		10.0
	FLUORIDE	ELECTRODE		1.4

FORM NO. 5460 (P) M

REV 8-83

ADDRESS +H.STECHNICIAN AB

TEST	PARAMETER	METHOD	RESULTS	MCL
	ARSENIC	ATOMIC ABSORPTION		0.05
	BARIUM	ATOMIC ABSORPTION		1.0
	CADMIUM	ATOMIC ABSORPTION		0.01
	CHROMIUM	ATOMIC ABSORPTION		0.05
	IRON	ATOMIC ABSORPTION		N/A
	LEAD	ATOMIC ABSORPTION		0.05
	MANGANESE	ATOMIC ABSORPTION		N/A
	MERCURY	FLAMELESS ATOMIC ABSORPTION		0.002
	SELENIUM	ATOMIC ABSORPTION		0.01
	SILVER	ATOMIC ABSORPTION		0.05
✓	NITRATE (AsN)	CADMIUM REDUCTION <u>Probe</u>	<u>1.92 mg/l NO<sub>3</sub>-N</u>	10.0
	FLUORIDE	ELECTRODE		1.4

REMARKS  
FORM NO. 5459

REPORT TO:



LABORATORY \_\_\_\_\_

LAB NUMBER DRG-166 #  
2/27/85

85-0166-B

LD Users Code No. \_\_\_\_\_

ALL CONTAINERS WHICH THIS FORM ACCOMPANIES ARE COLLECTIVELY REFERRED TO AS "SAMPLE".

## CERTIFICATE OF FIELD PERSONNEL

Sample Type: Water ☒ Soil ☐ Other \_\_\_\_\_Water Supply and/or Code No. PIT #2City & County DUNCAN OIL FIELD, SAN JUAN COUNTY, N.M.Collected (date & time) 050225(1300) By (name) ZAMANI, M., Payne

pH= \_\_\_\_\_; Conductivity= \_\_\_\_\_ umho/cm at \_\_\_\_\_ °C; Chlorine Residual= \_\_\_\_\_

Dissolved Oxygen= \_\_\_\_\_ mg/l; Alkalinity= \_\_\_\_\_; Flow Rate= \_\_\_\_\_

Sampling Location, Methods &amp; Remarks (i.e. odors etc.)

preserved w cyclohexane

I certify that the statements in this block accurately reflect the results of my field analyses, observations and activities. Signed \_\_\_\_\_

I certify that I witnessed these field analyses, observations and activities and concur with the statements in this block. Signed \_\_\_\_\_

Method of Shipment to Laboratory \_\_\_\_\_

THIS FORM ACCOMPANIES \_\_\_\_\_ septum vials with teflon-lined discs identified as: specimen \_\_\_\_\_; duplicate \_\_\_\_\_; triplicate \_\_\_\_\_; blank(s) \_\_\_\_\_,

and \_\_\_\_\_ amber glass jug(s) with teflon-lined cap(s) identified as \_\_\_\_\_,

and \_\_\_\_\_ other container(s) (describe) \_\_\_\_\_ identified as \_\_\_\_\_.

Containers are marked as follows to indicate preservation (circle):

NP: No preservation; sample stored at room temperature (~20°C).

P-ICE: Sample stored in an ice bath.

P-Na<sub>2</sub>O<sub>3</sub>S<sub>2</sub>: Sample preserved with 3 mg Na<sub>2</sub>O<sub>3</sub>S<sub>2</sub>/40 ml and stored at room temperature.

## CERTIFICATE(S) OF SAMPLE RECEIPT

I (we) certify that this sample was transferred from \_\_\_\_\_ to

\_\_\_\_\_ at (location) \_\_\_\_\_ on

(date &amp; time) \_\_\_\_\_ and that the statements in this block are correct.

Disposition of Sample \_\_\_\_\_. Seal(s) Intact: Yes ☐ No ☐.

Signature(s) \_\_\_\_\_

I (we) certify that this sample was transferred from \_\_\_\_\_ to

\_\_\_\_\_ at (location) \_\_\_\_\_ on

(date &amp; time) \_\_\_\_\_ and that the statements in this block are correct.

Disposition of Sample \_\_\_\_\_. Seal(s) Intact: Yes ☐ No ☐.

Signature(s) \_\_\_\_\_

## ANALYSES REQUESTED

LAB. No.: ORG- 166

PLEASE CHECK THE APPROPRIATE BOXES BELOW TO INDICATE THE TYPE OF ANALYTICAL SCREENS REQUIRED. WHENEVER POSSIBLE LIST SPECIFIC COMPOUNDS SUSPECTED OR REQUIRED.

QUALITATIVE	QUANTITATIVE	PURGEABLE SCREENS	QUALITATIVE	QUANTITATIVE	EXTRACTABLE SCREENS
		ALIPHATIC HYDROCARBON SCREEN			ALIPHATIC HYDROCARBONS
	X	AROMATIC HYDROCARBON SCREEN			CHLORINATED HYDROCARBON PESTICIDES
		HALOGENATED HYDROCARBON SCREEN			CHLOROPHENOXY ACID HERBICIDES
		GAS CHROMATOGRAPH/MASS SPECTROMETER			HYDROCARBON FUEL SCREEN
					ORGANOPHOSPHATE PESTICIDES
					POLYCHLORINATED BIPHENYLS (PCB's)
					POLYNUCLEAR AROMATIC HYDROCARBONS
					TRIAZINE HERBICIDES
		SPECIFIC COMPOUNDS			SPECIFIC COMPOUNDS

REMARKS:

## ANALYTICAL RESULTS

COMPOUND	[PPB]	COMPOUND	[PPB]
GC/MS Results			
Benzene	104		
<del>Toluene</del>	<del>N.D. ppm</del>		
Toluene	137		
Ethyl benzene	40		
p-Xylene	71		
m-Xylene	341		
o-Xylene	73		
		* DETECTION LIMIT	50 ppb

REMARKS: Aliphatic compounds, 3 carbon substituted benzenes and 1- and 2-methylnaphthylene were also detected but not quantitated.

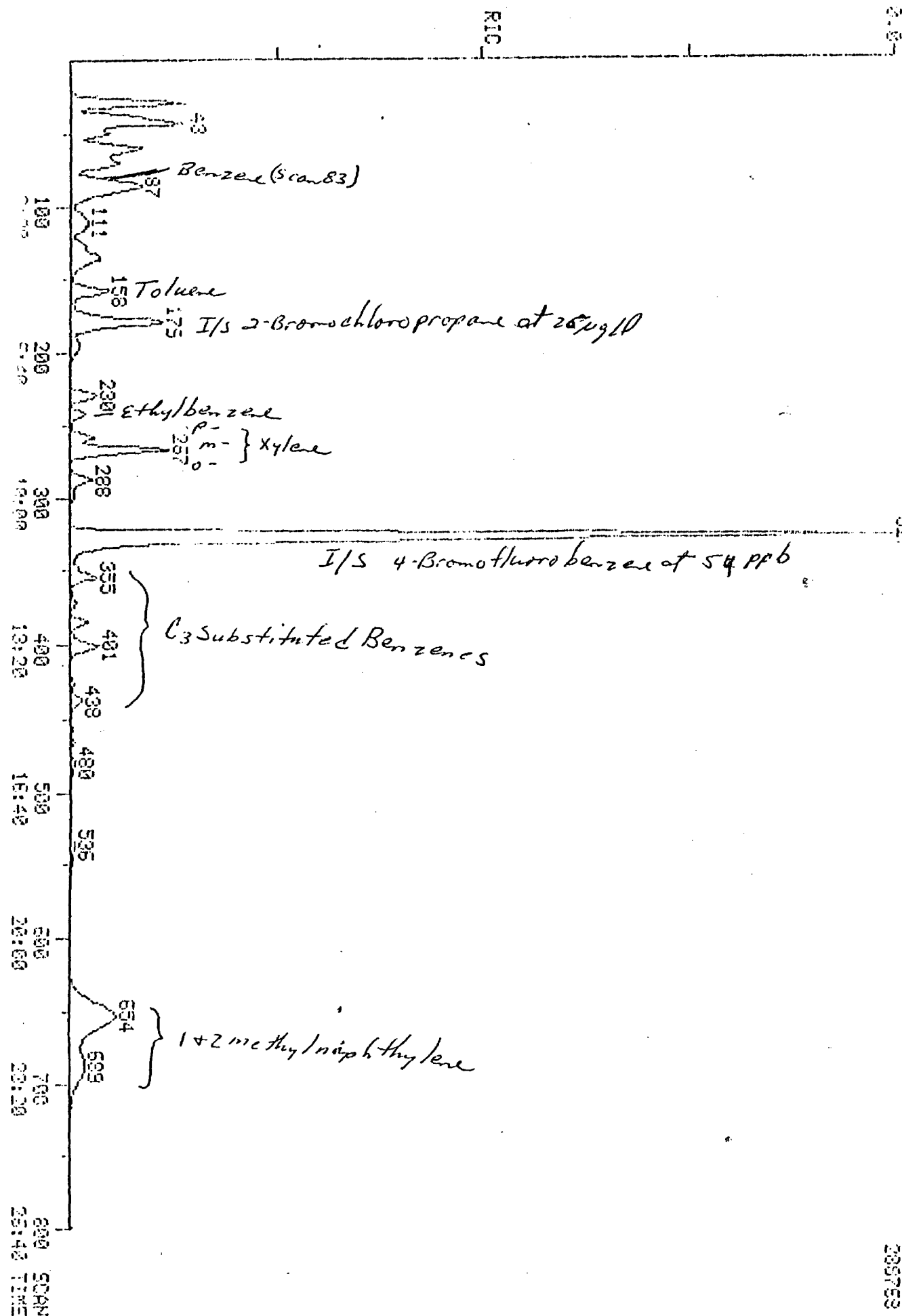
## CERTIFICATE OF ANALYTICAL PERSONNEL

Seal(s) Intact: Yes \_\_\_ NO X . Seal(s) broken by: \_\_\_\_\_ date: \_\_\_\_\_  
 I certify that I followed standard laboratory procedures on handling and analysis of this sample unless otherwise noted and that the statements in this block and the analytical data on this page accurately reflect the analytical results for this sample.  
 Date(s) of analysis: 3/13/85 . Analyst's signature: J. Meyerheim  
 I certify that I have reviewed and concur with the analytical results for this sample and with the statements in this block. Reviewers signature: \_\_\_\_\_

RIC  
 03/13/85 15:28:00  
 SAMPLE: ORG155 (100:5000 DILUTION)  
 RANGE: 0 1.000 LABEL: N 0.20  
 DATA: ORG155 #287  
 CALL: C011685 #4  
 QUAN: A 0.1.0 BASE: U 20. 3

SCANS 1 TO 800

285753



N.M. SCIENTIFIC LABORATORY DIVISION  
QUANTITATION REPORT FILE: ORG166

DATA: ORG166.11  
03/13/85 15:20:00  
SAMPLE: ORG166 100:5000 DILUTION  
SUBMITTED BY: EID ANALYST: RFM

AMOUNT=AREA(HIGHT) \* REF. AMNT/(REF. AREA(HIGHT) \* RESP. FACT)  
RESP. FAC. FROM LIBRARY ENTRY

NO NAME  
1 I/S (4-FLUOROBROMOBENZENE)  
2 I/S (2-BROMO-1-CHLOROPROPANE)  
3 BENZENE  
4 1,2-DICHLOROETHANE  
5 TOLUENE  
6 1,2-DIBROMOETHANE  
7 ETHYLBENZENE  
8 P-XYLENE  
9 M-XYLENE  
10 O-XYLENE

*This is a set of compounds we look for  
in gasoline contamination of water cases.*

NO	M/E	SCAN	TIME	REF	RR1	METH	AREA(HIGHT)	AMOUNT	%TOT	*50
1	174	327	10:54	1	1.000	A BB	397879.	54.210 UG/L	59.45	I/Sat 5
2	77	179	5:58	1	0.547	A BB	131186.	21.799 UG/L	23.91	I/Sat 25
3	78	83	2:46	1	0.254	A BB	23994.	1.904 UG/L	2.09	104
4	NOT FOUND									
5	91	158	5:16	1	0.483	A BB	36787.	2.735 UG/L	3.00	137
6	NOT FOUND									
7	91	242	8:04	1	0.740	A BB	12327.	0.910 UG/L	0.89	40
8	91	258	3:36	1	0.789	A BV	14214.	1.429 UG/L	1.57	71
9	91	267	8:54	1	0.817	A VB	69713.	6.823 UG/L	7.49	341
10	91	288	7:36	1	0.881	A BB	15455.	1.467 UG/L	1.61	73

OWNER COPY

SAMPLE NO. 6046

SAMPLE LOCATION DUNCAN FIELD Pit #2 DATE COLLECTED 2-25-85

DATE RECEIVED 2-26-85 COLLECTED BY DON PAYNE

DATE OF FINAL ANALYSIS 3-29-85 ADDRESS FHS

TECHNICIAN AB

TEST	PARAMETER	METHOD	RESULTS	mg/l
✓	ALKALINITY <u>CO<sub>3</sub> 8.70 / HCO<sub>3</sub> 15.73</u>	TITRAMETRIC	<u>1221.0 mg/L CaCO<sub>3</sub></u>	
✓	CALCIUM <u>.20</u>	TITRAMETRIC OR AA	<u>10.0 mg/L CaCO<sub>3</sub></u>	75-200
✓	CHLORIDE <u>3.45</u>	TITRAMETRIC	<u>122.5 mg/L Cl<sub>-</sub></u>	250
✓	TOTAL HARDNESS	TITRAMETRIC	<u>20.0 mg/L CaCO<sub>3</sub></u>	500
✓	MAGNESIUM <u>.20</u>	CALCULATED OR AA	<u>2.4 mg/L Mg</u>	9.8
	MANGANESE	SPECTROPHOTOMETRIC OR AA		0.05
	IRON	SPECTROPHOTOMETRIC OR AA		0.3
✓	pH	ELECTRODE	<u>8.7 @ 20C</u>	6.5-8.5
	PHOSPHATE	SPECTROPHOTOMETRIC		
✓	POTASSIUM <u>.04</u>	FLAME PHOTOMETER	<u>1.5 mg/L K</u>	1000-2000
✓	SODIUM <u>34.4</u>	FLAME PHOTOMETER	<u>791.2 mg/L Na</u>	
✓	SULFATE <u>4.64</u>	TITRAMETRIC	<u>223.0 mg/L SO<sub>4</sub></u>	250
✓	TOTAL DISSOLVED SOLIDS	ELECTRODE	<u>1701.5 mg/L CaCO<sub>3</sub></u>	500
	TURBIDITY	NEPHELOMETER		
	FLUORIDE	ELECTRODE		1.4

DATE OUT 3-21-85 ADDRESS FHS

TECHNICIAN AB

TEST	PARAMETER	METHOD	RESULTS	MCL
	ARSENIC	ATOMIC ABSORPTION		0.05
	BARIUM	ATOMIC ABSORPTION		1.0
	CADMIUM	ATOMIC ABSORPTION		0.01
	CHROMIUM	ATOMIC ABSORPTION		0.05
	IRON	ATOMIC ABSORPTION		N/A
	LEAD	ATOMIC ABSORPTION		0.05
	MANGANESE	ATOMIC ABSORPTION		N/A
	MERCURY	FLAMELESS ATOMIC ABSORPTION		0.002
	SELENIUM	ATOMIC ABSORPTION		0.01
	SILVER	ATOMIC ABSORPTION		0.05
✓	NITRATE (AsN)	CADMIUM REDUCTION	<u>1.95 mg/L NO<sub>3</sub>-N</u>	10.0
	FLUORIDE	ELECTRODE		1.4

FORM NO. 5460 (P) M

REV 8-83

DATE RECEIVED \_\_\_\_\_

DATE OUT 3-29-85 COLLECTED BY \_\_\_\_\_

TECHNICIAN AB ADDRESS \_\_\_\_\_

TEST	PARAMETER	METHOD	RESULTS	MCL
X	ARSENIC	ATOMIC ABSORPTION		0.05
X	BARIUM	ATOMIC ABSORPTION	<u>.2315</u>	1.0
X	CADMIUM	ATOMIC ABSORPTION	<u>1.2636</u>	0.01
X	CHROMIUM	ATOMIC ABSORPTION	<u>.0026</u>	0.05
	IRON	ATOMIC ABSORPTION	<u>.0227</u>	N/A
+	LEAD	ATOMIC ABSORPTION	<u>.0451</u>	0.05
	MANGANESE	ATOMIC ABSORPTION		N/A
	MERCURY	FLAMELESS ATOMIC ABSORPTION	<u>2.002</u>	0.002
X	SELENIUM	ATOMIC ABSORPTION	<u>.1745</u>	0.01
X	SILVER	ATOMIC ABSORPTION	<u>.0099</u>	0.05
	NITRATE (AsN)	CADMIUM REDUCTION		10.0
	FLUORIDE	ELECTRODE		1.4

FORM NO. 5460 (P) M

REV 8-83

REPORT TO: \_\_\_\_\_



LABORATORY \_\_\_\_\_

LAB NUMBER ORG-167-A  
2/27/85

85-0167-B

Users Code No. \_\_\_\_\_

ALL CONTAINERS WHICH THIS FORM ACCOMPANIES ARE COLLECTIVELY REFERRED TO AS "SAMPLE".

## CERTIFICATE OF FIELD PERSONNEL

Sample Type: Water ☒ Soil ☐ Other \_\_\_\_\_Water Supply and/or Code No. PMT #3City & County DURAN OIL FIELDCollected (date & time) 850225/130 By (name) M. ZAMAN, D. Payne

pH= \_\_\_\_\_; Conductivity= \_\_\_\_\_ umho/cm at \_\_\_\_\_ °C; Chlorine Residual= \_\_\_\_\_

Dissolved Oxygen= \_\_\_\_\_ mg/l; Alkalinity= \_\_\_\_\_; Flow Rate= \_\_\_\_\_

Sampling Location, Methods &amp; Remarks (i.e. odors etc.)

cyclohexane preservative

I certify that the statements in this block accurately reflect the results of my field analyses, observations and activities. Signed \_\_\_\_\_

I certify that I witnessed these field analyses, observations and activities and concur with the statements in this block. Signed \_\_\_\_\_

Method of Shipment to Laboratory \_\_\_\_\_

THIS FORM ACCOMPANIES \_\_\_\_\_ septum vials with teflon-lined discs identified as:

specimen \_\_\_\_\_; duplicate \_\_\_\_\_; triplicate \_\_\_\_\_; blank(s) \_\_\_\_\_,

and \_\_\_\_\_ amber glass jug(s) with teflon-lined cap(s) identified as \_\_\_\_\_,

and \_\_\_\_\_ other container(s) (describe) \_\_\_\_\_ identified as \_\_\_\_\_.

Containers are marked as follows to indicate preservation (circle):

NP: No preservation; sample stored at room temperature (~20°C).

P-ICE: Sample stored in an ice bath.

P-Na<sub>2</sub>O<sub>3</sub>S<sub>2</sub>: Sample preserved with 3 mg Na<sub>2</sub>O<sub>3</sub>S<sub>2</sub>/40 ml and stored at room temperature.

## CERTIFICATE(S) OF SAMPLE RECEIPT

I (we) certify that this sample was transferred from \_\_\_\_\_ to

\_\_\_\_\_ at (location) \_\_\_\_\_ on

(date &amp; time) \_\_\_\_\_ and that the statements in this block are correct.

Disposition of Sample \_\_\_\_\_. Seal(s) Intact: Yes ☐ No ☐.

Signature(s) \_\_\_\_\_

I (we) certify that this sample was transferred from \_\_\_\_\_ to

\_\_\_\_\_ at (location) \_\_\_\_\_ on

(date &amp; time) \_\_\_\_\_ and that the statements in this block are correct.

Disposition of Sample \_\_\_\_\_. Seal(s) Intact: Yes ☐ No ☐.

Signature(s) \_\_\_\_\_



## ANALYSES REQUESTED

LAB. No.: ORG- 167

PLEASE CHECK THE APPROPRIATE BOXES BELOW TO INDICATE THE TYPE OF ANALYTICAL SCREENS REQUIRED. WHENEVER POSSIBLE LIST SPECIFIC COMPOUNDS SUSPECTED OR REQUIRED.

QUALITATIVE	QUANTITATIVE	PURGEABLE SCREENS	QUALITATIVE	QUANTITATIVE	EXTRACTABLE SCREENS
		ALIPHATIC HYDROCARBON SCREEN			ALIPHATIC HYDROCARBONS
	X	AROMATIC HYDROCARBON SCREEN			CHLORINATED HYDROCARBON PESTICIDES
		HALOGENATED HYDROCARBON SCREEN			CHLOROPHENOXY ACID HERBICIDES
		GAS CHROMATOGRAPH/MASS SPECTROMETER			HYDROCARBON FUEL SCREEN
					ORGANOPHOSPHATE PESTICIDES
					POLYCHLORINATED BIPHENYLS (PCB's)
					POLYNUCLEAR AROMATIC HYDROCARBONS
					TRIAZINE HERBICIDES
		SPECIFIC COMPOUNDS			SPECIFIC COMPOUNDS

REMARKS:

## ANALYTICAL RESULTS

COMPOUND	[PPB]	COMPOUND	[PPB]
GC/MS Purgeables			
Cyclohexane	(Preservative)		
Benzene	220		
p-Xylene	3		
m-Xylene	9		
		* DETECTION LIMIT	~ 10 ppb

REMARKS: Some 3 carbon substituted benzenes were also detected but not quantitated

## CERTIFICATE OF ANALYTICAL PERSONNEL

Seal(s) Intact: Yes \_\_\_ NO X. Seal(s) broken by: \_\_\_\_\_ date: \_\_\_\_\_

I certify that I followed standard laboratory procedures on handling and analysis of this sample unless otherwise noted and that the statements in this block and the analytical data on this page accurately reflect the analytical results for this sample.

Date(s) of analysis: 3/14/85. Analyst's signature: [Signature]

I certify that I have reviewed and concur with the analytical results for this sample and with the statements in this block. Reviewers signature: \_\_\_\_\_

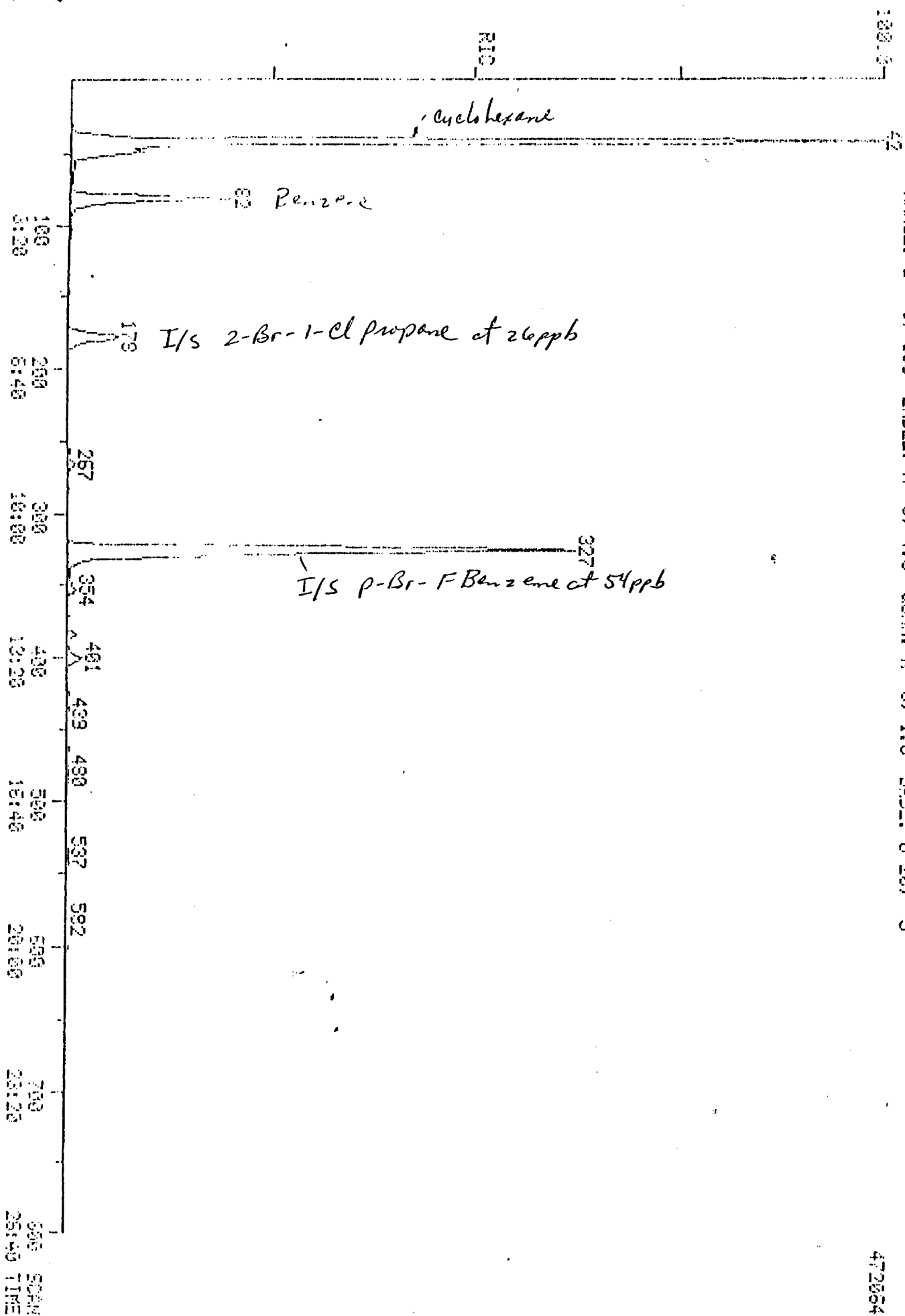
RIC  
03/14/85 11:03:00  
SAMPLE: ORG-157 DILUTED  
RANGE: G 1, 200 LABEL: N 3, 4.0 QLEN: A 0, 1.0 PORE: U 20, 3

\*10  
500:5000

DATA: ORG157 #287  
CALL: C011685 #4

SCANS 1 TO 800

472064



M. H. SCIENTIFIC LABORATORY DIVISION  
QUANTITATION REPORT . FILE: ORG167

DATA: ORG167.11

03/14/85 11:00:00

SAMPLE: ORG-167 DILUTED 500:5000

SUBMITTED BY: EDD

ANALYST: RFM

AMOUNT=AREA(HGHT) \* REF. AMNT/(REF. AREA(HGHT)\* RESP. FACT)  
RESP. FAC. FROM LIBRARY ENTRY

NO. NAME

- 1 I/S (4-FLUOROBROMOBENZENE)
- 2 I/S (2-BROMO-1-CHLOROPROPANE)
- 3 BENZENE
- 4 1,2-DICHLOROETHANE
- 5 TOLUENE
- 6 1,2-DIBROMOETHANE
- 7 ETHYLBENZENE
- 8 P-XYLENE
- 9 M-XYLENE
- 10 O-XYLENE

NO	M/E	SCAN	TIME	REF	RRT	METH	AREA(HGHT)	AMOUNT	ZTOT	*10
1	174	327	10:54	1	1.000	A BB	407671.	54.210 UG/L	56.06	
2	77	170	5:56	1	0.544	A BB	119718.	19.416 UG/L	20.08	
3	78	82	2:44	1	0.251	A BB	281330.	21.783 UG/L	22.53	220
4	NOT FOUND									
5	NOT FOUND									
6	NOT FOUND									
7	91	241	8:02	1	0.737	A BB	1184.	0.076 UG/L	0.08	
8	91	250	8:36	1	0.789	A BV	3041.	0.298 UG/L	0.31	3
9	91	267	8:54	1	0.817	A VB	7657.	0.723 UG/L	0.95	9
10	NOT FOUND									



SAMPLE NO. 6047  
 SAMPLE LOCATION DUNCAN FIELD PH #3 DATE COLLECTED 2-25-85  
 DATE RECEIVED 2-26-85 COLLECTED BY DON PAYNE  
 DATE OF FINAL ANALYSIS 3-29-85 ADDRESS FHS  
 TECHNICIAN AB

TEST	PARAMETER <i>mg/l</i>	METHOD	RESULTS	mg/l
✓	ALKALINITY <i>CO<sub>3</sub> 15.34 / HCO<sub>3</sub> 12.13</i>	TITRAMETRIC	<i>1,201.0 mg/L CaCO<sub>3</sub></i>	
✓	CALCIUM <i>.32</i>	TITRAMETRIC OR AA	<i>162.0 mg/L CaCO<sub>3</sub></i>	75-200
✓	CHLORIDE <i>5.62</i>	TITRAMETRIC	<i>199.5 mg/L Cl</i>	250
✓	TOTAL HARDNESS	TITRAMETRIC	<i>56.0 mg/L CaCO<sub>3</sub></i>	500
✓	MAGNESIUM <i>.80</i>	CALCULATED OR AA	<i>9.7 mg/L CaCO<sub>3</sub></i>	
	MANGANESE	SPECTROPHOTOMETRIC OR AA		0.05
	IRON	SPECTROPHOTOMETRIC OR AA		0.3
✓	pH	ELECTRODE	<i>7.7 @ 20C</i>	6.5-8.5
	PHOSPHATE	SPECTROPHOTOMETRIC		
✓	POTASSIUM <i>.04</i>	FLAME PHOTOMETER	<i>1.5 mg/L K</i>	1000-2000
✓	SODIUM <i>30.3</i>	FLAME PHOTOMETER	<i>696.9 mg/L Na</i>	
✓	SULFATE <i>7.64</i>	TITRAMETRIC <i>turbidity</i>	<i>367.0 mg/L SO<sub>4</sub></i>	250
✓	TOTAL DISSOLVED SOLIDS	ELECTRODE	<i>1,379.6 mg/L CaCO<sub>3</sub></i>	500
	TURBIDITY	NEPHELOMETER		
	FLUORIDE	ELECTRODE		1.4

DATE RECEIVED \_\_\_\_\_ COLLECTED BY \_\_\_\_\_  
 DATE OUT 3-29-85 ADDRESS \_\_\_\_\_  
 TECHNICIAN AB

TEST	PARAMETER	METHOD	RESULTS	MCL
X	ARSENIC	ATOMIC ABSORPTION	<i>.2404</i>	0.05
X	BARIUM	ATOMIC ABSORPTION	<i>.5385</i>	1.0
X	CADMIUM	ATOMIC ABSORPTION	<i>&lt;.001</i>	0.01
X	CHROMIUM	ATOMIC ABSORPTION	<i>.036</i>	0.05
	IRON	ATOMIC ABSORPTION		N/A
X	LEAD	ATOMIC ABSORPTION	<i>.0435</i>	0.05
	MANGANESE	ATOMIC ABSORPTION		N/A
	MERCURY	FLAMELESS ATOMIC ABSORPTION	<i>2.002</i>	0.002
X	SELENIUM	ATOMIC ABSORPTION	<i>.0810</i>	0.01
X	SILVER	ATOMIC ABSORPTION	<i>.0052</i>	0.05
	NITRATE (AsN)	CADMIUM REDUCTION		10.0
	FLUORIDE	ELECTRODE		1.4

FORM NO. 5460 (P) M

REV 8-82

DATE OUT 3-21-85 ADDRESS FHS  
 TECHNICIAN AB

TEST	PARAMETER	METHOD	RESULTS	MCL
	ARSENIC	ATOMIC ABSORPTION		0.05
	BARIUM	ATOMIC ABSORPTION		1.0
	CADMIUM	ATOMIC ABSORPTION		0.01
	CHROMIUM	ATOMIC ABSORPTION		0.05
	IRON	ATOMIC ABSORPTION		N/A
	LEAD	ATOMIC ABSORPTION		0.05
	MANGANESE	ATOMIC ABSORPTION		N/A
	MERCURY	FLAMELESS ATOMIC ABSORPTION		0.002
	SELENIUM	ATOMIC ABSORPTION		0.01
	SILVER	ATOMIC ABSORPTION		0.05
✓	NITRATE (AsN)	CADMIUM REDUCTION	<i>10.9 mg/L NO<sub>3</sub>-N</i>	10.0
	FLUORIDE	ELECTRODE		1.4

FORM NO. 5460 (P) M

REV 8-83

REPORT TO:

MASUD ZAMANLABORATORY STATE LABNAV. WATER RESOURCES DIVLAB NUMBER JRG-168-H  
2/27/85P.O. BOX 137485-0168-BWINDECK ROCK, AZ 865...

Users Code No.

ALL CONTAINERS WHICH THIS FORM ACCOMPANIES ARE COLLECTIVELY REFERRED TO AS "SAMPLE".

## CERTIFICATE OF FIELD PERSONNEL

Sample Type: Water ☒ Soil ☐ Other ☐Water Supply and/or Code No. PIT #4City & County DUNCAN OIL FIELD, SAN JUAN COUNTYCollected (date & time) 850225/1200 By (name) ZAMAN, M. PaynepH=       ; Conductivity=        umho/cm at        °C; Chlorine Residual=       Dissolved Oxygen=        mg/l; Alkalinity=       ; Flow Rate=       

Sampling Location, Methods &amp; Remarks (i.e. odors etc.)

cyclohexane preservativeI certify that the statements in this block accurately reflect the results of my field analyses, observations and activities. Signed       I certify that I witnessed these field analyses, observations and activities and concur with the statements in this block. Signed       Method of Shipment to Laboratory       THIS FORM ACCOMPANIES        septum vials with teflon-lined discs identified as:specimen       ; duplicate       ; triplicate       ; blank(s)       ,and        amber glass jug(s) with teflon-lined cap(s) identified as       ,and        other container(s) (describe)        identified as       .

Containers are marked as follows to indicate preservation (circle):

NP: No preservation; sample stored at room temperature (~20°C).

P-ICE: Sample stored in an ice bath.

P-Na<sub>2</sub>O<sub>3</sub>S<sub>2</sub>: Sample preserved with 3 mg Na<sub>2</sub>O<sub>3</sub>S<sub>2</sub>/40 ml and stored at room temperature.

## CERTIFICATE(S) OF SAMPLE RECEIPT

I (we) certify that this sample was transferred from        to       at (location)        on(date & time)        and that the statements in this block are correct.Disposition of Sample       . Seal(s) Intact: Yes ☐ No ☐.Signature(s)       I (we) certify that this sample was transferred from        to       at (location)        on(date & time)        and that the statements in this block are correct.Disposition of Sample       . Seal(s) Intact: Yes ☐ No ☐.Signature(s)

## ANALYSES REQUESTED

LAB. No.: ORG- 168

PLEASE CHECK THE APPROPRIATE BOXES BELOW TO INDICATE THE TYPE OF ANALYTICAL SCREENS REQUIRED. WHENEVER POSSIBLE LIST SPECIFIC COMPOUNDS SUSPECTED OR REQUIRED.

QUALITATIVE	QUANTITATIVE	PURGEABLE SCREENS	QUALITATIVE	QUANTITATIVE	EXTRACTABLE SCREENS
		ALIPHATIC HYDROCARBON SCREEN			ALIPHATIC HYDROCARBONS
	X	AROMATIC HYDROCARBON SCREEN			CHLORINATED HYDROCARBON PESTICIDES
		HALOGENATED HYDROCARBON SCREEN			CHLOROPHENOXY ACID HERBICIDES
		GAS CHROMATOGRAPH/MASS SPECTROMETER			HYDROCARBON FUEL SCREEN
					ORGANOPHOSPHATE PESTICIDES
					POLYCHLORINATED BIPHENYLS (PCB's)
					POLYNUCLEAR AROMATIC HYDROCARBONS
					TRIAZINE HERBICIDES
		SPECIFIC COMPOUNDS			SPECIFIC COMPOUNDS

REMARKS:

## ANALYTICAL RESULTS

COMPOUND	[PPB]	COMPOUND	[PPB]
Benzene <sup>o</sup>	Not pure Bz	Toluene	now detected
Ethyl-Benzene	5	ortho-xylene	now detected
para-xylene	8		
meta-xylene	170		
halogenated purgables	110 NO detected		
		* DETECTION LIMIT	1.0 µg/L

REMARKS: One peak equivalent to 85 ppb pure benzene. Twenty-two other peaks detected by PID. Sample was not taken provided in a proper septum vial bottle; but in a gallon jug with headspace. Unable to quantitate Benzene due to interference.

## CERTIFICATE OF ANALYTICAL PERSONNEL

Seal(s) Intact: Yes NO X. Seal(s) broken by: \_\_\_\_\_

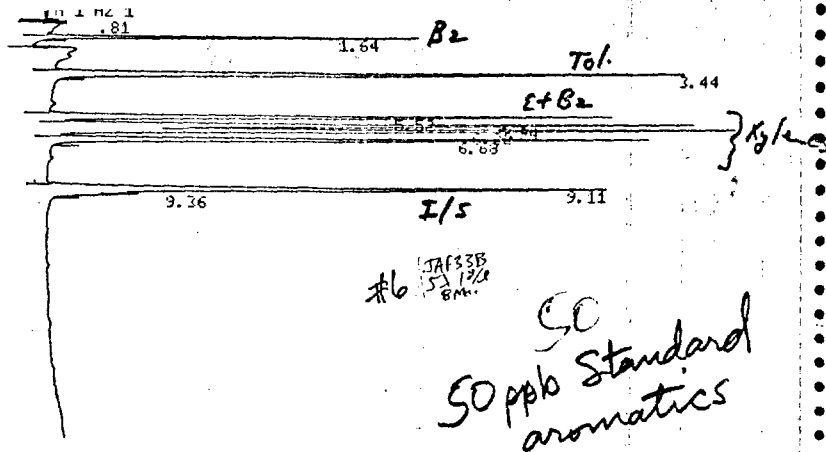
date: \_\_\_\_\_

I certify that I followed standard laboratory procedures on handling and analysis of this sample unless otherwise noted and that the statements in this block and the analytical data on this page accurately reflect the analytical results for this sample.

Date(s) of analysis: 228 7 3-13-85

Analyst's signature: Dr. Fumey

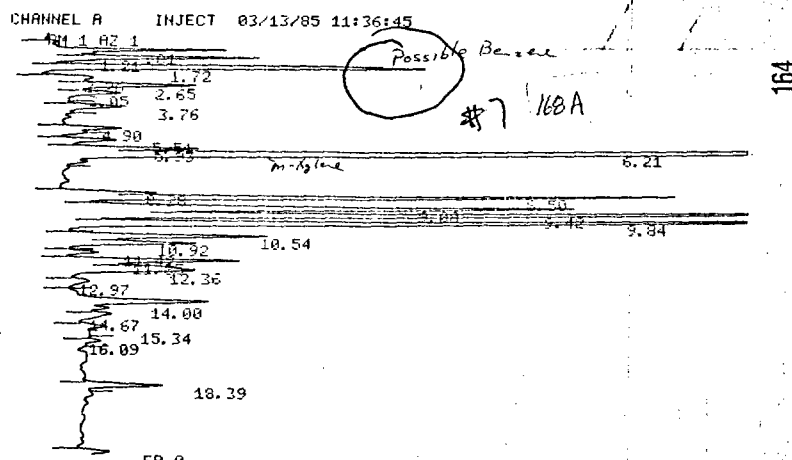
I certify that I have reviewed and concur with the analytical results for this sample and with the statements in this block. Reviewers signature: \_\_\_\_\_



PID 03/13/85 11:09:38 CH= "A" PS= 1.

FILE 1. METHOD 0. RUN 446 INDEX 446

PEAK#	AREA%	RT	AREA BC
1	5.667	1.64	23036 01
2	15.317	3.44	62265 01
3	13.794	5.52	56071 02
4	16.367	5.94	68531 02
5	17.236	6.22	70063 02
6	13.98	6.68	56329 03
7	15.466	9.11	62871 02
8	2.173	9.36	8932 03
TOTAL	100.		406498



PID 03/13/85 11:36:45 CH= "A" PS= 1.

FILE 1. METHOD 0. RUN 447 INDEX 447

PEAK#	AREA%	RT	AREA BC
1	2.079	0.81	20257 02
2	2.62	1.21	25529 02
3	5.302	1.72	51669 02
4	0.841	2.26	8197 02
5	2.232	2.65	21748 02
6	1.005	3.05	9790 03
7	0.666	3.76	6494 01
8	0.428	4.9	4167 01
9	0.718	5.51	6996 02
10	1.34	5.93	13056 02
11	29.862	6.21	291009 03
12	0.886	8.28	8633 02
13	7.085	9.5	69346 02
14	27.640	9.84	24511 02

## WATER CHEMICAL ANALYSIS

## NAVAJO TRIBAL UTILITY AUTHORITY

OWNER COPY

SAMPLE NO. 6048SAMPLE LOCATION DUNKIN Field Pit 4DATE COLLECTED 2-25-85DATE RECEIVED 2-26-85COLLECTED BY DON PAYNEDATE OF FINAL ANALYSIS 3-29-85ADDRESS FHSTECHNICIAN B Mag.

TEST	PARAMETER	METHOD	RESULTS	mg/l
✓	ALKALINITY $\text{CO}_3$ 2.23, $\text{HCO}_3$ 10.94	TITRAMETRIC	734.5 mg/l $\text{CaCO}_3$	
✓	CALCIUM 1.76	TITRAMETRIC OR AA	88.1 mg/l $\text{CaCO}_3$	75-200
✓	CHLORIDE 1.77	TITRAMETRIC	63.0 mg/l $\text{Cl}$	250
✓	TOTAL HARDNESS	TITRAMETRIC	108.0 mg/l $\text{CaCO}_3$	500
✓	MAGNESIUM .40	CALCULATED OR AA	9.8 mg/l $\text{Mg}$	
	MANGANESE	SPECTROPHOTOMETRIC OR AA		0.05
	IRON	SPECTROPHOTOMETRIC OR AA		0.3
✓	pH	ELECTRODE	7.6 @ 20C	6.5-8.5
	PHOSPHATE	SPECTROPHOTOMETRIC		
✓	POTASSIUM .06	FLAME PHOTOMETER	2.3 mg/l $\text{K}$	1000-2000
✓	SODIUM 13.9	FLAME PHOTOMETER	319.7 mg/l $\text{Na}$	
✓	SULFATE 1.39	TITRAMETRIC	67.0 mg/l $\text{SO}_4$	250
✓	TOTAL DISSOLVED SOLIDS	ELECTRODE	603.8 mg/l $\text{CaCO}_3$	500
	TURBIDITY	NEPHELOMETER		
	FLUORIDE	ELECTRODE		1.4

DATE RECEIVED

DATE OUT 3-29-85

ADDRESS

TECHNICIAN FW

TEST	PARAMETER	METHOD	RESULTS	MCL
X	ARSENIC	ATOMIC ABSORPTION	.1451	0.05
X	BARIUM	ATOMIC ABSORPTION	.8533	1.0
X	CADMIUM	ATOMIC ABSORPTION	.0028	0.01
X	CHROMIUM	ATOMIC ABSORPTION	.0157	0.05
	IRON	ATOMIC ABSORPTION		N/A
X	LEAD	ATOMIC ABSORPTION	.1475	0.05
	MANGANESE	ATOMIC ABSORPTION		N/A
+	MERCURY	FLAMELESS ATOMIC ABSORPTION	2.002	0.002
X	SELENIUM	ATOMIC ABSORPTION	.0968	0.01
X	SILVER	ATOMIC ABSORPTION	2.001	0.05
	NITRATE (AsN)	CADMIUM REDUCTION		10.0
	FLUORIDE	ELECTRODE		1.4

FORM NO. 5460 © M

REV 8-83

DATE OUT 3-29-85ADDRESS FHSTECHNICIAN B

TEST	PARAMETER	METHOD	RESULTS	MCL
	ARSENIC	ATOMIC ABSORPTION		0.05
	BARIUM	ATOMIC ABSORPTION		1.0
	CADMIUM	ATOMIC ABSORPTION		0.01
	CHROMIUM	ATOMIC ABSORPTION		0.05
	IRON	ATOMIC ABSORPTION		N/A
	LEAD	ATOMIC ABSORPTION		0.05
	MANGANESE	ATOMIC ABSORPTION		N/A
	MERCURY	FLAMELESS ATOMIC ABSORPTION		0.002
	SELENIUM	ATOMIC ABSORPTION		0.01
	SILVER	ATOMIC ABSORPTION		0.05
✓	NITRATE (AsN)	CADMIUM REDUCTION Probe	120.0 mg/l $\text{NO}_3 - \text{N}$	10.0
	FLUORIDE	ELECTRODE		1.4

FORM NO. 5460 © M

REV 8-83

REMARKS  
FORM NO. 5459





LABORATORY

LAB NUMBER ORG-169-A  
2/27/85

85-0169-B Users Code No.

ALL CONTAINERS WHICH THIS FORM ACCOMPANIES REFERRED TO AS "SAMPLE".

## CERTIFICATE OF FIELD PERSONNEL

Sample Type: Water ☒ Soil ☐ Other ☐Water Supply and/or Code No. PIT #5City & County DUNKAN OIL FIELD, SAN JUAN COUNTY, N.M.Collected (date & time) 8502251200 By (name) ZAMAN, M. PaynepH=       ; Conductivity=        umho/cm at        °C; Chlorine Residual=       Dissolved Oxygen=        mg/l; Alkalinity=       ; Flow Rate=       

Sampling Location, Methods &amp; Remarks (i.e. odors etc.)

cyclohexane preservative ??I certify that the statements in this block accurately reflect the results of my field analyses, observations and activities. Signed       I certify that I witnessed these field analyses, observations and activities and concur with the statements in this block. Signed       Method of Shipment to Laboratory       THIS FORM ACCOMPANIES        septum vials with teflon-lined discs identified as:specimen       ; duplicate       ; triplicate       ; blank(s)       ;and        amber glass jug(s) with teflon-lined cap(s) identified as       ;and        other container(s) (describe)        identified as       .

Containers are marked as follows to indicate preservation (circle):

NP: No preservation; sample stored at room temperature (~20°C).

P-ICE: Sample stored in an ice bath.

P-Na<sub>2</sub>O<sub>3</sub>S<sub>2</sub>: Sample preserved with 3 mg Na<sub>2</sub>O<sub>3</sub>S<sub>2</sub>/40 ml and stored at room temperature.

## CERTIFICATE(S) OF SAMPLE RECEIPT

I (we) certify that this sample was transferred from        to         
       at (location)        on       (date & time)        and that the statements in this block are correct.Disposition of Sample       . Seal(s) Intact: Yes ☐ No ☐.Signature(s)       I (we) certify that this sample was transferred from        to         
       at (location)        on       (date & time)        and that the statements in this block are correct.Disposition of Sample       . Seal(s) Intact: Yes ☐ No ☐.Signature(s)

LAB. No.: ORG-169

THE TYPE OF ANALYTICAL SCREENS  
SUSPECTED OR REQUIRED.

QUALITATIVE	QUANTITATIVE	PURGEABLE SCREENS	QUALITATIVE	QUANTITATIVE	EXTRACTABLE SCREENS
		ALIPHATIC HYDROCARBON SCREEN			ALIPHATIC HYDROCARBONS
	X	AROMATIC HYDROCARBON SCREEN			CHLORINATED HYDROCARBON PESTICIDES
		HALOGENATED HYDROCARBON SCREEN			CHLOROPHENOXY ACID HERBICIDES
		GAS CHROMATOGRAPH/MASS SPECTROMETER			HYDROCARBON FUEL SCREEN
					ORGANOPHOSPHATE PESTICIDES
					POLYCHLORINATED BIPHENYLS (PCB's)
					POLYNUCLEAR AROMATIC HYDROCARBONS
					TRIAZINE HERBICIDES
		SPECIFIC COMPOUNDS			SPECIFIC COMPOUNDS

REMARKS :

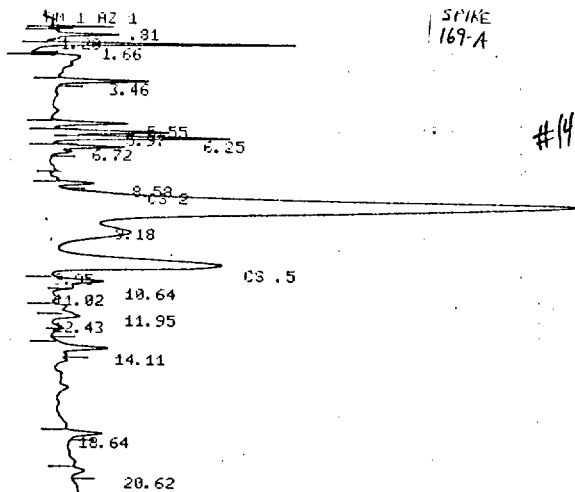
## ANALYTICAL RESULTS

COMPOUND	[PPB]	COMPOUND	[PPB]
halogenated paraffins	none detected	Toluene	none detected
Benzene	21	ethyl benzene	N.D.
meta-xylene	4	p-xylene	N.D. (trace)
		ortho xylene	N.D.
		* DETECTION LIMIT	2 ppm

REMARKS: Sample was not taken in a proper septum in bottle, but in a gallon bottle with a headspace. Seven other peaks detected by PID.

**CERTIFICATE OF ANALYTICAL PERSONNEL**

Seal(s) Intact: Yes NO. Seal(s) broken by: \_\_\_\_\_ date: \_\_\_\_\_  
I certify that I followed standard laboratory procedures on handling and analysis of this sample unless otherwise noted and that the statements in this block and the analytical data on this page accurately reflect the analytical results for this sample.  
Date(s) of analysis: 2-28-85. Analyst's signature: [Signature]  
I certify that I have reviewed and concur with the analytical results for this sample and with the statements in this block. Reviewers signature: \_\_\_\_\_



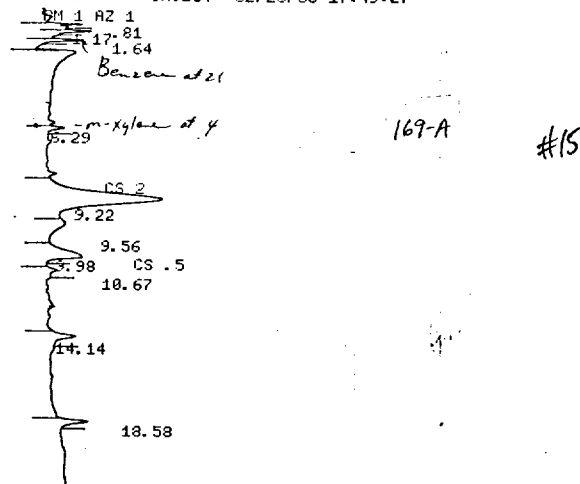
PID 02/28/85 17:21:16 CH= "A" PS= 1.

FILE 1. METHOD 0. RUN 286 INDEX 286

PEAK#	AREA%	RT	AREA BC
1	1.059	0.81	2297 01
2	4.218	1.2	9146 02
3	8.018	1.66	17384 03
4	3.987	3.46	8645 01
5	3.417	5.55	7408 01
6	5.17	5.97	11209 02
7	8.012	6.25	17372 03
8	4.024	6.72	8725 01
9	1.45	8.58	3143 01
10	33.792	9.18	73266 02
11	9.093	9.95	19715 02
12	4.174	10.64	9050 02
13	1.77	11.02	3838 02
14	1.687	11.95	3658 02
15	3.403	12.43	7378 03
16	3.634	14.11	7880 01
17	2.188	18.64	4743 01
18	0.902	20.62	1956 01

TOTAL 100. 216813

CHANNEL A INJECT 02/28/85 17:49:27



PID 02/28/85 17:49:27 CH= "A" PS= 1.

FILE 1. METHOD 0. RUN 287 INDEX 287

PEAK#	AREA%	RT	AREA BC
1	3.107	0.81	1134 01
2	5.572	1.17	2034 01
3	1.428	1.64	2602 01



SAMPLE NO. 6049  
 SAMPLE LOCATION DUNCAN FIELD Pit # DATE COLLECTED 2-25-85  
 DATE RECEIVED 2-26-85 COLLECTED BY DON PAYNE  
 DATE OF FINAL ANALYSIS 3-29-85 ADDRESS EHS  
 TECHNICIAN AB

TEST	PARAMETER	METHOD	RESULTS	mg/l
✓	ALKALINITY $CO_3$ & $HCO_3$ 2.49	TITRAMETRIC	152.0 mg/L $CaCO_3$	
✓	CALCIUM 3.52	TITRAMETRIC OR AA	176.3 mg/L $CaCO_3$	75-200
✓	CHLORIDE .34	TITRAMETRIC	12.1 mg/L Cl	250
✓	TOTAL HARDNESS	TITRAMETRIC	216.0 mg/L $CaCO_3$	500
✓	MAGNESIUM .80	CALCULATED OR AA	9.7 mg/L Mg	
	MANGANESE	SPECTROPHOTOMETRIC OR AA		0.05
	IRON	SPECTROPHOTOMETRIC OR AA		0.3
✓	pH	ELECTRODE	7.9 @ 20°C	6.5-8.5
	PHOSPHATE	SPECTROPHOTOMETRIC		
✓	POTASSIUM .02	FLAME PHOTOMETER	0.70 mg/L K	1000-2000
✓	SODIUM 1.6	FLAME PHOTOMETER	36.8 mg/L Na	
✓	SULFATE .33	TITRAMETRIC turbidity	15.8 mg/L $SO_4$	250
✓	TOTAL DISSOLVED SOLIDS	ELECTRODE	234.3 mg/L $CaCO_3$	500
	TURBIDITY	NEPHELOMETER		
	FLUORIDE	ELECTRODE		1.4

DATE RECEIVED 11/1/85 COLLECTED BY \_\_\_\_\_  
 DATE OUT 3-29-85 ADDRESS \_\_\_\_\_  
 TECHNICIAN AB

TEST	PARAMETER	METHOD	RESULTS	MCL
	ARSENIC	ATOMIC ABSORPTION		0.05
X	BARIUM	ATOMIC ABSORPTION	.0545	1.0
X	CADMIUM	ATOMIC ABSORPTION	.0012	0.01
X	CHROMIUM	ATOMIC ABSORPTION	.0091	0.05
	IRON	ATOMIC ABSORPTION		N/A
X	LEAD	ATOMIC ABSORPTION	.0193	0.05
	MANGANESE	ATOMIC ABSORPTION		N/A
	MERCURY	FLAMELESS ATOMIC ABSORPTION	2.002	0.002
X	SELENIUM	ATOMIC ABSORPTION	.0128	0.01
X	SILVER	ATOMIC ABSORPTION	2.001	0.05
	NITRATE (AsN)	CADMIUM REDUCTION		10.0
	FLUORIDE	ELECTRODE		1.4

FORM NO. 5460 © M

REV 8-8

TECHNICIAN AB

TEST	PARAMETER	METHOD	RESULTS	MCL
	ARSENIC	ATOMIC ABSORPTION		0.05
	BARIUM	ATOMIC ABSORPTION		1.0
	CADMIUM	ATOMIC ABSORPTION		0.01
	CHROMIUM	ATOMIC ABSORPTION		0.05
	IRON	ATOMIC ABSORPTION		N/A
	LEAD	ATOMIC ABSORPTION		0.05
	MANGANESE	ATOMIC ABSORPTION		N/A
	MERCURY	FLAMELESS ATOMIC ABSORPTION		0.002
	SELENIUM	ATOMIC ABSORPTION		0.01
	SILVER	ATOMIC ABSORPTION		0.05
✓	NITRATE (AsN)	CADMIUM REDUCTION	2.5 mg/L $NO_3-N$	10.0
	FLUORIDE	ELECTRODE		1.4

FORM NO. 5460 © M

REV 8-8



REPORT TO: \_\_\_\_\_

LABORATORY \_\_\_\_\_

LAB NUMBER ORG-170-17  
2/27/85

85-0170-B

SLD Users Code No. \_\_\_\_\_

ALL CONTAINERS WHICH THIS FORM ACCOMPANIES ARE COLLECTIVELY REFERRED TO AS "SAMPLE".

## CERTIFICATE OF FIELD PERSONNEL

Sample Type: Water ☒ Soil ☐ Other \_\_\_\_\_Water Supply and/or Code No. PIT #6City & County DUACAN OIL FIELD, SAN JUAN COUNTY, NMCollected (date & time) (850225/1400) By (name) ZAMAN, M.

pH= \_\_\_\_\_; Conductivity= \_\_\_\_\_ umho/cm at \_\_\_\_\_ °C; Chlorine Residual= \_\_\_\_\_

Dissolved Oxygen= \_\_\_\_\_ mg/l; Alkalinity= \_\_\_\_\_; Flow Rate= \_\_\_\_\_

Sampling Location, Methods &amp; Remarks (i.e. odors etc.)

*preserved w cyclohexane*

I certify that the statements in this block accurately reflect the results of my field analyses, observations and activities. Signed \_\_\_\_\_

I certify that I witnessed these field analyses, observations and activities and concur with the statements in this block. Signed \_\_\_\_\_

Method of Shipment to Laboratory \_\_\_\_\_

THIS FORM ACCOMPANIES \_\_\_\_\_ septum vials with teflon-lined discs identified as:

specimen \_\_\_\_\_; duplicate \_\_\_\_\_; triplicate \_\_\_\_\_; blank(s) \_\_\_\_\_,

and \_\_\_\_\_ amber glass jug(s) with teflon-lined cap(s) identified as \_\_\_\_\_,

and \_\_\_\_\_ other container(s) (describe) \_\_\_\_\_ identified as \_\_\_\_\_.

Containers are marked as follows to indicate preservation (circle):

NP: No preservation; sample stored at room temperature (~20°C).

P-ICE: Sample stored in an ice bath.

P-Na<sub>2</sub>O<sub>3</sub>S<sub>2</sub>: Sample preserved with 3 mg Na<sub>2</sub>O<sub>3</sub>S<sub>2</sub>/40 ml and stored at room temperature.

## CERTIFICATE(S) OF SAMPLE RECEIPT

I (we) certify that this sample was transferred from \_\_\_\_\_ to \_\_\_\_\_

\_\_\_\_\_ at (location) \_\_\_\_\_ on \_\_\_\_\_

(date &amp; time) \_\_\_\_\_ and that the statements in this block are correct.

Disposition of Sample \_\_\_\_\_. Seal(s) Intact: Yes ☐ No ☐.

Signature(s) \_\_\_\_\_

I (we) certify that this sample was transferred from \_\_\_\_\_ to \_\_\_\_\_

\_\_\_\_\_ at (location) \_\_\_\_\_ on \_\_\_\_\_

(date &amp; time) \_\_\_\_\_ and that the statements in this block are correct.

Disposition of Sample \_\_\_\_\_. Seal(s) Intact: Yes ☐ No ☐.

Signature(s) \_\_\_\_\_

## ANALYSES REQUESTED

LAB. No.: ORG-170

PLEASE CHECK THE APPROPRIATE BOXES BELOW TO INDICATE THE TYPE OF ANALYTICAL SCREENS REQUIRED. WHENEVER POSSIBLE LIST SPECIFIC COMPOUNDS SUSPECTED OR REQUIRED.

QUALITATIVE	QUANTITATIVE	PURGEABLE SCREENS	QUALITATIVE	QUANTITATIVE	EXTRACTABLE SCREENS
		ALIPHATIC HYDROCARBON SCREEN			ALIPHATIC HYDROCARBONS
	X	AROMATIC HYDROCARBON SCREEN			CHLORINATED HYDROCARBON PESTICIDES
		HALOGENATED HYDROCARBON SCREEN			CHLOROPHENOXY ACID HERBICIDES
		GAS CHROMATOGRAPH/MASS SPECTROMETER			HYDROCARBON FUEL SCREEN
					ORGANOPHOSPHATE PESTICIDES
					POLYCHLORINATED BIPHENYLS (PCB's)
					POLYNUCLEAR AROMATIC HYDROCARBONS
					TRIAZINE HERBICIDES
		SPECIFIC COMPOUNDS			SPECIFIC COMPOUNDS

REMARKS:

## ANALYTICAL RESULTS

COMPOUND	[PPB]	COMPOUND	[PPB]
halogenated purgables	none detected	Toluene	* Not detected
Benzene	100	ethyl benzene	N.D.*
		para-xylene	N.D.*
		meta-xylene	N.D.*
		ortho-xylene	N.D.*
		* DETECTION LIMIT	1 ppm

REMARKS: Sample was not taken in a proper septum vial bottle, but in a gallon jug bottle with a headspace. One other peak detected by PID.

## CERTIFICATE OF ANALYTICAL PERSONNEL

Seal(s) Intact: Yes NO X. Seal(s) broken by: \_\_\_\_\_ date: \_\_\_\_\_

I certify that I followed standard laboratory procedures on handling and analysis of this sample unless otherwise noted and that the statements in this block and the analytical data on this page accurately reflect the analytical results for this sample.

Date(s) of analysis: 2-28-85. Analyst's signature: [Signature]

I certify that I have reviewed and concur with the analytical results for this sample and with the statements in this block. Reviewers signature: \_\_\_\_\_

ER 0

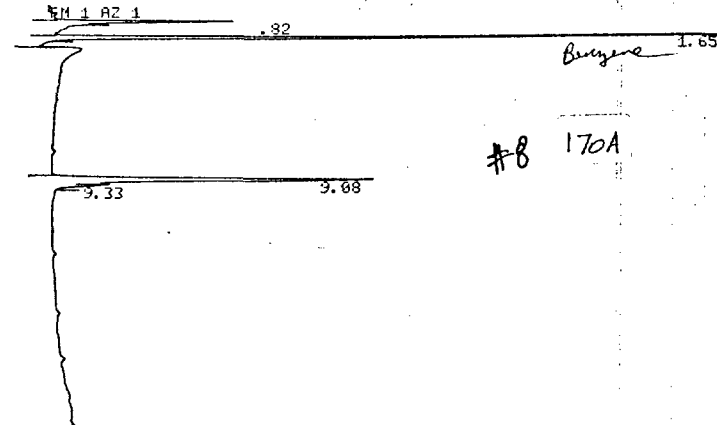
PID 03/13/85 11:36:45 CH= "A" PS= 1.

FILE 1. METHOD 0. RUN 447 INDEX 447

PEAK#	AREA%	RT	AREA BC
1	2.079	0.81	20257 02
2	2.62	1.21	25529 02
3	5.302	1.72	51669 02
4	0.841	2.25	8197 02
5	2.232	2.65	21748 02
6	1.005	3.05	9790 03
7	0.666	3.76	6494 01
8	0.428	4.9	4167 01
9	0.718	5.51	6996 02
10	1.34	5.93	13056 02
11	29.862	6.21	291009 03
12	0.886	8.28	8633 02
13	7.085	8.5	69046 02
14	7.649	9.08	74541 02
15	10.88	9.42	106024 02
16	10.896	9.84	106185 03
17	2.533	10.54	24638 02
18	1.826	10.92	17798 03
19	0.349	11.42	3400 02
20	2.372	11.85	23115 02
21	3.046	12.36	29685 02
22	0.393	12.97	3826 03
23	2.838	14.	27661 02
24	0.737	14.67	7187 02
25	0.322	15.34	3139 03
26	0.134	16.09	1303 01
27	0.961	18.39	9369 01

TOTAL 100. 974511

CHANNEL A INJECT 03/13/85 12:04:50



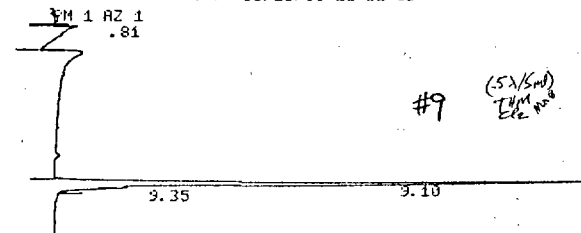
PID 03/13/85 12:04:50 CH= "A" PS= 1.

FILE 1. METHOD 0. RUN 448 INDEX 448

PEAK#	AREA%	RT	AREA BC
1	10.207	0.82	11456 01
2	51.722	1.65	58052 01
3	32.826	9.08	36243 02
4	5.245	9.33	5887 03

TOTAL 100. 112238

CHANNEL A INJECT 03/13/85 12:32:53



85-0253 -C

REPORT TO:

Masud Zu ZamanDivision of Water ResourcesP.O. Box 308, Window RockNavajo Nation (Arizona) 86515

LABORATORY

3/19/85

LAB NUMBER

OR. 253 A, B.

SLD Users Code No.

ALL CONTAINERS WHICH THIS FORM ACCOMPANIES ARE COLLECTIVELY REFERRED TO AS "SAMPLE".

## CERTIFICATE OF FIELD PERSONNEL

Sample Type: Water ☒ Soil ☐ Other \_\_\_\_\_

Water Supply and/or Code No. \_\_\_\_\_

City & County Window Rock, ArizonaCollected (date & time) 3/18/85 12:15 PM By (name) Zaman

pH= \_\_\_\_\_; Conductivity= \_\_\_\_\_ umho/cm at \_\_\_\_\_ °C; Chlorine Residual= \_\_\_\_\_

Dissolved Oxygen= \_\_\_\_\_ mg/l; Alkalinity= \_\_\_\_\_; Flow Rate= \_\_\_\_\_

Sampling Location, Methods &amp; Remarks (i.e. odors etc.)

Duncan Oil Field. Pit #3.

I certify that the statements in this block accurately reflect the results of my field analyses, observations and activities. Signed \_\_\_\_\_

I certify that I witnessed these field analyses, observations and activities and concur with the statements in this block. Signed \_\_\_\_\_

Method of Shipment to Laboratory \_\_\_\_\_

THIS FORM ACCOMPANIES 2 septum vials with teflon-lined discs identified as:

specimen \_\_\_\_\_; duplicate \_\_\_\_\_; triplicate \_\_\_\_\_; blank(s) \_\_\_\_\_,

and \_\_\_\_\_ amber glass, jug(s) with teflon-lined cap(s) identified as \_\_\_\_\_,

and \_\_\_\_\_ other container(s) (describe) \_\_\_\_\_ identified as \_\_\_\_\_.

Containers are marked as follows to indicate preservation (circle):

NP: No preservation; sample stored at room temperature (~20°C).

P-ICE: Sample stored in an ice bath.

P-Na<sub>2</sub>O<sub>3</sub>S<sub>2</sub>: Sample preserved with 3 mg Na<sub>2</sub>O<sub>3</sub>S<sub>2</sub>/40 ml and stored at room temperature.

## CERTIFICATE(S) OF SAMPLE RECEIPT

I (we) certify that this sample was transferred from \_\_\_\_\_ to \_\_\_\_\_  
\_\_\_\_\_ at (location) \_\_\_\_\_ on \_\_\_\_\_

(date &amp; time) \_\_\_\_\_ and that the statements in this block are correct.

Disposition of Sample \_\_\_\_\_. Seal(s) Intact: Yes ☐ No ☐.

Signature(s) \_\_\_\_\_

I (we) certify that this sample was transferred from \_\_\_\_\_ to \_\_\_\_\_  
\_\_\_\_\_ at (location) \_\_\_\_\_ on \_\_\_\_\_

(date &amp; time) \_\_\_\_\_ and that the statements in this block are correct.

Disposition of Sample \_\_\_\_\_. Seal(s) Intact: Yes ☐ No ☐.

Signature(s) \_\_\_\_\_



LAB. No.: ORG- 253

PURGEABLE SCREENS		EXTRACTABLE SCREENS	
QUALITATIVE	QUANTITATIVE	QUALITATIVE	QUANTITATIVE
	ALIPHATIC HYDROCARBON SCREEN		ALIPHATIC HYDROCARBONS
	AROMATIC HYDROCARBON SCREEN		CHLORINATED HYDROCARBON PESTICIDES
	HALOGENATED HYDROCARBON SCREEN		CHLOROPHENOXY ACID HERBICIDES
X	GAS CHROMATOGRAPH/MASS SPECTROMETER		HYDROCARBON FUEL SCREEN
			ORGANOPHOSPHATE PESTICIDES
			POLYCHLORINATED BIPHENYLS (PCB's)
			POLYNUCLEAR AROMATIC HYDROCARBONS
			TRIAZINE HERBICIDES
	SPECIFIC COMPOUNDS		SPECIFIC COMPOUNDS

REMARKS:

## ANALYTICAL RESULTS

COMPOUND	[PPB]	COMPOUND	[PPB]
m-Xylene	59 <del>15</del>		
Numerous Aliphatics	10 - 40 range		
Numerous C <sub>3</sub> -C <sub>6</sub> Substituted benzenes	20-40 ppb range		
		* DETECTION LIMIT	5 ug/l

REMARKS: No benzene or toluene detected

**CERTIFICATE OF ANALYTICAL PERSONNEL**

Seal(s) Intact: Yes NO ☒ . Seal(s) broken by: \_\_\_\_\_ date: \_\_\_\_\_

I certify that I followed standard laboratory procedures on handling and analysis of this sample unless otherwise noted and that the statements in this block and the analytical data on this page accurately reflect the analytical results for this sample. /

Date(s) of analysis: 3/25-6/85 . Analyst's signature: *K. M. ...*

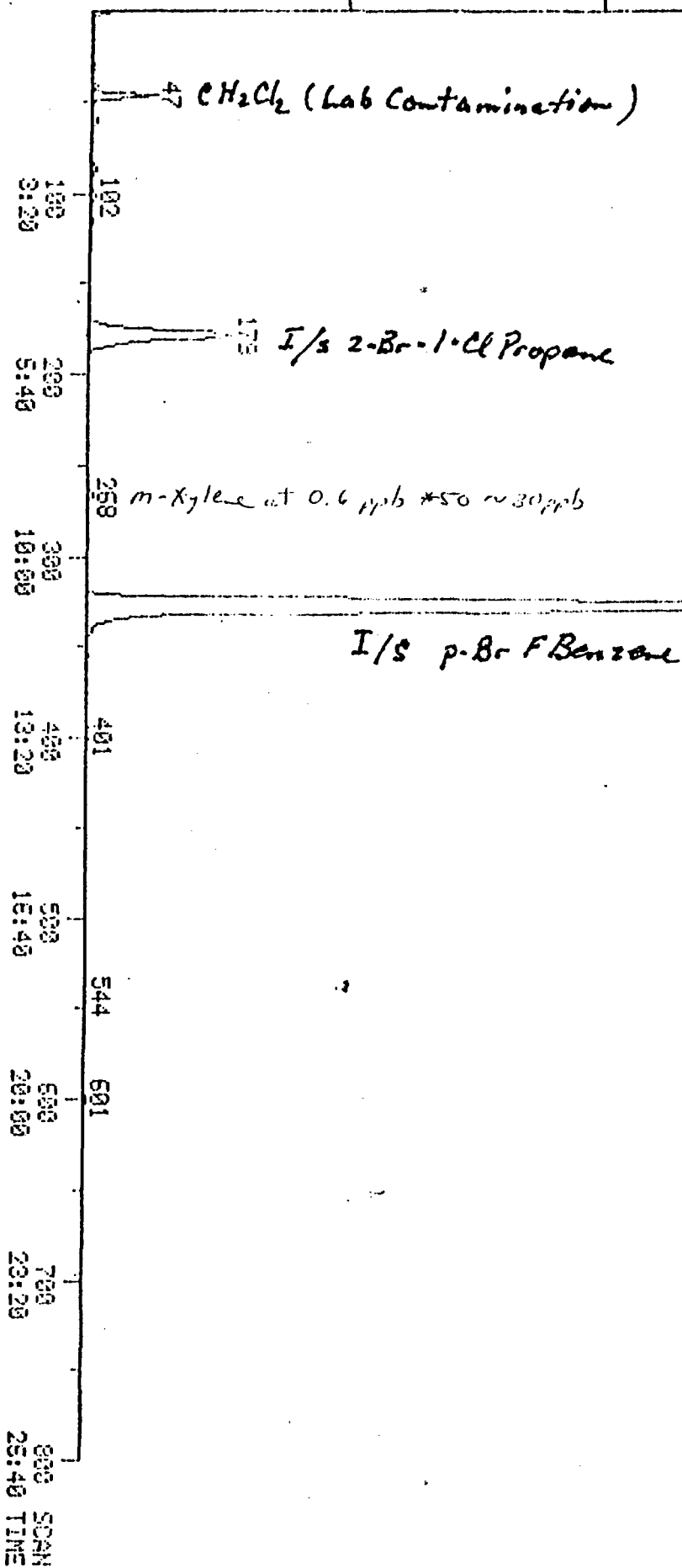
I certify that I have reviewed and concur with the analytical results for this sample and with the statements in this block. Reviewers signature:

RIC  
 03/25/85 14:38:39  
 SAMPLE: ORG-253 PIT#3 EUNCAN OIL FIELD 8503181215M2 100:5200 DILUTION  
 RANGE: C 1, 300 LABEL: N 0, 4.0 QUANT: A 0, 1.0 BASE: U 20, 3

DATA: ORG253 #297  
 CALL: 0850325 #4

SCANS 1 TO 800

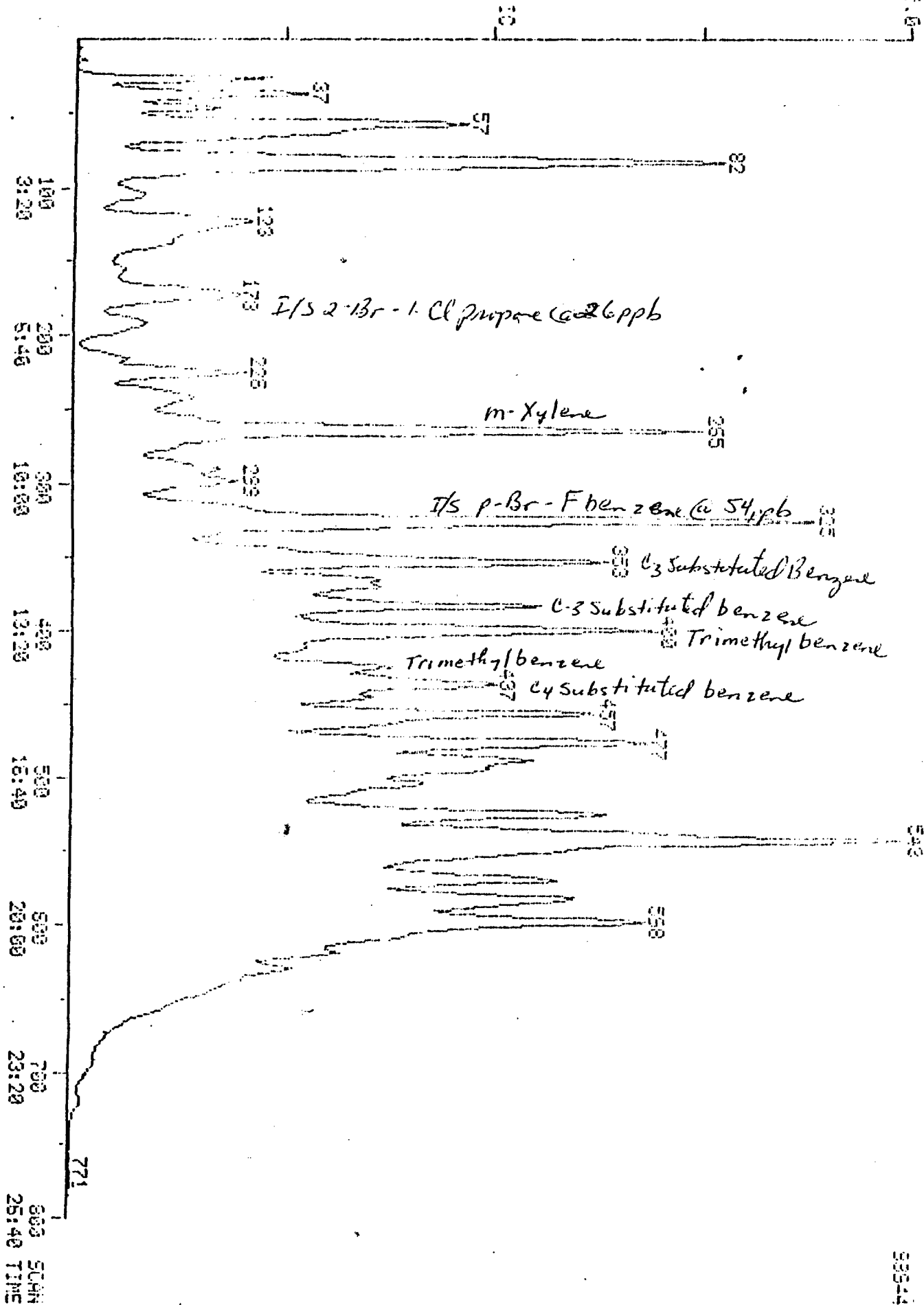
71153



RIC  
 03/23/85 15:43:00  
 SAMPLE: ORG-253 PIT#3 DUNCAN OIL FIELD 8503181215M2  
 RANGE: C 1, 800 LABEL: N 0, 4.0 QUAN: A 0, 1.0 BASE: U 20, 3  
 DATA: ORG253 #1  
 CALL: 0250325 #4

SCANS 1 TO 800

335-4



N. M. SCIENTIFIC LABORATORY DIVISION  
QUANTITATION REPORT - FILE: ORG253

DATA: ORG253.T1

03/25/85 15:43:00

SAMPLE: ORG-253 PIT#3 DUNCAN OIL FIELD 8503181215MZ

SUBMITTED BY: EID

ANALYST: RM

AMOUNT=AREA(HGHT) \* REF.AMNT/(REF.AREA(HGHT)\* RESP.FACT)  
RESP. FAC. FROM LIBRARY ENTRY

NO	NAME
1	I/S (4-FLUOROBROMOBENZENE)
2	I/S (2-BROMO-1-CHLOROPROPANE)
3	BENZENE
4	1,2-DICHLOROETHANE
5	TOLUENE
6	1,2-DIBROMOETHANE
7	ETHYLBENZENE
8	P-XYLENE
9	M-XYLENE
10	O-XYLENE

NO	M/E	SCAN	TIME	REF	RRT	METH	AREA(HGHT)	AMOUNT	%TOT
1	174	325	10:50	1	1.000	A BB	97693.	54.210 UG/L	36.85
2	77	175	5:50	1	0.538	A BB	47707.	32.287 UG/L	21.95
3	NOT FOUND								
4	NOT FOUND								
5	NOT FOUND								
6	NOT FOUND								
7	91	239	7:58	1	0.735	A BB	3365.	0.901 UG/L	0.61
8	NOT FOUND								
9	91	265	8:50	1	0.815	A BB	147865.	58.980 UG/L	40.09
10	91	284	9:28	1	0.874	A BB	1922.	0.743 UG/L	0.51

REPORT TO:

Masud Zu Zaman

LABORATORY

3/9/85

LAB NUMBER

OR 254 A, B.Division of Water ResourcesP.O. Box 308, Window RockNavajo Nation (Arizona) 86515

SLD Users Code No.

ALL CONTAINERS WHICH THIS FORM ACCOMPANIES ARE COLLECTIVELY REFERRED TO AS "SAMPLE".

## CERTIFICATE OF FIELD PERSONNEL

Sample Type: Water ☒ Soil ☐ Other \_\_\_\_\_

Water Supply and/or Code No. \_\_\_\_\_

City & County Window Rock, ArizCollected (date & time) 3/8/85 1:10 pm By (name) Zaman

pH= \_\_\_\_\_; Conductivity= \_\_\_\_\_ umho/cm at \_\_\_\_\_ °C; Chlorine Residual= \_\_\_\_\_

Dissolved Oxygen= \_\_\_\_\_ mg/l; Alkalinity= \_\_\_\_\_; Flow Rate= \_\_\_\_\_

Sampling Location, Methods &amp; Remarks (i.e. odors etc.)

Dineen Oil Field. Produced water pit

I certify that the statements in this block accurately reflect the results of my field analyses, observations and activities. Signed \_\_\_\_\_

I certify that I witnessed these field analyses, observations and activities and concur with the statements in this block. Signed \_\_\_\_\_

Method of Shipment to Laboratory \_\_\_\_\_

THIS FORM ACCOMPANIES 2 septum vials with teflon-lined discs identified as:

specimen \_\_\_\_\_; duplicate \_\_\_\_\_; triplicate \_\_\_\_\_; blank(s) \_\_\_\_\_

and \_\_\_\_\_ amber glass jug(s) with teflon-lined cap(s) identified as \_\_\_\_\_

and \_\_\_\_\_ other container(s) (describe) \_\_\_\_\_ identified as \_\_\_\_\_

Containers are marked as follows to indicate preservation (circle):

HP: No preservation; sample stored at room temperature (~20°C).

P-ICE: Sample stored in an ice bath.

P-Na<sub>2</sub>O<sub>3</sub>S<sub>2</sub>: Sample preserved with 3 mg Na<sub>2</sub>O<sub>3</sub>S<sub>2</sub>/40 ml and stored at room temperature.

## CERTIFICATE(S) OF SAMPLE RECEIPT

I (we) certify that this sample was transferred from \_\_\_\_\_ to \_\_\_\_\_ at (location) \_\_\_\_\_ on \_\_\_\_\_

(date &amp; time) \_\_\_\_\_ and that the statements in this block are correct.

Disposition of Sample \_\_\_\_\_. Seal(s) Intact: Yes ☐ No ☐

Signature(s) \_\_\_\_\_

I (we) certify that this sample was transferred from \_\_\_\_\_ to \_\_\_\_\_ at (location) \_\_\_\_\_ on \_\_\_\_\_

(date &amp; time) \_\_\_\_\_ and that the statements in this block are correct.

Disposition of Sample \_\_\_\_\_. Seal(s) Intact: Yes ☐ No ☐

Signature(s) \_\_\_\_\_

## ANALYSES REQUESTED

LAB. No.: ORG- 254

PLEASE CHECK THE APPROPRIATE BOXES BELOW TO INDICATE THE TYPE OF ANALYTICAL SCREENS REQUIRED. WHENEVER POSSIBLE LIST SPECIFIC COMPOUNDS SUSPECTED OR REQUIRED.

QUALITATIVE	QUANTITATIVE	PURGEABLE SCREENS	QUALITATIVE	QUANTITATIVE	EXTRACTABLE SCREENS
		ALIPHATIC HYDROCARBON SCREEN			ALIPHATIC HYDROCARBONS
		AROMATIC HYDROCARBON SCREEN			CHLORINATED HYDROCARBON PESTICIDES
		HALOGENATED HYDROCARBON SCREEN			CHLOROPHENOXY ACID HERBICIDES
	X	GAS CHROMATOGRAPH/MASS SPECTROMETER			HYDROCARBON FUEL SCREEN
					ORGANOPHOSPHATE PESTICIDES
					POLYCHLORINATED BIPHENYLS (PCB's)
					POLYNUCLEAR AROMATIC HYDROCARBONS
					TRIAZINE HERBICIDES
		SPECIFIC COMPOUNDS			SPECIFIC COMPOUNDS

REMARKS:

## ANALYTICAL RESULTS

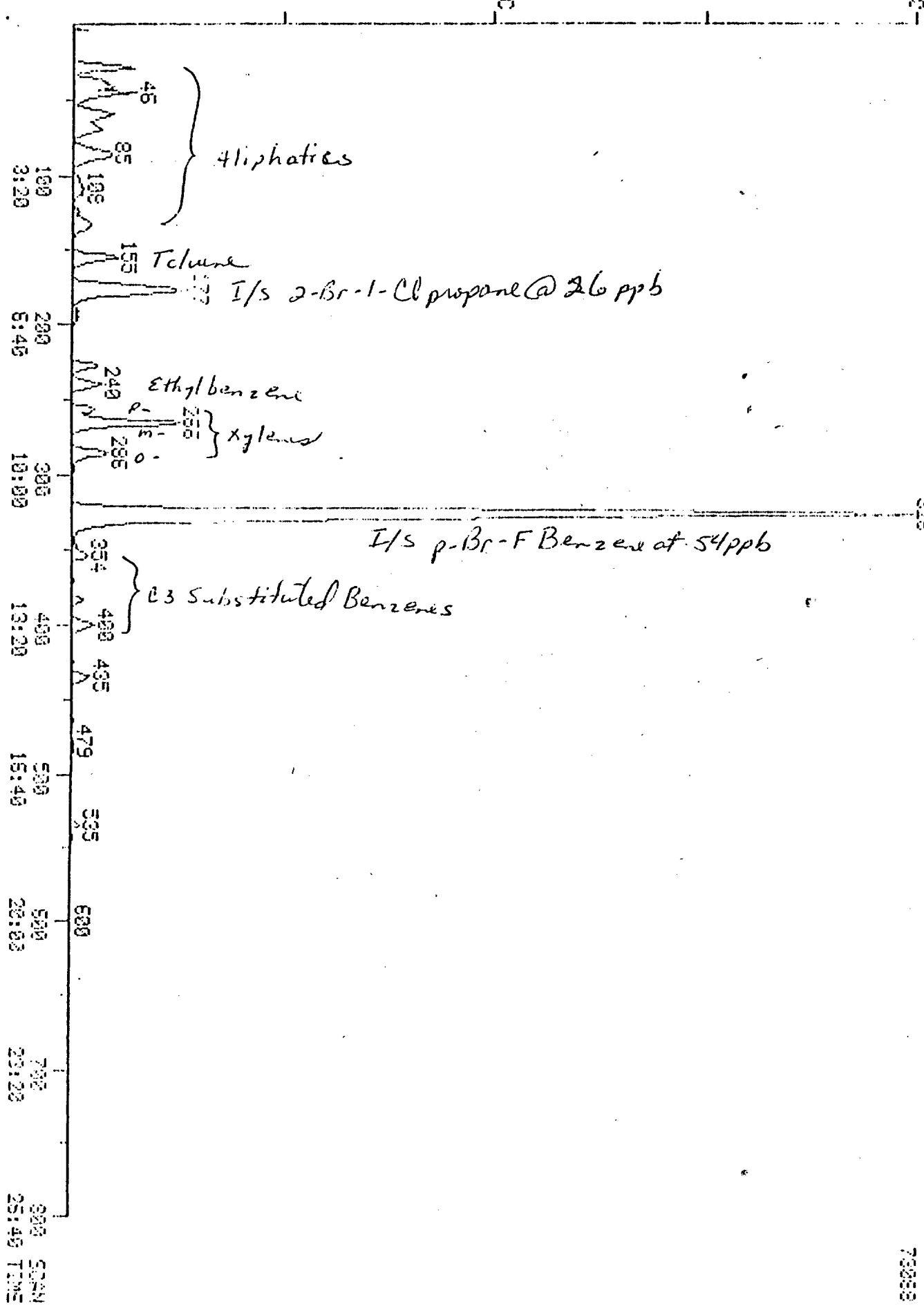
COMPOUND	[PPB]	COMPOUND	[PPB]
Benzene	less than 50 ug/l		
Toluene	180		
Ethylbenzene	106		
p-Xylene	80		
m-Xylene	390		
o-Xylene	150		
		* DETECTION LIMIT	50 ug/l

REMARKS: Some aliphatic compounds in the 6-8 carbon range at about the 10 ppb range.

## CERTIFICATE OF ANALYTICAL PERSONNEL

Seal(s) Intact: Yes NO X. Seal(s) broken by: \_\_\_\_\_ date: \_\_\_\_\_  
 I certify that I followed standard laboratory procedures on handling and analysis of this sample unless otherwise noted and that the statements in this block and the analytical data on this page accurately reflect the analytical results for this sample.  
 Date(s) of analysis: 3/25-6/85. Analyst's signature: R. Meyer  
 I certify that I have reviewed and concur with the analytical results for this sample and with the statements in this block. Reviewers signature: \_\_\_\_\_

RIC  
 03/26/85 16:33:00  
 SAMPLE: ORG-2548 PRODUCED WATER PIT DUNCAN OIL FIELD DIL 100:5000  
 RANGE: C 1, 800 LABEL: N 0, 4.0 QUAN: A 0, 1.0 BASE: U 20, 3  
 DATA: ORG2548 #286  
 CALL: C850325 #4  
 SCANS 1 TO 800



DATA: DRG254B.11

03/26/85 16:33:00

SAMPLE: DRG-254B PRODUCED WATER PIT DUNCAN OIL FIELD DIL 100:5000

SUBMITTED BY: EID

ANALYST: RM

AMOUNT=AREA(HGHT) \* REF. AMNT / (REF. AREA(HGHT) \* RESP. FACT)  
RESP. FAC. FROM LIBRARY ENTRY

NO NAME

- 1 I/S (4-FLUOROBROMOBENZENE)
- 2 I/S (2-BROMO-1-CHLOROPROPANE)
- 3 BENZENE
- 4 1,2-DICHLOROETHANE
- 5 TOLUENE
- 6 1,2-DIBROMOETHANE
- 7 ETHYLBENZENE
- 8 P-XYLENE
- 9 M-XYLENE
- 10 O-XYLENE

NO	M/E	SCAN	TIME	REF	RRT	METH	AREA(HGHT)	AMOUNT	WTOT
1	174	326	10:52	1	1.000	A BB	100880.	54.210 UG/L	53.17
2	77	177	5:54	1	0.543	A BB	44646.	29.260 UG/L	28.70
3	NOT FOUND								
4	NOT FOUND								
5	91	155	5:10	1	0.475	A BB	12727.	3.712 UG/L	3.64 180
6	NOT FOUND								
7	91	240	8:00	1	0.736	A BB	8360.	2.170 UG/L	2.13 106
8	91	257	8:34	1	0.788	A BV	4146.	1.644 UG/L	1.61 80
9	91	266	8:52	1	0.816	A VB	20558.	7.941 UG/L	7.79 390
10	91	206	9:32	1	0.877	A BB	8063.	3.018 UG/L	2.96 150



REPORT TO:

Masud UZ ZamanDivision of Water ResourcesP.O. Box 308Window Rock, Navajo Nation (AZ) 86515

LABORATORY

3/18/85

LAB NUMBER

OR 255 A, B.

SLD Users Code No.

ALL CONTAINERS WHICH THIS FORM ACCOMPANIES ARE COLLECTIVELY REFERRED TO AS "SAMPLE".

## CERTIFICATE OF FIELD PERSONNEL

Sample Type: Water ☒ Soil ☐ Other \_\_\_\_\_

Water Supply and/or Code No. \_\_\_\_\_

City & County Window Rock, Ariz.Collected (date & time) 3/18/85 11:45 AM By (name) Zaman

pH= \_\_\_\_\_; Conductivity= \_\_\_\_\_ umho/cm at \_\_\_\_\_ °C; Chlorine Residual= \_\_\_\_\_

Dissolved Oxygen= \_\_\_\_\_ mg/l; Alkalinity= \_\_\_\_\_; Flow Rate= \_\_\_\_\_

Sampling Location, Methods &amp; Remarks (i.e. odors etc.)

Duncan Oil FieldPit #1.

I certify that the statements in this block accurately reflect the results of my field analyses, observations and activities. Signed \_\_\_\_\_

I certify that I witnessed these field analyses, observations and activities and concur with the statements in this block. Signed \_\_\_\_\_

Method of Shipment to Laboratory \_\_\_\_\_

THIS FORM ACCOMPANIES 2 septum vials with teflon-lined discs identified as:

specimen \_\_\_\_\_; duplicate \_\_\_\_\_; triplicate \_\_\_\_\_; blank(s) \_\_\_\_\_,

and \_\_\_\_\_ amber glass jug(s) with teflon-lined cap(s) identified as \_\_\_\_\_,

and \_\_\_\_\_ other container(s) (describe) \_\_\_\_\_ identified as \_\_\_\_\_.

Containers are marked as follows to indicate preservation (circle):

NP: No preservation; sample stored at room temperature (~20°C).

P-ICE: Sample stored in an ice bath.

P-Na<sub>2</sub>O<sub>3</sub>S<sub>2</sub>: Sample preserved with 3 mg Na<sub>2</sub>O<sub>3</sub>S<sub>2</sub>/40 ml and stored at room temperature.

## CERTIFICATE(S) OF SAMPLE RECEIPT

I (we) certify that this sample was transferred from \_\_\_\_\_ to \_\_\_\_\_

\_\_\_\_\_ at (location) \_\_\_\_\_ on \_\_\_\_\_

(date &amp; time) \_\_\_\_\_ and that the statements in this block are correct.

Disposition of Sample \_\_\_\_\_. Seal(s) Intact: Yes ☐ No ☐.

Signature(s) \_\_\_\_\_

I (we) certify that this sample was transferred from \_\_\_\_\_ to \_\_\_\_\_

\_\_\_\_\_ at (location) \_\_\_\_\_ on \_\_\_\_\_

(date &amp; time) \_\_\_\_\_ and that the statements in this block are correct.

Disposition of Sample \_\_\_\_\_. Seal(s) Intact: Yes ☐ No ☐.

Signature(s) \_\_\_\_\_

LAB. No.: ORG- 255

QUALITATIVE		QUANTITATIVE	PURGEABLE SCREENS	QUALITATIVE	QUANTITATIVE	EXTRACTABLE SCREENS
			ALIPHATIC HYDROCARBON SCREEN			ALIPHATIC HYDROCARBONS
	X		AROMATIC HYDROCARBON SCREEN			CHLORINATED HYDROCARBON PESTICIDES
			HALOGENATED HYDROCARBON SCREEN			CHLOROPHENOXY ACID HERBICIDES
	X		GAS CHROMATOGRAPH/MASS SPECTROMETER			HYDROCARBON FUEL SCREEN
						ORGANOPHOSPHATE PESTICIDES
						POLYCHLORINATED BIPHENYLS (PCB's)
						POLYNUCLEAR AROMATIC HYDROCARBONS
						TRIAZINE HERBICIDES
			SPECIFIC COMPOUNDS			SPECIFIC COMPOUNDS

REMARKS:

COMPOUND	[PPB]	COMPOUND	[PPB]
<i>Aromatic purgeables</i>	<i>None Detected</i>		
		* DETECTION LIMIT	<i>50 ug/l</i>

REMARKS: Some aliphatic paraffins were detected in the 10 to 200 carbon range at about 100 to 500 ppb level.

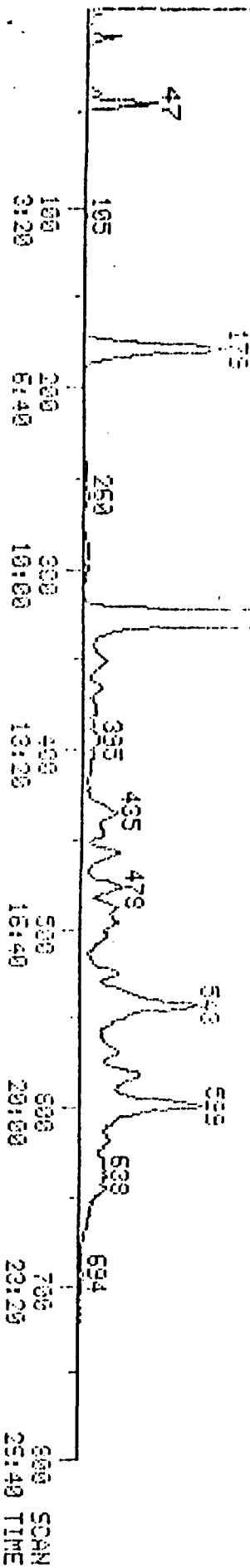
Seal(s) Intact: Yes \_\_\_ NO X. Seal(s) broken by: \_\_\_\_\_ date: \_\_\_\_\_  
I certify that I followed standard laboratory procedures on handling and analysis of this sample unless otherwise noted and that the statements in this block and the analytical data on this page accurately reflect the analytical results for this sample.  
Date(s) of analysis: 3/25-6/95. Analyst's signature: R Meyer  
I certify that I have reviewed and concur with the analytical results for this sample and with the statements in this block. Reviewer's signature: R Meyer

R1C  
 03/25/85 13:38:00  
 SAMPLE: ORG-255A PIT#1 DUNCAN OIL FIELD 65231814E2 100:5000 DILUTION  
 RANGE: 0 1, 800 LABEL: N 0, 4, 8 SUM: 9 0, 1, 2 BASE: 0 20, 3  
 327

DATA: ORG255A #287  
 CALL: C85825 #4

SCANS 1 TO 800

790000



~~N. M. SCIENTIFIC LABORATORY DIVISION~~

DATA: ORG255B, 11

03/26/85 13:38:00

SAMPLE: ORG-255A PIT#1 DUNCAN OIL FIELD 8503181145HZ 100:5000 DILUTION

SUBMITTED BY: EDD

ANALYST: RM

$$\text{AMOUNT} = \text{AREA}(\text{HEIGHT}) * \text{REF. AMNT} / (\text{REF. AREA}(\text{HEIGHT}) * \text{RESP. FACT})$$

RESP. FAC. FROM LIBRARY ENTRY

NO NAME

1 I/S (4-FLUOROBROMOBENZENE)

2 1/5 (2-BROMO-1-CHLOROPROPANE)

### 3 BENZENE

4 1,2-DICHLOROETHANE

## 5 TOLUENE.

## 6 1,2-DIURONIDEthane

7 ETHYLBENZENE

8 P--XYLENE

9 M-XYLENE

10. D-XYL FINE-

NO	M/E	SCAN	TIME	REF	RRY	METH	AREA(HGHT)	AMOUNT	%TOT
1	174	327	10:54	1	1.000	A BB	100236.	54.210 UG/L	63.74
2	77	177	5:58	1	0.547	A BB	46755.	30.837 UG/L	36.26
3	NOT FOUND								
4	NOT FOUND								
5	NOT FOUND								
6	NOT FOUND								
7	NOT FOUND								
8	NOT FOUND								
9	NOT FOUND								
10	NOT FOUND								