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BEFORE THE OIL CONSERVATION COMMISSION
OF THE STATE OF NEW MEXICO

IN THE MATTER OF THE HEARING CALLED BY THE
OIL CONSERVATION COMMISSION UPON ITS OWN
MOTION TO CONSIDER AN ORDER PROHIBITING THE
DISPOSAL OF OIL FIELD BRINES IN SURFACE
PITS IN LEA, CHAVES, ROOSEVELT, AND EDDY
COUNTIES, NEW MEXICO.

CASE No. 3551
Order No. R-3221

ORDER OF THE COMMISSION

BY THE COMMISSION:

This cause came on for hearing at 9 a.m. on April 19, 1967, at Hobbs, New Mexico, before the Oil Conservation Commission of New Mexico, hereinafter referred to as the "Commission."

NOW, on this 1st day of May, 1967, the Commission, a quorum being present, having considered the testimony presented and the exhibits received at said hearing, and being fully advised in the premises,

FINDS:

(1) That due public notice having been given as required by law, the Commission has jurisdiction of this cause and the subject matter thereof.

(2) That large amounts of water produced in conjunction with the production of oil or gas, or both, are being disposed of on the surface of the ground by means of unlined disposal pits located in Lea, Eddy, Chaves, and Roosevelt Counties, New Mexico.

(3) That said produced water contains high concentrations of chlorides.

(4) That fresh water supplies as designated by the state engineer exist in substantially all areas where there is surface pit disposal and in substantially all the area encompassed by Lea, Eddy, Chaves, and Roosevelt Counties, New Mexico.

(5) That the disposal of water produced in conjunction with the production of oil or gas, or both, on the surface of

the ground, or in any pit, pond, lake, depression, draw, streambed, or arroyo, or in any watercourse, constitutes a hazard to existing fresh water supplies, as designated by the state engineer, in the vicinity of such disposal.

(6) That in order to afford reasonable protection against contamination of fresh water supplies as designated by the state engineer, the disposal of water produced in conjunction with the production of oil or gas, or both, on the surface of the ground, or in any pit, pond, lake, depression, draw, streambed, or arroyo, or in any watercourse, or in any other place or in any manner which will constitute a hazard to any fresh water supplies existing in Lea, Eddy, Chaves, and Roosevelt Counties, New Mexico, should be prohibited in said Counties.

(7) That the testimony indicates that the volume of water produced in conjunction with the production of oil or gas, or both, from the North Bagley-Upper Pennsylvanian, North Bagley-Middle Pennsylvanian, North Bagley-Lower Pennsylvanian, North Bagley-Wolfcamp, and Northeast Bagley-Wolfcamp Pools, Lea County, New Mexico, and being disposed of into unlined pits is so great as to constitute an imminent threat to fresh water supplies designated by the state engineer and the surface disposal of said water should, therefore, be prohibited after October 31, 1967, in the area encompassed by Lea, Eddy, Chaves, and Roosevelt Counties, New Mexico.

(8) That large amounts of water are produced in conjunction with the production of oil from active waterflood projects and active water pressure maintenance projects.

(9) That one or more injection wells are present in each waterflood project.

(10) That in order to afford reasonable protection against contamination of fresh water supplies designated by the state engineer, the surface disposal of water produced in conjunction with the production of oil from active waterflood projects and active water pressure maintenance projects should be prohibited after December 31, 1967, in the area encompassed by Lea, Eddy, Chaves, and Roosevelt Counties, New Mexico.

(11) That complete prohibition of surface disposal in Lea, Eddy, Chaves, and Roosevelt Counties, New Mexico, of water produced in conjunction with the production of oil or gas, or both, should be accomplished by December 31, 1968, unless specifically exempted.

(12) That the surface disposal in pits of not more than one barrel per day for each developed 40-acre tract served by said pits, but limited to a maximum of 16 barrels per day, is so insignificant as to present little hazard to fresh water supplies and should be allowed in order to prevent waste caused by the premature abandonment of wells.

(13) That in order to prevent waste caused by the drowning out of oil or gas wells or burdensome delay or expenses, the District Supervisor of the appropriate District Office of the Commission should be empowered to authorize temporary disposal in surface pits for a period not to exceed 30 days for such contingencies as injection system failures and evaluation of wildcat wells.

IT IS THEREFORE ORDERED

(1) That effective November 1, 1967, the disposal of water produced in conjunction with the production of oil or gas, or both, from the North Bagley-Upper Pennsylvanian, North Bagley-Middle Pennsylvanian, North Bagley-Lower Pennsylvanian, North Bagley-Wolfcamp, and Northeast Bagley-Wolfcamp Pools, Lea County, New Mexico, or within one mile thereof, on the surface of the ground, or in any pit, pond, lake, depression, draw, streambed, or arroyo, or in any watercourse, or in any other place or in any manner which will constitute a hazard to any fresh water supplies, is hereby prohibited in that area encompassed by Lea, Eddy, Chaves, and Roosevelt Counties, New Mexico.

(2) That effective January 1, 1968, the disposal of water produced in conjunction with the production of oil from any waterflood project or water pressure maintenance project on the surface of the ground, or in any pit, pond, lake, depression, draw, streambed, or arroyo, or in any watercourse, or in any other place or in any manner which will constitute a hazard to any fresh water supplies is hereby prohibited in that area encompassed by Lea, Eddy, Chaves, and Roosevelt Counties, New Mexico.

(3) That effective January 1, 1969, the disposal of water produced in conjunction with the production of oil or gas, or both, on the surface of the ground, or in any pit, pond, lake, depression, draw, streambed, or arroyo, or in any watercourse, or in any other place or in any manner which will constitute a hazard to any fresh water supplies and said disposal has not

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previously been prohibited by Orders Nos. (1) or (2) above, or by Order No. R-1224-A, or by Order No. R-2526, or by Order No. R-2788, or by Order No. R-3164, is hereby prohibited in that area encompassed by Lea, Eddy, Chaves, and Roosevelt Counties, New Mexico.

(4) That in those areas subject to the provisions of Orders Nos. (1) and (3) above, surface pits may be utilized for the disposal of a maximum of one barrel of produced water per day for each developed 40-acre tract served by said pits, provided however, that in no event shall said surface pit disposal exceed 16 barrels per day, and provided further, that this authorization shall not apply to those areas affected by Orders Nos. R-1224-A, R-2526, R-2788, or R-3164.

(5) That nothing contained in this order shall be construed as prohibiting the disposal of water produced in conjunction with the production of oil or gas, or both, in impervious lined pits presently in use, provided said pits were inspected and approved by a Commission representative prior to use, and for so long as said pits are properly maintained to ensure their continued imperviousness.

(6) That each unlined pit used for the disposal of water produced in conjunction with the production of oil or gas, or both, and not servicing a well exempt under the provisions of Order No. (4) above shall be filled, leveled, and compacted within six months after its use for the disposal of produced water is prohibited or by November 1, 1967, whichever date is later.

(7) That nothing contained in this order shall be construed as prohibiting the use and maintenance of mud pits or burn pits.

(8) That the District Supervisor of the appropriate District Office of the Commission is hereby empowered to authorize temporary disposal in surface pits for a period not to exceed 30 days for such contingencies as injection system failures and evaluation of wildcat wells. Authority for said disposal shall only be granted on an individual case basis and only after the volume and quality of the water produced and the proximity of fresh water supplies have been taken into consideration.

(9) That the provisions of this order are in addition to the provisions of Order No. R-1224-A, Order No. R-2526, Order

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No. R-2788, and Order No. R-3164 of the Commission and nothing herein contained shall be construed as abridging or altering in any manner the provisions of said orders.

(10) That jurisdiction of this cause is retained for the entry of such further orders as the Commission may deem necessary.

DONE at Santa Fe, New Mexico, on the day and year hereinabove designated.

STATE OF NEW MEXICO
OIL CONSERVATION COMMISSION

DAVID F. CARGO, Chairman

GUYTON B. HAYS, Member

A. L. PORTER, Jr., Member & Secretary

S E A L

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BEFORE THE OIL CONSERVATION COMMISSION
OF THE STATE OF NEW MEXICO

IN THE MATTER OF THE HEARING CALLED BY THE OIL CONSERVATION COMMISSION UPON ITS OWN MOTION TO CONSIDER THE REVISION OF PARAGRAPH (1) OF ORDER NO. R-3221, TO PROVIDE THAT THE EFFECTIVE DATE FOR THE PROHIBITION OF SURFACE DISPOSAL OF PRODUCED WATER FROM THE NORTH BAGLEY-UPPER PENNSYLVANIAN, NORTH BAGLEY-MIDDLE PENNSYLVANIAN, NORTH BAGLEY-LOWER PENNSYLVANIAN, NORTH BAGLEY-WOLFCAMP, AND NORTHEAST BAGLEY-WOLFCAMP POOLS, LEA COUNTY, NEW MEXICO, OR WITHIN ONE MILE THEREOF, BE CHANGED FROM NOVEMBER 1, 1967, TO SOME EARLIER DATE.

CASE No. 3644
Order No. R-3221-A

ORDER OF THE COMMISSION

BY THE COMMISSION:

This cause came on for hearing at 9 a.m. on August 30, 1967, at Santa Fe, New Mexico, before the Oil Conservation Commission of New Mexico, hereinafter referred to as the "Commission."

NOW, on this 31st day of August, 1967, the Commission, a quorum being present, having considered the testimony presented and the exhibits received at said hearing, and being fully advised in the premises,

FINDS:

(1) That due public notice having been given as required by law, the Commission has jurisdiction of this cause and the subject matter thereof.

(2) That effective November 1, 1967, Order (1) of Order No. R-3221 forbids the disposal of water produced in conjunction with the production of oil or gas, or both, from the North Bagley-Upper Pennsylvanian, North Bagley-Middle Pennsylvanian, North Bagley-Lower Pennsylvanian, North Bagley-Wolfcamp, and Northeast Bagley-Wolfcamp Pools, Lea County, New Mexico, or within one mile thereof, on the surface of the ground, or in any pit, pond, lake, depression, draw, streambed, or arroyo, or in any water-course, or in any other place or in any manner which will constitute a hazard to any fresh water supplies in that area encompassed by Lea, Eddy, Chaves, and Roosevelt Counties, New Mexico.

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(3) That upon considering the evidence presented in Case No. 3551 which resulted in the issuance of Order No. R-3221, the Commission found the production of salt water in the North Bagley Field to be so great as to constitute an imminent threat to the fresh water supplies designated by the state engineer and found that it would be necessary to prohibit surface disposal of said salt water no later than November 1, 1967, in order to afford reasonable protection against contamination of said fresh water supplies.

(4) That the evidence presented in Case No. 3644 establishes that the volume of salt water being produced in conjunction with the production of oil or gas, or both, in the North Bagley Field greatly exceeds the volume of produced salt water anticipated by the Commission when it issued Order No. R-3221.

(5) That the evidence presented in Case No. 3644 establishes that the excessive amounts of water being produced in conjunction with the production of oil or gas, or both, in the North Bagley Field constitute an even more immediate threat to the fresh water supplies than anticipated by the Commission at the time Order No. R-3221 was issued.

(6) That the testimony presented in Case No. 3644 indicates that all salt water disposal systems in the North Bagley Area heretofore authorized by the Commission can be in operation by approximately the middle of October, 1967.

(7) That the aforesaid salt water disposal systems will be capable of handling all salt water being produced in the North Bagley Field in mid-October.

(8) That the prohibition of water produced in conjunction with the production of oil or gas, or both, in the North Bagley Field, or within one mile thereof, on the surface of the ground, or in any pit, pond, lake, depression, draw, streambed, or arroyo, or in any watercourse, or in any other place or in any manner which will constitute a hazard to any fresh water supplies on or after November 1, 1967, in the area encompassed by Lea, Eddy, Chaves, and Roosevelt Counties, New Mexico, will not afford reasonable protection against contamination of fresh water supplies designated by the state engineer.

(9) That in order to afford reasonable protection against contamination of fresh water supplies designated by the state engineer, Order (1) of Order No. R-3221 should be amended to

prohibit the surface disposal of water produced in conjunction with the production of oil or gas, or both, in the North Bagley Field, or within one mile thereof, on or after October 16, 1967.

IT IS THEREFORE ORDERED:

(1) That Order (1) of Order No. R-3221, dated May 1, 1967, is hereby amended to read in its entirety as follows:

"(1) That effective October 16, 1967, the disposal of water produced in conjunction with the production of oil or gas, or both, from the North Bagley-Upper Pennsylvanian, North Bagley-Middle Pennsylvanian, North Bagley-Lower Pennsylvanian, North Bagley-Wolfcamp, and Northeast Bagley-Wolfcamp Pools, Lea County, New Mexico, or within one mile thereof, on the surface of the ground, or in any pit, pond, lake, depression, draw, streambed, or arroyo, or in any watercourse, or in any other place or in any manner which will constitute a hazard to any fresh water supplies, is hereby prohibited in that area encompassed by Lea, Eddy, Chaves, and Roosevelt Counties, New Mexico."

(2) That jurisdiction of this cause is retained for the entry of such further orders as the Commission may deem necessary.

DONE at Santa Fe, New Mexico, on the day and year hereinabove designated.

STATE OF NEW MEXICO
OIL CONSERVATION COMMISSION

DAVID F. CARGO, Chairman

GUYTON B. HAYS, Member

A. L. PORTER, Jr., Member & Secretary

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BEFORE THE OIL CONSERVATION COMMISSION
OF THE STATE OF NEW MEXICO

IN THE MATTER OF THE HEARING CALLED BY THE OIL CONSERVATION COMMISSION ON ITS OWN MOTION TO CONSIDER THE AMENDMENT OF ORDER No. R-3221, THE COMMISSION'S SALT WATER DISPOSAL ORDER, TO PERMIT THE EXEMPTION OF CERTAIN PRESENTLY EXISTING AND FUTURE POOLS IN EDDY AND LEA COUNTIES, NEW MEXICO, FROM CERTAIN REQUIREMENTS OF SAID ORDER.

CASE No. 3806
Order No. R-3221-B

ORDER OF THE COMMISSION

BY THE COMMISSION:

This cause came on for hearing at 9 a.m. on July 17, 1968, at Santa Fe, New Mexico, before the Oil Conservation Commission of New Mexico, hereinafter referred to as the "Commission."

NOW, on this 25th day of July, 1968, the Commission, a quorum being present, having considered the testimony presented and the exhibits received at said hearing, and being fully advised in the premises,

FINDS:

(1) That due public notice having been given as required by law, the Commission has jurisdiction of this cause and the subject matter thereof.

(2) That effective January 1, 1969, Order (3) of Commission Order No. R-3221, dated May 1, 1967, prohibits in that area encompassed by Lea, Eddy, Chaves, and Roosevelt Counties, New Mexico, the disposal, subject to minor exceptions, of water produced in conjunction with the production of oil or gas, or both, on the surface of the ground, or in any pit, pond, lake, depression, draw, streambed, or arroyo, or in any watercourse, or in any other place or in any manner which will constitute a hazard to any fresh water supplies and said disposal has not previously been prohibited.

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(3) That within the area described as:

EDDY AND LEA COUNTIES, NEW MEXICO

TOWNSHIP 19 SOUTH, RANGE 30 EAST, NMPM
Sections 8 through 36

TOWNSHIP 20 SOUTH, RANGE 30 EAST, NMPM
Sections 1 through 36

TOWNSHIP 20 SOUTH, RANGE 31 EAST, NMPM
Sections 1 through 36

TOWNSHIP 20 SOUTH, RANGE 32 EAST, NMPM
Sections 4 through 9; Sections 16 through
21; and Sections 28 through 33

TOWNSHIP 21 SOUTH, RANGE 29 EAST, NMPM
Sections 1 through 36

TOWNSHIP 21 SOUTH, RANGE 30 EAST, NMPM
Sections 1 through 36

TOWNSHIP 21 SOUTH, RANGE 31 EAST, NMPM
Sections 1 through 36

TOWNSHIP 22 SOUTH, RANGE 29 EAST, NMPM
Sections 1 through 36

TOWNSHIP 22 SOUTH, RANGE 30 EAST, NMPM
Sections 1 through 36

TOWNSHIP 23 SOUTH, RANGE 29 EAST, NMPM
Sections 1 through 3; Sections 10 through
15; Sections 22 through 27; and Sec-
tions 34 through 36

TOWNSHIP 23 SOUTH, RANGE 30 EAST, NMPM
Sections 1 through 19

exist a number of oil and gas pools which produce varying amounts of salt water.

(4) That the major portions of Clayton Basin and North Draw, broad depressions caused by the slumping of the surface due to the

removal of the underlying salt by solution, lie within the above-described area.

(5) That the general direction of movement of both ground water and surface water in the subject area is toward and into said basins, thence southwest in Nash Draw toward Malaga Bend.

(6) That a substantial amount of water is produced in conjunction with the production of oil or gas, or both, by the oil and gas wells located in the above-described area.

(7) That said produced water is presently being disposed of in surface pits located in the above-described area.

(8) That a number of large surface ponds, or lakes, containing extremely high concentrations of chlorides are located in the above-described area.

(9) That in relation to said surface lakes, said disposal pits are inconsiderable in volume of water received and seepage underground.

(10) That the aforesaid disposal pits and surface lakes are located within the same surface and subsurface drainage system, as described in Finding (5) above.

(11) That the purpose of Order No. R-3221, to afford reasonable protection against contamination of fresh water supplies by surface disposal of produced water, would not be advanced by the enforcement of said order as to the above-described area.

IT IS THEREFORE ORDERED:

(1) That all oil and gas wells, both existing and prospective, located in the following-described area are hereby excepted from the provision of Order (3) of Order No. R-3221, to authorize the operators of said wells to dispose of water produced in conjunction with the production of oil or gas, or both, from said wells in unlined surface pits located in said following-described area until further order of the Commission:

EDDY AND LEA COUNTIES, NEW MEXICO
TOWNSHIP 19 SOUTH, RANGE 30 EAST, NMPM
Sections 8 through 36

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Order No. R-3221-B

TOWNSHIP 20 SOUTH, RANGE 30 EAST, NMPM
Sections 1 through 36

TOWNSHIP 20 SOUTH, RANGE 31 EAST, NMPM
Sections 1 through 36

TOWNSHIP 20 SOUTH, RANGE 32 EAST, NMPM
Sections 4 through 9; Sections 16 through
21; and Sections 28 through 33

TOWNSHIP 21 SOUTH, RANGE 29 EAST, NMPM
Sections 1 through 36

TOWNSHIP 21 SOUTH, RANGE 30 EAST, NMPM
Sections 1 through 36

TOWNSHIP 21 SOUTH, RANGE 31 EAST, NMPM
Sections 1 through 36

TOWNSHIP 22 SOUTH, RANGE 29 EAST, NMPM
Sections 1 through 36

TOWNSHIP 22 SOUTH, RANGE 30 EAST, NMPM
Sections 1 through 36

TOWNSHIP 23 SOUTH, RANGE 29 EAST, NMPM
Sections 1 through 3; Sections 10 through
15; Sections 22 through 27; and Sec-
tions 34 through 36

TOWNSHIP 23 SOUTH, RANGE 30 EAST, NMPM
Sections 1 through 19

(2) That the Commission may by administrative order rescind such authority as to any or all such wells whenever it reasonably appears to the Commission that such rescission would serve to afford reasonable protection against contamination of fresh water supplies.

(3) That jurisdiction of this cause is retained for the entry of such further orders as the Commission may deem necessary.

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CASE No. 3806

Order No. R-3221-B

DONE at Santa Fe, New Mexico, on the day and year hereinabove designated.

STATE OF NEW MEXICO
OIL CONSERVATION COMMISSION

DAVID F. CARGO, Chairman

GUYTON B. HAYS, Member

A. L. PORTER, Jr., Member & Secretary

S E A L

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BEFORE THE OIL CONSERVATION COMMISSION
OF THE STATE OF NEW MEXICO

IN THE MATTER OF THE HEARING CALLED BY THE OIL CONSERVATION COMMISSION ON ITS OWN MOTION TO CONSIDER THE AMENDMENT OF ORDER No. R-3221, THE COMMISSION'S SALT WATER DISPOSAL ORDER, TO PERMIT THE EXEMPTION OF CERTAIN PRESENTLY EXISTING AND FUTURE POOLS IN EDDY AND LEA COUNTIES, NEW MEXICO, FROM CERTAIN REQUIREMENTS OF SAID ORDER.

CASE No. 3806
Order No. R-3221-B-1

NUNC PRO TUNC ORDER

BY THE COMMISSION:

It appearing to the Commission that due to clerical error, Order No. R-3221-B, dated July 25, 1968, does not correctly state the intended finding of the Commission in one particular,

IT IS THEREFORE ORDERED:

(1) That the phrase "North Draw" is hereby stricken from the first line of Finding (4) on Page 2 of Order No. R-3221-B, dated July 25, 1968, and the phrase "Nash Draw" is hereby substituted in lieu thereof.

(2) That this order shall be effective nunc pro tunc as of July 25, 1968.

DONE at Santa Fe, New Mexico, on this 13th day of August, 1968.

STATE OF NEW MEXICO
OIL CONSERVATION COMMISSION

DAVID F. CARGO, Chairman

GUYTON B. HAYS, Member

A. L. PORTER, Jr., Member & Secretary

S E A L

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BEFORE THE OIL CONSERVATION COMMISSION
OF THE STATE OF NEW MEXICO

IN THE MATTER OF THE HEARING CALLED BY THE OIL CONSERVATION
COMMISSION ON ITS OWN MOTION TO CONSIDER THE AMENDMENT OF
ORDER NO. R-3221, THE COMMISSION'S SALT WATER DISPOSAL ORDER.

CASE No. 3807
Order No. R-3221-C

ORDER OF THE COMMISSION

BY THE COMMISSION:

This cause came on for hearing at 9 a.m. on July 17, 1968,
at Santa Fe, New Mexico, before the Oil Conservation Commission
of New Mexico, hereinafter referred to as the "Commission."

NOW, on this 10th day of September, 1968, the Commission,
a quorum being present, having considered the testimony presented
and the exhibits received at said hearing, and being fully advised
in the premises,

FINDS:

(1) That due public notice having been given as required by
law, the Commission has jurisdiction of this cause and the subject
matter thereof.

(2) That effective upon various dates, Orders (1), (2),
and (3) of Commission Order No. R-3221, dated May 1, 1967,
prohibits, in that area encompassed by Lea, Eddy, Chaves, and
Roosevelt Counties, New Mexico, the disposal, subject to minor
exceptions, of water produced in conjunction with the production
of oil or gas, or both, on the surface of the ground, or in any
pit, pond, lake, depression, draw, streambed, or arroyo, or in
any watercourse, or in any other place or in any manner which
will constitute a hazard to any fresh water supplies and said
disposal has not previously been prohibited.

(3) That Order (4) of said Order No. R-3221 authorizes
limited utilization of unlined surface pits in areas not affected
by Orders Nos. R-1224-A, R-2526, R-2788, or R-3164.

(4) That Order (5) of said Order No. R-3221 authorizes utilization of certain impervious lined pits in use at the effective date of said order.

(5) That Order (8) of said Order No. R-3221 authorizes temporary disposal in surface pits during certain contingencies.

(6) That in order to provide more uniform provisions among the various salt water disposal orders of the Commission and to ease the administration of said orders, Order (4) of said Order No. R-3221 should be amended to also authorize, in those areas subject to Orders Nos. (1) and (3) of said Order No. R-3221, the utilization of unlined surface pits in those areas affected by Orders Nos. R-1224-A, R-2526, and R-3164 for the disposal of limited quantities of water.

(7) That the utilization of lined evaporation pits is feasible and in the interest of good conservation practices, provided they are properly designed, constructed, and maintained.

(8) That the utilization of properly designed, constructed, and maintained lined evaporation pits should be authorized in all areas subject to Order No. R-3221.

(9) That in order to prevent waste caused by burdensome delay or expenses upon operators of development wells, Order (8) of said Order No. R-3221 should be amended to also authorize temporary storage or disposal of water in surface pits during the evaluation of all newly completed wells.

IT IS THEREFORE ORDERED:

(1) That Order (4) of Order No. R-3221 is hereby amended to read in its entirety as follows:

"(4) That in those areas subject to the provisions of Orders Nos. (1) and (3) above, and in those areas affected by Orders Nos. R-1224-A, R-2526, and R-3164, surface pits may be utilized for the disposal of a maximum of one barrel of produced water per day for each developed 40-acre tract served by said pits, provided however, that in no event shall said surface pit disposal exceed 16 barrels per day, and provided further, that this authorization shall not apply to that area affected by Order No. R-2788."

(2) That Order (5) of Order No. R-3221 is hereby amended to read in its entirety as follows:

"(5) The use of lined evaporation pits in those areas affected by Orders Nos. (1), (2), and (3) above, and in those areas affected by Orders Nos. R-1224-A, R-2526, R-2788, and R-3164 is hereby prohibited except as follows:

"A. Continued disposal of water in impervious lined pits which were previously constructed pursuant to the provisions of Orders Nos. R-1224-A, R-2526, R-2788, and R-3164, and which were inspected and approved by a Commission representative prior to use, shall be permitted after October 10, 1968, only if the operator of any such lined pit shall have obtained a permit for such use from the appropriate district office of the Commission. The permit shall be applied for in accordance with the provisions of paragraph B-8 below and shall be valid only for so long as the pit is properly maintained to ensure its continued imperviousness.

"B. ~~Under certain circumstances, the District Supervisor of the appropriate district office of the Commission may issue a permit authorizing the use of newly constructed lined pits for evaporation or storage of produced water.~~

To qualify for and to sustain a permit authorizing the operator to utilize newly constructed lined evaporation pits:

1. ~~The lease or leases served by the installation should have a settled or decreasing rate of water production.~~
2. ~~The installation must provide adequate storage capacity to safely contain all water produced, taking into account those months during which evaporation rates are normally at their minimum and must provide at least 600 square feet of evaporative surface for each barrel (42 U.S. Gallons) of produced water to be placed in said pit on a daily average basis throughout the year.~~
3. ~~The installation must provide a header pit, or other appropriate scheme, lined with a suitable oil-resistant material to trap any oil carried with the water, constructed and operated in a~~

manner to prevent said oil from reaching the evaporation pit, and the surface of the evaporation pit must be maintained free of oil.

4. Evaporation and header pits must be constructed with underlying gravel-filled sumps and laterals, or other suitable devices, for the detection of leakage; the Commission shall be given an opportunity to inspect same prior to being lined with an impervious material, at least 30 mils in thickness, which is resistant to hydrocarbons, salts, and aqueous acids and alkalis. The material must also be fungus- and rot-resistant and must be sun-resistant, or provision made to protect it from the sun.
5. Each lined pit installation shall be identified by a sign, posted on or near said installation which shall show the name of the lease, name of the operator, the location by quarter-quarter section, township and range, and the permit number of the permit authorizing the installation. In addition, the installation must be adequately fenced, with the corners securely braced, and the fence maintained in good repair.
6. Whenever there is evidence that leakage is occurring, the pit or pits must be emptied and repaired to the satisfaction of the Commission before disposal therein may be resumed.
7. Any salt remaining in a lined pit must, upon termination of use of said pit for disposal purposes, be disposed of in a manner that will afford reasonable protection against contamination of fresh water supplies, and the pit shall be filled, leveled, and compacted as soon as practicable after termination of such use.
8. Application for a permit to utilize a lined evaporation pit shall be in triplicate on a form prescribed by the Commission (a copy of which is attached hereto and made a part hereof as Exhibit "A") and shall be filed with and

approval obtained from the District Supervisor of the appropriate district office of the Commission prior to commencement of construction. Application forms and minimum specifications for the design and construction of lined evaporation pits are available at the district and Santa Fe offices of the Commission.

"C. The Commission may from time to time make such tests and require the furnishing of such evidence as it deems necessary to determine that any lined evaporation pit is maintained in satisfactory condition. The Commission may suspend or revoke by administrative order the permit authorizing a lined evaporation pit whenever it reasonably appears to the Commission that such suspension or revocation would serve to protect fresh water supplies from pollution."

(3) That Order (8) of Order No. R-3221 is hereby amended to read in its entirety as follows:

"(8) That the District Supervisor of the appropriate district office of the Commission is hereby empowered to authorize temporary storage or disposal in surface pits for a period not to exceed 30 days during such contingencies as injection system failures and evaluation of newly completed wells. Authority for said disposal shall only be granted on an individual case basis and only after the volume and quality of the water produced and the proximity of fresh water supplies have been taken into consideration. Any unlined pit used for temporary storage during an emergency must be emptied as soon as the emergency is ended."

(4) That Order (9) of Order No. R-3221 is hereby amended to read in its entirety as follows:

"(9) That subject to the provisions of Orders Nos. (4) and (5) above, the provisions of this order are in addition to the provisions of Orders Nos. R-1224-A, R-2526, R-2788, and R-3164 of the Commission and nothing herein contained shall be construed as abridging or altering in any manner the provisions of said orders."

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CASE No. 3807

Order No. R-3221-C

(5) That jurisdiction of this cause is retained for the entry of such further orders as the Commission may deem necessary.

DONE at Santa Fe, New Mexico, on the day and year hereinabove designated.

STATE OF NEW MEXICO
OIL CONSERVATION COMMISSION

DAVID F. CARGO, Chairman

GUYTON B. HAYS, Member

A. L. PORTER, Jr., Member & Secretary

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APPLICATION FOR PERMIT

PERMIT NO. _____

TO UTILIZE A LINED EVAPORATION PIT

New Mexico Oil Conservation Commission

Name of Operator _____

Address _____

Name of lease upon which evaporation pit will be located _____

Location of evaporation pit: Unit Letter _____ Section _____ Township _____ Range _____

Lease(s) which will be producing into pit _____

Pool(s) which will be producing into pit _____

Analysis of disposal water: Chlorides _____ ppm. Total dissolved solids _____ ppm.
(If more than one pool will be producing into pit, give water analysis for each pool.)

Quantity of water to be disposed of into this pit _____ barrels per day.

Water production from these same wells six months ago _____ bpd. Three months ago _____ bpd
(If more than one pool will be producing into pit, give water production data for each)

Method of hydrocarbon entrapment to be employed: Settling tank _____ Header pit _____

If settling tank is to be used, give size and number of barrels _____

) If header pit is to be used, give dimensions and depth _____

Header pit lining material _____ Thickness _____

Dimensions of Evaporation Pit ("A" and "B" on diagram) _____

Number of square feet contained in above _____

Depth (Top of levee to floor of pit--"D" on diagram) _____

Material to be used as liner _____ Thickness _____

Does manufacturer recommend protection of material from direct sunlight? Yes _____ No _____

If yes, what means will be provided to so protect the material? _____

Is material resistant to hydrocarbons? Yes _____ No _____

Is material resistant to acids and alkalis? Yes _____ No _____

Is material resistant to salts? Yes _____ No _____

Is material resistant to fungus? Yes _____ No _____

Is material rot-resistant? Yes _____ No _____

Will joints in material be fabricated in the field? Yes _____ No _____

If yes, describe method to be used in joining material _____

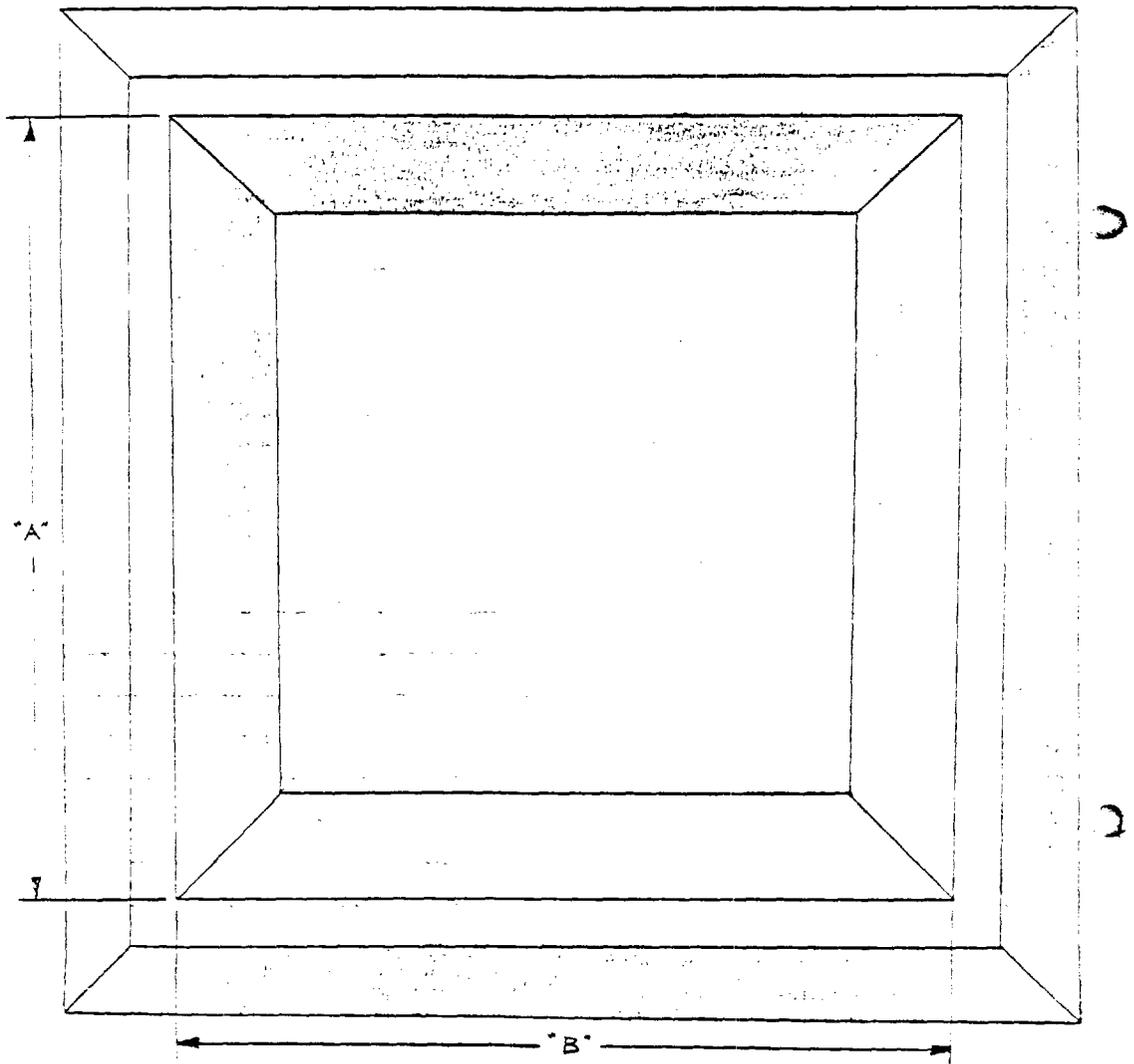
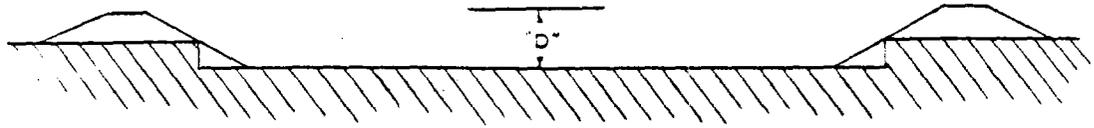
Attach manufacturer's brochure describing the qualities of the lining material.

Describe the leakage detection system to be used _____

I hereby certify that the information contained herein is true and complete to the best of my knowledge and belief, and further, that the subject evaporation pit and appurtenances, when installed, will be kept in good repair, and that all due diligence will be exercised in keeping the surface of the water free of oil and other debris.

Name _____ Title _____ Date _____

Approved by _____ Title _____ Date _____



Vol. 23, No. 17, April 26, 1984

customer whose business, along with a number of other persons with Insurors not to write for three years; the agreement. Danz had accepted the contract within that period. Testimony regarding similar transactions that Danzer had violated the June Insurors tendered that breach, and re-awarding compensation damages on its own refused.

Although Insur breach of the June matter was litigated Civ.P.R. 15(b) (Reple. 1980), provides that issues tried by express or implicit consent of the parties shall be treated as if they had been raised by the pleadings, and a failure to move to so amend "does not affect the result of the trial of these issues." See also *Fidelity Nat. Bank v. Lobo Hijo Corp.*, 92 N.M. 737, 594 P.2d 1193

Supreme Court Opinions, Chief Justice Federico (Ct.App.1979).

Duke City - Sub. Opinion

Since the evidence conclusively established that one of the respondents on the same day as to each of the respondents so also the judgment with days' notice of the respondent's is of

1981.
IT IS SO ORDERED.

s/MARY C. WALTERS, Justice

WE CONCUR:
s/WILLIAM RIORDAN, Justice
s/HARRY E. STOWERS, Jr., Justice

From The New Mexico Supreme Court

DUKE CITY LUMBER COMPANY,
Petitioner.

versus

**NEW MEXICO ENVIRONMENTAL IMPROVEMENT BOARD and
NEW MEXICO ENVIRONMENTAL IMPROVEMENT DIVISION,**
Respondents.

No. 15078 (filed April 4, 1984)

ORIGINAL PROCEEDING ON CERTIORARI
Administrative Appeal

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CLINTON
WILLIAM R. FEDERICI, Chief Justice.
The opinion of this Court heretofore filed on November 23, 1983 is withdrawn and the following opinion

is substituted therefor.
Duke City Lumber Company, petitioner, applied to the New Mexico Environmental Improvement Board (Board) for a one year variance from Air

Quality Control Regulation 402(A). The application was denied, and the denial was reviewed by the Court of Appeals. The court remanded "with instructions to the Board to conduct further proceedings to determine whether the wood smoke, in the volume being emitted from appellant's wigwam burner is 'injurious to health or safety.'" **Duke City Lumber Co. v. New Mexico Environmental Improvement Board**, 95 N.M. 401, 407, 622 P.2d 709, 715 (Ct.App.1980), cert. denied, 95 N.M. 426, 622 P.2d 1046 (1981) (emphasis added).

The Board held a second hearing and again denied the application for a variance. Duke City Lumber Company again appealed. The Court of Appeals, affirming the Board's second denial of the variance, held that a condition injurious to health or safety does not mean actual harm, but "only a condition that **tends** to cause harm to health or safety."

The Court of Appeals also upheld, but questioned, the current New Mexico standard of judicial review which is limited to substantial evidence in the record. We granted certiorari.

The questions presented are:

I. Whether judicial review of an Environmental Improvement Board decision should be limited to the current standard of substantial evidence in the record, or whether this Court should now adopt a "whole record review" standard.

II. Whether the showing by the Environmental Improvement Division that wood smoke from Duke City Lumber's woodwaste burner **tends** to cause harm is sufficient to constitute "emission . . . as may with reasonable probability injure human health," as required by NMSA 1978, Section 74-2-2(B) (Repl.Pamp.1983).

III. Whether there was competent evidence to support a finding that wood smoke from Duke City's woodwaste burner is injurious to health or safety.

I. Substantial Evidence Rule.

The Air Quality Control Act (Act), NMSA 1978, Sections 74-2-1 through 74-2-17 (Repl.Pamp.1983), specifically provides for judicial review of the Act. Section 74-2-9 states: "Any person to whom the board denies a variance, after a hearing, may appeal

to the court of appeals. All appeals shall be upon the record made at the hearing" The Court of Appeals may set aside the Board's denial of a variance if it is found to be arbitrary, capricious or an abuse of discretion, not supported by substantial evidence in the record, or otherwise not in accordance with law. *Id.*

The separation of powers doctrine directs administrative agencies to their duty of implementing legislation. The Legislature grants agencies the discretion of promulgating rules and regulations which have the force of law. The agencies must also determine whether there has been compliance with administrative decisions, and this is an adjudication. Therefore, agencies exercise in part functions of all three branches of government. This requires a carefully defined standard of review by the courts.

Some statutes, such as the New Mexico Administrative Procedures Act, provide for review of agency actions on the "entire record." NMSA 1978, Section 12-8-22(A). Other statutes, such as the New Mexico Air Quality Control Act before us, do not contain the words "entire record" or "as a whole." NMSA 1978, Section 74-2-9 (Repl.Pamp.1983). Certainly, in cases arising under statutes that expressly require whole record review of administrative decisions, the New Mexico Supreme Court has applied it. **Young v. Board of Pharmacy**, 81 N.M. 5, 462 P.2d 139 (1969). Also, in certain cases, even without express statutory provision for whole record review, we have engaged in a review of the record as a whole. **Ribera v. Employment Security Commission**, 92 N.M. 694, 696, 594 P.2d 742, 744 (1979) (deciding "based upon all of the evidence"); **Transcontinental Bus System v. State Corporation Commission**, 67 N.M. 56, 60, 352 P.2d 245, 247-248 (1959) (reviewing "the testimony of the witnesses, considered as a whole"); **Garrett Freight Lines v. State Corporation Commission**, 63 N.M. 48, 312 P.2d 1061 (1957) (considering testimony presented by both sides).

When the Legislature does not specify either standard the courts in New Mexico have followed the rule of

substantial evidence in the record, requiring the reviewing court to determine whether the record contains substantial evidence to support the agency decision and to ignore evidence to the contrary. Substantial evidence is "such relevant evidence as a reasonable mind might accept as adequate to support a conclusion." **Rinker v. State Corporation Commission**, 84 N.M. 626, 627, 506 P.2d 783, 784 (1973).

The United States Supreme Court addressed whole record review as early as 1951. It held that courts are to review and consider not only evidence in support of one party's contention to determine whether there was substantial evidence to support the agency finding, but courts are to look also to evidence which is contrary to the finding. The reviewing court would then decide whether on balance, the agency's decision was supported by substantial evidence. **Universal Camera Corp. v. NLRB**, 340 U.S. 474 (1951). The federal courts require administrative findings of fact to be supplemented by substantial evidence, and also that those findings be reviewed on the whole record. **Committee for an Independent P-1 v. Hearst Corp.**, 704 F.2d 467 (9th Cir.1983); **United States Soil Conditioning v. NLRB**, 606 F.2d 940 (10th Cir.1979).

The Court of Appeals was correct in applying to this case the more limited standard of review and ignoring all evidence unfavorable to the Board's decision. We have previously held that it is not proper for the Court of Appeals to change the standard of review, but their opinion properly addresses the shortcomings of this limited review and we agree with the Court of Appeals that it should be changed, insofar as administrative boards and agencies are concerned. See **Alexander v. Delgado**, 84 N.M. 717, 507 P.2d 778 (1973). New Mexico's present standard of review is not only outdated, but contrary to the rule followed by a majority of other jurisdictions and by the federal courts. Limited by our prior opinions the Court of Appeals concluded that because Section 74-2-9 provides for reversal of the Board's denial of variance if found not to be supported

by substantial evidence in the record as opposed to substantial evidence in the record as a whole it was compelled to ignore strong evidence of medical effects of wood smoke and deficiencies in the air quality model which was before the Board. Not only does this interpretation shroud the judgment of the reviewing courts with imposed ignorance of enlightening evidence, but it also causes uneven treatment among those who seek review of the actions of various administrative boards and agencies.

We have continued to follow the rule that if there is substantial evidence in the record to support a finding, the reviewing court is bound thereby. Also, in deciding whether the finding has substantial support, the court must view evidence in the light most favorable to support the findings, and any evidence unfavorable to the finding will not be considered. **Trujillo v. Romero**, 82 N.M. 301, 481 P.2d 89 (1971); **Tapia v. Panhandle Steel Erectors Company**, 78 N.M. 86, 428 P.2d 625 (1967). This Court has said it will not weigh conflicting evidence or determine credibility of witnesses. **Lujan v. Pendaries Properties, Inc.**, 96 N.M. 771, 635 P.2d 580 (1981); **Worthey v. Sedillo Title Guaranty, Inc.**, 85 N.M. 339, 512 P.2d 667 (1973). We confirm this rule for the judicial review of orders and judgments of trial courts.

However, for administrative appeals we now expressly modify the substantial evidence rule as heretofore adopted by this Court and supplement it with the whole record standard for judicial review of findings of fact made by administrative agencies. A review of the whole record is clearly indicated in those cases where the administrative agency serves not only as the factfinder but also as the complainant and prosecutor. See 73A C.J.S., **Public Administrative Law and Procedure** Section 213 (1983).

The new standard which we have pronounced in this case has been previously considered and applied by this Court in **New Mexico Human Services Department v. Garcia**, 94 N.M. 175, 608 P.2d 151 (1980), where we stated:

Whether the decision by HSD is

supported by substantial evidence in the [record as a] whole, is one of the standards for judicial review of administrative decisions by HSD as required by Section 27-3-4(F), N.M.S.A. 1978. The language employed in the statute does not abrogate the substantial evidence rule as that rule has existed in New Mexico. The language does point to the fact that the substantial evidence rule must be applied to the **entire record** and that segments of the record may not be ignored in applying the rule. The statute does not mean that upon judicial review of the findings by HSD, the Court may reweigh the evidence and reassign the preponderance of evidence.

In order to determine whether the decision by HSD is supported by substantial evidence in the record as a whole, we must view the evidence in the light most favorable to the decision by HSD.

While this rule is applicable to decisions of administrative boards and tribunals as well as to decisions of courts, it does not permit accepting part of the evidence and totally disregarding other convincing evidence in the record considered as a whole. Because of the minor departure from the customary substantial evidence rule in reviewing administrative decisions where the record as a whole must be considered, the reviewing court may act on other convincing evidence in the record and may make its own findings based thereon.

Id. at 176-177, 608 P.2d at 152-153 (emphasis added).

II. Tends to Cause Harm.

This case was originally remanded by the Court of Appeals to the agency solely to determine whether the wood smoke Duke City emitted was "injurious to health or safety." **Duke City Lumber Co. v. New Mexico Environmental Improvement Board**, 95 N.M. 401, 407, 622 P.2d 709, 715 (Ct. App.1980), cert. denied, 95 N.M. 426, 622 P.2d 1046 (1981).

The Air Quality Control Act defines air pollution as, "the emission . . . into the outdoor atmosphere of one or

more air contaminants in such quantities and duration as may with reasonable probability injure human health . . . NMSA 1978, Section 74-2-2(B) (Repl.Pamp.1983).

The Board may grant a variance when compliance with air quality regulations will result in arbitrary and unreasonable taking of property or will impose an undue economic burden, and will not result in a condition injurious to health or safety. NMSA 1978, Section 74-2-8 (Repl.Pamp.1983). This definition does not permit the denial by the Board of a variance upon a mere showing that a condition "tends to cause harm." The Board has power under the New Mexico Air Quality Control Act to deny the variance when the air pollution that would result from granting a variance would with "reasonable probability" injure health. The Legislature has provided the standard of reasonable probability, and this Court must adhere to it. The Board and the Court of Appeals erred by expanding the statutory definition of air pollution from emissions from a "reasonable probability" of injuring health to a condition which "tends to cause harm." The result we have reached under this Point II would be the same whether we applied the old substantial evidence test or the new whole record review standard.

III. Sufficiency of the Evidence (Residuum Rule).

Administrative hearings in this case were conducted under relaxed rules of procedure to obtain information upon which to arrive at a definitive result. Town meeting-type hearings were held by the Board to solicit public participation regarding Duke City's request for a variance. Unsworn testimony of citizens and testimony of an Espanola physician was presented in support of the Board's decision that smoke from Duke City's mill would be injurious to health. Testimony by the physician was general rather than specific. He referred to and read from source material but he did not relate to or apply that material to the facts in this case. Lay witnesses and the doctor testified that smoke from the burner caused asthma attacks and irritation of eyes, nose and throat.

The New Mexico Administrative

Procedures Act provides that evidence may be relied upon "if it is of a type commonly relied upon by reasonably prudent men in the conduct of their affairs." NMSA 1978, Section 12-8-11(A). The standard for admissibility in an administrative hearing under this Act is therefore one of whether the evidence has any probative value. However, New Mexico courts require that an administrative action be supported by some evidence that would be admissible in a jury trial. This has been referred to as the legal residuum rule. *Young v. Board of Pharmacy*, 81 N.M. 5, 462 P.2d 139 (1969). New Mexico has continued to require a residuum of competent evidence to support the findings of an administrative agency where a substantial right is at stake. *Trujillo v. Employment Security Commission*, 94 N.M. 343, 610 P.2d 747 (1980).

Our decision in this case to review the whole record does not negate the residuum rule. The substantial evidence rule and the whole record standard which we now adopt reaffirm the rule that some competent evidence is required to support an action by an administrative agency which affects a substantial right.

In this opinion we have not addressed the question of whether the Duke City burner emissions exceeded the NAAQS. Neither have we made a determination as to whether violation of this standard alone, or in

conjunction with medical evidence presented at trial, justifies denial of a variance. The agency decision on these questions should now be reviewed by the Court of Appeals, based upon the standard of review which we have pronounced in this opinion.

The Court of Appeals is reversed and the cause remanded to it for further proceedings consistent with this opinion.

IT IS SO ORDERED.

s/WILLIAM R. FEDERICI, Chief Justice

WE CONCUR:

s/WILLIAM RIORDAN, Justice
s/HARRY E. STOWERS, JR., Justice
s/MARY C. WALTERS, Justice

DAN SOSA, JR., Senior Justice,
dissenting.

DISSENT

DAN SOSA, JR., Senior Justice.

After rehearing was granted and the new majority opinion filed, I still cannot agree with the majority for the reason that the new standard announced allows this Court to substitute its judgment for the lower court or administrative body with impunity. I agree that the Court of Appeals should review the matter. To that extent, I favor this opinion over the previous opinion which merely reversed.

s/DAN SOSA, JR., Senior Justice

The Supreme Court of New Mexico

MANZANO INDUSTRIES, INC.,
Plaintiff-Appellant,

versus

EDWARD MATHIS, et al.,
Defendants-Appellees.

No. 15103 (filed April 4, 1984)

APPEAL FROM THE DISTRICT COURT OF DONA ANA COUNTY
JAMES T. MARTIN, JR., District Judge

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Transport of Organic Compounds Dissolved in Ground Water

by Joan M. Newsom

Abstract

Organic compounds, such as trichloroethylene (TCE) and chlorobenzene, that have been found in drinking water supplies are of public concern because they are possibly carcinogenic. These substances can now be routinely detected in trace amounts with gas chromatograph mass spectrometers. There are some polar organic compounds, which are not detectable individually by common methods and therefore little is known about them.

The transport of organic compounds is more difficult to predict than the flow of ground water because:

- Trace amounts of pollutants are difficult to measure
- Transport is complicated if the compound is partitioned into several phases
- The concentration of organics in ground water may vary due to aquifer heterogeneity and other hydrologic factors
- Reactions with other organic compounds and reactions with the aquifer material (such as adsorption) may affect the mobility of the organics
- Biodegradation may also affect net transport.

Adsorption is a factor in the attenuation of non-polar organics in aquifers with significant organic content (>0.1 percent organic carbon). The organic material adsorbs the non-polar organic chemicals. The mobility of a pollutant in such an aquifer depends on at least two parameters: the levels of dissolved organic matter and the content of organic carbon in the aquifer material. The partition coefficient of the chemical pollutant between the aquifer and water is commonly calculated as a function of the organic content of the aquifer and the partition coefficient between octanol and water.

Field and laboratory results reported in the literature indicate that the following organic compounds may be biodegradable under aerobic conditions: alkyl benzenes and chlorobenzenes. Under anaerobic conditions halogenated aliphatics, alkyl benzenes, several pesticides and phenolic compounds may be biodegradable. Halogenated aliphatics appear not to degrade under aerobic conditions and non-chlorinated aromatics and chlorobenzenes appear not to degrade under anaerobic conditions. Alkyl benzenes biodegrade more rapidly than their halogenated counterparts.

Introduction

Pollution of ground water by organic compounds is an important area of public concern, and hydrogeologists are increasingly required to evaluate hydrocarbon contamination in the subsurface. The methods of analysis have improved in recent years such that concentrations of less than one microgram per liter ($\mu\text{g/L}$) can be determined. The ability to measure more organic compounds, especially polar organics, will increase the number of different contaminants detectable in water.

Some of the organic compounds found in water are believed to be harmful in trace amounts. The health risks of the synthetic organics, however, are difficult to determine mainly because of the uncertainty in extrapolating the results of laboratory carcinogen tests on lab animals to humans. The health risks are not likely to become known very rapidly. References on health aspects of synthetic organics are found in Pearson (1982a, 1982b), and Merian and Zander (1982).

Man-made hydrocarbons are used in a wide range of industries and in household products. They are for the most part a product of technology used since the 1940s. Their solubility in non-polar substances and poor solubility in water account for their common and widespread use as degreasers. Trichloroethylene (TCE) is used, for example, to clean oil from industrial machines, to wash oils from airport runways, and to remove grease from clothes in dry cleaning.

Definitions

Hydrocarbon compounds, also called organic compounds, are composed of hydrogen and carbon. Aliphatic hydrocarbons are a group of hydrocarbons in which the carbon atoms are joined to form open chains. Aromatic hydrocarbons usually have structures that contain at least one benzene ring. Monocyclic aromatics, such as alkyl benzenes, have one ring. Polynuclear hydrocarbons possess more than one ring. This class of hydrocarbons can be divided into two groups. In the first, the rings are fused, which means at least two carbon atoms are shared between adjacent rings, e.g., naphthalene. In the second group, the aromatic rings are joined directly or through a chain of at least one carbon atom, e.g., biphenyl.

Many of the organic pollutants are halogenated;

that is, they contain halogen atoms in their molecular structure. Chlorine, bromine and fluorine are the most common halogens. Examples of halogenated aliphatics found in ground water include: trichloroethylene ($\text{ClCH}_2\text{CCl}_2$, commonly abbreviated TCE), which contains two carbon atoms joined by a double bond; 1,1,1-trichloroethane (CH_3CCl_3), which contains two carbon atoms joined by a single bond; and tetrachloroethylene ($\text{Cl}_2\text{C}=\text{CCl}_2$, commonly abbreviated PCE), which contains two carbon atoms joined by a double bond. Trihalomethanes (THMs) are a subgroup of the halogenated aliphatics that contain three halogens in the methane (CH_4) molecular structure. Examples include chloroform or trichloromethane (CHCl_3), bromoform or tribromomethane (CHBr_3), and dibromochloromethane (CHBr_2Cl). Halogenated aromatics found in ground water include: chlorobenzene (ClC_6H_5), dichlorobenzene ($\text{Cl}_2\text{C}_6\text{H}_4$, abbreviated in this paper, DCB), and trichlorobenzene ($\text{Cl}_3\text{C}_6\text{H}_3$, abbreviated in this paper, TCB).

Hydrocarbon compounds can also be generally divided into polar and non-polar groups. Polar molecules are electrically neutral molecules with concentrations of negative charge in one part of the molecule and of positive charge in another, producing an electric dipole.

Occurrence of Organic Pollutants in Ground Water

The extent of ground water pollution by organic compounds is difficult to estimate both for a given aquifer and in general. Specific studies are difficult to compare because of variations in analytical sensitivity and differences among the compounds studied. Even for a given aquifer, the extent of ground water pollution by organic compounds can only be estimated because such a small fraction of the ground water is usually sampled.

There are many sources of organic pollution. Contaminants may reach the aquifer by way of precipitation, by seepage of pesticides and herbicides from the surface, from pollutants in sanitary landfills, waste storage ponds, polluted streams and lakes, and from accidentally or deliberately spilled material. Organic pollution is found both in industrial areas and in rural areas.

Man-made compounds pose a ground water pollution problem in industrialized countries. One or two percent of ground water supplies in the United States are polluted based on estimates of point sources, but only a fraction of these are contaminated primarily by organic pollutants (Pye and Patrick 1983). The compounds that occur most frequently in ground water in the United States are the trihalomethanes (THMs), which are the halogenated organics produced by chlorination of water containing humic materials (Bouwer et al. 1981). The problem of THMs, such as chloroform, has received considerable attention beginning in 1974 and the maximum contaminant level allowed by the EPA is 100 $\mu\text{g/L}$ total THMs (Cotruvo 1981).

The extent of ground water pollution by organics in the Netherlands was measured by sampling all 232 ground water pumping stations in the Netherlands between 1976 and 1978. The samples from 54 of the 232 locations, 25 percent of the locations, contained concentrations $>0.1 \mu\text{g/L}$ of chlorinated hydrocarbons with 1 or 2 carbons (e.g., TCE) (Zoeteman et al. 1981). The Netherlands is at the end of the Rhine River and

receives pollutants from countries upstream. The compounds detected most frequently at concentrations greater than 0.01 $\mu\text{g/L}$ in Dutch ground water include: TCE (67 percent), chloroform (60 percent), tetrachloromethane (43 percent), PER (19 percent), and 1,1,1-trichloroethane (17 percent). These compounds are on the Environmental Protection Agency list of priority pollutants. The concentrations at higher levels ($>10 \mu\text{g/L}$) could always be associated with a specific source, i.e., local waste dumping. Concentrations at low levels (0.01 to 0.1 $\mu\text{g/L}$) may be due to volatile organics in rain water. Levels of substances such as chloroform and TCE are less than 1 $\mu\text{g/L}$ in rain water in the Netherlands.

Measurements of Organic Pollutants

Accurate measurements of the concentrations of organic pollutants in ground water are essential for understanding the behavior of the pollutants in aquifers. The problems of sampling an aquifer are especially severe for volatile organics, which are easily lost to the atmosphere (e.g., Pankow et al. 1984). Problems can arise from the type of well construction and the type of casing used. A study of the leaching of trace organics (0.5 ppb naphthalene and 0.5 ppb p-dichlorobenzene) into water from five common plastics used in well casing showed the following results: Teflon® (no leaching detected), nonglued PVC (0 to 0.1 ppb), Polyethylene (0.1 ppb), Polypropylene (0.5 ppb), glued PVC (0.5 ppb), and Tygon (1.0 ppb) (Curran and Tomson 1983).

Analytical results may be suspect because of the difficulty of analyzing water for trace concentrations of organics. In a comparison of analyses among certified private, state and university labs, large variations were reported even for relatively simple measurements of total dissolved solids (Keith et al. 1983). The following procedures were used to control the analytical precision and accuracy during an extensive investigation of a PCB spill site (Roberts, Cherry and Schwartz 1982). The concentrations of PCBs were determined by several analytical techniques. A standard with PCB concentrations similar to the samples being analyzed was run approximately every ten samples. Blanks were run during a switch from analysis of high PCB concentrations to low concentrations to ensure that the residual response of the system had returned to background levels.

The occurrence of some polar organic compounds in ground water has been much less studied than that of non-polar organic compounds. Very little is known about their health risk or their occurrence because they cannot be easily isolated and measured. The group parameter TOX (total organic halogen) provides a measure of the total amount of halogen in organic compounds and is determined by concentrating the organics by adsorption, and measuring halogen concentrations by titration, specific ion electrodes, or microcoulometer. TOX analyses are both relatively simple and quick compared to gas chromatography. The more polar, non-volatile and high molecular weight halogenated hydrocarbons presently can be detected by TOX and not by GC/MS (Jeckel and Roberts 1980). Field studies have shown that the TOX concentration is several times larger than the sum of halogenated organic compounds by gas chromatographic determination (Roberts, Schreiner and Hopkins 1982).

Advection and Dispersion

The mechanisms of advection and dispersion have an important control on the transport of organic pollutants. Total solute flow in porous media is composed of the portion that travels with the average ground water flow (advection) and the portion that deviates from the average ground water flow (dispersion). Dispersion causes a dilution of the solute concentration and a spreading of the contaminated area. Seen as a plot of concentration vs. the time to reach an observation point, dispersion causes the S-shaped breakthrough curve to broaden. The characteristic length of the porous medium, which is known as the dispersivity length, when multiplied with the ground water velocity, has been shown in the lab to yield the dispersion coefficient. This coefficient is used to determine the flux due to dispersive effects (Anderson 1979).

There are two types of dispersion: dispersion that occurs at the pore scale (microdispersion) and dispersion that occurs at the field scale due to aquifer heterogeneity (macrodispersion). Microdispersion is usually of not much significance for transport in relatively fast-flowing ground water. On the other hand, microdispersion and molecular diffusion are important in underground waste isolation site studies. Macrodispersion is significant due to the heterogeneity of the aquifer (e.g., Sudicky et al. 1983).

Lab dispersivity measurements do not agree with dispersivity measurements determined by field tracer tests because of scale factors. Lab measurements of dispersivity values for calculating microdispersion consist of determining breakthrough times at the outlet of cylindrical columns packed with porous media and then using the solute transport equation to determine dispersivity values. The field measurements of longitudinal dispersivity (in the direction of flow), which are on the order of 10 to 100m, are at least three orders of magnitude larger than lab measurements, 10^{-4} to 10^{-2} m (Anderson 1979). Field tracer tests show that longitudinal dispersivity is not constant for a given aquifer, but increases as the distance between the injection and observation well is increased. At some point, dispersivity stops increasing. This increase in dispersivity with increased travel distance or travel time of the solute is referred to as the scale effect in the literature (e.g., Molz 1983; Sudicky et al. 1983).

The cause of the variable dispersivity is the heterogeneity of the aquifer, leading to anisotropic distributions of horizontal hydraulic conductivity. Field data indicate that most compounds prefer to travel through more permeable pathways, such as through gravel lenses. The variation in concentration due to heterogeneity of the aquifer causes the distribution of the compound in a horizontal sense to sometimes deviate from the theoretical plume shape derived for homogeneous aquifer characteristics (e.g., Sudicky et al. 1983).

The problem of aquifer heterogeneity is as important on a vertical scale as on a horizontal scale. Field data have shown that when chemicals enter the aquifers do not mix to the full vertical extent of the ground water and are influenced by aquifer heterogeneities and density effects (Sudicky et al. 1983; Rea and Upchurch 1980; Schwartz et al. 1982). Even though some of the data in these studies are for ions and not organic compounds, one would expect the principles to apply.

Organic pollutants dissolved in water infiltrated from the Glatt River into the upper approximately 9m of a 20m thick Quaternary glaciofluvial valley fill aquifer composed of sand and gravel (Schwarzenbach et al. 1983). The contaminated water was detected several kilometers from the Glatt River in the upper half of the aquifer, while water in the lower half originated from less polluted sources. Monitoring of a PER-spill in glacial deposits in Michigan showed that the PER (density = 1.62 g/cm^3 at 20 C), which was well below saturation, migrated downward as it traveled away from the source (Minsley 1983).

Adsorption

Most aquifers have less than 0.1 percent organic content. Quantitative relationships have not been well established between sorption and the controlling factors, although the specific surface area and the nature of the mineral surface influence the degree of sorption. Some adsorption of non-polar organic compounds was experimentally observed in columns containing materials that contain no organic carbon, such as clean sand, limestone and montmorillonite clay (Schwarzenbach and Westall 1981a). Sand and gravel aquifers are likely to contain insignificant amounts of organic matter, although this parameter is usually not measured. The aquifer near the Glatt River in Switzerland, for example, contains less than 0.1 percent organic content (Schwarzenbach et al. 1983). The retention of hexachlorobenzene, for example, was small between the aquifer next to the Glatt River and observation wells, which are up to 120m away from the river, despite the fact that hexachlorobenzene has a high log Kow of 6.06, and therefore, would be expected to be strongly retained in an aquifer with significant carbon content. The mobility of hexachlorobenzene indicates the low sorption capacity of sandy gravel aquifers with insignificant organic content (Schwarzenbach et al. 1983).

Aquifers comprised of deposits where former living matter is likely to have accumulated, such as from peat deposits, slow-moving streams, lakes or bogs, tend to have significant organic content. Studies have shown that at least 0.1 percent carbon content in the aquifer (0.001 g of organic carbon per gram sorbent) is needed for carbon adsorption to be significant (e.g., Schwarzenbach and Westall 1981a). Instead of solubility, the octanol:water partition coefficient (Kow) is often used as a measure of the partitioning of pollutants between water and organic phases. The Kow is the ratio of the concentration of a compound in octanol, a readily available alcohol that is relatively non-polar, to that in water. An inverse correlation between log Kow values (ranging between 1 and 6) and log solubility values, ranging between -3 to 5 in mg/L, has been found for non-polar organic compounds (Mackay 1980; Zoeteman et al. 1981). Kow values are also used to predict the partitioning behavior of compounds into soil that contains organic matter, as well as into the fat bodies of fish and other biota. Measured values of Kow can be found in: Chiou, Porter and Schmedding (1983); Banerjee, Yalkowsky and Valvani (1980); Kenaga and Goring (1980); and Hutzinger (1982); and estimated Kow values are found in Hansch and Leo (1979); and Leo, Hansch and Elkins (1971). In addition, chemical properties of organic compounds can be found in Verscheuren (1983), Hutzinger (1982, 1980), Weast and Astle (1982).

An example from California illustrates how the order of breakthrough of several organic compounds correlated with solubility and K_{ow} such that the compounds that appear first have the highest solubility and lowest K_{ow} . The order of appearance at an observation well 11 m downstream from the injection well from first to last to appear was: chloride, chloroform, bromoform and dibromochloroform, 1,1,1-trichloroethane and chlorobenzene (Roberts, Schreiner and Hopkins 1982).

In another example from western Canada, TCB concentrations increased relative to that of PCB with depth as shown by the increase in the 1,2,4-TCB/PCB ratio from 0.02 in the surface fill to 0.19 in the underlying Regina clay (Roberts, Cherry and Schwartz 1982). The log K_{ow} of 1,2,4-TCB is 4.05 (Leo, Hansch and Elkins 1971) while that of 2,4,5,2',4',5'-PCB is 6.72 (Schwarzenbach and Westall 1981a). The increased mobility of TCB is reflected by the lower K_{ow} . Other indications of greater mobility are higher solubility, lower molecular weight and fewer chlorine atoms in the molecular structure in TCB compared with PCB.

Useful relationships have been found between the adsorption behavior of a pollutant and its K_{ow} value and the organic content of an aquifer. Preliminary work indicates that the partitioning behavior of a pollutant and its residence time can be calculated for aquifers containing sufficient organic material. Karickhoff et al. (1979) demonstrated that the degree to which a compound is adsorbed in a soil, as measured by the partition coefficient (K_p), depends on the K_{ow} and the "fraction organic content" (f_{oc}) of the soil by the relation:

$$K_p = 0.63 f_{oc} (K_{ow}) \quad (1)$$

The equation was developed by examining the adsorption of 10 organic pollutants, whose log K_{ow} ranged from 2 to 6, in river and pond sediments whose f_{oc} ranged from 0.1 to 3.3 percent. This equation applies when the pollutant concentration is less than half of the solubility limit in water. Based on surface and aquifer sediments, whose f_{oc} is greater than 0.001, Schwarzenbach and Westall (1981a) derived a similar equation:

$$K_p = 3.2 f_{oc} (K_{ow}^{0.72}) \quad (2)$$

This equation is also valid only for low concentrations of the pollutant. Means et al. (1980) derived a similar equation for PAHs. Figure 1 illustrates the relationship described by Equation 2 for four chlorinated benzenes with different K_{ow} coefficients. The equations establish the similar dependence of the parameters f_{oc} and K_{ow} on the partition coefficient between soil containing organic matter and water. These equations apply only for non-polar substances in material with greater than 0.1 percent carbon. K_{ow} provides a better estimate of sediment-water partitioning than does solubility, which gives at best an order of magnitude estimate of the partitioning behavior of a chemical in the organic fraction of the sediment medium (Karickhoff et al. 1979).

Schwarzenbach and Westall (1981a) found that more than 85 percent of the adsorption of the pollutants took place on particles of size less than 0.125 mm (fine sand) and Karickhoff et al. (1979) observed that most of the adsorption took place on the particle fraction smaller than 0.05 mm (silt or clay). More organic

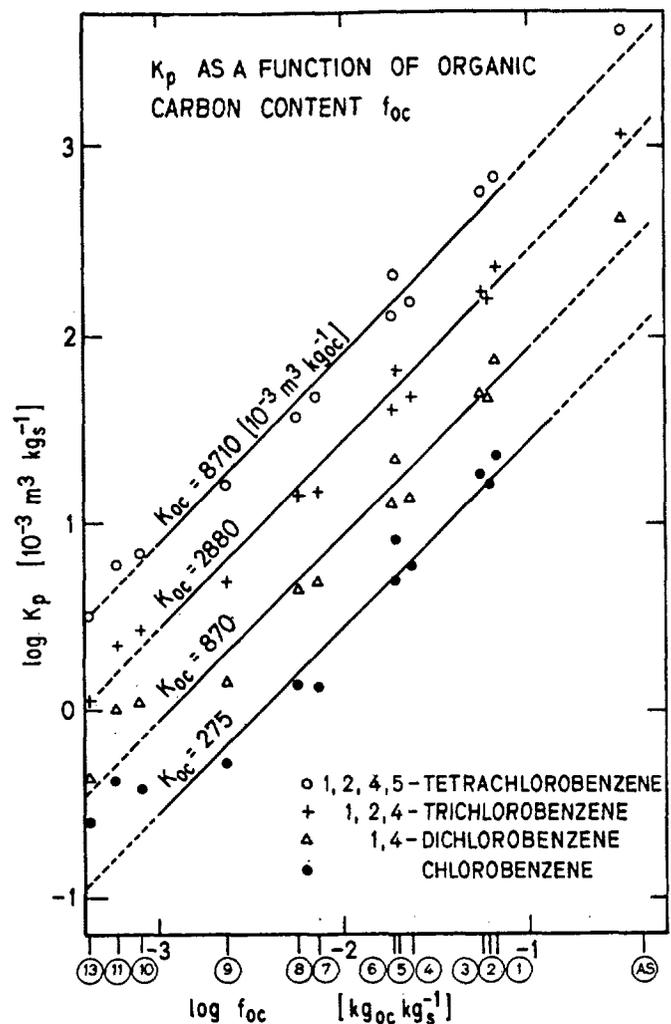


Figure 1. The sorbent to water partition coefficient (K_p) as a function of organic carbon fraction (f_{oc}) for four chlorobenzenes (Schwarzenbach and Westall 1981b). K_{oc} is the partition coefficient based on organic content and $K_{oc} = K_p/f_{oc}$. The circled symbols indicate the sorbents on which the data were obtained: AS, activated sludge; 1, 4, sea sediments (coastal zone); 2, detritus; 3, 5, lake sediments; 6, 8, river sediments; 7, 9, 10, 11, 13, aquifer material.

compounds were sorbed on the finer particle size fraction of sediments than on the coarse fraction principally because of the higher organic content as well as the larger surface area. Differences in sorption between silt and clay fractions depend on differences in f_{oc} rather than in sediment size (Karickhoff et al. 1979). Organic compounds also partition onto dissolved organic matter, such as fulvic and humic acids, such as in organic-rich water in landfill leachates (Cherry et al. 1984).

A pollutant that is adsorbed travels slower than the water containing the pollutant. The travel time of the solute divided by the travel time of the fluid is known as the retardation factor or the relative residence time (t_r), which based on Equation 1 is:

$$t_r = 1 + 0.63 f_{oc} (K_{ow}) \rho/\epsilon$$

where

ρ = average bulk density (g/cm^3)

ϵ = soil void fraction (unitless)

(Roberts, Reinhard and Valocchi 1982)

A comparison among τ values, which are dimensionless, calculated from the equation and those derived from the field show that τ values diverge for increasing values of K_{ow} . The τ values are 5 (field) and 6 (equation) for chloroform; 36 (field) and 41 (equation) for chlorobenzene; and greater than 200 (field) and 140 (equation) for 1,4-DCB (McCarty et al. 1981). K_{ow} values for these three compounds are 93, 692, and 2,400 respectively and the calculations are based on an average bulk density of 2 g/cm^3 , $\epsilon = 0.22$, and $f_{oc} = 1$ percent carbon (McCarty et al. 1981). Schwarzenbach et al. (1983) derived a similar equation but did not make a comparison with field results.

The common method of modeling the effects of sorption on solute transport is to assume that the solute and sorbent react in instant equilibrium, i.e., no kinetic effects, that the ratio of the sorbed solute to the solute dissolved in water is constant, i.e., linear isotherm, and that adsorption and desorption is a reversible process. The above equations are based on these assumptions.

Formulas for the calculation of limiting kinetic effects, non-linear isotherms and unequal sorption/desorption behavior are given in Miller and Weber (1984). Kinetic effects are important when the ground water velocity is too fast to allow equilibrium and the above equations are no longer valid. The ground water flow rate (approximately 0.014 cm/s) close to the Glatt River during storm water events was probably fast enough for kinetics to affect the transport of pollutants in the aquifer. Kinetic effects are also important when contaminants are newly introduced to a ground water system and when spike or plug contamination sources are appropriate. Under these conditions less material is sorbed onto the aquifer media and the material that is not sorbed travels farther. Kinetic effects were observed in column experiments when water containing chlorinated benzenes flowed through a column at a rate of 0.01 cm/s (Schwarzenbach and Westall 1981a, 1981b), which is well within the range of typical ground water velocities. The breakthrough times were faster than the breakthrough times of the same column experiment conducted at a velocity of less than 0.001 cm/s . The results of the column experiment at the slower rate (0.001 cm/s) matched those of an 18-hour long equilibrium batch experiment indicating that sorption equilibrium occurred at the slower rate.

Although numerous studies have shown that trace levels of dissolved organic compounds follow linear isotherms, one exception are trace levels of PCBs (Cherry et al. 1984). Non-linear isotherms are most likely to occur when the concentration of the dissolved solute nears the solubility limit. For example, at low concentrations (well below the solubility limit) pesticides showed linear isotherms, but at high concentrations several organic pesticides have very non-linear isotherms (Cherry et al. 1984).

An important source of data on adsorption is the treatment of waste water by artificial recharge of an aquifer. The advantage of studies on waste water recharge is that the rate and length of time that a contaminant was injected or allowed to infiltrate into the aquifer is known, in contrast to most pollution studies.

In one study, approximately 92 percent of the organics were removed from the waste water (Tomson et al. 1979). The highest initial concentration was only $4.05 \text{ } \mu\text{g/L}$ and the range in final concentrations was between 0.1 to $1 \text{ } \mu\text{g/L}$. Most removal rates for the 11

classes of compounds studied were between 90 to 100 percent, which included chloroaromatics and alkoxyaromatics, alkyl benzenes, naphthalenes, alcohols, ketones, indoles and indenenes. Those groups whose removal rate was below 90 percent include the alkylphenols (85 percent), alkanes (71 percent), and chloroalkanes (70 percent) and phthalates (2 percent). The phthalates was the only group not to exhibit a dramatic decrease in concentration, and it was concluded the observed decline of only 2 percent was in error. A study of dune infiltration in northern Holland actually showed a dramatic increase in phthalate concentration (Piet et al. 1981). Perhaps PVC tubing contamination influenced the phthalate concentrations in both cases.

Adsorption and volatilization were thought to be the significant transport mechanisms for the pollutants studied by Tomson et al. (1981). Biodegradation had a minimal impact for two reasons: (1) The injected fluid was effluent from an activated sludge plant and compounds that easily biodegrade would not have been present. (2) Biodegradation does not occur for low pollutant concentrations. Tomson found that in the lab sewage bacteria reduced 2,3-dimethylnaphthalene from 1.3 mg/L to $40 \text{ } \mu\text{g/L}$ in one day and that there was no further degradation for several days.

Under equilibrium conditions the net ratio of the rates of adsorption and desorption do not change and the reaction is said to be reversible. Sorption was reversible in several column studies (Schwarzenbach and Westall 1981a; Karickhoff et al. 1979). The reversibility of the reactions indicated that the initial removal of the compounds from solution was due to sorption and not to other factors such as biodegradation, which would cause the amount removed to be greater than the amount desorbed. A study by Horzempa and Di Toro (1983), however, showed that sorption of PCBs is not readily reversible under field conditions. The amount of sorption correlated with sediment surface area and organic content. The sorption effects were not felt to be attributable to biodegradation because PCBs are not readily biodegraded.

The restoration of aquifers depends upon the ability to remove contaminants adsorbed onto the subsurface material. One method is to flush the aquifer via injection and extraction wells. If the ground water velocity is too fast for equilibrium to be established, the concentration of the pollutant in ground water will decrease below the equilibrium concentration. Once the flushing stops, equilibrium conditions may become established and the concentration of dissolved pollutants may increase as desorption takes place. In such a case, the concentration of the pollutant at the extraction well decreases as the aquifer is flushed and then increases when the flushing is stopped. In addition to desorption during flushing as an important mechanism, the concentrations may also be affected by biodegradation rates of adsorbed, in-phase and dissolved pollutants.

Polar organics appear to be more mobile than non-polar organics, as shown by a study in an aquifer with significant amounts of organic carbon because they are poorly retained in the organic material in the soil (Roberts, Schreiner and Hopkins 1982). Piet et al. (1981) also found that the polar compounds were not as well adsorbed as non-polar compounds in soil column experiments using 50cm-long columns of soil composed of peat and sand layers. Those non-polar chlorine organics that were retained include: nitro-

benzene, nitrotoluene and chloronitrobenzene. Similarly, studies with granulated activated carbon (GAC) exhibit less adsorption of the polar organics than the non-polar organics.

Biodegradation

Biodegradation is the breakdown of chemical compounds by microorganisms and is controlled by such environmental parameters as temperature, pH, dissolved oxygen, Eh, salinity, nutrients, competing organisms, toxicity to organisms, and the concentrations of the organisms and compounds. Lab studies have shown that under steady-state conditions a pollutant must be present in concentrations of milligrams per liter to be broken down directly by microorganisms (McCarty et al. 1981). In a similar study it was found that the pollutant concentration must be at least 100 $\mu\text{g/L}$ to sustain a microbe population (Wilson and McNabb 1983). If the pollutant concentrations are not sufficiently high to sustain the microorganisms biodegradation will not occur (Kobayashi and Rittman 1982). Sewage bacteria reduced 2,3-dimethylnaphthalene from 1.3 mg/L to 40 $\mu\text{g/L}$ and no further reduction was observed for several days (Tomson et al. 1981). A lower limit for biodegradation of 10 $\mu\text{g/L}$ has also been found by Wilson and McNabb (1983). Trace levels of a compound can sometimes be broken down as a secondary result of the breakdown of another compound, which is present at much higher concentrations (Rittmann et al. 1980; McCarty et al. 1979).

Biodegradation depends on essential metabolic requirements, such as oxygenated water for aerobic processes. Metabolism can deplete the oxygen or other metabolic requirements in ground water at pollutant concentrations greater than 1,000 to 10,000 $\mu\text{g/L}$ (Wilson and McNabb 1983). Thus, pollutants at high concentrations may be only partially degraded when oxygen is depleted.

Results of lab and field biodegradation studies under aerobic and anaerobic conditions for different classes of organic pollutants are presented below. Most of the priority pollutants have been shown to be biodegradable under laboratory conditions (Kobayashi and Rittman 1982). This does not, however, mean that these pollutants are necessarily biodegradable under field conditions. Aerobic conditions generally occur in the unsaturated zone and may be found below the water table at shallow depths as well as at great depths (Winograd and Robertson 1982).

Halogenated Aliphatics. Field and lab results show that several halogenated aliphatics may biodegrade slowly under anaerobic conditions, but not under aerobic conditions. CH_2Cl_2 does, however, degrade under aerobic conditions (R. Schwarzenbach, personal communication 1983). Halogenated aliphatics at low concentrations in treated waste water decreased in concentration when injected into a coastal aquifer in California (Roberts, Schreiner and Hopkins 1982). THMs degraded 10 times faster than the other halogenated aliphatics although the rate of anaerobic degradation was slow for both. The THMs concentration declined from 100 $\mu\text{g/L}$ to less than 0.1 $\mu\text{g/L}$ at a rate of 0.03 per day. The decline was attributed to anaerobic biodegradation and not adsorption because the sorption capacity of the aquifer was saturated before the injection experiment began. Batch culture tests in the lab supported the field results that THMs degrade at low concentrations under anaerobic

conditions (Bouwer et al. 1981). Similarly the THM bromodichloromethane degraded slowly under anaerobic conditions of a shallow fluvial aquifer in Oklahoma (Wilson and Enfield 1983). Halogenated aliphatics that have been reported to biodegrade under anaerobic lab conditions include: TCE, trichlorethane, methyl chloride, chloroethane, dichlorobromoethane, vinylidene chloride, PER, methylene chloride and the THMs chloroform, dibromochloromethane, bromodichloromethane (Kobayashi and Rittman 1982).

No degradation was observed in studies of several compounds under anaerobic conditions, but the rate of degradation may have been too slow to be detected during the period of investigation. Bouwer et al. (1981) observed THMs but not TCE or PER to biodegrade in batch culture tests in the lab under anaerobic conditions. Wilson et al. (1983) did not observe degradation below the water table for several aliphatics: 1,2-dichloroethane, 1,1,2-trichloroethane, TCE or PER, but the period of study may not have been long enough to observe slow rates of degradation. Slow rates of degradation, therefore, cannot be ruled out. Similarly, Schwarzenbach et al. (1983) observed that TCE, PER, 1,1,1-trichloroethane, and hexachlorethane were persistent in the aquifer up to several kilometers away from the river, but the wide error bars on their figures may not rule out slow rates of degradation.

The decomposition of halogenated aliphatics under aerobic lab or field conditions has not been observed. No significant degradation of halogenated aliphatics (THMs, TCE, PER) was found under aerobic lab conditions (Bouwer et al. 1981; Bouwer and McCarty 1984). The persistence of chloroform, under aerobic conditions was reported in a study of ground water recharge, a study of chloroform passage through GAC columns, a study of bank filtration in Germany and a study of waste water percolation in soil columns (Bouwer et al. 1981). Wilson et al. (1983) in a field study in Oklahoma did not observe degradation of several halogenated aliphatics, 1,2-dichloroethane, 1,1,2-trichloroethane, TCE, or PER, above the water table.

Alkyl benzenes. Alkyl benzenes are known to degrade under aerobic conditions and may degrade under anaerobic conditions. Field observations show that toluene degraded rapidly in a shallow aquifer composed of flood-plain sediments in Oklahoma both above and below the water table (Wilson and Enfield 1979; Wilson et al. 1983). Schwarzenbach et al. (1983) observed a sharp decrease in non-halogenated compounds transported from the Glatf River to any of the ground water observation wells, the closest being 2.5m from the river. The alkyl benzenes included: toluene, 1,3-dimethyl benzene, and other 2 and 3 carbon benzene isomers. Aerobic respiration and nitrification occurred predominantly in the first few meters of infiltration, thus supporting the theory that the decrease in concentration was caused by biological processes under aerobic conditions. The biological processes that removed the organic compounds were efficient, considering the short residence time between the river and the closest well and the small retardation factors of the compounds. The decline was observed at different temperature throughout the year, including 5°C in winter. Alkyl benzenes degrade quicker than halogenated aromatics under aerobic conditions, probably because of the breaking of the halogen bond for halogenated aromatics is relatively slow.

Naphthalene and methyl-naphthalene also decreased in concentration but the decrease in

naphthalene, however, may be due to adsorption based on the results of Ehrlich et al. (1982). Ehrlich et al. (1982) observed that naphthalene did not biodegrade under anaerobic conditions, but was slightly sorbed. Bouwer and McCarty (1984) observed that several non-chlorinated aromatics are removed under aerobic but not anaerobic conditions.

Chlorobenzenes. Chlorobenzenes have been observed to degrade under aerobic but not anaerobic conditions (e.g., Bouwer and McCarty 1984). The chlorobenzenes, 1,4-DCB, 1,2,4-TCB and 1,2,3-TCB decomposed under aerobic conditions in the aquifer near the Glatt River, and are suggested to have degraded to chlorinated phenols and catechols (Schwarzenbach and Westall 1981b). The rate of decrease was slower than for the alkyl aromatics, perhaps because the breaking of the halogen bond slows the process (Schwarzenbach et al. 1983). Halogenated aromatics do not degrade under anaerobic conditions. The concentrations of 1,4-DCB did not decrease in July and August of 1979, 1980 and 1981 between the river and 5m from the river, as it did the rest of the year because conditions were anaerobic during these summer months and the compounds did not decompose. During the rest of the year the conditions were aerobic and the chlorobenzenes decomposed. Chlorobenzenes in another Swiss study persisted for at least seven years under anaerobic conditions (Giger and Schaffner 1981). Chlorobenzenes (1,4-DCB, 1,2,4-TCB and 1,2,3-TCB) decomposed above, but not below the water table in a shallow fluvial aquifer in Oklahoma (Wilson et al. 1983). The failure of chlorobenzene to decompose in autoclaved (i.e., sterilized) lab samples established microorganisms as the likely agent of destruction.

Pesticides. Lab studies on sewer sludge indicated that pesticides such as lindane degraded more quickly under active anaerobic lab conditions than under corresponding aerobic conditions, probably due to bacteria (Hill and McCarty 1967). DDT, for example, converted rapidly to DDD under anaerobic conditions, but persisted as DDT under aerobic conditions of several mg/L of dissolved oxygen. Similarly, more than 20 species of bacteria were found to reductively dechlorinate DDT under anaerobic conditions, whereas aerobic conditions apparently did not promote dechlorination (Kobayashi and Rittman 1982). Other pesticides that were dehalogenated under anaerobic conditions in lab culture tests include: toxaphene by bacteria, lindane by soil bacteria and parathion by bacteria (Kobayashi and Rittman 1982). These lab results indicate that pesticides are easier to break down under anaerobic than under aerobic conditions. The breakdown process is relatively easy once the halogen bond is broken.

Phenolic compounds have been shown to biodegrade under anaerobic conditions in an aquifer composed of glacial drift material in Minnesota (Ehrlich et al. 1982). Methane and CO₂ were formed by the anaerobic bacteria breaking down the phenolic compounds. Lab studies supported the field results, and also indicated that principally biodegradation and not sorption account for the decline in concentration (Ehrlich et al. 1982). Glass column experiments showed that chlorophenols can biodegrade under aerobic conditions (Zullei 1981).

Biodegradation is an appealing cleanup method because expensive cleanup methods could be avoided and the pollutant is destroyed rather than transferred

to another part of the environment, such as to the atmosphere via air stripping. In some cases, however, the degradation products could be as toxic or worse than the original compound. Management of some of the parameters that affect biodegradation, such as nitrate supply, may allow biodegradation to occur in situ in the vadose zone or aquifer. Limitations include the difficulty of managing environmental parameters that promote biodegradation and the difficulty in maintaining biodegradation as environmental conditions change.

Geological Considerations

The detailed structure and mineralogic composition of aquifers is critical to the transport of pollutants. One example is a PCB spill in a glacial till area in western Canada (Schwartz et al. 1982; Roberts, Cherry and Schwartz 1982). Between 6,800 and 21,000 liters of transformer oil containing PCBs and chlorobenzenes were spilled at a transformer plant. The PCBs traveled mainly in-phase because of the low solubility of PCBs (0.05 mg/L). The laboratory-determined conductivities of the till zone, between 10⁻⁵ and 10⁻⁹ cm/s, are too low to explain the observed vertical migration. Vertical movement is primarily through fractures in the clay, silt and till units, as indicated by the high PCB concentrations measured on fracture surfaces. Tritium was also found along fracture surfaces and used to calculate the rate of solute migration. This rate is a minimum because, unlike PCBs, some of the small tritium atoms diffuse into the sedimentary units. The geological units also have a low organic content, 0.2 to 0.9 percent carbon, minimizing the role of organic carbon in absorbing the PCBs.

Conclusions and Recommendations

Although progress is being made in understanding how organic compounds travel in the subsurface, large gaps and unknown important parameters exist. Several recommendations are given below on areas that need research.

- Some polar organic compounds are not commonly detectable by present methods. They appear to be persistent in ground water, able to travel significant distances and be resistant to degradation. Perhaps the increased ability to identify these polar organics will provide a better understanding of this type of contamination. Group parameter methods, such as TOX, may be attractive compliments to the commonly used GC/MS method because of the lower cost and because the measurements include classes of compounds, e.g., polar halogenated organics in the case of TOX, which are not readily identifiable individually.

- In cases where the aquifer might contain sufficient carbon for adsorption to be significant, the empirical relationships that have been developed may be useful for determining the partitioning behavior of organic pollutants. Further study of the effect of grain size, organic content, solute concentrations, dissolved organic matter and other controls on adsorption will help clarify how solutes are transported.

- Some elements, such as N, S, or P-compounds, when injected into pollution plumes may promote microbial degradation. The field conditions under which biodegradation of different compounds is promoted is not well understood. The phase in which the pollutant biodegrades might also be considered, i.e., dissolved in water, in-phase, or adsorbed onto the

matrix.

• More work is needed to determine how flushing of an aquifer via injection and extraction wells affects those pollutants sorbed onto aquifer or soil material. Travel of solutes in-phase during flushing, such as droplets within the water, may be an important mechanism.

Ground water flow models in porous media are useful for understanding a flow regime and for planning the placement of wells. Solute transport models assume constant dispersivity values and the solute is assumed to be dissolved, which in some cases may not be reasonable assumptions. Resolution problems with numerical models may occur in some cases, such as for modeling trace concentrations of a solute, high concentration gradients, or radial flow from a pulse on a rectangular grid. The mechanisms of adsorption and biodegradation are not well enough understood to model satisfactorily. The effects of such mechanisms will probably be lumped together in models because their effects will be difficult to separate in practice.

Although the technology may exist to clean up polluted ground water and pollution sites, the costs are often high. A water policy is needed to encourage prevention and set priorities for what should be cleaned up. The cost of cleanup can be several orders of magnitude larger than that of preventive measures. Monitoring of areas containing organic compounds has begun only recently, and as monitoring continues the understanding of solute transport will improve.

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Biographical Sketch

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Behavior of Organic Compounds during Infiltration of River Water to Groundwater. Field Studies

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■ The behavior of organic micropollutants during infiltration of river water to groundwater has been studied at two field sites in Switzerland. In agreement with predictions from model calculations, persistent organic chemicals exhibiting octanol/water partition coefficients smaller than about 5000 moved rapidly with the infiltrating river water to the groundwater. The biological processes responsible for the "elimination" of various micropollutants (e.g., alkylated and chlorinated benzenes) occurred predominantly within the first few meters of infiltration. Alkylated benzenes were "eliminated" at faster rates than 1,4-dichlorobenzene. Anaerobic conditions in the aquifer near the river hindered the biological transformation of 1,4-dichlorobenzene. Among the compounds that were found to be persistent under any conditions were chloroform, 1,1,1-trichloroethane, trichloroethylene, and tetrachloroethylene. With respect to such chemicals, bank filtration is ineffective as a first step in the treatment of river water for water supplies.

Since in many European countries a significant fraction of the groundwater is recharged through infiltration of river water (1, 2), the impact of river pollution on groundwater quality is of major concern. In addition, many waterworks use natural or artificial bank filtration as a first step in the treatment of river water for water supplies (3, 4). Therefore, the behavior of organic pollutants during infiltration is of great interest.

The transport and fate of organic pollutants in a river water-groundwater infiltration system is determined by several interacting processes, including advection, dispersion, (ad)sorption/desorption, hydrolysis, redox reactions, and biological transformations. In laboratory experiments, individual processes may be studied under controlled conditions (5, 6), and mathematical models may be developed to predict the effect of a particular process on the transport and fate of a compound in the environment (7, 8). However, comprehensive field investigations are needed to evaluate the applicability of laboratory studies and model calculations to natural systems.

To date, most of the field studies on natural river water-groundwater infiltration systems have been conducted with respect to the use of bank filtrate for public water supplies (e.g., ref 3). These studies have usually been confined to monitoring selected water constituents in the river and in groundwater wells near the region of infiltration. The temporal and spatial variations in concentration of organic compounds along the infiltration path have not been thoroughly investigated. Consequently, the results of such investigations provide only very limited insights into the behavior of individual compounds during infiltration.

In this paper, we report the results of two field studies aimed at investigating the transport and fate of organic micropollutants, including chlorinated hydrocarbons, alkylated benzenes, and chlorinated phenols during natural infiltration of river water to groundwater. In the near

fields of two rivers, a network of observation wells was installed that allowed the contaminants in the infiltrating water to be traced from the river to the groundwater. The results of this 2-year field study contribute significantly to the limited field data on the behavior of trace organics in the groundwater environment (9-11).

Theoretical Section

Prediction of Retardation Factors for Hydrophobic Organic Compounds in the Ground. A rough estimate of the retention behavior of a given hydrophobic organic compound during infiltration may be obtained by treating transport through the river bed and in the aquifer in a first approximation as a one-dimensional process with constant flow in a homogeneous porous medium. Assuming that only the fine fraction of the aquifer material is relevant for sorption (5, 12) and assuming a linear sorption isotherm, an average retardation factor (R_f^z = ratio of the residence time τ_z of the solute to the residence time τ_w of the water) can then be calculated for compound z for a given segment of the aquifer (e.g., ref 8):

$$R_f^z = \tau_z / \tau_w = 1 + fK_p^z \rho(1 - \epsilon) / \epsilon \quad (1)$$

where f = fraction of the aquifer material responsible for sorption (e.g., grain size range $\phi < 125 \mu\text{m}$; assumption: homogeneous distribution), K_p^z = equilibrium partition coefficient of the compound z between water and the fine fraction of the aquifer material at a given location in the ground (cm^3/g), ρ = density of the aquifer material (g/cm^3), and ϵ = total porosity. As we have shown in a previous study (5), for the compounds reported here, the equilibrium partition coefficient, K_p^z , may be estimated from the organic carbon content of the fine fraction of the aquifer material, f_{oc} , and from the octanol/water partition coefficient of the compound, K_{ow}^z :

$$K_p^z = 3.2f_{oc}(K_{ow}^z)^{0.72} \quad (2)$$

Similar relationships have been found for other types of compounds and natural sorbents (12, 13). Note that eq 2 is valid only for sorbents exhibiting organic carbon contents of greater than about 0.1% ($f_{oc} > 0.001$). For organic-poor sorbents, interactions of the chemical with the inorganic matrix of the sorbent may become important (5). Combining eq 1 and 2 yields

$$R_f^z = 1 + 3.2ff_{oc}(K_{ow}^z)^{0.72}\rho(1 - \epsilon) / \epsilon \quad (3)$$

Retardation factors calculated from eq 3 are valid only at sorption equilibria. At high groundwater-flow velocities, e.g., such as those encountered in the near field of a river during stormwater events (0.5 m/h; see ref 14), due to slow sorption kinetics, the compounds may be transported even faster than would be assumed from equilibrium considerations (5, 15). However, relationships such as eq 3 are very valuable for predicting the magnitude of the velocity at which a specific hydrophobic organic compound is transported in a given aquifer.

Experimental Section

Description of the Field Sites. The main field site of this investigation (field site I) is located in the lower

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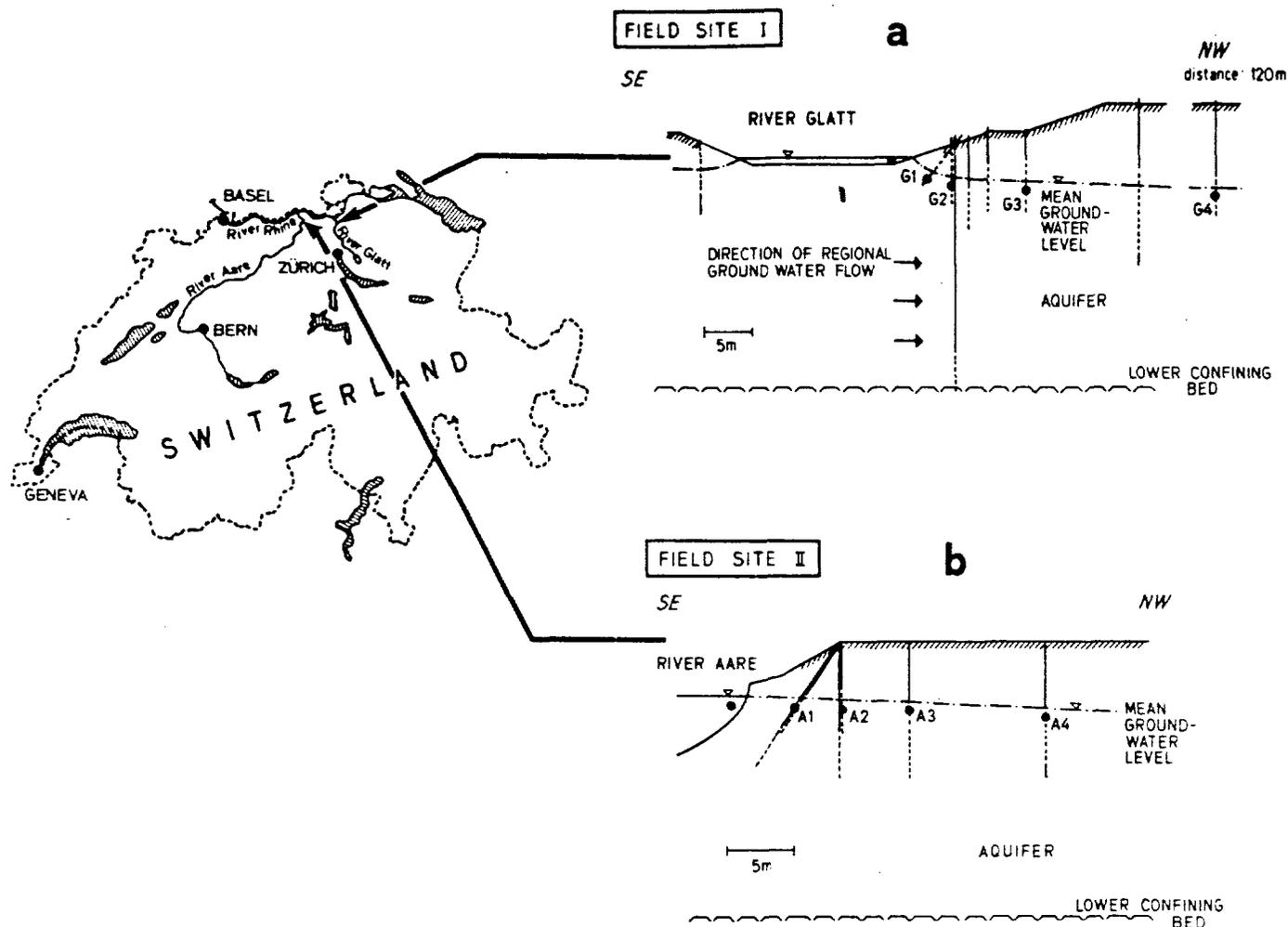


Figure 1. Locations and layouts of the two field study sites: (a) lower Glatt Valley, Switzerland (field site I); (b) lower Aare Valley, Switzerland (field site II); (● = sampling locations).

Glatt Valley, Switzerland (see Figure 1). In this region, the River Glatt infiltrates over a distance of about 5 km into a quaternary fluviglacial valley fill aquifer composed of layers of gravel and sand containing very little organic carbon (<0.1%). The River Glatt is a small, rather heavily polluted perialpine river which has been studied extensively (16, 17). The average discharge of the river is approximately $8 \text{ m}^3/\text{s}$, of which 15–20% is effluent from a number of mechanical-biological sewage treatment plants. These treatment plants are the major source for organic micropollutants in the river. At the study site, permanent infiltration of the River Glatt through a saturated zone can be assumed.

Figure 1a gives a cross-sectional view of the study site on the right bank of the River Glatt. The groundwater flows beneath the river at an angle between 60 and 90° to the flow of the river. The results presented in this paper have been obtained primarily from measurements in wells G1–G4 (see Figure 1a). These wells gave access to freshly infiltrated water that stratified in the top layers of the aquifer. Some data from observation well G15, which is screened throughout the saturated thickness of the aquifer, will also be discussed. G15 is located in the center of the valley about 60 m downstream from G4. A detailed description of this field site is presented elsewhere (18).

To check the general validity of conclusions drawn from results obtained from the main field site, a second study was conducted on a different type of river system: River Aare in the lower Aare Valley (field site II; see Figure 1b). River Aare is a moderately polluted alpine river with an average flow of $550 \text{ m}^3/\text{s}$ in the study area. The aquifer into which the river loses water is of the same geological

formation as the one in the lower Glatt Valley (19). At the study site, the River Aare infiltrates through a saturated zone. Figure 1b shows the network of observation wells that were installed on the left bank of the river. At this location the regional groundwater flows beneath the River Aare at an angle of between 45 and 90° to the flow direction of the river.

Groundwater Observation Wells. All wells were lined with hard PVC tubes. In laboratory experiments, the PVC material was found neither to contaminate the samples nor to (ad)sorb the organic water constituents of interest. For technical details, see Hoehn et al. (18).

Sample Collection and Analytical Program. Between May 1979 and Apr 1980 (field site I) and between Nov 1980 and Oct 1981 (field site II), a program was conducted to determine temporal and spatial variations in the water composition of the rivers and of the groundwater in the observation wells shown in Figure 1. Samples were collected at approximately monthly intervals. In addition to the trace organic compounds, a variety of other chemical parameters were determined, mainly to characterize the river water and the groundwater, as well as to study the biogeochemical processes occurring during infiltration. Results of these measurements are discussed elsewhere (20).

The groundwater was sampled by using a small underwater plunger pump as described by Käss (21). The small discharge rate of this pump, typically between 0.5 and 1 L/min, allowed sampling of the groundwater without causing a measurable drawdown of the groundwater level. For sampling the upper layer of the groundwater, the pump was usually placed 0.5 m below the groundwater

Table I. Ranges of Retardation Factors (Lower Limit-Upper Limit) Calculated for Some Hydrophobic Organic Compounds Detected in the River Glatt

compound	octanol/water partition coeff (log K_{ow})	calcd retardation factors (R_f) ^{a,b}		
		river sediment ^c (~0.1 m)	aquifer close to river bed ^d (<5 m)	aquifer far from river bed ^e (>5 m)
chloroform	1.97 ^f	2.7-8	1.2-4	1-1.2
1,1,1-trichloroethane	2.17 ^g	3.4-11	1.2-6	1-1.2
trichloroethylene	2.29 ^f	4-13	1.3-7	1-1.3
toluene	2.69 ^f	7-23	1.6-12	1-1.6
tetrachloroethylene	2.88 ^h	9-31	1.8-16	1-1.8
1,3-dimethylbenzene	3.15 ^f	13-48	2.2-24	1-2.2
naphthalene	3.30 ⁱ	16-62	2.5-31	1-2.5
1,4-dichlorobenzene	3.38 ^f	18-70	2.7-35	1-2.7
α -hexachlorocyclohexane	3.72 ^k	32-123	4-62	1-4
hexachlorobenzene	6.06 ^f	1500-6000	150-3000	~10-150

^a Equation 3; $\rho = 2.5 \text{ g cm}^{-3}$, $\epsilon = 0.2$. ^b Ranges in values for f and f_{oc} are based on experimental data (see ref 5 and 18). ^c $f_{oc} = 0.01-0.02$, $f = 0.2-0.4$. ^d $f_{oc} = 0.001-0.01$, $f = 0.2-0.4$. ^e $f_{oc} < 0.001$, $f < 0.2$. ^f Reference 39. ^g Reference 40. ^h Reference 41. ⁱ Reference 26. ^k Reference 42.

table. The water was pumped through stainless steel tubing (5 mm i.d.) into 1-L glass bottles. The bottles were filled completely and closed without headspace. The samples were stored at 4 °C within 6 h of collection and analyzed within 48 h.

Analytical Methods. Volatile organic compounds were concentrated from the water samples by the closed-loop gaseous stripping/adsorption/elution procedure developed by Grob (22, 23). The water samples (typically 1 L) were stripped for 90 min at 30 °C, and the organic compounds were trapped by adsorption on a filter of 1.5 mg of activated charcoal. The filter was then extracted with 20 μL of carbon disulfide (CS_2) and the extract analyzed by high-resolution glass capillary gas chromatography and, when necessary, by gas chromatography/mass spectrometry. The gas chromatographic equipment and parameters used have been described elsewhere (24).

Purgeable organochlorine compounds (POCI) were determined by the method described by Zürcher (25). The compounds were purged with oxygen from 1-L water samples for 30 min at 60 °C. The purged compounds were continuously combusted at 950 °C, and the resultant chloride was trapped and quantified by ion chromatography.

Pentachlorophenol was determined by a method based on the procedure described by Renberg and Lindström (26). The lipophilic phenols were extracted by percolating 0.5 L of the acidified water sample (pH 2) through a SepPak C_{18} cartridge (Waters Inc., Milford, MA). The adsorbed phenols were eluted with 1.5 mL of acetone and acetylated by adding 50 μL of acetic anhydride. The excess anhydride was then destroyed by adding 3 mL of 0.1 M aqueous K_2CO_3 , and the acetylated phenols were extracted with 2 mL of pentane. The pentane extract was analyzed by glass capillary gas chromatography using electron capture detection. 2,4,6-Tribromophenol was used as internal standard.

α - and γ -hexachlorocyclohexane and hexachlorobenzene were determined by the method of Müller (27). For all three compounds, the detection limit of the method used was 0.02 ng/L.

Dissolved Organic Carbon (DOC). Fractionation of the DOC was carried out with the method described by Schneider et al. (28). The DOC is operationally separated into three fractions: a "hydrophilic" fraction, an "acidic" fraction, and a "hydrophobic" fraction (see Figure 2). The fractionation is based on the retention of the organic constituents on a column (50 mm \times 4 mm) packed with octadecylsilica (LiChrosorb RP 18) and connected to an on-line DOC detector (29). The "hydrophilic" fraction 1

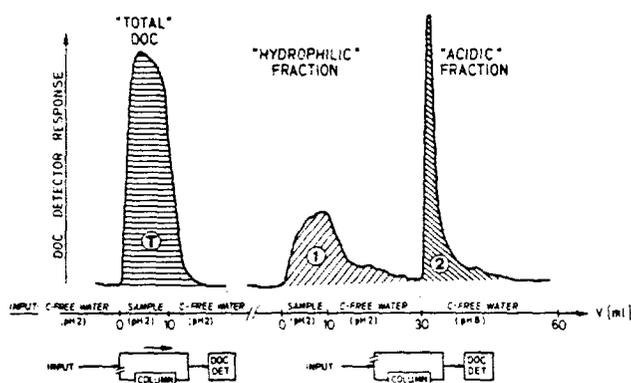


Figure 2. DOC fractionation by the method of Schneider et al. (28).

is not retained on the column at pH 2 (peak 1 in Figure 2). The "acidic" fraction 2 is retained at pH 2 but is eluted from the column at pH 8 (peak 2 in Figure 2). This fraction includes all fulvic acid type materials (28). The "lipophilic" fraction cannot be eluted from the column with water at any pH. Its concentration is calculated by subtracting fraction 1 and 2 from the total DOC, which is determined by passing the adsorption column (see peak T in Figure 2).

Results and Discussion

Field Site I. The names and octanol/water partition coefficients of some of the hydrophobic organic compounds found in the River Glatt are given in Table I. A more comprehensive inventory of the trace organics detected in this river has been published previously (17). The concentrations of individual compounds were usually between 0.01 and 2 $\mu\text{g/L}$.

Figure 3 depicts the temperature values, the concentrations of two representative volatile organic compounds, and the concentrations of oxygen, ammonium, and dissolved organic carbon determined over the course of 1 year in the River Glatt and in the wells G2 and G3, located 5 and 14 m, respectively, from the river. The average concentrations of some water constituents in the River Glatt and in wells G1-G4 are presented in Table II and Figure 4.

On the basis of the results of tracer experiments (18) and from the temperature data presented in Figure 3a, it can be assumed that the residence time of the water between the river and the two wells G2 and G3 was usually in the order of hours to a few days (well G2) and days to a few weeks (well G3). Figure 3b shows that for tetrachloroethylene, large fluctuations in concentration were observed

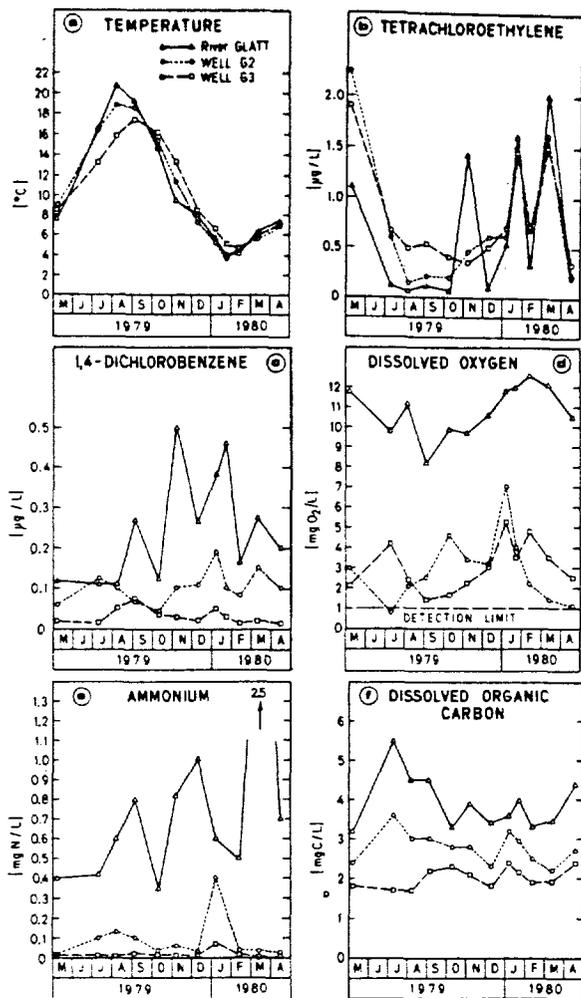


Figure 3. Field site I: monthly determined values for temperature, tetrachloroethylene, 1,4-dichlorobenzene, dissolved oxygen (detection limit = 1 mg of O₂/L), ammonium, and DOC, in the River Glatt and in observation wells G2 and G3.

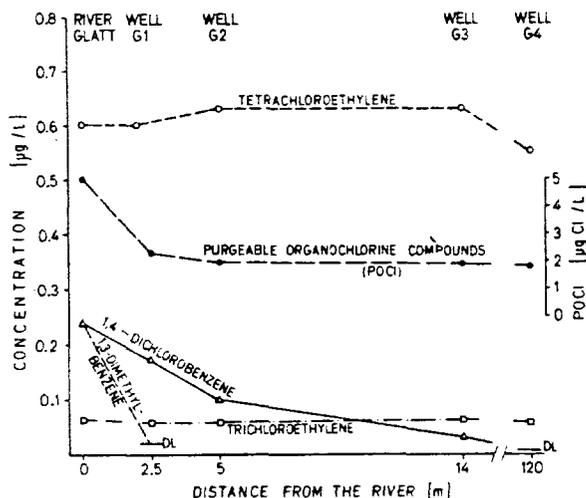


Figure 4. Field site I: average concentrations of selected organic micropollutants in the River Glatt and in the upper layers of the groundwater at various distances from the river (DL = detection limit).

in the River Glatt and in both observation wells, indicating a rapid response in the groundwater to concentration changes in the river. From the very similar average concentrations found for this compound in the river and in the groundwater in the top layers of the aquifer at different distances to the river (see Figure 4), one can conclude that tetrachloroethylene was not significantly affected by any elimination processes. The effect of the sorptive and dispersive processes, i.e., the attenuation of concentration

Table II. Yearly Average Concentrations and Standard Deviations of DOC, POCl, and Four Individual Organic Micropollutants in the River and in the Upper Layers of the Groundwater at Different Distances from the River

sampling location ^a	dist from river, m	DOC, mg/L	POCl, ^b µg of Cl/L	tetrachloroethylene, ^c µg/L	1,4-dichlorobenzene, ^d µg/L	1,3-dimethylbenzene, ^e µg/L	pentachlorophenol, ^f µg/L
field site I							
River Glatt	0	3.9 ± 0.7 {16}	5.0 ± 4.2 {7}	0.60 ± 0.70 {16}	0.23 ± 0.13 {16}	0.23 ± 0.25 {16}	0.08 ± 0.005 {4}
well G1	2.5	3.0 ± 0.7 {9}	2.3 ± 1.8 {7}	0.60 ± 0.59 {9}	0.17 ± 0.07 {9}	<0.02 {9}	0.08 ± 0.02 {4}
well G2	5	2.8 ± 0.6 {16}	2.0 ± 1.0 {7}	0.63 ± 0.57 {16}	0.10 ± 0.04 {16}	<0.02 {16}	0.05 ± 0.04 {4}
well G3	14	2.1 ± 0.4 {16}	1.9 ± 0.9 {7}	0.63 ± 0.47 {16}	0.03 ± 0.02 {16}	<0.02 {16}	<0.02 {4}
well G4	120	1.5 ± 0.4 {9}	1.8 ± 0.9 {7}	0.55 ± 0.14 {9}	<0.005 {9}	<0.02 {9}	<0.02 {4}
field site II							
River Aare	0	2.3 ± 0.5 {12}	1.6 ± 0.9 {7}	0.24 ± 0.12 {12}	0.035 ± 0.015 {12}	0.05 ± 0.03 {12}	nd
well A1	4	1.0 ± 0.4 {12}	0.9 ± 0.6 {7}	0.25 ± 0.09 {12}	<0.01 {12}	<0.02 {12}	nd
well A2	7.5	0.8 ± 0.4 {12}	0.8 ± 0.5 {7}	0.24 ± 0.10 {12}	<0.005 {12}	<0.02 {12}	nd
well A3	13	0.7 ± 0.4 {12}	0.7 ± 0.3 {7}	0.20 ± 0.08 {12}	<0.005 {12}	<0.02 {12}	nd
well A4	23	0.7 ± 0.3 {12}	0.7 ± 0.3 {7}	0.20 ± 0.07 {12}	<0.005 {12}	<0.02 {12}	nd

^a See Figure 1. ^b Purgeable organochlorine compounds (group parameter), detection limit 0.1 µg of Cl/L. ^c Detection limit 0.01 µg/L. ^d Detection limit 0.005 µg/L. ^e Detection limit 0.02 µg/L. ^f Detection limit 0.02 µg/L; nd = not determined. ^g The number of measurements is given in brackets.

fluctuations (see, e.g., ref 8), are reflected in the decreasing standard deviations observed with increasing flow distance (see Table II). Very similar results were found for chloroform, trichloroethylene (see Figure 4), and 1,1,1-trichloroethane.

The strong response in the groundwater to concentration changes in tetrachloroethylene in the River Glatt (Figure 3b) suggests that during infiltration this compound was not strongly retained in the ground. This finding is in agreement with the rather small average retardation factor predicted for tetrachloroethylene from eq 3 for this field site (see Table I and footnotes in Table I). It should be noted that for the aquifer in the near field of the River Glatt (<5 m; see Table I), the average retardation factors for the various compounds must be assumed to be closer to the indicated lower limits, since a sharp drop in the organic carbon content of the aquifer material occurs within a few meters distance from the river (18). Thus, at this field site, rapid transport in the ground can be expected for persistent organic compounds exhibiting octanol/water partition coefficients of smaller than about 5000 (see Table I). Unfortunately, because of the short residence time of the water between the river and the observation wells G2 and G3, no quantitative information on the actual retardation of the various compounds can be obtained from our data, since samples were taken only at monthly intervals.

Among the volatile organic compounds that were always present in the River Glatt but were almost never detected in any of the groundwater observation wells, were the aromatic hydrocarbons including toluene, various C_2 - and C_3 -benzene isomers (e.g., 1,3-dimethylbenzene; see Table II and Figure 4), and naphthalene. Since under the conditions typical for the groundwater environment these aromatic hydrocarbons, as well as all other volatile organic compounds listed in Table I (e.g., 1,4-dichlorobenzene), do not undergo chemical reactions at significant rates, and since these compounds are also only weakly sorbed, any observed "elimination" during infiltration must be attributed to biological transformation and/or mineralization. Parts d-f of Figure 3 indicate that the major biologically mediated processes, i.e., aerobic respiration and nitrification, occurred predominantly within the first few meters of infiltration. Therefore, it could be expected that the biological "elimination" of xenobiotic compounds would also take place primarily in the near field of the river.

Although the volatile aromatic hydrocarbons are biodegradable under simulated groundwater conditions (30), it is interesting to note that they were always eliminated between the river and well G1. Thus, considering the short residence time of the water between the river and this well and the small retardation factors of the compounds, the biological processes responsible for their removal were quite efficient, even at temperatures as low as 5 °C.

The observed significant decrease in concentration of 1,4-dichlorobenzene with increasing distance to the river (see Figures 3b and 4, Table II) indicates that this compound was also affected by biological processes. However, compared with the aromatic hydrocarbons, 1,4-dichlorobenzene was "eliminated" at a much slower rate, such that it was still detected in well G3. In addition, in July and Aug 1979 (Figure 3c) and during several short-term investigations in July and Aug 1980 and 1981 (31), no decrease in the concentration of 1,4-dichlorobenzene was found between the River Glatt and well G2. These findings suggest that, considering the steep concentration gradients usually detected between the river and G2, 1,4-dichlorobenzene was not significantly eliminated during parts of

the summer. Since denitrification and manganese reduction were also observed during this time of the year (20, 31), the persistence of 1,4-dichlorobenzene may be explained by the anoxic conditions that prevailed in parts of the aquifer near the river. This hypothesis is corroborated by other field observations (32) and by the results of laboratory experiments that suggest that halogenated benzenes are not biotransformed under anaerobic conditions (33). Whether, under aerobic conditions, 1,4-dichlorobenzene is completely mineralized or only transformed to products not detected by the analytical techniques used (e.g., to dichlorophenol and dichlorocatechol; see ref 34) is presently under investigation.

The results of the measurements of the group parameter "purgeable organochlorine compounds" (POC; Table II, and Figure 4), which includes substances such as tetrachloroethylene and 1,4-dichlorobenzene, showed the same picture as the results of the single-component measurements: (i) the major "elimination" of volatile organochlorine compounds occurred predominantly within the first few meters of infiltration; (ii) for compounds not affected by any transformation processes, similar average concentrations were found throughout the upper layers of the aquifer.

From the few data obtained for pentachlorophenol (Table II), no final conclusions can be drawn as to whether or not this compound was "eliminated" during infiltration. At a pH of between 7.4 and 8, pentachlorophenol is present predominantly as phenolate anion ($pK_a = 4.75$). Thus, retardation factors of smaller than 50 in the near field of the river and smaller than 10 in the actual aquifer would be expected for this compound (35). The fact that the concentration of pentachlorophenol in well G3 was always below the detection limit could be an indication that this compound underwent some chemical and/or biological transformation reactions.

To date, only two sets of measurements have been conducted for α - and γ -hexachlorocyclohexane and for the highly lipophilic compound hexachlorobenzene. Although detected at very low concentrations, these compounds were found in the River Glatt and in all of the observation wells. The concentrations determined for α - and γ -hexachlorocyclohexane were 4 ng/L in the river, about 2 ng/L in the near field of the river (G2, G3), and less than 1 ng/L in G4. For hexachlorobenzene, very similar concentrations (between 0.1 and 0.2 ng/L) were found in the River Glatt and in all observation wells including G4. These findings demonstrate that, especially in aquifers composed of materials of low organic carbon content, even highly lipophilic compounds may be transported over long distances.

Field Site II. Compared to the River Glatt, the River Aare is a large river exhibiting much smaller short-term fluctuations in water composition (31). Also, in contrast to the field site in the lower Glatt Valley, the residence time of the water in the ground between the river and the observation wells (see Figure 1b) is generally much longer (in the order of weeks). This is evident from the temperature data shown in Figure 5a. Figure 5a also shows that very similar residence times can be assumed between the river and the two wells A1 and A3. Since a detailed investigation of the flow directions and velocities of the groundwater in the near field of the River Aare has not been conducted, it is not possible to give exact values for linear flow distances of the infiltrating water between the river and the wells. However, the results of the year-round study at this field site (see Figure 5 and Table II) can be qualitatively compared to those obtained in the lower Glatt Valley.

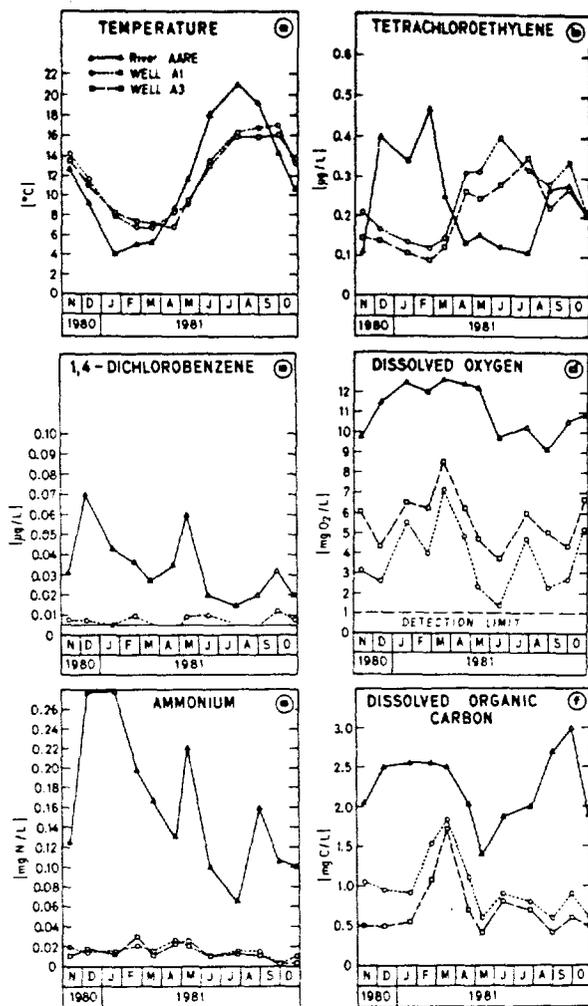


Figure 5. Field site II: monthly determined values for temperature, tetrachloroethylene, 1,4-dichlorobenzene, dissolved oxygen (detection limit = 1 mg of O₂/L), ammonium, and DOC, in the River Aare and in observation wells A1 and A3.

The data presented in Figure 5d-e show that the biologically mediated processes respiration and nitrification always occurred between the river and the wells A1 and A3. Reducing conditions were never observed in the groundwater at this field site. With respect to the organic micropollutants, all compounds exhibited behaviors similar to those observed at the Glatt site; that is, no elimination of tri- and tetrachloroethylene during infiltration, and degradation and/or transformation of the alkylbenzenes and 1,4-dichlorobenzene (see Figure 5b,c; Table II). It should be noted that the slightly lower average concentrations of tetrachloroethylene in A3 and A4 (see Table II) may be attributed to dilution of the infiltrated water with less polluted groundwater from other sources.

From the data shown in Figure 5b, it is possible to determine an average retardation factor for tetrachloroethylene. Figure 5b shows that significantly higher concentrations of this compound were observed in the river between Dec 1980 and Mar 1981. The response in the groundwater (i.e., in wells A1 and A3) to the high concentrations in the river was observed about 4 months later (τ_{tetra}). Thus, when an average residence time (τ_w) of the water between the River Aare and A1 and A3 of approximately 3 weeks to 1 month is assumed (see Figure 5a), an average retardation factor (τ_{tetra}/τ_w) of about 5 is obtained. This value is rather low when compared to the retardation factors determined by Roberts et al. (11) for compounds of similar lipophilicity in an aquifer in the Palo Alto Baylands (e.g., $R_f = 33$ for chlorobenzene). The result is, however, not surprising considering the much lower organic

Table III. Average Concentrations of Total DOC and of the Three DOC Fractions Determined by the Method of Schneider et al. (28)

sampling location ^a	dist from river, m	dissolved organic carbon, ^b mg of C/L			
		total	fraction 1 ^c	fraction 2 ^d	fraction 3 ^e
River Glatt	0	4.0	1.5	1.0	1.5
well G1	2.5	2.7	1.0	0.9	0.8
well G2	5	2.6	1.0	0.9	0.7
well G3	14	2.0	0.8	0.8	0.4
well G4	120	1.5	0.9	0.5	0.1

^a See Figure 1. ^b Average values from four measurements conducted between Sept 1979 and Dec 1979.

^c Hydrophilic at pH 2. ^d "Acidic" fraction: hydrophobic at pH 2, hydrophilic at pH 8. ^e Hydrophobic at pH 2 and pH 8.

carbon content of the aquifer materials at this field site as compared to the Palo Alto site, and it is consistent with predictions from model calculations for the very similar type of aquifer at the study site in the Glatt Valley (see Table I).

Behavior of Dissolved Organic Carbon (DOC) during Infiltration. Although this study focused on the behavior of individual organic micropollutants, a few remarks may be made on compositional changes in the bulk DOC during infiltration. Table III contains the average values for DOC and the three DOC fractions determined in the River Glatt and in wells G1-G4. During the first few meters of infiltration, the concentrations of the hydrophilic (fraction 1) and the hydrophobic fraction (fraction 3) of the DOC were significantly reduced. These reductions may be primarily attributed to microbial mineralization (20). The "acidic" fraction 2, which includes all fulvic acid type materials (28), was not significantly affected by these processes. With increasing distance from the river (G1 → G4), the decrease in concentration of the "acidic" fraction might have been caused by the formation of insoluble complexes with metal ions (e.g., Ca²⁺) and/or by adsorption onto clay minerals (36-38). Between G1 and G4, no significant changes in concentrations were observed in the hydrophilic fraction, whereas the lipophilic fraction was, to a great extent, removed and/or transformed into compounds appearing in one of the other fractions. It is not possible to identify the processes responsible for the removal of the lipophilic fraction of DOC from the available data.

Long-Range Effects of River Water Infiltration.

Figure 6 shows the vertical concentration profiles of oxygen, DOC, and two persistent volatile organic compounds determined in well G15, which provides a representative picture of the water composition at various depths in the aquifer in the lower Glatt Valley (20). From the data in Figure 6 and from the results of the measurements of other parameters (20), it can be assumed that the upper half of the aquifer contained water that had predominantly been infiltrated from the River Glatt, whereas the water in the bottom half of the aquifer originated mostly from less polluted sources.

It is interesting to note that throughout the upper half of the aquifer, the concentrations of tri- and tetrachloroethylene were very similar to the average concentrations detected in the River Glatt (see Figure 4). Thus, when considering that the deeper layers of the upper half of the aquifer contained water that had been infiltrated from the river at distances of up to several kilometers from this well

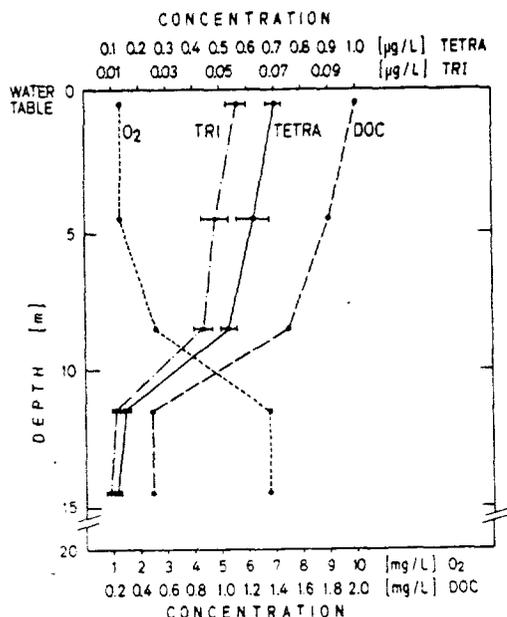


Figure 6. Field site I: vertical concentration profiles of selected parameters in well G15: (---) = dissolved oxygen (O_2); (-.-) dissolved organic carbon (DOC); (---) trichloroethylene (tri); (—) tetrachloroethylene (tetra).

(18), these findings again clearly demonstrate the great mobility of such persistent compounds in these types of aquifers and hence their potential to contaminate large groundwater areas.

Summary and Conclusions

The transport and fate of organic pollutants, including various volatile organic compounds, during infiltration of river water to groundwater has been studied in year-round investigations at two different field sites in Switzerland. The most important results and conclusions of this field investigation follow.

(1) As predicted by model calculations, volatile organic compounds move rapidly with infiltrating water from rivers to groundwaters. If a river is permanently charged with such chemicals, large groundwater areas may be contaminated, unless the compounds are eliminated during infiltration by biological processes.

(2) Among the volatile organic compounds for which no evidence of biological transformation under any conditions was found were chloroform, 1,1,1-trichloroethane, trichloroethylene, and tetrachloroethylene. With respect to such persistent chemicals, bank filtration is ineffective as a first step in the treatment of river water for water supplies.

(3) The compounds for which biotransformation was observed (e.g., all alkylated C_1 - C_4 -benzenes, naphthalene, the methylnaphthalenes, and 1,4-dichlorobenzene) were "eliminated" during infiltration to concentrations below their detection limits. Alkylated benzenes were always "eliminated" within the first few meters of infiltration, even at temperatures below 5 °C. The biotransformation of 1,4-dichlorobenzene occurred at a slower rate.

(4) There is strong evidence that certain organic micropollutants (e.g., 1,4-dichlorobenzene) were only biotransformed under aerobic conditions. The elimination of such compounds may therefore be hindered if anaerobic conditions prevail in the aquifer in the near field of a river.

(5) The retention of even highly lipophilic compounds such as hexachlorobenzene is rather small in aquifers composed of materials of low organic carbon content (i.e., $f_{oc} < 0.001$).

The results of this study show that long-term field measurements are useful (i) to gain relevant insights into the behavior of organic micropollutants in a natural river water-groundwater infiltration system and (ii) to check the general validity of conclusions drawn from laboratory investigations.

Acknowledgments

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Registry No. 1,4-Dichlorobenzene, 106-46-7; toluene, 108-88-3; 1,3-dimethylbenzene, 108-38-3; naphthalene, 91-20-3; hexachlorobenzene, 118-74-1; chloroform, 67-66-3; 1,1,1-trichloroethane, 71-55-6; trichloroethylene, 79-01-6; tetrachloroethylene, 127-18-4.

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OH Radical Rate Constants and Photolysis Rates of α -Dicarbonyls

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■ Photolysis rates of glyoxal, methylglyoxal, and biacetyl and OH radical reaction rate constants for glyoxal and methylglyoxal have been determined at 298 ± 2 K in an environmental chamber, by using the photolysis of CH_3ONO -air mixtures to generate OH radicals. The OH radical rate constants obtained were $(1.15 \pm 0.04) \times 10^{-11}$ and $(1.73 \pm 0.13) \times 10^{-11} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$ for glyoxal and methylglyoxal, respectively. The photolysis rates of glyoxal, methylglyoxal, and biacetyl increased throughout this series, and average quantum yields for the wavelength region $\geq 290 \text{ nm}$ of 0.029 ± 0.018 , 0.107 ± 0.030 , and 0.158 ± 0.024 were derived for glyoxal, methylglyoxal, and biacetyl, respectively. In addition, upper limits to the rate constants for the reaction of O_3 with glyoxal and methylglyoxal of $<3 \times 10^{-21}$ and $<6 \times 10^{-21} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$, respectively, were obtained at 298 ± 2 K. These data will serve as needed input to chemical kinetic computer modeling studies of the aromatic hydrocarbons.

Introduction

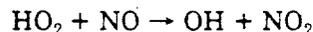
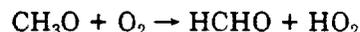
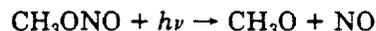
The α -dicarbonyls glyoxal, methylglyoxal, and biacetyl are important ring-cleavage products in the NO_x -air photooxidations of the aromatic hydrocarbons (1-6), and the photolysis of methylglyoxal to radical species is postulated to lead to the observed photochemical reactivity of toluene and the higher aromatics (4, 6). In addition, methylglyoxal is postulated to be an intermediate product in the NO_x -air photooxidation of the naturally emitted hydrocarbon isoprene (7, 8). Under atmospheric conditions, these α -dicarbonyls, besides photolyzing, may also react with OH radicals or with O_3 . On the basis of the data for methylglyoxal (9, 10), their reactions with O_3 are expected to be negligible, as is the reaction of OH radicals with biacetyl (2). However, both glyoxal and methylglyoxal are expected to react rapidly with OH radicals with rate constants of $\sim(1-2) \times 10^{-11} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$ at 298 K (4), although the only OH radical rate constant available for glyoxal or methylglyoxal is a recent value of $(7.1 \pm 1.6) \times 10^{-12} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$ at 297 K for methylglyoxal (11). In addition, the rates and products of the photolysis of

these α -dicarbonyls under atmospheric conditions are not well-known (4, 6, 12), and hence there is clearly a need to investigate the atmospheric loss processes of these α -dicarbonyls in more detail.

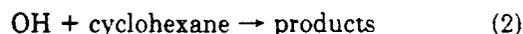
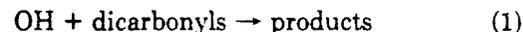
In this work, rate constants for the reactions of OH radicals with glyoxal have been determined at 298 ± 2 K, relative to the rate constant for the reaction of OH radicals with cyclohexane, and the photolysis rates of glyoxal, methylglyoxal, and biacetyl have been determined in 1 atm of air in an environmental chamber. In addition, upper limits to the rate constants for the reaction of O_3 with glyoxal and methylglyoxal have been determined at 298 ± 2 K.

Experimental Section

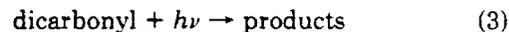
Photolysis and OH Radical Reactions. The technique for the determination of relative OH radical rate constants and of photolysis rates was essentially identical with those described previously (13, 14). Hydroxyl radicals were generated by the photolysis of methyl nitrite in air at $\geq 290 \text{ nm}$, at part-per-million concentrations:



In order to minimize the formation of O_3 during these irradiations, NO was included in the reaction mixtures. In the presence of an α -dicarbonyl and a reference organic (cyclohexane), the OH radicals can, besides reacting with CH_3ONO , NO, NO_2 , and the organic reaction products, react with these organics:



Additionally, the α -dicarbonyls also photolyze:



Under the experimental conditions employed, reactions of the dicarbonyls and cyclohexane with $\text{O}(^3\text{P})$ atoms and O_3 were negligible, and since dilution due to sampling was also negligible ($<0.2\%$), then

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Comoro
McCoy Gas Com. A

Operator: Monthly Report

Not Available:

1985 Feb + March

SOUTHERN JUNIOR EXPLORATION COMPANY

JICARILLA K	1292	1034	1211	1519	894	967	22	317	287	63	1937	1955	10594	315238
E 1301125N 5W GAS	658	604	466	469	413	392	382	413	392	308	520	521	5337	205954
E 1401125N 5W GAS	545	325	182	215	95	209	282	145	82	82	908	661	7757	91271
E 1601225N 5W GAS	800	816	592	723	337	534	781	317	830	830	3371	2842	757	215229
E 18M 225N 5W GAS	3295	2779	2451	2926	1718	2123	1467	779	745	1283	3371	2842	25779	827692

SOUTHLAND ROYALTY COMPANY

E 1101528N10W GAS	3665	3427	3014	3267	2138	2877	2559	268	2772	4006	3620	3243	34856	198688
E 23A1628N10W GAS	3345	3209	2917	3031	2162	3331	1652	3295	2658	2798	2899	2587	33884	104295
E 24H1528N10W GAS	6374	6253	4136	2868	4295	337	1652	4	4	9492	4636	4565	43362	70643
E 3402 GAS	2820	2820	2253	2803	1568	2306	2237	163	2073	3997	3281	2773	29676	142750
E 3894 GAS	4178	4178	2972	3198	2777	2783	3035	3227	2822	2637	2978	3303	37804	130409
E 1257 GAS	1011	1011	844	852	853	856	68	637	1138	637	1132	1138	8648	69124
E 531 GAS	1130	1130	1057	1335	773	894	880	1211	917	1160	1076	912	11876	33288
E 51 GAS	55	55	59	4747	45	57	17	11	11	16	17	17	419	39796

MANGUM

E 402829N11W GAS	2824	3400	2592	2152	2286	2226	3234	3100	2781	2036	3955	2717	33303	112112
E 542929N11W GAS	28	26	28	21	28	27	27	3	3	3	3	3	176	39796

MCLANAHAN

E 2557 GAS	2400	2400	2166	2319	2377	2617	2287	2674	2475	2292	2059	2227	28450	95892
E 2181528N10W GAS	7445	5566	5041	4747	4144	872	872	1	1	1	1	9885	37700	65831

REID

E 2111928N 9W GAS	2813	2923	1647	1944	2185	2216	2055	2051	1761	2441	2539	2161	26736	118184
E 26N 728N 9W GAS	7721	6702	5502	5204	3607	481	7	4	3	7871	5715	4462	47265	72162
E 45828 GAS	171	168	180	154	160	173	35	31	27	39367	33890	39973	373560	1253174

TEMNECO OIL COMPANY

E 2741 GAS	1552	1552	606	9	2809	277	1956	885	2376	1691	1846	1593	18341	97965
E 1402 GAS	1997	1997	1917	1540	624	1132	1985	1726	1521	1313	1751	614	17522	93100
E 2166 GAS	1974	1974	2382	2003	2101	1795	1068	1133	445	1824	1485	614	18990	147758
E 1985 GAS	1969	1969	1866	1789	1583	1713	1783	1486	2408	1578	1729	1913	19394	54670
E 2701 GAS	2701	2701	2900	2267	2268	1030	2611	2370	2408	2779	2227	2667	26228	68589
E 1201 GAS	1993	1993	6292	3010	1881	4859	2790	2557	4491	2699	2494	1075	24191	81044
E 131 GAS	107	892	62	84	203	98	571	330	176	285	174	173	1381	121652
E 615 GAS	892	892	226	665	969	969	571	330	176	415	38	115	5008	18260

JICARILLA B

E 35 GAS	81	81	803	2572	1784	827	1100	1122	1467	2085	2311	1802	13767	44584
E 427 GAS	1503	1503	289	515	1698	1583	1100	1122	1151	1263	46	548	11245	22477
E 3638 GAS	1495	1495	1960	1841	1152	1551	5099	3961	3183	3167	2996	2758	30841	175612
E 1407 GAS	1147	1147	1267	1473	1013	843	2132	1813	1666	1666	644	2307	22422	109238
E 3308 GAS	1789	1789	96	1921	1217	1028	962	1008	854	774	644	1409	13190	784958
E 7200 GAS	1370	1370	1262	1509	1421	582	9127	3775	1197	4439	3396	3846	46876	190224
E 1332 GAS	1370	1370	1262	1509	1421	582	9127	3775	1197	4439	3396	3846	46876	190224

Chacra 1984

CONTINUED OPERO CHACRA (GAS)

WELL S T R	JAN	FEB	MAR	APRIL	MAY	JUNE	JULY	AUG	SEPT	OCT	NOV	DEC	1984	PROD	MP	ACCU
11L2628N10W GAS	1432	1669	824	1435	1605	962	1793	1485	978	927	814	786	14710	15394	35708	35986
12A1628N11W GAS	1930	1132	1525	1773	1659	1731	1333	1254	957	888	865	847	15394	14710	35708	35986

WELL	DATE	PROD.	DATE	10/79	11/80	12/81	1/82	2/82	3/82	4/82	5/82	6/82	7/82	8/82	9/82	10/82	11/82	12/82	1/83	2/83	3/83	4/83	5/83	6/83	7/83	8/83	9/83	10/83	11/83	12/83			
DUPLEX CORNELL A	12	7	3	15	28	19	26	17	42	19	252	1123																					
E 1D 129N12W OIL	305	8	3385	5852	7858	5448	4367	2422	6030	8911	53696	2792686																					
E 1S 129N12W OIL	386	8	3750	6110	9919	6835	5916	2755	6700	7792	65301	3465982																					
DAWSON A	1357	1438	1659	1529	1510	1735	1346	1794	1523	1360	18004	494892																					
M 1D +27N 8W GAS	3976	3750	4379	3960	4152	4404	3460	4469	4268	3431	47846	266491																					
DAWSON FEDERAL	33	50	61	37	73	45	50	100	73	99	676	2885																					
M 1N2627N 8W GAS	2064	2016	2069	1983	1797	1791	1968	1904	1802	1623	22688	101463																					
DAY	739	587	734	1461	1342	1067	1057	972	768	1127	12033	506939																					
E 1P 729N 8W GAS	940	2948	3385	3651	3652	2766	2736	3075	2340	3333	35532	44554																					
E 2G 829N 8W GAS	2235	4	1523	2308	15933	13177	10613	11448	11448	145	57386	594422																					
E 2I 829N 8W GAS	12264	11961	2429	7936	22013	11094	12412	11515	16697	1331	72219	453384																					
E 3A1 729N 8W GAS	7	20	3	20	22	8	13	15	16	14552	69485	591694																					
E 3I1 729N 8W GAS	7	21947	65	11204	5993	3255	8893	9908	25177	20101	74157	359264																					
E 4F 829N 8W GAS	3629	15023	11204	1794	2594	2056	1	2	2	2	2	2																					
E 4N 829N 8W GAS	7407	8407	5115	1794	2594	2056	1	2	2	2	2	2																					
E 5P1 829N 8W GAS	1320	2248	1521	3	3	3	3	3	3	3	3	3																					
E 5A1 829N 8W GAS	2512	3381	2269	3535	2056	1	2	2	2	2	2	2																					
DELHI TAYLOR	2741	3422	3342	2565	2385	2413	2343	2412	2328	2234	31029	3180791																					
5A1 726N11W OIL	21	15	45	25	20	24	8	12	17	103	307	4150621																					
6K1 726N11W GAS	2862	2648	2530	3415	1887	2704	2125	2760	2663	2972	17456	4280037																					
E 6C1 726N11W GAS	1934	2062	2530	2440	3422	2704	2125	2760	2663	2972	30483	72632																					
DELHI TAYLOR D	11	20	15	20	26	18	20	14	18	22	207	561																					
E 1D 326N11W GAS	1116	1062	2109	1759	1506	1508	1773	1579	1748	1636	19113	3294114																					
DELHI TAYLOR E	3239	288	328	316	313	326	322	300	282	274	6615	25672																					
E 1J 420N11W OIL	848	827	862	1058	865	819	750	861	652	999	10046	990948																					
DRYDEN	1	11	1	1	1	1	1	1	1	1	1	1																					
1M2628N 8W GAS	86	7	3074	3258	1216	2704	2125	2760	2663	2972	19235	628497																					
E 112 828N 8W GAS	221	4	7232	7716	5153	3832	3311	5005	5834	40974	40974	65807																					
EATON A	1501	2252	1867	2336	2081	2130	2558	1698	1647	2091	24134	2859640																					
1P2529N11W OIL	2871	4082	3595	2980	2623	2068	4237	2465	3092	1946	35340	132030																					
E 1B2529N11W GAS	46	19	19	13	15	15	30	20	20	212	212	1209																					
FIELDS	3645	2857	2862	2804	6537	1086	2997	7025	6866	5091	55774	394753																					
1M2932N11W GAS	8264	3742	3671	6010	11718	2346	4608	7814	6852	4597	76105	248770																					
E 112932N11W OIL	3557	4733	6892	2543	3663	2272	3087	4343	4776	3045	47754	356928																					
E 2E2932N11W OIL	9394	9992	9315	5808	5611	2799	4385	8464	7911	8284	87210	307280																					

BASIN DAKOTA 1984

COMPANY: THE STATE OIL COMPANY ADDRESS: P.O. BOX 3249, BALEWOOD, COLORADO 80135 MONTH: JANUARY, 1985 PAGE 76 OF

LEASE NO. 91-007454
 WELL ID: 29-29N-11W-F
 FIELD: A
 COUNTY: GARFIELD
 STATE: CO

DATE: 1/31/85
 TIME: 10:00 AM
 BY: [Signature]

OPERATOR: [Name]
 ADDRESS: [Address]
 PHONE: [Phone]

LESSOR: [Name]
 ADDRESS: [Address]
 PHONE: [Phone]

DATE: 1/31/85
 TIME: 10:00 AM
 BY: [Signature]

OPERATOR: [Name]
 ADDRESS: [Address]
 PHONE: [Phone]

LESSOR: [Name]
 ADDRESS: [Address]
 PHONE: [Phone]

DATE: 1/31/85
 TIME: 10:00 AM
 BY: [Signature]

OPERATOR: [Name]
 ADDRESS: [Address]
 PHONE: [Phone]

LESSOR: [Name]
 ADDRESS: [Address]
 PHONE: [Phone]

DATE: 1/31/85
 TIME: 10:00 AM
 BY: [Signature]

OPERATOR: [Name]
 ADDRESS: [Address]
 PHONE: [Phone]

LESSOR: [Name]
 ADDRESS: [Address]
 PHONE: [Phone]

DATE: 1/31/85
 TIME: 10:00 AM
 BY: [Signature]

OPERATOR: [Name]
 ADDRESS: [Address]
 PHONE: [Phone]

LESSOR: [Name]
 ADDRESS: [Address]
 PHONE: [Phone]

DATE: 1/31/85
 TIME: 10:00 AM
 BY: [Signature]

OPERATOR: [Name]
 ADDRESS: [Address]
 PHONE: [Phone]

LESSOR: [Name]
 ADDRESS: [Address]
 PHONE: [Phone]

DATE: 1/31/85
 TIME: 10:00 AM
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LESSOR: [Name]
 ADDRESS: [Address]
 PHONE: [Phone]

DATE: 1/31/85
 TIME: 10:00 AM
 BY: [Signature]

OPERATOR: [Name]
 ADDRESS: [Address]
 PHONE: [Phone]

ILLEGIBLE

COMPANY: CHAGRA OIL COMPANY ADDRESS: 1000 S. W. 10th Ave, Denver, Colorado 80202 MONTH: JANUARY, 1985 PAGE 143 OF 143

WATER INJECTION: 0 BBL PER DAY
 MONTHLY ACTUAL OIL: 0 BBL
 MONTHLY ACTUAL GAS: 0 BBL

DISPOSITION OF GAS: 0 BBL
 DISPOSITION OF OIL: 0 BBL

WATER INJECTION: 0 BBL PER DAY
 MONTHLY ACTUAL OIL: 0 BBL
 MONTHLY ACTUAL GAS: 0 BBL

DISPOSITION OF GAS: 0 BBL
 DISPOSITION OF OIL: 0 BBL

WATER INJECTION: 0 BBL PER DAY
 MONTHLY ACTUAL OIL: 0 BBL
 MONTHLY ACTUAL GAS: 0 BBL

DISPOSITION OF GAS: 0 BBL
 DISPOSITION OF OIL: 0 BBL

WATER INJECTION: 0 BBL PER DAY
 MONTHLY ACTUAL OIL: 0 BBL
 MONTHLY ACTUAL GAS: 0 BBL

DISPOSITION OF GAS: 0 BBL
 DISPOSITION OF OIL: 0 BBL

WATER INJECTION: 0 BBL PER DAY
 MONTHLY ACTUAL OIL: 0 BBL
 MONTHLY ACTUAL GAS: 0 BBL

DISPOSITION OF GAS: 0 BBL
 DISPOSITION OF OIL: 0 BBL

WATER INJECTION: 0 BBL PER DAY
 MONTHLY ACTUAL OIL: 0 BBL
 MONTHLY ACTUAL GAS: 0 BBL

DISPOSITION OF GAS: 0 BBL
 DISPOSITION OF OIL: 0 BBL

WATER INJECTION: 0 BBL PER DAY
 MONTHLY ACTUAL OIL: 0 BBL
 MONTHLY ACTUAL GAS: 0 BBL

DISPOSITION OF GAS: 0 BBL
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WATER INJECTION: 0 BBL PER DAY
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WATER INJECTION: 0 BBL PER DAY
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DISPOSITION OF GAS: 0 BBL
 DISPOSITION OF OIL: 0 BBL

WATER INJECTION: 0 BBL PER DAY
 MONTHLY ACTUAL OIL: 0 BBL
 MONTHLY ACTUAL GAS: 0 BBL

WELL: CHAGRA (CHAGRA)
 SUNC: 001

2363 31

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WELL: CHAGRA (CHAGRA)
 SUNC: 001

757 51

STATUS CODE: 0
 GAS CODE: 0
 OIL CODE: 0
 NAME: URSULA SULZBACH
 POSITION: DISTRICT ADMINISTRATOR
 SIGNATURE: DATE: 02/27/85
 PHONE: (303) 740-2590

ILLEGIBLE

DISPOSITION OF GAS

DISPOSITION OF OIL

WATER INJECTION	MONTHLY ACTUAL	PRODUCED DAYS	PROD	OTHER	TER	PORTER	TER	OTHER	TER	OTHER	TER

WELLS

RO (CHACRA) (CHACRA)

UNCE COM

C 19 29N 10W F 1698 31

LEASE TOTAL 1698 1698 EPG 40 U 10 10

RO (CHACRA) (CHACRA)

ATION A

COMM. AGREE NO. 91-007454

IE R 25 29N 11W F 803 20

STATUS CODE

F...FLOWING

P...PUMPING

G...GAS LIFT

S...SHUT IN

T...TEMP ABANDONED

I...INJECTION

D...DISCONTINUED

X...USED OFF LEASE

D...USED FOR DRILLING

G...GAS LIFT

L...LOST (MCF EST)

R...REPRESS-PRESS MAINT

V...VENTED

U...USED ON LEASE

G...CIRCULATING OIL

L...LOST

S...SEDIMENTATION

E...EXPLANATION ATTACH

NAME : URSULA SULZBACH

POSITION : DISTRICT ADMINISTRATOR

SIGNATURE: DATE: 04/24/85

PHONE : (303) 740-2590

ILLEGIBLE

COUNTY	BASIN	DATE	JAN	FEB	MAR	APRIL	MAY	JUNE	JULY	AUG	SEPT	OCT	NOV	DEC	PROD	RES
E	1P2729N10W	GAS	4066	4395	5495	4357	5221	609	1966	6139	6494	4109	5424	3529	52297	211237
		OIL	1	2	2	3	4			26	38	12	2	4	29	
		WAT														
MARS	ROW GAS CUM															
	1F1529N12W	GAS	6820	5984	4483	4480	3087	3	1	3	1	2		109	24963	3239151
		OIL	45	3	3	27	7								83	18865
		WAT	18	15	15	1886	1169							45	9264	106361
E	1L1529N12W	GAS	2487	2160	1517	1886	1169	3	1			4			103	359
		OIL	6	15	1	20	21									
		WAT	18													
MARTIN	GAS CUM															
	1A1527N10W	GAS	2488	2571	1408	5622	3153	1345	865	4533	4309	3439	4576	2914	37223	2670220
		OIL	42	31	2	85	65		40	28	119	1	85	65	582	41021
		WAT	1	1	1	1	1	1	1	1	1	1	1	1	1	1
E	1J1527N10W	GAS	4217	7144	1464	8010	5579	1332	1586	10481	9103	5247	8879	4920	67962	317429
		OIL	125	82	63	105	100	10	20	122	140	88	192	61	1108	4671
		WAT	1	1	1	1	1	1	1	1	1	1	1	1	1	1
MARTIN	GAS CUM															
	1G5128N10W	GAS	4056	4798	4657	5081	4664	4796	4378	3047	4654	4556	4731	4917	54335	3017793
		OIL	30	43	24	43	27	28	25	14	29	34	33	34	359	33369
		WAT														
E	1P3128N10W	GAS	7542	8022	7545	4951	8440	6408	6753	6261	6043	5964	5902	5819	79650	244684
		OIL	40	67	80	67	72	40	62	28	52	25	44	44	624	2642
		WAT														
MARTIN	GAS CUM															
	1K1127N10W	GAS	6978	3763	3274	3668	5812	4598	3665	3506	5286	3546	4034	4600	52730	1682279
		OIL	73	57	1	54	31	40	21	26	35	32	25	28	449	20454
		WAT	1	1	1	1	1	1	1	1	1	1	1	1	1	1
E	1O1127N10W	GAS	5584	5363	4613	3515	3747	4178	4014	3822	3696	3714	2988	4865	50093	2152239
		OIL	62	83	61	82	77	54	48	50	15	74	34	34	673	2933
		WAT	1	1	1	1	1	1	1	1	1	1	1	1	1	1
MARTIN	GAS CUM															
	1L1527N10W	GAS	9488	8950	3951	3823	10347	954	672	5738	8460	12190	7246	7931	79750	2846062
		OIL	102	52	39	13	92	3	1	53	42	132	72	107	707	39084
		WAT	1	1	1	1	1	1	1	1	1	1	1	1	1	1
E	1C1527N10W	GAS	4129	7129	1559	4778	3967	771	447	2324	8217	4665	5208	2155	45349	241279
		OIL	155	176	42	88	65	13	7		171	197	48	80	1042	4925
		WAT	1	1	1	1	1	1	1	1	1	1	1	1	1	1
MARTIN	GAS CUM															
	1E1427N10W	GAS	132	97											1320	2701551
		OIL	9	2											97	56548
		WAT														
E	1N1427N10W	GAS	6451	8303	8158	7592	4120	7213	9855	7158	6831	219	882	882	66782	258766
		OIL	118	95	84	43	52	45	52	74	53	19	1	1	635	5163
		WAT	1	1	1	1	1	1	1	1	1	1	1	1	1	1
R	1F1427N10W	GAS	16		1621	5411	3575	3772	6702	7813	4380	37	762	762	34073	34073
		OIL			259	225	30	67	117	115	51	31	1	1	880	880
		WAT			1	1	1	1	1	1	1	2	1	1	10	10
MARTIN	GAS CUM															
	1B1427N10W	GAS	2923	4174	5865	852	6893	8473	6965	3130	36	36	284	284	39631	2045835
		OIL	39	15	85	22	44	73	59	33	5	5	1	1	376	23124
		WAT	1	1	1	1	1	1	1	1	1	1	1	1	13	13
E	1J1427N10W	GAS	1149	6701	8610	1350	11151	13889	13405	6652	1	1	1	1325	64235	415721
		OIL	29	93	120	226	1	186	126	95	1	1	1	1	731	4525
		WAT	1	1	1	1	1	1	1	1	4	1	1	1	15	15
MARTINEZ	GAS CUM															
	1L2429N10W	GAS	1890	3062	2189	3295	1918	2444	3312	3281	3399	3649	4208	4045	36692	2224411
		OIL	2	20	45	20	3	5	17	13	15	14	10	46	213	9835
		WAT													36	36
E	1P2429N10W	GAS	5166	6231	4902	3711	3499	5019	2370	2974	4029	4285	5100	5538	52824	430020
		OIL	30	27	37	15	33	32	34	18	30	25	18	15	184	2854
		WAT	20	30	20	15	30								268	268
MARTINEZ	GAS CUM															
	1A2429N10W	GAS	3491	2826	3371	2744	2951	3542	2419	3685	523	819	2359	5793	34523	2590492
		OIL	18	2	30	13	13	24	26	22	11	4	8	32	203	11237
		WAT													12	12
E	1O2429N10W	GAS	4556	8586	8104	7164	5213	216	1088	7121	7590	6403	5398	6412	67851	3080198
		OIL	10	38	2	3	8								17	29009
		WAT													79	79
E	1D2829N11W	GAS	2445	2618	1919	5871	3843	74	437	5273	5980	3604	4475	4681	41220	183928
		OIL	9	37	5	94	20		4	54	57	35	44	21	376	1541
		WAT	6	5	4	3	5		4	3	8	10			56	56
C A	MCADAMS	B														

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	MAY	JUNE	JULY	AUG	SEPT	OCT	NOV	DEC	TOTAL
112628N10W GAS	1432	1605	1793	1485	527	814	14710	786	35793
120502N10W GAS	1930	1659	1333	1254	585	865	15394	847	35233
144252N10W GAS	2204	1283	1244	1636	1025	866	12362	784	40355
151362N10W GAS	1782	5018	5449	3931	2653	2397	34365	2874	34373
178262N10W GAS	11410	654	5449	3470	2009	2397	36773	2650	36773
PAYNE A									
E 1M1929N10W GAS	1960	1383	1234	865	770	1101	14374	980	87445
ROMERO CUM									
E 1P2526N 6W GAS	2730	2909	2906	2456	2143	1856	22805	1850	60353
SULLIVAN A									
E 1M2527N11W GAS	818	469	551	279	178	294	5285	345	24047
SULLIVAN FRAME A									
E 1A3029N10W GAS	3710	2169	2551	2257	2162	2145	30802	2487	103578
SULLIVAN FRAME CUM B									
E 103629N10W GAS	2935	1469	2663	2299	2261	2129	21916	2274	76674
BRUCE SULLIVAN CUM B									
E 1C2529N11W GAS	4803	3784	3276	3242	2995	2885	33768	3786	173754
VALDEZ A									
E 192429N11W GAS	3002	2822	3161	2604	2218	1696	21782	1610	93609
VALDEZ CUM B									
E 102429N11W GAS	5185	3763	2779	1830	14	4711	35025	3786	182431
COMPANY TOTAL OIL	64087	55681	79828	57403	51102	54714	674573	57715	2616178
TURNER PRODUCTION CO.									
TURNER 26									
112626N 7W GAS								254	2605
UNION TEXAS PETROLIUM CORPORATION									
ALBRIGHT N10W GAS	2401	1527	2012	1462	853	1647	12169	1385	99520
J 2E2221N10W GAS									
ANGEL PEAK									
E 2M 72N10W GAS	402	1150	321	1096	432	2040	20803	2744	145172
ANGEL PEAK B									
E 25E242N11W GAS	1295	1186	1174	838	1195	404	1870	1174	5520
E 30B132N11W GAS									
GUNGMESS									
E 4E3529N11W GAS	2089	1224	1248	226	408	1522	9684	751	72041
E 5P3529N11W GAS	1049	2377	1789	2484	563	1477	9381	1335	66056
E 6P3529N11W GAS	3511	3011	2757	3976	2162	1292	27063	2150	122501
E 7A3529N11W GAS	1031	3917	3414	2162	537	2474	99938	537	99938
E 8A3529N11W GAS	1891	1396	1589	1716	1527	1964	283999	1487	283999
E 9A2629N11W GAS	1851	2045	1826	1792	1537	2359	15279	1918	110045
E 10N3529N11W GAS	3431	2962	2604	2243	3036	3016	31038	1976	170471
GARLAND									
E 3M2729N11W GAS	2205	1298	832	98	1390	1942	14060	1056	96152
JICARILLA									
E 9J3429N 5W GAS	1371	389	114	538	13	200	2528	65	11430
E 7A3429N 5W GAS	731	623	447	171	171	416	5527	476	21464
MCCLANAHAN A									
E 3M2529N10W GAS	1828	1675	1589	1856	1728	2267	20962	1833	138750
NEW MEXICO B CUM									
E 101629N11W GAS	2	5559		6435	5685	5263	45931	5488	96284
REID B									
E 113129N10W GAS	2764	2764	2711	1110	2638	3582	29809	3040	157886
E 2E3129N10W GAS	2831	2587	1804	1586	1524	1329	89099	1206	157886
E 3M3129N10W GAS	2764	2764	2630	1845	1545	2779	28194	2557	166338
SUMMIT									
E 9A3329N11W GAS	2105	1794	2794	1537	691	2201	19844	2046	137362
E 10G3329N11W GAS	1333	1333	1440	867	1141	1243	17802	1402	42792
E 11K3329N11W GAS	2809	1997	1634	1845	117	2753	140443	2502	140443
WILSON									
E 253129N10W GAS	4442	5362	4682	5036	4897	5225	52424	5390	322221
MITT									
E 1P3332N11W GAS	2085	1187	1998	13	2053	1393	15313	1556	71872
ZACHRY									
E 19B3329N10W GAS	1765	4176	4875	3025	5828	6627	52434	6035	171596
E 19B3329N10W GAS	1765	4176	4875	3025	5828	6627	52434	6035	171596
E 1801129N10W GAS	2200	2402	1510	1025	1151	1611	12579	2248	167983
E 20N3329N10W GAS	3394	3980	3490	3271	2696	2667	38959	2342	173611
E 21K3329N10W GAS	4442	1929	1475	1987	2120	2329	24870	2372	182015
E 22B3329N10W GAS	1600	1588	1475	1159	1506	1432	16936	1368	150102
E 23M3329N10W GAS	4073	3335	3478	4101	1976	4416	39547	3896	200755
E 24N3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 25P3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 26R3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 27S3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 28T3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 29U3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 30V3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 31W3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 32X3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 33Y3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 34Z3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 35AA3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 36AB3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 37AC3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 38AD3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 39AE3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 40AF3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 41AG3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 42AH3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 43AI3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 44AJ3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 45AK3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 46AL3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 47AM3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 48AN3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 49AO3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 50AP3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 51AQ3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 52AR3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 53AS3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 54AT3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 55AU3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 56AV3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 57AW3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 58AX3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 59AY3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 60AZ3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 61BA3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 62BB3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 63BC3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 64BD3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 65BE3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 66BF3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 67BG3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 68BH3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 69BI3329N10W GAS	3350	3354	3478	4101	1976	2986	47254	2378	147254
E 70BJ3329N10W GAS	3350</								

36	430N	8M	15	123	1048	6141	2680	3309	2580	996	5894	22771	173	6043804	1509
A	31	430N	8M	1668	18895	23979	18895	6679	22955	14822	17999	133052	52	1854910	117
A	4H	230N	8M	1692	23130	23769	23130	21000	17624	15650	18816	162465	52	6835146	383
A	4P	530N	8M	601	340	5726	340	14051	348	2230	12675	46739	82	5160	182
A	5N	930N	8M	2656	1197	4009	1794	1794	3407	3682	7931	30521	13	782354	333
A	5U	930N	8M	9023	8747	24671	21310	16400	16989	23679	21903	144264	2	5274367	4818
LINDA NYE															299
A	1B2030N	3M	9102	5631	2824	6768	4265	8294	5664	2539	9910	68652	52	3366873	4871
A	1D2030N	8M	14936	9202	5028	11563	8164	15853	7050	6259	11148	102954	105	4871	105
E	1M192910M	10M	148	30	3	43	24	60	20	8	15	265	265	1251139	5637
PRITCHARD															59726
A	1M130N	9M	6792	2089	11859	14245	331	61119	129763	125	16760	53310	22	4100353	1289
A	1F130N	9M	9628	2558	4128	11388	10568	10927	4208	125	1636	45821	49	1265882	474
A	2H130N	9M	142810	110033	89464	42776	10568	65	129763	71642	121331	1119149	2	5408051	1225
A	2I130N	9M	262	136	88	11388	64	10927	4208	155	234	2032	2	47206	1225
A	3H3129N	8M	2	1167	353	41	3521	5070	3608	4383	6863	54047	2	526531	47206
A	3P3129M	8M	6485	4658	5085	5625	8055	5294	2977	3415	4491	50618	30	3871	3871
A	3P3129M	8M	40	12	18	7	8055	5294	11	1274	3267	3072266	169	5976	5976
A	4L3129N	8M	117	4668	5686	2	2257	2299	24	1274	3267	46623	296	490022	490022
A	4L3129N	8M	3031	955	2070	2340	2257	2299	1703	1167	2090	23223	296	2809381	2884
A	4F3129N	8M	12568	2274	11815	2308	151M	14989	5697	3699	7759	84995	231	620909	1211
RIDDLE															1211
A	1B2130N	9M	22025	4079	2558	10015	3848	12771	4020	4053	12977	83974	10829553	10829553	10829553
A	1E2130N	9M	86	3	163	24	351	7437	18	390	5199	31404	842079	842079	842079
A	2M1730N	9M	969	1552	2712	4343	351	7437	1726	390	5199	31404	842079	842079	842079
A	2J1730N	9M	2288	6001	2712	12	1	25	1726	390	5199	31404	842079	842079	842079
A	2J1730N	9M	20210	4773	331	12	1	25	1726	390	5199	31404	842079	842079	842079
RIDDLE															1211
A	1A1927N	9M	1047	927	1593	1628	303	708	1019	976	907	11652	632416	632416	632416
RIDDLE															2678
M	111927N	9M	1960	3117	2217	508	1072	1254	1902	1447	1030	16800	26229	26229	26229
RUMERS															438
A	1P2526N	6M	7157	296	4351	9874	370	176	2801	13	535	24723	97985	97985	97985
STATE															7215
A	1M3230N	9M	605	1145	2092	2144	1014	1059	228	3	3803	10806	1888488	1888488	1888488
A	1E5230N	9M	10316	17048	15330	12665	6096	1059	228	125	20600	82180	987894	987894	987894
STATE															722
A	2M1630N	9M	836	84	38	1247	4891	114	114	3599	16795	26745	4733888	4733888	4733888
A	2C1650N	9M	35	1530	114	307	1530	114	114	15	5569	7523	12349	12349	12349
STATE															2139
A	3A1630N	9M	6314	2803	794	1405	2412	3989	2938	2316	3654	34587	4633181	4633181	4633181
A	3P1650N	9M	13015	809	534	1582	3480	1169	2556	15591	19085	91544	943016	943016	943016
STATS															1851
A	105227N	9M	9138	10432	6927	7184	6286	3020	119	13	355	45124	98633	98633	98633
A	105227N	9M	133	119	100	23	119	100	119	13	355	45124	98633	98633	98633

Mesa Verde 1984

DISPOSITION OF GAS

DISPOSITION OF OIL

LEASE NAME	STATUS	DATE	ACTUAL	PRODUCED	WATER	GAS	PRODUCED	TRANS	PORT-	OTHER	MONTH	OTHER	MONTH
1	B	20	30N	3W	F	37	37	9138	31				
1A	2	20	30R	3X	F	81	81	13011	27				
LEASE TOTAL			118	113		22149	22085	506	276	258	PLA		136

BLANCO (MESAVERDE) (MESAVERDE)
 PRITCHARD 1,1-1-1-2,2-A

LEASE TOTAL 71 71 114644 109138 5456 U 418 489

1L N 19 29N 10W F 2 2 722 31

LEASE TOTAL 2 2 722 687 35 U 119 121

1 1 1 20N 2W F 31 31 14006 29

STATUS CODE: X...USED OFF LEASE, U...USED FOR DRILLING, P...PUMPING, G...GAS LEFT, S...GAS LEFT, L...LOST LEASE, E...EXPLANATION ATTACHED, R...REPRESS-PRESS MAINT, V...VENTED, U...USED FOR LEAS

GAS CODE: USED OFF LEASE, USED FOR DRILLING, PUMPING, GAS LEFT, LOST LEASE, EXPLANATION ATTACHED, REPRESS-PRESS MAINT, VENTED, USED FOR LEAS

OIL CODE: C...CIRCULATING OIL, L...LOST, S...SEQUESTRATION, L...EXPLANATION ATTACHED

NAME: URSULA SULZBACH, POSITION: DISTRICT ADMINISTRATOR, SIGNATURE: [Signature], DATE: 02/20/85

PHONE: (303) 740-2590

ILLEGIBLE

COMPANY: TELECO OIL COMPANY ADDRESS: P.O. BOX 3247 EAGLECLIFF, COLORADO 80155 MONTH: JANUARY, 1985 PAGE 145 OF 145

LEASE NO.	WATER INJECTION	TOTAL LIQUIDS PRODUCED		DISPOSITION OF GAS		DISPOSITION OF OIL	
		WATER	INJECTION	GAS	WATER	OIL ON HAND	BARRELS TO TRANS-PORT
1							
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97							
98							
99							
100							

GILRO (CHACRA) (CHACRA)
JACKSON 272A, 82L

LE N 34 294 9w F 557 31

LE N 33 29N 9w F 2913 31

LE N 19 29N 10w F 883 31

LE N 25 26w 6w F 1698 31

LE N 34 294 9w F 557 31

LE N 33 29N 9w F 2913 31

LE N 19 29N 10w F 883 31

LE N 25 26w 6w F 1698 31

GILRO (CHACRA) (CHACRA)
JACKSON 272A, 82L

LE N 34 294 9w F 557 31

LE N 33 29N 9w F 2913 31

LE N 19 29N 10w F 883 31

LE N 25 26w 6w F 1698 31

GILRO (CHACRA) (CHACRA)
JACKSON 272A, 82L

LE N 34 294 9w F 557 31

ILLEGIBLE

PHONE : (303) 740-2590

DATE: 02/26/85

SIGNATURE: _____

POSITION: DISTRICT ADMINISTRATOR

NAME: URSULA SUEZBACH

OIL CODE: _____

CIRCULATING OIL: _____

LOSS: _____

SEEDIMENTATION: _____

EXPLANATION ATTACH: _____

EXPLANATION ATTACHED: _____

REPRESS-MAINT: _____

VENTED: _____

USED ON LEASE: _____

USED OFF LEASE: _____

FOR DRILLING: _____

GAS LIFT: _____

GAS LIFT (MCF EST): _____

EXPLANATION ATTACHED: _____

REPRESS-MAINT: _____

VENTED: _____

USED ON LEASE: _____

USED OFF LEASE: _____

FOR DRILLING: _____

GAS LIFT: _____

GAS LIFT (MCF EST): _____

EXPLANATION ATTACHED: _____

REPRESS-MAINT: _____

VENTED: _____

USED ON LEASE: _____

USED OFF LEASE: _____

FOR DRILLING: _____

GAS LIFT: _____

GAS LIFT (MCF EST): _____

EXPLANATION ATTACHED: _____

REPRESS-MAINT: _____

VENTED: _____

USED ON LEASE: _____

USED OFF LEASE: _____

FOR DRILLING: _____

GAS LIFT: _____

GAS LIFT (MCF EST): _____

EXPLANATION ATTACHED: _____

REPRESS-MAINT: _____

VENTED: _____

USED ON LEASE: _____

USED OFF LEASE: _____

FOR DRILLING: _____

GAS LIFT: _____

GAS LIFT (MCF EST): _____

EXPLANATION ATTACHED: _____

REPRESS-MAINT: _____

VENTED: _____

USED ON LEASE: _____

USED OFF LEASE: _____

FOR DRILLING: _____

GAS LIFT: _____

GAS LIFT (MCF EST): _____

EXPLANATION ATTACHED: _____

REPRESS-MAINT: _____

VENTED: _____

USED ON LEASE: _____

USED OFF LEASE: _____

FOR DRILLING: _____

GAS LIFT: _____

GAS LIFT (MCF EST): _____

EXPLANATION ATTACHED: _____

REPRESS-MAINT: _____

VENTED: _____

USED ON LEASE: _____

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FOR DRILLING: _____

GAS LIFT: _____

GAS LIFT (MCF EST): _____

EXPLANATION ATTACHED: _____

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FOR DRILLING: _____

GAS LIFT: _____

GAS LIFT (MCF EST): _____

EXPLANATION ATTACHED: _____

REPRESS-MAINT: _____

VENTED: _____

USED ON LEASE: _____

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FOR DRILLING: _____

GAS LIFT: _____

GAS LIFT (MCF EST): _____

EXPLANATION ATTACHED: _____

REPRESS-MAINT: _____

VENTED: _____

USED ON LEASE: _____

USED OFF LEASE: _____

FOR DRILLING: _____

GAS LIFT: _____

GAS LIFT (MCF EST): _____

EXPLANATION ATTACHED: _____

REPRESS-MAINT: _____

VENTED: _____

USED ON LEASE: _____

USED OFF LEASE: _____

FOR DRILLING: _____

GAS LIFT: _____

GAS LIFT (MCF EST): _____

EXPLANATION ATTACHED: _____

REPRESS-MAINT: _____

VENTED: _____

USED ON LEASE: _____

USED OFF LEASE: _____

FOR DRILLING: _____

GAS LIFT: _____

GAS LIFT (MCF EST): _____

EXPLANATION ATTACHED: _____

REPRESS-MAINT: _____

VENTED: _____

USED ON LEASE: _____

USED OFF LEASE: _____

FOR DRILLING: _____

GAS LIFT: _____

GAS LIFT (MCF EST): _____

EXPLANATION ATTACHED: _____

REPRESS-MAINT: _____

VENTED: _____

USED ON LEASE: _____

USED OFF LEASE: _____

FOR DRILLING: _____

GAS LIFT: _____

GAS LIFT (MCF EST): _____

EXPLANATION ATTACHED: _____

REPRESS-MAINT: _____

VENTED: _____

USED ON LEASE: _____

COMPANY: TENJICO OIL COMPANY ADDRESS: P.O. BOX 3249 ENGLEWOOD, COLORADO 80155 MONTH: FEBRUARY, 1985 PAGE 71 OF 71

WATER INJECTION: TOTAL LIQUIDS PRODUCED: 7324 28
 MONTHLY ACTUAL OIL: 65 65
 GAS: 7324 28
 IRANS: 36 U 185
 PORT: 7288 EPG
 D BEG. OF TRANS: 36 U 185
 OTHER F MONTH: 1549 EPG
 PURTIER: 72 U 97
 OTHER F MONTH: 124

WELLS

IN (DAKOTA) (DAKOTA)

WYNE A

65 65 7324 28

LAST TOTAL 65 65 7324 2888 EPG 36 U 185 250

IN (DAKOTA) (DAKOTA)

WYNE A

1585 28

LAST TOTAL 1585 28 1549 EPG 36 U 124

IN (DAKOTA) (DAKOTA)

WYNE A

21 21 2510 28

6 6 212 28

LAST TOTAL 21 21 2722 2650 EPG 72 U 97 124

IN (DAKOTA) (DAKOTA)

WYNE A

19 19 1295 28

LAST TOTAL 19 19 1295 28

STATES-GHRT 645 600E 811 600F

X.....USED OFF LEASE C.....CIRCULATING OIL NAME : URSULA SULZBACH

O.....PUMPING O.....USED FOR DRILLING L.....LOST POSITION : DISTRICT ADMINISTRATOR

G.....GAS LIFT G.....GAS LIFT S.....SEDIMENTATION

S.....SHUT IN L.....LOST (MCF-EST) F.....EXPLANATION ATTACH SIGNATURE: DATE: 03/22/85

T.....TEMP ABANDONED E.....EXPLANATION ATTACHED

R.....REPRESS V.....VENTED PHONE : (303) 740-2590

O.....DISCONTINUED U.....USED ON SE

ILLEGIBLE

NEW MEXICO OIL CONSERVATION COMMISSION
OPERATOR'S MONTHLY REPORT

APART: TEMPECO OIL COMPANY ADDRESS: P.O. BOX 3249 ENGLEWOOD, COLORADO 80155 MONTH: FEBRUARY, 1985 PAGE 115 OF 115

LEASE NAME: WATER INJECTION
TOTAL LIQUIDS PRODUCED: 18026 18
MONTHLY ACTUAL: OIL 18026 18
WATER 0
ALLOW. PRODUCED (BBL) 18026 18
PRODUCED DAYS (MCE) 18
DISPOSITION OF GAS: 489
DISPOSITION OF OIL: 518

WELLS: 19 30N 8W F 5 5 18026 18
20 30N 8W F 29 29 86948 82464 SUG 4484 U 489 518

WELLS: 23 23 7884 28
22 22 10925 28
EASE TOTAL 55 55 18809 18747 SUG 62 U 136 191

WELLS: 19 29N 10W F 9 9 798 28
EASE TOTAL 9 9 798 767 FPG 31 U 121 130

WELLS: 20 20 11065 24
EASE TOTAL 20 20 11065 24

WELLS: 20 20 11065 24
EASE TOTAL 20 20 11065 24

WELLS: 20 20 11065 24
EASE TOTAL 20 20 11065 24

STATUS CODE: F... FLOWING X... USED OFF LEASE
P... PUMPING D... USED FOR DRILLING
G... GAS LIFT G... GAS LIFT
S... SHUT IN L... LIFT (MCF-FSI)
E... EXPLANATION ATTACHED
D... DISCONTINUED R... REPRESS PRESS MAINT
V... VENTED U... USED ON ASE

NAME: URSULA SULZBACH
POSITION: DISTRICT ADMINISTRATOR
SIGNATURE: DATE: 03/22/85
PHONE: (303) 740-2590

ILLEGIBLE

COMPANY: TENVEDO OIL COMPANY ADDRESS: P.O. BOX 3249 ENGLEWOOD, COLORADO 80155 MONTH: MARCH 1985 PAGE 72 OF

WATER INJECTION
 LEASE NAME: S
 MONTHLY ACTUAL
 OIL OIL GAS
 ALLOW. PRODUCED PRODUCED DAYS
 (BBL) (BBL) (MCF) PRD SOLD TER OTHER E MUNT E MUNT H

DISPOSITION OF GAS: 1916 1876 EPG 40 U
 DISPOSITION OF OIL: C OIL ON BARRELS O HAND TO TRANS C OIL ON
 O HAND TO TRANS O HAND TO TRANS D BEG: OF TRANS PORT D END OF
 OTHER E MUNT PORTER TER OTHER E MUNT H

LEASE TOTAL 1916 1876 EPG 40 U

WELLS

PATTERSON

LEASE TOTAL 1916 1876 EPG 40 U

WELLS

STATUS CODE: F.....FLOWING X.....USED OFF LEASE GAS CODE: C.....CIRCULATING OIL OIL CODE: ORSOLA SOLZBACH
 P.....PUMPING D.....USED FOR DRILLING L.....LOST
 S.....GAS LIFT G.....GAS LIFT S.....SEDIMENTATION POSITION: DISTRICT ADMINISTRATOR
 SHUT IN L.....LOST (MCF EST) E.....EXPLANATION ATTACH SIGNATURE: DATE: 04/24/85
 TEMP ABANDONED E.....EXPLANATION ATTACHED
 INJECTION R.....REPRESS-PRESS MAINT
 DISCONTINUED V.....VENTED U.....USED ON LEASE PHONE: (303) 740-2590

LEASE NAME: _____ WATER INJECTION _____ TOTAL LIQUIDS PRODUCED _____ MONTHLY ACTUAL _____ DISPOSITION OF GAS _____ DISPOSITION OF OIL _____

STATUS CODE: _____ GAS CODE: _____ OIL CODE: _____

WELLS: _____

PERO (CHACRA) (CHACRA) JACKSON 1,1A,8E

LEASE TOTAL 1916 25 1888 EPG 28 U

PERO (CHACRA) (CHACRA) JACKSON 2,2A,8E

LEASE TOTAL 1927 8 1917 EPG 10 U

PERO (CHACRA) (CHACRA) DAYNE A

LEASE TOTAL 927 25 884 EPG 43 U

PERO (CHACRA) (CHACRA) SOMERO CUM

LEASE TOTAL 1657 31

STATUS CODE: _____ GAS CODE: _____ OIL CODE: _____

F.....FLOWING X.....USED OFF LEASE C.....CIRCULATING OIL NAME: _____

P.....PUMPING D.....USED FOR DRILLING L.....LUST S.....SEDIMENTATION POSITION: DISTRICT ADMINISTRATOR

G.....GAS LIFT G.....GAS LIFT L.....LUST (MCF EST) E.....EXPLANATION ATTACH SIGNATURE: _____ DATE: 04/24/85

S.....SHUT IN L.....LUST (MCF EST) E.....EXPLANATION ATTACHED R.....REPRESS-MAINT U.....VENTED ON LEASE

T.....TEMP ABANDONED E.....EXPLANATION ATTACHED V.....VENTED ON LEASE

I.....INJECTION R.....REPRESS-MAINT U.....USED ON LEASE

U.....DISCONTINUED V.....VENTED ON LEASE

PHONE: (303) 740-2590

CHAPTER III FIELD OPERATIONS

RULE 3-100 POLLUTION ABATEMENT

RULE 3-101 PROHIBITION OF POLLUTION

(a) All operators, contractors, drillers, service companies, pipepulling and salvaging contractors, or other persons shall at all times conduct their operations and drill, equip, operate, produce, plug and abandon all wells drilled for oil or gas, service wells or exploratory wells (including seismic, core and stratigraphic holes) in a manner that will prevent pollution and the migration of oil, gas, salt water or other substance from one stratum into another, including any fresh water bearing formation. Pollution of surface or subsurface fresh water by deleterious substances used in connection with the exploration, drilling, producing, refining, transporting or processing of oil or gas is hereby prohibited.

(b) Sections 305, 306, 307 and 308 of Title 52, Oklahoma Statutes Annotated, governing the drilling, operation and plugging of oil and gas wells in workable coal beds are hereby adopted as rules of the Commission as fully as if set out verbatim herein.

RULE 3-102 ADMINISTRATION AND ENFORCEMENT OF RULES

The Manager of Pollution Abatement shall supervise and coordinate the administration and enforcement of these rules under the direction of the Director of Conservation and the Commission.

RULE 3-103 COOPERATION WITH OTHER AGENCIES

(a) These rules shall not be construed as modifying the rights, obligations or duties of any person under any law of this State, or under any order, rule or regulation of the Oklahoma Water Resources Board, State Department of Health, Oklahoma Wildlife Conservation Commission, State Board of Agriculture, Department of Pollution Control, or any other agency of this State with respect to the pollution of fresh water.

(b) Whenever a written complaint against any person is filed with the Commission, alleging pollution as prohibited by Rule 3-101, the Manager of Pollution Abatement shall immediately initiate such action as may be necessary or appropriate to abate the pollution.

RULE 3-104 PITS AND TANKS

(a) Pits and tanks for drilling mud or deleterious substances used in the drilling, completion and recompletion of wells shall be constructed and maintained so as to prevent pollution of surface and subsurface fresh water.

(b) Deleterious fluids other than fresh water drilling fluids that were used in drilling or workover operations, which are displaced or produced in well completion or stimulation procedures such as from

fracturing, acidizing, swabbing, drill stem tests, and any other well stimulation process, shall be collected into a plastic lined pit of at least 30 mil, or metal tank and maintained separate from above-mentioned drilling fluids to allow for separate and legal disposal. (3-30-82)

RULE 3-105 SURFACE AND PRODUCTION CASING

(a) Owners, operators and drilling contractors shall comply with Rule 3-206, "Drilling and Casing Procedures" and Rule 3-301, "Approval of Enhanced Recovery Injection Wells or Disposal Wells". (3-16-81)

(b) In the event a rupture, break or opening occurs in the surface or production casing, the owner, operator or drilling contractor shall take immediate action to repair it, and shall report the occurrence to the appropriate District Office or the Manager of Pollution Abatement.

RULE 3-106 FRACTURE AND ACIDIZING

In the completion of an oil, gas, injection, disposal or service well, where acidizing or fracture processes are used, no oil, gas or deleterious substances shall be permitted to pollute any surface and subsurface fresh water.

RULE 3-107 SWABBING AND BAILING

In swabbing, bailing or purging a well, all deleterious substances removed from the bore hole shall be placed in adequate pits or tanks, and no such substances shall be permitted to pollute any surface and subsurface fresh water.

RULE 3-108 PRODUCING OIL AND GAS WELLS

All wellhead connections, surface equipment and tank batteries shall be maintained at all times so as to prevent leakage of oil, gas, salt water or other deleterious substances.

RULE 3-109 OIL STORAGE

Oil storage tanks shall be constructed so as to prevent leakage; and dikes or walls, where necessary, shall be constructed so as to prevent oil or deleterious substances from polluting surface and sub-surface water.

RULE 3-110 USE OF EARTHEN PITS

RULE 3-110.1 USE OF ON-SITE EARTHEN PITS

(a) An earthen pit serving only the lease or unit on which it is located is defined as an on-site pit. An on-site earthen pit used for the handling, storage or disposal of any deleterious substance produced, obtained, or used in connection with the drilling or

operation of wells, shall be constructed of, or sealed with, an impervious material, and shall be used and operated at all times so as to prevent any escape of any deleterious substance. (4-2-81)

(b) No on-site earthen pit shall be constructed, enlarged, reconstructed, or used until the District Office has issued a written permit for its use and assigned a permit number. The operator shall file Form 1014, in triplicate, with the appropriate District Office. When approved, one copy will be returned to the operator as a permit which shall bear the permit number assigned. The operator shall post a waterproof sign bearing the name of the operator and the permit number within twenty-five (25) feet of the pit. (4-2-81)

(c) Every on-site earthen pit not having a permit and permit number shall be emptied and leveled. (4-2-81)

(d) Paragraph (b) and (c) above, shall not apply to:

(1) An emergency pit constructed solely to prevent escape of substances. Provided, an emergency pit shall not be constructed in pervious soil unless lined, and shall never be used for the storage of any substance. (4-2-81)

(2) A circulating, frac or reserve mud pit used in drilling, deepening, testing, reworking or plugging a well while such operations are in progress. Each reserve pit shall be leveled within twelve (12) months after drilling operations cease. One six-month extension may be granted by the District Manager for reasonable cause. Each circulating pit shall be emptied and leveled within sixty (60) days after the drilling operations cease. Each fracture pit shall be emptied and leveled within sixty (60) days after completion of fracture operations. Provided, however, upon application, notice and hearing, and not less than ten (10) days notice by restricted mail to the occupying owner or tenant of the land upon which the pit is located, and for good cause shown, reasonable extensions of the times set out above may be granted. (4-2-81)

(3) A burn pit used solely to burn waste oil or other flammable material. Provided, a burn pit shall never be used for storage of any substance. (4-2-81)

(e) Notice of construction of an on-site emergency pit or burn pit shall be filed, in triplicate, with the appropriate District Office on Form 1014. The appropriate District Office shall be notified in writing of each use of an emergency pit. (4-2-81)

(f) No on-site earthen pit shall be constructed or maintained so as to receive outside runoff water and the fluid level of each earthen pit shall be maintained at all times at least eighteen (18) vertical inches below the lowest point of the embankment. (3-30-82)

(g) The appropriate District Office shall be notified in writing whenever an on-site earthen pit is abandoned. (4-2-81)

RULE 3-110.2 USE OF OFF-SITE EARTHEN PITS

(a) Any earthen pit not defined in Rule 3-110.1 is defined as an off-site earthen pit. An off-site earthen pit used for the handling, storage or disposal of any deleterious substance produced, obtained, or used in connection with the drilling or operation of wells, shall be constructed of, or sealed with, an impervious material, and shall be used and operated at all times so as to prevent any escape of any deleterious substance. (3-30-82)

(b) No off-site earthen pit shall be constructed, enlarged, reconstructed, or used until the District Office has issued a written permit for its use and assigned a permit number. The operator shall file Form 1014, in triplicate, with the appropriate District Office. When approved, one copy will be returned to the operator as a permit which shall bear the permit number assigned. The operator shall post a waterproof sign bearing the name of the operator and the permit number within twenty-five (25) feet of the pit. If Form 1014 is not approved by the appropriate District Office, or if a protest is received at the district level, the operator may file an application for hearing with the Commission, which shall be set for hearing. (4-2-81)

(c) Notice that an application has been filed with the Commission shall be published by the applicant in a newspaper of general circulation and published in the county in which the pit is located and not less than ten (10) days notice by restricted mail to the occupying owner or tenant of the land upon which the pit is located. The applicant shall file proof of publication prior to the hearing. (4-2-81)

(d) Every off-site earthen pit not having a permit and permit number shall be emptied and leveled. (4-2-81)

(e) Every off-site earthen pit shall be completely enclosed by a permanent woven wire fence of at least four (4) feet in height. (4-2-81)

(f) No off-site earthen pit shall be constructed or maintained so as to receive outside runoff water and the fluid level of each earthen pit shall be maintained at all times at least eighteen (18) vertical inches below the lowest point of the embankment. (3-30-82)

(g) The appropriate District Office shall be notified in writing whenever an off-site earthen pit is abandoned. (4-2-81)

(h) The provisions of Rule 3-110.2 shall not apply to an off-site reserve pit used for primary drilling operations. (4-2-81)

(i) Use of off-site earthen pits designed specifically for disposal of deleterious substances from more than one well site shall meet the additional following requirements: (3-30-82)

ILLEGIBLE

- (1) No off-site earthen pit shall be constructed or maintained so as to receive outside runoff water and the fluid level in the off-site earthen pit shall be maintained at all times at least twenty-four (24) vertical inches below the lowest point of the embankment. (3-30-82)
- (2) No off-site earthen pit shall be constructed in the 100 year flood plain of any drainage basin. (3-30-82)
- (3) No off-site earthen pit shall contain fluids with a chloride content greater than 3500 MG/L. (3-30-82)
- (4) No off-site earthen pit shall contain a soil seal less than 12 inches thick with the co-efficient of permeability no greater than 10^{-7} cm/sec. If a Bentonite seal is to be used, the Bentonite shall be mixed to form the previously mentioned permeability requirement into the soil to a uniform depth of at least 6 inches. (3-30-82)
- (5) Two test borings shall be drilled to a minimum depth of 25' below the bottom of the earthen pit, and to be located outside of and near the low elevation side of the pit. The borings shall be submitted with the application to demonstrate the subsurface profile of the proposed pit. (3-30-82)
- (6) Any earthen pit that contains deleterious substances shall be lined so as to prevent contamination of the fresh water. The type of liner proposed shall be approved by the Commission's District Manager and Manager of Pollution Abatement. (3-30-82)
- (7) Written certification that the seal was provided and constructed in accordance with Commission-approved specifications shall be furnished by the supplier, project engineer, or independent soils laboratory. (3-30-82)
- (8) All off-site earthen pits shall be filled and leveled within one (1) year after abandonment. (3-30-82)
- (9) No abandoned mines or strip pits shall be used for disposal of oilfield waste unless the geology and hydrology demonstrate that such disposal will not contaminate the fresh water of the state. (3-30-82)
- (10) No off-site earthen pit shall contain deleterious substances unless the geology and hydrology demonstrate that such disposal will not contaminate the fresh water of the state. (3-30-82)

RULE 3-110.3 AGRICULTURAL USE OF OIL FIELD WASTE PROHIBITED

Any spreading and/or soil farming of oil field drilling waste shall be prohibited.

RULE 3-111 REFINING AND PROCESSING OF OIL AND GAS

(a) All deleterious substances obtained or used in the processing and refining of oil and gas shall be disposed of in a manner that will prevent the pollution of fresh water.

(b) Chemicals, gasolines, oils and other deleterious substances shall be stored, where necessary, in tanks or containers of a material and of a construction and in a manner that will prevent the escaping, seepage, or draining of such liquids into any fresh water.

RULE 3-114 PROTECTION OF MUNICIPAL WATER SUPPLIES

The Commission, upon application of any municipality or other governmental subdivision, may enter an order establishing special field rules within a defined area to protect and preserve fresh water and fresh water supplies.

RULE 3-120 INSPECTION AND ENFORCEMENT

RULE 3-121 INFORMAL COMPLAINTS

If, upon information or inspection, it is found that an operator, processor, refiner, or transporter of oil or gas is violating any rule or order of the Commission or causing damage or pollution to any oil or gas formation, surface or underground fresh water, the Conservation Division shall cause an investigation to be made and shall file a written administrative complaint, in duplicate, on Form 1036, and one copy of Form 1036 shall be delivered or mailed to the operator. If, upon subsequent inspection it is determined that the operator has taken the corrective actions specified the complaint shall be dismissed; otherwise, formal application will be made to the Commission for an order shutting down the lease or well, and for any other appropriate remedy; pending the outcome of the final determination of the Commission on the formal application, any District Manager shall, after an on-site inspection, have the authority to shut down those operations where conditions appear obvious that surface or underground pollution is occurring.
(4-2-61)

RULE 3-200 DRILLING AND DEVELOPMENT

RULE 3-201.1 OPERATORS AGREEMENT, FINANCIAL STATEMENT, ETC.

(a) Each person who drills or operates any well within the State of Oklahoma for the exploration, development or production of oil or gas, or as an injection or disposal well, shall furnish his agreement in writing to plug the well at the time and in the manner prescribed by the Rules and Regulations of the Commission and the laws of the State of Oklahoma. The agreement shall provide that if the Commission determines that he has neglected, failed or refused to plug any well in compliance with the Commission's Rules and Regulations, he will forfeit or pay to the State, through the

RAILROAD COMMISSION OF TEXAS
OIL AND GAS DIVISION



STATE OF TEXAS
RAILROAD COMMISSION
JAMES E. (JIM) HENDERSON, Commissioner
JAMES E. (JIM) HENDERSON, Commissioner
CAPTION: STAFFERS - P. O. CHAMBERLAIN, JR.
ALVIN TEXAS 771

NOTICE OF RULE ADOPTION

The following is a copy of amended Statewide Rule 8 relating to Water Protection (16 TAC §3.8) as amended by the Railroad Commission of Texas on March 5, 1984. These amendments will go into effect on May 1, 1984.

William H. Barnes
William H. Barnes
Legal Counsel
Underground Inspection Control

Railroad Commission of Texas
Oil and Gas Division

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§ 3.8. Water Protection.

(a) Definitions. The following words and terms, when used in this section, shall have the following meanings, unless the context clearly indicates otherwise:

- (1) Basic sediment pit -- Pit used in conjunction with a tank battery for storage of basic sediment removed from a production vessel or from the bottom of an oil storage tank. Basic sediment pits were formerly referred to as burn pits.
- (2) Brine pit -- Pit used for storage of brine which is used to displace hydrocarbons from an underground hydrocarbon storage facility.
- (3) Collecting pit -- Pit used for storage of saltwater prior to disposal at a tidal disposal facility, or pit used for storage of saltwater or other oil and gas wastes prior to disposal at a disposal well or fluid injection well. In some cases one pit is both a collecting pit and a skimming pit.
- (4) Completion/workover pit -- Pit used for storage or disposal of spent completion fluids, workover fluids, and drilling fluid, silt, debris, water, brine, oil scum, paraffin, or other materials which have been cleaned out of the well bore of a well being completed or worked over.
- (5) Drilling fluid disposal pit -- Pit, other than a reserve pit, used for disposal of spent drilling fluid.
- (6) Drilling fluid storage pit -- Pit used for storage of drilling fluid which is not currently being used but which will be used in future drilling operations. Drilling fluid storage pits are often centrally located among several leases.
- (7) Emergency saltwater storage pit -- Pit used for storage of produced saltwater for limited period of time. Use of the pit is necessitated by a temporary shutdown of a disposal well or fluid injection well and/or

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associated equipment, by temporary overflow of saltwater storage tanks on a producing lease, or by a producing well loading up with formation fluids such that the well may die. Emergency saltwater storage pits may sometimes be referred to as emergency pits or blowdown pits.

- (8) Flare pit -- Pit which contains a flare and which is used for temporary storage of liquid hydrocarbons which are sent to the flare during equipment malfunction but which are not burned. A flare pit is used in conjunction with a gasoline plant, natural gas processing plant, pressure maintenance or repressurizing plant, tank battery, or a well.
- (9) Fresh makeup water pit -- Pit used in conjunction with drilling rig for storage of water used to make up drilling fluid.
- (10) Gas plant evaporation/retention pit -- Pit used for storage or disposal of cooling tower blowdown, water condensed from natural gas, and other wastewater generated at gasoline plants, natural gas processing plants, or pressure maintenance or repressurizing plants.
- (11) Mud circulation pit -- Pit used in conjunction with drilling rig for storage of drilling fluid currently being used in drilling operations.
- (12) Reserve pit -- Pit used in conjunction with drilling rig for collecting spent drilling fluids; cuttings, sands, and slits; and wash water used for cleaning drill pipe and other equipment at the well site. Reserve pits are sometimes referred to as slush pits or mud pits.
- (13) Saltwater disposal pit -- Pit used for disposal of produced saltwater.
- (14) Skimming pit -- Pit used for skimming oil off saltwater prior to disposal of saltwater at a tidal disposal facility, disposal well, or fluid injection well.

- (15) Washout pit -- Pit located at truck yard, tank yard, or disposal facility for storage or disposal of oil and gas waste residue washed out of trucks, mobile tanks, or skid-mounted tanks.
- (16) Water condensate pit -- Pit used in conjunction with a gas pipeline strip or gas compressor station for storage or disposal of fresh water condensed from natural gas.
- (17) Generator -- Person who generates oil and gas wastes.
- (18) Carrier -- Person who transports oil and gas wastes generated by a generator. A carrier of another person's oil and gas wastes may be a generator of his own oil and gas wastes.
- (19) Receiver -- Person who stores, handles, treats, reclaims, or disposes of oil and gas wastes generated by a generator. A receiver of another person's oil and gas wastes may be a generator of his own oil and gas wastes.
- (20) Director -- Director of the Oil and Gas Division or his staff delegate designated in writing by the Director of the Oil and Gas Division or the commission.
- (21) Person -- Natural person, corporation, organization, government or governmental subdivision or agency, business trust, estate, trust, partnership, association, or any other legal entity.
- (22) Affected person -- Person who, as a result of the activity sought to be permitted, has suffered or may suffer actual injury or economic damage other than as a member of the general public.
- (23) To dewater -- To remove the free water.
- (24) To dispose -- To engage in any act of disposal subject to regulation by the commission including, but not limited to, conducting, draining, discharging, emitting, throwing, releasing, depositing, burying, landfarming, or allowing to seep, or to cause or allow any such act of disposal.

- (25) Landfarming -- A waste management practice in which oil and gas wastes are mixed with or applied to the land surface in such a manner that the waste will not migrate off the landfarmed area.
- (26) Oil and gas wastes -- Materials to be disposed of or reclaimed which have been generated in connection with activities associated with the exploration, development, and production of oil or gas or geothermal resources, or activities associated with underground storage of hydrocarbons. The term oil and gas wastes includes, but is not limited to, saltwater, other mineralized water, sludge, spent drilling fluids, cuttings, waste oil, spent completion fluids, and other liquid, semi-liquid, or solid waste material.
- (27) Oil field fluids -- Fluids to be used or reused in connection with activities associated with the exploration, development, and production of oil or gas or geothermal resources, or activities associated with underground storage of hydrocarbons. The term oil field fluids includes, but is not limited to, drilling fluids, completion fluids, surfactants, and chemicals used to deaerate oil and gas wastes.
- (28) Pollution of surface or subsurface water -- The alteration of the physical, thermal, chemical, or biological quality of, or the contamination of, any surface or subsurface water in the state that renders the water harmful, detrimental, or injurious to humans, animal life, vegetation, or property, or to public health, safety, or welfare, or impairs the usefulness or the public enjoyment of the water for any lawful or reasonable purpose.
- (29) Surface or subsurface water -- Groundwater, percolating or otherwise, suitable for domestic or livestock use, irrigation of crops, or industrial use, and lakes, bays, ponds, impounding reservoirs, springs, rivers, streams, creeks, estuaries, marshes, inlets, canals, the Gulf of Mexico inside the territorial limits of the state, and all other bodies of surface water.

- natural or artificial, inland or coastal, fresh or salt, navigable or nonnavigable, and including the beds and banks of all watercourses and booms of surface water, that are wholly or partially inside or bordering the state or inside the jurisdiction of the state.
- (b) No pollution. No person conducting activities subject to regulation by the commission may cause or allow pollution of surface or subsurface water in the state.
- (c) Exploratory wells. Any oil, gas, or geothermal resource well or well drilled for exploratory purposes shall be governed by the provisions of statewide or field rules which are applicable and pertain to the drilling, safety, casing, production, abandonment, and plugging of wells.
- (d) Pollution control.
- (1) Prohibited disposal methods. Except for those disposal methods authorized for certain wastes by paragraph (3) of this subsection or subsection (e) of this section, or disposal methods permitted pursuant to § 3.9 of this title (relating to Disposal Wells) or § 3.46 of this title (relating to Fluid Injection into Productive Reservoirs) (Rules 9 or 46), no person may dispose of any oil and gas wastes by any method without obtaining a permit to dispose of such wastes. The disposal methods prohibited by this paragraph include, but are not limited to, the unpermitted discharge of oil field brines, geothermal resource waters, other mineralized waters, or drilling fluids, into any watercourse or drainageway, including any drainage ditch, dry creek, flowing creek, river, or any other body of surface water.
- (2) Prohibited pits. No person may maintain or use any pit for storage of oil or oil products, except as authorized by paragraph (4) of this subsection, no person may maintain or use any pit for storage of oil field fluids, or for storage or disposal of oil and gas wastes, without obtaining a

permit to maintain or use the pit. A person is not required to have a permit to use a pit if a receiver has such a permit, if the person complies with the terms of such permit while using the pit, and if the person has permission of the receiver to use the pit. The pits required by this paragraph to be permitted include, but are not limited to, the following types of pits: saltwater disposal pits; emergency saltwater storage pits; collecting pits; skimming pits; brine pits; drilling fluid storage pits (other than mud circulation pits); drilling fluid disposal pits (other than reserve pits or slush pits); washout pits; and gas plant evaporation/retention pits. If, after the effective date of this subsection, a person maintains or uses a pit for storage of oil field fluids, or for storage or disposal of oil and gas wastes, and the use or maintenance of the pit is neither authorized by paragraph (4) or (7)(C) of this subsection nor permitted, then the person maintaining or using the pit shall backfill and compact the pit in the time and manner required by the director. Prior to backfilling the pit, the person maintaining or using the pit shall, in a permitted manner or in a manner authorized by paragraph (3) of this subsection, dispose of all oil and gas wastes which are in the pit.

(3) Authorized disposal methods.

(A) Fresh water condensate. A person may, without a permit, dispose of fresh water which has been condensed from natural gas and collected at gas delineating drills or gas compressor stations, provided the disposal is by a method other than disposal into surface water of the state.

(B) Inert wastes. A person may, without a permit, dispose of inert and essentially insoluble oil and gas wastes including, but not limited to, concrete, glass, wood, and wire, provided the disposal is by a method other than disposal into surface water of the state.

(C) Low chloride drilling fluid. A person may, without a permit, dispose of the following oil and gas wastes by landfarming, provided the wastes are disposed of on the same lease where they are generated, and provided the person has the written permission of the surface owner of the tract where landfarming will occur: water base drilling fluids with a chloride concentration of 3,000 milligrams per liter (mg/l) or less; drill cuttings, sands, and slits obtained while using water base drilling fluids with a chloride concentration of 3,000 milligrams per liter (mg/l) or less; and wash water used for cleaning drill pipe and other equipment at the well site.

(D) Other drilling fluid. A person may, without a permit, dispose of the following oil and gas wastes by burial, provided the wastes are disposed of at the same well site where they are generated: water base drilling fluids which had a chloride concentration in excess of 3,000 milligrams per liter (mg/l) but which have been dewatered; drill cuttings, sands, and slits obtained while using oil base drilling fluids or water base drilling fluids with a chloride concentration in excess of 3,000 milligrams per liter (mg/l); and those drilling fluids and wastes allowed to be landfarmed without a permit.

(E) Completion/workover pit wastes. A person may, without a permit, dispose of the following oil and gas wastes by burial in a completion/workover pit, provided the wastes have been dewatered, and provided the wastes are disposed of at the same well site where they are generated: spent completion fluids, workover fluids, and the materials cleaned out of the well bore of a well being completed or worked over.

(F) Effect on backfilling. A person's choice to dispose of a waste by methods authorized by this paragraph shall not extend the time allowed for backfilling any reserve pit, mud circulation pit, or completion/workover pit whose use or maintenance is authorized by paragraph (4) of this subsection.

(4) Authorized pits. A person may, without a permit, maintain or use reserve pits, mud circulation pits, completion/workover pits, basic sediment pits, flare pits, fresh makeup water pits, and water condensate pits on the following conditions:

(A) Reserve pits and mud circulation pits. A person shall not deposit or cause to be deposited into a reserve pit or mud circulation pit any oil field fluids or oil and gas wastes other than the following:

(i) drilling fluids, whether fresh water base, saltwater base, or oil base;

(ii) drill cuttings, sands, and slits separated from the circulating drilling fluids;

(iii) wash water used for cleaning drill pipe and other equipment at the well site;

(iv) drill stem test fluids; and

(v) blowout preventer test fluids.

(B) Completion/workover pits. A person shall not deposit or cause to be deposited into a completion/workover pit any oil field fluids or oil and gas wastes other than spent completion fluids, workover fluids, and the materials cleaned out of the well bore of a well being completed or worked over.

(C) Basic sediment pits. A person shall not deposit or cause to be deposited into a basic sediment pit any oil field fluids or oil and gas wastes other than basic sediment removed from a production vessel or from the bottom of an oil storage tank. Although a person may store basic sediment in a basic sediment pit, a person may not deposit oil or free saltwater in the pit. The total capacity of a basic sediment pit shall not exceed 50 barrels. The area covered by a basic sediment pit shall not exceed 250 square feet.

(D) Flare pits. A person shall not deposit or cause to be deposited into a flare pit any oil field fluids or oil and gas wastes other than the hydrocarbons designed to go to the flare during upset conditions at the well, tank battery, or gas plant where the pit is located. A person shall not store liquid hydrocarbons in a flare pit for more than 48 hours at a time.

(E) Fresh makeup water pits. A person shall not deposit or cause to be deposited into a fresh makeup water pit any oil field fluids or oil and gas wastes.

(F) Water condensate pits. A person shall not deposit or cause to be deposited into a water condensate pit any oil field fluids or oil and gas wastes other than fresh water condensed from natural gas and collected at gas separator drops or gas compressor stations.

(G) Backfill requirements.

(1) A person who maintains or uses a reserve pit, mud circulation pit, fresh makeup water pit, completion/workover pit, basic sediment pit, flare pit, or water condensate pit shall dewater, backfill, and compact the pit according to the following schedule:

(i) Reserve pits and mud circulation pits which contain fluids with a chloride concentration of 6,100 milligrams per liter (mg/l) or less and fresh makeup water pits shall be dewatered, backfilled, and compacted within one year of cessation of drilling operations.

(ii) Reserve pits and mud circulation pits which contain fluids with a chloride concentration in excess of 6,100 milligrams per liter (mg/l) shall be dewatered within 30 days and backfilled and compacted within one year of cessation of drilling operations.

(iii) All completion/workover pits used when completing a well shall be dewatered within 30 days and backfilled and compacted within 120 days of

well completion. All completion/workover pits used when working over a well shall be dewatered within 30 days and backfilled and compacted within 120 days of completion of workover operations.

(iv) Basic sediment pits, flare pits, and water condensate pits shall be dewatered, backfilled, and compacted within 120 days of final cessation of use of the pits.

(v) If a person constructs a sectioned reserve pit, each section of the pit shall be considered a separate pit for determining when a particular section should be dewatered.

(i) A person who maintains or uses a reserve pit, mud circulation pit, fresh makeup water pit, or completion/workover pit shall remain responsible for dewatering, backfilling, and compacting the pit within the time prescribed by clause (1) of this subparagraph (G), even if the time allowed for backfilling the pit extends beyond the expiration date or transfer date of the lease covering the land where the pit is located.

(ii) The director may require that a person who uses or maintains a reserve pit, mud circulation pit, fresh makeup water pit, completion/workover pit, basic sediment pit, flare pit, or water condensate pit backfill the pit sooner than the time prescribed by clause (1) of this subparagraph (G) if the director determines that oil and gas wastes are likely to escape from the pit or that the pit is being used for improper disposal of oil and gas wastes.

(iv) Prior to backfilling any reserve pit, mud circulation pit, completion/workover pit, basic sediment pit, flare pit, or water condensate pit whose use or maintenance is authorized by this paragraph (4), the person maintaining or using the pit shall, in a permitted manner or in a manner

authorized by paragraph (3) of this subsection, dispose of all oil and gas wastes which are in the pit.

(5) Responsibility for disposal.

(A) Permit required. No generator or receiver may knowingly utilize the services of a carrier to transport oil and gas wastes if the carrier is required by this rule to have a permit to transport such wastes but does not have such a permit. No carrier may knowingly utilize the services of a second carrier to transport oil and gas wastes if the second carrier is required by this rule to have a permit to transport such wastes but does not have such a permit. No generator or carrier may knowingly utilize the services of a receiver to store, handle, treat, reclaim, or dispose of oil and gas wastes if the receiver is required by statute or commission rule to have a permit to store, handle, treat, reclaim, or dispose of such wastes but does not have such a permit. No receiver may knowingly utilize the services of a second receiver to store, handle, treat, reclaim, or dispose of oil and gas wastes if the second receiver is required by statute or commission rule to have a permit to store, handle, treat, reclaim, or dispose of such wastes but does not have such a permit. Any person who plans to utilize the services of a carrier or receiver is under a duty to determine that the carrier or receiver has all permits required by the Oil and Gas Division to transport, store, handle, treat, reclaim, or dispose of oil and gas wastes.

(B) Improper disposal prohibited. No generator, carrier, receiver, or any other person may improperly dispose of oil and gas wastes or cause or allow the improper disposal of oil and gas wastes. A generator causes or allows the improper disposal of oil and gas wastes if:

(1) the generator utilizes the services of a carrier or receiver who improperly disposes of the wastes, and

(11) the generator knew or reasonably should have known that the carrier or receiver was likely to improperly dispose of the wastes and failed to take reasonable steps to prevent the improper disposal.

(6) Permits.

(A) Standards for permit issuance. A permit to maintain or use a pit for storage of oil field fluids or oil and gas wastes may only be issued if the commission determines that the maintenance or use of such pit will not result in the waste of oil, gas, or geothermal resources or the pollution of surface or subsurface waters. A permit to dispose of oil and gas wastes by any method, including disposal into a pit, may only be issued if the commission determines that the disposal will not result in the waste of oil, gas, or geothermal resources or the pollution of surface or subsurface water. A permit to maintain or use any unlined pit, other than an emergency saltwater storage pit, for storage or disposal of oil field brines, geothermal resource waters, or other mineralized waters may only be issued if the commission determines that the applicant has conclusively shown that use of the pit cannot cause pollution of surrounding productive agricultural land nor pollution of surface or subsurface water, either because there is no surface or subsurface water in the area of the pit, or because the surface or subsurface water in the area of the pit would be physically isolated by naturally occurring impervious barriers from any oil and gas wastes which might escape or migrate from the pit. Permits issued pursuant to this paragraph will contain conditions reasonably necessary to prevent the waste of oil, gas, or geothermal resources and the pollution of surface and subsurface waters. A permit to maintain or use a pit will state the conditions under which the pit may be operated, including the conditions under which the permittee shall be required to dewater, dewater, backfill, and compact the pit. Any permits issued pursuant to this paragraph may contain requirements

concerning the design and construction of pits and disposal facilities, including requirements relating to pit construction materials, dike design, liner material, liner thickness, procedures for installing liners, schedules for inspecting and/or replacing liners, overflow warning devices, leak detection devices, and fences. However, a permit to maintain or use any lined pit for storage or disposal of oil field brines, geothermal resource waters, or other mineralized waters will contain requirements relating to liner material, liner thickness, procedures for installing liners, and schedules for inspecting and/or replacing liners.

(B) Application. An application for a permit to maintain or use a pit or to dispose of oil and gas wastes shall be filed with the commission in Austin. The applicant shall mail or deliver a copy of the application to the appropriate district office on the same day the original application is mailed or delivered to the commission in Austin. A permit application shall be considered filed with the commission on the date it is received by the commission in Austin. When a commission-prescribed application form exists, an applicant shall make application on the prescribed form according to the instructions on such form. The director may require the applicant to provide the commission with engineering, geological, or other information which the director deems necessary to show that issuance of the permit will not result in the waste of oil, gas, or geothermal resources or the pollution of surface or subsurface water.

(C) Notice. The applicant shall give notice of the permit application to the surface owner of the tract upon which the pit will be located or upon which the disposal will take place. When the tract upon which the pit will be located or upon which the disposal will take place lies within the corporate limits of an incorporated city, town, or village, the applicant shall

also give notice to the city clerk or other appropriate official. Where disposal is to be by discharge into a watercourse other than the Gulf of Mexico or a bay, the applicant shall also give notice to the surface owner of each waterfront tract between the discharge point and 1/2 mile downstream of the discharge point except for those waterfront tracts within the corporate limits of an incorporated city, town, or village. When one or more waterfront tracts within 1/2 mile of the discharge point lie within the corporate limits of an incorporated city, town, or village, the applicant shall give notice to the city clerk or other appropriate official. Notice of the permit application shall consist of a copy of the application together with a statement that any protest to the application should be filed with the commission within 15 days of the date the application is filed with the commission. The applicant shall mail or deliver the required notice to the surface owners and the city clerk or other appropriate official on or before the date the application is mailed or delivered to the commission in Austin. If in connection with a particular application the director determines that another class of persons, such as offset operators, adjacent surface owners, or an appropriate river authority, should receive notice of the application, the director may require the applicant to mail or deliver notice to members of that class. If the director determines that, after diligent efforts, the applicant has been unable to ascertain the name and address of one or more persons required by this subparagraph (C) to be notified, then the director may authorize the applicant to notify such persons by publishing notice of the application. The director shall determine the form of the notice to be published. The notice shall be published once each week for two consecutive weeks by the applicant in a newspaper of general circulation in the county where the pit will be located or the disposal will take place. The applicant shall file proof of publication with the commission in Austin.

(0) Protests and hearings. If a protest from an affected person is made to the commission within 15 days of the date the application is filed, then a hearing shall be held on the application after the applicant requests a hearing. If the director has reason to believe that a person entitled to notice of an application has not received such notice within 15 days of the date an application is filed with the commission, then the director shall not take action on the application until reasonable efforts have been made to give such person notice of the application and an opportunity to file a protest to the application. If the director determines that a hearing is in the public interest, a hearing shall be held. A hearing on an application shall be held after the commission provides notice of hearing to all affected persons, or other persons or governmental entities, who express an interest in the application in writing. If no protest from an affected person is received by the commission, the director may administratively approve the application. If the director denies administrative approval, the applicant shall have a right to a hearing upon request. After hearing, the hearings examiner shall recommend a final action by the commission.

(E) Modification, suspension, and termination. A permit granted pursuant to this paragraph (E), or a renewal permit granted pursuant to paragraph (7) of this subsection, or a permit which has been issued by the commission prior to the effective date of this subsection but which does not expire pursuant to paragraph (7) of this subsection, may be modified, suspended, or terminated by the commission for good cause after notice and opportunity for hearing. A finding of any of the following facts shall constitute good cause:

(1) pollution of surface or subsurface water is occurring or is likely to occur as a result of the permitted operations;

(11) waste of oil, gas, or geothermal resources is occurring or is likely to occur as a result of the permitted operations;

(111) the permittee has violated the terms and conditions of the permit or commission rules;

(1v) the permittee misrepresented any material fact during the permit issuance process;

(v) the permittee failed to give the notice required by the commission during the permit issuance process;

(vi) a material change of conditions has occurred in the permitted operations, or the information provided in the application has changed materially.

(f) Emergency permits. If the director determines that expeditious issuance of the permit will prevent or is likely to prevent the waste of oil, gas, or geothermal resources or the pollution of surface or subsurface water, the director may issue an emergency permit. An application for an emergency permit to use or maintain a pit or to dispose of oil and gas wastes shall be filed with the commission in the appropriate district office. Notice of the application is not required. If warranted by the nature of the emergency, the director may issue an emergency permit based upon a verbal application, or the director may verbally authorize an activity before issuing a written permit authorizing that activity. An emergency permit is valid for up to 30 days, but may be modified, suspended, or terminated by the director at any time for good cause without notice and opportunity for hearing. Except when the provisions of this subparagraph (f) are to the contrary, the issuance, denial, modification, suspension, or termination of an emergency permit shall be governed by the provisions of subparagraphs (A) - (E) of this paragraph.

(6) Minor permits. If the director determines that an application is for a permit to store only a minor amount of oil field fluids or to store or dispose of only a minor amount of oil and gas waste, the director may issue a minor permit provided the permit does not authorize an activity which results in waste of oil, gas, or geothermal resources or pollution of surface or subsurface water. An application for a minor permit shall be filed with the commission in the appropriate district office. Notice of the application shall be given as required by the director. The director may determine that notice of the application is not required. A minor permit is valid for 30 days, but a minor permit which is issued without notice of the application may be modified, suspended, or terminated by the director at any time for good cause without notice and opportunity for hearing. Except when the provisions of this subparagraph (6) are to the contrary, the issuance, denial, modification, suspension, or termination of a minor permit shall be governed by the provisions of subparagraphs (A) - (E) of this paragraph.

(7) Existing permits and pits.

(A) Existing permits. Each permit to maintain or use a lined or unlined pit for storage or disposal of oil field brines, geothermal resource waters, or other mineralized waters, which has been issued by the commission prior to the effective date of this subsection (d), shall expire 180 days after the effective date of this subsection. Every other permit to store oil field fluids or oil and gas wastes or to dispose of oil and gas wastes, which permit has been issued by the commission prior to the effective date of this subsection (d), shall remain in effect until modified, suspended, or terminated by the commission pursuant to paragraph (6)(E) of this subsection. The permits which will expire pursuant to this paragraph (7) include, but are not limited to,

permits for the following types of pits: saltwater disposal pits, emergency saltwater storage pits, skimming pits, and brine pits.

(8) Renewal permits. Any person holding a permit scheduled to expire pursuant to subparagraph (A) of this paragraph may apply to the commission for renewal of the permit. If a person makes timely and sufficient application for renewal of a permit, then, notwithstanding the provisions of subparagraph (A) of this paragraph, the permit shall not expire until final commission action renewing or denying renewal of the permit. An application for renewal of a permit shall be filed with the commission in Austin within 180 days of the effective date of this subsection. No notice of the application is required. The director may administratively approve an application for renewal of a permit. No hearing shall be held on an application for renewal of a permit unless the applicant requests a hearing or the director determines that a hearing is necessary. No renewal permit will be issued unless the standards for permit issuance stated in paragraph (6)(A) of this subsection have been met.

(C) Operating existing unpermitted pits. If, as of the effective date of this subsection, a person is maintaining or using a pit, which is required by this subsection to be permitted but which was not required to be permitted prior to the effective date of this subsection, then the person maintaining or using the pit may continue to maintain or use the pit for 180 days after the effective date of this subsection. If a person makes timely and sufficient application for a permit to maintain or use such an existing but unpermitted pit, then the person may continue to use the pit until final commission action denying the permit. An application for a permit shall be considered timely if it is filed with the commission within 180 days of the effective date of this subsection. The issuance or denial of the permit shall be governed by the provisions of paragraph (6) of this subsection. The

unpermitted pits, whose use or maintenance is authorized by this subparagraph (C), include, but are not limited to, the following types of pits: drilling fluid storage pits, gas plant evaporation/retention pits, and washout pits.

(9) Backfilling existing pits. If, as of the effective date of this subsection, a person is maintaining or using a basic sediment pit which does not meet the 50 barrel size limitation of paragraph (4)(C) of this subsection, then that person shall dewater, backfill, and compact the pit or rebuild the pit to comply with the 50 barrel size limitation within 180 days of the effective date of this subsection. Any person who, as of the effective date of this subsection, is maintaining or using a lined or unlined pit for storage or disposal of oil field brines, geothermal resource waters, or other mineralized waters, which pit was permitted prior to the effective date of this subsection, shall dewater, backfill, and compact the pit within 270 days of the effective date of this subsection unless the person applies for a renewal permit pursuant to subparagraph (6) of this paragraph. If a person applies for a renewal of a permit to maintain or use a lined or unlined pit for storage or disposal of oil field brines, geothermal resource waters, or other mineralized waters, the director may extend the time for dewatering, backfilling, and compacting the pit to up to 90 days after final commission action denying renewal of the permit. If, as of the effective date of this subsection, a person is maintaining or using a pit, which is required by this subsection to be permitted but which was not required to be permitted prior to the effective date of this subsection, then the person maintaining or using the pit shall dewater, backfill, and compact the pit within 270 days of the effective date of this subsection unless the person applies for a permit to maintain or use the pit within the 180-day period allowed by subparagraph (C) of this paragraph. If a person applies for such a permit to maintain or use a previously unpermitted pit, the director may

extend the time for dewatering, backfilling, and compacting the pit to up to 90 days after final commission action denying issuance of the permit. The director may require that pits required to be backfilled by this subparagraph be dewatered, backfilled, and compacted sooner than the time prescribed by this subparagraph if the director determines that oil and gas wastes are likely to escape from the pit or that the pit is being used for improper disposal of oil and gas wastes.

(e) Pollution prevention. [reference Order Number 20-59,200, effective May 1, 1989].

(1)-(4) (No change.)
(f) Saltwater haulers.

(1)-(2) (No change.)
(g) Record keeping.

(1) Produced water. When produced water is hauled by truck from the lease where it is produced to an off-lease disposal facility, the person producing the water shall keep, for a period of two years from the date of water production, the following records:

(A) identity of the property from which the produced water is hauled;

(B) identity of the commission-approved disposal facility to which the produced water is delivered;

(C) name, address, and permit number (MP No.) of saltwater hauler transporting the water from producing lease to disposal facility; and

(D) volume of produced water transported each day from producing lease to disposal facility by saltwater hauler.

(2) Retention of run tickets. A person may comply with the requirements of paragraph (1) of this subsection by retaining run tickets or

other billing information created by the saltwater hauler, provided the run tickets or other billing information contain all the information required by paragraph (1).

(3) Examination and reporting. The person keeping any records required by this subsection (3) shall make the records available for examination and copying by members and employees of the commission during reasonable working hours. Upon request of the commission, the person keeping the records shall file such records with the commission.

(4) Penalties. Violations of this section may subject a person to penalties and remedies specified in Title 3 of the Texas Natural Resources Code and any other statutes administered by the commission. The certificate of compliance for any oil, gas, or geothermal resource will may be revoked in the manner provided in § 3.68 of this title (relating to Pipeline Connection and Severance) (Rule 73) for violation of this section.

Instructions to Pit Application
Authority: Statewide Rule 8, Water Protection

- A. File the application, including all attachments, with the Railroad Commission, Oil and Gas Division, P.O. Drawer 12967, Capitol Station, Austin, Texas 78711. On the same day file one copy of the application and its attachments with the appropriate District Office. This form is not required for a minor permit.
- B. Notify the surface owner of the land where the pit will be located by mailing or delivering a copy of the application form, both front and back, but excluding the attachments. If the land where the pit is proposed is within corporate limits, also notify the city clerk or other appropriate city official. If application is for renewal of an existing permit, notice is not required.
- C. Attach a plat showing the size of the lease or tract and the location of the pit within the lease or tract. Give approximate perpendicular distance to nearest intersecting lease/unit lines and section/survey lines. To avoid confusion, distinguish between the two sets of lines. Indicate scale on this plat.
- D. Attach a county highway map (scale: 1" = 4 miles) showing the location of the pit. County highway maps are available from the Texas Department of Highways and Public Transportation, P. O. Box 5051, Attn: Map Distribution File D-10, Austin, TX 78763.
- E. If application is for renewal of a permit for an existing pit, attach a copy of your current authority to use the pit.
- F. Identify the type of pit in Item 13 using one of the following as defined in Statewide Rule 8(a): Emergency Saltwater Storage Pit, Collecting Pit, Gas Plant Evaporation/Retention Pit, Brine Pit (located at underground hydrocarbon storage facilities only), Saltwater Disposal Pit, Skimming Pit, Washout Pit, Drilling Fluid Disposal Pit, Drilling Fluid Storage Pit, or other (specify in item 13 and explain in item 15a).
- G. Attach a drawing of two perpendicular, sectional views of the pit showing the pit bottom, sides, dikes and the natural grade. For an existing pit, dimensions below fluid level may be approximated. If the pit length and width are irregular, include a top view to show pit dimensions and dike widths. Indicate scale on all views.
- H. If pit is lined, attach data on liner material, thickness, and installation procedures.
- I. Attach an identification and description of the soil or subsoil that will make up the pit bottom and sides. The information shall describe the soil by typical name, appropriate proportion of grain sizes, texture, consistency, moisture condition, and other pertinent characteristics. (Example: clayey silt, slightly plastic, small percentage of fine sand, firm and dry in place.) Identify the source of soil information. Information on how to classify soils is available from the District Office or Austin Office upon request. If application is for renewal of a permit for an existing emergency saltwater storage pit or a lined pit with a leak detection system, this attachment is not required.
- J. If pit is equipped with a leak detection system, attach engineering design drawing of the pit and leak detection system.
- K. If lined pit is not equipped with a leak detection system, describe procedures for periodic maintenance and determining liner integrity, including any special monitoring.
- L. If pit is an emergency salt water storage pit, attach justification for pit size based on water production, lease water storage capacity, and anticipated well or equipment shut-down time.

Note: The Director of the Oil and Gas Division may require the applicant to provide the Commission with any additional engineering, geological, or other information which the Director deems necessary to show that issuance of the permit will not result in the waste of oil, gas, or geothermal resources or the pollution of surface or subsurface water.

Protests and hearings.

An affected person may file a protest to the application and request a hearing. Any protest to the application should be filed with the Commission in Austin within fifteen days of the date the application is filed with the Commission. Any such protest shall be made in writing and shall include (1) the name, mailing address, and phone number of the person making the protest; and (2) a brief description of how the protestant would be adversely affected by the granting of the permit. If the Commission determines that a valid protest has been received, or that a hearing would be in the public interest, a hearing will be held after the issuance of proper and timely notice of the hearing by the Commission. If no protest is received within fifteen (15) days of receipt of the application in Austin, the application may be processed administratively.