

RECEIVED NOV 27 1979

DEC 12 1979

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ARTESIA, OFFICE

LABORATORY NO. 1179190
SAMPLE RECEIVED 11-19-79
RESULTS REPORTED 11-23-79

NO. 1 Recovered water - taken from Federal #1. 11-12-79

NO. 2

NO. 3

NO. 4

REMARKS:

CHEMICAL AND PHYSICAL PROPERTIES				
	NO. 1	NO. 2	NO. 3	NO. 4
Specific Gravity at 60° F.	1.2160			
pH When Sampled				
pH When Received	7.3			
Bicarbonate as HCO ₃	2,245			
Supersaturation as CaCO ₃				
Undersaturation as CaCO ₃				
Total Hardness as CaCO ₃	4,300			
Calcium as Ca	488			
Magnesium as Mg	748			
Sodium and/or Potassium	135,190			
Sulfate as SO ₄	17,280			
Chloride as Cl	197,433			
Iron as Fe	3.7			
Barium as Ba				
Turbidity, Electric				
Color as Pt				
Total Solids, Calculated	353,384			
Temperature °F.				
Carbon Dioxide, Calculated				
Dissolved Oxygen, Winkler				
Hydrogen Sulfide	1,750			
Resistivity, ohms/m at 77° F.	0.041			
Suspended Oil				
Filtrable Solids as mg/l				
Volume Filtered, ml				
*Lithium, as Li	285			

Results Reported As Milligrams Per Liter

Additional Determinations And Remarks: Determined by Atomic Absorption Spectrophotometry.

We are not familiar with the circumstances involved herein, but this shows an unusually high sulfate for a heavy brine. It should be further brought out that this water would be significantly incompatible with spent acid coming off limestone or dolomite (it would cause rapid gyping with calcium sulfate).

By

Waylan C. Martin, M. A.