(JUNE 1990) BUREAU SUNDRY NOTION	Image: State Stat	FORM APPROVED Budget Bureau No. 1004-0135 Expires: March 31, 1993 5. Lease Designation and Serial No. NM-17103 6. If Indian, Allottee or Tribe Name	
Use "APPLICATIO			
SUL	7. If Unit or CA, Agreement Designation		
1. Type of Well Oil Gas X Well Well Other		8. Well Name and No.	
2. Name of Operator	Lee Federal #7		
3. Address and Telephone No. P.O. Box <sup>-</sup> Roswell, N A leasting of Well (Feature Sec. T. B. Mars Sec.	9. API Well No. 30-015-28146 10. Field and Pool, or Exploratory Area Scoplan Delevers		
4. Location of Well (Footage, Sec., T., R., M., or Sur 1980'FN Section 2	Scanlon Delaware 11. County or Parish, State		
·····		Eddy County, New Mexico	
	(s) TO INDICATE NATURE OF NOTICE, REPOI	RI, OR OTHER DATA	
TYPE OF SUBMISSION	TYPE OF SUBMISSION TYPE OF ACTION		
<ul> <li>Notice of Intent</li> <li>X Subsequent Report</li> <li>Final Abandonment Notice</li> </ul>	Abandonment         Recompletion         Plugging Back         Casing Repair         Altering Casing         OTHER	Change of Plans New Construction Non – Routine Fracturing Water Shut – Off Conversion to Injection X Dispose Water (Note: Report results of multiple completion on Well Completion or Recompletion Report and Log form.)	
	all pertinent details, and give pertinent dates, including estimated date of star as for all markers and zones pertinent to this work.)* Pents of NonCompliance dated May 22,1996 reg eview and approval:	ing any proposed work. If well is directionally drilled,	
<ol> <li>Production is 1</li> <li>Produces an a</li> <li>Water analysis</li> <li>Produced wate</li> </ol>	rom the Delaware formation. verage of four (4) barrels of water per day.		

14. I hereby certify that the foregoing is true and correct Signed <u>arol</u> J. Parcia	Title	Production Records Manager	Date_	7/9/96
(This space for Federal or State office use) Approved by (ORIG. SGD.) DAVID & CIANG Conditions of approval, if any: SEE ATTACHED FOR Title 18 US.C. CONTRACT OF A PERSON AND AND A WORK OF A PERSON AND AND A WORK OF A PERSON ADDRESS AND A PERSON ADDRESS AND A WORK OF A PERSON ADDRESS AND A PERSON ADDRESS AND A WORK OF A PERSON ADDRESS AND A PERSON ADDRE	Title villfully to	PETROLEUM ENGINEER	Date_	SEP 2.4 1995
*	Soo I	activition on Devence Cide		

Enviro-Ch	em, Inc.						
WATER ANALYSIS REPORT Sample							
Cil Co. : Strata Oil & Gas San Lease : Lee Federal Dat	mple Loc. : te Analyzed: 12-August-1996 te Sampled :						
ANALYSIS							
1. pH 7.100 2. Specific Gravity 60/60 F. 1.103 3. CaCO3 Saturation Index @ 80 F. +1.428 @ 140 F. +2.328							
Dissolved Gasses	MG/L EQ. WT. *MEQ/L						
4. Hydrogen Sulfide 5. Carbon Dioxide Not De 6. Dissolved Cxygen Not De	180 termined termined						
Cations							
7. Calcium (Ca++) 3. Magnesium (Mg++) 9. Sodium (Na+) (Calculated) 10. Barium (Ba++) Not De	7,415 / 20.1 = 368.91 2,006 / 12.2 = 164.43 45,855 / 23.0 = 1,993.70 termined						
Anions							
11. Hydroxyl (CH <sup>-</sup> ) 12. Carbonate (CO <sub>3</sub> =) 13. Bicarbonate (HCO <sub>3</sub> <sup>-</sup> ) 14. Sulfate (SO <sub>4</sub> =) 15. Chloride (Cl <sup>-</sup> )	0 / 17.0 = 0.00 0 / 30.0 = 0.00 771 / 61.1 = 12.62 1,700 / 43.8 = 34.84 87.980 / 35.5 = 2,478.31						
16. Total Dissolved Solids 17. Total Iron (Fe) 18. Total Hardness As CaCO3 19. Resistivity @ 75 F. (Calculated) @	145,727 2 / 18.2 = <b>0.11</b> 25,774 0.042 /cm.						
LOGARITHMIC WATER PATTERN PROBABLE MINERAL COMPOSITION *meq/L. COMPOUND EQ. WT. X *meq/L = mg/L.							
Na	Ca(HCC <sub>3</sub> ) <sub>2</sub> 81.04 12.62 1,023						
Ca	3 CaSO4 68.07 34.84 2,371						
	CaCl <sub>2</sub> 55.50 321.45 17,841						
Fe here have see for the think think the CO3 12000 1000 100 100 10 1 10 100 1000 100	Mg(HCO <sub>3</sub> ) <sub>2</sub> 73.17 0.00 0						
<u>Calcium Sulfate Solubility Profile</u>	MgSO4 60.19 <b>0.00</b> 0						
	MgCL <sub>2</sub> 47.62 164.43 7,830						
	NaHCO3 84.00 0.00 C						
	NaSO4 71.23 0.00 0						
	NaCI 58.46 1,992.43 116,478						
This water is mildly corrosive due to the pH observed or apalysis							

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This water is mildly corrosive due to the pH observed on analysis. The corrosivity is increased by the content of mineral salts, and the presence of H2S in solution.