

NEW MEXICO OIL CONSERVATION COMMISSION
Santa Fe, New Mexico

(Form C-104)
Revised 7/1/57

REQUEST FOR (OIL) - ~~(GAS)~~ ALLOWABLE

New Well
~~Recompletion~~

This form shall be submitted by the operator before an initial allowable will be assigned to any completed Oil or Gas well. Form C-104 is to be submitted in QUADRUPLICATE to the same District Office to which Form C-101 was sent. The allowable will be assigned effective 7:00 A.M. on date of completion or recompletion, provided this form is filed during calendar month of completion or recompletion. The completion date shall be that date in the case of an oil well when new oil is delivered into the stock tanks. Gas must be reported on 15.025 psia at 60° Fahrenheit.

Farmington, New Mexico March 31, 1958

(Place)

(Date)

WE ARE HEREBY REQUESTING AN ALLOWABLE FOR A WELL KNOWN AS:

El Paso Natural Gas Products Company Malco-Copple

Well No. 6, in SW 1/4 NE 1/4,

(Company or Operator)

G

3

T. 30-N

(Lease)
15-W

NMPM,

Verde-Gallup

Pool

Unit Letter

San Juan

County. Date Spudded 1-22-58

Date Drilling Completed 3-6-58

Please indicate location:

Elevation 3415'

Total Depth 3462' ~~XXXX~~ CORD 3450'

Top Oil/Gas Pay 3341'

Name of Prod. Form. Gallup

PRODUCING INTERVAL -

Perforations 3341' - 3450'

Open Hole - -

Depth 3450' Depth Tubing 3445'

OIL WELL TEST -

Natural Prod. Test: _____ bbls. oil, _____ bbls water in _____ hrs, _____ min. Size _____

Test After Acid or Fracture Treatment (after recovery of volume of oil equal to volume of load oil used): 105 bbls. oil, _____ bbls water in 24 hrs, _____ min. Size 2"

GAS WELL TEST -

Pumping 14 SPM - 40" Stroke.

Natural Prod. Test: _____ MCF/Day; Hours flowed _____ Choke Size _____

Method of Testing (pitot, back pressure, etc.): _____

Test After Acid or Fracture Treatment: _____ MCF/Day; Hours flowed _____

Choke Size _____ Method of Testing: _____

Acid or Fracture Treatment (Give amounts of materials used, such as acid, water, oil, and sand): 13,396 gals. oil & 15,000# sand, Flush w/3,440 gals. oil - 200 gals.

Casing Tubing Date first new March 27, 1958 mud acid ahead
Press. Press. oil run to tanks

Oil Transporter El Paso Natural Gas Products Co. by Truck of feet:

Gas Transporter None

Remarks:

I hereby certify that the information given above is true and complete to the best of my knowledge.

Approved: MAR 31 1958, 19____

El Paso Natural Gas Products Company

(Company or Operator)

By:

Petroleum Engineer

Title

Send Communications regarding well to:

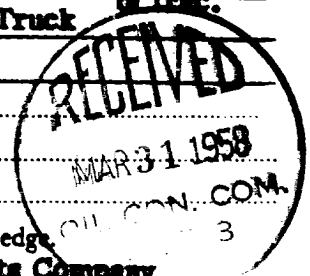
Name: Ewell N. Walsh

Address: Box 1565, Farmington, New Mexico

OIL CONSERVATION COMMISSION

By: Original Signed Emery C. Arnold

Title: Supervisor Dist. # 3



1. Name of the person or organization: [REDACTED]
2. Address: [REDACTED]
3. City: [REDACTED]
4. State: [REDACTED]
5. Zip: [REDACTED]
6. Phone: [REDACTED]
7. Fax: [REDACTED]
8. E-mail: [REDACTED]
9. Other: [REDACTED]

1. $\text{C}_2\text{H}_5\text{Br}$ (1.00 mol) + NaOH (1.00 mol) \rightarrow $\text{C}_2\text{H}_5\text{OH}$ (1.00 mol) + NaBr (1.00 mol)
 2. $\text{C}_2\text{H}_5\text{OH}$ (1.00 mol) + H_2SO_4 (1.00 mol) \rightarrow $\text{C}_2\text{H}_5\text{H}_2\text{O}$ (1.00 mol) + H_2O (1.00 mol)
 3. $\text{C}_2\text{H}_5\text{H}_2\text{O}$ (1.00 mol) + H_2SO_4 (1.00 mol) \rightarrow $\text{C}_2\text{H}_5\text{H}_2\text{O}$ (1.00 mol) + H_2O (1.00 mol)
 4. $\text{C}_2\text{H}_5\text{H}_2\text{O}$ (1.00 mol) + H_2SO_4 (1.00 mol) \rightarrow $\text{C}_2\text{H}_5\text{H}_2\text{O}$ (1.00 mol) + H_2O (1.00 mol)
 5. $\text{C}_2\text{H}_5\text{H}_2\text{O}$ (1.00 mol) + H_2SO_4 (1.00 mol) \rightarrow $\text{C}_2\text{H}_5\text{H}_2\text{O}$ (1.00 mol) + H_2O (1.00 mol)
 6. $\text{C}_2\text{H}_5\text{H}_2\text{O}$ (1.00 mol) + H_2SO_4 (1.00 mol) \rightarrow $\text{C}_2\text{H}_5\text{H}_2\text{O}$ (1.00 mol) + H_2O (1.00 mol)
 7. $\text{C}_2\text{H}_5\text{H}_2\text{O}$ (1.00 mol) + H_2SO_4 (1.00 mol) \rightarrow $\text{C}_2\text{H}_5\text{H}_2\text{O}$ (1.00 mol) + H_2O (1.00 mol)
 8. $\text{C}_2\text{H}_5\text{H}_2\text{O}$ (1.00 mol) + H_2SO_4 (1.00 mol) \rightarrow $\text{C}_2\text{H}_5\text{H}_2\text{O}$ (1.00 mol) + H_2O (1.00 mol)
 9. $\text{C}_2\text{H}_5\text{H}_2\text{O}$ (1.00 mol) + H_2SO_4 (1.00 mol) \rightarrow $\text{C}_2\text{H}_5\text{H}_2\text{O}$ (1.00 mol) + H_2O (1.00 mol)
 10. $\text{C}_2\text{H}_5\text{H}_2\text{O}$ (1.00 mol) + H_2SO_4 (1.00 mol) \rightarrow $\text{C}_2\text{H}_5\text{H}_2\text{O}$ (1.00 mol) + H_2O (1.00 mol)

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