EXHIBIT 2

Section C. Downhole Commingling item 1. (a)(4)

The fluids from each zone are compatible with the fluids from the other, and combining the fluids will not result in the formation of precipitates which might damage either (any) reservoir.

Attached are copies of current water analysis from the Belcher 'A' Well No. 1 and the Belcher Well No. 1. The Belcher 'A' well has only the Blinebry and Tubb zones open, while the Belcher Well has all three zones open at this time (Blinebry, Tubb, and Drinkard). Also attached is a water analysis on the Belcher well prior to commingling of all three zones. Examination of each set of analysis shows no appreciable difference in the compositions of the produced water from each well or time period (before or after commingling).

EEFORE DYNAMINATION OF THE ONE OF THE PROPERTY HO. 2

CASE NO. 857/

P. O. BOX 1468 MONAHANS, TEXAS 79756 PH. 943-3234 OR 563-1040

709 W. INDIANA MIDLAND. TEXAS 79701 PHONE 683-4521

RESULT OF WATER ANALYSES

	L	ABORATORY NO385481					
To: Mr. Mark Mladenka		MPLE RECEIVED 3-26-85					
901 Petroleum Building, Midland,	rexas B	SULTS REPORTED	4-3-8	5			
RESULTS REPORTED							
COMPANY Mabee Petroleum Corporation LEASE Belcher							
FIELD OR POOL T-22S & COUNTY Lea STATE NM							
FIELD OR POOL SECTION 7 BLOCK SURVEY T-22S & COUNTY Lea STATE NM SOURCE OF SAMPLE AND DATE TAKEN: R-38E							
SOURCE OF SAMPLE AND DATE TAKEN:							
NO. 1 Produced (Blinebry, Tubb, & Drinkard) water - taken from Belcher #1. 3-26-85							
NO. 2 Produced (Blinebry & Tubb) water - taken from Belcher "A" #1. 3-26-85							
NO. 3							
NO. 4							
REMARKS: Samples taken by G	reg Ogden, Mart	in Water Labo	ratories, In	c			
	L AND PHYSICAL F						
	NO. 1	NO. 2	NO. 3	NO. 4			
Specific Gravity at 60° F.	1.1097	1.1122					
pH When Sampled							
pH When Received	6.54	7.01					
Bicarbonate as HCO3	519	470					
Supersaturation as CaCO3	10	50					
Undersaturation as CaCO3							
Total Hardness as CaCO ₃							
Calcium as Ca	31,250 8,100	29,000 6,700	· · · · · · · · · · · · · · · · · · ·				
Magnesium as Mg	2,673	2,977					
Sodium and/or Potassium	50,547	53,768					
Sulfate as SO4	1,460	2,217					
Chloride as CI	98,716	101,557					
Iron as Fe	5.5	31.8					
Barium as Ba							
Turbidity, Electric		ļ					
Color as Pt		<u> </u>					
Total Solids, Calculated	162,015	167,689					
Temperature °F.		 					
Carbon Dioxide, Calculated		ļ					
Dissolved Oxygen, Winkler		 					
Hydrogen Sulf-de	0.0	0.0					
Resistivity, ohms/m at 77° F.	0.066	0.065					
Suspended Oil							
Filtrable Solids as mg/1							
Volume Filtered, ml							
			· · · · · · · · · · · · · · · · · · ·				
			 _				
Results Reported As Milligrams Per Liter							
	Additional Determinations And Remarks A study of the above reveals no clear-cut evidence of any						
incompatibility between these waters. We do find the water from Belcher "A" #1 to							
be just at the saturation point to calcium sulfate and also the combination of the							
waters are just at the saturation point. However, we do not find evidence to sug-							
gest that they are supersaturated and therefore do not consider any concern in mix-							
ing them. Therefore, in general, we classify these two waters as compatible. It should be added that we do not have the ability to effectively evaluate compatibil-							
ity between oils. However, we would not expect any problem in this regard.							

Form No. 3

Way Yan C. Martin, M.A.

RESULT OF WATER ANALYSES

		BORATORY NO	434187			
ro: Mr. Mark Mladenka			/ 17 0/			
ro: Mr. Mark Miadenka 901 Petroleum Building, Midland, Texas		MPLE RECEIVED				
JUL TOUTOLGUM DOLLGLING TAGETHE	Te sao Re	SUL IS REPURIEUL	<u>,</u>			
COMPANY Mabee Petroleum Corporation	n = 4.55	Reicher				
	LEASE _	2,02,02,04				
FIELD OR POOL	1		NM			
SECTION / BLOCK SURVEY 1-225 G	_ COUNTY	stasta	ATE			
SOURCE OF SAMPLE AND DATE TAKEN -38E	· · · · · · · · · · · · · · · · · · ·					
No. 1 Produced fluid - taken from Belcher #1. 4-17-84						
NO. 2 Produced fluid - taken from Belcher "A" #1. 4-17-84						
NO. 3						
			· · · · · · · · · · · · · · · · · · ·			
REMARKS: 1. Tubb & Dri	nkard 2.	Tubb & Bline	bry			
OREMIOAL A	ND PHYSICAL F	NO. 2				
Specific Gravity at 60° F.	1.1014	1.0990	NO. 3	NO. 4		
pH When Sampled	T*T0T4	1.0750				
-,	6 6 /2	7.47				
pH When Received	6.64					
Bicarbonate as HCO ₃	1,354	409				
Supersaturation as CaCO3	150	65				
Undersaturation as CaCO3						
Total Hardness as CaCO ₃	44,500	29,250				
Calcium as Ca	12,600	7,100				
Magnesium as Mg	3,159	2,795				
Sodium and/or Potassium	43,193	47,212				
Sulfate as SO ₄	1.058	1,318				
Chloride as Cl	96.586	92,325				
Iron as Fe	71.7	24.1				
Barium as Ba						
Turbidity, Electric						
Color as Pt						
Total Solids, Calculated	157,949	151,158				
Temperature °F.						
Carbon Dioxide, Calculated						
Dissolved Oxygen, Winkler						
Hydrogen Sulfide	0.0	0.0				
Resistivity, ohms/m at 77° F.	0.066	0.070				
Suspended Oil						
Filtrable Solids as mg/						
Volume Filtered, ml						
Calcium Carbonate Scaling Tendency	NONE	MILD				
Calcium Sulfate Scaling Tendency	NONE	NONE				
Oil Gravity, OAPI	40.2	35.5				
	eported As Milligram					
		of these two	waters reve	als no		
evidence of any incompatibility between these two waters. This is to say that the combination of these waters should not result in any precipitation or scaling						
potential in the resulting combination.						
potential in the resulting complianton.						
						
						

Form No. 3

Waylan C. Martin, M.A.