3R - <u>249</u>

GENERAL CORRESPONDENCE

YEAR(S): 1992-1982



OIL CONSERSE ON DIVISION

RECTOR CED P. O. BOX 1492 EL PASO, TEXAS 79978 PHONE: 915-541-3071

192 FEH 2H AM 9 20

JAMYE BOONE WARD ATTORNEY AT LAW

February 21, 1992

Ms. Kathy M. Brown Geologist Oil Conservation Division P.O. Box 2088 Santa Fe, New Mexico 87504

> Re: Contamination at Manana - Mary Wheeler No. 1E Gas Well Site, Flora Vista, New Mexico

Dear Ms. Brown:

Last October, El Paso Natural Gas Company (El Paso) completed nearly every aspect of the remediation of the contaminated ground water at the site named above. In your November 15, 1991, letter you told El Paso that no further remediation of the site was necessary at this time. You also advised us that the OCD has no jurisdiction over the operation of the Water Well Sl, which, as part of the pre-approved remediation plan, El Paso was to plug and abandon.

Your recommendation with respect to plugging Well Sl was to "contact both the State Engineer and the Flora Vista Water Users Association (FVWUA) prior to any activities involving Well Sl." El Paso did as you The State Engineer advised Anu Pundari, recommended. El Paso's environmental compliance engineer in Farmington, that there are no specific regulations on shallow well abandonment procedures. The State Engineer advised that the FVWUA should complete the bottom portion of the well record form regarding the plugging method. El Paso's preferred plugging method is to weld a steel cap on Well S1.

I contacted the FVWUA, through their attorney, seeking their approval of El Paso's effort to cap Well Sl. Mr. Chester Miller, the FVWUA's attorney, informed me yesterday that the FVWUA will not allow El Paso to cap Well Sl until El Paso agrees to several other conditions which are totally unrelated to capping the well. Ms. Kathy M. Brown February 21, 1992 Page 2

By way of this letter, I am informing OCD that El Paso, at this time, is not able to plug and abandon Water Well S1 as your agency directed.

Very truly yours,

Janue Boone Ward

c: Chester Miller III, P.C. 304 South Lake Street Farmington, New Mexico 87401



P. O. BOX 1492 EL PASO, TEXAS 79978 PHONE: 915-541-3071

JAMYE BOONE WARD ATTORNEY AT LAW

February 21, 1992

F. Chester Miller III, Esq. 304 South Lake Street Farmington, New Mexico 87401

Re: Contamination at Manana-Mary Wheeler #1E Gas Well Site, Flora Vista, New Mexico

Dear Mr. Miller:

El Paso Natural Gas Company ("El Paso") believes that it has made every effort to accommodate the Flora Vista Water Users Association ("FVWUA"). As part of the pre-approved plan to remediate the ground water contamination at the site named above, El Paso was directed to plug and abandon Water Well Sl. Please refer to the enclosed copy of the letter to David Boyer, dated March 19, 1991. In the paragraph numbered 22 in Exhibit A to the letter, El Paso agrees to plug and abandon Water Well Sl.

Then, at the direction of the New Mexico Oil Conservation Division ("OCD"), El Paso contacted the FVWUA (through you) and the State Engineer's office seeking approval of its efforts to plug Well S1 by welding a metal cap on the well opening. El Paso and the OCD have repeatedly informed the FVWUA that the safe completion of the ground water contamination cleanup requires plugging and abandoning Well S1. Unfortunately, the FVWUA refuses to allow El Paso to plug and abandon Well Sl.

Notwithstanding the fact that Well SI should be plugged and abandoned as a step in the complete remediation of the ground water, El Paso is very concerned that the open and accessible condition of Well SI raises some serious safety issues. The well may be used for unauthorized disposal or dumping as it now exists. El Paso will not accept any responsibility for the safety or protection of anyone receiving water service from Well S1. El Paso will not be liable for any injury, damage, death, or further contamination arising from the use of Well S1.

If the FVWUA reconsiders its present position and decides to allow El Paso to cap Well Sl, please contact

F. Chester Miller III, Esq. February 21, 1992 Page 2

me immediately. Thank you for conveying this message to your client.

Very truly yours, Januge Boone Ward

Enclosures

c (w/o enclosures):

Kathy M. Brown, OCD ~

STATE OF NEW MEXICO

ENERGY, MINERALS AND NATURAL RESOURCES DEPARTMENT

OIL CONSERVATION DIVISION



BRUCE KING GOVERNOR POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87504 (505) 827-5800

November 15, 1991

Mr. Thomas D. Hutchins North Region Compliance Manager El Paso Natural Gas Company P.O. Box 1492 El Paso, Texas 79978

RE: Remediation at Manana-Mary Wheeler #1E Gas Well Site, Flora Vista, New Mexico

Dear Mr. Hutchins:

The Oil Conservation Division (OCD) has received your October 11, 1991 summary of remediation activities at the referenced site. In this document El Paso Natural Gas Company (EPNG) has requested permission from the OCD to plug and abandon Well S1. Because Well S1 is a public water supply well, the OCD has no jurisdiction over the operations of this well. The OCD recommends that EPNG contact both the State Engineer and the Flora Vista Water Users Association prior to any activities involving Well S1.

Based on the data submitted to the OCD by EPNG, the OCD requires no further remediation at this time at the Manana-Mary Wheeler #1E Gas Well Site, Flora Vista, New Mexico.

If you have any questions, please contact me at (505) 827-5884.

Sincerely,

n. Drown

Kathy M. Brown Geologist

xc: OCD Aztec District Office



9830 S. 51st Street Suite B-113 Phoenix, AZ 85044 (602) 496-4400

ATI I.D. 105554

May 16, 1991

El Paso Natural Gas Company P.O. Box 4990 Farmington, NM 87499

Project Name/Number: Mary Wheeler

Attention: John Lambdin

On 05/03/91, Analytical Technologies, Inc. received a request to analyze soil sample(s). The sample(s) were analyzed with EPA methodology or equivalent methods. The results of these analyses and the quality control data, which follow each set of analyses, are enclosed.

If you have any questions or comments, please do not hesitate to contact us at (602) 496-4400.

jabeth,

Elizabeth Proffitt Senior Project Manager

Cohert V. Wood

Robert V. Woods Laboratory Manager

RVW:clf Enclosure

Corporate Offices: 5550 Morehouse Drive San Diego, CA 92121 (619) 458-9141

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 ATI #	CLIENT DESCRIPTION	MATRIX DATE COLLECTED
	: MARY WHEELER ATI I.D. : 105554	REPORT DATE : 05/09/91
CLIENT PROJECT #	: EL PASO NATURAL GAS, NEW MEXICO : (NONE)	D DATE RECEIVED : 05/03/93

01	N10615	SOIL	05/02/91
02	N10616	SOIL	05/02/91
03	N10617	SOIL	05/02/91
04	N10618	SOIL	05/02/91



---- TOTALS -----

MATRIX	# SAMPLES
SOIL	4

S 56 77 6

ATI STANDARD DISPOSAL PRACTICE

The samples from this project will be disposed of in thirty (30) days from th date of this report. If an extended storage period is required, please contac our sample control department before the scheduled disposal date.



GENERAL CHEMISTRY RESULTS

	ATI I.D. : 105554 —
CLIENT : EL PASO NATURAL GAS, NEW MEXICO PROJECT # : (NONE)	DATE RECEIVED : 05/03/91
PROJECT NAME : MARY WHEELER	REPORT DATE : 05/09/91
PARAMETER UNITS 01 02	03 04
PETROLEUM HYDROCARBONS, IR MG/KG <20 40	74 <20



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ATI I.D. : 10555402

TEST : FUEL HYDROCARBONS (MODIFIED EPA METHOD 8015)

CLIENT : EL PASO NATURAL GAS, NEW MEXICO PROJECT # : (NONE) PROJECT NAME : MARY WHEELER CLIENT I.D. : N10616 SAMPLE MATRIX : SOIL	DATE SAMPLED : 05/02/91 DATE RECEIVED : 05/03/91 DATE EXTRACTED : 05/03/91 DATE ANALYZED : 05/06/91 UNITS : MG/KG DILUTION FACTOR : 1
COMPOUNDS	RESULTS
FUEL HYDROCARBONS HYDROCARBON RANGE HYDROCARBONS QUANTITATED USING SURROGATE PERCENT RECOVERIES	<5 - -

DI-N-OCTYL-PHTHALATE (%)

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109





ATI I.D. : 10555401

TEST : FUEL HYDROCARBONS (MODIFIED EPA METHOD 8015)

CLIENT PROJECT # PROJECT NAME CLIENT I.D. SAMPLE MATRIX		NEW MEXICO	DATE SAMPLED DATE RECEIVED DATE EXTRACTED DATE ANALYZED UNITS DILUTION FACTOR	: 05/02/91 : 05/03/91 : 05/03/91 : 05/06/91 : MG/KG : 1
COMPOUNDS		R	ESULTS	
FUEL HYDROCAR	BONS	<	5	
HYDROCARBON R. HYDROCARBONS	ANGE DUANTITATED USING	-		
	GATE PERCENT RECOVERIES			

DI-N-OCTYL-PHTHALATE (%)

116

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and the



ATI I.D. : 10555403

TEST : FUEL HYDROCARBONS (MODIFIED EPA METHOD 8015)

CLIENT PROJECT # PROJECT NAME CLIENT I.D. SAMPLE MATRIX	: MARY WHEELER : N10617	NEW MEXICO DATE SAMPLED DATE RECEIVED DATE EXTRACTE DATE ANALYZED UNITS DILUTION FACT	<pre>b : 05/03/91 cD : 05/03/91 c) : 05/06/91 c) : MG/KG</pre>
COMPOUNDS		RESULTS	
FUEL HYDROCAR HYDROCARBON R HYDROCARBONS		<5 -	
	GATE PERCENT RECOVERIES	_	

DI-N-OCTYL-PHTHALATE (%)





ATI I.D. : 10555404

TEST : FUEL HYDROCARBONS (MODIFIED EPA_METHOD 8015)

CLIENT PROJECT # PROJECT NAME CLIENT I.D. SAMPLE MATRIX	ICO DATE SAMPLED : 05/02/91 DATE RECEIVED : 05/03/91 DATE EXTRACTED : 05/03/91 DATE ANALYZED : 05/06/91 UNITS : MG/KG DILUTION FACTOR : 1
COMPOUNDS	RESULTS
FUEL HYDROCAR HYDROCARBON R HYDROCARBONS	 <5 - -

SURROGATE PERCENT RECOVERIES

100

DI-N-OCTYL-PHTHALATE (%)

141

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GENERAL CHEMISTRY - QUALITY CONTROL

CLIENT : E PROJECT # : (PROJECT NAME : N		GAS, NEW	MEXICO	ATI	I.D.	: 10555	54	
PARAMETER	UNITS	ATI I.D.	SAMPLE RESULT			SPIKED SAMPLE		۶ REC
PETROLEUM HYDROC	CARBONS MG/KG	10555402	40	48	18	230	190	100

Acceptule!

 COPV

% Recovery = (Spike Sample Result - Sample Result) ______ X 100 Spike Concentration RPD (Relative Percent Difference) = (Sample Result - Duplicate Result) ______ X 10 Average Result

* 1.00



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GAS CHROMATOGRAPHY - RESULTS

REAGENT BLANK

TEST : FUEL HYDROCARBONS (MODIFIED EPA METHOD 801)	,
CLIENT : EL PASO NATURAL GAS, NEW MEXICO PROJECT # : (NONE) PROJECT NAME : MARY WHEELER CLIENT I.D. : REAGENT BLANK	ATI I.D. : 105554 DATE EXTRACTED : 05/03/91 DATE ANALYZED : 05/03/91 UNITS : MG/KG DILUTION FACTOR : N/A
COMPOUNDS	ESULTS
FUEL HYDROCARBONS < HYDROCARBON RANGE - HYDROCARBONS QUANTITATED USING -	5

SURROGATE PERCENT RECOVERIES

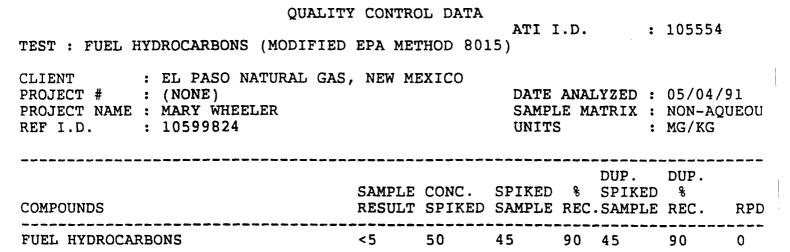
DI-N-OCTYL-PHTHALATE (%)

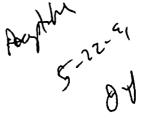
100



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Analytical Technologia





% Recovery = (Spike Sample Result - Sample Result) Spike Concentration RPD (Relative % Difference) = (Spiked Sample - Duplicate Spike) Result Sample Result

Average of Spiked Sample

MANANA MARY WHEELER #1E

MW.WK3 10-08-91

					Analytica	l Data Sumr	mary - Total F	Petroleum Hy	arbon			John Lambdi	in
								FIELD	-	Aromatic H			
Field	Sample	Sample		Date		Modified	Carbon	TPH by IR		EPA 602/8			
ID	Number	Location	Time	(DDMMYY)	Matrix	8015	Range	Mod. 418.1			Ethyl-	Total	
						(MG/KG)	•	(MG/KG)	Benzene	Toluene	Benzene	Xylenes	MTBE
1	N11667	78N, 36E, 9'	1210	15-8-91	Soli	<5	N/A	42	< 0.025	<0.025	<0.025	< 0.025	<0.12
2	N11668	85N, 32E, 9'	1215	15-8-91	Soil	<5	N/A	19					
3	N11669	62N, 02W, 9'	1628	15-8-91	Soil	<5	N/A	46					
4	N11670	60N, 24E, 9'	840	16-8-91	Soil	<5	N/A	18					
5	N11671	68N, 06W, 9'	920	16-8-91	Soil	<5	N/A	108					
6	N11672	53N, 22W, 9'	1020	16-8-91	Soil	<5	N/A	22					
7	N11673	39N, 33W, 9'	1140	16-8-91	Soil	<5	N/A	68					
8	N11674	Field Blank, South, 0'	1145	16-8-91	Soil	<5	N/A	4		+			
c	N11675	Field Blank, North, 0'	1140	16-8-91	Soil	10	(C22-C36)	12					
13	N11676						· · · · · ·		<0.025	<0.025	< 0.025	< 0.025	<0.1
		50N, 16E, 11'	1415	19-8-91	Soil	<5	N/A	89	< 0.025	< 0.025	<0.025	<0.025	
14	N11677	41N, 09W, 12'	1437	19-8-91	Soil	<5	N/A	86		· ·			
15/16	N11678	40N, 5E/34N, 2W, 10'	848	20-8-91	Soil	<5	N/A	50/111			· · · ·		
17	N11679	Reserve pit, 100S,29E, 6'	1015	20-8-91	Soil	<5	N/A	87					
18	N11680	Reserve pit, 90S,10E, 6'	1025	20-8-91	Soil	<5	N/A	54.5					
19/20/21	N11681	42N,36E/50N,28E/10N,19E, 7'	1140	20-8-91	Soil	<5	N/A	42.5/102/109	< 0.025	< 0.025	< 0.025	< 0.025	< 0.1
22	N/A	Clay Layer, 90S, 10E, 6'	1411	20-8-91	Soil	Not Run	N/A	286					
23	N11682	32N,31E, 10'	1500	20-8-91	Soil	<5	N/A	49					
24	N11683	30N,15E, 8'	1525	20-8-91	Soil	29	(C10-C23)	130.5					
25	N11684	20N,27E, 10'	1540	20-8-91	Soil	<5	N/A	42.5		1			
26/27/28	N11685	31N,45E/18N,37E/23N,33E, 8'	750	21-8-91	Soil	<5	N/A	36/75/36	< 0.025	<0.025	< 0.025	< 0.025	< 0.1
29	N11686	04N,28E, 8'	1059	21-8-91	Soil	<5	N/A	24	<0.025	< 0.025	0.026	0.20	<0.1
30	N11687	Duplicate of #29	1100	21-8-91	Soil	<5	N/A	11	< 0.025	<0.025	<0.025	< 0.025	<0.1
31	N11688	10N,10E, 8'	1215	21-8-91	Soil	<5	N/A	20					
33/35/37	N11689	20N,35E/10N,50E, 6'	1245	21-8-91	Soil	<5	N/A	15/24/7					
34/36/38	N11690	20N,35E/10N,50E/0N,45E, 7.5'	1250	21-8-91	Soil	<5	N/A	16/15/9					
39	N11691	15S,35E, 7.5'	1320	21-8-91	Soil	<5	N/A	11					
40	N11692	15S,35E, 9'	1325	21-8-91	Soil	<5	N/A	8					-
41	N11693	55S,5W, 6'	1425	21-8-91	Soil	<5	N/A	8					
42	N11694	35S,30W, 6'	1430		Soil	<5	N/A	17					
43	N11695	15N,BW, 9'	1501	21-8-91	Soil	<5	N/A	12					
44/45	N11743	10N,15W/10N,30W, 8.5'	820	22-8-91	Soil	<5	N/A	104/88				· · · · ·	
46	N11744	20N,0E, 9'	905	22-8-91	Soil	<5	N/A	83.5					
47/48	N11745	5N,25W/0N,15W, 8'	1047	22-8-91	Soil	<5	N/A	99.5/60	< 0.025	< 0.025	< 0.025	< 0.025	< 0.1
47/48									<0.025	<0.025	<0.025	CO.025	
	N11746	10S,10E, 10'	1110	22-8-91	Soil	<5	N/A	45	-0.005	-0.005	-0.005	40.005	
50	N11747	25S,5E, 6'	1145		Soll	<5	N/A	42.5	< 0.025	< 0.025	< 0.025	< 0.025	<0.1
51	N11748	85,40W, 8'	1300		Soil	<5	N/A	50		+			
52	N11749	30S,5E, 7.5'	1340		Soit	15	(C10-C32)	87		<u> </u>			
53	N11750	10S,15E, 8'	1440		Soil	<5	N/A	70		-			
54	N11751	50S,8E, 7.5'	1500	22-8-91	Soil	<5	N/A	47					
55/56	N11752	30S,5W/40S,5W, 7'	815	23-8-91	Soil	48	(C8-C32)	83/75					
57	N11753	20S,20W, 7.5'	1013	23-8-91	Soil	<5	N/A	48.5	< 0.025	< 0.025	< 0.025	< 0.025	<0.1
60	N11754	Clean Cobble, Field Blank, 0'	1300	23-8-91	Soil	<5	N/A	Not Run					
61	N11755	35S,15W, 9'	1310	23-8-91	Soil	6	(C10-C22)	117					
62	N11756	35S,15 (#61 Dup.), 9'	1312	23-8-91	Soil	6	C8	110					
63	N11757	30W, Hay Pile, Field Blank, 0'	1300	23-8-91	Soil	<5	N/A	Not Run					
64	N11758	20S,M.W.#1 Field Blank, 0'	1308	23-8-91	Soil	10	(C10-C24)	Not Run	<0.025	<0.025	< 0.025	< 0.025	<0.1
65	N11759	Alfalfa Field Edge, Field Blank,	1315	23-8-91	Soil	<5	N/A	Not Run					
66	N11760	50S,5W, 7'	1450	23-8-91	Soil	<5	N/A	113.5					
67	N11761	50S,20W, 8'	800	26-8-91	Soil	<5	N/A	99					
68	N11762	50S,35W, 8'	845		Soil	<5	N/A	28	< 0.025	< 0.025	< 0.025	< 0.025	< 0.1
69	N11763	58S,38W, 8'	910		Soil	<5	N/A	92.5			1		
70	N11713	Reserve pit, 3' (Clay Layer)		26-8-91	Soil	47	(C12-C36)	252		1	· · · · ·		
71	N11714	Reserve pit, 4' (Clay Layer)	1020		Soil	26	(C12-C36)	307		1	1	1	+ ·

NOTE: Field Sample #8 - Location Unknown (TPH=41 Mg/Kg).

Sample #9,10,11,12 were at the same location as #14 but at a shallower depth.

Sample #32 was at the same location as #46 but at a shallower depth.

Sample #58 was at the same location as #61 but at a shallower depth.

MANANA MARY WHEELER #1E

MW.WK3 10-08-91

					Analytica	I Data Sumr	nary – Total F	Petroleum Hy	arbon			John Lambdi	in
								FIELD		Aromatic H	lydrocarbo	ns	
Field	Sample	Sample		Date		Modified	Carbon	TPH by IR		EPA 602/8	020 BETX	(UG/L)	
ID	Number	Location	Time	(DDMMYY)	Matrix	8015	Range	Mod. 418.1			Ethyl –	Total	acesare Gadicoo
						(MG/KG)		(MG/KG)	Benzene	Toluene	Benzene	Xylenes	MTBE
73	N11765	Ballard Clean Fill, 0'	1055	26-8-91	Soil	<5	N/A	Not Run					
74	N11820	70S, 30W, 7'	1105	26-8-91	Soil	<5	N/A	75					
75/76	N11821	62S, 8W/62S, 5W, 8'/7'	1310	26-8-91	Soil	6	(C18-C32)	61.5/81.5					
78	N11822	76S, 13W, 6'	1440	26-8-91	Soll	18	(C18-C32)	53					
79	N/A	76S, 18W, 8'	1515	26-8-91	Soil	Not Run	N/A	53.5					
80	N11823	70S, 20W, 7'	1545	26-8-91	Soll	60	(C5-C32)	221	< 0.025	< 0.025	< 0.025	0.16	<0.1
81	N11824	70S, 20W, 8'	800	27-8-91	Soil	<5	N/A	45.5				- 0.10	
82	N11825	67S, 45W, 7'	810		Soil	<5	N/A	61.5					
83	N11826	Duplicate of 82		27-8-91	Soil	<5	N/A	87					
84	N11827		810					37					
		80S, 8W, 7'	816	27-8-91	Soil	<5	N/A		10.005	-0.005	-0.005	10.005	-0.1
85	N11828	Duplicate of 84	816	27-8-91	Soil	<5	N/A	56	< 0.025	< 0.025	< 0.025	< 0.025	<0.1
86	N11829	86S, 17W, 7'	930	27-8-91	Soil	<5	N/A	40					
87	N11830	87S, 40W, 7.5'	1155	27-8-91	Soil	<5	N/A	84.5					
88	N11831	Field Blank, 0'	1255	27-8-91	Soil	18	(C18-C36)	72					+
89	N11832	92S, 25W, 7'		27-8-91	Soil	<5	N/A	90					
90	N11833	84S, 18W, 6'	1415	27-8-91	Soil	22	(C18-C32)	27	< 0.025	< 0.025	< 0.025	< 0.025	<0.1
91	N11834	80S, 50W, 7'	1420	27-8-91	Soil	<5	N/A	94					
92	N11835	87S, 55W, 7'	1500	27-8-91	Soil	<5	N/A	38.5					
93	N11836	Duplicate of 92, 7'	1500	27-8-91	Soil	<5	N/A	69					
94	N11837	100S, 47W, 8'	1520	27-8-91	Soil	<5	N/A	68					
95	N11838	Duplicate of 94, 8'	1520	27-8-91	Soil	<5	N/A	67.5	< 0.025	< 0.025	< 0.025	< 0.025	<0.1
96/98	N11839	105S, 56W/110S, 58W, 8'/8'	800	28-8-91	Soil	<5	N/A	41.5/41.5					
97	N11840	1+48S, 0+64W, 5'	910	28-8-91	Soil	<5	N/A	83.5					
99	N11841	108S, 65W, 7'	1045	28-8-91	Soil	<5	N/A	33.5					
100	N11842	Duplicate of 99, 7'	1045	28-8-91	Soll	<5	N/A	73.5					
101	N11843	110S, 38W, 7'	1140	28-8-91	Soil	<5	N/A	82.5	< 0.025	< 0.025	< 0.025	< 0.025	< 0.1
102	N11844	122S, 34W, 5'	1155	28-8-91	Soil	45	(C18-C32)	73			-		
103	N11845	180S, 92W, 4'	1255	28-8-91	Soil	<5	N/A	90					
104	N11846	Duplicate of 103, 4'	1255	28-8-91	Soil	<5	N/A	46					
105	N11847	1+255, 0+38W, 5'	1310	28-8-91	Soil	29	(C18-C32)	71				1	
106	N11848	129S, 52W, 5'	1340	28-8-91	Soil	<5	N/A	72	< 0.025	< 0.025	< 0.025	< 0.025	< 0.1
107	N11849	1+30S, 0+57W, 6.5'	1425	28-8-91	Soil	<5	N/A	119	40.020	40.020	40.020	10.020	
107	N11850	1+52S, 0+73W, 6'	807	29-8-91	Soil	<5	N/A	96				1	+
100	N11851		807	29-8-91								+	
		Duplicate of 109, 6'			Soil	<5	N/A	81					
110	N11852	1+60S, 0+75W, 5'	825	29-8-91	Soil	<5	N/A	92					
111	N11853	1+64S, 0+87W, 5'		29-8-91	Soil	10	(C20-C32)	71.5	<0.025	< 0.025	< 0.025	< 0.025	<0.1
112	N11854	1+61S, 0+91W, 7'		29-8-91	Soil	<5	N/A	86.5					
113	N11855	1+70S, 0+91W, 5'		29-8-91	Soil	13	(C12-C32)	294					
114	N11856	1+73S, 0+92W, 5'	1120	29-8-91	Soil	<5	N/A	69.5					
115	N11857	1+70S, 1+08W, 5'	1130	29-8-91	Soll	<5	N/A	58					
116	N11858	1+80S, 1+18W, 6'	1325	29-8-91	Soil	<5	N/A	102	< 0.025	< 0.025	<0.025	<0.025	<0.1
117	N11859	1+66S, 1+12W, 6'	1335	29-8-91	Soil	<5	N/A	88	ļ		ļ		
118	N11860	Duplicate of 117, 6'	1335	29-8-91	Soil	<5	N/A	88.5					
119	N11861	1+70S, 1+05W, 8'	1400	29-8-91	Soli	<5	N/A	83					
120	N11862	1+85S, 1+40W, 10'	1455	29-8-91	Soil	<5	N/A	86					
121	N11863	1+75S, 1+40W, 10'	1457	29-8-91	Soil	12	(C12-C32)	133	< 0.025	< 0.025	< 0.025	< 0.025	< 0.1
122	N11864	1+34S, 0+11W, 6'	1525	29-8-91	Soil	<5	N/A	94					
123	N11865	1+95S, 0+00E, 7'	1545	29-8-91	Soil	<5	N/A	122.5		1			1
124	N11866	1+04S, 0+12E, 7'	1555	29-8-91	Soil	<5	N/A	109.5				1	1
125	N11867	1+25S, 0+23W, 6'	815	30-8-91	Soil	<5	N/A	82		<u> </u>			1
126	N11868				Soil	<5	N/A	106		<0.025	<0.025	< 0.025	<0.
		1+10S, 0+00E, 6'		30-8-91	1			1			<u><u> </u></u>	<u> \0.025</u>	
127	N11869	0+96S, 0+00E, 6'	910		Soil	<5	N/A	85.5		<u> </u>		+	+
128	N11870	Duplicate of 127, 6'	910	******	Soil	<5	N/A	72.5					
N/A	N11912	CDK Clean Fill	1530	03-09-91	Soil	<5	N/A	Not Run					

NOTES: Sample #77 was at the same location as #79 but at a shallower depth.

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nalytical Technologies, Inc. 9830 S. 51st Street Suite B-113 Phoenix, AZ 85044 (602) 496-4400

ATI I.D. 108741

August 23, 1991

El Paso Natural Gas Company P.O. Box 4990 Farmington, NM 87499

Project Name/Number: M. Wheeler 1E

Attention: John Lambdin

On 08/19/91, Analytical Technologies, Inc. received a request to analyze aqueous sample(s). The sample(s) were analyzed with EPA methodology or equivalent methods. The results of these analyses and the quality control data, which follow each set of analyses, are enclosed.

If you have any questions or comments, please do not hesitate to contact us at (602) 496-4400.

Mary A. Cyce Mary Tyer

Project Manager

E. Proffitt for:

Robert V. Woods Laboratory Manager

RVW:jat Enclosure

Corporate Offices: 5550 Morehouse Drive San Diego, CA 92121 (619) 458-9141

Analytical Technologies, Inc.

CLIENT	: EL PASO NATURAL GAS, NEW MEXICO	DATE RECEIVED : 08/19/91
PROJECT #	: (NONE)	
PROJECT NAME	: M.WHEELER 1E	REPORT DATE : 08/23/91
	ATI I.D. : 108741	

ATI #	CLIENT DESCRIPTION	MATRIX	DATE COLLECTED
01	N11636	AQUEOUS	08/15/91

Arration, Observation trench split sayde with OCD,



---- TOTALS -----

MATRIX

SAMPLES 1

AQUEOUS

ATI STANDARD DISPOSAL PRACTICE

The samples from this project will be disposed of in thirty (30) days from the date of this report. If an extended storage period is required, please contact our sample control department before the scheduled disposal date.



ATI I.D. : 10874101

TEST : BTEX & MTBE (EPA METHOD 8020)

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CLIENT PROJECT # PROJECT NAME CLIENT I.D. SAMPLE MATRIX	: EL PASO NATURAL GAS, : (NONE) : M.WHEELER 1E : N11636 : AQUEOUS	DATE RECEIVED DATE EXTRACTED DATE ANALYZED UNITS DILUTION FACTOR	: 08/15/91 : 08/19/91 : N/A : 08/21/91 : UG/L : 1
		RESULTS	
BENZENE TOLUENE ETHYLBENZENE TOTAL XYLENES METHYL-t-BUTYI	ETHER	<0.5 <0.5 <0.5 <0.5 <0.5 <2.5	
SURROC	GATE PERCENT RECOVERIES		

BROMOFLUOROBENZENE (%)



REAGENT BLANK

TEST : BTEX & MTBE (EPA METHOD 8020)

PROJECT # PROJECT NAME	: EL PASO NATURAL GAS, : (NONE) : M.WHEELER 1E : REAGENT BLANK	ATI I.D. NEW MEXICO DATE EXTRACTE DATE ANALYZED UNITS DILUTION FACT	: 08/21/91 : UG/L
COMPOUNDS		RESULTS	
BENZENE TOLUENE ETHYLBENZENE TOTAL XYLENES METHYL-t-BUTYL	ETHER	<0.5 <0.5 <0.5 <0.5 <0.5 <2.5	

SURROGATE PERCENT RECOVERIES

BROMOFLUOROBENZENE (%)

Aceptablu 8-26-91 AS



QUAL	ITY CONTROL DATA		100741
TEST : BTEX & MTBE (EPA METHOD 8	020)	ATI I.D. :	108741
CLIENT : EL PASO NATURAL G PROJECT # : (NONE) PROJECT NAME : M.WHEELER 1E REF I.D. : 10899831	AS, NEW MEXICO	DATE ANALYZED : SAMPLE MATRIX : UNITS :	
COMPOUNDS	SAMPLE CONC. RESULT SPIKED	DUP. SPIKED % SPIKED SAMPLE REC.SAMPLE	
BENZENE TOLUENE ETHYLBENZENE TOTAL XYLENES METHYL-t-BUTYL ETHER	<pre><0.5 10 <0.5 10 <0.5 10 <0.5 10 <0.5 30 <2.5 20</pre>	8.1 81 8.1 7.9 79 7.8 7.8 78 7.9 23 77 24 17 85 17	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$

Aceptalla 8-26-01 0-J-

% Recovery = (Spike Sample Result - Sample Result) Х 100 Spike Concentration RPD (Relative % Difference) = (Spiked Sample - Duplicate Spike) Result Sample Result Х 100 -----Average of Spiked Sample

Image: NEEDED BY: Im	OSAL INSTRUCTIONS LAB NUMBER $OS 7 - Q$ RIOR AUTHORIZATION IS REQUIRED FOR RUSH PROJE L) $\chi_{\mathscr{L}}$ 5 (RUSH) \mathbb{Z}^4 \mathbb{Z}^8
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9830 S. 51st Street Suite B-113 Phoenix, AZ 85044 (602) 496-4400

ATI I.D. 109558

September 12, 1991

El Paso Natural Gas Company P.O. Box 4990 Farmington, NM 87499

Project Name/Number: Mary Wheeler

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Attention: John Lambdin

On 09/05/91, Analytical Technologies, Inc. received a request to analyze aqueous & soil sample(s). The sample(s) were analyzed with EPA methodology or equivalent methods. The results of these analyses and the quality control data, which follow each set of analyses, are enclosed.

If you have any questions or comments, please do not hesitate to contact us at (602) 496-4400.

Mary A. Lyer

Mary Tyer Project Manager

RVW:clf Enclosure

Jobert V. Woods

Robert V. Woods Laboratory Manager

Corporate Offices: 5550 Morehouse Drive San Diego, CA 92121 (619) 458-9141

2017月11日11日

Analytical **Technologies**, Inc.

CLIENT :	EL PASO NATURAL GAS, NEW MEXICO	DATE RECEIVED : 09/05/91
PROJECT # : PROJECT NAME :		REPORT DATE : 09/11/91
	ATI I.D. : 109558	

ATI #	CLIENT DESCRIPTION	MATRIX	DATE COLLECTED
01	N11911	AQUEOUS	09/03/91
02	N11912	SOIL	09/03/91



----- TOTALS -----

MATRIX	# SAMPLES
SOIL	1
AQUEOUS	1

ATI STANDARD DISPOSAL PRACTICE

The samples from this project will be disposed of in thirty (30) days from the date of this report. If an extended storage period is required, please contact our sample control department before the scheduled disposal date.



ATI I.D. : 10955802

TEST : FUEL HYDROCARBONS (MODIFIED EPA METHOD 8015)

CLIENT : EL PASO NATURAL GAS, NEW MEXICO PROJECT # : (NONE) PROJECT NAME : MARY WHEELER CLIENT I.D. : N11912 SAMPLE MATRIX : SOIL	DATE SAMPLED : 09/03/91 DATE RECEIVED : 09/05/91 DATE EXTRACTED : 09/05/91 DATE ANALYZED : 09/06/91 UNITS : MG/KG DILUTION FACTOR : 1
COMPOUNDS	RESULTS
FUEL HYDROCARBONS HYDROCARBON RANGE HYDROCARBONS QUANTITATED USING SURROGATE PERCENT RECOVERIES	<5 - -

DI-N-OCTYL-PHTHALATE (%)

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129

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Analytical Technologies, Inc.

GAS CHROMATOGRAPHY - RESULTS

REAGENT BLANK

TEST : FUEL HYDROCARBONS (MODIFIED EPA METHOD 8015) ATI I.D. : 109558 : EL PASO NATURAL GAS, NEW MEXICO DATE EXTRACTED : 09/05/91 CLIENT PROJECT # : (NONE) DATE ANALYZED : 09/06/91 PROJECT NAME : MARY WHEELER UNITS CLIENT I.D. : REAGENT BLANK DILUTION FACTOR : N/A _ _ _ _ _ _ _ _ _ _ _____

COMPOUNDS	RESULTS
FUEL HYDROCARBONS	<5
HYDROCARBON RANGE	-
HYDROCARBONS QUANTITATED USING	-

SURROGATE PERCENT RECOVERIES

DI-N-OCTYL-PHTHALATE (%)

1. 1. 9. 200 - 201 -

96

1914

: MG/KG



QUALITY CONTROL DATA ATI I.D. : 109558 TEST : FUEL HYDROCARBONS (MODIFIED EPA METHOD 8015) : EL PASO NATURAL GAS, NEW MEXICO CLIENT PROJECT # : (NONE) DATE ANALYZED : 09/06/91 **PROJECT NAME : MARY WHEELER** SAMPLE MATRIX : NON-AQUEOUS REF I.D. UNITS : 10999915 : MG/KG DUP. DUP. SAMPLE CONC. SPIKED & SPIKED & COMPOUNDS RESULT SPIKED SAMPLE REC. SAMPLE REC. RPD FUEL HYDROCARBONS <5 50 53 106 55 110 4

% Recovery = (Spike Sample Result - Sample Result) ----- X 100 Spike Concentration RPD (Relative % Difference) = (Spiked Sample - Duplicate Spike)

Result Sample Result Average of Spiked Sample

21 1800 A 11 10



ATI I.D. : 10955801

TEST : BTEX & MTBE (EPA METHOD 8020)

CLIENT PROJECT # PROJECT NAME CLIENT I.D. SAMPLE MATRIX	: EL PASO NATURAL GAS, NEW : (NONE) : MARY WHEELER : N11911 : AQUEOUS	DATE RECEIVED DATE EXTRACTED	: 09/03/91 : 09/05/91 : N/A : 09/09/91 : UG/L : 1
COMPOUNDS		RESULTS	
BENZENE TOLUENE ETHYLBENZENE TOTAL XYLENES METHYL-t-BUTY		<0.5 <0.5 <0.5 <0.5 <0.5 <2.5	
SURRO	GATE PERCENT RECOVERIES		

107

1566.2

11

BROMOFLUOROBENZENE (%)



REAGENT BLANK

TEST : BTEX & MTBE (EPA METHOD 8020)

PROJECT # PROJECT NAME	: EL PASO NATURAL GAS, NEW MEXIG : (NONE) : MARY WHEELER : REAGENT BLANK	ATI I.D. : 109558 CO DATE EXTRACTED : 09/09/91 DATE ANALYZED : 09/09/91 UNITS : UG/L DILUTION FACTOR : N/A	
COMPOUNDS		RESULTS	
BENZENE TOLUENE ETHYLBENZENE TOTAL XYLENES METHYL-t-BUTYL	ETHER	<0.5 <0.5 <0.5 <0.5 <0.5 <2.5	

SURROGATE PERCENT RECOVERIES

BROMOFLUOROBENZENE (%)

105



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-	LITY CONTROL DAT	A ATI I.D.	:	109558	
TEST : BTEX & MTBE (EPA METHOD	8020)				
CLIENT : EL PASO NATURAL PROJECT # : (NONE) PROJECT NAME : MARY WHEELER REF I.D. : 10999804	GAS, NEW MEXICO	DATE ANAI SAMPLE MA UNITS	ATRIX :		
COMPOUNDS	SAMPLE CONC RESULT SPIK	. SPIKED % ED SAMPLE REC.	DUP. SPIKED SAMPLE	DUP. % REC.	RPD
BENZENE TOLUENE ETHYLBENZENE TOTAL XYLENES METHYL-t-BUTYL ETHER	<0.5 10 <0.5 10 <0.5 10 <0.5 30 <2.5 20	8.7 87 8.9 89 10 100 31 103 18 90	8.9 8.8 9.7 30 18	89 88 97 100 90	2 1 3 3 0

* Recovery = (Spike Sample Result - Sample Result)		
Spike Concentration		
RPD (Relative % Difference) = (Spiked Sample - Duplicate Spike) Result Sample Result		
	Х	100

Average of Spiked Sample

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9830 S. 51st Street Suite B-113 Phoenix, AZ 85044 (602) 496-4400

ATI I.D. 109619

September 12, 1991

El Paso Natural Gas Company P.O. Box 4990 Farmington, NM 87499

Project Name/Number: Mary Wheeler

Attention: John Lambdin

On 09/07/91, Analytical Technologies, Inc. received a request to analyze aqueous sample(s). The sample(s) were analyzed with EPA methodology or equivalent methods. The results of these analyses and the quality control data, which follow each set of analyses, are enclosed.

If you have any questions or comments, please do not hesitate to contact us at (602) 496-4400.

Mary L

Mary Tyer Project Manager

RVW:clf Enclosure

Robert V. Woods Laboratory Manager

Corporate Offices: 5550 Morehouse Drive San Diego, CA 92121 (619) 458-9141

Analytical Technologies , In	ic.
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CLIENT : EL PASO NATURAL GAS, NEW MEXICO DATE	RECEIVED :	09/07/91
PROJECT # : (NONE)		
PROJECT NAME : MARY WHEELER REPO	RT DATE :	09/12/91
ATI I.D. : 109619		

ATI #	CLIENT DESCRIPTION	MATRIX	DATE COLLECTED
01	N11938	AQUEOUS	09/05/91



----- TOTALS -----

MATRIX # SAMPLES -----AQUEOUS 1

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ATI STANDARD DISPOSAL PRACTICE

The samples from this project will be disposed of in thirty (30) days from the date of this report. If an extended storage period is required, please contact our sample control department before the scheduled disposal date.



ATI I.D. : 10961901

TEST : BTEX & MTBE (EPA METHOD 8020)

CLIENT PROJECT # PROJECT NAME CLIENT I.D. SAMPLE MATRIX	: EL PASO NATURAL GAS, NEW MEXICO : (NONE) : MARY WHEELER : N11938 : AQUEOUS	DATE SAMPLED : 09/05/91 DATE RECEIVED : 09/07/91 DATE EXTRACTED : N/A DATE ANALYZED : 09/10/91 UNITS : UG/L DILUTION FACTOR : 1
COMPOUNDS		RESULTS
BENZENE TOLUENE ETHYLBENZENE TOTAL XYLENES METHYL-t-BUTY]		<0.5 <0.5 <0.5 <0.5 <2.5

SURROGATE PERCENT RECOVERIES

BROMOFLUOROBENZENE (%)



REAGENT BLANK

TEST : BTEX & MTBE (EPA METHOD 8020)

PROJECT # : PROJECT NAME :	EL PASO NATURAL GAS, (NONE) MARY WHEELER REAGENT BLANK	NEW MEXICO	ATI I.D. DATE EXTRACTED DATE ANALYZED UNITS DILUTION FACTOR	: 109619 : 09/10/91 : 09/10/91 : UG/L : N/A
COMPOUNDS			RESULTS	
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SURROGATE PERCENT RECOVERIES

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ETHYLBENZENE

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QUALITY	CONTROL	DATA
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TEST : BTEX & MTBE (EPA METHOD 8020)		ATI 1	.D.	:	109619	
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% Recovery = (Spike Sample Result - Sample Result) 100 _____ Х Spike Concentration RPD (Relative % Difference) = (Spiked Sample - Duplicate Spike) Result Sample Result ____ 100

Х Average of Spiked Sample

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TO: Anu Pundari

FROM: John Lambdin

DATE: September 12, 1991

PLACE: North Engineering Laboratory-Farmington

RE: MANANA MARY WHEELER #1E TESTING

Sample Number:N11940Sample Date:9-5-91Sampled By:Dennis BirdLocation:Mary WheelerSample Point:Monitor Well S-1Sample Time:1545 HoursMatrix:Water

The sample was tested for total dissolved solids (TDS) and found to contain 752 mg/L. A duplicate analysis yielded 750 mg/L TDS.

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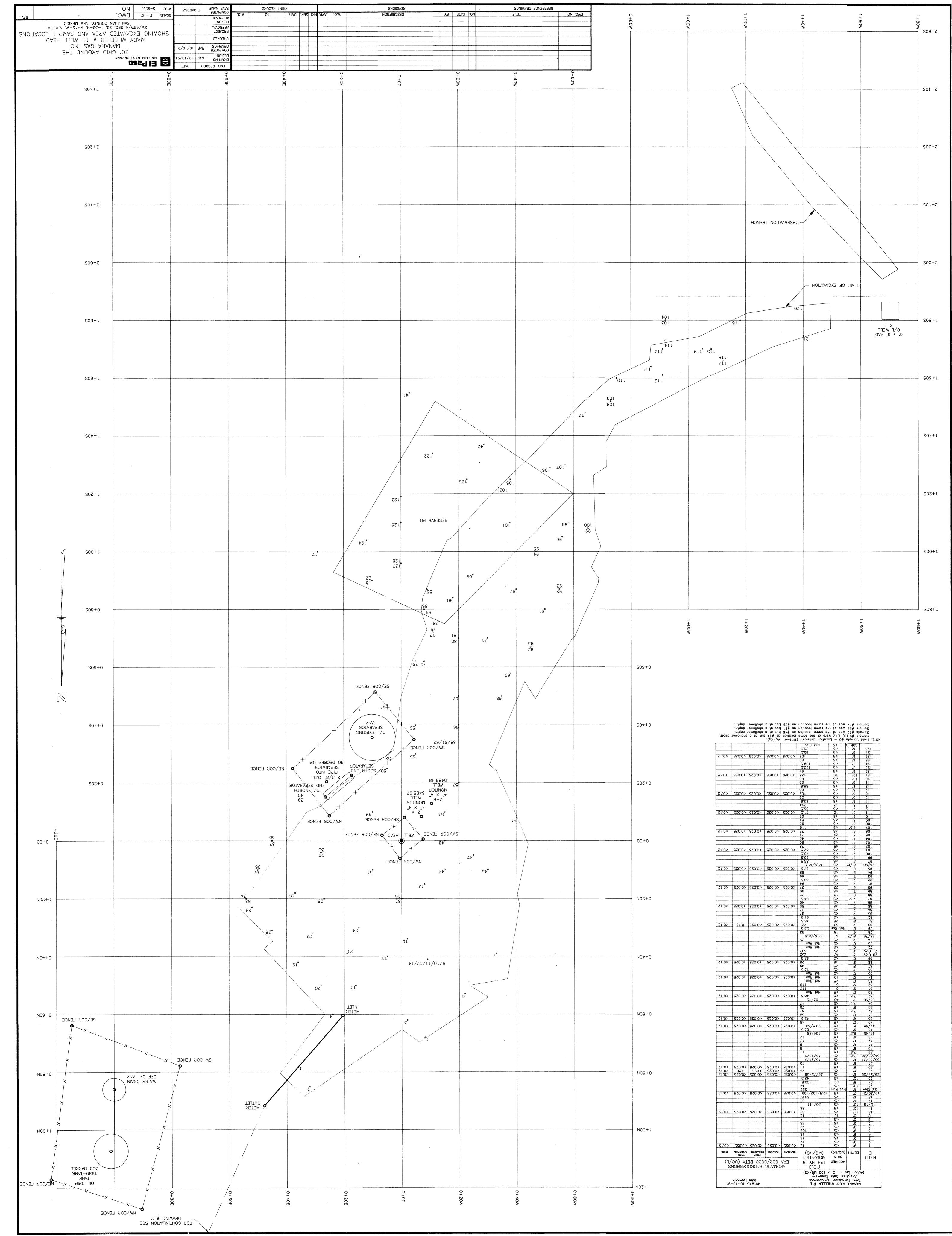
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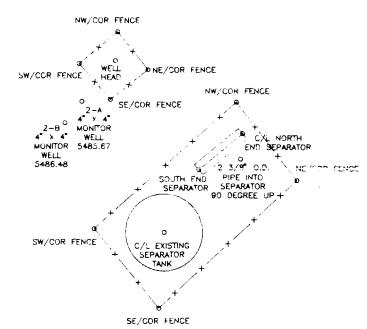
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OIL CONSERVATION DIVISION RECATED '91 OCT 22 AM 8 45

P. O. BOX 1492 EL PASO, TEXAS 79978 PHONE: 915-541-2600

October 17, 1991

Mr. David Boyer Environmental Bureau Chief New Mexico Oil Conservation Division P. O. Box 2088 State Land Office Building Santa Fe, NM 87504

Re: Remediation at Manana-Mary Wheeler #1E Gas Well Site, Flora Vista, New Mexico

Dear Mr. Boyer:

On October 11, we sent you the Manana-Mary Wheeler #1E Site Remediation package. After mailing, we discovered a line was missing on Page 4. Please replace with the attached. We apologize for the inconvenience.

Very truly yours,

thomas D. Hutchis

Thomas D. Hutchins Manager North Region Compliance Engineering Page 4 Manana Mary Wheeler #1E Remediation

III. Reserve Pit

During the 1989 K.W. Brown investigation, the reserve pit area was identified. The size of the pit was approximate and based on sizes of similar pits within the San Juan Basin. During remediation, EPNG dug trenches on the north and south edges of the reserve pit. By digging the trenches and locating the clay layer, EPNG determined the size of the reserve pit.

Dave Boyer directed EPNG to remove all clays and drilling mud in the reserve pit. Approximately one half of the reserve pit area was within the plume. In the area outside of the plume, clean overburden and uncontaminated substrate material was excavated and stockpiled for use as clean backfill. The contractor removed the clay zone and soil beneath the clay zone. Field verification samples were obtained at a depth of six feet in the reserve pit area. The clay layer varied in depth but in most cases, it was approximately three feet from grade and one foot thick.

IV. Cleanup and Closure of Observation Trench

Water samples from the observation trench were obtained on August 15 and September 3. The results are under Tab 3. The results were below WQCC standards. During the excavation activities, there was no sign of hydrocarbons in the observation trench. Therefore, on September 3, the fence around the observation trench was removed and the trench was backfilled.

V. Eight Hour Pump Test

Since the observation trench showed no signs of hydrocarbon contamination during the excavation activities, Dave Boyer agreed to an eight hour pump test. Originally, it was planned to pump the water into "frac" tanks, analyze the water and truck the water to EPNG's Kutz lined pond.

However, since the contractor had the 6000 gallon portable water tank at the site, EPNG decided to pump the water directly into the portable tank instead of a "frac" tank. A water truck hauled the water from the portable tank to the Kutz Lined Pond.

On September 5, an eight hour pump test was conducted. Well S1 was pumped at a rate of approximately 66 gallons per minute. During the beginning, middle and end of the test, the water did not show any signs of hydrocarbon contamination. The water was clear and did not have a hydrocarbon odor.

After the pump test, the water was analyzed for BETX and TDS. The results are under Tab 4. The BETX results are below WQCC standards.



October 11, 1991

P. O. BOX 1492 EL PASO, TEXAS 79978 PHONE: 915-541-2600

RECEIVED

OCT 1 5 1991

OIL CONSERVATION DIV. SANTA FE

Mr. David Boyer Environmental Bureau Chief New Mexico Oil Conservation Division P.O. Box 2088 State Land Office Building Santa Fe, New Mexico 87504

Re: Remediation at Manana-Mary Wheeler #1E Gas Well Site, Flora Vista, New Mexico

El Paso Natural Gas Company(EPNG) recently completed remediation activities at the referenced site. Attached is a synopsis of excavation activities, field Total Petroleum analyses results, Modified 8015 analyses results and results from the eight hour pump test.

As noted in Section VI of the synoposis, EPNG requests permission to plug and abandon Well S1. This request is based on the successful soil and groundwater remediation, lack of any visible discolored water appearing in the observation trench, the clear water observed during the pump test and the results of the samples collected during the pump test.

We look forward to your early consideration of this matter.

Sincerely yours,

Thomas D. Hulchims

Thomas D. Hutchins North Region Compliance Manager

Encl: As stated

MANANA MARY WHEELER #1E SITE REMEDIATION

I. Introduction and Background Information

Manana Gas, Inc. operates a producing natural gas well in Flora Vista, New Mexico. Manana Gas facilities include an aboveground oil storage tank, a separator and below grade separator liquids storage tank. El Paso Natural Gas Company (EPNG) currently has a meter house and meter run at the site. In the 1980's, EPNG also had dehydration facilities at the site. The facilities were removed in 1987. The well is in an area directly adjacent to the Animas River. Hydrocarbons were detected in one water well located hydrologically downgradient of the gas production units. El Paso Natural Gas Company conducted a site investigation of the site in 1989.

Based on OCD's authority to require Manana Gas and EPNG to conduct the necessary cleanup activities, EPNG removed hydrocarbon contaminated soil and drilling muds near the wellsite.

II. Excavation Activities

Diamond D Construction of Kirtland, New Mexico was the project contractor. Manana Gas removed their ancillary production units and shut-in the well prior to excavation. In order to assure worker safety and access to the site, EPNG removed the meter house and meter run prior to excavation activities.

On August 15, 1991, EPNG conducted a pre-excavation safety meeting at the site. First, a temporary trench, located approximately ten feet southwest of Well S1, was constructed to serve as an observation trench. The observation trench allowed natural volatilization of any dissolved hydrocarbons. The observation trench was approximately eighty five feet long, eight feet wide and five feet deep. The contractor initially installed a portable air compressor and slotted PVC pipe to act as a "air stripper" in the trench. Although the air compressor was equipped with a dry element filter, the filter was not sufficient. After discussion with Dave Boyer, EPNG requested the contractor to recirculate the water in the trench with a portable pump. A five foot high fence was installed around the observation trench.

After construction of the observation trench, the contractor began excavation with a Komatsu 200 LC backhoe in an area north of the meter house. During excavation, the top three to five feet of clean overburden was removed and stockpiled adjacent to the excavation area. The clean overburden was used for backfill after all hydrocarbon contaminated soil was removed.

The contractor excavated below the groundwater table. The groundwater table at the site varied from 5 1/2 feet from grade in the northern half of the excavation to 4 feet from grade in the southern half of the excavation. Water infiltrating into the

Page 2 Manana Mary Wheeler #1E Remediation

excavation area was pumped into a portable 6000 gallon water tank. From the portable tank the water was transferred into an 80 barrel water truck. The water was discharged into EPNG's Kutz Plant lined pond.

Kutz Plant is located approximately ten miles from the project site. Two absorbent sweeps were stored at Kutz lined pond. The sweeps were on hand to move any freephase hydrocarbons to one side of the pond. If there were hydrocarbons, EPNG planned to collect and recycle the hydrocarbons. Not enough hydrocarbons were detected in the water discharged to require skimming.

Four 20 yard end dumps and one belly dump hauled excavated soil from the job site to EPNG's Ballard Plant. Trucks were allowed to leave the site after most of the water had been drained from the soil, provided the water was not oily. The soil was fully drained prior to leaving the site if oily water was encountered. Belly dump trucks hauled only dry soils. The truck hauling procedures were approved by NMOCD.

Although EPNG constructed a trough to collect water drainage from the end dumps, the trough was not used at the site. During the first three days of the of the project, a twenty four inch diameter PVC pipe was cut in half and was placed underneath the dump truck tailgate. The water collected in the PVC pipe was diverted back to the open excavation trench. Later, a John Deere 544 Loader bucket was used to collect any water drainage from the end dumps. The John Deere 544 Loader returned the water back into the excavation trench.

One hundred oil absorbent pads were stored at the site in case large volumes of hydrocarbons were liberated during excavation. The pads were not used since no significant volumes of hydrocarbons were encountered in the excavation area.

The Farmington Division Laboratory conducted field total petroleum hydrocarbons (TPH) test with a Horiba OCMA-220 Oil Content Analyzer using the 418.1 IR method. Since the background TPH concentration was 35 ppm near the site, an action level of 135 ppm was set by NMOCD. The background sample analytical data is under Tab 1. Figure 2, under Tab 6, shows the location of the background samples. During remediation, the soil was excavated until the field TPH was below 135 ppm. Although every effort was made to obtain samples in a grid , it was difficult since the progress of excavation followed the hydrocarbon contamination.

Each field sample was analyzed twice by the Horiba OCMA-220. The two results were averaged and noted under Field TPH in the Analytical Data Summary table under Tab 2. When the field results were less than 135 ppm, verifications samples were sent to Analytical Technologies Laboratories in Phoenix, Arizona for Modified 8015 verification analysis. Every fifth sample was analyzed

Manana Mary Wheeler #1E Remediation

Page 3

for BTEX by Method 8020. Duplicate and blank samples were collected and analyzed for quality assurance/quality control. A summary of the analytical results is presented in the Table under Tab 2. A map of the excavated area showing the sample locations, is under Tab 5.

On the west edge of the excavation, visual and field TPH tests verified the western extent of contamination. When the soil turned from black to gray and the field TPH tests were below 135 ppm, further westward excavation was discontinued. On the east edge of the excavation, the soil remained black. Therefore, excavation continued until the field TPH was below 135 ppm.

The excavated soil was hauled to EPNG's Ballard Plant and spread in a six inch layer. Ballard Plant is located approximately 45 miles from the excavation site. The shallowest useable aquifer is about 440 feet deep at Ballard Plant. The soil was disced twice a week for the duration of the excavation activities. A Caterpillar D-6 with a ripper disced the soil. The entire soil remediation site was bermed to prevent runoff from entering and leaving.

Clean sandy backfill from Ballard Plant and a local contractor, CDK Inc., was used to backfill the site. Below a depth of four feet from grade, riverstone cobble and coarse grained sandy backfill was required. This material was placed to reduce the probability of future settling in the area. Clean, sandy backfill was utilized for the top four feet of the excavated area.

On August 28th, the excavation was at the southern edge of the plume, near the large cottonwood trees. The landowner did not want his large cottonwood trees cut unless it was absolutely necessary. Dave Boyer asked for a representative of Flora Vista Water Users Association to be at the site. The representative, Mr. Leon Knowlton, approved the plan to not cut the large cottonwood trees. EPNG cut down small cottonwoods and Russian olive trees in order to follow the plume to Well S1.

The contractor excavated, hauled and spread 3621.82 tons of contaminated soil. The backfill from Ballard Plant and CDK totalled 1849.45 tons of cobbly backfill and 1474.40 tons of sandy backfill for the site. The contractor transported 2590 barrels of water to Kutz lined pond from the excavation area. In addition, 2590 cubic yards of clean overburden was removed and reinstalled in the excavation area.

After excavation was completed, the excavated area was graded with a A550 Gallion Grader to meet the tenant, Mr. Eli Velasquez, specifications.

Manana Mary Wheeler #1E Remediation

III. Reserve Pit

Page 4

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Water samples from the observation trench were obtained on August 15 and September 3. The results are under Tab 3. The results were below WQCC standards. During the excavation activities, there was no sign of hydrocarbons in the observation trench. Therefore, on September 3, the fence around the observation trench was removed and the trench was backfilled.

V. Eight Hour Pump Test

Since the observation trench showed no signs of hydrocarbon contamination during the excavation activities, Dave Boyer agreed to an eight hour pump test. Originally, it was planned to pump the water into "frac" tanks, analyze the water and truck the water to EPNG's Kutz lined pond.

However, since the contractor had the 6000 gallon portable water tank at the site, EPNG decided to pump the water directly into the portable tank instead of a "frac" tank. A water truck hauled the water from the portable tank to the Kutz Lined Pond.

On September 5, an eight hour pump test was conducted. Well S1 was pumped at a rate of approximately 66 gallons per minute. During the beginning, middle and end of the test, the water did not show any signs of hydrocarbon contamination. The water was clear and did not have a hydrocarbon odor.

results are under Tab 4. The BETX results are below WQCC standards. Page 5 Manana Mary Wheeler #1E

VI. Future Work At Site

All EPNG and NMOCD monitoring wells within the excavated area were removed during excavation activities. EPNG also removed all NMOCD's monitoring wells located south of the excavated area. Since the BETX results were below WQCC standards after the eight hour pump test, EPNG plans to plug and abandon Well S1. STATE OF NEW MEXICO

ENERGY, MINERALS AND NATURAL RESOURCES DEPARTMENT

OIL CONSERVATION DIVISION



BRUCE KING GOVERNOR

October 11, 1991

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87504 (505) 827-5800

Mr. Thomas D. Hutchins, Manager North Region Compliance Engineering El Paso Natural Gas Company P.O. Box 1492 El Paso, TX 79978

RE: EPNG FLORA VISTA REMEDIATION SAN JUAN COUNTY, NEW MEXICO

Dear Mr. Hutchins:

Enclosed please find a copy of a BTEX water analysis for the Flora Vista aeration trench taken on August 15. Please note that the sample was taken after the air compressor had been activated. As you may know, we suspected during this time that the compressor filters were not removing hydrocarbons because a blue haze was observed above the water in the trench. The haze dissipated after the compressor was shut down. It was replaced by a water pump that circulated the water. Because of the circumstances of this incident, I do not believe the low level of benzene is significant and that the value represents failure of the compressor filters and not evidence of other contamination.

If you have any questions regarding this analysis please call Bill Olson at (505) 827-5885 after October 21.

Sincerely,

David G. Boyer / Environmental Bureau Chief

enc.

cc: OCD Aztec Office

OIL CO	TE OF NEW MEXICO DINATURAL RESOURCE EPARTMENT DINSERVATION DIVISION
Contract Lab IML	Contract No
OCD Sample No. 9108151105	
Collection Date Collection Time Collected by —Person/Agency	
9108151105 Boyen	/OCD
SITE INFORMATION	
Sample location FLORA UISTA - Aero	ation Tsench
Collection Site Description	sol the sh
Sample Scom N. Suce center Approx 90 minutes ofter # C Shut ofter	
SEND ENVIRONMENTAL BUREAU FINAL NM OIL CONSERVATION DIVISION REPORT DO Box 2008	SAMPLE FIELD TREATMENT — Check proper boxes
TO \$ Santa Fe, NM 87504-2088	No. of samples submitted:
SAMPLING CONDITIONS Water level Bailed Pump Discharge 'Dipped Tap	☑ NF: Whole sample (Non-filtered) □ F: Filtered in field with 0.45 µmembrane filter □ PF: Pre-filtered w/45 µmembrane filter
pH(00400) Sample type Description Conductivity (Uncorrected) Water Temp. (00010) "mho	
Conductivity at 25° C	FIELD COMMENTS:

LAB ANALYSIS REQUESTED:

ITEM	DESC	METHOD	<u>ITEM</u>	DESC	METHOD	ITEM	DESC	METHOD
X 001	VOA	8020	013	PHENOL	604	026	Cd	7130
002	VOA	602	014	VOC	8240	027	РЪ	7421
003	VOH	8010	015	voc	624	028	Hg(L)	7470
004	VOH	601	016	SVOC	8250	031	Se	7740
× 005	SUITE	8010-8020	017	SVOC	625	032	ICAP	6010
006	SUITE	601-602	018	VOC	8260	033	CATIONS/ANI	IS
007	HEADSPACE		019	SVOC	8270	034	N SUITE	-
008	PAH	8100	020	O&G	9070	035	NITRATE	į
009	PAH	610	022	AS	7060	036	NITRITE	:
010	PCB	8080	023	Ba	7080	037	AMMONIA	
011	PCB	608	024	Cr	7190	038	TKN	
012	PHENOL	8040	025	Cr6	7198		OTHER	



2506 West Main Street Farmington, New Mexico 87401 Tel. (505) 326-4737

VOLATILE AROMATIC HYDROCARBONS

Client: NMOCD		Report Date:	09-26-91
Sample ID:	Flora Vista Trench	Date Sampled:	08-15-91
Laboratory Number:	6949	Date Received:	08-16-91
Analysis Requested	: 8020	Date Analyzed:	08-28-91
Sample Matrix:	Water	Preservative:	Cool & HgCl2
Condition:	Cool & Intact		

Parameter	Concentration (ug/L)	Det. Limit (ug/L)
Benzene	1.4	0.5
Toluene	ND	0.5
Chlorobenzene	ND	0.5
Ethylbenzene	ND	0.5
p,m-Xylene	ND	0.5
o-Xylene	ND	0.5
1,3-Dichlorobenzene	ND	0.5
1,4-Dichlorobenzene	ND	0.5
1,2-Dichlorobenzene	ND	0.5

SURROGATE	RECOVERIES :	Parameter	Percent Recovery
		4-Bromofluorobenzene	e 93.2 %

Method: Method 8020, Aromatic Volatile Organics, SW-846, USEPA, (Sept. 1986). ND - Parameter not detected at the stated detection limit.

Comments:

Tony Tristano Analyst

STATE OF NEW MEXICO

ENERGY, MINERALS AND NATURAL RESOURCES DEPARTMENT

OIL CONSERVATION DIVISION



BRUCE KING

GOVERNOR

August 21, 1991

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87504 (505) 827-5800

<u>CERTIFIED MAIL -</u> <u>RETURN RECEIPT NO. P-756-666-155</u>



Mr. Thomas D. Hutchins, Manager North Region Compliance Engineering El Paso Natural Gas Company P. O. Box 1492 El Paso, Texas 79978

RE: Remediation of Manana Mary Wheeler #1E Gas Well Site Flora Vista, New Mexico

Dear Mr. Hutchins:

The Oil Conservation Division (OCD) has approved additional changes or clarifications to EPNG's work plan for remediation at the above site. These changes were requested in a phone call on August 20 to OCD from Anu Pundari and John Lambdin of EPNG's Farmington Office.

Based on the information provided in that phone conversation, I approved the following modifications to the site work plan:

- 1. Three trench samples, each one taken every ten feet, can be composited to be tested for method 8015 verification.
- 2. Excavation of deep contaminated material exceeding 135 ppm TPH is to be continued until the depth limitation of on-site equipment is reached. At that point, a soil verification sample should be obtained, along with a water sample for BTEX testing. The apparent aeral extent of the unreclaimed material and grid location should be noted and the area backfilled. Further evaluation will be made after test results are received.
- 3. Removal of clays, and material exceeding the 135 ppm TPH limit, is required at the drilling mud/reserve pit. Removal of clean overburden or uncontaminated substrate material is not necessary. However, trenching below the obvious clay zone should be performed periodically to confirm the absence of mixed material at depth.

Mr. Thomas D. Hutchins August 21, 1991 -2-

Copies of this letter and yesterday's (8-20) letter are being faxed to Ms. Pundari in Farmington. If you have any questions please contact me at (505) 827-5812.

Sincerely,

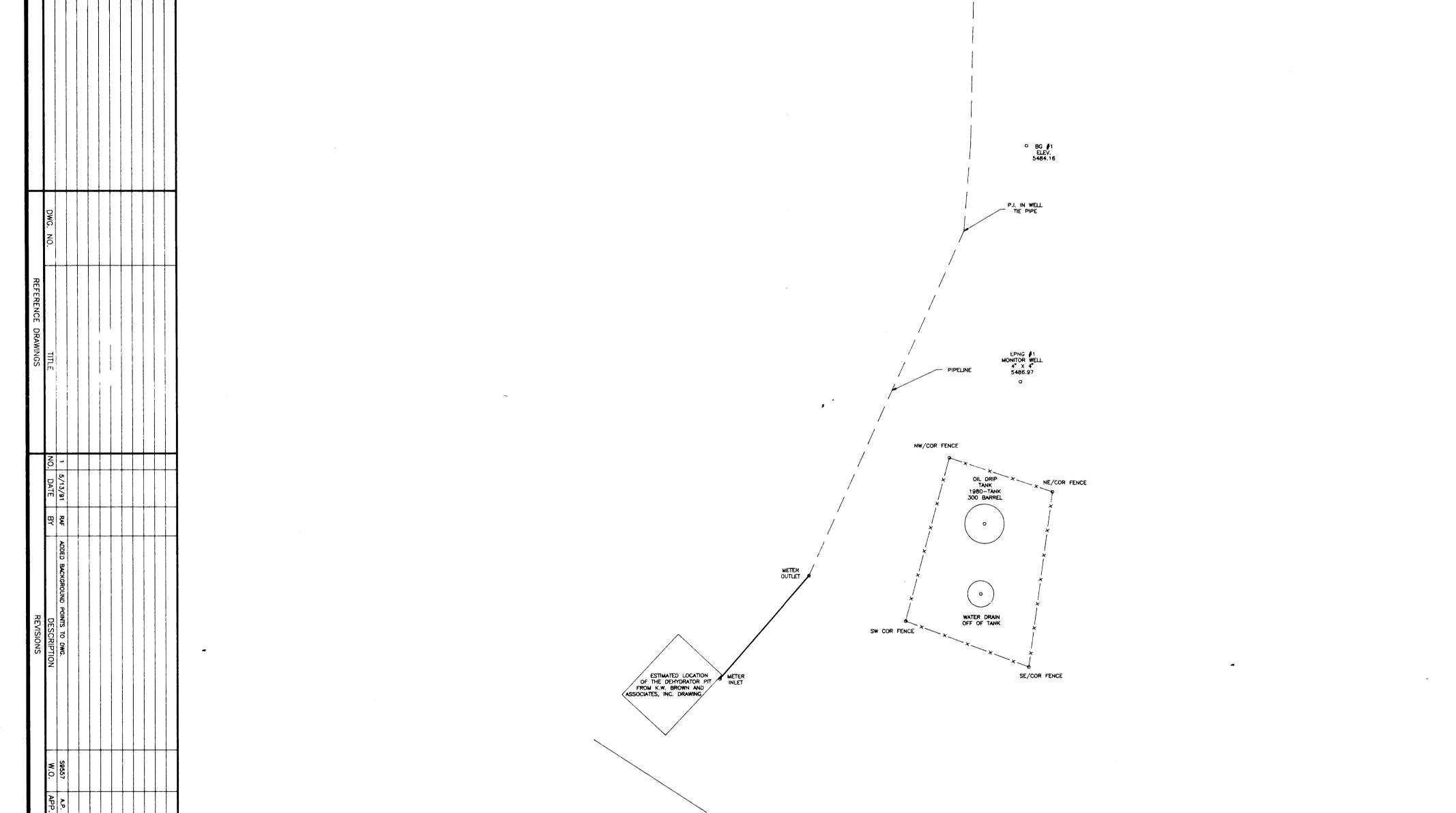
J. Barry David G. Boyer, Hydrogeologist

Environmental Bureau Chief

DGB/sl

cc: OCD Aztec Office Anu Pundari, EPNG, Farmington

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° BC #3 ELX, 5465.40	W
O BG ∦2 ELEV. 5484.85	



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S. RECORD DATE ING RAF 2/12/91 IUTER RAF 2/12/91 IUTER RAF 2/12/91 ICS MATURAL GAS COMPANY PLAT SHOWING ON THE MARY WHI SCALE: 1"-20' DWG. NO.: S-9557 NO.	RECEIVED AUG 1 3 1991 OL CONSERVATION DIV SANTA FE	SE/COR FENCE	
PLAT SHOWING LOCATION OF FACILITIES ON THE MANANA GAS INC MARY WHEELER # 1E WELL SW/4SW/4 SEC. 23, T-30-N, R-12-W, N.M.P.M. SAN JUAN COUNTY, NEW MEXICO NO. NO. 1	NOTE PRELIMINARY DRAWING FOR APPROVAL ONLY FINISH DRAWING AND DRAWING NUMBER TO DRAWING NUMBER TO DRAWING NUMBER TO		

STATE OF NEW MEXICO

ENERGY, MINERALS AND NATURAL RESOURCES DEPARTMENT

OIL CONSERVATION DIVISION



BRUCE KING GOVERNOR

August 20, 1991

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87504 (505) 827-5800

<u>CERTIFIED MAIL -</u> <u>RETURN RECEIPT NO. P-756-666-150</u>

Mr. Thomas D. Hutchins, Manager North Region Compliance Engineering El Paso Natural Gas Company P. O. Box 1492 El Paso, Texas 79978

RE: Remediation of Manana Mary Wheeler #1E Gas Well Site Flora Vista, New Mexico

Dear Mr. Hutchins:

Your letter of August 9, 1991, containing the results of the background tests for Total Petroleum Hydrocarbons (TPH) using the 418.1 IR method has been received. I have reviewed the data and concur with EPNG's request to use a TPH of 35 ppm as a background value using this method.

Additionally, in a conversation on August 13, Ms. Anu Pundari requested that the number of verification samples for BTEX be reduced. Because of the expected large number of samples (in excess of 60) and because BTEX at the site was not a major contributor to the contamination, I concurred with her request. We agreed that only every fifth (5) sample is to be analyzed for BTEX.

I was at the site on Thursday and Friday of last week and verbally approved a number of further changes in the work plan:

- 1. Composites of four soil samples were approved for method 8015 verification tests.
- 2. Trucks were allowed to leave the site after most water had been drained from the dirt load provided the water was not oily. Full draining is required for oily water. Bellydump trucks for dry soils were approved Monday morning (8-19) after a conversation with Nancy Prince.

Mr. Thomas D. Hutchins August 20, 1991 Page 2

3. A water pump was approved for use in the aeration trench. It replaced an air compressor which was imparting a blue haze to the trench indicating that the air filters were not completely effective.

I was very impressed with the portable laboratory which Norman Norvelle has set up. It worked very well showing that much forethought was put into its construction and operation. It's operation will greatly speed up excavation work.

Kathy Brown of this office will be on site Thursday and Friday of this week to observe excavation of the soils. I expect to return for several days during the week of August 26.

Sincerely,

David G. Boyer, Hydrogeologist Environmental Bureau Chief

DGB/sl

OCD Aztec Office cc: Anu Pundari, EPNG



P. O. BOX 1492 EL PASO, TEXAS 79978 PHONE: 915-541-2600

August 9, 1991

RECEIVED

Mr. David Boyer Environmental Bureau Chief New Mexico Oil Conservation Division P.O. Box 2088 Land Office Building Santa Fe, New Mexico 87504-2088

AUG 1 3 1991

OIL CONTRACTION DIV. SANTA FE

RE: Remediation of Manana-Mary Wheeler #1E Gas Well Site, Flora Vista, New Mexico

Dear David:

In accordance with the plan approved by the New Mexico Oil Conservation Division (NMOCD), attached is a copy of Anu Pundari's memo concerning the sampling to determine a background Total Petroleum Hydrocarbon level at the Mary Wheeler #1E well site.

The El Paso lab will perform the 418.1 IR Method and an outside lab will perform the analysis by the Modified 8015 Method. Based on the attached results, El Paso proposes the following background levels be used for the remediation at Mary Wheeler #1E:

- 418.1 IR Method (20+40+74+20)/4 = 38.5 ppm, Use 35 ppm. Therefore, when the results from the field detection unit are less than 135 ppm, a verification sample will be caught and the area backfilled.
- Modified 8015 Method The background level will be zero (0). The results should not exceed 100 ppm.

I will be out of the office August 12 - 15. Therefore, please contact Anu Pundari, at (505) 599-2176 if you have any questions or need additional information.

Sincerely,

J. D. Hatchino, gb

Thomas D. Hutchins Manager, North Region Compliance

Attachments: As stated c: Anu Pundari

FLORA VISTA CONTAMINATION MANANA-MARY WHEELER

TESTING

- Using the portable IR (infrared) spectrophotometer for field screening of TPH in the soils (418.1). Pick up additional organics so determined a background level. Background level determined is 35 ppm. Soils must be below the UST standard of 100 ppm. Therefore, the remediation level is 135 ppm TPH. Anything below this level can be left in place.
- 2. Use modified 8015 for TPH and 8020 for BTEX. Composite samples are taken from each area = to the backhoe swing width x 50' or the length of the excavation (smaller). OCD requires 1 sample per 10 feet are to be composited for 8015 verification
- 2. Highest contamination is TPH. BTEX is not a major contamination contributor. The reserve pit is the major source of the contamination.
- 3. The excavation will begin in the area farthest up gradient. Any clean (nonstained) overburden soils can be stockpiled and used for backfill. Any excavated soils less than 135 ppm TPH can be used for backfill. For soils over 135 ppm TPH (field analyzed) must have a lab verification. If excavations proceed to a depth of 8 feet, need to stop operations and assess the need for further excavations.

EL PASO NATURAL GAS COMPANY

MEMORANDUM

To: Tom Hutchins

Date: June 4, 1991

From: A.N. Pundari

Place: North Region Engineering

Re: Mary Wheeler #1E Background Samples

In order to determine the background TPH level, four soil samples were collected in an area no less than 100 feet up gradient from the beginning of the excavation area. At each of the four locations, soil was collected at the surface, at a depth of 12 inches and at a depth of 24 inches. The soil was composited into a single sample for analysis by 418.1 IR and Modified 8015 Methods. The results are noted below:

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		418.1 IR Method	Modified 8015 Method
	and the second states of the second states in	Search and the second	
•		(mg/l)	(mg/l)
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•••	Sample #1	<20	<5
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¢1	Sample #4	<20	and the second
	Dembre 44	tan an ∖∠u .	
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The average of the four sample results will be used for the background TPH level, if approved by NMOCD. A map showing the area and location of each sample is attached. In addition, a copy of the analytical results are also attached.

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Pundan_ A.N. Pundari

Pundari

cc: S.Miller/File 5216-Mary Wheeler #1E



9830 S. 51st Street Suite B-113 Phoenix, AZ 85044 (602) 496-4400

ATI I.D. 105554

May 16, 1991

El Paso Natural Gas Company P.O. Box 4990 Farmington, NM 87499

Project Name/Number: Mary Wheeler

Attention: John Lambdin

On 05/03/91, Analytical Technologies, Inc. received a request to analyze soil sample(s). The sample(s) were analyzed with EPA methodology or equivalent methods. The results of these analyses and the quality control data, which follow each set of analyses, are enclosed.

If you have any questions or comments, please do not hesitate to contact us at (602) 496-4400.

ijabeth

Elizabeth Proffitt Senior Project Manager

1.0

Sheat V. Woon

Robert V. Woods Laboratory Manager

RVW:clf Enclosure

Corporate Offices: 5550 Morehouse Drive San Diego, CA 92121 (619) 458-9141

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Analyt Analyt	ical lechnologues, Inc.	-
CLIENT PROJECT #	: EL PASO NATURAL GAS, NEW MEXICO : (NONE)	DATE RECEIVED : 05/03/91
	: MARY WHEELER ATI I.D. : 105554	REPORT DATE : 05/09/91
	CLIENT DESCRIPTION	

ATT #	CLIENI DESCRIPTION		DATE CODECTED
01	N10615	SOIL	05/02/91
02	N10616	SOIL	05/02/91
03	N10617	SOIL	05/02/91
04	N10618	SOIL	05/02/91



----- TOTALS -----

MATRIX	
SOIL	

儿

SAMPLES -----4

ATI STANDARD DISPOSAL PRACTICE

The samples from this project will be disposed of in thirty (30) days from the date of this report. If an extended storage period is required, please contactor our sample control department before the scheduled disposal date.



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GENERAL CHEMISTRY RESULTS

	ATI I.D. : 105554
CLIENT : EL PASO NATURAL GAS, NEW MEXICO PROJECT # : (NONE)	DATE RECEIVED : 05/03/91
PROJECT NAME : MARY WHEELER	REPORT DATE : 05/09/91
PARAMETER UNITS 01 02	03 04
PETROLEUM HYDROCARBONS, IR MG/KG <20 40	74 <20

and the first

113



ATI I.D. : 10555401

TEST : FUEL HYDROCARBONS (MODIFIED EPA METHOD 8015)

PROJECT # :	: MARY WHEELER : N10615	DATE RECEIVED DATE EXTRACTED DATE ANALYZED	05/02/91 05/03/91 05/03/91 05/06/91 MG/KG 1
COMPOUNDS		RESULTS	
FUEL HYDROCARBO HYDROCARBON RAN HYDROCARBONS QU		<5 - - -	

SURROGATE PERCENT RECOVERIES

DI-N-OCTYL-PHTHALATE (%)

116





ATI I.D. : 10555402

TEST : FUEL HYDROCARBONS (MODIFIED EPA METHOD 8015)

CLIENT PROJECT # PROJECT NAME CLIENT I.D. SAMPLE MATRIX	XICO DATE SAMPLED : 05/02/9 DATE RECEIVED : 05/03/9 DATE EXTRACTED : 05/03/9 DATE ANALYZED : 05/06/9 UNITS : MG/KG DILUTION FACTOR : 1	1 1
COMPOUNDS	RESULTS	
FUEL HYDROCAR HYDROCARBON R HYDROCARBONS	 <5 - -	-

SURROGATE PERCENT RECOVERIES

3 sacretter

DI-N-OCTYL-PHTHALATE (%)

109





ATI I.D. : 10555403

TEST : FUEL HYDROCARBONS (MODIFIED EPA METHOD 8015)

CLIENT PROJECT # PROJECT NAME CLIENT I.D. SAMPLE MATRIX	: EL PASO NATURAL GAS, NEW MEXICO : (NONE) : MARY WHEELER : N10617 : SOIL	DATE RECEIVED DATE EXTRACTED DATE ANALYZED	: 05/02/91 : 05/03/91 : 05/03/91 : 05/06/91 : MG/KG : 1
COMPOUNDS		RESULTS	
FUEL HYDROCARI HYDROCARBON RA	-	<5	

HYDROCARBONS QUANTITATED USING

SURROGATE PERCENT RECOVERIES

10.00

DI-N-OCTYL-PHTHALATE (%)

Acephilu GI



ATI I.D. : 10555404

TEST : FUEL HYDROCARBONS (MODIFIED EPA METHOD 8015)

CLIENT PROJECT # PROJECT NAME CLIENT I.D. SAMPLE MATRIX	CO DATE SAMPLED : 05/02/91 DATE RECEIVED : 05/03/91 DATE EXTRACTED : 05/03/91 DATE ANALYZED : 05/06/91 UNITS : MG/KG DILUTION FACTOR : 1
COMPOUNDS	RESULTS
FUEL HYDROCAR HYDROCARBON R HYDROCARBONS	 <5 - -

SURROGATE PERCENT RECOVERIES

DI-N-OCTYL-PHTHALATE (%)

141

partial single

Analytical **Technologies**, Inc.

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GENERAL CHEMISTRY - QUALITY CONTROL

CLIENT : PROJECT # : PROJECT NAME :	(/	GAS, NEW	MEXICO	ATI	I.D.	: 1055	54	
PARAMETER	UNITS	ATI I.D.	SAMPLE RESULT	DUP. RESULT		SPIKED SAMPLE		% REC
PETROLEUM HYDE	ROCARBONS MG/KG	10555402	40	48	18	230	190	100

Acceptule!

COPV

% Recovery = (Spike Sample Result - Sample Result) ---- X 100 Spike Concentration RPD (Relative Percent Difference) = (Sample Result - Duplicate Result) Х 101 Average Result

79001-0175



REAGENT BLANK

TEST : FUEL HYDROCARBONS (MODIFIED EPA METHOD 8015)

CLIENT PROJECT # PROJECT NAME CLIENT I.D.	: EL PASO NATURAL GAS, NEW : (NONE) : MARY WHEELER : REAGENT BLANK	ATI I.D. MEXICO DATE EXTRACTED DATE ANALYZED UNITS DILUTION FACTOR	: 105554 : 05/03/91 : 05/03/91 : MG/KG : N/A
COMPOUNDS		RESULTS	
FUEL HYDROCARI HYDROCARBON RJ		<5	
	QUANTITATED USING	-	

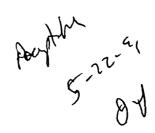
SURROGATE PERCENT RECOVERIES

DI-N-OCTYL-PHTHALATE (%)

100



Analytical Technologies , Inc.	
QUALITY CONTROL DA TEST : FUEL HYDROCARBONS (MODIFIED EPA METHOD	ATI I.D. : 105554
CLIENT : EL PASO NATURAL GAS, NEW MEXICO PROJECT # : (NONE) PROJECT NAME : MARY WHEELER REF I.D. : 10599824	DATE ANALYZED : 05/04/91 SAMPLE MATRIX : NON-AQUEOU UNITS : MG/KG
COMPOUNDS SAMPLE CONC RESULT SPIN	DUP. DUP. C. SPIKED % SPIKED % KED SAMPLE REC.SAMPLE REC. RPD
FUEL HYDROCARBONS <5 50	45 90 45 90 0



% Recovery = (Spike Sample Result - Sample Result) Spike Concentration RPD (Relative % Difference) = (Spiked Sample - Duplicate Spike) Result Sample Result

Average of Spiked Sample

7. . 7.00

DE CONSERVE ON DIVISION REPORTED



191 JUL 25 6月 10 14

P. O. BOX 1492 EL PASO, TEXAS 79978 PHONE: 915-541-2600

July 24, 1991

Mr. David Boyer Environmental Bureau Chief New Mexico Oil Conservation Division P.O. Box 2088 Land Office Building Santa Fe, New Mexico 87504-2088

RE: Remediation at Manana-Mary Wheeler #1E Gas Well Site, Flora Vista, New Mexico

Dear Mr. Boyer

This letter is to provide you an update on the status of the subject project. On July 18, 1991, a job showing was held with contractors to advise them of the Scope of Work and allow an opportunity to get any clarifications needed prior to submitting bids for the work. We anticipate the contract for the remediation project will be awarded on July 29, with work commencing on or about August 5, 1991.

Attached for your files are the following:

- 1. A copy of the Bid Package provided to the potential contractors;
- 2. A copy of a June 10, 1991 letter to the landowner outlining the agreement needed between the landowner and EPNG;
- 3. A copy of a July 12, 1991 memorandum from an EPNG Right of Way representative advising that Mr. Velasquez, son-in-law of the landowner, has agreed to the requests outlined in EPNG's June 10, 1991 letter to Mr. Thurstonson;
- 4. A copy of a July 17, 1991 letter to Mr. Ed Hartman, President of Manana Gas, Inc. advising him of the timeframe for removing Manana's facilities; and,
- 5. A copy of a July 17, 1991 letter to Mr. William F. Carr, attorney for Manana Gas, Inc.

If you have any questions or need additional information please advise.

Very truly yours,

Thomas D. Hutchin's

Thomas D. Hutchins, Manager North Region Compliance Engineering



DIE CONSER, IN DIVISION RECT VED

P. O. BOX 1492 EL PASO, TEXAS 79978 PHONE: 915-541-2600

'9 July 15, 1991 10 14

BID INVITATION

Re: Remediation at Manana Mary Wheeler #1E at Flora Vista, New Mexico

Gentlemen:

El Paso Natural Gas Company invites you to submit your proposal for work as referenced above and more particularly described in Exhibit "A" of the proposal Contract, two (2) copies of which are attached.

Contractors are asked to meet at 9:00 A.M. (MDT) on Thursday, July 18, 1991 at Company's Farmington Office located at 614 Reilly, Farmington, New Mexico.

Your bid is to be prepared in accordance with the proposed Contract Specification and Drawings. With the exception of the executed date, you are to fill in all blanks including Cost Schedule in Exhibit "B" in the Contract. Please properly execute, sign and witness the Contract in the places and return both complete originals.

Contractors are asked to bring and wear a hard hat, safety glasses with sideshields and a "fire retardant" suit for the job showing. El Paso Natural Gas Company will not supply these items!

Project start Date August 5, 1991 and Completion Date August 30, 1991.

All bidders must be licensed and authorized to transact business in the State (if applicable) where the work is to be performed. ALL APPLICABLE TAXES ARE NOT TO BE INCLUDED ON THE PROPOSAL.

Your two (2) executed copies fo the Contract must be received by the undersigned no later than 2:00 P.M. (MDT) on Thursday, July 25, 1991. Proposals not received by that time will not receive consideration.

Gentlemen: July 15, 1991 Page 2

Please clearly mark the envelope: "BID PROPOSAL -REMEDIATION AT MANANA MARY WHEELER #1E".

Mail Bids to: El Paso Natural Gas Company Attention: Mr. Joe A. Fierro, Jr. Contracting 304 Texas Avenue El Paso, Texas 79901

After receiving your proposal, the Company will review the prices and, upon management review and concurrence, enter into a Contract with the successful bidder, if any.

A preconstruction safety meeting will be held with the successful Contractor prior to move-in.

If you have any questions concerning the document, contact the undersigned. Questions concerning the "Scope of Work" should be directed to Anu Pundari, Compliance Engineer, at (505) 599-2176, Farmington, New Mexico.

Yours very truly,

Joe A. Fierro, Jr. Analyst, Contracting (915) 541-5298 FAX (915) 541-3390

JAF:gh Attachment

cc:

- M. L. Ayoub
- G. E. Bauer
- M. L. Gallegos
- A. E. Gilmore
- T. L. Hutchins
- D. M. Kelsey
- A. Pundari
- H. A. Shaffer
- K. L. Steelhammer (w/o attachment)
- L. B. Tinker

File

7-15-91 MAT205 Flint Engineering & Construction Co. Attention: Mr. Bobby J. Brown 2 County Road 5569 Farmington, New Mexico 87401 505 325-5081 2 Bid Invitation-Remediation at Manana Mary Wheeler #1E at Flora Vista, New Mexico VERNIGHT

7-15-91 MAT544 Diamond D Construction, Inc. Attention: Mr. Dennis Fieldsted 36 County Rd. 6500 Kirkland, New Mexico 87417 505 598-5836 2 Bid Invitation-Remediation at Manana Mary Wheeler #1E at Flora Vista, New Mexico OVERNIGHT

7-15-91 MAT600 Taft Construction Attention: Mr. Wesley Taft #32 Road 5219 B Bloomfield, New Mexico 87413 505 632-8822 2 Bid Invitation-Remediation at Manana Mary Wheeler #1E at Flora Vista, New Mexico -OVERNIGHT-

7-15-91 MAT025 Four-Four, Inc. Attention: Mr. B. N. Strunk 3000 E. Bloomfield Highway Farmington, NM 87401 505 327-3500 2 Bid Invitation-Remediation at Manana Mary Wheeler #1E at Flora Vista, New Mexico OVERNIGHT

7-15-91 MAT028 Foutz & Bursum Construction Attention: Mr. Rob Mitchell 3620 E. Main Farmington, NM 87401 505 325-3712 2 Bid Invitation-Remediation at Manana Mary Wheeler #1E at Flora Vista, New Mexico OVERNIGHT

7-15-91 MAT718 Yucca Welding & Excavating Attention: Henry B. Armenta 1901 W. Blanco Blvd. Bloomfield, NM 87413 (505) 632-3728 2 Bid Invitation-Remediation at Manana Mary Wheeler #1E at Flora Vista, New Mexico OVERNIGHT

7-15-91 MAT119 CDK Contracting Company Attention: Bill Etie 800 S. Hutton Road Farmington,NM 87401 505 326-4722 2 Bid Invitation-Remediation at Manana Mary Wheeler #1E at Flora Vista, New Mexico OVERNIGHT

7-15-91 MAT796 ADD TO LIST Envirotech Inc. Attention: Morris D. Young 5796 U. S. Highway 64-3014 Farmington, New Mexico 87401 (505) 632-0615 2 Bid Invitation-Remediation at Manana Mary Wheeler #1E at Flora Vist, New Mexico OVERNIGHT CONSTRUCTION CONTRACT REMEDIATION AT MANANA MARY WHEELER AT FLORA VIST, NEW MEXICO

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REMEDIATION AT MANANA MARY WHEELER AT FLORA VISTA, NEW MEXICO

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Contract No. W/O No. S-9557

CONSTRUCTION CONTRACT

EXCAVATING AND BUCKFILLING AT MANANA MARY WHEELER #1E WELL

THIS CONTRACT, made and entered into this _____ day of _____, 1991, by and between EL PASO NATURAL GAS COMPANY, a Delaware Corporation, whose address is Post Office Box 1492, El Paso, Texas 79978, hereinafter referred to as "Company," and ______, whose address is ______, and who is authorized and licensed to do business in the State of Texas hereinafter referred to as "Contractor";

WITNESSETH:

That for and in consideration of payments, covenants and agreements hereinafter set forth, Company and Contractor do hereby contract, covenant and agree with each other as follows:

PART I

DESCRIPTION OF WORK, PRICE AND SPECIAL PROVISIONS

ARTICLE I. STATEMENT OF WORK

1

Contractor shall excavate and transport contaminated soils to Envirotech's Disposal Site or to Company's facility for disposal. Company will select one of the two sites as the disposal location.

Site of the work is Manana Mary Wheeler #1E Well Site located in Flora Vista, New Mexico.

All such work is more particularly described in Exhibit "A" attached hereto and made a part hereof, and Company and Contractor desire that Contractor shall perform all such work in accordance with drawing and

Company's specifications, all of which Company has furnished to Contractor or will furnish to Contractor in sufficient time to enable Contractor to perform work, and all of which hereinafter referred to as the "Work".

ARTICLE II. SCOPE OF WORK

To complete the work, Contractor shall furnish the following:

A. <u>Construction</u>

(1) <u>Personnel</u>

Contractor shall furnish all necessary superintendents, supervisors, foremen, skilled and unskilled labor, and all other persons.

(2) Construction Equipment. Tools and Supplies

Contractor shall furnish construction tools and equipment, small tools and consumable and expendable supplies necessary to perform the Work in accordance with the diagrams, drawings and Company's Manual of Engineering Standards ("MES") and shall arrange

for transportation to the site of the Work of all such items.

ARTICLE III.

COMPENSATION

In full consideration of the undertakings of Contractor pursuant to the terms of this Contract, Company shall pay Contractor the sum of the following:

Section 1.

As full compensation for the performance of the Work, inclusive of all taxes and other charges, Contractor shall be paid the according with cost schedule in Exhibit "B":

> Work Order No. S-9557 * \$_____ *See Cost Schedule Referenced Above

Section 2. Compensation for Delays in Performance of Work

A. By Contractor

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All delays in the performance of the Work resulting from causes other than those attributable to Company shall be at the cost and expense of Contractor where the Work involves coordination of schedules with other contractors on the site. Contractor shall use its best efforts to minimize delay to the Work, provided, however, that where delay occurs through no fault of the Contractor, such delay shall be either compensable or excused or both, as the case may warrant.

B. By Company

For delays in the performance of the Work attributable to Company, it is agreed that the compensation and/or amounts due Contractor in full and complete settlement of such delays shall be as follows:

- A lump sum settlement, inclusive of all taxes, mutually agreed upon by Company and Contractor; or
- (2) Reimbursement at the hourly rates contained in Exhibit "D" hereof, or if no specific rate is listed in Exhibit "D", at rental rates prevailing in the area in which the work is being performed for all construction equipment and tools owned by Contractor which are placed in "stand-by" status as a result of such delays; and/or

- (3) Reimbursement at actual costs for all construction equipment and tools rented from third parties, inclusive of taxes, which are placed in a "stand-by" status as a result of such delays; provided, however, such actual costs shall not exceed the current rates prevailing in the area where the Work is being performed; and, provided further, that a copy of all such rental agreements are furnished Company; and/or
- (4) Reimbursement in accordance with the wage rates shown in the attached Exhibit "C" for Contractor's labor placed in a "stand-by" status as a result of such delays; and/or
- (5) Reimbursement of costs and expenses, inclusive of all taxes, incurred by Contractor for subcontractors placed in a "stand-by" status as a result of such delays; and/or
- (6) Reimbursement of actual costs, exclusive of all taxes, incurred by Contractor as a result of such delays for which Contractor is not reimbursed by the terms of (2) through (5) above.

Company shall have the right to mitigate the amount of compensation payable by it for delays in the performance of the Work by requiring Contractor to dismiss from the Work such labor, construction equipment and tools, and subcontractors as Company may in its discretion deem necessary.

ARTICLE IV. COMMENCEMENT AND COMPLETION OF WORK

The move-in and completion dates for the Work are as follows:

MOVE-IN DATE:	August 5, 1991
COMPLETION DATE:	August 30, 1991

"Move-In" is the date which Contractor is approved to move onto the project, and "Completion" is the date that all Work is completed to the satisfaction of Company and Contractor moves off the site. ANY DELAYS IN THE PERFORMANCE OF THE WORK attributable to Company shall be added to the term hereof.

Notwithstanding any provisions of this Article IV., as to the Move-In date, it is the understanding of Company and Contractor that Contractor shall not enter upon the premises where the Work is to be performed without the written or oral permission of Company first having been obtained; provided, however, that Contractor shall be reimbursed on a lump sum basis pursuant to Section 2.B (1) of Article III. hereof for any delay by Company which prevents Contractor from moving in on the above said Move-In date.

ARTICLE V. COMPANY'S RESPONSIBILITY

Company shall, at its own expense and at such times as may be required for the successful and continuous prosecution of the Work, furnish the following:

- (1) All materials, equipment, and apparatus described in Exhibit "A" attached hereto.
- (2) All drawings and specifications.
- (3) An area immediately adjacent to the site of the Work sufficient in size to accommodate Contractor's temporary facilities (including office, warehouse, shop building, materials storage, and employee parking). Improvements of the site required to adapt the same to Contractor's use shall be at the expense of Contractor.

In the event of conflict between the terms and provisions of Part I. and II. of this Contract, the terms and provisions of Part I. shall govern.

PART II

GENERAL CONDITIONS

ARTICLE VI. METHOD OF PAYMENT

A. Cost Reimbursement

On or after the first day of each month during the performance of the Work, Contractor shall submit to Company a statement of the costs incurred by Contractor in the performance of the Work for the previous month. Costs incurred in the construction of the Work shall be submitted in accordance with instructions of and on forms requested by Company's Accounting Department. Such statement shall be addressed as follows: El Paso Natural Gas Company, Post Office Box 4990, Farmington, New Mexico 87499, Attention: Anu Pundari, North Region Engineering. Company shall pay such statement amount within twenty (20) days following receipt by Company's South Region Engineering of Contractor's statement. (Separate invoices shall be prepared by Contractor and submitted to Company to cover Changes in Scope of Work items.) Upon approval of such statement by Company's representative, Company shall promptly pay Contractor the statement amount in full until the sum of such payments equals ninety percent (90%) of the lump sum price set out in Article III. hereof. Thereafter Company shall retain an amount equal to ten percent (10%) of such lump price, in addition to retainage, if any being withheld with respect to compensation for Changes in Scope of Work items.

The unpaid balance of the lump price set in Article III, hereof, shall be paid to Contractor when all the following conditions have been met:

- Satisfactory completion of all performance tests described in Article IX. hereof;
- i (2) Acceptance of Work by Company as provided in Article X. hereof; and
 - (3) Receipt of acceptable evidence by Company, including Contractor's affidavit, that all bills for labor, supplies, and equipment and other claims incurred by Contractor in the performance of the Work have been paid.

B. Record Keeping and Company's Audit Rights

Contractor shall maintain books and records of all costs and expenditures incurred by in such detail that all such costs and expenditures may be readily computed and audited. Company shall have the right to inspect and audit such books and records of Contractor for the purpose of determining the accuracy of such costs and expenditures charged to Company for the purpose of answering any inquiries or requirements made by a regulatory authority to Company.

ARTICLE VII. PERFORMANCE OF WORK

Contractor is an independent contractor having full power and authority to select the mode, means, methods and manner of performing the Work, including the selection of sufficient, competent employees, suitable construction equipment and tools, and qualified subcontractors, and shall follow the desires of Company only as to the results to be achieved; subject, however, to the following terms and conditions:

A. Company shall have the right, subject to applicable laws, regulations and agreements in effect at the time, to require the removal from the Work of any employee of Contractor, or its subcontractors, who in Company's opinion is incompetent, careless, unqualified,

insubordinate, or guilty of improper conduct. Contractor agrees to indemnify and hold harmless Company against any liens filed by or other claims of any employee of Contractor so removed.

- B. Contractor shall be responsible for the selection and maintenance of all construction tools and equipment necessary for the proper performance of the Work. Such equipment shall be serviceable, kept in a first-class operating condition and must be satisfactory to Company's Representative. Any equipment that is not satisfactory shall be removed and satisfactory equipment substituted therefor.
- C. Company may perform work not covered by this Contract, perform changes provided for in Article VIII. hereof, perform Work uncompleted in the event of termination of the Contract, and award contracts to others for any part thereof. Contractor shall fully cooperate with Company or with such other contractors and carefully fit its own Work with the work of Company and such other contractors as may be directed by Company. Contractor shall not commit any act which will interfere with the performance of the Work by Company or by any of its subcontractors.
- D. Contractor represents that it has had an opportunity to examine, and has carefully examined, all of the Contract documents and has fully acquainted itself with the Scope of Work, the general topography, soil structure, substructure conditions, obstructions, and all other conditions pertaining to the Work, the site of the Work and its surroundings; that it has made all investigations essential to a full understanding of the difficulties which may be encountered in performing the Work; and that anything in any of the Contract documents or in any

representations, statements or information made or furnished by Company or its representatives notwithstanding, Contractor will regardless of any such conditions pertaining to the Work, the site of the Work or its surroundings, complete the Work for the compensation stated in this Contract, and pursuant to the extent of Contractor's liability under this contract, assume full and complete responsibility for any such conditions pertaining to the Work, the site of the Work or its surroundings, and all risks in connection therewith. In addition thereto, Contractor represents that it is fully qualified to do the Work in accordance with the terms of this Contract within the time specified.

- E. Contractor shall comply with all state and federal laws and regulations pertaining to the hours of employment and rates of pay applicable to the employees of Contractor and all governmental regulations and orders pertaining hereto, in performing the Work covered hereby, and shall comply with any agreements, stipulations orrestrictions applicable to the site of the Work when copies thereof are furnished to Contractor by Company.
- F. Contractor shall properly protect the property of Company and others at the site of or adjacent to the Work. Contractor shall take all necessary precautions for the safety of the employees on the Work and shall comply with all applicable provisions of federal, state, and municipal safety laws to prevent accidents or injuries to persons or damage to property on, or about, or adjacent to the premises where the Work is being performed. Hard hats and safety glasses shall be worn at all times in designated areas. Firearms, alcohol and drugs, and the use thereof, shall be prohibited from any job site.

Contractor specifically agrees to comply, in all respects, to the G. Drug Testing Regulations adopted by the U.S. Department of Transportation that appear at 49 CFR Part 199. Contractor, in addition, understands and agrees that Company shall have the right to audit Contractor's records to verify compliance with these regulations and Contractor acknowledges that Company has retained a Consultant to conduct these audits as requested by Company. Contractor agrees to, upon request by Company, provide access to Contractor's records as required by the Consultant to verify Contractor's compliance with the DOT Regulations. If Contractor's policy and practices do not, in the opinion of the Consultant, comply with the DOT Regulations and if, after reasonable opportunity to correct such deficiencies, Contractor does not comply with such Regulations this Contract may be terminated by Company.

ARTICLE VIII. CHANGES IN SCOPE OF WORK

A. As defined herein, "Change in Scope of Work" shall mean the addition to or elimination from the Work described in Article I., hereof. A "Change of Scope of Work" can only be made following the issuance of a "Contract Change Order," which shall be in writing, signed by Company's Authorized Representative, and delivered to Contractor's Authorized Representative, who shall acknowledge receipt therefor in writing.

Company may at any time by such Change in Scope of Work, without notice to sureties, if any, issued additional instructions, make changes in the specifications and drawings, omit certain Work and require other Work to be performed by Contractor. In any event, Company shall specify the amount and kind of Work to be done or omitted, the materials to be used and the equipment to be furnished.

Where a Change in Scope of Work is authorized, Contractor shall make every effort to provide such additional personnel and equipment to complete said Work within the time specified herein, it being agreed that the date for the completion of the Work shall not be extended unless approved in writing by Company.

Contractor shall make no additions, alterations or omissions, perform no extra work nor supply or use extra materials or equipment of any kind except upon a prior "Contract Change Order" of Company.

- B. If any of the additions, changes, alterations or omissions due to a change in Scope of Work shall increase or decrease the Contract Compensation, such increase or decrease shall be subject to Company's right to retain 10% of billed amounts and shall be in accordance with one or more of the following methods acceptable to Company:
 - Referencing Article III., Section 1., a proposal acceptable to Company excluding of all applicable taxes.
 - (2) Charges for direct labor costs, including applicable insurance, all taxes, overhead and profit based on the schedule of wage rates which is attached as Exhibit "C". Company will not reimburse Contractor for superintendent's time under this item unless such superintendent is on the project solely for the purpose of supervising work which is a result of a "Contract Change Order".
 - (3) Charges for the use of equipment based on a schedule of equipment rental rates which is attached as Exhibit "D". The rental rates shown thereon shall be on a fully maintained basis, exclusive of all operating labor, and shall be considered to expressly include all gasoline, oil, supplies, repair parts, repair labor, all taxes,

license fees, rentals, supervision, overhead and profit and any and all other costs incidental to the use of such equipment and applicable to the extra work involved.

(4) The actual costs, less all discount, plus fifteen percent (15%) for Contractor's overhead, profit and taxes, of materials, supplies and outside services (with the exception of materials, supplies and services included in "Equipment Rates" as set forth in Exhibit "D") furnished by Contractor in connection with extra work performed in accordance with items (2) and/or (3) above.

Charges made for the foregoing terms are to be supported by reports thereof (forms to be furnished by Company) signed by representatives of Contractor and Company and shall be subject to audit by Company.

Company reserves the right to check the labor and equipment time devoted to the charges provided for hereunder by the use of time checkers and equipment checks as Company may deem necessary. In the event Company exercises this right, Contractor shall furnish and make available to such time checkers or equipment checkers complete information and records relating to such labor and equipment time.

ARTICLE IX. ACCEPTANCE OF WORK

When Contractor deems that is has completed all Work hereunder, it shall notify Company in writing that the Work is ready for final inspection and acceptance. Company shall make such inspection within ten (10) days after receipt of such notice. When Company's final inspection has been completed and all defects, if any, have been remedied. Company shall promptly execute and cause to be delivered to Contractor a letter of acceptance stating that, subject to continuing

obligations under Articles IX. and XVII. hereof, Contractor has fully performed all its obligations under the terms of this Contract, and that the Work is accepted by Company as being completed in accordance with the terms and conditions hereof. When accepted, care, custody, and control of the Work shall pass to Company.

ARTICLE X. SUBCONTRACTORS

Contractor shall procure Company's written approval of all subcontractors used by Contractor and of all subcontracts let. Such approval shall not relieve Contractor from any of the obligations of this Contract to Company. No subcontract shall bind or purport to bind Company but shall contain certain provisions permitting the assignment thereof from Contractor to Company. Contractor shall check subcontractor's work and keep such records and furnish reports and information relative to subcontractor as Company may request.

Notwithstanding any of the provisions set forth above, Company agrees that Contractor may subcontract those items set forth in Exhibit "F" attached hereto and made a part hereof.

ARTICLE XI. REPRESENTATIVES OF COMPANY AND CONTRACTOR

Company hereby designates Mr. L. R. Tarver, Vice President of El Paso Natural Gas Company, its "Representative" for the purpose of exercising Company's rights hereunder. In addition to Company's Inspector for the Work, who is hereby designated to act as substitute for Representative, Representative may appoint from time to time another substitute or other substitutes to act in his stead or delegate portions of Company's supervision of the Work to third persons for exercise. Such substitute or delegation shall be made in writing by Representative

to Contractor detailing the rights which may be exercised by the substitute or delegates and shall continue in effect until expiration by the terms of the writing or revocation by the Representative.

Representative shall be the sole, exclusive judge of the interpretation and meaning of the diagrams, plans, drawings, and specifications furnished by Company hereunder when there is doubt as to their meaning, and shall have the right to make reasonable changes in said drawings, plans, diagrams, and specifications, providing that such changes are in writing and are for the purpose of correcting errors therein, including minor design errors.

Representative shall have the right of ingress and egress with respect to the site of the Work and full privileges of inspection and testing for the purpose of insuring that the Work is being performed in accordance with the conditions this terms and of Contract. Representative may require Contractor to conform strictly with the terms and conditions hereof and to make any change or alteration as may become necessary to constitute compliance herewith. Upon Representative's request, Contractor at any time before acceptance of the Work shall remove or uncover such portions of the finished Work as may be directed. After examination, Contractor shall restore said portions of the Work to the standards required by the specifications.

All instructions, requests, and decisions made by Representative, his substitute or delegates in the performance of their duties shall be binding upon Company.

Contractor at all times shall have a designated representative and a superintendent as its agents on the Work. Such representatives and

superintendent shall have full authority to execute the orders or directions of the Company's Representative without delay; shall supply promptly such materials, tools, equipment, and labor as may be required; and shall be furnished by Contractor irrespective of the amount of Work sublet. Contractor's Superintendent shall not be removed from the Work or replaced without written permission by Company.

ARTICLE XII. REPORTS AND ACCOUNTING PROCEDURES

As soon as practicable, Contractor shall prepare and submit to Company a monthly cost estimate and revisions and a completion schedule of all Work it shall perform hereunder. Contractor shall also prepare and submit to Company a monthly progress report showing the status of all construction to be performed by Contractor. In addition, Contractor shall make such other reports and furnish such other information as the Company shall request on forms prescribed by Company. In the preparation and submissions of the foregoing reports and information, Company's procedures shall be followed.

ARTICLE XIII. ACCOUNTING

A. Record Keeping

All record keeping shall be performed by Contractor in accordance with procedures set forth in Company's Field Accounting Procedures for lump sum Contract Plant Construction (copies of which shall be provided to Contractor). All billings from Contractor to Company shall be prepared by Contractor at the site of the Work, itemized, in accordance with Company's Gas Plant Catalog and Plant Accounting Manual. Contractor shall present its billings to Company's Field Auditor, or if no Field Auditor has been assigned to the Work, to Company's Inspector,

who shall secure Company's necessary field approval to same and shall forward such billings to Company for payment.

B. Deficiencies

If at any time during the performance of the Work it becomes the opinion of Company's Field Auditor and Material Inspector that Contractor's record keeping to be performed pursuant to this Article XIV. has become deficient to a critical point, Company's Representative shall give written notice of such deficiencies to Contractor's Representative designated in Article XXII. hereof. If after fifteen (15) days Contractor has not taken immediate steps to correct such deficiencies, Contractor hereby consents and agrees that payments to Contractor of Fixed Fee installments pursuant to Article VI. hereof may be suspended by Company until, in the opinion of Company's Representative, proper action has been taken by Contractor to remedy such deficiencies.

ARTICLE XIV. INSURANCE

A. Contractor and each of its subcontractors hereunder, if any, shall, each at its own expense, obtain insurance as provided below from reliable insurance companies acceptable to Company and authorized to do business in the state in which the Work is to be performed. Each policy shall provide for a waiver of subrogation rights against Company. Company shall be designated as additional insured on each policy required by this contract provision except Workers' Compensation and Employer's Liability policies. Contractor and its subcontractors shall maintain in force during the entire period of this Contract, the following insurance:

- Statutory Workers' Compensation and Employer's Liability Insurance with a limit of not less that \$100,000.
- (2) Comprehensive General Public Liability policy, including crossliability endorsement, covering premises and operations, contractual liability, contractor's protective liability, products and completed operations and, if an exposure exists, elevator liability and/or water craft liability, with Bodily Injury limits of not less than \$500,000 per occurrence, and Broad Form Property Damage Liability Coverage with limits of not less than \$500,000 per occurrence, and ston, ston, coverage for blasting and explosions, collapse, and underground property damage.
 - (3) Comprehensive Automobile Liability Insurance including hired cars and nonowner coverages, with Personal Injury limits of not less than \$300,000 per person and not less than \$500,000 per occurrence and Automobile Property Damage with limits of not less than \$500,000 per occurrence.
 - (4) If using an owned or nonowned aircraft or employing or borrowing an aircraft in connection with the Work performed under this Contract, a combined single limit Bodily Injury and Property Damage Liability coverage shall be maintained, with a limit of liability of not less than \$5,000,000 per occurrence, and Voluntary Settlement Coverage of not less than \$500,000 per passenger seat.
 - (5) Such additional insurance as may be required by Company, the cost of which shall be reimbursed by Company.

B. Before Work is commenced, Certificates evidencing that satisfactory coverage of the type and limits set forth above in Paragraph A. with respect to Contractor are in effect shall be furnished to Company. Such certificates shall be in a form acceptable to Company and shall contain provisions that no reductions, cancellations, or material changes in the policies shall become effective except upon thirty (30) days' written notice to Company; provided, however, that no such reduction, cancellation, or material changes in any policy shall relieve Contractor of its obligations to maintain coverages in accordance with Paragraph A. above.

C. Irrespective of the requirements as to insurance to be carried by Contractor as provided herein, insolvency, bankruptcy, or failure of any insurance company to pay all claims accruing, shall not be held to relieve Contractor of any obligations hereunder.

ARTICLE XV. PERMITS

Permits and licenses necessary for the performance of the Work shall be secured and paid by Contractor. Permits, licenses and easements for permanent structures or authorization of permanent modification thereof shall be secured and paid by Company.

ARTICLE XVI. INDEMNITY

A. Without excluding or limiting in any way Company's right to indemnity and/or contribution for any loss arising from any cause whatsoever growing out of or in connection with the Work, as such right may exist under applicable laws, Contractor agrees to indemnify and hold harmless Company, its parent, subsidiaries and affiliates and all their respective officers, agents and employees (all hereinafter collectively

referred to as the "Indemnitee"), from all and every kind and character of liability, damages losses, costs, expenses, demands, claims, suits, actions and causes of action on account of illness, personal injury or death to employees or any other persons, damage to property of Company or others or other loss or liability arising from any cause whatsoever growing out of or in connection with Contractor's negligent performance of the Work. Further, Contractor, at its own expense, shall defend any demand, claim, suit, action or cause of action brought against the Indemnitee where such demand, claim, suit, action or cause of action arises from any cause for which the Indemnitee may be entitled to be indemnified and held harmless pursuant to this Article XVII., and Contractor shall pay all damages, losses, costs and expenses (including attorneys' fees), growing out of or in connection with such demand, claim, suit, action or cause of action; provided, however, that the Indemnitee shall be entitled to participate in such defense at the Indemnitee's own expense. Notwithstanding the fore going, in no event shall Contractor be liable to indemnify and hold harmless the Indemnitee from any liability, damages, losses, costs, expenses, demands, claims, suits, actions or causes of action arising out of the sole negligence of the Indemnitee.

B. Contractor further agrees that each of its subcontractors performing activities in connection with the Work shall enter into a written contract containing the indemnification and hold harmless clauses in this Article XVII. in favor of the Indemnitee, such contract to be entered into by the subcontractor prior to the performance of any work by the subcontractor.

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ARTICLE XVII. TERMINATION

Should Contractor at any time refuse or neglect to supply a Α. sufficiency of properly skilled workmen or materials of the proper quality or quantity, fail in any respect to prosecute the Work or any portion thereof in an efficient, workmanlike, skillful and careful manner or fail to prosecute the Work with such speed as in the judgment of Company is necessary to complete the same within the time herein specified, fail to comply with any of the terms of this Contract, perform in bad faith, or become insolvent, then in any such event Company may give written notice to Contractor stating the event and, if pertinent, the respect or respects in which Contractor is failing to comply with the terms of this Contract. If Contractor does not remedy such event or failure to proceed with such speed as is required by Company with five (5) days after such notice is given, then Company shall have the right to provide any such labor or materials as may be required and to deduct the cost thereof from any money due or thereafter to become due Contractor under this Contract. Additionally, Company may terminate Contractor's right to proceed with the Work or any part thereof regardless of its stage of completion and without prejudice to any claim that Company may have hereunder. In such case Contractor shall not be entitled to receive any further payment until the Work is finished.

If the unpaid balance of the amount to be paid on the Contract shall exceed the expense of finishing the Work, compensation for additional managerial or administrative services and such other costs and damages as Company may suffer, such excess shall be paid to

Contractor. In such expenses, compensation, costs, and damages shall exceed such unpaid balance, Contractor and it sureties, if any, shall be liable for and shall pay the difference to Company.

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If by reason of labor disputes, strikes, lock-outs, or any similar reason, a shutdown or cessation of Work should occur and continue for as long as fourteen (14) days, then Company at its option, may give Contractor written notice of Company's desire to terminate the Contract and to take over the Work and perform or cause the contracted Work to be performed and completed. If Contractor has failed to resume full performance of the Work within five (5) days after such notice, then Company, at its option, may terminate the Contract insofar as future Work is concerned and take over the Work and itself perform or cause the contracted Work to be performed by others. Company will not interfere with Contractor in fulfilling its obligations, if any, with labor unions.

C. Should conditions arise, which in the opinion of Company make it advisable or necessary to cease Work under this Contract, Company may terminate this Contract upon twenty-four (24) hours' written notice to Contractor, and all amounts due Contractor in full and complete settlement of this Contract shall be the sum of the following:

- (1) Pro rata compensation for the portion of the Contract already performed, adjusted for the amount of any changes which may have been authorized, approved and performed under Article VIII. of these General Conditions prior to the date of termination:
- (2) The net cost of material for which Contractor has made firm contracts it being understood that Company shall be entitled to and receive such material, and

(3) An allowance for the unabsorbed portion of Contractor's moving in and moving out costs. The unabsorbed portion of such costs shall be determined based on the ratio of the Contract Compensation for Work not competed to the total Contract Compensation. Moving costs shall be limited to equipment only and for actual distance moved but in no case for a greater distance than from the Work to Contractor's General Offices.

Upon determination of the compensation above specified, Company shall promptly pay the amount thereof to Contractor upon delivery by it of the evidence required by Section A. (3) of Article VI. of the Contract.

ARTICLE XVIII. FORCE MAJELRE

Neither party hereto shall be liable for any failure to perform the Α. terms of this Contract when such failure is due to "force majeure" as hereinafter defined. The term "force majeure" as used in this Contract shall mean any delay or default in performance hereunder due to any cause beyond the control of the parties and without their fault or negligence, including but not restricted to acts of God or the public, civil disturbances, arrests and restraints by rulers and people, acts, requests or interruptions of the federal, state or local government or any agency thereof, or of any federal, state or local officer purporting to act under duly constituted authority, court orders, present and future valid orders of any governmental authority, or any officer, agency or any instrumentality thereof, floods, fires, acts of the public enemy, wars, storms, strikes, lockouts, or industrial disturbances, interruption of transportation, freight embargoes or failures,

exhaustion or unavailability or delays in delivery equipment or service necessary to the performance of any provision hereof, including inability to secure materials as a result of allocations promulgated by authorized governmental agencies, riots, rebellions, blockade, insurrections, sabotage, inability to secure right-of-way, labor shortages, epidemics, invasions, landslides, lightning, earthquakes, quarantine, restrictions, washouts, explosions, breakage or accident to machinery or lines of pipe, or any other cause, whether of the kind herein enumerated or otherwise, not reasonably within the control of the party claiming "force majeure." Nothing herein contained, however, shall be construed to require either party to settle a labor dispute against its will.

B. In the event either party is unable, wholly or in part, to carry out its obligation under this contract when caused by force majeure, other than the obligation to make payment of money due hereunder, the party claiming a suspension which, by the exercise of due diligence, such party shall not be able to overcome or avoid, then, upon such party's giving notice and full particulars of such cause in writing to the other party as soon as possible after the occurrence of the cause relied on, the obligation of the party giving such notice, so far as it is affected by the cause specified in such notice, shall be suspended during the continuance of any inability so caused by for no longer period, and such cause shall, as far as possible, be remedied with all reasonable dispatch; provided, however, that nothing herein contained shall be construed to limit Company's rights as set forth in Article XVIII. (Termination) hereof.

PERFORMANCE BONDS

If requested by Company, Contractor shall obtain, at Company's expense, and furnish Company within fifteen (15) days after execution of this Contract and prior to commencing work, a performance and payment bond in form and amount satisfactory to Company with surety approved by Company and payable to Company, its successors and assigns, conditioned that Contractor will faithfully perform promptly and with due diligence all of the terms, covenants, and conditions of this Contract on its part to be performed, including all of the undertakings, obligations, and liabilities of Contractor or subcontractors, protect Company against loss by reason of breach of or default in the performance of any obligations Contractor or such subcontractors may have incurred, and protect Company against all claims for the payment of labor and materialbils incurred in the Work by Contractor, respectively. Failure of Contractor to furnish such bond within the time provided shall entitle Company at its option to cancel and terminate this Contract.

ARTICLE XX. LIENS

Contractor shall indemnify and hold Company harmless from all costs, damages, or expenses arising out of any charge or encumbrance in the nature of a laborer's, mechanic's, or materialman's lien in connection with the Work as performed by Contractor or its subcontractors.

ARTICLE XXI NOTICES

All notices, orders, statements, reports, or other correspondence required or made necessary by the terms of this Contract shall be in writing and shall be considered as having been given (a) to Company if

delivered permanally to its designated Representative at the site of the Work with authority to act for it and if mailed by registered mail, postage prepaid, to Mr. L. R. Tarver, Vice President, El Paso Natural Gas Company, Post Office Box 1492, El Paso, Texas 79978, or (b) to Contractor if delivered personally to its designated Representative at the site of the Work and if mailed by registered mail, postage prepaid, to:

of

ARTICLE XXII. NON-WAIVER OF TERMS OR CONDITIONS

Any failure by Company at any time or from time to time, to enforce or require the strict keeping and performance of any of the terms or conditions of this Contract shall not constitute a waiver of such terms or conditions and shall not affect or impair such terms or conditions in any way or the right of Company at any time to avail itself of such remedies as it may have for any breach or breaches of such terms or conditions.

ARTICLE XXIII. CONFLICTS

In the event of conflict between the terms of this Contract and the Exhibits attached hereto, such conflict shall be resolved by applying the following priorities:

- (1) Exhibit "A"
- (2) Exhibit "B"

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- (3) Exhibit "C"
- (4) Exhibit "?" G

(5) The body of the Contract.

(6) The remaining Exhibits.

ARTICLE XXIV. DISPUTES

If, at any time, there is a difference of opinion between the parties hereto in regard to their respective rights, duties, and obligations, under such provisions of this Contract, the question or questions in dispute shall be referred to Mr. L. R. Tarver, Vice President of El Paso Natural Gas Company, whose decision shall be binding upon Company.

ARTICLE XXV. (GOVERNMENTAL AUTHORITY

In the performance of the Work, Contractor shall comply with all federal and state laws and all rules, regulations, and orders of federal and state agencies having jurisdiction in the premises.

ARTICLE XXVII. ASSIGNMENT

All covenants and agreements herein contained shall be extended to and be binding upon the successors and assigns of Contract and Company. Contractor shall not assign this Contract or any moneys to become due hereunder without prior written consent of Company; provided, however, that no conveyance or transfer of any interest of Contractor shall be binding upon Company until Company has been furnished with written notice and true copy of such conveyance and transfer. Company shall at all times have the right to assign all or any part of its interest in this Contract.

ARTICLE XXVII. TITLE

The title to all Work completed and in the course of construction at the site of the Work and of all materials furnished by Company,

irrespective of the location thereof, as between Company and Contractor or Contractor's subcontractor, shall be in Company.

ARTICLE XXVIII. SURPLUS MATERIALS

In the performance of work of this nature, there will be certain overages of materials. All such overages will belong to Company. Contractor, if requested by Company, will dispose of any such overages and will credit the proceeds of any such sales to the Work.

ARTICLE XXIX. NONDISCRIMINATION

Contractor agrees as follows:

- (1) In the event that the amount of this Contract equals or exceeds \$10,000, Contractor agrees that, unless this Contract or the Work performed hereunder is exempt under Executive Order 11246, as amended, or under the rules and regulations issued thereunder, there are hereby incorporated by reference the provisions of Section 202 of Executive Order 11246, as amended, codified in 41 C.F.R. Section 60-1.4.
- (2) In the event that the amount of this Contract equals or exceeds \$50,000, and Contractor has fifty (50) or more employees, Contractor agrees that, unless this Contract or the Work performed hereunder is exempt pursuant to Executive Order 11246, as amended, or under the rules and regulations issued thereunder, the following terms and conditions shall be applicable during the performance of this Contract:
 - (a) Contractor acknowledges that it is required to file Standard Form 100 (EEO-1), pursuant to 41 C.F.R. Section 60-1.7, within

thirty (30) days after the Contract award, unless such a report has been filed by Contractor within 12 months preceding the date of the award, and to otherwise comply with or file such compliance reports as may be required under Executive Order 11246, as amended, or under the rules and regulations issued thereunder.

- (b) Contractor further acknowledges that it is required to develop

 a written affirmative action compliance program, pursuant to
 41 C.F.R. Section 60-1.40.
- (3) In the event that the amount of this Contract equals or exceeds \$10,000, Contractor further agrees that, unless this Contract or the Work performed hereunder is exempt under 41 C.F.R. Part 60-1 or 41 C.F.R. Part 60-250, as appropriate, the following terms and conditions shall be applicable:
 - (a) During the performance of this Contract, Contractor certifies that it does not and will not maintain any facilities it provides for its employees in a segregated manner, or permit its employees to perform their services at any location, under its control, where segregated facilities are maintained; and that it will obtain a similar certification in the form approved by the Director of the OFCCP, prior to the award of any nonexempt subcontract.
 - (b) There are hereby incorporated by reference the provisions of
 41 C.F.R. Section 60-250.4, pertaining to affirmative action for veterans.

- (4) In the event that the amount of this Contract equals or exceeds \$2,500, Contractor further agrees that, unless this Contract or the Work performed hereunder is exempt under 41 C.F.R. Part 60-741, there are hereby incorporated by reference the provisions of 41 C.F.R. Section 60-741.4, pertaining to affirmative action for handicapped persons.
- (5) Upon the request of the Company, Contractor shall provide the Company with copies of any or all plans and programs which Contractor uses to satisfy the requirements of the preceding paragraphs.

ARTICLE XXX. GOVERNING LAWS

Contractor and Company agree that the laws of the State of Texas shall govern the interpretation and application of this Contract and the rights and obligations of the parties hereunder.

ARTICLE XXXI. SEVERABILITY OF PROVISIONS

Contractor and Company agree that, in the event any provision (or portion thereof) of this Contract is determined to be invalid for any reason, such invalidity shall not affect the validity of any remaining language which can be given effect without the invalid provision. IN WITNESS WHEREOF, the parties hereto have executed this Contract in several counterpart originals as of the day and year first above written.

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EXHIBIT "A"

SCOPE OF WORK

FOR REMEDIATION PROJECT AT

MANAMA MARY WHEELER #18 WELL SITE

LOCATED IN FLORA VISTA, NEW MEXICO

1.0 BACKGROUND

Manana Gas operates a producing natural gas well, Mary Wheeler #1E, in Flora Vista, New Mexico. The alluvial aquifer became contaminated with hydrocarbon liquids. Figure 1 is a map of the site.

The water table at the site is approximately five to six feet from grade. Due to the shallow water table, hydrocarbons migrated with the water and contaminated an area approximately 100 feet by 260 feet. The New Mexico Oil Conservation Division (NMOCD) requested EPNG to implement a plan to remediate the site. The approximate area to be remediated is indicated on Figure 1.

2.0 INTRODUCTION

EPNG desires to contract for the excavation of contaminated soils and transportation of the excavated soils. EPNG would like the contractor to bid on transportation of excavated soils to: (1) Envirotech's disposal site; or (2) to EPNG's facility for disposal. After reviewing the costs for both sites, EPNG will select one of the two sites as the disposal location.

2.1 Ingress and Egress to the Work Site

The well, Mary Wheeler #1E is located in Flora Vista, New Mexico. The site may be reached by going east from Farmington on Highway 550. Then, take a right at the Flora Vista stop light onto County Road 3500. After approximately 3/4 mile, take a left onto Road 3490. The turnoff from County Road 3500 is before the Animas River Bridge. Road 3490, a 1/2 mile dirt road, leads to the site.

3.0 REMEDIATION SCOPE OF WORK

The Scope of Work describes the work necessary to remediate the site. This section is comprised of the following :

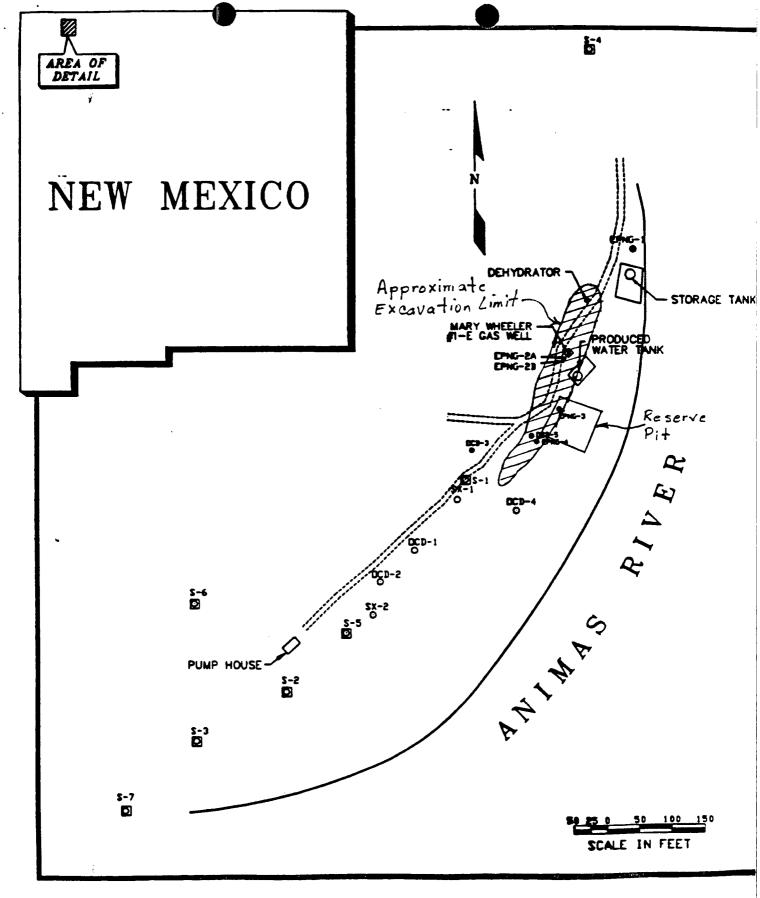


Figure 1

MANANA-MARY WHEELER #1-E Well Site Page 2 Scope of Work

3.1 Soil Excavation and Disposal

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3.1.1 Clean Overburden Excavation and Stockpiling

3.1.2 Contaminated Soil Excavation

- 3.1.3 Saturated/Wet Soil
- 3.1.4 Laboratory Support
- 3.1.5 Backfill
- 3.1.6 Observation Trench
- 3.1.7 Soil Disposal

3.2 Groundwater Disposal

3.3 Other Issues

3.1. EXCAVATION

3.1.1 Clean Overburden Excavation and Stockpiling

A "strip mining" approach will be used for the project, whereby the top three to four feet of clean overburden is removed and stockpiled adjacent to the area prior to excavation of the visibly stained underlying sediments. The clean overburden shall be used for backfill. In the reserve pit area, the top four feet is not considered "clean" and must be excavated and disposed.

Only clean overburden shall be stockpiled for backfill. Any stained soil encountered during excavation must be taken to the disposal location.

3.1.2 Contaminated Soil Excavation

The site can be subdivided into three different excavation areas.

Area I - Reserve Pit

The first area is the reserve pit, an area approximately 70 feet by 70 feet. The reserve pit should be excavated from grade to a depth of approximately six feet.

Area II - "Highly" Contaminated Area

The second area is the highly contaminated area approximately 100 feet by 50 feet. The depth of excavation may range from six to eight feet. Results from field laboratory analyses, discussed in Section 3.1.4, will determine the final excavation depth.

The contaminated soil must be transported to the selected disposal site, EPNG'S Ballard Plant or to Envirotech's facility. It is estimated that approximately two to four feet of contaminated soil will be excavated. Water may collect in the excavation Page 3 Scope of Work

trench. A portable pump to be furnished by contractor will be needed during soil excavation to pump any water encountered from the excavation trench into water trucks or "frac" storage tanks furnished by contractor.

Area III - "Less" Contaminated Area

The third area is the "less" contaminated area of approximately 15,700 square feet. This area will most likely be excavated to a depth of six feet. A final excavation depth will be determined by field soil analyses during the excavation.

The contaminated soil must be transported to the selected disposal site. It is estimated that approximately two feet of contaminated soil will be excavated. Water may collect in the excavation trench. A portable pump to be furnished by the contractor will be needed during soil excavation to pump any water encountered from the excavation trench into the contractors water trucks or "frac" storage tanks.

3.1.3 Saturated/Wet Soil

If groundwater is encountered during excavation, wet soil must be placed into the dump trucks. Oil and water drainage from the excavated soil must be intercepted prior to the discharge onto the ground. Wet soil loaded directly into the trucks must be drained prior to the soil being transported to the disposal location. Oil and water must not discharge on-site or enroute to the disposal location.

To collect oil and water drainage, EPNG will provide troughs which will be placed underneath the trucks. The truck's tailgate will be pinned and the bed tilted over the trough until all residual water is removed from the soils. This is estimated to take approximately 15 minutes. The contractor will be responsible for pumping collected liquids from the troughs to frac tanks or water trucks.

The contractor may offer other options for removing the water prior to transportation of the soil.

3.1.4 - Laboratory Support

NMOCD requires EPNG to collect soil samples at the bottom of the excavation area. EPNG will conduct field total petroleum hydrocarbon (TPH) analysis. Although a specific sampling plan has not been determined, if a 50 foot long, 3 to 5 foot wide trench is excavated, approximately four to five samples will be needed. On average, it will take approximately 45 minutes to analyze the four to five samples. The contractor must assist EPNG by providing the use of the backhoe for collection of soil samples.

The contractor **will mot** be allowed to backfill the site until the TPH is below the limits set by NMOCD. The contractor will be required to excavate until the TPH is below NMOCD limits or to a depth of eight feet. If the TPH is above NMOCD limits and the excavation depth is at eight feet, EPNG will require the contractor to keep the excavation open until clearance for closure is given by NMOCD. The NMOCD may be at the site and request excavation beyond the eight foot depth at some locations.

3.1.5 Backfill and Compaction

Two different types of backfill are required at the site. Riverstone cobble and coarse grained sandy backfill (Type I) is required for all areas below a depth of four feet from grade. The reason for the cobble backfill, similar to the existing soil, is to decrease the probability of future settling in the area. Clean, sandy backfill (Type II) is required for the top four feet of the entire excavated area.

Area I - Reserve Pit - 70 foot by 70 foot by 6 foot depth

The contractor is responsible for furnishing riverstone cobble and coarse grained sandy (Type I) backfill for the bottom two feet of the reserve pit area. The top four feet of the excavated area should be clean sandy backfill (Type II).

Area II - "Highly" Contaminated Area - 5,000 square foot by 8 foot depth

The top four feet of the excavation area can be reused as clean backfill. Assuming a four foot contaminated soil depth, the contractor is responsible for furnishing four additional feet of riverstone cobble and coarse grained sandy (Type I) backfill in the excavation area.

Area III - "Less" Contaminated Area - 15,700 square foot by 6 foot depth

The top four feet of the excavation area can be reused as clean backfill. The contractor is responsible for furnishing two feet of clean sandy backfill (Type II) for the top two feet of the excavation area.

EPNG will provide a borrow site near Ballard Plant for clean sandy backfill if Ballard is chosen as the disposal site. The contractor will be responsible for loading, transporting and placing backfill in the excavated area. Page 5 Scope of Work

If Envirotech's site is chosen, Envirotech has agreed to provide and load clean sandy backfill at a cost of \$1.98 per cubic yard to be paid by the contractor. The contractor will be responsible for transporting and placing backfill in the excavated area.

The contractor is responsible for providing the cobble, coarse grained sandy backfill. The contractor must provide a representative sample and identify the borrow source of the riverstone cobble backfill. The top two feet of Area I,II and III must be machine compacted.

3.1.6 Trench South of Contaminated Soil Excavation Area

A temporary trench must be constructed approximately 20 feet downgradient of the southwestern most area of excavation to serve as an observation trench. The trench would provide visual verification that no floating hydrocarbons escaped the excavation. It will also allow natural volatilization of any remaining dissolved hydrocarbons. The trench will be approximately 75 feet in length, three feet wide and five feet in depth or until groundwater is reached. The trench will be kept open for a minimum of 60 but not more than 120 days.

The contractor is responsible for excavating the trench and providing and installing a temporary six foot high fence around the trench with a three foot wide locked gate. In addition, the contractor is responsible for installing and operating an agitator device in the trench. The agitator and fence can be removed after completion of the project.

The agitator will be a small portable air compressor attached to a perforated hose. The agitator will assist in volatilization of any hydrocarbons. The air compressor must be equipped with a dry element filter that prevents any oil, water or combustion by products from entering the air discharge stream.

3.1.7 Contaminated Soil Transport to Disposal Site

The contractor will transport the contaminated soil to the selected disposal site. EPNG will consider the following two disposal sites. The first disposal site is an area immediately east of EPNG's Ballard Plant, located approximately 45 miles from the excavation site. To get to the plant, the contractor must go south on Highway 44 and turn east just before Huerfano Mesa on County Road 7425. An approximately nine mile dirt road, County Road 7425, leads to the plant.

The second site is Envirotech's facility near Bloomfield. The facility is approximately 26 miles, of paved road, from the remediation site. Envirotech has quoted a price of \$12.80 per

Page 6 Scope of Work

cubic yard for contaminated soil disposal. Envirotech would be responsible for discing the soil.

If the Ballard site is chosen, the contractor must be spread out at the disposal site in six inch lifts. In addition, the contractor must disced the soil two times per week for the duration of the remediation project.

3.2 GROUNDWATER DISPOSAL

The excavation site is near the Animas River. The Animas River may be at a "low level" this year. Therefore, the water table may be deeper than six feet from grade and the contractor may not have to dispose of any groundwater.

However, if the water table is higher than the excavation depth, the contractor must pump the water from the excavation trench into waiting water trucks and/or "frac" storage tanks. The amount of water that will need to be pumped out is difficult to accurately determine due to unknown aquifer characteristics.

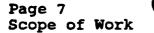
The best estimates of groundwater disposal ranges from 300,000 to 500,000 gallons of water for the project. The actual groundwater disposal quantities may be greater or less than the estimates.

The contractor is responsible for assuring that the water trucks and "frac" storage tanks are adequately steam cleaned prior to project use. The contractor is also responsible for obtaining pumps for groundwater pumping during excavation.

The contractor must haul water to the lined pond at EPNG's Kutz Plant. Kutz Plant is approximately ten miles from the project site. If the Kutz lined pond is filled to capacity, the contractor must truck the water to Southwest Disposal, approximately 20 miles from the project site.

Although it is not anticipated that large volumes of hydrocarbons will be liberated during excavation, the contractor must have absorbent pads on site. The pads will be used to capture any free-phase hydrocarbons that may be in the excavation trench. EPNG will provide drums for the used absorbent pads. In addition, EPNG will dispose of the pads.

In addition, the contractor must have an absorbent "sweep" at the Kutz lined pond. The "sweep" will be used to move, capture in the pond, remove any freephase hydrocarbons from the surface of the pond. The hydrocarbons phase must be trucked to EPNG's Blanco Plant near Bloomfield. At the end of each working day, the contractor is responsible for assuring that the pond surface is free of any oil.



The "sweep" and pads are available from Moore Engineering and Sales, Inc. of Englewood, Colorado. Two absorbent 3M "sweeps" or equivalent, T-126, which is 17" X 100' X 3/8" is adequate for the project. One "sweep" costs \$86.40 plus shipping. One bale of absorbent 3M pads or equivalent, T-156, which is 17"X 19"X 3/8" is adequate for the project. One bale, 100 pads, costs \$68.55 plus shipping.

3.3 OTHER ISSUES

Prior to project startup, EPNG and Manana Gas will remove the well from service. The meter house, meter run, separator, fiberglass storage tank and fencing around the separator area will be removed.

Contractor working hours are limited to Monday through Friday from 7 a.m. to 5 p.m. All efforts should be made to close all excavations by the end of the working day. If excavations are left open overnight, contractor shall install as a minimum, appropriate safety barriers i.e. saw horse with lights, red tape, temporary fence, around the open excavation.

The project site is on private property. The landowner lives in a house near the project site. The existing road to the house is in the excavation area. The contractor will be responsible for constructing and maintaining a temporary road (approximately 400 feet) around the project area to provide access for the owner.

In order to allow access to the site, an alternate route, adjacent to County Road 3490, will be utilized. A section of wooden fence(approximately 15 feet) must be removed and reinstalled by the contractor to allow access to the site along the alternate route. In addition, the contractor is responsible for assuring the integrity of an existing six inch irrigation pipe underneath the access road and preventing damage due to heavy vehicular traffic.

The contractor must exercise extreme care to ensure that private property is not damaged during the execution of this project. EPNG will not compensate the contractor for any property damage due to contractor negligence.

The contractor must purchase/rent good quality pipe, fittings, pumps and other water handling hardware. The contractor must utilize good housekeeping practices so that contaminated water does not leak or spill onto clean soil. If the contractor is negligent and allows contaminated water to discharge onto clean soil, the contractor is responsible for the cleanup of contaminated soil at the contractors expense. The discharge of any water into the Animas River is strictly prohibited. The contractor shall provide general cleanup after completion of the project.

All trucks with contaminated soil and clean backfill must be weighed at Doug Foutz Construction's scale near Crouch Mesa Road or an alternate scale approved by EPNG's Project Engineer. The turnoff from Crouch Mesa Road to the scale is approximately 3-1/2 miles from the site. An additional 3-1/2 mile road leads to the scale. The scale is on the way to either Ballard Plant or Envirotech. Mr. Doug Foutz has agreed to a \$3.00 fee to weigh each truck. The fee in the contractor's bid shall be included.

The contractor is required to follow New Mexico Motor Transportation Division and Department of Transportation rules regarding truck weight limitations and other applicable transportation rules.

All work performed by the contractor shall conform to applicable industry codes and standards and the EPNG Manual of Engineering Standards. The EPNG project engineer will have sole authorization to make changes in the material or specified procedures. Field and approved by the project engineer. All excavation activities must be in accordance with applicable OSHA and EPNG standards. No changes shall be made without the written approval of the Project Engineer.

The contractor shall be responsible for providing the required protection or security for equipment or materials on the job site. EPNG will not assume any liability for losses of materials or equipment.

The contractor must provide a list of all subcontractors with the bid. The contractor must receive written approval from EPNG before using subcontractors for any portion of the work. Such approval will not relieve the contractor of any obligations with EPNG.

4.0 COST SCHEDULE

The contractor must provide costs as required by Figure 2.

Page 9 Scope of Work

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EXHIBIT "B"

FIGURE 2

COST SCHEDULE

1.	Soil Excavation/to (Includes a temporary fencing and assistance in sampling)		
2.	Soil Hauling,Spreading and Discing at Ballard Plant/ton (Include \$3 scale fee.)		
3.	Soil Hauling and Disposal at/ton Envirotech (Include \$12.80/cubic yard Envirotech fee) (Include \$3 scale fee)		
4.	Clean Riverstone Cobble Backfill (Includes cost of material,loading,transportation, \$3 scale fee and placement.)		
5.	Clean Sandy Backfill/ton from Ballard Plant (Includes cost of loading,transportation \$3 scale fee and placement.)		
6.	Clean Sandy Backfill from Envirotech (Include \$1.98/cubic yard material and loading fee, transportation \$3 scale fee and placement. Envirotech will load the oil.)		
7.	South Trench Excavation & Agitator(Lump Sum) (Includes fencing)		
8.	Haul Water to Kutz Lined Pond(per barrel)		
9.	Haul Hydrocarbon Liquids to Blanco Plant(per barrel)		
10.	Haul Water to Southwest Disposal(per barrel)		
11.	Two Oil Absorbent "Sweeps" (Lump Sum)		
12.	100 Oil Absorbent Pads(Lump Sum)		
13.	Temporary Road Construction (Lump Sum)		
14.	Fence Removal and Reinstallation (Lump Sum)		

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Page 10 Scope of Work

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Please Note: One barrel is equivalent to 42 gallons

- : It is difficult to accurately determine the weight of the excavated soil but the following are our best estimates.
- : Clean Backfill is approximately 1.35 tons/cubic yard
- : Moist/Wet Cobble Backfill is approximately 1.75 tons/cubic yard : Dry Cobble Backfill is approximately
- 1.50 tons/cubic yard

EXHIBIT "C"

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RATE SCHEDULE

* Equipment rates shall be on a fully maintained basis, inclusive of all taxes, exclusive of operating labor, in accordance with Exhibit "B".

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EXHIBIT "D"

BOUIPMENT RATE SCHEDULE

EXHIBIT "E"

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 SAFETY MANUAL



EXHIBIT "F"

LIST OF SUBCONTRACTOR

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11. E

EXHIBIT "G"

PROJECT SHOWING NOTES

THE FOLLOWING CHANGES, ADDITIONS. OR CORRECTIONS WERE MADE IT THE SHOWING. ALL CONTRACTOR'S, INSPECTORS AND INTERESTED EPNG PERSONNEL WERE MADE AWARE OF THE CHANGES AT THE SHOWING. THIS DOCUMENT IS TO BE MADE PART OF THE CONTRACT.



P. O. BOX 1492 EL PASO, TEXAS 79978 PHONE: 915-541-2600

July 22, 1991

Re: Project Showing Notes - REMEDIATION AT MANANA MARY WHEELER #1E AT FLORA VISTA, NEW MEXICO

Gentlemen:

Enclosed are "Project Showing Notes" for the reference project. These notes are to be placed under Exhibit "G" and will become a part of the Contract.

If you have any questions, please call the undersigned at (915) 541-5298.

Sincerely,

Joe A. Fierro, Jr. Analyst, Contracting

JAF:gh Attachment

cc:

Messrs.: M. L. Ayoub G. E. Bauer M. L. Gallegos A. E. Gilmore T. L. Hutchins D. M. Kelsey A. Pundari H. A. Shaffer K. L. Steelhammer (w/o attachment) L. B. Tinker Ms.: M. L. Gallegos

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EXHIBIT "G"

PROJECT SHOWING NOTES

THE FOLLOWING CHANGES, ADDITIONS, OR CORRECTIONS WERE MADE AT THE SHOWING. ALL CONTRACTOR'S, INSPECTORS AND INTERESTED COMPANY PERSONNEL WERE MADE AWARE OF THE CHANGES AT THE SHOWING. THIS DOCUMENT IS TO BE MADE PART OF THE CONTRACT.

Re: Soil remediation at Manana Mary Wheeler #1E at Flora Vista, New Mexico

1. Section 3.1.6. - Trench south of contaminated soil excavation area

The trench must be kept open for 60 days. The agitator must be operating from 7 A. M. till one hour after excavation. (example: excavation ends at $^{-}4$ P. M. each day, the agitator must be running until 5 P. M. each day)

 Section 3.1.7. - Contaminated soil transport to disposal site

Company will obtain a separate contract with Envirotech if their disposal site is selected.

If the Ballard Plant site is chosen, Contractor will be responsible for utilizing proper equipment in order to aerate the soil two times per week for the duration of the remediation project (i.e. duration of excavation and project cleanup activities). The contaminated soil contains various sizes of riverstone cobbles.

3. Section 3.3 - Other issues

A) Company has sent a letter to Manana Gas, Inc. asking them to remove all underground piping in the excavation area. If Manana does not remove the piping, Contractor must take every precaution to protect the integrity of their lines.

B) The landowner has agreed to allow trucks and vehicles to travel in an area which is approximately 60 feet west of the meter house. This will allow additional area for vehicle traffic.

C) The temporary road around the project area should be immediately west of the 60 - foot easement mentioned above. The road should be smooth and free of riverstone cobbles. There should be a small, approximately six to twelve inches depth, trench on the west side of the road to allow water drainage from the landowner's irrigation system. In addition, the Contractor shall provide and place a small section (approximately 10 - foot) of four inch PVC pipe to allow Showing Notes Page Two

drainage from the west side of the temporary road to the east side.

D) If the four inch PVC drainage pipe on the existing road to the landowner's house is damaged, Contractor will be responsible for replacing the section of pipe.

E) Contractor is not required to excavate within the small fenced area around the wellhead.

F) Contractor must assure that dust from the roads is not a nuisance to nearby property owners and must follow the guidelines of the on-site Company inspector or at a maximum, water the roads three times per day.

- 4. Figure 2 Cost Schedule
- Item 1. Temporary fencing means any fencing around open excavations
- Item 3. Do not include the \$12.80/cubic yard Envirotech fee, if their disposal site is elected Company will negotiate a separate Contract with Envirotech.

ADD:

Item 15. Portable Scale for project (Lump sum)_____ (If a portable scale is not used, include the \$3.00 scale fee in your cost estimate.)





P. O. BOX 4990 FARMINGTON, NEW MEXICO 87499 PHONE: 505-325-2841

June 10, 1991

Mr. Richard G. Thurstonson P.O. Box 400 Flora Vista, New Mexico 87415

Re: R/W 89320 Soil Contamination: Manana Gas-Mary Wheeler No. 1E San Juan, Co., New Mexico

Dear Mr. Thurstonson:

El Paso Natural Gas Company(EPNG) is working closely with New Mexico Oil Conservation Division(NMOCD) to remediate the Mary Wheeler No. 1E site. Work on this site is scheduled to begin in late July or early August of this year and should be completed with as little inconvenience to you as possible.

Last year, EPNG discussed the project with your son-in-law, Mr. Eli Velasquez. We discussed the extent of excavation, access to the site, discharge of water onto the field and other project topics. Due to unforeseen problems, the project had to be delayed until this year.

EPNG needs to reach an agreement with you and your son-in-law regarding access to the site, discharge of water and other project topics. A contractor will be hired to excavate and remove contaminated soil. The contractor working hours will be limited to Monday through Friday from 7 a.m. to 5 p.m.. All efforts will be made to close all excavations by the end of the working day. If excavations are left open overnight, the contractor will install appropriate safety barriers around the excavation.

Since the existing road to your son-in-law's house is within the excavation area, the contractor will be responsible for constructing and maintaining a temporary road around the project area. The location of the road will be based on your judgment and approval. As discussed with Mr. Eli Velasquez, in order to allow access to the site, a section of wooden fence on the west side of the property will need to be removed and later replaced after project completion. Mr. Thurstonson June 10, 1991 Page 2

After soil remediation activities, NMOCD requested EPNG to pump groundwater from Well S-1. If approved by you, EPNG wants to temporarily place frac tanks for groundwater storage near Well S-1. Then, the water will be analyzed. If the sample results are less than Water Quality Control Commission(WQCC) limits for benzene, ethylbenzene, toluene, xylene(BETX) and 1000 ppm for Total Dissolved Solids, EPNG asks your permission to discharge the water onto your property either to the west or south of the remediation site.

Attached is a detailed explanation of the project.I will work with you to reach an agreement on this project. If you have any questions, please call me at 599-2253.

Very Truly Yours,

onard Stort

Leonard I. Lord Senior Right of Way Negotiator

cc: Mr. Eli Velasquez

bc: Gail Bauer Tom Hutchins Anu Pundari Jamye B. Ward S. D. Miller/File 5216-Mary Wheeler #1E

REMEDIATION PLAN FOR MANANA - MARY WHEELER #1E

The following outlines the procedures and plan for the remediation activities at the subject well site.

1. The reserve pit and the area of soil visibly stained by hydrocarbon will be excavated. The area was delineated during the site investigation study. A "strip mining" approach will be used, in which clean overburden is removed and stockpiled prior to excavation of the underlying stained soil.

2. The excavation will begin up gradient. Overburden to a depth of four to five feet, or until hydrocarbon stained soil is encountered, will be removed and stockpiled for reuse as final backfill after the remediation is completed. Only clean overburden will be stockpiled and used for backfill. Any soil mixed with stained soil will not be used for backfill.

3. Before commencing the excavation activities, El Paso will remove all of its ancillary equipment from the site. After the remediation is complete, El Paso will reinstall its equipment.

4. In order to assure worker safety, access to the site and a thorough and final cleanup, Manana Oil will perform the follow-ing:

a. Shut-in the well for the duration of the excavation activities;

b. remove their ancillary production units (e.g., the separator and storage tank) prior to excavation activities

5. Any above or below ground tanks installed by Manana Oil after the remediation is complete will be equipped with OCD approved leak detection equipment.

6. To augment the clean overburden that will be placed back into the excavation area, a borrow site will be located where clean, coarse-grained sand can be obtained.

7. A small amount of groundwater will probably be encountered. All water observed to have a hydrocarbon sheen will be removed during excavation. A portable pump or vacuum truck will be used to pump the water from the excavation area during soil removal. Page 2 - Remediation Plan - Manana - Mary Wheeler #1E

8. In the event it is necessary to excavate below the water table, the saturated soils will be placed into dump trucks. Each dump truck tailgate will be pinned and the bed tilted over a trough designed to collect and separate the hydrocarbon/water mixtures that drain from the saturated soil. This procedure will result in effective drainage of the wet soils and avoidance of any leakage during transport. The collected water will be trucked to the lined pond at Kutz Plant or a permitted disposal site.

9. It is not anticipated that large volumes of hydrocarbons will be liberated. However, absorbent pads will be available on site to facilitate the capture of any free-phase hydrocarbons.

10. The water from the excavation area will be pumped into water trucks and hauled to the Kutz lined pond or a permitted disposal site.

11. Hydrocarbon contaminated soil will be removed and spread in six inch lifts on EPNG property approved by NMOCD.

12. A temporary trench will be constructed just down gradient of the excavation area to serve as an observation trench. An agitator will be installed in the trench to act as a final treatment "air stripper" and aeration system. The agitator will be kept in use for a short period (60 - 90 days) after soil remediation is complete to allow flushing of the replacement soil. Water samples will be collected at the beginning of the excavation, at seven day intervals during excavation or if a hydrocarbon sheen is observed and at thirty day intervals after remediation until the trench is closed.

13. All former NMOCD monitoring wells will be removed. In addition, all EPNG monitoring wells in the excavation area will be removed.

14. After remediation, well S-1 will be pumped at a constant rate of about 65 gallons per minute (gpm) for 24 hours and sampled. Visual observations of the color, odor, and turbidity of the water will be noted at the beginning and end of the 24 hour The samples will be analyzed for BETX and TDS. period. The water discharged during the 24 hour pumping will be contained in tanks until the sample results are received. If the sample results are less that the WQCC limits for BETX and 1000 ppm for TDS and EPNG receives approval from the landowner, the water will be discharged on to either the west or south of the remediation site. If the results of the samples exceed the WQCC limits and/or TDS limits, the water will be treated prior to discharge or disposed at the Kutz lined pond.

Page 3 - Remediation Plan - Manana - Mary Wheeler #1E

15. Following item 14 above, water well S-1 will be pumped for 48 hours at a constant rate of about 65 gpm and sampled at the beginning and end of the 48 hour period. Visual observations will be noted at the beginning, middle and end of the 48 hour period. If the results from the 24 hour pumping were less than the above limits and no visual problems are noted, the water will be discharged as discussed in Item 14. Otherwise, the water will be contained in tanks and either treated prior to discharge to the landowner's property or disposed of at the Kutz lined pond.

16. Every thirty(30) days thereafter for a period of six(6) months, water well S-1 will be pumped for 48 hours at a constant rate of about 65 gpm. Samples will be collected at the beginning and end of the 48 hour period and analyzed for the NMOCD recommended constituents. Visual observations will be noted at the beginning, middle and end of the 48 hour period. If the results of the previous sampling are below the WQCC limits and no visual problems are noted and EPNG receives approval from the landowner, water will be discharged onto the landowner's property. Otherwise, the water will be stored in tanks for treatment prior to discharge to the property or disposed at the Kutz Plant lined pond. If the sample results at the end of the six month period are satisfactory, no additional monitoring will be performed.

17. After remediation is complete and the groundwater is demonstrated to be clean, water well S-1 will be plugged and abandoned.



TO: Tom Hutchins

DATE: July 12, 1991

FROM: Leonard I. Lord

PLACE: Right of Way Dept. Farmington Division

RE: R/W 890320 Manana Gas Company -Mary Wheeler #1E San Juan County, New Mexico

On July 10, 1991 Anu Pundari, Henry Bluejacket and I met with Mr. Eli Velasquez, son-in-law of Richard G. Thurstonson and occupant of the subject land, to discuss our June 10, 1991 letter.

We discussed the contents of the June 10th letter as well as some of the construction activities that are proposed during this project. Mr. Velasquez had no objections to our proposed project and offerred any assistance that he may be to the project.

Mr. Velasquez was invited to the "Contractor's Showing" that is to be held the week of July 15th.

The settlement of surface damage for this project will be after construction is completed. Mr. Velasquez was in agreement with this manner of settlement.

eonard I. Lord

LIL/sj

cc: Anu Pundari Glen E. Orr



P. O. BOX 4990 FARMINGTON, NEW MEXICO 87499 PHONE: 505-325-2841

July 17, 1991

Mr. Ed Hartman Manana Gas, Inc. P.O. Box 14069 Albuquerque, NM 87191-4069

Re: Manana Mary Wheeler #1E Gas Well Site, Flora Vista,NM

Dear Mr. Hartman:

El Paso Natural Gas Company has prepared a Scope of Work for the remediation project and plans to conduct a Bid Showing on July 18, 1991.

The project start date is August 5, 1991 and excavation should be completed by August 30. Please shut-in the gas well and remove your ancillary production units(e.g., the separator and storage tank) during the week of July 29th. In addition, it would be helpful if you could remove all of the underground piping connecting to and from the production units in the excavation area.

EPNG will remove our facilities during the week of July 29th. You may want to coordinate the removal of facilities with our Kutz Pipeline Superintendent, Walt Mosley or Lead Technician, Glen Zimmerman .Both Mr. Mosley and Mr. Zimmerman can be reached at 632-2470 or 632-1180.

Since worker safety and access are important to this project, the shut-in of the well and removal of the production facilities will help to assure a safe working environment.

We will notify you after project completion. If you have any questions regarding the project, please call me at (505) 599-2176 or Mr. Tom Hutchins at (915) 541-3531.

Sincerely,

a.n. Pundani

A.N. Pundari

cc: Mr. Tom Hutchins

bc: Jamye Boone Ward
Walt Mosley
Glen Zimmerman
S. Miller/R. Duarte/File 5216-Mary Wheeler #1E

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P. O. BOX 1492 EL PASO, TEXAS 79978 PHONE: 915-541-3071

JAMYE BOONE WARD ATTORNEY AT LAW

July 17, 1991

William F. Carr Campbell & Black, P.A. P.O. Box 2208 Santa Fe, New Mexico 87504-2208

> Re: Contamination at Manana-Mary Wheeler #1E Gas Well Site, Flora Vista, New Mexico

Dear Bill:

Tomorrow, July 18, 1991, El Paso Natural Gas Company ("EPNG") will have a bid showing, the first step in hiring a contractor to remediate the ground water contamination at the Manana-Mary Wheeler #1E gas well site. Anu Pundari, EPNG's engineer overseeing the project, is notifying Mr. Hartman of EPNG's proposed remediation schedule. Actual remediation activities are scheduled to begin August 5, 1991. In order to complete the remediation as scheduled, Mr. Hartman should shut in the well and remove the production equipment no sooner than July 29, 1991, and no later than August 4, 1991.

EPNG appreciates Mr. Hartman's cooperation in this matter, but must emphasize once more that EPNG is not responsible for the contamination and is not acting as a volunteer. You attended several meetings during which the New Mexico Oil Conservation Division discussed the ground water contamination near the Manana-Mary Wheeler #1E well site. During those meetings and by written notice, OCD directed EPNG and Manana Gas, Inc., jointly, to remediate the contamination at the site.

Unfortunately, the only way to legally determine responsibility for the contamination as between the two companies is to allow OCD to take the matter to public hearing. EPNG believes that taking this matter to a public hearing is unwise and against the best interests of both companies. Therefore, EPNG is left with no choice but to conduct the remediation.

Very truly yours,

Jamy Boone Ward

William F. Carr July 17, 1991 Page 2

C: L. R. Tarver Vice President El Paso Natural Gas Company

W. P. Pearce--Montgomery & Andrews

William F. Carr July 17, 1991 Page 3

bc: G. E. Bauer T. D. Hutchins G. J. Odegard A. Pundari

WPPENV32:27



	JIL CONSER.	N DIVISION
	P. O. BOX 1492 🛛 🕅 🗧 🥻 EL PASO, TEXAS 79978	. ED
May 17,	1991 '91 15.541-2600 1991 '91 1141' 2;;	AM 9 06

Mr. David Boyer Environmental Bureau Chief New Mexico Oil Conservation Division P.O. Box 2088 Land Office Building Santa Fe, New Mexico 87504-2088

RE: Contamination at Manana-Mary Wheeler #1E Gas Well Site, Flora Vista, New Mexico

Dear David:

In accordance with the schedule provided by my letter dated April 24, 1991, attached is a copy of a letter from Bill Carr, the attorney for Manana Gas, Inc., acknowledging that Manana will cooperate with EPNG's request to shut in the well and remove the production equipment prior to EPNG's remediation of the site. Manana has advised that they are not willing to provide substitute gas service to the house adjacent to the well. However, they will notify the property owner of the service interruption.

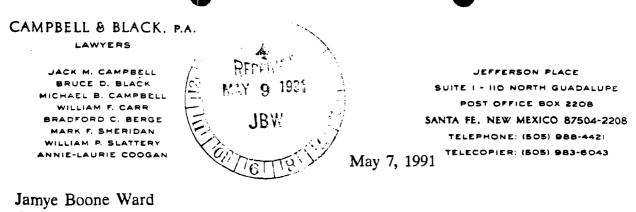
The negotiations with the landowner for the right-of-way required to perform the cleanup and authorization to dispose of the water from the pump test is continuing. We anticipate the negotiations and agreement to be complete by May 31, 1991. A copy of the agreement will be provided for your files.

Also, the equipment needed to establish the background Total Petroleum Hydrocarbon (TPH) levels has not been received. Therefore, the background TPH will not be established until sometime in June. You will be provided a report when the work is completed.

Please let me know if you have any questions.

Thomas D. Hatchis

Thomas D. Hutchins Manager North Region Compliance Eng.



Jamye Boone Ward Attorney at Law El Paso Natural Gas Company Post Office Box 1492 El Paso, Texas 79978

Re: Contamination at Manana-Mary Wheeler No. 1E Gas Well Site, Flora Vista, New Mexico

Dear Jamye:

I have reviewed your letter of April 25, 1991 concerning the above referenced matter with Mr. Ed Hartman and can advise that Manana will cooperate with El Paso Natural Gas Company in its efforts to clean up the contamination at the above referenced well site.

Manana will shut in the Mary Wheeler No. 1E Gas Well during the excavation activities and will remove the production equipment and the storage tank for produced water at the well site. Manana does not believe it is its responsibility to provide substitute gas service to the house presently receiving gas from the Manana Mary Wheeler No. 1E Well but will advise the owner of this residence of the interuption in gas supply that will occur during your clean up operation. El Paso should coordinate its efforts at this well site directly with Mr. Hartman at Manana Gas, Inc.

Manana is willing to assist El Paso in these remediation efforts but this assistance should not be considered a change in its position on this matter. Manana Gas, Inc. continues to believe that it is not responsible for the contamination at this well site and it is its position that in proceeding with the proposed clean up operation, El Paso is acting as a volunteer.

truly yours /

WILLIAM F. CARR ATTORNEY FOR MANANA GAS, INC.

WFC:ys cc: Mr. Ed Hartman Mr. W. Perry Pearce. OIL CONSERVE ON DIVISION RECEIVED

Natural Gas Company 1 APR 29 AM 9 28

P. O. BOX 1492 EL PASO, TEXAS 79978 PHONE: 915-541-2600

April 24, 1991

Mr. David Boyer Environmental Bureau Chief New Mexico Oil Conservation Division P.O. Box 2088 Land Office Building Santa Fe, New Mexico 87504-2088

RE: Contamination at Manana-Mary Wheeler #1E Gas Well Site, Flora Vista, New Mexico

Dear David:

El Paso Natural Gas Company (EPNG) has received Mr. Morrow's letter dated, March 26, 1991, approving El Paso Natural Gas Company's remediation plan for the Mary Wheeler #1E site. Your assistance and your staff's was most helpful in finalizing the plan.

I would like to provide the remediation schedule for performing the work outlined in the plan as follows:

- Reach agreement with Manana Oil, Inc., Item 7. Begin on April 8, estimated completion by May 10;
- Obtain Right-of-Way agreement with adjacent landowner, Item
 20. Begin April 15, estimated completion by May 17;
- Determine background Total Petroleum Hydrocarbon level, Item
 2. Begin April 29, estimated completion by May 31;
- Select a borrow site, Item 9. Begin April 29, estimated completion by May 31;
- 5. Perform soil excavation, groundwater disposal and backfilling - Begin July 29, estimated completion by August 30;
- 6. Sample Water Well S-1 Begin September 3, estimated completion by February 1992.

We will provide you data on Nos. 1, 2, 3 and 4, as soon as possible and notify you of any schedule changes in advance.

Sincerely yours,

Thomas D. Hutchins

Thomas D. Hutchins Manager North Region Compliance Eng.



OIL GONSERS TOON DIVISION RECTIVED

P. O. BOX 1492 EL PASO, TEXAS 79978 PHONE: 915-541-3071

JAMYE BOONE WARD ATTORNEY AT LA 91 APR 29 AM 9 58

April 25, 1991

William F. Carr Campbell & Black, P.A. P.O. Box 2208 Santa Fe, New Mexico 87504-2208

> Re: Contamination at Manana-Mary Wheeler #1E Gas Well Site, Flora Vista, New Mexico

Dear Bill:

Enclosed is a copy of the remediation plan El Paso Natural Gas Company submitted to the New Mexico Oil Conservation Division by letter dated March 19, 1991. Jim Morrow sent you and Mr. Hartman copies of his letter dated March 26, 1991, in which OCD accepts and approves the plan as submitted. El Paso intends to begin soil excavation in late July 1991, and complete it by 1991. Due to potential weather related August 30, complications, it is very important that El Paso begin the cleanup as scheduled. In preparation of the soil excavation and total remediation of the site, El Paso is seeking the cooperation and assistance of Manana Gas, Inc.

Please refer to the paragraph numbered 7 in the attached letter. El Paso cannot begin the cleanup of the contamination at Manana-Mary Wheeler #1E until Manana takes its wellhead out of service, removes its ancillary production units from the well site, and provides the adjacent house with an alternate source of natural gas. As soon as El Paso completes the cleanup of the site, Manana may replace its production equipment and resume gas production and delivery to the adjacent house.

You will receive a copy of the remediation schedule submitted to OCD. Manana's cooperation through completion of the three items listed above is crucial to the proposed remediation schedule. Within a week from El Paso's submission of its proposed schedule to OCD, I will contact you for assurance of Manana's cooperation. William F. Carr April 25, 1991 Page 2

Feel free to contact me or Perry Pearce if you have any questions regarding this matter. Thank you in advance for your cooperation.

> Very truly yours, Jamey Ward

Enclosure

c (w/enclosure):

J. F. Eichelmann--Burlington Resources Inc.

W. P. Pearce--Montgomery & Andrews

Dave Boyer Environmental Bureau Chief New Mexico Oil Conservation Division P.O. Box 2088 State Land Office Building Santa Fe, NM 87504



P. C. ECK 1423 EL PASC (TEKAC) TERT PHONE: 915341 CK

March 19, 1991

Mr. David Boyer Environmental Bureau Chief New Mexico Oil Conservation Division P.O. Box 2088 State Land Office Building Santa Fe, New Mexico 87504

> Re: Contamination at Manana-Mary Wheeler #1E Gas Well Site, Flora Vista, New Mexico

Dear David:

El Paso Natural Gas Company ("El Paso") recognizes that the New Mexico Oil Conservation Division ("OCD") has the jurisdiction necessary to demand that El Paso and all other responsible parties remediate the site of the Manana-Mary Wheeler #1E gas well located in Unit M (SW/4, SW/4) Section 23, Township 30 North, Range 12 West, NMPM, San Juan County, New Mexico. Further, El Paso recognizes that OCD has the power to enforce such a requirement through a properly issued order.

Based on OCD's authority to require El Paso to conduct the necessary cleanup activities, El Paso is prepared to do so. El Paso will provide a schedule for the cleanup activities as soon as El Paso receives OCD's written acknowledgement that El Paso's remediation plan, contained herein, is sufficient and adequate to accomplish the required remediation.

In response to your demands that remediation be conducted at this site, El Paso will cleanup the contamination at the Manana-Mary Wheeler #1E gas well near Flora Vista, New Mexico according to the following procedures set forth in Exhibit A, attached hereto and incorporated herein.

El Paso's actions taken in response to your demands to remediate the Manana-Mary Wheeler #lE gas well site should not be construed as a waiver of its rights to contribution from other responsible parties. Mr. David: Boyer March 19, 1991 Page 2

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If you have any questions with regard to this plan, please feel free to contact me or Jamye Boone Ward.

Very truly yours,

Homas D. Hatchins

Thomas D. Hutchins Manager, North Region Engineering

EXHIBIT A

The outline below includes the plan and procedures which El Paso Natural Gas Company will follow pursuant to the clean up of the groundwater contamination at the Manana-Mary Wheeler #1E Gas Well Site near Flora Vista, New Mexico:

- 1. The reserve pit and the area of soil visibly stained by hydrocarbon will be excavated until the field detection unit, Miran 1FF Analyzer, indicates a TPH level less than 100 ppm above background level. Said area was delineated during the site investigation study. A "strip mining" approach will be used, in which clean overburden soil is removed and stockpiled prior to excavation of the underlying stained soil.
- In order to determine the background TPH level, 2. four (4) soil samples will be collected in an area no less than 100 feet up gradient from the beginning of the excavation area. At each of the four (4) locations, soil will be collected at the surface, at a depth of 12 inches, and at depth of 24 inches. The soil will be а composited into a single sample for analysis by the 418.1 IR and Modified 8015 Methods. The average of the four sample results will be used for the background TPH level, if approved by OCD. A map showing the area and location of each sample will be provided to OCD along with the sample results.
- 3. The excavation will begin up gradient. Overburden to a depth of four to five feet, or until hydrocarbon stained soil is encountered, will be removed and stockpiled for reuse as final backfill after the remediation is complete. Only clean overburden soil will be stockpiled and used for backfill. Any clean soil mixed with stained soil will not be used for backfill.
- 4. Composite samples will be collected in each excavation zone. An excavation zone is defined as the backhoe swing width by a maximum length of 50 feet or the length of the excavation, whichever is less. The number of samples collected for each composite will be determined

in accordance with the SW846 field sampling protocol. The soil samples will be analyzed for BETX, EPA 8020 Method and TPH, Modified 8015 Method.

5. The zones will be backfilled when the field detection results for TPH are less than 100 ppm above background. If the field results are close to, but slightly greater than, 100 ppm above background, the zone may be sampled for laboratory verification, left open until results are received, and then backfilled if the lab verification results are acceptable. In no case will a zone be backfilled before composite verification samples are collected. If the excavation depths reach eight feet and the field and lab sample results are greater than 100 ppm above background, excavation will cease until El Paso and OCD assess the need for additional excavation.

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- 6. Before commencing the excavation activities, El Paso will remove all its ancillary equipment from the site. After the remediation is complete, El Paso will reinstall its ancillary equipment.
- 7. In order to assure worker safety and access to the site, the operator must perform the following:
 - a. shut-in the gas well for the duration of the excavation activities;
 - b. remove the operator's ancillary production units (e.g., the separator and storage tank) prior to excavation activities;
 - c. assure that the house located near the well which presently receives gas directly from the Manana-Mary Wheeler #1E well is provided substitute gas service during the shut-in period.

El Paso will seek to reach agreement with the operator for its cooperation in attaining items a, b, and c, above. A copy of such agreement will be provided to OCD. However, if El Paso and the operator are unable to reach agreement, OCD will be notified as soon as possible.

- 8. Any above or below ground tanks installed after the remediation is complete will be equipped with OCD approved leak detection systems.
- 9. To augment the clean overburden which will be placed back into the excavated area, a borrow site with coarse-grained sand will be designated. Most likely, the borrow site will be near Ballard Plant, which is also the disposal site for the contaminated soil. Once a borrow site is designated, El Paso will notify OCD as to the site's location and a representative sample of the borrow material will be provided.

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- 10. A small amount of ground water will probably be encountered. All water observed to have a hydrocarbon sheen will be removed during excavation. A portable pump or vacuum truck will be used to pump the water from the excavation area during soil removal.
- 11. In the event it becomes necessary to excavate below the water table, the saturated soil will be placed in dump trucks. Each dump truck tailgate will be pinned and the bed tilted over a trough designed to collect and separate the hydrocarbon/water mixtures draining from the saturated soil. This procedure will result in effective drainage of the wet soil and avoidance of any leakage during transport.
- 12. Absorbent pads will be available on site to facilitate the capture of any free-phase hydrocarbons.
- 13. The collected water from the excavation area will be pumped into water trucks and hauled to the Kutz Plant lined pond or some other legally permitted disposal site. In accordance with OCD recommendations the Kutz Plant pond liner was inspected for leaks, tears, etc. and none were found.

A containment "sweep" will be available at the Kutz pond. If necessary, the sweep will be used to move and capture any free-phase hydrocarbons to one side of the pond. The hydrocarbons will then be collected in a pump truck and transported to the Blanco Plant for recycling.

- 14. Hydrocarbon contaminated soil will be removed to Ballard Station and spread in six inch lifts. The soil will be disced twice a week for the duration of the excavation activities. After excavation is complete, random samples of the disced soil will be collected and analyzed for TPH and BETX to ensure successful If the TPH, BETX or benzene exceed aeration. 100 ppm, 50 ppm, or 10 ppm, respectively, the soil will be disced until the sample results do not exceed the above limits.
- 15. A temporary trench will be constructed down gradient of the excavation area to serve as an observation trench. The observation trench will provide visual verification that floating hydrocarbons are removed during the excavation process. In addition, the trench will allow volatilization natural of any remaining dissolved hydrocarbons. An agitator will be installed in the trench to act as a final treatment "air stripper" and aeration system. The agitator will be kept in use for a short period (60 - 90 days) after soil remediation is complete to allow flushing of the replacement soil. A fence will be placed around the trench for the period it is open. The fence will be removed and the area returned to its original grade after the 60 - 90 day period is over. Water samples will be collected at the beginning of the excavation activities, at excavation seven day intervals during activities or more often if a hydrocarbon sheen is observed, and at thirty day intervals once remediation is complete. The samples will be analyzed for BETX and TPH by the 8020 and Modified 8015 Methods, respectively.
- 16. All existing OCD monitoring wells will be removed. In addition, all El Paso monitoring wells in the excavation area will be removed.
- 17. After remediation is complete, well S-1 will be pumped at a constant rate of about 65 gallons per minute (gpm) for 24 hours and sampled. Visual observations of the color, odor, and turbidity of the water will be noted at the beginning and end of the 24 hour period. The samples will be analyzed for BETX and TDS. The water discharged during the 24 hour pumping

will be contained in tanks until the sample results are received. If the sample results are less than WQCC limits for BETX and 1000 ppm for TDS, the water will be discharged to the farmer's field west of the remediation site. If the results of the samples exceed the WQCC and/or TDS limits, the water will be treated prior to discharge to the field or disposed of at the Kutz Plant lined pond.

- Following item 17 above, water well S-1 will be 18. pumped for 48 hours at a constant rate of about 65 gpm and sampled at the beginning and end of the 48 hour period. Visual observations will be noted at the beginning, middle and end of the 48 hour period. If the results from the 24 hour pumping were less than the above limits and no visual problems are noted, the water will be discharged directly onto the farmer's field. Otherwise, the water will be contained in tanks and either treated prior to discharge to the farmer's field or disposed of at the Kutz Plant lined pond. The samples collected during the 48 hour pumping will be analyzed for BETX, TPC, EC, sodium, calcium, potassium, magnesium, chloride, sulfate, nitrate, carbonate, and bicarbonate.
- 19. Every thirty (30) days thereafter for a period of six (6) months, water well S-1 will be pumped for 48 hours at a constant rate of about 65 gpm. Samples will be collected at the beginning and end of the 48 hour period and analyzed for the constituents listed in item 18 above. Visual observations will be noted at the beginning, middle and end of the 48 hour If the results of the previous period. sampling are below the WQCC limits and no visual problems are noted, the water will be discharged directly to the farmer's field. Otherwise, the water will be stored in tanks for treatment prior to discharge to the field or disposed of at the Kutz Plant lined pond. If the sample results at the end of the six month period are satisfactory, no additional monitoring will be performed.
- 20. The approval of the landowner and OCD will be obtained prior to any discharge to the farmer's field. El Paso will provide OCD a copy of the

agreement with the landowner, noting approval for discharge onto the field.

- 21. The necessity for additional monitoring wells will be assessed with OCD after remediation activities are complete and sample results received. OCD will be provided a summary report on the remediation activities, sampling, and sample results.
- 22. After remediation is complete and the groundwater is demonstrated to be clean, water well S-1 will be plugged and abandoned.

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STATE OF NEW MEXICO

ENERGY, MINERALS AND NATURAL RESOURCES DEPARTMENT

OIL CONSERVATION DIVISION



BRUCE KING GOVERNOR POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87504 (505) 827-5800

March 26, 1991

<u>CERTIFIED MAIL -</u> <u>RETURN RECEIPT NO. P-327-278-089</u>

Mr. Thomas D. Hutchins, Manager North Region Compliance Engineering El Paso Natural Gas Company P. O. Box 1492 El Paso, Texas 79978

RE: Contamination at Manana-Mary Wheeler #1E Gas Well Site, Flora Vista, New Mexico

Dear Mr. Hutchins:

The Oil Conservation Division (OCD) has received your letter, dated March 19, 1991, to Mr. David Boyer together with the proposed remediation plan (attached as your Exhibit A) for the Mary Wheeler #1E site. This remediation plan has been reviewed by Mr. Boyer and other OCD staff.

Based on information previously provided OCD as part of the site investigation study and information provided in the plan, we believe the plan to be sufficient and adequate to accomplish the required remediation and hereby approve the plan.

You are requested to provide OCD with the proposed schedule for conducting the cleanup and notify OCD 14-days prior to beginning soil removal to allow OCD viewing of this procedure.

Please be advised that OCD approval does not relieve you from taking additional measures should the approved plan fail to remediate petroleum contaminated ground water at this site. In addition, OCD approval does not relieve you of responsibility for compliance with other city, county, state and federal laws and/or regulations.

Mr. Thomas Hutchins March 26, 1991 Page -2-

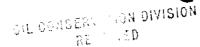
We appreciate El Paso Natural Gas Company's cooperation in undertaking this remediation activity and believe that voluntary compliance with our agency's regulatory requirements provides the fastest and most efficient use of agency and company time and resources. We look forward to working with you as groundwater cleanup and site remediation proceeds. If you have any questions, contact Mr. David Boyer of my staff at (505) 827-5812.

Sincerely

Jim Morrow Deputy Director

JM/DGB

cc: OCD Aztec Office William Carr - Campbell & Black Perry Pearce - Montgomery & Andrews John Eichelmann - Burlington Resources Ed Hartman - Manana Oil





'91 MAR 21 AM 9 15

P. O. BOX 1492 EL PASO, TEXAS 79978 PHONE: 915-541-2600

March 19, 1991

Mr. David Boyer Environmental Bureau Chief New Mexico Oil Conservation Division P.O. Box 2088 State Land Office Building Santa Fe, New Mexico 87504

> Re: Contamination at Manana-Mary Wheeler #1E Gas Well Site, Flora Vista, New Mexico

Dear David:

El Paso Natural Gas Company ("El Paso") recognizes that the New Mexico Oil Conservation Division ("OCD") has the jurisdiction necessary to demand that El Paso and all other responsible parties remediate the site of the Manana-Mary Wheeler #1E gas well located in Unit M (SW/4, SW/4) Section 23, Township 30 North, Range 12 West, NMPM, San Juan County, New Mexico. Further, El Paso recognizes that OCD has the power to enforce such a requirement through a properly issued order.

Based on OCD's authority to require El Paso to conduct the necessary cleanup activities, El Paso is prepared to do so. El Paso will provide a schedule for the cleanup activities as soon as El Paso receives OCD's written acknowledgement that El Paso's remediation plan, contained herein, is sufficient and adequate to accomplish the required remediation.

In response to your demands that remediation be conducted at this site, El Paso will cleanup the contamination at the Manana-Mary Wheeler #1E gas well near Flora Vista, New Mexico according to the following procedures set forth in Exhibit A, attached hereto and incorporated herein.

El Paso's actions taken in response to your demands to remediate the Manana-Mary Wheeler #1E gas well site should not be construed as a waiver of its rights to contribution from other responsible parties. Mr. David Boyer March 19, 1991 Page 2

If you have any questions with regard to this plan, please feel free to contact me or Jamye Boone Ward.

Very truly yours,

Homas D. Hatchins

Thomas D. Hutchins Manager, North Region Engineering

EXHIBIT A

The outline below includes the plan and procedures which El Paso Natural Gas Company will follow pursuant to the clean up of the groundwater contamination at the Manana-Mary Wheeler #IE Gas Well Site near Flora Vista, New Mexico:

- 1. The reserve pit and the area of soil visibly stained by hydrocarbon will be excavated until the field detection unit, Miran 1FF Analyzer, indicates a TPH level less than 100 ppm above background level. Said area was delineated during the site investigation study. A "strip mining" approach will be used, in which clean overburden soil is removed and stockpiled prior to excavation of the underlying stained soil.
- 2. In order to determine the background TPH level, four (4) soil samples will be collected in an area no less than 100 feet up gradient from the beginning of the excavation area. At each of the four (4) locations, soil will be collected at the surface, at a depth of 12 inches, and at depth of 24 inches. The soil will be а composited into a single sample for analysis by the 418.1 IR and Modified 8015 Methods. The average of the four sample results will be used for the background TPH level, if approved by OCD. A map showing the area and location of each sample will be provided to OCD along with the sample results.
- 3. The excavation will begin up gradient. Overburden to a depth of four to five feet, or until hydrocarbon stained soil is encountered, will be removed and stockpiled for reuse as final backfill after the remediation is complete. Only clean overburden soil will be stockpiled and used for backfill. Any clean soil mixed with stained soil will not be used for backfill.
- 4. Composite samples will be collected in each excavation zone. An excavation zone is defined as the backhoe swing width by a maximum length of 50 feet or the length of the excavation, whichever is less. The number of samples collected for each composite will be determined

in accordance with the SW846 field sampling protocol. The soil samples will be analyzed for BETX, EPA 8020 Method and TPH, Modified 8015 Method.

- 5. The zones will be backfilled when the field detection results for TPH are less than 100 ppm above background. If the field results are close to, but slightly greater than, 100 ppm above background, the zone may be sampled for laboratory verification, left open until results are received, and then backfilled if the lab verification results are acceptable. In no case will a zone be backfilled before composite verification samples are collected. If the excavation depths reach eight feet and the field and lab sample results are greater than 100 ppm above background, excavation will cease until El Paso and OCD assess the need for additional excavation.
- 6. Before commencing the excavation activities, El Paso will remove all its ancillary equipment from the site. After the remediation is complete, El Paso will reinstall its ancillary equipment.
- 7. In order to assure worker safety and access to the site, the operator must perform the following:
 - a. shut-in the gas well for the duration of the excavation activities;
 - b. remove the operator's ancillary production units (e.g., the separator and storage tank) prior to excavation activities;
 - c. assure that the house located near the well which presently receives gas directly from the Manana-Mary Wheeler #1E well is provided substitute gas service during the shut-in period.

El Paso will seek to reach agreement with the operator for its cooperation in attaining items a, b, and c, above. A copy of such agreement will be provided to OCD. However, if El Paso and the operator are unable to reach agreement, OCD will be notified as soon as possible.

- 8. Any above or below ground tanks installed after the remediation is complete will be equipped with OCD approved leak detection systems.
- 9. To augment the clean overburden which will be placed back into the excavated area, a borrow site with coarse-grained sand will be designated. Most likely, the borrow site will be near Ballard Plant, which is also the disposal site for the contaminated soil. Once a borrow site is designated, El Paso will notify OCD as to the site's location and a representative sample of the borrow material will be provided.
- 10. A small amount of ground water will probably be encountered. All water observed to have a hydrocarbon sheen will be removed during excavation. A portable pump or vacuum truck will be used to pump the water from the excavation area during soil removal.
- 11. In the event it becomes necessary to excavate below the water table, the saturated soil will be placed in dump trucks. Each dump truck tailgate will be pinned and the bed tilted over a trough designed to collect and separate the hydrocarbon/water mixtures draining from the saturated soil. This procedure will result in effective drainage of the wet soil and avoidance of any leakage during transport.
- 12. Absorbent pads will be available on site to facilitate the capture of any free-phase hydrocarbons.
- 13. The collected water from the excavation area will be pumped into water trucks and hauled to the Kutz Plant lined pond or some other legally permitted disposal site. In accordance with OCD recommendations the Kutz Plant pond liner was inspected for leaks, tears, etc. and none were found.

A containment "sweep" will be available at the Kutz pond. If necessary, the sweep will be used to move and capture any free-phase hydrocarbons to one side of the pond. The hydrocarbons will then be collected in a pump truck and transported to the Blanco Plant for recycling.

- 14. Hydrocarbon contaminated soil will be removed Ballard Station and spread in six inch to lifts. The soil will be disced twice a week for the duration of the excavation activities. After excavation is complete, random samples of the disced soil will be collected and analyzed for TPH and BETX to successful ensure If the TPH, BETX or benzene exceed aeration. 100 ppm, 50 ppm, or 10 ppm, respectively, the soil will be disced until the sample results do not exceed the above limits.
- 15. A temporary trench will be constructed down gradient of the excavation area to serve as an observation trench. The observation trench will provide visual verification that floating hydrocarbons are removed during the excavation In addition, the trench will allow process. of volatilization remaining natural any dissolved hydrocarbons. An agitator will be installed in the trench to act as a final treatment "air stripper" and aeration system. The agitator will be kept in use for a short period (60 - 90 days) after soil remediation is complete to allow flushing of the replacement soil. A fence will be placed around the trench for the period it is open. The fence will be removed and the area returned to its original grade after the 60 - 90 day period is over. will be collected Water samples at the beginning of the excavation activities, at during seven day intervals excavation activities or more often if a hydrocarbon sheen is observed, and at thirty day intervals once remediation is complete. The samples will be analyzed for BETX and TPH by the 8020 and Modified 8015 Methods, respectively.
- 16. All existing OCD monitoring wells will be removed. In addition, all El Paso monitoring wells in the excavation area will be removed.
- 17. After remediation is complete, well S-1 will be pumped at a constant rate of about 65 gallons per minute (gpm) for 24 hours and sampled. Visual observations of the color, odor, and turbidity of the water will be noted at the beginning and end of the 24 hour period. The samples will be analyzed for BETX and TDS. The water discharged during the 24 hour pumping

will be contained in tanks until the sample results are received. If the sample results are less than WQCC limits for BETX and 1000 ppm for TDS, the water will be discharged to the farmer's field west of the remediation site. If the results of the samples exceed the WQCC and/or TDS limits, the water will be treated prior to discharge to the field or disposed of at the Kutz Plant lined pond.

- 18. Following item 17 above, water well S-1 will be pumped for 48 hours at a constant rate of about 65 gpm and sampled at the beginning and end of the 48 hour period. Visual observations will be noted at the beginning, middle and end of the 48 hour period. If the results from the 24 hour pumping were less than the above limits and no visual problems are noted, the water will be discharged directly onto the farmer's Otherwise, the water will be contained field. in tanks and either treated prior to discharge to the farmer's field or disposed of at the Kutz Plant lined pond. The samples collected during the 48 hour pumping will be analyzed for BETX, TPC, EC, sodium, calcium, potassium, magnesium, chloride, sulfate, nitrate, carbonate, and bicarbonate.
- Every thirty (30) days thereafter for a period 19. of six (6) months, water well S-1 will be pumped for 48 hours at a constant rate of about Samples will be collected at the 65 qpm. beginning and end of the 48 hour period and analyzed for the constituents listed in item 18 above. Visual observations will be noted at the beginning, middle and end of the 48 hour If the results of the previous period. sampling are below the WQCC limits and no visual problems are noted, the water will be discharged directly to the farmer's field. Otherwise, the water will be stored in tanks for treatment prior to discharge to the field or disposed of at the Kutz Plant lined pond. If the sample results at the end of the six month period are satisfactory, no additional monitoring will be performed.
- 20. The approval of the landowner and OCD will be obtained prior to any discharge to the farmer's field. El Paso will provide OCD a copy of the

agreement with the landowner, noting approval for discharge onto the field.

- 21. The necessity for additional monitoring wells will be assessed with OCD after remediation activities are complete and sample results received. OCD will be provided a summary report on the remediation activities, sampling, and sample results.
- 22. After remediation is complete and the groundwater is demonstrated to be clean, water well S-1 will be plugged and abandoned.

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OK w/phone 15 Revised 3/27/91 TDH 2/10/8

SUBJECT: Site Remediation for the Manana - Mary Wheeler #12 (Letty L)

GNON TH The following outlines the procedures and plan for the remediation activities at the subject well site.

1. The reserve pit and the area of soil visibly stained by hydrocarbon will be excavated until the field detection unit, Miran 1FF Analyzer, indicates a TPH level less than 100 ppm above background. The area was delineated during the site investigation study. A "strip mining" approach will be used, in which clean overburden is removed and stockpiled prior to excavation of the underlying stained soil.

2. To determine the background TPH level, four soil samples will be collected in an area a minimum of 100 feet up gradient from the beginning of the excavation area. At each of the four locations, soil will be collected at the surface, a depth of 12 inches and a depth of 24 inches. The soil will be composited into a single sample for analysis by the 418.1 IR and Modified 8015 Methods. The average of the four sample results will be used for the background TPH level, if approved by OCD. A map shoving the area and location of each sample will be provided to OCD along with the sample results.

3. The excavation will begin up gradient. Overburden to a depth of four to five feet, or until hydrocarbon stained soil is encountered, will be removed and stockpiled for reuse as final backfill after the remediation is complete. Only clean overburden will be stockpiled and used for backfill. Any soil mixed with stained soil will not be used for backfill.

 Composite samples will be collected in each excavation zone. An excevation zone is defined as the backhoe swing width by a meximum length of 50 feet or the length of the excavation, whichever is less. The number of samples collected for each composite will be determined in accordance with the SW846 field sampling protocol. The soil samples will be analyzed for BETX, EPA 8020 Nethod and TPH, Modified 8015 Nethod.

5. The zones will be backfilled when the field detection results for TPH are less than 100 ppm above background. If the field results are close to but slightly greater than 100 ppm above background, the zone may be sampled for laboratory verification, left open until results are received and then backfilled if the lab verification results are acceptable. In no case will a zone be backfilled before composite verification samples are collected. If the excavation depths reach eight feet and the field and



lab sample results are greater than 100 ppm above background, excavation will cease until El Paso and OCD assess the need for additional excavation.

6. Before commencing the excavation activities, El Paso will remove all of it's ancillary equipment from the site. After the remediation is complete, El Paso will reinstall it's equipment.

7. In order to assure worker safety, access to the site and a thorough and final cleanup, the operator must perform the follow-ing:

a. Shut-in the well for the duration of the excavation activities;
b. remove the operator's ancillary production units prior to excavation activities;
c. assure that the house located near the well, which presently receives gas directly from the Manana-Mary Wheeler #1E well, is provided substitute gas service during the shut in period.

El Paso vill seek to reach an agreement with the operator for the above items and a copy of the agreement will be provided to OCD. However, if El Paso is unable to reach an agreement with the operator, OCD will be notified as soon am possible.

8. Any above or below ground tanks installed after the remediation is complete must be installed with leak detection equipment that is approved by OCD.

9. To augment the clean overburden that will be placed into the excavation area, a borrow site with coarse-grained sand will be designated. Most likely the borrow site will be near Ballard Plant, which is also the disposal site for the contaminated soil. Once a borrow site is located, OCD will be notified of the location and a representative sample of the borrow material will be provided.

10. A small amount of ground water will be encountered. Any water observed during excevation with a hydrocarbon sheen will be removed. A portable pump or vacuum truck will be used to remove the water from the excavation area during excavation. The water -will be discharged taken to the lined pond at Kutz.

11. In the event it is necessary to excavate below the water table, the saturated soils will be placed into dump trucks. The dump truck tailgate will be pinned and the bed tilted over a trough designed to collect and separate the hydrocarbon/water mixtures that drain from the saturated soil. This procedure will result in effective drainage of the wet soils and avoidance of any loskage during transport. The collected water will be trucked to the lined pond at Kutz Plant or another permitted dispesal site.

12. Absorbent pads will be available on site to facilitate the

capture of any free-phase hydrocarbons.

13. The water from the excavation area will be pumped into water trucks and hauled to the Kutz lined pond or another permitted disposal site. In accordance with OCD recommendations the Kutz pond liner was inspected for leaks, tears, etc. and none were found.

A containment "sweep" will be on site at the Kutz pond. The sweep will be used to move and capture any free-phase hydrocarbons to one side of the pond. The hydrocarbons will be collected in a pump truck and transported to the Blanco Plant for recycling.

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15. A temporary trench will be constructed just down gradient of the excavation area to serve as an observation trench. It will provide visual verification that no floating hydrocarbons escaped during the excevation process. It will also allow natural volatilization of any remaining dissolved hydrocarbons. An agitator will be installed in the trench to act as a final treatment "air stripper" and aeration system. It will be kept in use for a short period (60 - 90 days) after soil remediation is complete to allow flushing of the replacement soil. A fence will be placed around the trench for the period it is open. The fence will be removed and the area returned to it's original grade after the 60 90 day period. Water samples will be collected at the beginning of the excavation, at seven day intervals during excavation or if a hydrocarbon sheen is observed and at thirty day intervals after remediation until the trench is closed. The samples will be analyzed for BETX and TPH by the 8020 and Modified 8015 Methods, respectively.

16. All former OCD monitoring wells will be removed. In addition, all EPNG monitoring wells in the excavation area will be removed.

17. After remediation, well 5-1 will be pumped at a constant rate of about 65 gallons per minute (gpm) for 24 hours and sampled. Visual observations of the color, odor and turbidity of the vater will be noted at the beginning and end of the 24 hour period. The sample will be analyzed for BETX and TDS. The water discharged during the 24 hour pumping will be contained in tanks until the sample results are received. If the sample results are less than WQCC limits for BETX and 1000 ppm for TDF the water will be discharged to the farmers field. If the results exceed the WGCC and/or TDS limits the water will be treated prior to discharge to the field or disposed of at the Kutz lined pond. 18. Well 5-1 will then be pumped for 48 hours at a constant rate of about 65 gpm and sampled at the beginning and end of the 48 hour period. Visual observations will be noted at the beginning, middle and end of the 48 hour period. If the results from the 24 hour pumping were less than the above limits and no visual problems are noted, the water will be discharged directly onto the farmer's field. Otherwise, the water will be contained in tanks and either treated prior to discharge to the farmer's field or disposed of at the Kutz lined pond. The samples collected during the 48 hour pumping will be analyzed for BETX, TPH, EC, sodium, calcium, potassium, magnesium, chloride, sulfate, nitrate, carbonate and bicarbonate.

19. Every thirty days thereafter for a period of six months, well S-1 will be pumped for 48 hours at a constant rate of about 65 gpm. Samples will be collected at the beginning and end of the 48 hour period. Visual observations will be noted at the beginning, middle and end of the 48 hour period. If the results of the previous sampling were below the WOCC limits and no visual problems are noted, the water will be discharged directly to the farmer's field. Otherwise, the water will be stored in tanks for treatment prior to discharge to the field or disposed of at the Kutz pond. If the sample results at the end of the six month period are satisfactory, no additional monitoring will be performed.

20. The approval of the landowner and OCD will be obtained prior to any discharge to the farmer's field. El Paso will provide OCD a copy of the agreement with the landowner, noting approval for discharge on the field.

21. The necessity for additional monitoring wells will be assessed with OCD after remediation activities are completed and sample results are received. OCD will be provided a summary report on the remediation activities, sampling and mample results.

22. After remediation is complete and the groundwater is shown to be clean, well S-1 will be plugged and abandoned.

EPAIG - OCD Flora Vista Meeting 3/11/91 - 1000 hrs personal - Dan Boyer - OP - Bill Olson - OCD - Hanry Van - EPNG - Anan Pundari - EPNG - Tom Hutchings EPNG. TH Hand out page on site remediation addressing OCD comments DB. Is Manane voogenating in romalisti I.H. So far, will move forward until obstales an P.D. OCP will send letter to Manana it trankle with cooperation arises & item 14 med commitment from kondomm to discharge an OCD, approvid prior to discharge not included IH. O.b. These need to be included as wella as discharge color turbitive, odor & taste during discharge as in Och letter 1/11/91 item 13. These absorbedies reeded at least one per dey

P.B. Meed Letter on file blue landourne and EPN6 Itom 16 can omit refere to dereplacement water well and state that an agreement will be reached will Flora Vista on Cate et S-1 is cappily and abandonment ion dis mange Ø Will Manana be responsible for removing their own equipment? T.H. Yes but may be love by EPA6. P.B. Change item 3 to reflect removal it all andlary ognipment includy Manana's J.H. already correct in item 4 D.D. Othenis the terms 2-16 ok P.D. Dily soils in ceserve pit area still reamain A Originally this are analysed for oil of great later dome with FID/6C nothing for TPH Shrowed not good correlation between methods (see copy of results)

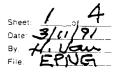
We have field 418.1 equipment that can be used for field continuation T.H. What Jevel TPH used in field WO., D.M. 100 ppm 15 action land P.B. Consider file, or upgradmit sample for field 418.1 method then subtract out from mormal samples to take out bet netword cheets T.A. Could leave open while confirmation big being done. A. Dig to eight feet on TPH field reading which aren p. B. Don't expert to reach air & donth. But need confirmation at actual continuits remaining W.D. Keep close comminication with DCD during excavation to make determination of amount remained required H.V. Will do TH. May consider downlering, for execution and discharge D.D. Needs NPDES parmit, don't went that Will check on discharge to swamp instead at

A.P. What I A confirmation supples needed and type at malysis P.B. TPH & BTEX for analysis Isolate into cells and set sampler from each cell on W.O. Sample as par Str #46 compositions techniques U.O. Try doing tidd contirmation as tool for Leterning benchilling i.e. it clean with field method wont to break each trench in 50 cells with composite trem each cell as below D.B. 1 - = /00' --I saysh cell for ft frile anca

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	MARY Wheeler # IE Gas Well Site in Floravista, N.M.
	· Participanta
MW gals/mol 16.04 C1 6.4 30.07 C2 10.12 44.10 C3 10.42 58.12 iC4 12.38 58.12 iC4 11.93 72.15 iC5 13.85 72.15 nC5 13.71 86.18 iC6 15.57 100.21 iC7 17.2 100.21 C7 17.46 114.23 C8 19.39 28.05 C2' 9.64 42.08 C3' 9.67	- David Boyer - OGD - Dill Olsen - OGD - Tom Hutchins - EPAG - Ann Rundari - " - Henry Van - " Rurpace of Meeting - Toroview draft remediation workplan
	· Discussion Highlights
	- Manana should be notified about the temperory moment of the
	ocil to procure approval for N/W disposal. - Item 14
MISC. <u>MW</u> gals/mol 32.00 O2 3.37 28.01 CO 4.19	· Notify the owner and receives authorization from OCD should
44.01 CO2 6.38 64.06 SO2 5.50 34.08 H2S 5.17 28.01 N2 4.16 2.02 H2 3.38	be atomas stipulated
· · ·	. Thysical barameters should be included as visual abourvation.
-	- Turbidity) once per day, at lost
	- Item 16
	· The agreement is between EPNG & Flora- vista.
	- Item 3 meent about EPrile will attempt to

Form 7-1 (Rev. 9-77) 3/11/21 ENGINEERING CALCULATION By: H. Van File: EPNE Mary Whenler # 1-E (Floravista) Remediation Proposal reach agreement with Manane to remove this equipment to allow мw gais/moi 16.04 C1 6.4 30.07 C2 10.12 remediation and EPNE will notify OCT 44.10 C3 10.42 58.12 12.38 58.12 nC4 11.93 if Marana does not agree. If agreement 72.15 13.85 is reached capey of the agreement should 86.18 C6 15.57 100.21 iC7 17.2 C7 100.21 17.46 114.23 C8 19.39 be sent to OCD. C2[:] 28.05 9.64 42.08 <u>C3</u>: 9.67 IFens:# 5 \$ 6,7 are OK 7 woond "small" te 8 ok 井 g OK # 10 0K MISC # 11 @K MW gals/mo 3.37 32.00 02 HIZ OK 28.01 CO 4.19 44.01 6.38 64.06 SO2 5.50 # 13 OK 34.08 H₂S 5.17 N2 28.01 4,16 2.02 H2 3.38 # 190K Item#1: - The KW Brown indicates vous low values. of what we can measure. - What was left was Hydrocondon - Need to sing ani Fundari's 1990 testing which show no correlation

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ENGINEERING CALCULATION



Maey Wheler #1-E (Flovavista) Kenediahim Trapson between the Method, (418) for TPM and Oil & Grease. oals/mol C1 6.4 C2 We could use Field TPH "MIRAN 10.42 C3 12.38 IFF and JACVF. Kemediatie 13.85 iC5 13.71 TPH luce to quide remediation 15 57 17.2 iC7 17.46 C7 could be : C8 19.39 Cleanup Level 9.64 C2[:] 9.67 C3: Propered Take complex of background TTAZ 100 minus background. 1) Forminge in the Tab from vorification (live the 418 FID method) Leave open for about two weeks while thating MISC gals/mo 2) 2 ft below 02 3.37 co 6.38 CO₂ 3) Take samples -> L=6 5.50 SO2 H₂S 5.17 4.16 N2 4) Review results 3.38 5) gake more empirimation OCT believes that the excountion may not have to be beyond & feet. level these should be discussed at that print.

gais/mol

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10.42 CB

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C7 19.39

C8

C3: 9.67

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gals/mol

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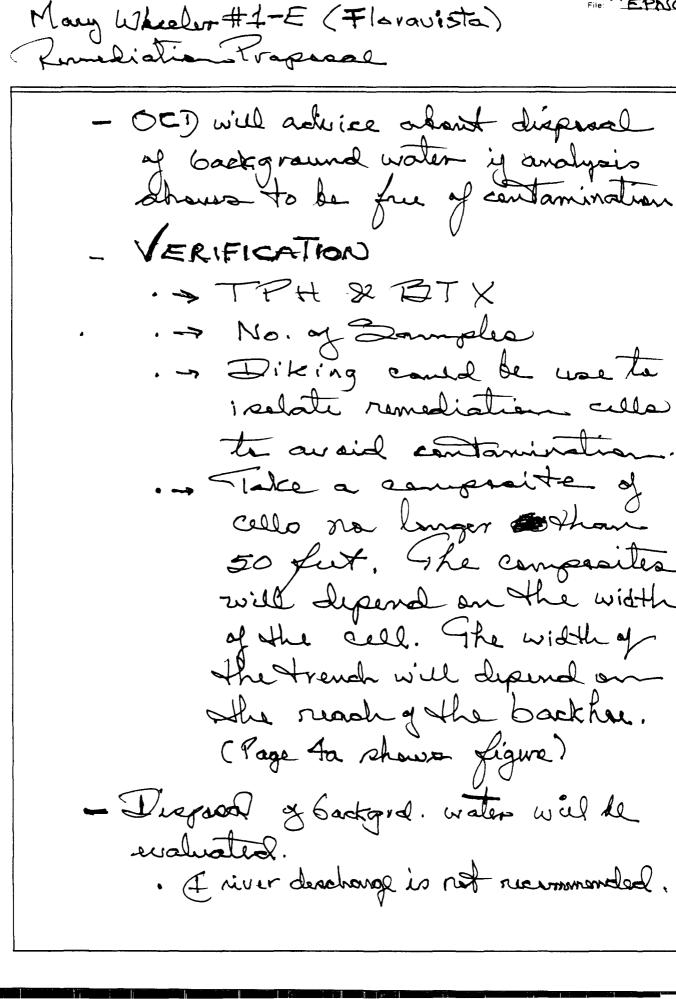
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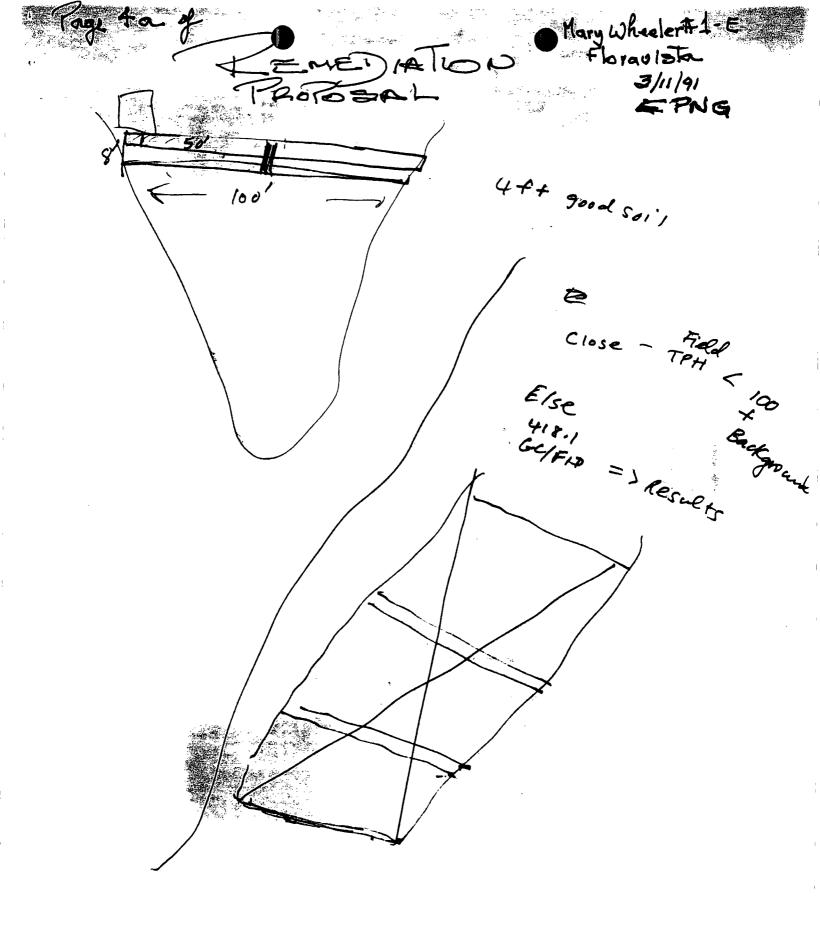
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ENGINEERING CALCULATION





3/11/91

SUBJECT: Site Remediation for the Manana = Mary Wheeler #1E Gas Well Site in Flora Vista, New Mexico

The following outlines the procedures and plan for the remediation activities at the subject well site.

1. The reserve pit and the area of soil visibly stained by hydrocarbon will be excavated to a depth that is two feet below groundwater. The area was delineated during the site investigation study. A "strip mining" approach will be used, in which clean overburden is removed and stockpiled prior to excavation of the underlying stained soil.

2. The excavation will begin up gradient. Overburden to a depth of four to five feet, or until hydrocarbon stained soil is encountered, will be removed and stockpiled for reuse as final backfill after the remediation is completed. Only clean overburden will be stockpiled and used for backfill. Any soil mixed with stained soil will not be used for backfill.

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b. remove the operator's ancillary production units prior
 to excavation activities;

c. assure that the house located near the well, which presently receives gas directly from the Manana-Mary Wheeler #1E well, is provided substitute gas service during the shut in period.

5. Any tanks installed after the remediation is complete must be installed with leak detection equipment that is approved by NMOCD.

6. To augment the clean overburden that will be placed back into the excavation area, a borrow site with coarse-grained sand will be designated. Most likely the borrow site will be near Ballard Plant, which is also the disposal site for the contaminated soil. Once a borrow site is located, OCD will be notified of the location and a representative sample of the borrow material will be provided.

7. A small amount of ground water will be encountered. However, any water observed with a hydrocarbon sheen will be filtered and water with soluble hydrocarbons will be removed during excava-A portable pump will be used to pump the water from the tion. excavation area during soil removal. The water will be discharged into tank trucks and taken to the lined pond at Kutz. 8. In the event it is necessary to excavate below the water table, the saturated soils will be placed into dump trucks. The dump truck tailgate will be pinned and the bed tilted over a trough designed to collect and separate the hydrocarbon/water mixtures that drain from the saturated soil. This procedure will result in effective drainage of the wet soils and avoidance of

any leakage during transport. The collected water will be trucked to the lined pond at Kutz Plant or a permitted disposal site.

9. Absorbent pads will be available on site to facilitate the capture of any free-phase hydrocarbons.

10. The water from the excavation area will be pumped into water trucks and hauled to the Kutz lined pond or a permitted disposal site. In accordance with OCD recommendations the Kutz pond liner was inspected for leaks, tears, etc. and none were found.

A containment "sweep" will be on site at the Kutz pond. The sweep will be used to move and capture any free-phase hydrocarbons to one side of the pond. The hydrocarbons will be collected in a pump truck and transported to the Blanco Plant for recycling.

11. Hydrocarbon contaminated soil will be removed to Ballard Station and spread in six inch lifts. The soil will be disced twice a week for the duration of the excavation activities. After excavation is completed, random samples will be collected and analyzed for TPH and BETX to ensure aeration was successful. If the TPH, BETX or benzene results exceed 100 ppm, 50 ppm or 10 ppm, respectively, the soil will be disced until the results do not exceed these limits.

12. A temporary trench will be constructed just down gradient of the excavation area to serve as an observation trench. It will provide visual verification that no floating hydrocarbons escaped during the excavation process. It will also allow natural volatilization of any remaining dissolved hydrocarbons. An agitator will be installed in the trench to act as a final treatment "air stripper" and aeration system. It will be kept in use for a short period (60 - 90 days) after soil remediation is completed to allow flushing of the replacement soil. Water samples will be collected at the beginning of the excavation, at seven day intervals during excavation or if a hydrocarbon sheen is observed and at thirty day intervals after remediation until the trench is closed. The samples will be analyzed for BETX.

13. All former NMOCD monitoring wells will be removed. In addition, all EPNG monitoring wells in the excavation area will be removed.

14. After remediation, well S-1 will be pumped for 72 hours and sampled. Every thirty days thereafter, S-1 will be pumped for 72 hours and sampled. This will continue for a period of six months. The samples will be analyzed for BETX, TPH, EC, sodium, calcium, potassium, magnesium, chloride, sulfate, nitrate, carbonate and bicarbonate. If the sample results are satisfactory, no additional sampling will be performed. The water from S-1 will be discharged into the field west of the location.

15. The necessity for additional monitoring wells will be assessed with NMOCD after remediation activities are completed and sample results are received.

16. A replacement water well will be provided for Flora Vista to replace S-1. S-1 will be capped and abandoned and the replacement well located downstream of the existing well field.



TO: Tom Hutchins

FROM: John Lambdin

MEMORANDUM

DATE: March 6, 1991

PLACE: North Engineering Laboratory-Farmington

RE: TOTAL PETROLEUM HYDROCARBON METHOD AND EQUIPMENT FOR FIELD TESTING

Please find enclosed the reference material for the Method and equipment which I have found to be suitable for our application at Manana Mary Wheeler.

The cost of the unit described is approximately \$4,000. To set up in the field, we will need approximately an additional \$1,000 worth of equipment. We would need approximately two (2) weeks to get set up to perform this analysis. The unit would be a nice addition to our capabilities and would pay for itself in about 50 analyses.

Let me know, if you want us to set this up and how we can pay for it.

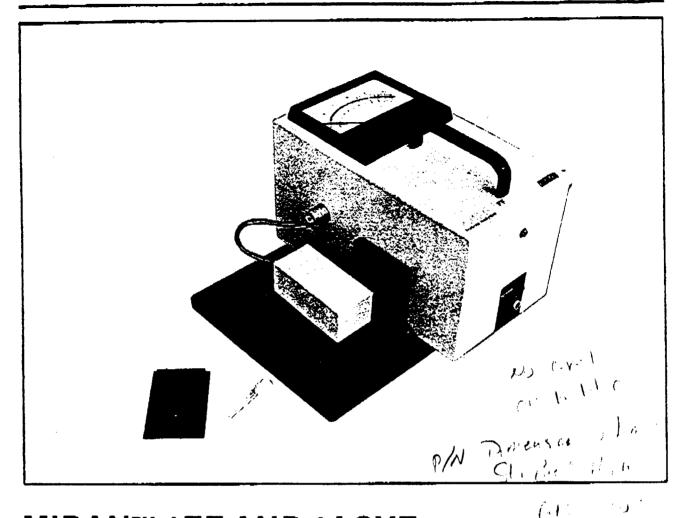
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Attachment

cc: Anu Pundari File

EM 10 0003

Laboratory Application Data



MIRAN™ 1FF AND 1ACVF LABORATORY ANALYZERS Field Screening Method for Total Petroleum Hydrocarbons (TPH) in Soil

INTRODUCTION

Contaminated soil is present at many industrial and commercial sites. It is the job of regulatory agencies, plant personnel, environmental consultants, and hazardous materials personnel to assess the degree of contamination in an accurate and efficient manner.

Screening methods have generally involved headspace or vapor analysis. The problem with using vapor analysis as a screening method arises when dealing with severely weathered samples, or samples that are not sufficiently volatile. Another method of assessing contamination involves sampling suspected contaminated areas, and transporting these samples to a laboratory. The problem with this method is slow response, and more importantly, a potential waste of money by sending samples that are un contaminated.

Foxboro has equipment designed for portable use which makes it easy to perform field screening analysis for petroleum hydrocarbons. In general, the method involves a modification of EPA Method 418.1. The modification was developed by Mobil Oil Corporation and appeared in the American Petroleum Institute Publication Number 4449.



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INSTRUMENT DESCRIPTION

For professionals who want real time answers to soil contaminated with hydrocarbons, the portable in-field solution is the MIRAN 1FF (front page) or the MIRAN 1ACVF (Figure 1) Laboratory Analyzer. These instruments can be set up to run in the field when connected to a 12 V dc inverter. This allows the unit to be effectively and accurately switched from ac to dc current. The inverter can be obserted from a vehicle's 12-volt battery.

The MIRAN 1FF is a single beam spectrophotometer with direct meter readout in Absorbance Units (AU) or percent transmission (%T). The MIRAN 1ACVF comes equipped with a circular variable filter which is user selectable in the range of 2.5 to 14.1 micrometres.

FIELD USE

Both the MIRAN 1FF and 1ACVF infrared spectrophotometers are relatively small and rugged. Each unit has an obtional carrying case for use in the field. Because these instruments weigh approximately 5.4 kg (12 lb) and are approximately 140 x 140 x 250 mm (5.5 x 5.5 x 10 in) in size, they can be easily oberated from a van or the trunk of a car. Each unit has a sturdy steel base platform which allows the unit to be placed on the floor of a van or on another flat surface. When the MIRAN 1FF or 1ACVF is taken into the field, the 110 V ac needed is supplied by a power inverter connected to a dc source such as a car or truck battery. These inverters are lightweight and small.

This screening method was designed specifically for field use. Therefore, the amount of equipment required is limited. All sample extraction is done in the sample container. No separatory funnels or elaborate extraction apparatus are needed. The Freon 113 and hydrocarbon extract is separated from the soil and mixed with silica gel. The extract is then placed in a cuvette for analysis. Most importantly, however, is the fact that this analysis is easily and accurately performed by technicians with no chemistry background. The entire analysis time is approximately 15 to 20 minutes per sample.

The information derived from this method will aid the user in determining the presence of any cetroleum hydrocarbons. Chiefly however, this method will aid the user in determining the presence of nonvolatile constituents in soil or sludge. Users of this method, we believe, will have already screened the areas for the presence of volatile organics with an instrument such as the Organic Vapor Analyzer (OVA) manufactured by The Foxboro Company. If high readings are generated by an OVA, then there would be no reason to continue; the sample would be collected and sent to the laboratory.

The MIRAN 1FF and 1ACVF will, according to the following laboratory documentation, accurately detect the presence of nonvolatile and volatile hydrocarbons. With this information the users can systematically and cost effectively screen soil samples for determination of additional laboratory testing.

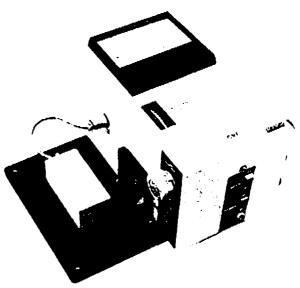


Figure 1. MIRAN 1ACVF

CALIBRATION STANDARDS

Several calibration standards were prepared by diluting stock solutions and bringing to volume with Freon 113. The formula is:

(Eq. 1) (C1)(V1) = (C2)(V2)

where, C1 = Stock Concentration (ppm)

V1 = Stock Volume Pipetted (mL)

C2 = Prepared Concentration (ppm)

V2 = Final Standard Volume (mL)

Experimental data has shown that the parts per million divided by Absorbance Unit (ppm/AU) relationship is linear. We recommend, therefore, that users generate a one point calibration curve at 500 ppm oil in Freon. Calibration curves were then generated by measuring the absorbance (1A range), on our MIRAN 1FF at room temperature, and plotting it versus the prepared concentrations in ppm (mg hydrocarbon/kg Freon 113). The calibration curves for various hydrocarbons have been generated and are shown in Figures 2, 3, and 4.

Equipment Required

- 1. Reaction vessels: 8 oz or 15 oz french square bottles with teflon lined caps (1/analysis)
- 2. Scale: Accurate to \pm 0.1 g and having a minimum capacity of 50 g
- 3. Ring Stand Assembly with clamps
- 4. Glass Funnel: 10 cm top diameter
- 5. Whatman #41 Filter Circles: 15 cm
- 6. Glass Wool
- 7. Volumetric Container: 50 mL (1/analysis)
- 8. ptfe Tape: One roll of 1/2 in wide (teflon)





DISCUSSION

The differing recovery rates shown in Table 1 for sand versus soil matrices is primarily due to the porosity of the matrix. The sandy mixtures will provide a greater degree of extraction capability. In rough terms, the Freon 113 comes in contact with the particles of sand more easily, thereby allowing the hydrocarbons to be extracted by the Freon 113 solvent more efficiently. Recovery rates and analytical data are presented in Tables 2, 3, and 4 for a variety of compounds.

Table 1.
Hydrocarbon Recovery Data for Sand vs Soil Matrices

Hydrocarbon	Matrix	Percent	Standard
Type	Type	Recovered (%)	Deviation
Used No. 2	Sand	88.4%	0.30
Fuel Oil	Soil	63.7%	1.20
Used 10W-40	Sand	78.2%	1.12
Mobil Oil	Soil	74.2%	0.66
Weathered	Sand	85.8%	2.38
Gas (50%)	Soil	71.8%	1.40

Table 2.
50% Weathered Unleaded Gasoline Data for Sand and Soil Matrices

Sand Matrix			Soil Matrix			
Prepared Standard (ppm)	Absorbance 1 A	ppm from Calibration Curve	Percent Recovered (%)	Absorbance 1 A	ppm from Calibration Curve	Percent Recovered (%)
102	0.0780	88	86.3%	0.0660	74	* 72.5%
205 409	0.164	182 348	88.8%	0.124	139	67.8% 69.7%
512	0.314	425	83%	0.256	285 396	77.3%
			Avg: 85.8%	1		Avg: 71.8%

Table 3.

10W-40 Mobil Motor Oil Data for Sand and Soil Matrices

Sand Matrix			Soil Matrix			
Prepared Standard (ppm)	Absorbance 1 A	ppm from Calibration Curve	Percent Recovered (%)	Absorbance 1 A	ppm from Calibration Curve	Percent Recovered (%)
52	0.0552	42	80.8%	0.0532	38	75.0%
105	0.108	80	76.2%	0.104	75	72.4%
315	0.341	248	78.7%	0.316	234	74.3%
525	0.553	405	77.1%	0.540	395	75.2%
			Avg: 78,2%	1	· .	Avg: 74.2%

Table 4. No. 2 Fuel Oil Data for Sand and Soil Matrices

	Sand Matrix			Soil Matrix		
Prepared Standard (ppm)	Absorbance 1 A	ppm from Calibration Curve	Percent Recovered (%)	Absorbance 1 A	ppm from Calibration Curve	Percent Recovered (%)
26				0.0220	16	61.5%
51	0.0580	45	88.2%	0.0440	33	64.7 %
77				0.0600	46	59.7%
102				0.0880	66	64.7%
128				0.116	69	61.8%
153	0.172	135	88.2%			
256	0.296	228	89.1%	0.216	165	64 8%
512	0.586	450	87.9%	0.456	351	68.5%
`*			Avg: 88.4%	1		Avg: 63.7 %



Note

The calibration curve that was generated for (mg oil)/(kg Freon) is linear. This allows the user to set up a one point calibration curve. We suggest a calibration point representing 500 ppm oil in Freon. This curve, although only a single point, will provide a very accurate calibration which will be the basis for back-calculating the milligrams of hydrocarbon in the soil.

One observation not noted in the analytical charts is the additional percent recovery achieved by performing a secondary extraction of the soil. The second extraction step requires shaking for five additional minutes. Although this step does increase the recovery rate, the significance of the second extraction may not be necessary if the user suspects the sample to be highly contaminated.

Finally, the volatility of the petroleum hydrocarbons was compared. The results indicate that each of the three hydrocarbons contains only a small fraction of volatile components. This suggests that to analyze the sample properly, an extraction procedure must be carried out.

CONCLUSION

The MIRAN 1FF and MIRAN IACVF Portable Spectrophotometers are effective tools for screening soil samples for TPH. An extraction technique is a cost effective method of detection for identifying the presence of nonvolatile hydrocarbons. With the ever increasing need for fast and reliable analytical information, the analysis of samples at the site not only provides fast and reliable information, but will immediately save the client and consultant money by making more efficient use of excavation equipment and personnel. The screening method described in this document will provide that information.

REFERENCES

D. DeAngelis, Mobil Oil Corporation; Manual of Sampling and Analytical Methods for Petroleum Hydrocarbons in Ground Water and Soil;

American Petroleum Institute Publication No. 4449 Health and Environmental Sciences Department.

PORTABLE IR SPECTROPHOTOMETERS FOR FIELD SCREENING OF TOTAL PETROLEUM HYDROCARBONS IN SOIL

Mark T. Grant, Jay E. Godleski, W. Douglas Stakem The Foxboro Company, Foxboro, Massachusetts

ABSTRACT

Field screening for non-volatile organic compounds is a much needed tool for hazardous waste site investigators. It not only delivers real time answers which can help to develop a remediation scheme in a more timely fashion, but it also saves a great deal of money for the owners of the site in the way of consulting costs, unnecessary removal of soil, and downtime. The method presented here is not meant to replace the valuable tool of organic vapor analyzers, but instead is meant to provide a method to enhance the entire process. The method is designed for use by unskilled technicians as well as highly technical experts and it is meant to be performed in a field vehicle. The method is a solid-liquid extraction using Freon The extracted solution of Freon 113 and hydrocarbon is 113. analyzed by a small portable infrared spectrophotometer. .. The amount of equipment is limited and the run time of a single analysis is about 20 minutes. Sample time can be significantly reduced by employing a mechanical shaker. Analytical results have been achieved which produce a recovery rate of between 64% and 88% depending on the matrix and hydrocarbon of interest. Analysis of the costs associated with the screening method indicates that the costs of start-up and labor are covered after the analysis of as few as 45 samples.

INTRODUCTION

Contaminated soil is present at many industrial and commercial sites. It is the job of regulatory agencies, plant personnel, environmental consultants, and hazardous materials (HAZMAT) personnel to assess the sites in an accurate and efficient manner.

To accurately evaluate samples that have been substantially weathered or samples that contain a significant portion of non-volatile hydrocarbons a method other than headspace evaluation needs to be instituted.

The other issue that is at the top of the list of all the concerned parties is the timeliness. Sending samples to analytical labs generally takes 2 weeks. There is always the option of priority testing but the costs are generally doubled.

A method has been devised which employs the basic techniques of EPA Method 418.1 (Petroleum Hydrocarbons, Total Recoverable) which is a Freon 113 extraction from a liquid sample followed by infrared analysis.

The method that will be discussed here represents a Freon 113 extraction of soil followed by infrared analysis. The method is designed for use by personnel with little or no chemistry background and uses a limited number of supplies. The analytical methodology requires the use of a small portable infrared spectrophotometer. The MIRAN 1FF from The Foxboro Company, Foxboro, Massachusetts was used for the experiments discussed within this paper. The Miran 1FF can be powered by a DC inverter. Inverters are lightweight and inexpensive.

FIELD ANALYSIS

The percent recovery for each hydrocarbon is dependent upon the matrix, Lot No. of Freon 113, and the extraction time. Test samples were prepared using various concentrations of each hydrocarbon in sand and in soil matrices. The procedure used to extract the hydrocarbons with Freon 113 varies slightly depending on the condition of the matrix. A damp or wet matrix requires 3 to 5 grams of anhydrous sodium sulfate to be added to the filter system or reaction vessel. Additional anhydrous sodium sulfate may be needed for the wet matrices. If there is a distinct water layer in the sample, then the pH should be checked and adjusted to a pH of 2 or lower by using hydrochloric acid (50% by volume). Nonpetroleum hydrocarbons (decaying vegetation, for example) may be present in the samples. These more polar hydrocarbons can be separated by adding 3 to 5 grams of silica gel to the extract solution just prior to measuring its absorbance. For the purposes of this report, dry soil and sand samples were used and negligible interferences were encountered from nonpetroleum hydrocarbons. All joints/seals involved in the extraction process should have 1 to 2 wraps of 1/2 inch teflon tape to reduce sample loss error. All glassware, glass wool, and filter paper should be rinsed with clean solvent prior to their use in a particular step. The following is a list of equipment needed.

EQUIPMENT

Frank square pointer

- 1. Reaction vessels 8 oz. or 16 oz. bottles with teflon lined caps (1/analysis)
- Scale: accurate to +/- 0.1 grams and having minimum capacity of 500 grams
- 3. Ring stand with clamps
- 4. Nalgene funnel 10 cm
- 5. Whatman #41 filter circles: 15cm
- 6. Glass wool
- 7. Graduated cylinder: 100ml
- 8. Teflon tape
- 9. Silical gel: Grade 922 200 mesh (3-5g/analysis)
- 10. Anhydrous sodium súlfate: (3-5g/analysis)
- 11. Hydrochloric acid: 50% by volume (0.5ml/analysis)
- 12. Portable infrared spectrophotometer

PROCEDURE

1. Collect a 20 gram sample of the contaminated soil/sand in a reaction vessel and add 20 ml of Freon 113. Mix for 15 seconds and vent the vessel, then continue mixing for 10 minutes.

2. Slowly decant the extract, through the larger filter paper and glass wool plug into a volumetric container.

3. Add 20 ml additional solvent to the vessel, mix for 5 minutes, and then slowly decant as in step 2 above. Combine the extracts.

4. Wash the funnel system by pouring two aliquots of clean solvent (5-10 ml) through the filter paper and glass wool. Collect the washings in the same volumetric as the extract and then bring to final volume by adding clean solvent directly to the volumetric container.

5. Measure the absorbance of this solution using an infrared spectrophotometer.

6. From the calibration curve, determine the corresponding ppm reading (mg hydrocarbons/kg Freon 113) of the sample.

7. Convert the reading from the calibration curve to the actual concentration of petroleum hydrocarbons in the soil from the site. Use Equation 1.

CALCULATION

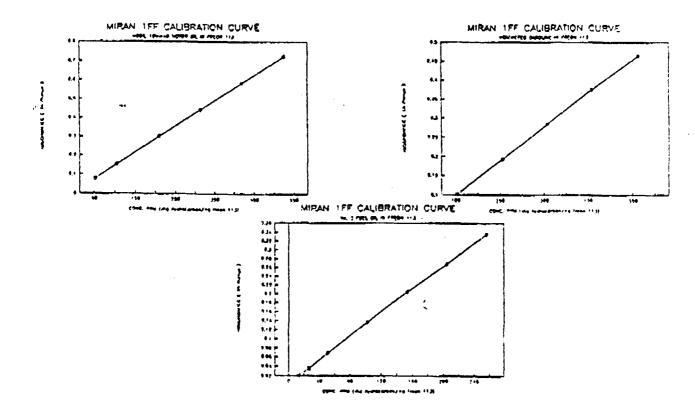
The final analysis of TPH is made in Freon 113. The calibration data from the calibration curve is central to calculating the concentration of TPH in the soil. The user will be essentially back calculating the TPH in soil concentration form the calibration curve.

There are several variables involved in the sample formula. The formula is as follows:

TABL	E 1	
CALIBRATION	CURVE	DATA

[r		Filt No. 2 F	50% Weathered Unleaded Gasoline		-	Used 1 Mobil Mo
	Measured Absorb. (1A)	Prepared Standard (PFM)	Measured Absorb. (1A)	Prepared Standard (PPM)	Measured Absorb. (1A)	Prepared Standard (PPM)
	.005	13	-	-	-	-
Ī	.009	26	~	*	.076	52
	.017	51	.100	102	. 152	105
	.034	102	. 192	205	.300	210
	.051	153	.284	307	. 444	315
	.067	204	.376	409	.584	420
	.084	256	.464	512	.724	525

The above chart represents the analytical information generated when creating the calibration curves for the three types of hydrocarbons. The graphical representations are shown below. The MIRAN 1FF was operated in the 1A range setting.



4

From the formula it is shown that the user must accurately record the weight of the soil used and the final volume of the Freon 113 used.

TABLE 2

Hydro- carbon Type	Matrix	Average Percent Recovery	Standard Deviation
NO. 2	Sand	88.4*	0.30
Fuel Oil	Soil	63.7%	1.20
Used 10W-40	Sand	78.2*	1.117
Mobil Oil	Soil	74.2%	0.66
Weathered	Sand	85.8%	2.38
Gas (50%)	Soil	71.8*	1.40

HYDROCARBON RECOVERY DATA

This chart represents average recovery rates for various matrices. The recovery rates were generated with the MIRAN 1FF and were analyzed via the described screening method.

DISCUSSION

The differing recovery rates for sand versus soil matrices is primarily due to the porosity of the matrix. The sandy mixtures will provide a greater degree of extraction capability. In rough terms, the Freon 113 comes in contact with the particles of sand more easily, thereby allowing the hydrocarbons to be extracted by the Freon 113 solvent. It should also be noted that the calibration curve that was generated for mg oil per mg Freon is linear. This allows the user to set up a one point calibration curve. We suggest a calibration point representing 500ppm hydrocarbon in Freon. This curve, although only a single point, will provide a very accurate calibration which will be the basis for backcalculating the concentration of hydrocarbons in the soil.

One observation not noted in the analytical charts is the additional gain achieved by performing a secondary extraction of the soil. The second extraction step requires shaking for 5 additional minutes. Although this step does increase the recovery rate, the significance of the secondary shaking is questionable. Obviously for a user who believes the soil to be highly contaminated, a secondary extraction is not necessary. The extraction with the first volume will put the TPH concentration over the limit.

EQUIPMENT PRICES

The user will need only a few limited non-disposable items The disposable items generally are inexpensive and should provide the users a long period of use before replenishing the supply.

NON-DISPOSABLES

1. A portable infrared spectrophotometer complete with sturdy base, cuvette holder, and cuvette. A system similar to the MIRAN 1FF or MIRAN 1ACVF from The Foxboro Company, Foxboro, MA is suitable. The price for the unit is approximately \$4,300.

- 2. A beam balance that can measure accurately to 0.1g with a range up to 500g is suitable. The price for a unit in this range is approximately \$100.
- 3. An inverter to change the DC power source of a vehicle battery to the required AC power of the MIRAN spectrophotometer. The price for a unit in this range is approximately \$60.
- 4. <u>OPTIONAL</u>: Mechanical shaker that holds up to 4 flasks for more efficient use of the time will cost approximately \$550.
- 5. Graduated cylinder will be needed for measuring Freon 113. A 100 ml Nalgene graduated cylinder will cost approximately \$8.
- 6. Beakers will be needed for mixing the silica gel. A box of twelve 100 ml beakers will cost approximately \$25.
- 7. Supports and clamps for holding funnels. A support and a ring stand will cost approximately \$30.

 Funnels for pouring and filtering solutions will be needed. A package of 6 funnels will cost approximately \$10.

DISPOSABLES

- 1. Sample jars with teflon lined lids. The volume of the jars should be preferably 500ml (250ml will also work). These jars are reusable if the user washes them and then bakes them to drive off residual hydrocarbons. The price for a case of 250ml bottles (48 bottles per case) is approximately \$70.
- 2. Freen 113 is used in a quantity of approximately 80-100ml per run. A 4 liter jug will provide enough sample for 40-50 analyses. This includes using Freen for blanks and rinses. The price for a 4 liter jug is approximately \$100.
- 3. Silica gel is used to remove polar hydrocarbons (mostly biological hydrocarbon such as fats, waxes, vegetable oils). The substance is a powder and 250 grams costs about \$50. This should be enough to last for approximately 80 analyses.
 - Solium sulfate anhydrous is a reagent used to remove trace amounts of water. The substance is a powder and

250 grams costs about \$20.

5. Filter paper is required to filter the liquid from the soil and for filtering the solvent through the anhydrous sodium sulfate. A case of 1000 sheets will cost approximately \$20.

COSTS

As with any new procedure or piece of equipment, the costs need to be compared to the benefits that are to be realized from the change. This section of the report will attempt to show all the costs associated with performing the analysis described in this paper.

DISPOSABLE COSTS

The cost for items that will wear out or chemicals that will be consumed is approximately \$260. This initial outlay will cover the analysis of 40-50 analyses. For an additional \$170 the user will have enough supply to double the number of samples to analyze. The disposable cost breaks down to \$6.50 per sample.

NON-DISPOSABLE COSTS

The cost for all non-disposable items, items that can be reused, is approximately \$4,600. If the user chooses to purchase a mechanical shaker which can be used to shake up to 4 flasks at a time, the efficiency of performing analyses in the field will be dramatically increased. The cost will increase to \$5,200.

LABOR COSTS

Generally environmental consulting firms like to have each member of their staff approach a billable rate of \$75 per hour. Therefore, this will be the associated labor cost that will be applied to the overall cost analysis.

Each analysis takes approximately 20 minutes if the analysis is being done one at a time. That is, the user is carrying out the entire analysis of one sample prior to starting another. This breaks the labor cost down to approximately \$25 per sample.

Under most circumstances the user will be analyzing multiple samples and will, therefore, be using an automatic shaker. By using an automatic shaker the analytical time is reduced to approximately 12 minutes per analysis. The labor rate is, therefore, reduced to \$15/sample. The total variable cost (labor plus materials) is \$21.50 per sample. BREAK-EVEN ANALYSIS

The goal of this method is to avoid taking samples to a lab that are over the state defined action limit or well below the action limit. Therefore, any analysis that is done needs to be based on the savings that are generated from not taking samples to a laboratory. A typical environmental lab will charge \$80-\$100 for Method 9073 or Method 418.1 (TPH analysis in soil via IR). The price of \$80 will be the base price used for the analysis. The Break-Even Volume is calculated as follows:

> FC + TVC (X) S FC = Fixed Cost: \$5,200 TVC = Total Variable Cost/Sample (labor & materials) \$21.50 X = Break-Even Volume S = Savings generated (labcost-TVC): \$58.50

For our example:

In many cases lab analyses are required on a priority basis. Although most labs provide a sliding scale, 48 hour turnaround generally carries a 100% surcharge. In these cases the Break-Even Volume is significantly decreased. Using the above formula the Break-Even Volume is:

The user can also calculate the Break Even Volume based on a combination of priority cases and normal turnaround cases. In this example we will assume 25% require priority treatment of 48 hours and 75% require normal treatment. The Break-Even Volume equation now becomes:

\$5,200 + \$21.50 X ------ = X X = 92 samples .25(\$160-\$21.50) + .75(\$80-\$21.50)

CONCLUSION

The use of a portable spectrophotometer for field analytical screening of Total Petroleum Hydrocarbons is an extremely valuable tool for identifying sources of non-volatile hydrocarbons at contaminated sites. The method of solidliquid extraction is the only true means of identifying high boiling or non-volatile organics.

The equipment required to perform the analysis of TPH in soils is limited, making the analysis all the more easily performed in a field vehicle. The analytical time required to perform the analysis can be significantly reduced by using a portable shaker. Actually the time associated with each analysis can be significantly reduced by decreasing the shaking time for each sample. Analytical work has been done which indicates

FOXBORO The Foxboro Company

Environmental Monitoring Operations

P.O. Box 500

600 North Bedford Street Message No. 525-MTG East Bridgewater, MA 02333 EL Paso Natura (Ges (Date Company, Male Grant ď Signed To - 226, О 21 Fax No This telefax consists of ____ _ pages -- including this page. MESSAGE any 9 18 W The Foxboro Company MARK T. GRANT Merketing Specialist P. O. 807 500 600 N. Bedford St. East Bridgewaler, MA (2333) Telephona 508-378-5548 Fax 508-378 5503 OXBORU

Telephone: (508) 378-5400

Fax No: (508) 378-5505



OIL CONSERVED ON DIVISION RECEIVED

P. O. BOX 1492 EL PASO, TEXAS 79978 PHONE: 915-541-2600

'91 FEB 25 AM 9 51

February 21, 1991

Mr. David Boyer Environmental Bureau Chief New Mexico Oil Conservation Division P.O. Box 2088 State Land Office Building Santa Fe, New Mexico 87504

Re: Contamination at Manana-Mary Wheeler #1E Gas Well Site, Flora Vista, New Mexico

Dear David:

El Paso Natural Gas Company ("El Paso") recognizes that the New Mexico Oil Conservation Division ("OCD") has the jurisdiction necessary to demand that El Paso and all other responsible parties remediate the site of the Manana-Mary Wheeler #1E gas well located in Unit M (SW/4, SW/4) Section 23, Township 30 North, Range 12 West, NMPM, San Juan County, New Mexico. Further, El Paso recognizes that OCD has the power to enforce such a requirement through a properly issued order.

Based on OCD's authority to require El Paso to conduct the necessary cleanup activities, El Paso is prepared to do so. Weather permitting, El Paso will begin cleanup activities less than one month from the time El Paso receives OCD's written acknowledgement that El Paso's remediation plan, contained herein, is sufficient and adequate to accomplish the required remediation.

In response to your demands that remediation be conducted at this site, El Paso will cleanup the contamination at the Manana-Mary Wheeler #1E gas well near Flora Vista, New Mexico according to the following procedures:

1. The reserve pit and the area of soil visibly stained by hydrocarbon will be excavated until all of the visibly stained soil is removed or to one foot below groundwater, whichever is most efficient. Said area was delineated during the site investigation study. A "strip mining" approach will be used, in which clean

> overburden soil is removed and stockpiled prior to excavation of the underlying stained soil.

- 2. The excavation will gradient. begin up Overburden to a depth of four to five feet, or until hydrocarbon stained soil is encountered, will be removed and stockpiled for reuse as final backfill after the remediation is complete. Only clean overburden soil will be stockpiled and used for backfill. Any clean soil mixed with stained soil will not be used for backfill.
- 3. Before commencing the excavation activities, El Paso will remove all its ancillary equipment from the site. After the remediation is complete, El Paso will reinstall its ancillary equipment.
- 4. In order to assure worker safety and access to the site, the operator must perform the following:
 - a. shut-in the gas well for the duration of the excavation activities;
 - b. remove the operator's ancillary production units (e.g., the separator and storage tank) prior to excavation activities;
 - c. assure that the house located near the well which presently receives gas directly from the Manana-Mary Wheeler #1E well is provided substitute gas service during the shut-in period.

If the operator refuses to perform the above activities, El Paso may seek OCD's assistance to achieve the above.

Once the excavated area has been backfilled, the operator may return its ancillary units and metering equipment to the appropriate locations and place the gas well back in service.

5. El Paso understands that buried tanks installed after the remediation is complete must be equipped with leak detection monitors and all above ground tanks must be set on concrete pads with curbing.

- To augment the clean overburden which will be 6. placed back into the excavated area, a borrow will site with coarse-grained sand be Most likely, the borrow site will designated. be near Ballard Plant, which is also the disposal site for the contaminated soil. Once a borrow site is designated, El Paso will notify OCD as to the site's location and a representative sample of the borrow material will be provided.
- 7. A small amount of ground water will probably be encountered. However, any water observed to have a hydrocarbon sheen will be removed during excavation. A portable pump will be used to pump the water from the excavation area during soil removal. The water will be discharged into tank trucks and taken to the lined pond at El Paso's Kutz Plant.
- 8. In the event it becomes necessary to excavate below the water table, the saturated soil will be placed in dump trucks. The dump truck tailgates will be pinned and the beds tilted over troughs designed to collect and separate the hydrocarbon/water mixtures draining from the saturated soil. This procedure will result in effective drainage of the wet soil and avoidance of any leakage during transport. The collected water will be trucked to the lined pond at Kutz Plant or some other legally permitted disposal site.
- 9. Absorbent pads will be available on site to facilitate the capture of any free-phase hydrocarbons.
- 10. The water from the excavated area will be pumped into water trucks and hauled to the Kutz lined pond or some other legally permitted disposal site. In accordance with OCD recommendations the Kutz pond liner was inspected for leaks, tears, etc. and none were found.

A containment "sweep" will be available at the Kutz pond. If necessary, the sweep will be used to move and capture any free-phase hydrocarbons in the pond. The hydrocarbons

> will then be collected in a pump truck and transported to the Blanco Plant for recycling.

- 11. Hydrocarbon stained soil will be removed to Ballard Station and spread in six inch lifts. The soil will be disced twice a week for the duration of the excavation activities. After excavation is complete, random samples of the disced soil will be collected and analyzed for TPH and BETX to ensure aeration. If the TPH, BETX or benzene exceed 100 ppm, 50 ppm or 10 ppm, respectively, the soil will be disced until the sample results do not exceed the above limits.
- A temporary trench will be constructed down 12. gradient of the excavation area to serve as an observation trench. The observation trench will provide visual verification that floating hydrocarbons are removed during the excavation process. In addition, the trench will allow natural volatilization of any remaining dissolved hydrocarbons. An agitator will be installed in the trench to act as a final treatment "air stripper" and aeration system. The agitator will be kept in use for a short period (60 - 90 days) after soil remediation is complete to allow flushing of the replacement soil. While the trench is open, water samples will be collected at seven day intervals during excavation and more often if a hydrocarbon sheen is observed and at thirty day intervals once remediation is complete. The samples will be analyzed for BETX.
- 13. All OCD monitoring wells will be removed. In addition, all El Paso monitoring wells in the excavation area will be removed.
- 14. After remediation is complete, well S-1 will be pumped for 72 hours and sampled. Every thirty days thereafter, S-1 will be pumped for 72 hours and sampled. This will continue for a period of six months. If the sample results are satisfactory, no additional sampling will be performed. The water from S-1 will be discharged into the field west of the location.

1

- 15. The necessity for additional monitoring wells will be evaluated mutually by El Paso and OCD after remediation activities are complete and sample results received.
- 16. Either an alternate source of water or a replacement water well will be provided for the community of Flora Vista. Water Well S-1 will be capped and abandoned and any replacement well will be located downstream of the last well in the existing well field.

El Paso's actions taken in response to your demands to remediate the Manana-Mary Wheeler #1E gas well site should not be construed as a waiver of its rights to contribution from other responsible parties.

If you have any questions with regard to this plan, please feel free to contact me or Jamye Boone Ward.

Very truly yours,

Thomas D. Autchine

Thomas D. Hutchins Manager, North Region Engineering

WPPENV32:3

STATE OF NEW MEXICO



ENERGY, MINERALS AND NATURAL RESOURCES DEPARTMENT

OIL CONSERVATION DIVISION

January 11, 1991

BRUCE KING GOVERNOR POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87504 (505) 827-5800

CERTIFIED MAIL RETURN RECEIPT NO. P918-402-436

Mr. Thomas D. Hutchins, Manager North Region Compliance Engineering El Paso Natural Gas Company P.O. Box 1492 El Paso, TX 79978

> Re: Site Investigation/Remediation for the Manana-Mary Wheeler #1E Gas Well Flora Vista, New Mexico

Dear Mr. Hutchins:

On December 27, 1990, I met with you and other representatives of EPNG, and Mr. William Carr representing Manana Oil, to discuss remediation at the above site. At that time an undated draft document entitled "Replacement Water Well Option" was presented to OCD for review and comment. The draft proposal was discussed with me during the meeting and has also been reviewed by other staff members. This letter transmits our agency's comments and requirements on the proposal. Comments are numbered as presented in the draft proposal.

- 1. The lower boundary depth of six feet is approximate. The actual depth to which evacuation will be performed is dependent on the presence of hydrocarbon; at some locations it may exceed six feet. Likewise, hydrocarbons may be present for some distance beneath the water table and be released during excavation. If hydrocarbons do exist beneath the water table, soil and hydrocarbons need to be excavated as discussed in Item #7.
- 2. As discussed at the meeting, it is necessary to remove the reserve pit. Also, due to previous investigative digging and trenching, overburden may not be clean since the trenches were not backfilled in any particular sequence.

- 3. Leak detection will be required on all condensate or produced water storage tanks installed below or partially below grade. Above grade tanks should be placed on a pad such that bottom leaks are detected at the edge of the pad. Submit plans and specifications for all storage units to OCD prior to reinstallation.
- 4. Propose a procedure if the random samples show high **TPH** or **BTEX**. Are the pits to remain open to await receipt of the test results? If they are closed and results are high, what action is proposed? (A possible solution is portable onsite sampling/detection equipment calibrated for the known constituents. Laboratory testing would only be used for final confirmation of field tests.)
- 5-9. Acceptable.
- 10. At the conclusion of discing, hydrocarbon levels of the random soil samples analyzed for TPH, total BTEX and Benzene, should not exceed 100 ppm, 50 ppm and 10 ppm, respectively (UST standards).
- 11. OCD requests that the observation trench be sampled only for BTEX and conductivity at the beginning of the excavation and at seven day intervals during active site work, or if floating hydrocarbons or hydrocarbon sheens are observed. Upon completion of work, samples will be taken at thirty day intervals until closure.
- 12. Accepted.
- 13. During the 30-day period of testing Well S-1, how will the water be disposed? Disposal into a water course or surface water body requires an NPDES permit. Although I believe that quality will not be a problem, EPNG should perform some physical observation of the pumped water and note color, turbidity, odor, and taste. If any significant deterioration is observed, pumping should cease and OCD notified.
- 14. It is likely that some monitoring wells will be necessary to monitor at least one complete seasonal cycle. However, long term use of these wells will be necessary if water wells continue to supply the water system, but sampling would be infrequent. Final decisions on numbers, locations and sampling can await completion of remedial actions.
- 15. This item is dependent on separate negotiations between the parties and not part of OCD's required remediation.

Based on these comments, you are requested to present a revised proposal to OCD within thirty (30) days of receipt of this letter. Review of final plans and approval of remediation procedures will allow the cleanup to begin in late winter this year as planned.

If you have any questions on this letter, please contact me at (505) 827-5812.

Sincerely,

L. Boys DAVID G. BOYER, Hydrogeologist

Environmental Bureau Chief

DGB/dp

cc: David Catanach - OCD OCD District Office - Aztec John Eichelmann, EPNG William Carr, Campbell & Black Perry Pearce - Montgomery & Andrews

EPNG meeting 12/27/20 EL Paso Natural Gat: John Erchelmann Tom WRight Tom Hutchins Jayme Ward Manona Oil: Bill CARR Meeting to discuss septement status of remediation efforts at flora Vista. EPNG required to present site investigation remediation for review, and to preven OCA from Scheduling a hearing requiring parties to show calle why enforcement action should not be taken to require clean up (removal of soils, ground water Treatment) EPNG presenter plant remediation, io, tha replacement well option for the community, but stated that the actual replacement 1 option had not and been decided being of stalled negotiations with adjecent communities Since the actual physical cleanup does not Lepens on the type of replacement water supply, EPNG proposed to more check without cen equement soon the community

З.

Boye gave his first-cut opinion of the proposal and stated that, to although 5 some changes would be necessary and The stoff need to review the baje structure was sound. Och will comment within two weeks. EPNG toid They expect to freable to start work in Elereer ofter receiving OCD approval. 5 (Notes of A Boys) -· · · · · · · . . · · · · · · · · · ·

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REPLACEMENT WATER WELL OPTION

Mr. David Boyer Environmental Bureau Chief New Mexico Oil Conservation Division P.O. Box 2088 State Land Office Building Santa Fe, New Mexico 87504

Keciever OCP 12/27/90

SUBJECT: Site Investigation/Remediation for the Manana -Mary Wheeler #1E Gas Well Site in Flora Vista, New Mexico

Dear Mr. Boyer:

The following outlines the proposed procedures and plan for the remediation activities at the subject well site.

1. A 15,000 square foot area of visibly hydrocarbon stained soil will be excavated to a depth of about six feet. The area was delineated during the site investigation study.

A "strip mining" approach will be adopted, whereby clean overburden is removed and stockpiled prior to excavation of the underlying hydrocarbon stained soil. The upper boundary of the hydrocarbon zone should be coincident with the high water level. The lower boundary of the zone is expected to be approximately six feet below grade as discussed with your staff.

- 2. The excavation will begin up gradient. Overburden to a depth of four to five feet, or until hydrocarbon stained soil is encountered, will be removed and stockpiled for reuse as final backfill after the remediation is completed.
- 3. For safety and access measures, the gas well will be shut-in for the duration of the excavation portion of the remediation effort. All ancillary production units, such as the separator and fiberglass storage tank will be removed prior to initiating excavation operations. The meter house and meter run will also be removed. Once the excavated area has been backfilled, these units will be returned to their locations and the gas well will be returned to service.
- 4. Soil from the bottom of the excavation will be randomly sampled for TPH and BETX. The excavation will take approximately twenty working days.
- 5. To augment the clean overburden that will be placed back into the excavation area, a borrow site will be located where clean, coarse-grained sand can be obtained. We would like to identify a borrow site near Ballard Plant, the disposal site for the contaminated soil. Once a borrow site

is located, OCD will be notified of the location and a representative sample of the borrow material will be provided.

6. Since the excavation depth is estimated to be about six feet, we anticipate a small amount of water will be encountered. A portable pump will be used to pump the water from the excavation area during soil removal.

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7. If it is possible to excavate the entire hydrocarbon zone without reaching the water table, the problem of water handling and disposal will not be an issue.

However, in the event it is necessary to excavate below the water table, the saturated soils will be placed into waiting dump trucks. The dump truck tailgate will be pinned and the bed tilted over a trough designed to collect and separate the hydrocarbon/water mixtures that drain from the saturated soil. This procedure will result in effective drainage of the wet soils and avoidance of any leakage during transport. The collected hydrocarbon contaminated water will be trucked to the lined pond at Kutz Plant or a permitted disposal site.

- 8. It is not anticipated that large volumes of hydrocarbons will be liberated. However, absorbent pads will be available on site to facilitate the capture of any free-phase hydrocarbons.
- 9. The water from the excavation area will be pumped into water trucks and hauled to the lined pond or a permitted disposal site. In accordance with OCD recommendations the Kutz pond liner was inspected for leaks, tears, etc. and none were found.

A containment "sweep" will be on site at the Kutz pond. The sweep will be used to move and capture any free-phase hydrocarbons to one side of the pond. The hydrocarbons will be collected in a pump truck and transported to the Blanco Plant for recycling.

- "10. Hydrocarbon contaminated soil will be removed to Ballard Station and spread in six inch lifts. The soil will be disced twice a week for the duration of the excavation activities. Random samples will be collected and analyzed for TPH and BETX to ensure aeration was successful.
- 11. A temporary trench will be constructed just down gradient of the excavation area to serve as an observation trench. It will provide visual verification that no floating hydrocarbons escaped during the excavation process. It will also allow natural volatilization of any remaining dissolved

hydrocarbons. An agitator will be installed in the trench to act as a final treatment "air stripper" and aeration system. It will be kept in use for a short period (60 - 90 days) after soil remediation is completed to allow flushing of the replacement soil.

- 12. All former NMOCD monitoring wells will be removed. In addition, all EPNG monitoring wells in the excavation area will be removed.
- 13. After remediation, well S-1 will be sampled. The well will then be pumped at rates representative of those used during the time when the well was used as a water production well. S-1 will be pumped for 30 days and then sampled immediately. The well will then be shut in for a period of two weeks and again sampled. The groundwater samples will be analyzed for BETX, TPH, pH, EC, sodium, calcium, potassium, magnesium, chloride, sulfate, nitrate, carbonate and bicarbonate.
- 14. The necessity for additional monitoring wells will be assessed with NMOCD after completion of the remediation project.
- 15. A replacement water well will be provided for Flora Vista to replace S-1. We will provide the Flora Vista Municipal Water Users Association funds to drill, complete and tie in a well and leave it to their discretion to locate the well.

We believe that the remediation plan outlined above will result in successful site remediation. Most of the remediation requirements discussed with your staff and stated in Mr. Catanach's letter dated July 18, 1990 will be met after completion of the project.

Since groundwater in much of the San Juan basin is known to exceed inorganic WQCC and/or USEPA standards regardless of the public water well's proximity to a natural gas production facility, we believe that the requirement for meeting WQCC or USEPA drinking water standards for inorganic contaminants is beyond the scope of this project.

We hope the remediation plan outlined above meets with your approval. If you have any questions regarding the remediation project, please call me at (915) 541-3531.

Sincerely,

Thomas D. Hutchins North Region Compliance Manager



CAMPBELL & BLACK, P.A.

JACK M. CAMPBELL BRUCE D. BLACK MICHAEL B. CAMPBELL WILLIAM F. CARR BRADFORD C. BERGE MARK F. SHERIDAN WILLIAM P. SLATTERY JEFFERSON PLACE SUITE I - IIO NORTH GUADALUPE POST OFFICE BOX 2208 SANTA FE, NEW MEXICO 87504-2208 TELEPHONE: (505) 988-4421 TELECOPIER: (505) 983-6043

November 8, 1990

HAND-DELIVERED

RECEIVED

NOV 8 1990

OIL CONSERVATION DIV. SANTA FE

Mr. David G. Boyer, Chief Environmental Bureau New Mexico Oil Conservation Division Energy, Minerals and Natural Resources Department State Land Office Building Santa Fe, New Mexico 87503

Re: Contamination at Mary Wheeler No. 1-E Gas Well Site, Flora Vista, New Mexico

Dear Mr. Boyer:

This letter confirms our conversations of November 5, 1990 in which Perry Pearce and I advised you that negotiations were continuing between representatives of Manana Gas Inc. and El Paso Natural Gas Company concerning the above-referenced matter. It is our belief that progress is being made toward an agreement whereby action can be undertaken to remedy this situation, subject to OCD approval.

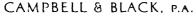
The parties are reviewing information concerning the details of a proposed clean-up action and hope to meet again during the week of November 19, 1990 to conclude their negotiations and to finalize a proposal for your consideration.

I will keep you advised of all further developments.

ry truly yours

WILLIAM F. CARR ATTORNEY FOR MANANA GAS INC. WFC:mlh cc: Robert G. Stovall, Esg.

Robert G. Stovall, Esq.
 W. Perry Pearce, Esq.
 Jayme Boone Ward, Esq.



LAWYERS

JACK M. CAMPBELL BRUCE D. BLACK MICHAEL B. CAMPBELL W LIAM F. CARR BRADFORD C. BERGE MARK F. SHERIDAN WILLIAM P. SLATTERY PATRICIA A. MATTHEWS

RECEIVED

AUG 1 6 1990

OIL CONSERVATION DIV. SANTA FE JEFFERSON PLACE SUITE 1 - 110 NORTH GUADALUPE POST OFFICE BOX 2208 SANTA FE, NEW MEXICO 87504-2208 TELEPHONE: (505) 988-4421 TELECOPIER: (505) 983-6043

August 16, 1990

HAND-DELIVERED

Mr. David R. Catanach Deputy Director Oil Conservation Division New Mexico Department of Energy, Minerals and Natural Resources State Land Office Building Santa Fe, New Mexico 87503

> Re: Contamination at Manana-Mary Wheeler No. 1E Gas Well Site, Flora Vista, New Mexico

Dear Mr. Catanach:

This letter confirms conversations which I had on this date with Mr. David Boyer and Mr. Bob Stovall of the Oil Conservation Division staff concerning the above-referenced matter.

I have been retained to represent Manana Gas Inc. in this case and have been in negotiations with W. Perry Pearce, attorney for El Paso Natural Gas Company, concerning the clean-up of the above-referenced well site. We have scheduled a meeting between El Paso and Manana representatives on September 7, 1990 to review the plans that have been developed between the Oil Conservation Division and El Paso Natural Gas Company and discuss those actions necessary for a successful clean-up of this site. This meeting is scheduled on this date since certain key individuals are out of town until that time. Therefore, we will be unable to respond to your July 18th letter until after that date.

Mr. David R. Catanach Deputy Director Oil Conservation Division New Mexico Department of Energy, Minerals and Natural Resources August 16, 1990 Page Two

Accordingly we request an extension until September 12, 1990 to respond to the Oil Conservation Division with an outline for a remediation plan.

Very truly yours,

WILLIAM F. CARR WFC:mlh cc: Mr. Ed Hartman, President Manana Gas, Inc. Post Office Box 14069 Albuquerque, New Mexico 87191

W. Perry Pearce, Esq.Montgomery & Andrews, PA325 Paseo de PeraltaSanta Fe, New Mexico 87501

David Boyer
 Environmental Bureau Chief
 State Land Office Building
 Santa Fe, New Mexico 87503

Robert G. Stovall, Esq. General Counsel Oil Conservation Division State Land Office Building Santa Fe, New Mexico 87503

Oil Conservation Division District III Office 1000 Rio Brazos Road Aztec, New Mexico 87410 Attn: Frank Chavez Hand-Delivered

Hand-Delivered

Hand-Delivered

CAMPBELL & BLACK, P.A.

LAWYERS

JACK M. CAMPBELL BRUCE D. BLACK MICHAEL B. CAMPBELL WILLIAM F. CARR BRADFORD C. BERGE MARK F. SHERIDAN WILLIAM P. SLATTERY PATRICIA A. MATTHEWS JEFFERSON PLACE SUITE I - 110 NORTH GUADALUPE POST OFFICE BOX 2208 SANTA FE, NEW MEXICO 87504-2208 TELEPHONE: (505) 988-4421 TELECOPIER: (505) 983-6043

August 16, 1990

HAND-DELIVERED

Mr. David R. Catanach Deputy Director Oil Conservation Division New Mexico Department of Energy, Minerals and Natural Resources State Land Office Building Santa Fe, New Mexico 87503 RECEIVED

AUG 1 6 1990

OIL CONSERVATION DIVISION

Re: Contamination at Manana-Mary Wheeler No. 1E Gas Well Site, Flora Vista, New Mexico

Dear Mr. Catanach:

This letter confirms conversations which I had on this date with Mr. David Boyer and Mr. Bob Stovall of the Oil Conservation Division staff concerning the above-referenced matter.

I have been retained to represent Manana Gas Inc. in this case and have been in negotiations with W. Perry Pearce, attorney for El Paso Natural Gas Company, concerning the clean-up of the above-referenced well site. We have scheduled a meeting between El Paso and Manana representatives on September 7, 1990 to review the plans that have been developed between the Oil Conservation Division and El Paso Natural Gas Company and discuss those actions necessary for a successful clean-up of this site. This meeting is scheduled on this date since certain key individuals are out of town until that time. Therefore, we will be unable to respond to your July 18th letter until after that date.

Mr. David R. Catanach Deputy Director Oil Conservation Division New Mexico Department of Energy, Minerals and Natural Resources August 16, 1990 Page Two

Accordingly we request an extension until September 12, 1990 to respond to the Oil Conservation Division with an outline for a remediation plan.

Very truly yours,

WILLIAM F. CARR WFC:mlh cc: Mr. Ed Hartman, President

Manana Gas, Inc. Post Office Box 14069 Albuquerque, New Mexico 87191

W. Perry Pearce, Esq.Montgomery & Andrews, PA325 Paseo de PeraltaSanta Fe, New Mexico 87501

David Boyer Environmental Bureau Chief State Land Office Building Santa Fe, New Mexico 87503

Robert G. Stovall, Esq. General Counsel Oil Conservation Division State Land Office Building Santa Fe, New Mexico 87503

Oil Conservation Division District III Office 1000 Rio Brazos Road Aztec, New Mexico 87410 Attn: Frank Chavez Hand-Delivered

Hand-Delivered

Hand-Delivered



OIL CONSERV ON DIVISION RECE: ≠ED P. O. BOX 1492 EL PASO, TEXAS 79978 PHONE: 915-541-3071 9 00

JAMYE BOONE WARD ATTORNEY AT LAW

August 15, 1990

Mr. David Catanach Deputy Director Oil Conservation Division P. O. Box 2088 Santa Fe, New Mexico 87504

> Re: Contamination at Manana - Mary Wheeler No. 1E Gas Well Site, Flora Vista, New Mexico

Dear Mr. Catanach:

We are in receipt of your letter to Kenneth E. Beasley dated July 18, 1990, dealing with the contamination at the Mary Wheeler No. 1E Well Site situated in SW/4, SW/4, Section 23, Township 30 North, Range 12 West, NMPM, San Juan County, New Mexico which is in the vicinity of community of Flora Vista.

As you know, El Paso Natural Gas Company has performed extensive study of the area surrounding the well site but at this time is not prepared to submit a remediation plan. If you or other members of the Oil Conservation Division staff would like to discuss this matter with us, we will be happy to meet at your convenience.

If you have any questions about this matter, please do not hesitate to contact me or Perry Pearce at 505/986-2506.

Sincerely,

James Boone Ward

c: W. Perry Pearce, Montgomery & Andrews Kenneth E. Beasley, III, No. Region Engineering Thomas L. Wright, Office of General Counsel

WPPENV14:17

STATE OF NEW MEXICO



ENERGY, MINERALS AND NATURAL RESOURCES DEPARTMENT

OIL CONSERVATION DIVISION

GARREY CARRUTHERS

July 18, 1990

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87504 (505) 827-5800

CERTIFIED MAIL RETURN RECEIPT NOS. P-918-402-294 P-918-402-296

Mr. Kenneth E. Beasley, Manager North Region Compliance Engineering EL PASO NATURAL GAS COMPANY P. O. Box 1492 El Paso, Texas 79978

Mr. Ed Hartman, PresidentManana Gas, Inc.P. O. Box 14069Albuquerque, New Mexico 87191-4069

RE: Contamination at Manana - Mary Wheeler #1E Gas Well Site, Flora Vista, New Mexico

Dear Mr. Beasley:

This letter is to notify you that the New Mexico Oil Conservation Division (OCD) is initiating enforcement action to require removal of oil, drilling muds, contaminated soil and other materials that has caused and continues to cause ground water contamination at the Manana - Mary Wheeler #1E gas well site located in Unit M (SW/4, SW/4) Section 23, Township 30 North, Range 12 West, NMPM, San Juan County, New Mexico.

Information in OCD files shows contamination first reported in 1983 requiring shut down of a Flora Vista public water supply well because of floating oil and odors. Subsequent investigation by OCD and others in 1986, 1987 and 1989 determined the following water contaminants were present in soils and ground water:

1. In water: Benzene, ethylbenzene, toluene, xylene and polynuclear aromatic hydrocarbons. Benzene exceeded the federal drinking water standard in some samples.

- 2. In soil: Benzene, toluene, xylene, polynuclear aromatic hydrocarbons (including naphalene), chromium, and sodium chloride.
- 3. In water and soil: Oily hydrocarbons that impart a taste and odor to the water.

The presence of these contaminants in soils and ground water is due to activities at the lease by Manana and El Paso Natural Gas that include but are not limited to spills and/or disposal of drilling muds and additives, oil and grease, condensate, produced water, glycol and/or other contaminants into unlined drilling/reserve pits, dehydration pits, condensate tank drain pits, and onto the surface of the ground. The remediation requirements that must be met for successful cleanup at the site are shown in Attachment A. An outline for a remediation plan must be presented to OCD within thirty (30) days from receipt of this letter. The plan outline should consider various alternative remediation schemes in the event one or more of the Flora Vista water wells will not be used as a public water supply in the future.

This enforcement action is taken under authority granted to the OCD under the Oil and Gas Act (70-2-1 to 38 NMSA 1978), the Water Quality Act (74-6-1 to 12, NMSA 1978) and under authority to abate the public nuisance of polluting water (30-8-1, 30-8-2 and 30-8-8, NMSA, 1978). Failure to comply with this directive may result in additional enforcement action.

If you have any questions regarding the remediation measures contact David Boyer at 827-5812; please direct other questions to Robert Stovall, General Counsel, at 827-5805.

Sincerely,

David Catanach Deputy Director

Attachment

DC/DGB/sl

cc: OCD Aztec Office

Attachment A

Remediation Requirements

Successful site remediation must meet the following criteria:

- 1. Dissolved, emulsified, and free-floating petroleum, and other organic and inorganic water contaminants, must be removed from the groundwater such that:
 - a) the water quality standards of Section 3-103 of the New Mexico Water Quality Control Commission Regulations (as amended through December 24, 1987) are met; and
 - b) the USEPA drinking water standards for public water supplies in effect as of June 15, 1988; and
 - c) undesirable odors attributable to loss of petroleum or other fluids from the site are not present in ground water.
- 2. The unsaturated (vadose) zone in the vicinity of the contamination shall not contain drilling muds, inorganic salts, heavy metals, or hydrocarbons in quantities sufficient to recontaminate ground water. Contamination shall be deemed to occur if State or Federal numerical standards are exceeded, or if there is continued presence or reappearance of oil or grease, or undesirable odors in the water supply. Such recontamination shall not be allowed to occur as a result of seasonal rises in the water table, drainage, recharge events, or in any other manner. Soils contaminated with drilling muds, salts, heavy metals, or hydrocarbons, and needing to be removed to prevent continued or future contamination of groundwater, shall be excavated and disposed in locations approved in advance by OCD. Clean fill material may need to be provided to replace contaminated soils. Pumping of the affected water supply well for some length of time may be necessary to verify successful clean-up.



P. O. BOX 4990 FARMINGTON, NEW MEXICO 87499

'SG 22111 AM 9 29

June 6, 1990

Mr. David G. Boyer Environmental Bureau Chief New Mexico Oil Conservation Division P.O. Box 2088 Santa Fe, New Mexico 87504

Re: Remedial Action Plan for the Manana -Mary Wheeler #1E Gas Well Site, Flora Vista, New Mexico

Dear Mr. Boyer:

This letter confirms our conversation regarding dewatering and disposition of groundwater at the Manana- Mary Wheeler #1E Site. El Paso Natural Gas Company (EPNG) plans to construct a slurry wall or trench around the remediation site in order to restrict groundwater flow into the excavation area. At the beginning of the project, groundwater will be pumped out of the remediation area. Later, pumps will be strategically placed to remove groundwater flow into the remediation area.

We do not expect large amounts of oil to be liberated during excavation. The groundwater will be pumped into "frac" tanks and allowed to settle for an appropriate amount of time until any oil separates.

Presently, we are investigating groundwater disposal options. A disposal option is to irrigate an adjacent field. At this time, Mr. Eli Velasquez, the tenant, is considering our watering proposal.

If we obtain approval from him, we will follow your guidelines set forth during our phone conversation on June 5, 1990. This was a conference call with you, Mr. Eli Velasquez, Mr. Leonard Lord (EPNG Right-of-Way Department) and I. Mr. David G. Boyer June 6, 1990 Page 2

The guidelines set on June 5, 1990 are listed below:

1. Prior to groundwater discharge onto the alfalfa fields, the groundwater will be analyzed for Benzene(B), Ethylbenzene(E), Toluene(T), Xylene(X) and Electrical Conductivity(EC). EC will be used as an indicator of Total Dissolved Solids.

2. The benzene limit is 10 micrograms per liter. The ethylbenzene limit is 750 micrograms per liter. The Toluene limit is 750 micrograms per liter. The Xylene limit is 620 micrograms per liter. The EC limit is 2250 micromhos per cm

(Corrected to 25 C). The EC limit was set by considering the existing Total Dissolved Solids level of the background well.

3. Groundwater will not be discharged onto the fields unless it meets the requirements set above.

4. EPNG will conduct additional tests every five "frac" tank volumes. In addition to the previously mentioned constituents, as you requested, we will analyze the water for Chlorides, Sulfates, Total Dissolved Solids, Chromium, Cadmium and Barium.

If you have any questions, please contact me at (505) 599-2176.

Sincerely,

a.n. Pundani

A. N. Pundari Compliance Engineer

cc: Mr. Kenneth E. Beasley Mr. Leonard Lord Mr. Eli Velasquez STATE OF NEW MEXICO

ENERGY, MINERALS AND NATURAL RESOURCES DEPARTMENT

OIL CONSERVATION DIVISION

GARREY CARRUTHERS

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87504 (505) 827-5800

May 29, 1990

CERTIFIED MAIL RETURN RECEIPT NO. P-918-402-255

Mr. Kenneth E. Beasley, Manager North Region Compliance Engineering EL PASO NATURAL GAS COMPANY P. O. Box 1492 El Paso, Texas 79978

RE: Finalization of Site Investigation Report/Remedial Action Plan for the Manana - Mary Wheeler #1E Gas Well Site, Flora Vista, New Mexico.

Dear Mr. Beasley:

The staff of the Oil Conservation Division (OCD) Environmental Bureau have reviewed the May 16, 1990 response document that was presented and discussed at our meeting in Santa Fe on May 17th. Unless otherwise discussed herein, the responses in that document are accepted and become part of the Report and Remedial Plan.

- 1. Section 6.1, Reclamation Technologies: When final decisions are made regarding dewatering of fluids, soil excavation and reserve pit removal, provide the following information for OCD review and concurrence prior to beginning site excavation:
 - a. Specifics on the method and planned depth of dewatering and soil excavation (e.g. one or two feet below river bottom elevation at the site?), slurry wall installation (if chosen), water treatment, batch testing, and disposal. If land application of the treated water is chosen, a discharge plan is <u>not</u> necessary if the length of time of the discharge does not exceed 120 days. If a slurry wall is installed, it must be breached upon completion of restoration.
 - b. Specifics on disposal of soil removal from the excavation including disposal location, land application and treatment procedures (depth of lifts, frequency of discing), testing parameters (TPH) and proposed limits.
 - c. Additional specifics on encapsulation if that method of treatment is chosen.

Mr. Kenneth E. Beasiey May 29, 1990 Page -2-

- 2. Resampling of S-5 should include analysis of all volatile aromatics (i.e. BTEX).
- 3. At the conclusion of the first month's test pumping of S-1 after cleanup, a sample should be taken for the specified parameters at the time the well is shut-in.
- 4. If removal of some existing OCD monitoring wells is necessary as part of the excavation, OCD would prefer that all be removed since they were installed hastily and do not meet current monitoring well criteria.
- 5. OCD requests that EPNG perform random sampling of soils from the bottom of the excavations for TPH and BTEX prior to backfilling, in lieu of the sampling originally proposed to determine clean levels.

As stated at our meeting OCD will separately notify Manana of your planned activities, request cooperation in removing equipment, and notify them of deficiencies with the wastewater collection system(s). I will send you copies of our correspondence.

Thank you for your prompt and complete response to my letter and comments of March 30th. EPNG and K.W. Brown have obviously put a lot of thought and effort into this planned remediation and I look forward to its being implemented by mid-summer.

If you have any questions, please contact me at (505) 827-5812.

Sincerely,

David G. Boyer, Hydrogeologist Environmental Bureau Chief

DGB/sl

cc: OCD Aztec Office J. Eichelmann, Burlington Northern Resources S. Johnson, K.W. Brown

OCD-Eti, Flora Vista Remainin Meeting 5/17/90 10:00 in Handal and Respinse to Commits on Flore Vista Threetighton Report (OCD comments) Slarry nall to limit 6. W. 6" lift for soil disposed at at Bellind Plant dean levels in saturation 2000? 6. 1-2 fait below W.T. then take confirmation samples? One tout below level at since them All the confirmation. 3. p. 62 - Will use compressor to blow it into "1 pg 62 - sample will be taken immediately atta shortfig down pump instead at atta 2 weeks Project start date June 15, 1990 Look at reserve pit examplion. from Hor-Wester?

RESPONSES TO COMMENTS ON THE SITE INVESTIGATION OF THE MANANA -MARY WHEELER #1-E WELL SITE NEAR FLORA VISTA, NEW MEXICO

MAY 1 7 1990

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OIL CONSERVATION DIV. SANTA FE

I. Comments on Site Investigation Report

1. p.41, Table 3-4. The values for Benzo(b)fluoranthene and Indeno(1,2,3-cd)pyrene should be shown as 2 and $6 \mu g/l$ (instead of 0.002 and 0.006 $\mu g/l$). The correct values for WQCC standards for Ethylbenzene, Toluene and Xylenes (total) are 750, 750 and 620 mg/l, respectively.

Response: The table has been corrected with the appropriate values. A corrected Table 3-4 is enclosed in Attachment 1.

2. p.45. A discussion of oil and grease concentrations found in soil samples taken from trenches 6 and 7 is presented. Additional investigation is necessary to verify the actual presence of a "slug" of oil in this area, especially if analytical techniques may have contributed to false positive readings (ref. "A Comparison of Methods of Measuring Total Petroleum Hydrocarbons in Soil," 1989 NWWA/API Proceedings of the Conference on Petroleum Hydrocarbons and Organic Chemicals in Ground Water). Present a plan to perform the necessary verification.

Response: Additional soil samples will be collected in the area to provide more control with respect to delineating the impacted area. The soil samples will be collected by excavating pits, not trenches. Proposed locations are illustrated on Figure 1 (see Attachment 2). Sampling methods will involve the use of a backhoe and standard sampling equipment as described in the September, 1989, report.

With respect to analytical techniques, it is proposed that both oil and grease, and total petroleum hydrocarbons (TPH) analyses be performed. To reduce the likelihood that matrix interference is occurring, the GC analytical technique will be specified.

3. p.48. Please provide a copy of the referenced 1983 paper on risk assessment by Crouch ("The Risk of Drinking Water").

Response: The requested reference is included in Attachment 3.

4. p.48, Comment on Section 4.0 Risk Assessment. As stated in this section, risk assessment techniques are used to estimate chronic toxic potential from contaminants dissolved in ground water. In addition to dissolved contaminants, we have an oil phase that imparts visual and hydrocarbon odor characteristics to the water. The assessment technique would seem more appropriate if used after the cleanup to evaluate the risk of any remaining dissolved constituents.

Response: The goal of presenting the risk assessment prior to cleanup activities was to illustrate the current risk. It was never intended that the risk assessment would not be performed following remedial activities. The intent was to secure baseline data for comparison purposes (i.e., pre-remediation versus post-remediation).

5. p.52, Table 4.2. The contaminant concentrations listed in Table 4.2 are 1,000 times greater that actually observed (e.g., benzene concentration in EPNG-2A is 8.1 ppb not 8,100 ppb). I suspect these are transcription errors. Please verify that the risk calculations shown are for the actual concentrations and not those shown in the table.

Response: The presented risk assessment calculations were based on the water contaminant concentrations listed in the original Table 4-2. Thus, the risk factors presented in the original report are in error. The source for the error was hand-written drafts which preceded the final plates included in the report. The hand-written notation for $\mu g/l$ was mistakenly read as mg/l and then converted to $\mu g/l$; hence, the 1,000 factor error.

The contaminant levels have been corrected and are presented as a revised Table 4-2. Also, the risk assessment has been re-run, with a commensurate decrease in risk as depicted in the table. Hence, the discussion included in Section 4.2 of the report needs to be revised. Attached are revised text and tables for Section 4.2. which should be inserted into the report (Attachment 4).

6. p.57. In New Mexico, buried reserve pits commonly contain waste debris from drilling operations. It would not surprise me to find concrete fragments and scrap pipe in the pit.

Response: None.

II. Remedial Action Plan

A. Section 6.1.1. Contaminated Soil

1. p.59. Describe further the procedure for visual inspection verification of removal of contaminated soil, especially since contamination extends to sands and gravels beneath the water table. Is temporary dewatering proposed as part of the soil removal process?

Response: It is envisioned that a "strip mining" approach will be adopted, whereby clean overburden is removed and stock-piled prior to excavation of the oily/stained underlying sediments. Identifying the upper boundary of the oily zone should not pose a problem since it should be coincident with the high water level.

Identification of the lower boundary of the oily/stained zone, on the other hand, may prove to be more difficult. To aid in discerning the lower boundary, excavation operations should be undertaken during the summer when the Animas River is at its lowest stage level.

Due to an exceptionally dry summer in 1989, and low winter precipitation receipts in the mountainous region to the north of the Flora Vista area, it is anticipated that the Animas River will be at a low stage during the upcoming summer months, a condition favoring excavation efforts.

In lieu of a dewatering program, EPNG intends to construct a slurry wall around the lateral and upgradient portions of the area to be excavated in an effort to limit the volume of water to be handled during the remedial effort.

2. p.59. Removal of hydrocarbon stained soil should extend to the vicinity of well S-1 since staining was noted in that location during the 1987 excavation (See the discussion on page 2 of Dr. Blair's 1987 report).

Response: KWB&A Plate 6 illustrates the distribution of oil and grease in the sediments at the site. Comparing this plate with Blair's sketch indicates a reasonable correlation between the observed oil staining. It is anticipated that excavation will extend to the vicinity of S-1.

3. p.59. What procedures are proposed to determine the extent and the necessity for soil removal under lease structures (eg. oil/water separator, produced water tank, meter house).

Response: For safety and access measures, the gas well will be shut-in for the duration of the excavation portion of the remediation effort. All ancillary production units, such as the produced water and oil storage tanks, will be removed from the area prior to initiating excavation operations. Once the excavated area has been back-filled, these units will be returned to the production location, and the gas well will be placed back into service.

4. p.59. Describe the type and source of the material that will be deposited in the excavation to replace the contaminated soil.

Response: Much of the soil excavated from the site will not be contaminated, and will therefore be returned to the excavated area. To augment those clean sediments that are placed back into the excavations, a borrow site will be located where clean, coarse-grained sand can be obtained. This material is specified since it is easy to handle, and closely resembles those materials to be removed from the site.

At this time, a borrow pit or quarry has not been identified. Once such a facility has been located, OCD will be notified of its location and the type of materials to be procured for use at the Manana - Mary Wheeler #1-E Well Site. In addition, EPNG will acquire a representative sample of this material which will be submitted to OCD for inspection.

5. p. 60. If the contaminated soil is stored on-site prior to offsite removal, it must be stored such that oil and water drainage is intercepted prior to discharge to the bare ground. Likewise, saturated soil cannot be loaded directly into trucks such that oily water will discharge on-site or en route to the disposal location.

Response: All of the contaminated soil above the water table (i.e., in the unsaturated zone) will be excavated first, thereby minimizing drainage problems during the initial phases of the excavation effort. Moreover, if it is possible to excavate the entire oily zone without reaching the water table, the problem of drainage water will not be an issue.

In the event that it is necessary to excavate below the water table, saturated soils will be placed into waiting dump trucks. At this point, the truck bed tailgate will be pinned and the bed tilted over a trough designed to collect and separate the oil/water mixtures that drain from the saturated, contaminated soils. This procedure will result in effective drainage of the wet soils and avoidance of any leakage during transport. The collected fluids will be disposed at a location agreed upon by OCD and EPNG.

B. Section 6.1.2 Contaminated Groundwater

1. p.60. Explain the impacts to the area adjacent to well S-5. Is the reference to the oil and grease analyses or the low levels of dissolved hydrocarbons found during the 1989 investigation sampling?

Response: The reference to groundwater being impacted in the area adjacent to S-5 stems from the fact that low levels of dissolved hydrocarbons were detected in a groundwater sample taken from that well. As Plate 3 of the report indicates, 0.26 μ g/l benzene and 0.32 μ g/l toluene were found in a sample of groundwater taken from S-5 on June 16/17, 1989.

2. p.60. Additional sampling of S-5 is necessary to verify if benzene and toluene are present at trace levels shown in the 1989 site sampling. (1986 sampling also detected these at about the same levels.)

Response: Resampling of S-5 will be conducted at a date to be arranged prior to the remediation of the site. The requested analysis will be limited to benzene and toluene.

3. p.60. During the excavation, water will be made turbid with the disturbance and residual oil may be freed from the soil and float on top of the water. Equipment should be available on site during the excavation to skim and remove floating oil.

Response: It is not anticipated that large volumes of oil will be liberated during excavation. Arrangements will be made to have an oil-containment boom and/or absorbent pads as necessary to facilitate the capture of any free-phase oils that may develop. A holding tank will be made available on-site during excavation to accommodate the collection of any oil (see the response to comment A.5. above).

C. Section 6.3 Monitoring Soil

1. p.61. Soil. Explain how representative samples are to be collected from the bottom of the excavation if the excavation is not dewatered.

Response: It is presupposed that the bottom of the excavation will be accessible to samplers since either: (1) the Animas River will be at a sufficiently low stage that the contaminated soil can be fully excavated without encountering saturated conditions, or (2) the slurry wall will provide a sufficient level of groundwater containment to permit ready access to the bottom of the excavation.

It is anticipated that the sidewalls of the excavation will be separated a sufficiently large distance apart to permit the sampling team to walk out onto the floor of the excavation without incurring the risk of sidewall failure. Thus, representative soil samples will be collected with decontaminated augers or stainless-steel utensils. If sidewall stability becomes an issue, a backhoe or other piece of heavy equipment may be employed to acquire a sample of soil from the bottom of the excavation. From this, a sub-sample will be taken in a safe area, such as the original grade level on the banks of the excavation.

2. p.61. Groundwater. The December, 1989 EID/OCD sampling of well S-1 detected acetone. Pre-cleanup sampling should verify the presence or absence of that contaminant.

Response: In January, 1990, S-1 was purged and samples were collected for volatile analysis. The laboratory results do not confirm the presence of acetone. Attached are the lab results for this sampling event (see Attachment 5).

3. p.62. In addition to replacement monitoring wells, I would like to discuss digging a temporary trench just downgradient of the southwestern-most area of excavation to serve as an observation trench. It would provide visual verification (absence of product or sheens) that no floating hydrocarbons escaped the excavation. More importantly, with ground water flow at 2 to 3 feet per day, it would allow natural volatilization of any remaining dissolved hydrocarbons. With a minimum of agitation and circulation the trench could act as a final treatment "air stripper" and aeration system. It could be kept in use for a short period of time (90-120 days) after excavation and soil replacement has been completed to allow flushing of the replacement soil.

Response: EPNG is agreeable to a discussion on the use of a temporary observation trench. If used, all necessary safety and monitoring protocols will be formulated and observed.

4. p.62. Suggest a schedule for short and long term pumping and testing of well S-1 after completion of the excavation.

Response: It is recommended that S-1 be pumped at rates considered to be representative of those used during the time when that well was used as a water-production well (e.g., 50 GPM). S-1 should begin pumping no sooner than the date when all remediation activities have been completed. This approach will insure that no false positives are generated in response to the potential mobilization of contaminants due to excavation activities.

As for a pumping schedule, it is proposed that S-1 be pumped continuously for the first month. At the end of this period, it is additionally proposed that a water sample be taken and analyzed for the list of parameters specified in the response to comment C.6. below.

After the first month of pumping, to allow for the diffusion of any residual-level contaminants from clayey subsurface deposits, it is suggested that S-1 be shut in for a period of two weeks, with a sample taken at the end of this dormant period using an appropriate sampling methodology.

Once it has been demonstrated that the detected contaminant levels are below WQCC section 3-103 standards, S-1 will then be turned over to the Flora Vista Water Users Association for re-introduction into its water-distribution network.

5. p.62. Provide the number and location of the proposed monitoring wells.

Response: At this time, EPNG feels it to be inappropriate to specify the exact number and location of post-remediation groundwater monitoring wells. OCD can be assured, however, that a sufficient number of wells will be used or installed that will provide the necessary groundwater quality data to monitor conditions at the Manana - Mary Wheeler #1-E Well Site. The number and location of these wells will be established through joint communications between OCD and EPNG prior to entering into the field to implement the post-remediation monitoring program.

6. p.62. Analytical Parameters. In addition to the proposed sampling constituents, analyses of water should include major cations and anions. Soil analyses should include chromium and those PAH's detected and shown in Table 3.5 (p. 43).

Response: The following tables indicate the proposed suite of analytical parameters to be employed during the remedial activities:



Analytical Parameters for Water Samples to be Taken During Remedial Activities.

VOLATILE	
AROMATICS:	CATIONS:
Benzene	Sodium
Toluene	Calcium
Ethylbenzene	Potassium
Total Xylene	Magnesium
INDICATORS:	ANIONS:
pH	Chloride
EC	Sulfate
	Nitrate
{	Carbonate
	Bicarbonate

Analytical Parameters for Soil Samples to be Taken During Remedial Activities.

VOLATILE AROMATICS:	PAHs:
Benzene Toluene Ethylbenzene Total Xylene	Acenapthene Fluoranthene Napthalene
INDICATORS:	METALS:
pH TPH ¹	Chromium (TCLP)

7. p.63. Sampling Frequency. Monitor wells shall be removed, or properly plugged and abandoned at the conclusion of their use. This will be no later than at the time of decommissioning of the gas well, or an earlier time as approved by OCD.

Response: During excavation, several of the existing OCD- and EPNG-series monitoring wells will have to be removed since they lie in the affected area. Otherwise, the monitoring wells will be maintained, or plugged, as prescribed in the report.

D. Additional Comments

Provide a proposed timetable for the start and completion of the remedial action, including the time expected to be required for the excavation phase.

Response: EPNG continues to assemble information on project costs and available resources. It is expected that sufficient information will be available to provide NMOCD with a project start date on June 15, 1990.



¹ Total Petroleum Hydrocarbons to be determined via Gas Chromatography.

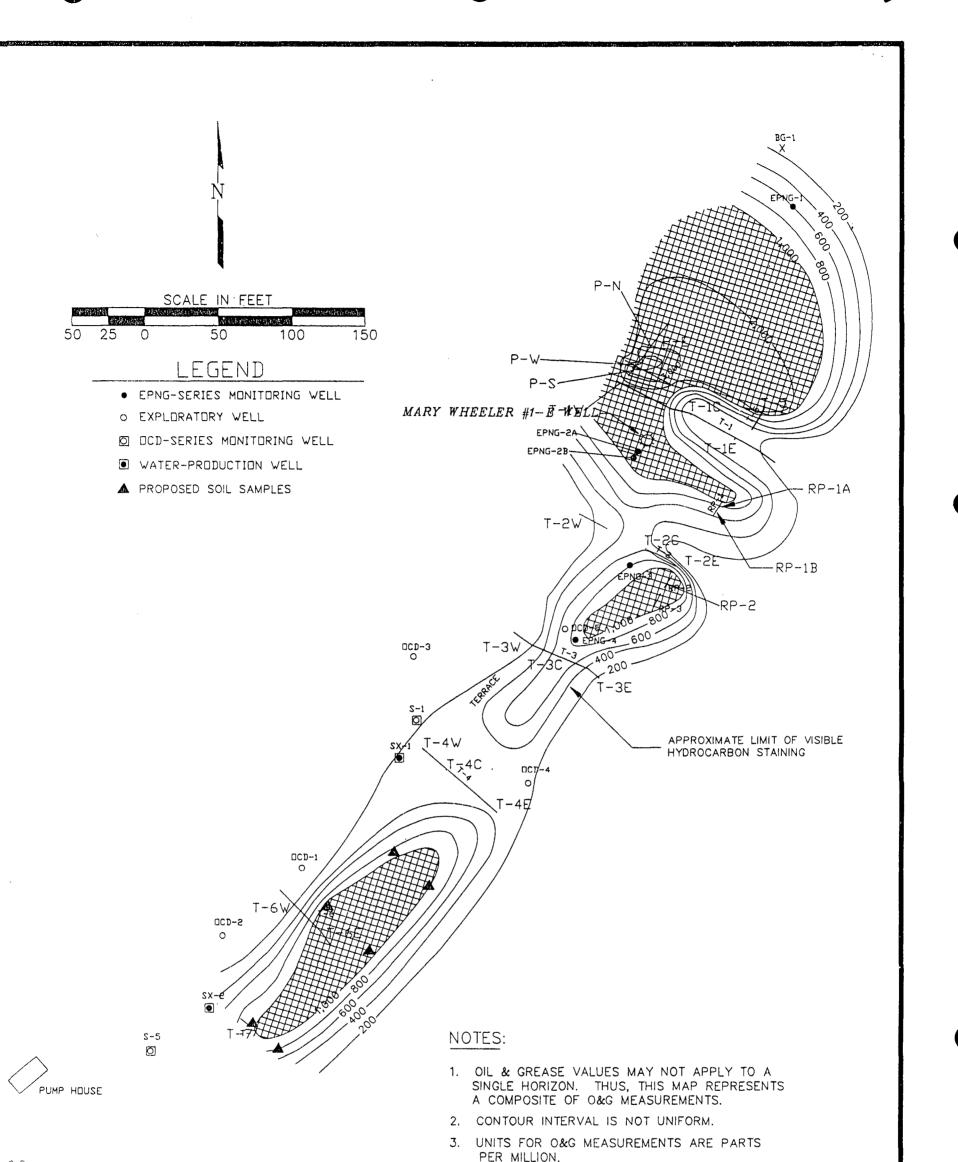
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Table 3-4. Analytical Results for Groundwater Samples.

	NM WGCC												
	Standard					Ũ	Concentration (ug/	tion (ug/	[]				
Constituent	(ug/1)	EPNG-1 EPN	EPNG-2A	EPNG-2B	EPNG-3	EPNG-4	OCD-1	OCD-2	OCD-3	OCD-4	OCD-5	S-1	S-5
Aromatic Volatile Organics:	anics:												
Benzene	10	QN	8.1	1.6	0.4	0.95	QN	QN	NS	NS	NS	ND	0.26
Ethylbenzene	750	QN	37.4	1.2	4.3	Ð	Q	QN	NS	NS	NS	QN	QN
Methylene Chloride	100	2.1	QN	QN	QN	Ð	1.3	QN	NS	NS	SN	1.6	1.2
Toluene	750		QN	QN	0.33	Ð	QN	QN	NS	NS	NS	DN	0.32
Xylene (total)	620	0.34	192.4	12.1	13.54	QN	DN	DN	NS	NS	NS	ND	ND
Polynuclear Aromatic Hydrocarbons:	Hydrocarbe	:suc											
Benzo(b)fluoranthene	see note 3	DN	DN	QN	an	2	QN	an	SN	SN	NS	ND	DN
Indeno(1,2,3-cd)pyrene		ND	ND	QN	ND	6	DN	QN	NS	NS	NS	ΟN	ND
Notes:													

WQCC standard, Section 3-103, as amended through December 24, 1987
 ND - not detected, NS - well not sampled
 PAHs: total napthalene plus monomethylnapthalenes = 30 ug/l

Note: This table revised 4/26/90.



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WATER RESOURCES

The Risks of Drinking Water

TUNGES 159

E. A. C. CROUCH,¹ R. WILSON,¹ AND L. ZEISE²

Energy and Environmental Policy Center, Harvard University

Exposures to the low levels of organic compounds found in drinking water might cause a variety of health effects, cancer being one of great concern. The estimation of cancer risk is uncertain, which may in part explain why cancer risk is not properly considered in establishing drinking water standards. Current standards for a civen chemical or class of chemicals do not account for the presence of other pollutants, chemicals which have not been tested in long term oral bioassays are generally ignored, and the uncertainties of risk calculations are not explicitly considered. To better quantily risks from drinking waters, we apply procedures which incorporate measures of uncertainty into the risk calculation and consider all chemicals included in a water analysis. This procedure is demonstrated by calculating risks for several U.S. water supplies. Our results show that some might pose moderately high risks.

INTRODUCTION

Current drinking water standards are not directly related to > health risks posed. Usually it is cancer risk which is of greatest concern in the setting of guidelines for low levels of organic compounds, and almost always estimates of cancer risk are made in an atmosphere of uncertainty. Over a period of years we have developed procedures which account for the uncertainties involved in risk calculations so that such risks can be more accurately determined. We illustrate these methods by calculating risks for some U.S. water supplies.

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We begin 1, discussing how to estimate the cancer risk from chemicats in cases in which there is no unequivocal evidence of adverse response on people and outline the procedure that we use for this purpose. Then we compare this procedure to that used by the Carcinogen Assessment Group (CAG) of the Environmental Protection Agency (EPA), emphasizing some of the assumptions in both procedures. Our method, which treats uncertainties more consistently, results in a probability distribution rather than a single value for risk. Upper and lower bounds as well as the mean risk can be derived from the distribution. Thirdly, we uncover inconsistencies w thin the proposed regulations for volatile organic chemicals [Federal Reaster, 1982] and the final regulations for halomethe les [Federai Register, 1979]; we note that if the acceptable risk level is rigidly set at 10. " life for a conservatively calculated risk, one of the possibilities suggested in the proposed regulations, the EPA would be forced to ban as unfit for human consumption most of the drinking water supplies meastruct by EPA in the National Organics Monitoring Survey of 1977 [EP.4, 1977]. We then suggest that drinking water risks be considered in light of societal behavior toward better known risks. Public attitudes indicate that risks of one ir, a million per year trisks averaged over a lifetime) and under might be small enough to be ignored. Finally, we note that many of the U.S. water supplies may impose much higher risks, usually because low levels of halomethanes are present. This suggests caution until risks of these chemicals can be more accurately evaluated through animal bioassay and other experiments

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Paper number 3W (490) (6)12-1397 83 (603W-1490505 (6)

CANCER RISK

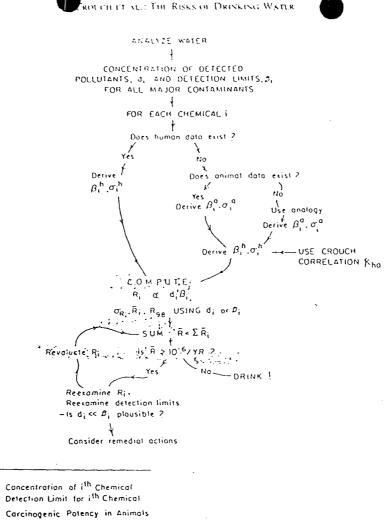
The risks addressed will be risks of cancer and not of other toxic effects. The reason for this restriction is that most chemical toxicity is characterized by a sharp threshold in dose or dose rate, so that for the low doses discussed here, toxic effects are very small. For carcinogenic effects for some chemicals and/or types of cancer there may be no thresholds, so that for public policy purposes it is prudent to assume that there are none. It is possible that teratogenic effects are also important, but information is limited. To the extent that teratogens are carcinogens, this is not a huge omission.

The word risk in itself implies uncertainty. The uncertainties dealt with here are of two types, first, those of the carcinogenic process which is, at least in part, a stochastic process, and second, uncertainties in the estimating the parameters of the mathematical models assumed.

It should be recognized that the cancer risks calculated differ from some of the other risks of everyday life. Since cancer risk from a single chemical is usually small and the latency period for cancer is large, and since a variety of circumstances and chemicals prevalent in the environment could have caused cancer of a particular type, causality is usually difficult if not impossible to establish. Often, even when it is possible to link an exposure with human calleer, quantification of the relationship remains clusive. For most of the unnking water contaminants there are no epidemiological studies which demonstrate a direct association with human calleer. However, it would be unethical to ignore the possibility. Many of the contaminants are known animal carcinogens, and since the exposure is so widespread, thousands of human cancers may have already been caused and many more may be caused in the future. The public health should be protected. but the cost of reducing the chemicals to the lowest levels possible in all water supplies would be enormous. We must find a way to determine when a water supply is reasonably safe. This first involves estimating the risk as well as is possible, which is the main purpose of this paper. Such riss can then be compared with other risks that society endures, some of which are readily accepted, others avoided, and some reduced through public expenditures.

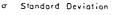
The procedure is to use all relevant and available data to estimate the lifetime cancer risk for drinking water with various levels of contamination. Measures of risk for the individual constituents are developed, then combined to establish the overall risk of the supply. For a very few chemicals there are human data which can be used to evaluate cancer risk

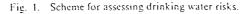
¹Also with the Department of Physics, Harvard University, ²Also with the Division of Applied Sciences, Harvard University,



- β.
- β
- βi^h Corcinogenic Potency in Humans
- Kha Interspecies Conversion Factor (Animal to Human)
- Human Concer Risk R

1 100





Usually, the most relevant data available are from studies on animals. Without exception the 30 or so materials or processes known to be carcinogenic in man have also been found to be carcinogenic in animals, so that this can be taken as a general rule. The converse of this general rule has been used for public policy purposes to proscribe any chemical found to be carcinogenic in animals (a good example is the Delaney Clause). However, there are so many animal carcinogens present in such small but detectable amounts that such a "zero risk" approach is unworkable.

-Since the zero risk option is not feasible, the next step is to recognize that chemicals differ widely in their ability to cause cancer. If it is noted that chemicals are usually acutely toxic in animals and man at similar relative values, the question arises as to whether the qualitative agreement in carcinogenicity be-/tween species can be made into a quantitative statement; that a chemical which is potent in one species is likely to be potent in man and that a chemical which has a low potency in an animal has a low potency in man. Most attempts to estimate carcinogenic risks rest on the assumption that this is to some _gxtent true.

Animal carcinogenesis studies have not been run on all the chemicals of concern. To estimate risk when this is the case, we propose the use of mutagenicity and toxicity information as well as carcinogenesis data from studies in animals of chemicals with like structure. We call this method of risk assessment "risk by analogy." The general scheme for assessing risks of drinking water is illustrated in Figure 1.

METHODOLOGY

We are concerned with estimating in humans the carcinogenic effects of a chemical at low concentrations. To do this we usually use data on animals exposed to much higher levels of the chemical, so a two-fold extrapolation is required: from high to low dose and from animal to human. Logically, the two orders of extrapolation might give different results, but in practice the assumptions made assure their equivalence.

The usual approach for extrapolation between doses has been to use a mathematical dose response model:

$$R = F(d, x, y, z, \dots; a, b, c, \dots)$$
(1)

Here R is a measure of response which is presumably related to the risk of interest and which can be observed in the animal experiments. The functional form F is usually chosen by guesswork, or on the basis of some hypothesis, or to fit various constraints. The response is expected to be a function of dose d measured in some way and also of various other variables (x, x, z, --), for example, age. Included in the model

are a set of parameters (a, b, c, \cdots) which measure how strongly the response R depends upon the variables (d, x, y, z, \cdots) . The dose response model may be fitted to experimental animal results to give estimates of the parameters (a, b, c, \cdots) , one of which (or some combination of which) may be considered a potency of the material under test. With such estimates of the parameters the dose response model allows extrapolation to low doses.

The next problem is extrapolation to another species, in this case humans. Once again a dose response model

$$Z = G(d, x, y, z, \cdots; x, \beta, \gamma, \cdots)$$
(2)

is specified. Here G, Z, $(\alpha, \beta, \gamma, \cdots)$ have the same general interpretations as F, R, (α, b, c, \cdots) . However, the response Z may be a response different in nature from R, and the functional forms F and G are not necessarily identical. For example, R may represent the incidence of liver cancer in mice with F chosen to be an S-shaped dose response function, and Z may represent bladder cancer in humans caused by the same carcinogen, with a linear form for G.

The interspecies extrapolation is specified by relations between the parameters (a, b, c, ...) and $(z, \beta, \gamma, ...)$:

$$g_1(a, b, c, \dots; x, \beta; \gamma, \dots) = 0$$

$$g_2(a, b, c, \dots; x, \beta; \gamma, \dots) = 0$$

$$(3)$$

Given these relationships, information about the response in one species, contained in the parameters \mathbf{x} , β , γ , \cdots can be used to obtain information in the second species through the parameters a, b, c, \cdots .

In practice the same functional form in the assumed dose response functions is being used for animals and humans (F = G). This is the origin of the logical equivalence between the two routes of extrapolation mentioned above. Some go so far as to suggest that the numerical values of the parameters are identical, so that the set of relations $(g_1 = 0, g_2 = 0, \cdots)$ becomes

$$a = x$$
 $b = \beta$ $c = \gamma$ etc. (4)

The CAG adopts this approach.

The above summary has been written with considerable generality to emphasize the following separate elements of the extrapolation: (1) a dose response function F in one species; (2) a logically separate dose response function G in another species, for a logically separate response, and (3) relations $(g_1 = 0, g_2 = 0, \dots)$ between the parameters of such dose response functions. It is easy to become confused in the practical cases where the same symbols are used to represent parameters or variables referring to different species, and the same mathematical dose response function is used to model dose response in both species of interest.

Risk Calculation

For both animals and humans, Crouch and Wilson [1979, 1981] use as a dose-response model a formula which can, inter alia, be derived from a one-hit model:

$$R = 1 - (1 - x) \exp - [\beta D_i (1 - x)]$$
(5)

K= 1-1-0. 11- K

Here *R* is the probability of dying with a tumor induced by a dose *D*: 5 and β are the parameters of this model, and both are taken to be constant. By expanding the exponential we see that the right-hand side of (5) reduces to $z + \beta D$ at low doses.

to the spontaneous tumor rate x at D = 0, and tends to unity at high doses. The parameter β is called the carcinogenic "potency," since it is the ratio of the excess lifetime cancer incidence in a population caused by the carcinogen to the average daily dose received by the population (provided all individual doses are small enough; this is assumed from here onl. Dose is the lifetime average daily intake per unit bodyweight, measured in milligrams per kilogram day; hence potency is expressed in kilogram days per milligram. For an individual the product of potency and dose will be the lifetime probability of cancer, or "risk," imposed by the carcinogen. It is found that a chemical's potency, as defined above, is roughly independent of species [Crouch and Wilson, 1979].

We can use the carcinogenic potency in a risk calculation in the following way. The risk is the product of potency and dose $(R = \beta D)$. For example, if the amount of pollutant in the water if $(R = \beta D)$ is the daily water intake is 2 liters, then the intake is 1 days for a 70-kg man the dose, in the required units, is

$$\int AN = \frac{2 \mu g (kg \, day)}{1984} = 3 \times 10^{-5} \text{ mg} (kg \, day) < \frac{4}{3}$$

TEXAS Area tild cilled VERSITY, the probability of developing a cancer in a lifetime, is $\beta \times 3 \times 10^{-5}$ for an intake of 2 1/day for life of water contaminated with 1 µg 1.

For convenience in making comparisons, we define the "average annual cancer risk" R to be the lifetime probability of cancer divided by the life expectancy, which is approximately 70 years. (We do not mean to imply that the risk is equally spread over age: it probably is not, even if ingestion of the carcinogen occurs at a constant rate or in a single dose. Hence average annual cancer risk is only a convenient way of expressing lifetime risk; it is a nominal risk and not a true probability.) So average annual cancer risk R for a lifetime intake of 2 l/day of water contaminated with $d \mu g/l$ of pollutant is

$$R = [3 \times 10^{-5} (\beta d)].70$$

= 4 × 10⁻⁷ (\beta d) (6)

For a mixture of chemicals we calculate the quantity

$$R = 4 \times 10^{-7} \left[\sum_{i} \left(\beta_i \ d_i \right) \right]$$
(7)

where d_i is the concentration of chemical *i* in units of microgram per liter and β_i is the potency of that chemical in humans. If there are uncertainties in any of the values in (7), as is always the case, the procedure is a little more complicated. Before detailing that procedure, four elements of uncertainty plaguing most risk calculations are discussed.

Uncertainties

Uncertainties in measuring potency in animal experiments, σ_x . In fitting the dose response curve (equation (5)) to experimental results, the potency β cannot be found exactly but is subject to uncertainty. This uncertainty can be approximately modelled by a lognormal distribution, so that if

$$\mathbf{x} = \log_{\mathbf{r}}\left(\boldsymbol{\beta}\right) \tag{S}$$

then the probability distribution for x is given by

1+ 21 - 1- 0. - 1- ED

$$p(x) dx = \frac{1}{\sigma_1 (2\pi)^{1/2}} \exp \left[-\frac{(x-\bar{x})^2}{2\sigma_1^2} dx \right]$$
(9)



where π_{i} is usually about 0.35 for the ingestion bioassays tep-sted on in the National Cancer Institute's Carcinogenesis Technical Report Series [Zeise et al., 1982] and \hat{x} is the median value of log_e (β), corresponding to the point estimates given by Crouch and Wilson [1979]. When the only studies available are insensitive and the carcinogenic potency β differs insignificantly from zero, the distribution of β is not lognormal. For these cases we obtain upper confidence bounds on potency using techniques described by Crouch [1983].

Uncertainties in extrapolating from animals to humans, σ_{a} . We estimate that the potency in people is the same as the potency in animals, when expressed in units of kilogram days per milligram, although this statement of equality must be tempered by the uncertainty associated with this interspecies extrapolation. The most systematic study of this uncertainty was begun by Crouch and Wilson [1979] and extended by Crouch [1983], wherein they compared the carcinogenic potency of several chemicals in rats and mice, and where data was available, in people also. They found about the same proportion of cancers would be induced in different species of animals fed the same daily dose, expressed as lifetime average daily intake as fraction of bodyweight. The animals studied had been treated for an appreciable proportion of their lifespans. Crouch and Wilson express this formally by assigning an interspecies sensitivity factor K to convert from potency in one species to potency in another:

$$\beta_{h} = K_{ha}\beta_{a} \tag{10}$$

where β_a and β_k are the potencies in species a and h, respectively. They find that K fits a lognormal distribution. If

$$y = \log_{e}(K) \tag{11}$$

then the probability distribution p(x) is

$$p(y) \, dy = \frac{1}{\sigma_y (2\pi)^{1/2}} \exp - \frac{(y - \dot{y})^2}{2\sigma_y^2} \, dy \tag{12}$$

For comparisons of rats and mice, they found that $\sigma_x \approx 1.5$, corresponding to the factor 4.5, since exp (1.5) ≈ 4.5 and $\dot{y} \approx 0$ [*Crouch*, 1983]. We assume that potency in humans is, on average, the same as in mice or rats and that the error in extrapolating from rodents to humans is the same as extrapolating from mice to rats or vice versa. The available data on human cancers indicates that this is not an unreasonable assumption [*Crouch and Wilson*, 1979].

Uncertainties in exposure, σ_{2n} . Estimates of the dose d are derived from drinking water analyses and information on drinking water habits. We assume that the doses to the population are lognormally distributed, as is often the case for environmental contaminants, so that for

$$z = \log_{z} (d)$$
 $p(z) dz = \frac{1}{\sigma_{z} (2\pi)^{1/2}} \exp -\frac{(z-z)^{2}}{2\sigma_{z}^{2}} dz$ (13)

where ξ is the median.

Uncertainties in high to low dose extrapolation. E. The dose response curve (equation (5)), here used in making estimates of potency, incorporates an assumption of linear extrapolation from high to low doses. All the estimates of potency have been made at high dose, and other dose response models which may differ drastically in their low dose behavior would give similar estimates of high-dose potency and similar results for estimate errors. The work on interspecies extrapolation quoted above has little bearing on the correct extrapolation to low dose.

The data on dose response is equivocal. We do not co into detail here but note that there is evidence for certain chemicals and types of cancer that the dose response relationship is linear; for others a threshold response is more plausible. For most chemicals the data to distinguish the forms are inadequate Therefore the matter of high to low dose extrapolation is one of public policy, at least for the purposes of risk assessment, rather than of science. It has been conventional to assume that the dose response relationship is linear. The FDA in its ruling on saccharin implicitly assumed linearity, while the EPA now consistently allows for sublinear response at low doses. Because the cost of being wrong is so great, we contend that unless the data strongly indicates otherwise, linear extrapolation should be used. However, we formally account for those who disagree by introducing an extrapolation factor E. which can be assigned a probability distribution based on prior belief.

The distribution for E is subjectively derived. Those who believe in a threshold assume that E is near zero, a few believe that at low doses the response is greater than linear (E > 1), and those who, like ourselves, believe in linearity set E to one. Indeed, if decisions are based on average risk and if the chance that there is no threshold exceeds, say, 1 in 3, the risk will be greater than $\frac{1}{3}$ of that calculated under the linear assumption and there will be little difference in any policy implications.

The Probability Distribution for Risk

Provided the doses are small enough, the excess annual risk due to a carcinogen js thus

$$R = 4 \times 10^{-7} \, (\beta \ K \ d \ E) \tag{14}$$

where

 β potency;

K interspecies factor;

d contaminant concentration;

E dose extrapolation factor.

The uncertainties in each of the factors in (14) must be taken into account in the Crouch-Wilson procedure. As noted above, this may be done by assuming a probability distribution for each factor in R. For simplicity we take the extrapolation factor E = 1, corresponding to no threshold. For reasons discussed above the effect on any decisions using our risk estimation procedure is likely to be small, although it unavoidably introduces a small coupling between the risk assessment and risk decisions, which is generally considered undesirable. Since E is assumed to be equal to 1 with certainty, and since the factors β , K, and d are independent and lognormally distributed. R also obeys a lognormal distribution, with a standard deviation σ given by

$$\sigma^2 = \sigma_x^2 + \sigma_y^2 + \sigma_z^2 \tag{15}$$

so that for

$$r = \log_{e}(R)$$
 $p(r) dr = \frac{1}{\sigma(2\pi)^{1/2}} \exp -\frac{(r-\dot{r})^{2}}{2\sigma^{2}} dr$ (16)

where \vec{r} is the median value, the probability distribution for the annual risk R is

$$p(\mathbf{R}) | d\mathbf{R} | = \frac{1}{\sqrt{\mathbf{R}(2\pi)^{1/2}}} \exp\left(-\frac{(\log_{\mathbf{R}} |\mathbf{R}| + i)^2}{\sum_{k=1}^{1/2}} | d\mathbf{R} | = -(17)$$

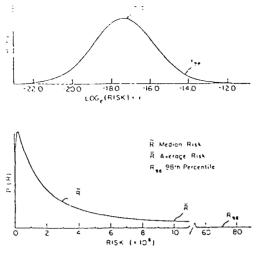


Fig. 2. Lognormal risk distribution (average annual cancer risk from 100 µg.1 trichloroethylene in drinking water).

The distributions for r and R are illustrated in Figure 2. Note that the average, or mean value, of R is not the same as the median but larger by a factor of exp ($\sigma^2/2$). This is because a lognormal distribution is skewed to the right.

Since risk is represented by a distribution, rather than one value, a summary statistic must be derived. It is a matter of judgment what measure of the probability distribution is selected, and the choice may depend on the type of action considered. One of us argues that an individual cannot distinguish between the different types of uncertainty: first, the stochastic variability arising since R itself expresses only a probability of harm or second, the uncertainties in β , K, or d. For him the parameter of interest is an average value of R, \overline{R} . This is given by

$$\vec{R} = (4 \times 10^{-7}) \, \vec{\beta} \, \vec{K} \, \vec{d} \, \vec{E} = (4 \times 10^{-7}) \, \vec{\beta} \, \vec{K} \, \vec{d} \, \vec{E} \exp(\sigma^2/2)$$
(18)

where $\vec{\beta} = \exp(i\vec{x})$, $\vec{k} = \exp(i\vec{y})$, $\vec{d} = \exp(i\vec{z})$. A risk-averse person may instead decide on the basis of an upper quantile of the distribution, say the 98th percentile.

For public policy decisions, where one must be conservative, it may also be appropriate to use an upper limit on the risk; here the stochastic nature of the risk is averaged out by the large population but the uncertainties in β . K, and dose remain. In a population of size N, the upper 98th percentile on the annual number of cancers is $NR_{48} = (4 \times 10^{-7}) N \beta \tilde{K} \tilde{d}$ $E \exp (2\sigma)$ and the mean estimate is $N\bar{R} = (4 \times 10^{-7}) N \beta \tilde{K} \tilde{d}$ $E \exp (\sigma^2/2)$. Using an upper limit corresponds in intent to the CAG procedure of taking the 95th percentile of the various terms before multiplying them but accounts more correctly and completely for the uncertainty.

This procedure is slightly more complicated when the effect of a mixture of chemicals is to be determined. In that case, as indicated in (7), the overall risk is obtained by summing several variables, each with a lognormal distribution. The mean value is simply the sum of the means, but in order to obtain the percentage points \tilde{R} and R_{og} , the distribution of the sum must be determined.

An Example: Risk Calculated for Trichloroethylene

Using the National Cancer Institute (NCI) ingestion bioassay results. Crouch and Wilson evaluated the carcinogenic potency of trichloroethylene [*Crouch and Wilson*, 1979]. The value of potency used is the highest significant value calculated for any site among the four experiments (rats and mice of both sexes). Since a chemical may cause cancer in more than one organ, this method may underestimate potency, as it does when applied to eigarette smoking in humans, where the potency is understated by a factor of 2, since it is that of lung cancer and the effects of smoking on other organs are omitted. Underestimation is somewhat compensated for by taking the highest value in the four experiments.

The model fits the trichloroethylene data well, and the median estimate for β is 7.3×10^{-4} kg day/mg and for σ_x is 0.22. As discussed above, for the interspecies comparison factor K we assume that the uncertainty is measured by the spread of the points in the comparison of *Crouch* [1983] and obtain $\sigma_y = 1.5$, while for the dose measure we assume for illustrative purposes that $\sigma_z = 1/2$.

The overall uncertainty in the risk is then $\sigma^2 = \sigma_x^2 + \sigma_z^2 + \sigma_z^2 = 2.6$, and so the average risk \bar{R} is then greater than \bar{R} by a factor of 3.6 (exp $\sigma^2/2 = \exp(1.3) = 3.6$) and the 98th percentile of the distribution differs from \bar{R} by 25 (exp $(2\sigma) = \exp(3.2) = 25$). The median estimate of annual risk of drinking 2 liters of water a day containing trichlorethylene at a level of 100 μ g/l is then (see (6))

$$\tilde{R} = 4 \times 10^{-7} \times 100 \times 7.3 \times 10^{-4}$$
$$= 2.9 \times 10^{-8}$$
(19)

so that the mean estimate of risk is

$$\bar{R} = \exp(\sigma^2/2) \times \bar{R} = 1 \times 10^{-7}$$
 (20)

and the upper 98th percentile is

$$R_{98} = \exp(2\sigma) \times \tilde{R} = 7 \times 10^{-7}$$
⁽²¹⁾

These values as well as the risk distribution for irichloroethylene are illustrated in Figure 2.

Comparison with CAG Methodology

The Carcinogen Assessment Group follows a procedure similar to that discussed [EPA, 1981], but although they often derive upper bound estimates of risk similar to our own, there are important differences in the details. The CACi uses a procedure developed by K. Crump, in which the dose response relationship is a "modified multistage model" in that lifetime probability of tumor, R, is related to dose D by:

$$R = 1 - \exp\left[-(a + bD + cD^2 + \cdots)\right]$$
(22)

Dose is measured in a different way from that discussed above. The parameter b is very similar to the potency β described above, and the CAG use an estimate, from the animal experiments, of the 95% upper confidence limit on this parameter, h_{95} , as an estimator to be used in deriving an upper limit on risk. In practice, b_{95} and β_{95} usually differ little when account is taken of the differences in definition of dose.

Whereas the CAG chooses the 95% confidence bound on potency for use in the risk calculation, with the intention of obtaining a conservative estimate of risk, we instead strive to estimate risk and its distribution as precisely as possible in order to reliably estimate the upper bounds of risk. Indeed, the CAG is much less conservative in these important respects.

1. Insignificant findings in animal experiments are used to imply zero risk.

2. The modified multistage model can result in estimates of b_{gs} , which are substantially smaller than of β_{gs} .

TABLE 14 - Effetime Average Annual Risk of 100 µg 1 Trichloroethylene

	Value
R, average risk	1×10^{-7}
Ros 98th percentile	7×10^{-7}
CAG risk estimate	3.6×10^{-7}

Assumes 24 day of water consumption.

3. The CAG does not attempt to estimate the risk for chemicals on which neither human epidemiological studies nor animal bioassays exist and, in general, ignore the potential risk from these chemicals.

4. Estimates of the parameters of the multistage model corresponding to experiments with several dose groups are unstable; therefore extracting the single parameter b for use in estimating risk can produce widely varying estimates of risk for similar experimental results.

The CAG have not yet discussed in detail the relationship between carcinogenicity in animals and carcinogenicity in man, which is accounted for by our factor K. Implicit in their methodology are the assumptions that the dose response relationships are functionally identical in man and experimental animal and that the values of the parameters are numerically identical in the two relationships. They thus effectively assume K to be a fixed exact value. Thus animals and men are considered to respond identically when dosed at an equal amount per unit surface area (not per unit weight), which is equivalent to taking $K \approx 5$ for rat to man comparisons and $K \approx 13$ for mouse to man comparisons. Although the data do not suggest these values, CAG justify them as being conservative.

Despite the many dissimilarities in computation, our method and that used by the CAG often produce similar results. In Table 1a, CAG estimates of risk are compared with the risk calculated above for a lifetime intake of 2 l/day of water contaminated with 100 μ g/l of trichloroethylene (TCE). (The number quoted in Table 1a for the CAG estimate of risk differs slightly from that derived from Table 3 of the Federal Register notice, which is reproduced as Table 7. From the first line, the lifetime risk of TCE is 10^{-4} at 280 µg/l, or $10^{-4}/2.8$ at 100 μ g/l. The annual risk is $10^{-4}/(2.8 \times 70)$ at 100 μ g/l or 5×10^{-7} . We do not know the reason for the discrepancy, except that the Federal Register may use a different version of the CAG report.) For some compounds tested in the NCI Carcinogenesis Technical Report Series (numbers 2-239, National Cancer Institute, Bethesda, Md.) the two methods produce different results, as evidenced in Table 1b.

RISKS FROM VARIOUS U.S. WATER SUPPLIES

The methods described above may be used to estimate risks from carcinogens in the various water supplies monitored by the EPA in the National Organic Monitoring Survey of 1976 -1977 [EPA, 1977]. Table 2 lists the 27 chemicals monitored, together with our estimators of potency and their uncertainty. For estimates of exposure we assume that an individual drinks 2 liters daily of the polluted water for life and that the water remains contaminated at the maximum levels found in the 1976-1977 survey (Table 3).

In most cases we derive values of potency from bioassay data and in one case from a human epidemiological study. However, such data is not available for all chemicals or is so poor as to be of no use. We estimate these potencies by "analogy," deriving a value based on the carcinogenicity of related chemicals. For example, the potency of bromoform may be estimated by comparing the carcinogenicity of brominated alkanes to their chlorinated analogues [Wilson and Fiering, 1983]. In most cases the brominated compounds are more carcinogenic. The only direct evidence for the carcinogenicity of bromoform comes from a comparatively short and insensitive study using intraperitoneal injection to induce pulmonary tumors in A/St mice [Theiss et al., 1977]. Although many compounds were tested, only bromoform induced a significant number of tumors. This test included chloroform, which was shown to be carcinogenic in the more sensitive NCI tests. However, the bromoform results are considered inconclusive because the significant result was in the middose group and insignificant numbers of tumors were induced in mice with the same survival rate but receiving a higher dose. In another short term study of alkyl halides using intraperitoneal injection, this time into A/He mice, butyl bromides were slightly more potent than or as potent as their butyl chloride analogues [Poirer et al., 1975]. We also note that 1.2-dibromoethane has a carcinogenic potency which is 10 to 200 times as large as 1,2-dichloroethane, and that the only other brominated alkane tested in the NCI series, 1,2dibromo-3-chloropropane, was highly carcinogenic, with potency greater than that found for any of the other halogenated alkanes in these tests. (Other halopropanes were not tested.) In addition, bromoform and dibromochloromethane are mutagenic in salmonella typhimurium under conditions in which chloroform is not mutagenic [Simmon et al., 1978]. These facts suggest that brominated hydrocarbons may be substantially more potent than their chlorinated analogues.

For bromoform, which has three bromine atoms, it would be prudent to assume it is at least a hundred times more potent than chloroform. For dibromochloromethane (two bromine atoms) we assume that the potency is 20 times that of chloroform and for dichlorobromomethane (one bromine atom) 5 times. The uncertainties in these estimates have to include the uncertainty in extrapolation from animals to humans and the uncertainties in the animal experiments, as well as the uncertainty due to the use of analogy. As a crude estimate we add an extra 0.3 to σ for each bromine substituted for chlorine, leading to the values shown in Table 2 for bromodichloromethane, dibromochloromethane, and bromoform. The upper bound on the carcinogenic potency of iodoform derived from the NCI bioassay is less than the potency measured for chloroform, so we assume that dichloroiodomethane is less carcinogenic than chloroform. Notice that the order of carcinogenicity of the brominated, chlorinated, and iodinated alkanes may be the same as the order of electronegativity of the halogen ions.

Using analogy, potencies were derived for several of the other chemicals listed in Table 2. The analyses and assumptions made for these chemicals are discussed in the appendix.

Some people have suggested that 1,1,1-trichloroethane is

TABLE 1b. Lifetime Average Risk from Water Contaminated With 100 µg/l

	Clonitralid	Lead Dimethyldithiocarbamate
Ř. median risk	1 × 10 ⁻⁸	0
R ₈₆ , 98th percentile	2 × 10"'	2×10^{-3}
CAG risk estimate	0	0

Assumes 21/day of water consumption

CROUCH FT ALL THE RISKS OF DRINKING WATER

1	Chemical	Potency. kg day/mg	ο,	Total o Assumed	Reference
1	Chloroform	1.1×10^{-2}	0.14	1.51	(NCI)*
2	Bromodichloromethane	5.5×10^{-2}	• • •	1.8	text
-1-	Dibrumochloromethane			21	text
4	Brenieform	$1.1 \times 10^{\circ}$		2.4	text
5	Dichloroiodomethane	2.2×10^{-3}		1.8	lext
	1.2-Dichloroethane	9.4×10^{-3}	0.27	1.53	(NCI)*
<u>()</u> 7	Carbon tetrachloride	4.0×10^{-3}	0.22	1.52	(NCI)*
8	Methylene chloride	2.5×10^{-3}	2.1	2.5	text (app.)t
9		2.0×10^{-2}			(Maltoni);
10	Trichloroethylene	7.3 × 10 ⁻⁴	0.22	1.52	(NCI)
11	Tetrachloroethylene	1.6×10^{-3}	0.16	1.51	(NCI)
12	1.1.1-Trichloroethane	1.7×10^{-3}	0.75	1.68	(NCI)
13	Bis-(2-chloroethyl)-ethcr	4.2×10^{-2}	0.34	1.68	(Innis)§
14	Bis-(2-chloroisopropyl)-ether	$< 6.7 \times 10^{-4}$	NAI	(1.5)	(NCI)*
15	Benzene	1.0×10^{-3}		1.5	CW (human)**
16	p-Dichlorobenzene	1.4×10^{-3}	1.9	2.3	text (app.)†
17	m-Dichlorobenzene	$< 8.8 \times 10^{-2}$	2.0	2.5	text (app.)†
18	o-Dichlorobenzene	6.9×10^{-3}	17	2.3	text (app.)†
19	1.2.4-Trichlorobenzene	4.4×10^{-3}	1.5	2.1	text (app.)*
20	2.4-Dichlorophenol	1.7×10^{-3}	1.7	2.2	text (app.)†
21	Pentachlorophenol	4.0×10^{-2}	1.7	2.1	text (app.)†
22	Polychlorinated biphenyls	5.3×10^{-2} tt	0.57	1.6	(NCI)*
<u>22</u> 23	Fluoranthene	3.1×10^{-2}	2.5	2.9	lext (app.1t
24	Benzofluoranthene	9.3×10^{-2}	1.8	2.0	text (app.)*
25	1.12-Benzperylene	1.9×10^{-2}	2.5	2.9	text (app.)t
26	3.4-Benzopyrene	9.3×10^{-1}	0.87	1.7	text (2pp.)†
27	Indeno (1.2.3-cd) pyrene	9.3×10^{-2}	2.5	2.9	text (app.)*

17801-1	University	Potency of	Drinking	Water	Contaminants.	Monitored	by the EPA in the	
		1976-1977	National (Organic	Monitoring Su	irvey		

*(NCI): Analysis of NCI Carcinogenesis Technical Report Series data (Numbers 2-239, 1976-1982, National Cancer Institute, Bethesda, Md.).

ftext (app.): Analysis appears in the appendix of this paper.

(Maltoni): Analysis of Maltoni's ingestion data [Maltoni et al., 1979].

(Innis): Analysis of data from Innis et al. [1969].

NA: Not applicable.

** CW (human): Analysis of human data [Crouch and Wilson, 1979].

##The potency is of the PCB Arochlor 1254.

not carcinogenic. This statement is inaccurate; a more correct statement would be that it has not been found to be carcinogenic, for the simple reason that the tests may have not been sensitive enough. The NCI test was considered inconclusive, in part because toxic effects killed the treated animals early, before carcinogenic effects might be manifest. However, we still can estimate a potency figure, with uncertainty, from the 1.1.1 trichloroethane data.

Table 4 shows the risk calculated for the selected cities. To arrive at these values, contributions for each of the chemicals are calculated separately (Tables 5a-5c) and then combined. In Tables 5a-5c we have only tabulated risk values for chemicals which have been detected and significantly contribute to the risk and not for all of the chemicals listed in Table 2. The value of \hat{R} , the mean estimate, can be obtained simply by arithmetically summing the individual Table 5a entries which are the mean annual risks for each of the chemicals:

$$\tilde{R} = \sum_{i} \tilde{R}_{i} \left[\exp\left(\sigma_{i}^{2} 2t\right) \right]$$
$$= 4 \times 10^{-7} \left[\sum_{i} \tilde{\beta}_{i} \tilde{d}_{i} \exp\left(\sigma_{i}^{2} 2t\right) \right]$$
(23)

where \sum_{i} represents the summation over all chemicals *i*. However, to obtain the median and the 98th percentile estimates of risk, it is necessary to determine the probability distribution of the sum expressed in (7). This has been done, using a Monte Carlo technique, to arrive at the entries \tilde{R} and R_{qg} in Table 4. In most cases the median of the sum is approximately twice the sum of the medians, while the 98th percentile of the sum is approximately 10-20% less than the sum of the 98th percentiles.

Table 4 shows that all cities have average risks greater than 10^{-6} per year and that several have values of 10^{-5} and above. The value d_i used is the highest concentration measured for the chemical in the appropriate city, and so these estimates of risk may be a bit pessimistic. However, most of the large estimates are dominated by large concentrations of bromoform, where the carcinogenic potency is very uncertain since it has not been directly measured. Notice especially the cities of Brownsville (Texas), Fort Worth (Texas), Melbourne (Florida), Topeka (Kansas), and Dayton (Ohio), where bromoform concentrations are particularly high so that, since for bromoform the value of β is moderately high and of σ is very high, the 98th percentile risk is very high indeed (Table 5c). This indicates a need for caution while awaiting the results of the ongoing NCI bioassay on bromoform. In Tables 4 and Sa-Sc, entries greater than 10⁻³ are not precise and slightly overestimate the risk, because the expression in (5) is not linear at these levels. Appropriately alarming, though not precisely calculated, these large numbers indicate unacceptably high risks by most standards.

Although we have tried to be systematic and logical in estimating risks for drinking water, we have not been entirely complete. We have not discussed how two pollutants added to the water can produce a multiplicative synergistic effect such as follows, for example, exposure to both asbestos and tobac-

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Polychlorinated biphenyls		
ιουομόστομοπισο	$\begin{array}{c} 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0$	
ionsdavoldsib-4.5	0.17 0.01 0.01 0.01 0.01 0.01 0.01 0.01	
sasznedoroldsin-4. <u>C.</u> 1		1
sussusdovoldsib-o	(0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.005) (0.0	· • · · · · · · · · · ·
sussusdoroldsib-m	$ \begin{array}{c} (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) \\ (0.005) $	1
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Parentheses indicate that the chemical was not detected. The number inside the parentheses is the detection limit. All values are in micrograms per liter. None detected: detection limit 0.0. None detected: detection limit 0.0. So not detected present: detection limit 0.05. So not detected but not quantified. Sample taken for survey was lost.

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CROUCH ET ALL: THE RISKS OF DRINKING WATER

TABLE 4	Mean Median and 98th Percentile Estimates of Annual
Risk from	Organic Chemicals in Drinking Water from Selected
•	U.S. Cities

	0.5. Cines		
City	R. Average	R. Median	R _{ox} , 98th Percentile
Annandale, Va.	7.0E - 06	3.2E-06	4.21E - 05
Atlanta, Ga.	7.5E-06	2.6E - 06	4.5E-05
Baltimore, Md,	7.4E - 06	2.4E - 06	5.3E - 05
Billings, Mont.	2.6E - 06	7.5E-07	1.9E – 05
Birmingham, Ala.	5.5E - 06	2.1E-06	3.2E-05
Brownsville_Tex.	2.5E - 03*	2.8E-04	1.9E - 02*
Cape Girardeau, Mo.	2.9E - 05	1.1E - 05	1.5E - 04
Columbus, Ohio.	1.9E - 05	6.7E – 06	1.2E-04
Dayton, Ohio	1.2E - 04	1.6E - 05	7.5E - 04
Elizabeth, N. J.	1.2E - 05	4.5E - 06	7.8E – 05
Fort Worth, Tex.	2.7E - 04	3.2E - 05	2.0E 031
Huntington, W. Va	7.9E – 05	2.0E - 05	5.5E-04
Jackson, Miss,	1.1E – 05	4.5E-06	6.2E – 05
Jersev City, N. J.	5.8E - 06	2.1E - 06	3.1E-05
Kunsas City, Kan.	3.5E - 06	1.2E-06	2.0E-05
Lincoln, Neb.	8.1E-05	9.8E-06	6.6E - 04
Los Angeles, Calif.	7.6E-06	2.2E – 06	5.3E – 05
Manchester, N. H.	3.8E - 06	1.6E – 06	2.5E-05
Melbourne, Fla.	2.4E-04	5.9E - 05	1.6E - 03
Montgomery, Ala.	6.5E-06	2.9E - 06	3.9E – 05
Newport, R. L.	4.7E-05	1.4E-05	3.0E - 04
Norfolk, Va.	1.4E - 05	5.5E - 06	8.6E – 05
Passaic Valley, N. J.	1.1E-05	3.9E-06	6.9E-05
Sacramento, Calif.	2.4E-06	9.6E - 07	1.7E – 05
Topeka, Kan.	1.5E-04	2.9E-05	1.1E-03

1.0E = 06 means  $1.0 \times 10^{-6}$  in ordinary scientific notation. *Entries greater than  $1.0E = 03 (10^{-3})$  are not precise; see text.

co smoke. However, in most cases we are concerned with small doses of one pollutant added to a "sea" of many pollutants. For these small doses, a multiplicative effect is not expected. In addition, we have included only the 27 chemicals analyzed by the EPA in its organics survey and have not accounted for inorganic chemicals.

This analysis has been extended to include risks for those chemicals not detected at any site. We assume that the chemicals which were not detected were present at the detection limits and rederived the risk distribution using the same methods described above. In doing this it was found that at the very most, average annual risks were doubled. The cases of doubling were those where we would least expect undetected contaminants to be at concentrations near the detection limits: the least polluted water supplies with average risks of  $2 \times 10^{-6}$  and less. For the more polluted water supplies the values of risks were unchanged.

### SELECTION OF AN ACCEPTABLE LEVEL OF RISK

Once the risk distributions are derived, the obvious next step is to question whether or not the waters are safe to drink. The EPA has solicited public opinion as regards an acceptable level of risk [Federal Register, 1982]. All told, there has been much written on this subject, wherein most agree on one point: there are no simple answers. There is also a general consensus that judgements on public safety have no place in a risk assessment, especially since risk perception and public acceptance cannot be exactly predicted from actuarial statistics, benefits, costs of risk reduction, and the degree to which the risk is voluntary. It is without the intention of suggesting a risk level to be used as a strict safety standard that we make the following remarks. First, an acceptable level cannot be determined in vacuo: an assessed risk is only meaningful in

relation to like risks, similarly determined, and the risks and costs of the other options under consideration. Second, some general statements can be made about public behavior toward known cancer risks (e.g., X rays, smoking) and other non-cataclysmic risks of death (e.g., car accidents and bee stings; not reactor meltdowns and nuclear war). Some risks are so small that people tend to ignore them and regulations are not developed to reduce them. There are other, larger risks that the public attends to through regulations, public expenditures, and individual action. Pochin [1976] and the United Kingdom Royal Commission on Environmental Pollution [1976] point out that many risks of  $10^{-6}$ /year are familiar and casually accepted. This is not surprising when we consider that a person exposed to a risk of 1 in  $10^6$  per year has life expectancy reduced by only one day [Crouch and Wilson, 1982].

Risks of cancer from drinking water can be compared to a number of risks indicated in Table 6 taken from <u>Crouch and</u> <u>Wilson</u> [1982]. Data on hazards other than cancer were derived from annual death rates on exposed populations. The cancer risks shown are calculated in the same way as drinking water risks in this paper, and so the same caveats and uncertainties apply. Table 6 gives further evidence that risks of 1 in  $10^6$  and less are generally ignored, while risks above  $10^{-6}$ /year are of concern. Dangerous occupations, in which workers are somewhat compensated for risks assumed, carry risks of  $10^{-4}$  and greater (Table 6); while many popular sports, several considered risky, carry risks of  $10^{-5}$  and more [Crouch and Wilson, 1982]. Cancer risks from air pollution are approximately  $10^{-5}$ /year [Crouch and Wilson, 1982].

With these remarks in mind we note that, except for the very polluted water supplies indicated in Table 4, drinking water risks are comparable to other risks society puts up with but tries to control. The mean and median estimates are moderate but less than  $10^{-4}$ /year, apart from the five extreme cases. The 98th percentiles of the distributions are high, however, especially where there is bromoform present, and the uncertainty is large. Given that this is the case, none of the risks for drinking water supplies we calculated should be ignored. More careful experiments and the attendard risk analyses, as well as further regulatory action, should be considered.

### EPA'S DRINKING WATER STANDARDS

Although cancer risk is an important consideration to the EPA in the setting of standards, it is not entirely consistent in its use of risk estimates. The federal standard for the total concentration of trihalomethanes is 100 µg/i [Federal Register, 1979]. The EPA defines total trihalomethanes as the sum of concentrations of the four halomethanes chloroform, bromoform, dibromomethane, and bromodichloromethane. The mean value of the risk for 100  $\mu$ g 1 of chloroform is 1.4  $\times 10^{-6}$ /year or  $1 \times 10^{-4}$ /lifetime. The 98th percentile is 6  $\times 10^{-4}$ /lifetime. We calculate the 98th percentile risk from 100 µg/l of bromoform as 600 times greater, partly because of the extreme uncertainty, and so a supply just meeting the standard and contaminated solely with chloroform poses much less risk than if it were just meeting the standard but contaminated mostly with bromoform. Bromoform is likely to be more carcinogenic than chloroform, which is 10 to 100 times as carcinogenic as trichloroethylene in animals. The EPA should put the allowed maximum contaminant levels (MCE's) in this order. Yet the proposed standards are not

				IVBLE 20		i ne Averag.	C KISK OI L	It, The Average Kisk of Drinking water of Selected V.S.	valac lo Jalis		CILICS					
	Chloro- form	Bronsodi- chloro- rnethane	Dibromo- chloro- methane	Bronio- form	1,2-Di- chloro- cthanc	Carbon 1etra- chloride	Methyl- ene chloride	Benzene	<i>m</i> -Di- chloro- benzene	<i>n</i> -Di- chloro- benzene	1,2,4 tri- chloro- benzene	Penta- chloro- phenol	Fluor- anthene	1,12- Benz- perylene	3,4-llen zopyrene	Indeno- (1,2.J-ed) pyrene
							10		100	101 10	101	11 00	1012 001	, 100 BC1	101 221	116 1171
Annandale, Ya.	41: -0.6	21: -0.6	91: - 01/	(00 - 77)	1	11-09	26-70	(4 E - U3)	(4 G – UV)	(1 - 27)	(11 - 20)		100-00		100-20	
Atlanta, Ga.	1 E - 06	213 - 06	46-06	(2E-06)	(6E — 09)	(1E-09)	(5E-08)	(4E-08)	(4 E - 09)	(2E - 10)	(8E – 11)	3E - 08	(8 E - 09)	(3E – 08)	(2E - 0X)	(16-07)
Battimore, Md.	9.6 0.7	21: - 06	515-06	(2E-06)	(6E – 09)	(1 E – 09)	(2E08)	(4E-03)	(4 E – 09)	(2E – 10)	(8E – 11)	1E – 08	(8E – 09)	(3E – 08)	(SE – 0X)	(1E - 07)
Billings, Mont.	215 - 07	61: - 07	215-06	(2E-06)	(6E - 10)	(1E - 09)	(2E - 08)	(4E-08)	(4E-09)	(2E - 10)	(8 E - 11)	d	((10 – 18)	(3E - 08)	(5E - 08)	(1 E = 07)
Birmingham, Ma.	1E - 06	2E - 06	26-06	(2E - 06)	(1E - 08)	(1E - 09)	(2E – 08)	(8E – 08)	(4 E - 09)	(2E - 10)	(8E – 11)	3E – 09	(8E – 09)	(3E - 08)	T	(1E - 07)
Brownsville, Tex.	5E-07	11: - 05	215 - 04	2E - 0.3	(6E - 09)	(1E - 09)	(215 - 08)	(4 E - 03)	(4E - 09)	(2E - 10)	(8 E – 11)	(IE – 09)	(8 E – 09)	(3E – 08)	(5E-08)	(1E – 07)
Cupe Girardeau, Mo.	3E - 06	61E – 06	1E05	6E-06	(1E - 08)	1E - 07	(5E – 08)	6E – 07	2E – 08	7E – 09	2E — 07	6E – 09	(SE – 09)	(3E-08)	(5E – 03)	(1E - 07)
Columbus, Ohio	3E - 06	51: - 06	11505	(215 - 06)	(1 E - 08)	• d	(5E - 08)	(4 E - 08)	(4 E – 09)	(2E – 10)	(8 E – 11)	(1E – 09)	(XE – 09)	(JE-0S)	(5E - 0X)	(1E - 07)
Dusten, Ohio	4E - 07	215 – OA	2E = 0.5	16 - 04	(1E - 09)	$(1 \pm -09)$	(2E - 08)	(4E ~ 0S)	(4E-09)	(2E-10)	(8 E - 11)	1E – 09	(8 E – 09)	(3E - 08)	(SE – 0X)	(16-07)
Elizabeth, N. J.	2E – 06	315 - 06	715 - 016	(2E – 06)	(6E - 09)	(1E - 09)	(SE - 08)	(4E - 0N)	(4E – 09)	(2E-10)	(8E – 11)	6E - 09	(SE – 09)	(3E - 08)	(SE - 0X)	(1 E 07)
Tort Worth, Tex.	6E – 08	315 – 06	315-05	213 - 04	(01 - 10)	115-09	(2E - 0S)	(3E-03)	(4E - 09)	(2E-10)	(8E – 11)	(1 E - 09)	2E - 08	(3E-08)	( <b>5</b> E - 08)	(1E – 07)
Huntington, W. Va.	2E - 06	61: 0.6	31E – 05	4E-05	215 - 0.8	2E – 07	(2E - 08)	4 E - 08	1 E – 07	4E - 07	5E - 10	1E – 08	(SE – 09)	(3E-08)	(SE – 0X)	(1E - 07)
lackson, Miss.	4E - 06	31: - 06	915 — UK	(2E – 06)	(1 E - 03)	d	3E - 08	(4 E - 08)	(4E-09)	(2E – 10)	(8 E – 11)	2E-08	(SE – 09)	(3E-08)	(SE – 08)	(IE – 07)
Jersey City, N. J.	9E - 07	2E - 06	ME = 0.6	(2E - 06)	(6E - 09)	(1E – 09)	(2E - 08)	(SE0S)	(4E – 09)	(2E-10)	(8E - 11)	1E – 07	2E – 08		(SE - 0%)	(1 E - 07)
Sunsas City, Mo.	61: - 07	81:07	215 06	(215 - 06)	(61:09)	(11: -0.0)	(5E - 08)	(8E - 0k)	(4 E – 09)	11: - 09	(8 E - 11)	1E – 08	(81: - 09)	(36 - 08)	(5E-08)	(1 E - 07)
incoln Neb.	1E – 07	1E – 06	SE 06	7E - 05	(6E - 10)	(1 E - 09)	(2E - 08)	(4E - 03)	(4E-09)	(2E - 10)	(8 E - 11)	1E – 09	(8E – 09)	(3E – 0S)	(SE - 0S)	11E - 071
Los Angeles, Calif.	7E - 07	11: - 06	515 - 06	(2E - 06)	(6E - 09)	(1 E - 09)	(2E - 0S)	1	(4E - 09)	(2E - 10)	(3E-11)	7E - 09	(8E - 07)	(3E - 08)	(SE - 08)	$\{1E = 07\}$
Munchester, N. H.	115 - 06	2E - 06	715 - 0.7	(2E - 06)	(1 E - 03)	(1E - 09)	4 E - 08	(4 E - 08)	(4 E – 09)	1	(8E - 11)	6E – 09	(SE - 09)	(3E-08)	(5E - 0%)	(1E - 07)
Melbourne, Fla.	8E – 06	21: -05	1E = 0.4	1E - 04	(1E - 08)	(1 E - 09)	(4 E - 0S)	(4 E – 08)	(4 E – 09)	(2E – 10)	(SE - 11)	2E-08	(8E – 09)	(3E – 08)	(5E - 0S)	(1E - 07)
Montgomery, Ala.	3E - 06	213 06	213 06	(2E - 06)	(1E - 03)	(1 E – 09)	(7 E - 08)	(4 E - 08)	1.	(2E-10)	(8 E - 11)	6E – 09	(8E – 09)	(3E-08)	1	(1E - 0.7)
Newport, R. I.	3E - 06	7E = 0.6	21: -05	2E - 05	(1E - 08)	(1E – 09)	Ь	(SE – OS)	(4E-09)	(2E - 10)	(8E-11)	7E – 09	8E – 09	(3E - 08)	1	(1E - 07)
Norfolk, Va.	3E - 06	7E - 06	415 - 0.6	(2E-06)	(1 E - 0.8)	(1E - 09)	(2E – 0S)	(4 E - 08)	1	(2E-10)	(8 E - 11)	3E – 09	(8 E – 09)	(JE-08)	(SE – 08)	(1E - 07)
Passaic Valley, N. J.	2E – 06	3E - 06	6E - 06	(215 - 06)	(6E - 09)	9E - 09	(SE - 08)	(4E - 0S)	(4E - 09)	(2E-10)	5E-10	IE – 08	(8E-09)	(JE – 08)	1	(1E - 07)
Sucramento, Calif.	5E-07	815-07	115 - 0.6	(2E – 06)	(61: - 09)	(1 E ~ 09)	(2E - 08)	(8E-08)	(4 E - 09)	(2E-10)	(8E-11)	9E - 09	(8E – 09)	(3E-08)	1	(IE – 07)
Topeka, Kan.	2E-06	7E - 06	45-05	1E – 04	(1 E - 08)	4E - 08	(5E – 08)	(SE – 08)	(4E – 09)	(2E – 10)	(8E – 11)	4E-09	(8E - 09)	(JE-03)	(5E - 08)	(1E - 07)
1F - 06 represents 1 × 10 ⁻⁶ in ordinary scientific notation. Chemicals	× 10°° in	ordinary so	ientific notat	ion. Chemi	cats listed :	ire those fo	r which th	ie average r	isk exceeds	1 IE - 08 in	at least on	ic city. Pari	entheses in	dicate that	listed are those for which the average risk exceeds 1E-08 in at least one city. Parentheses indicate that the chemical was not	al was not

TABLE 5a. R. The Average Risk of Drinking Water of Selected U.S. Cities

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Indicale one city. Parentneses - US in at least 1E = 06 represents 1 × 10⁻⁶ in ordinary scientific notation. Chemicals listed are those for which the average risk exceeds 1Edetected; inside the parentheses is the average risk for exposure at the detection limit. • P indicates that the chemical was detected but that the quantity present was not given. CROCHERT MIL THE RISKS OF DRINKING WATER

TABLE 5h. R. The Median Risk of Drinking Water of Selected U.S. Cities

								0								
	Chluro- furnt	Bromodi- chloro- methane	Dibromo- chloro- methane	Bromo- form	1,2-Di- chloro- cthane	Carbon tetra- chloride	Methyl- cne chloride	Bis(2- chioro- cthyl)cther	Benzene	<i>m</i> -Di- chloro- benzene	<i>u</i> -Di- chloro- benzene	1,2,4 (ri- chloro- benzene	Penta- chloro- phenol	J,4 Benzo- Nuor- anthene	3.4-Ben- zapyrene	Indeno- (1.2.3-ed) Pyrene
Aerondale: Va	90 11	15 - 07	11: - 07	(11: -07)	(4 F = 09)	3E - 10	2E-09	(2E - 10)	(4E - 08)	(2E - 10)	(16 - 11)	(9E - 12)	3E - 09	(1 E - 09)	(1 E - ()8)	(21: - 0.1)
Atlanta Ga	4.6 - 07	4E - 07	51: - 07	(1E - 07)	(2E - 09)	(3E-10)	(2E - 09)	(2E - 10)	(4E - 0S)	(2E-10)	(1E - 11)	(9E - 12)	4E – 09	(1E - 09)	(1E – 08)	(2E – 0'1)
Baltimore, Md.	3E-07	36-07	51: 07	(11E - 07)	(21: 09)	(JE - 10)	(1E-09)	(2E - 10)	(4 E - 08)	(215 - 10)	(1E - 11)	ī	1E 09	(1 E – 09)	(1E - 0.8)	(21: - 0'')
Bulings, Mont.	6.E - 0.8	11 - 07	215 - 07	(115 - 07)	(2E - 10)	(JE - 10)	(1E-09)	(2E - 10)	(4 L - 08)	(2E-10)	(1E - 11)	(9 E - 12)	Ь	(1 E - 09)	(IE-08)	(2E - 1))
Beeningham, Ala.	4E 07	4E - 07	3E = 07	(1E - 07)	(4E - 09)	(3E - 10)	(IE – 09)	(2E - 10)	(8E - 08)	(2E - 10)	(16 - 11)	(9 E – 12)	3E – 10	1	(1E-08)	(2E - 00)
Brownsville, Tex.	215 - 07	31! - 0.6	31: - 05	111 - 0.4	(2E - 09)	(36 - 10)	(1E - 09)	(2E - 10)	(4 E - 08)	(21: - 10)	(11 - 11)	(9 E – 1 2)	(2E - 10)	(1E - 09)	(1E - 08)	(215 - 09)
Cape Girardenn, Mo.	1E-06	1E = 0.6	1E = 0.6	M:  – 07	(41: - 09)	51208	(212 – 09)	- 31 09	6E - 07	16~09	5E - 10		7E - 10	(1E - 09)	(1E - 03)	(2E - 09)
Compus, Ohio		11! - 06	11] = 06	(115 - 07)	(415 - 09)	• 1	(2E – 09)	9E – 10	(415 - 03)	(2E-10)	(11 - 31)	- 12)	(2E - 10)	(11) - 04)	(115 - 08)	(215 - 09)
Dayton, Ohio	1E - 07	4 E - 07	213 - 06	5E – 06	(vi E - 10)	(3E-10)	(1E – 09)	(2E - 10)	(4 E ~ 0S)	(2E - 10)	1	_	2E-10	(1 E - 09)	(1 = -0.8)	(2E - 09)
Elizabeth, N. J.	7E – 07	6E - 07	SE = 07	(1E – 07)	(2E – 09)	(JE – 10)	(2E-09)	(2E - 10)	(4 E – 08)	(2E - 10)	(11 - 31)	ĥ	7E - 10	1	(1 = -0.8)	(21: -0.9)
Fort Worth, Tex.	21E – 08	7E - 07	3E – 06	1E-05	(2E-10)	4E - 10	(IE – 09)	(2E – 10)	(8 E - 08)	(2E - 10)	(1E-11)	(9 E - 12)	(2E - 10)	1	(1 E - 08)	(215 - 0.9)
Huntington, W. Va.	6E - 07	1 E – 0á	3E – 06	2E – 06	5E – 09	5E-0S	(1E – 09)	(2E - 10)	4E-08	6E – 09	3E - 08	5E - 11	1E - 09	1	(1E - 03)	(2 E - 09)
Unckson, Miss.	IE - 06	6E – 07	4E - 07	(1E – 07)	(4 E – 00)	Ы	2E – 09	(21E – 10)	(4 E - 08)	(21: - 10)	(11 - 31)	(9E - 12)	2E - 09	(1 E - 09)	(1 = -0.8)	(2E - 19)
Jersev City, N. J.	3E - 07	3E - 07	3E – 07	(115 - 07)	(2E-09)	(3E - 10)	(1 E 09)	(2E – 10)	(SE - US)	(2E-10)	(11 - 11)	(9E - 12)	1E - 08	(1E - 09)	(1E - 0S)	(2E - 0'')
Kansas City, Kan.	213-07	2E - 07	21: - 07	(11: - 07)	(2 LE – 09)	(3E - 10)	(2E09)	(2E – 10)	(815 - 08)	1	SE-11	(9E - 12)	1E - 09	1	(1E - 0S)	1
L. acota, Neb.	4E - 08	2E – 07	9E – 07	415-06	(2E 10)	(3E - 10)	(1E-09)	2E – 10	(4 E ~ 08)	(2E - 10)	(1E - 11)	1	2E - 10	1	(1 E - 03)	(
Los Angeles, Calif.	26 - 07	3E 07	61: - 07	(1E-07)	(2E – 09)	(JE-10)	(1 E – 09)	(2E – 10)	(4 E - 03)	(2E - 10)	(1E - 11)	1	8E - 10	1	(1E - 08)	
Machener, N. H.	JE = 07	4E = 07	$S\Gamma = 0S$	(115 - 07)	(41: 09)	(31 - 10)	2E – 09	(2E – 10)	(4 E - 08)	(2E - 10)	(1E - 11)	1	7E - 10	1	(1E - 03)	(
Melbourne, Fla.	21E – 06	4E - 06	11:05	61: - 06	(415 - 09)	(JE – 10)	(2E-09)	(2E – 10)	(41: -08)	(2E - 10)	(1 E - 11)	1	$\frac{2E}{2E} - 09$	1	(1 E - 03)	1
Nonigomery, Ala.	9E – 07	45-07	2E – 07	(1E-07)	(41:-09)	() 2 2 - 10)	(3E-09)	(2E-10)	(4E - 08)	(2E - 10)	(1 - 2)	1	76-10	1	(1 E - 08)	(A) = 32)
Newport, R. I.	1E – 06	11: - 06	2E = 06	91: - 07	(4 E – 09)	(3E - 10)	۵.	(2E - 10)	(8E - 08)	(2E - 10)	(11 - 21)		86 - 10	1		(50 - 03)
Zerfolk, Va.	9E-07	1E - 06	415 - 07	(11: - 07)	(네티 - 09)	(01 - 30)	1	(2E - 10)	(4E - 08)	(2E - 10)	(112 - 11)	(71 - 17)	JE - 10	1	(1 E - 03)	(A) - 37)
Passaic Valley, N. J.	6E-07	6E-07	6E-07	(1E-07)	(2E – 09)	JE – 09	t	(2E-10)	(4E - 08)	(2E - 10)	(11 - 31)	5E - 11	1E - 09	1	(1E + 08)	(2E - 09)
Sacramento, Calif.	2E - 07	2E – 07	115 - 07	(113 - 07)	(215 – 09)	(315 - 10)	1	(2E-10)	(8 E - 08)	(2E - 10)	(11 - 11)	1	1E - 09	1	(11: -03)	(7E - UV)
Topeka, Kan.	6E – 07	1E – 06	4E - 06	SE - 06	(4E – 09)	1E – 08	(2E – 09)	(2E-10)	(8E - 08)	(2E - 10)	(1E - 11)	(9t - 12)	56-10	(1 E - UV)	(1 = - 02)	17
in the sum 15-06 tenescents 1 x 10-6 in ordinary scientific						the street of the	m of the	madiane See	· Table 4 fr	sr the med	ian of the	16 - 1	06 represen	-01 × 1 str	6 in ordinal	y scientific
Note that the median of the sum of the suffice for a given cut is not under the sufficiency of the median of the decided in field of the former we not the median field of the former of the median field of the former of the second	an-of the s			given city	15 810001 1	wice the su		Duronthave	indicate the	i the chird	nical was no	of detected	: inside the	Darenthese	s is the med	ian rísk for
notation. Chemicals II	sted are Inc	Did which	enone out e	III LISK CYCC			Olle cut.	ו מו רווווורזרז								

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action. Chemicals listed are those for which the median risk exceeds 1.5 +09 in exposure at the detection limit.
• P indicates that the chemical was present but that the quantity was not given.

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5c. R., 98th Percentile of the Distribution for Annual Risk of Drinking Water of Selected Cities

			TABLE	TABLE Sc. R., 98th P	18th Percer	uile of the	ercentile of the Distribution for Annual Risk of Drinking Water of Selected Cities	for Annual	l Risk of D	rinking W:	uter of Sele	cted Cities			.¥*	
	Chloro- form	Bromodi- chloro- methane	Dibromo- chloro- methane	Bromo- form	1,2-Di- chloro- cthane	Carbon tetra- chloride	Methyl. ene chloride	Benzene	<i>m</i> -Di- chloro- benzene	<i>o</i> -Di- chloro- benzene	1,2,4 tri- chloro- benzene	Pentar chloro- phenol	Fluor- anthene	1,12. Henzperylene	3,4 Ben- zopyrene	Indeno- (1,2,3-cd) pyrene
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	36 - 05 56 - 06 56 - 06 56 - 06 56 - 05 37 - 06 56 - 05 56	1E     05       1E     05       1E     05       1E     05       1E     05       2E     05       3E     05       3E     05	7E - 06 3E - 05 3E - 05 3E - 05 2E - 01 2E - 04 8E - 05 5E - 04 5E - 04 5E - 04 5E - 04 5E - 05 5E - 04 5E - 05 5E - 04 5E - 05 5E - 06 5E - 0	(255-05) (255-05) (255-05) (255-05) (255-05) (255-05) 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-04 (255-05) 755-04 (255-05) 755-03 755-04 (255-05) 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-03 755-0	(3 E - 08)       (4 E - 08)       (4 E - 08)       (3 E - 08)       (4 E - 08)       (5 E - 08)       (6 E - 08)       (7 E - 08)       (8 E - 08)       (8 E - 08)       (8 E - 08)       (8 E - 08)       (	$\begin{array}{c} 7E = 09\\ 1E = 06\\ 8E = 09\\ 8E = 09\\ 8E = 09\\ 1E = 06\\ 1E = 06\\ 7E = 09\\ 7E = $	$\begin{array}{c} 3E - 07\\ 3E - 07\\ (2E - 07)\\ (2E - 07)\\ (2E - 07)\\ (3E - 07$	(4E - 08)       (3E - 08)	136     136     136       136     136     136       136     136     136       136     136     138       136     136     138       136     136     138       136     136     138       136     136     138       136     136     138       136     136     138       136     136     138       136     136     138       136     138     138       136     138     138       136     138     138       136     138     138       136     138     138       136     138     138       136     138     138       136     138     138       136     138     138       136     138     138       136     138     138       136     138     138       136     138     138       136     138     138       136     138     138       137     138     138       138     138     138       139     138     138	(110 - 09) (110 - 09)	(6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (6.6.1 - 10)         (7.1 - 10)         (8.6.1 - 10)         (9.6.1 - 10)         (16.1 - 10)         (16.1 - 10)	$\begin{array}{c} 2E + 07 \\ 3E - 07 \\ 3E - 07 \\ 3E - 07 \\ 7E - 08 \\ 2E - 08 \\ 2E - 08 \\ 2E - 08 \\ 3E - 08 \\$	(4 (2 (1)))       (4 (2 (1)))       (4 (2 (1)))       (4 (2 (1)))       (4 (2 (1)))       (4 (2 (1)))       (4 (2 (1)))       (4 (2 (1)))       (4 (2 (1)))       (4 (2 (1)))       (4 (2 (1)))       (4 (2 (1)))       (4 (2 (1)))       (4 (2 (1)))       (4 (2 (1)))       (4 (2 (1)))       (4 (2 (1)))       (4 (2 (1)))       (4 (2 (1)))       (4 (2 (1)))       (4 (1))       (4 (1))       (4 (1))       (4 (1))       (4 (1))       (4 (1))       (4 (1))       (4 (1))       (5 (1))       (6 (1))       (7 (1))       (7 (1))       (7 (1))       (7 (1))       (7 (1))       (7 (1))       (7 (1))       (7 (1))       (7 (1))       (7 (1))       (7 (1))       (7 (1))       (7 (1))       (7 (1))       (8 (1))       (9 (1))       (1))       (1))       (2))       (2))       (3))       (4))       (4))       (4))       (1))      <	(16 - 07) (16 - 07) (17 -	00000000000000000000000000000000000000	<ul> <li>(66 - 07)</li> </ul>
$h_{\mu}$ indicates that the chemical way present out not during $d_{\mu}$	כ כחכוחונים	may prevent	מה וומו והם	יוווחוויים.												

CROUCH ET AL.: THE RISKS OF DRINKING WATER

## CROUCH ET AL.: THE RISKS OF DRINKING WATER

TABLE	6.	Risks of Other Activities	
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	Everyday Risks			
	Time to Accumulate a One in a Million Risk of Death	Average Annua Risk per Capita		
	Living in the United States			
Motor vehicle accident	LS days	$2 \times 10^{-4}$		
Falls	6 days	$6 \times 10^{-3}$		
Drowning	10 days	$4 \times 10^{-3}$		
Fires	13 days	$3 \times 10^{-3}$		
Firearms	36 days	$1 \times 10^{-3}$		
Electrocution	2 months	$5 \times 10^{-6}$		
Tornados	20 months	$6 \times 10^{-7}$		
Floods	20 months	6 × 10 ⁻⁷		
Lightning	2 years	$5 \times 10^{-7}$		
Animal bite or sting	4 years	$2 \times 10^{-7}$		
General	Occupational Risks			
manufacturing	4.5 days	$S \times 10^{-3}$		
trade	7 days	$5 \times 10^{-5}$		
service and government	3.5 days	1 × 10 ⁻		
transport and public utilities	1 day	$4 \times 10^{-4}$		
agriculture	15 hours	$6 \times 10^{-4}$		
construction	14 hours	$6 \times 10^{-4}$		
mining and quarrying	9 hours	$1 \times 10^{-3}$		
Specific	7 110013	1 ~ 10		
coal mining (accidents)	14 hours	$6 \times 10^{-4}$		
police duty	1.5 days	$2 \times 10^{-4}$		
railroad employment	1.5 days	$2 \times 10^{-4}$		
fire fighting	11 hours	8 × 10 ⁻⁴		
······································	Other One in a Million Risks			
Source of Risk	Type and Amount of Exp	osurc:		
	Examples			
Cosmic rays	one transcontinental round trip by air living 1.5 months in Colorado compare camping at 15,000 feet for 6 days comp			
Other radiation	20 days of sea level natural background 2.5 months in masonry rather than woo	I radiation		
	‡ of a chest X ray using modern equipn	nent		
Eating and drinking	40 diet sodas (saccharin)			
	6 pounds of peanut butter (aflatoxin)			
	180 pints of milk (aflatoxin)			
	200 gallons of drinking water from Mia			
	90 pounds of broiled steak (cancer ris	k only)		
Smoking	2 cigarettes			

From Crouch and Wilson [1982].

exactly related to probable risk, as they should be (compare Table 7 and Table 8).

For individual contaminants, a  $10^{-6}$  risk per year leads to the concentrations shown in Table 7. We note that the concentrations of chloroform giving an average risk of  $10^{-6}$ /year is 30". less than the federal standard [Federal Register, 1979]. Also, rarely is only one important contaminant, such as chloroform, present in drinking water, and so the risk from drinking water in compliance with federal regulations may be substantially higher. For mixtures the calculation should be adjusted to limit the total risk to  $10^{-6}$  per year, by using the techniques described above.

If the EPA wish to be inconsistent in the setting of riskbased standards, they should state why. Among possible reasons are that balomethanes are costly to reduce and enter water supplies as byproducts of a desirable technology, e.g., chlormation of water supplies to prevent disease. Trichlotoethylene and tetrachloroethylene usually enter the supplies

as industrial pollution, which might more easily and cost effectively be controlled. According to this idea, different riskbased standards may be in order; for example, use of  $10^{-5}$ /year for the mixture produced through chlorination and

TABLE 7. Contaminant Concentration for  $R = 10^{-6}$ /year

	Rag	Ŕ	Ř	EPA Potential MCL's*
Trichloroethylene	165	1100	3400	5 to 500
Tetrachloroethylene	75	500	1500	5 to 500
Carbon Tetrachloride	30	200	625	5 to 500
1.1.1-Trichloroethane	5000	36,000	145,000	1000
1.2-Dichloroethane	10	80	250	1 to 100
Vinyl Chloride	5	35	125	1 to 100

Values in pp 1

"From Luck & el Federal Register [1982]. MCL is the maximum contaminant level permissible.



TABLE 8 - Inviced Upper Limit Lifetime Cancer Risk for Indicated Drinking Water Concentrations by Two Concentrations by Two Cancer Risk Calculations

	Projected Upper Limit Excess Lifetime	Concentr Drinking µg	Water,
	Cancer Risk	CAG	NAS
Trichloroethylene	10-6	280	450
-	10 - *	28	45
	10 - 4	2.8	4.5
Tetrachloroethlyene	10-*	90	350
	10 - 5	9	35
	10	0.9	3.5
Carbon Tetrachloride	10 - *	40	450
	10 - 5	1	45
	10-*	0.4	4.5
1.2-Dichloroethane	10 - 6	95	70
	10 - 5	9.5	7
	10 - 4	0.95	0.7
Vinyl Chloride	10-6	200	100
-	10-5	20	10
	10-4	2	1

Assumes: Lifetime exposure (70 years) by 70-kg adult. Consumption of 2 liters of water per day. Nonthreshold toxicity mechanism is operative at low doses in humans. Assimilation in humans at low doses is the same as animals at experimental doses. Interspecies (animal.human) dose scaling is proportional to body surface area. Note: CAG is the EPA Carcinogen Assessment Group. NAS is the National Academy of Sciences Safe Drinking Water Committee. Copied verbatim from Federal Register [1982, Table 3].

 $10^{-6}$  year for all industrial chemicals. Other reasons for choosing different risk levels for different types of chemicals may be to account for persistence or the use of certain chemicals as surrogates for others, possibilities not included in the scheme here propounded.

## APPENDIX: ORAL POTENCIES FOR PAH AND SOME OF THE CHLORINATED HYDROCARBONS LISTED IN TABLE 2

Risk estimates are conventionally obtained from human data or from lifetime experiments on animals in which the animals are exposed to the test chemical in the same manner as humans would be. When lifetime animal results or human data are not available, it has usually been maintained that risks cannot be estimated, as was the case when the National Academy of Sciences chose not to estimate the risks of 3.4benzopyrene in drinking water [NAS, 1977]. However, we believe that the available data are adequate for risk estimation in many cases where human or lifetime animal studies cannot be found.

For all of the chemicals listed in Table 2 we believe there is sufficient data for risk estimation. The chemicals listed in Table 2 for which we have yet to derive values of potency are all either chlorinated hydrocarbons or polycyclic aromatic hydrocarbons (PAH).

#### Polycyclic Aromatic Hydrocarbons (PAH)

The potencies of these compounds are derived from animal carcinogenesis studies. Cancer has been induced in laboratory animals with the PAH 3.4-benzopyrene in numerous experiments. In several of these experiments, 3.4-benzopyrene was given orally and the experiment lasted for a long period of time, albeit not for the entire lifetime of the animals (Table A1). We obtain a potency for 3.4-benzopyrene from these ex-

periments by applying a factor to correct for the length of the experiment. For the rest of the PAH listed in Table 2, we use results from mouse skin tests, which were performed to compare the carcinogenic activities of the various PAH. The potency of a particular PAH is extrapolated from the potency of 3,4-benzopyrene by comparing its skin activity with that of 3,4-benzopyrene.

When given orally, 3,4-benzopyrene consistently induces tumor growth, R. H. Rigdon and coworkers have extensively studied its induction of leukemias and lymphomas, as well as cancers of the stomach and lung, in laboratory mice [Rigdon et al., 1967, 1969; Riydon and Neal, 1966]. Unfortunately, their experiments were too short to be of use in potency calculations. There are, however, several rodent experiments lasting a year or longer, and we are much more comfortable using these long experiments for potency estimation. J. Borneff (Borneff [1963], as reported by U.S. Department of Health; Education and Welfare [1973a], and Borneff et al. [1968], as reported by U.S. Department of Health, Education and Welfare [1973b]) induced lung adenomas and stomach papillomas and carcinomas with 3,4-benzopyrene in experiments that lasted 500 days (see Table A1). He was able to affect the activity at a particular site by adding mineral oil or detergents to the feed or drinking water and administering the chemical either in the drinking water or feed. However, the potency, which we define as the highest significant value for any site in a particular experiment, did not vary by more than a factor of 4 in the Borneff experiments (Table A1).

Several other investigators also induced cancer of the forestomach in long-term experiments. All of the experiments we know of which lasted a year or longer where 3,4-benzopyrene was administered orally are listed in Table A1. We have estimated potencies from these experiments, using methods described in the body of the text. The values obtained are listed in the first column of Table A1. None of the experiments used were run for the lifetime of the animals, and so the potencies listed in the first column are underestimates; had the experiment run for the lifetime of the animals, more cancers probably would have been observed.

To take this into account, potency values are multiplied by a compensating factor f:

$$\beta_{\text{corrected}} = \beta_{\text{uncorrected}} f \tag{A1}$$

Armitage and Doll [1954] have observed that for many human tumors, incidence R by a given age t is proportional to  $t^2$ , where  $\lambda$  is characteristic of a tumor:

$$R \sim t^2$$
 (A2)

Hence the potency from a short experiment is underestimated by a factor

$$f = \left(\frac{t_1}{t_c}\right)^t \tag{A3}$$

where  $t_i$  is the natural lifetime of the animal and  $t_i$  is the age of the animal at the end of the experiment. For rodents,  $t_i$  is approximately 2 years. For human tumors of the type listed in Table A1,  $\lambda$  is approximately 5.6. In rodents, we assume that it is the same and so correct the values in Table A1 by computing:

$$\beta_{corrested} = \beta_{uncorrected} \left(\frac{24 \text{ months}}{l_e}\right)^{5.4} \quad (A4)$$

	Lumor Type	Method of BAP Administration	Remarks	Duration of	Animal Age at the End of Experiment	ſ	R _{correcte} d
			Chouroulinkov et al. [1967]				
0.123	forestomach papilloma	feed	·	14 months	496 days	8.7	1.07
			Borneff et al. [1968]				
0.130	stomach papilloma			(500.1			0.60
0.245	stomach carcinoma	drinking water	surfactants added to water	(500 days)*	(556 days)	(4.6)	1.13
0.182	stomach papilloma ]	drinking water	surfactants and potassium	(500 days)*	(556 davs)	(4.6)	0.84
0.219	stomach carcinoma	utitiking water	chromate added to water	(Stor days)	(550 days)	(4.0)	1.01
			Borneff [1963]				
0.367	lung adenoma	¢ .		60.0 I			1.68
016	stomach pap. +ker.t	feed	mineral oil and detergent	500 days	(556 days);	(4.6)	0.74
0159	stomach carcinoma [	drinking water	added to drinking water detergent added to water	(0) da	(556 davs)		0.73
0.274	stomach papilloma J	drinking water	detergent added to water	500 days	(220 03/2)	(4.6)	1.26
0.066	lung adenoma	drinking water	mineral oil and detergent	·500 davs	(556 days)	(4.6)	0.30
0.546	stomach papilloma	uninking water	added to drinking water	Sco days	().0 02)3/	(4.0)	2.51
0.138	lung adenoma	feed	mineral oil added to feed	500 days	(556 davs)	(4.6)	0.63
0.021 0.143	stomach papilloma ( lung adenoma				(000 00,0)	(	0.10 10.66
0.143	stomach papilloma	drinking water	mineral oil added to feed;	500 days	(556 days)	(4.6)	0.86
0.12.5	stomach paphionia j	-	detergent added to water	-	•		{0.50
			Fedorenko et al. [1967]				
0.149	stomach carcinoma	gavage: 10 times	to fasting mice	19 months	21-22 months	1.9	0.28
0.807	stomach papilloma	Entage: 10 miles	ç	17 110/11/13	21 22 months	1.5	1.49
			Gibel [1964]				
0.015	esophagus and fore- stomach papilloma	daily gavage for 1 year	to Sprague Dawley rats	12 months	67 weeks	11.7	0.17
			Biancinifiori et al. [1967]				
0.27	stomach carcinoma	gavage, 30 times	· · · ·	65 weeks	73 weeks	7.3	1.96

TASI ( AL	Animal Potency of 3.4 Benzopyrene, Ora	I Route, Derived From Animal Bioassa	is Longer Than One Year

Except for the Gibel experiment, all experiments were performed on mice, of a variety of strains.

*It was assumed that the 1968 Borneff experiment was run under the same protocol as the 1963 Borneff experiment.

+For stomach papillomas and keratosis combined.

‡It was assumed that the experiment began when the animals were 8 weeks of age.

where  $t_e$  is expressed in months. The corrected potencies are listed in the last column of Table A1. The values of potency obtained are surprisingly consistent, considering the variety of strains and experimental conditions, with the largest only a factor of 15 from the smallest value. To obtain a single value for 3.4-benzopyrene, we took a geometric mean. Using a single value of potency for each of the references, the geometric mean was found to be 0.93 with a logarithmic standard deviation of 0.87.

No long-term oral bioassays were found for any of the other PAH listed in Table 2, although mouse skin-painting experiments were available on all of them. In one skin-painting experiment, in which both 3.4-benzofluoranthene (B(b)F) and 3.4-benzopyrene were repeatedly applied, B(b)F was less potent by about a factor of 10 [Wynder and Hoffman, 1959]. In an initiation experiment, B(b)F was about 1/40th as powerful as 3.4-benzopyrene [Van Duaren et al., 1966]. We assume that B(b)F would be about 1/10th as potent as 3.4benzopyrene in an oral study. In applying this assumption, we note that we may be underestimating the relative potency of B(b)F by a factor of 5 or more ( $\beta = 9 \times 10^{-2}$ , but  $\sigma_x^{-2} = [\ln (5)]^2 + (0.87)^2 = 1.83^2$ ).

Haffman and Wynder [1966], as reported by International Agency for Research on Cancer [1973], topically applied indeno(1.2,3-cd)pyrene (IP) as well as 3,4-benzopyrene thrice weekly for 1 year. In this experiment, 3,4-benzopyrene was at least 20 times as active as IP. We assume that IP is less than 1 10th as potent as 3,4-benzopyrene through the oral route, but to account for the looseness of the extrapolation, assume a large uncertainty ( $\beta = 0.09$ ,  $\sigma_x^2 = [\ln (10)]^2 + (0.87)^2 = 2.46^2$ ). In an 80-week skin painting experiment, *Horton and Christian* [1974] tested the complete and cocarcinogenic properties of several PAH, including 3,4-benzopyrene and fluoranthene. In this experiment 3,4-benzopyrene was at least 40 times more potent than fluoranthene as a co-arcinogen and 30 times more potent as a complete carcinogen. Hence we assume a potency of  $3 \times 10^{-2}$  for fluoranthene, with large uncertainty ( $\sigma_x = 2.5$ ).

In one skin painting experiment, 1,12-benzperylene was not active at doses in which other PAH were highly active [Lijinsky and Saffiotti, 1965]. From the Van Duuren et al. [1966, 1970] experiments we can deduce that 1,12-benzperylene may be less potent than 3,4-benzopyrene in initiation by a factor of 250; however, 1,12-benzperylene can also act as a cocarcinogen and was found to be half as potent as pyrene in cocarcinogenic activity [Van Duuren et al., 1973]. Pyrene was about 1/20th as potent as benzpyrene as a cocarcinogen in the Horton and Christian [1974] experiment. These facts indicate that, orally, 1,12-benzperylene may be less than 1/50th as potent as 3,4-benzopyrene ( $\beta = 0.02, \sigma_x = 2.5$ ).

## Chlorinated Hydrocarbons

A number of chlorinated hydrocarbons tested in the standard NCI bioassays have produced noncarcinogenic effects, such as liver and kidney toxicity, as well as liver cancers or leukemia in mice and kidney and/or endocrine cancer in rats. In fact, acute oral toxicity was found to be strongly correlated with carcinogenic potency for those chlorinated hydrocarbons

Chemical	LD _{se}	a <u>,</u>	ß. mg kg day	σ.	LD 30 References
Methylene Chloride	1270	1.36	$2.5 \times 10^{-3}$	2.13	Kimura et al. [1971]; NIOSH [1980]
p-Dichloro- benzene	2132	1.00	$1.4 \times 10^{-3}$	1 88	Spencer [1973]
a-Dichloro- benzene	500	0.36*	$6.9 \times 10^{-3}$	1.70	NIOSH [1980]
m-Dichforo- benzene	> 50		< 8.8 × 10 ⁻²	(2(X))	we assume that the meta isomer is no more than 10 times as toxic as the p- and o- isomers
1.2.4-tri- chloro- benzene	761	0.36*	$4.4 \times 10^{-3}$	1.70	Brown et al. [1969]
2.4-dichloro- phenol	1765	0.73	$1.7 \times 10^{-3}$	1.72	Kobayuski et al. [1972]; NIOSH [1980]; Vernot et al. [1972]
pentachloro- phenol	102.5	0 68	4.0 × 10 ⁻¹	1.69	NIOSH [1980]; Deichmann et al. [1942]; Gaines [1969]; Stohlman [1951]; McGaweck et al. [1941]

TABLE A2 Potency & Estimated from LD so Values for Some Chlorinated Hydrocarbons

•Since only one  $LD_{so}$  value was found, we assume that the uncertainty in the  $LD_{so}$  is the same as that for an arbitrary  $LD_{so}$  found in the literature [Zeise et al., 1982].

tested by the NCI [Zeise et al, 1983]. The relationship derived is used to predict carcinogenic activity for the untested chlorinated hydrocarbons in Table 2 given their acute toxicity. The relationship is

$$\tilde{\beta} = 6.5(LD_{50})^{-1.1}$$
 (A5)

where  $\beta$  is lognormally distributed with median log,  $(\tilde{\beta})$  and standard deviation  $(\sigma_1)$  1.52. Uncertainty measures on potency  $(\sigma_x)$  must include both the uncertainty from using the relationship  $(\sigma_1)$  and the uncertainty in the value of the LD₅₀ used  $(\sigma_2)$ :

$$\sigma_x^2 = \sigma_1^2 + (1.1^2)\sigma_2^2 \tag{A6}$$

The potency values derived, together with the  $LD_{50}$  values, their references, and the uncertainties, are listed in Table A2.

#### NOTATION

- b EPA Carcinogen Assessment Group's estimate of potency.
- $d_i$  average concentration of chemical *i* in drinking water,  $\mu$ gA.
- D average daily dose, mg/kg of bodyweight/day.
- E high to low dose extrapolation factor.
- $K_{ba}$  interspecies extrapolation factor, from species a to species b.
- r( ) probability density function.
  - R lifetime cancer risk.
  - R average annual cancer risk.
  - w median value of w.
  - w average value of w.
- w⁹⁸ 98th percentile of w.
  - x backround lifetime tumor incidence.
  - $\beta$  carcinogenic potency.
  - $\sigma$  standard deviation of log, (R).
- $\sigma_{1}$  standard deviation of log ( $\beta$ ).
- $\sigma_{1}$  standard deviation of log₂ (K).
- $a_1$  -standard deviation of  $\log_2(d)$  -

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E. Crouch, R. Wilson, and L. Zeise, Energy and Environmental Policy Center, Jefferson Physical Laboratory, Harvard University, Cambridge, MA 02138.

> (Received May 6, 1983) revised September 9, 1983) accepted September 13, 1983.)

#### 4.0 RISK ASSESSMENT

In establishing priorities for remediation or monitoring, it is important to consider the health risk associated with a specific sample or location. In most cases, relatively high concentrations of contaminants must be present to constitute an acute health hazard. However, extremely low dose levels of chronic toxicants can present a significant risk if exposure occurs over the lifetime of the organism. The chemical constituents and concentrations of soil and water samples have been reviewed in order to estimate their chronic toxic potential. The lifetime cancer risk associated with consumption of the various chemicals detected in soil and groundwater samples at the Manana - Mary Wheeler #1-E well site has been estimated using the procedures of Crouch et al. (1983). This procedure can be used to estimate the risk of individual chemicals, as was done for the soil samples; or, the procedure can be used to estimate the cumulative cancer risk for mixtures of chemicals, as was done for the groundwater samples. The risk equation takes into account the carcinogenic potential of each chemical, the variability of the data used to estimate carcinogenic potential, and the concentration of the chemical. The major limitations of this equation are that extrapolations must be made from the high dose used in animal testing to the low dose which occurs in most environmental samples; and, extrapolations must be made from the animal species in which toxicity testing was conducted and humans for which the risk is being estimated.

For groundwater, the risk equation assumes that a 70 kg human will consume 2 liters of water per day for an entire lifetime. The equation can be used to estimate the mean, median, and 98th percentile risk. The mean risk is typically used to assess the hazard of individual chemicals or

mixtures in most groundwater samples. The median risk is usually less than the mean risk, and should only be used if the risk of human exposure is extremely low; while, the 98th percentile risk is a conservative estimate and can be used if potable water wells are believed to be within the plume of contaminated groundwater. For the purposes of this review, the mean risk will be used.

#### 4.1 SOIL

The results from the estimation of risk for individual chemicals detected in the soil are provided in Table 4-1. The oral rat lethal dose or tumorigenic dose for chrysene and acenaphthene could not be located in the literature. However, chrysene is described as a potential carcinogen (Lewis and Tatken, 1984). Thus, the carcinogenic potential for chrysene should be comparable to that for fluoranthene. In addition, a comparison based on chemical structure suggests that the carcinogenic potential of acenaphthene should be comparable to naphthalene. The adsorptive properties of soil should prevent a significant amount of these chemicals from reaching groundwater. However, in order to provide a higher level of security, soil samples will be considered to present a potential risk to human health if the soil concentration incurs an estimated risk of greater than 10 in a million.

Three soil samples contained a concentration of chemical with an estimated risk above a mean of 100 in a million. Soil sample P-N contained 178 ppb fluoranthene which incurs an estimated risk of 148 in a million (Table 4-1). The risk associated with chrysene is likely equal to that for fluoranthene in this soil sample. The estimated risk associated with fluoranthene in soil sample T-1W is 137 in a million; this sample also contained xylene at a concentration which incurs an estimated risk of 1 in

	#1-E Well Site.				,,
Sample I.D.	Constituent	Conc. (ppm)	Mean R	Median R	98th Percentile R98
P-N	Fluoranthene	0.178	1.48E-04	2.21E-06	7.29E-04
	Chrysene	0.118			
T-1W	Fluoranthene	0.165	1.37E-04	2.05E-06	6.76E-04
T-2E	Napthalene	3.901	2.79E-05	2.65E-06	2.04E-04
	Acenapthene	3.584			
BG-1	1,2-Dichloropropane	0.61	7.07E-06	1.03E-07	3.45E-05
P-N	Toluene	1.2	1.61E-06	4.23E-07	1.10E-05
	Xylene	17.8	2.54E-05	7.05E-06	1.73E-04
P-W	Benzene	2.6	3.20E-06	1.04E-06	2.09E-05
	Toluene	34.3	4.59E-05	1.24E-05	3.15E-04
	Xylene	177.2	2.52E-04	7.02E-05	1.72E-03
P-S	Chlorobenzene	0.37	3.37E-06	1.48E-07	2.20E-05
	Xylene	0.76	1.08E-06	3.0 <u>1E-0</u> 7	7.38E-06
P-E	Xylene	2.1	2.99E-06	8.32E-07	2.04E-05
<u>T-1</u> W	Xylene	0.73	1.04E-06	2.89E-07	7.09E-06
T-1C	Toluene	0.16	2.14E-07	5.76E-08	1.47E-06
<u>T-1</u> E	Xylene	0.98	1.40E-06	3.88E-07	9.52E-06
<u>RP-1B</u>	Vinyl chloride	4.3	1.14E-04	3.44E-05	7.64E-04
T-2C	Toluene	0.26	3.48E-07	9.36E-08	2.39E-06
	Xylene	0.94	1.40E-06	3.72E-07	9.13E-06
T-2E	Benzene	0.22	2.71E-07	8.80E-08	1.77E-06
	Toluene	0.35	4.68E-07	1.26E-07	3.47E-06
T-3C	Chlorobenzene	0.19	<u>1.73E-06</u>	7.60E-08	1.13E-05

# Table 4-1. Risk Factors for Soil Samples Collected at the Manana - Mary Wheeler #1-E Well Site.

Note: This table revised 4/17/90.

a million (Table 4-1). The maximum risk associated with a soil sample at the Manana - Mary Wheeler #1-E well site was 252 in a million for xylene in soil sample P-W. This sample also had a relatively high risk for toluene (45.9 in a million). The concentration of vinyl chloride in soil sample RP-1B incurs an estimated risk of 114 in a million (Table 4-1). This soil may present the greatest risk of those investigated, as vinyl chloride is a suspected human carcinogen with a high vapor pressure. Thus, the potential for exposure via either air or groundwater exists.

Other soil samples with chemicals at a concentration which incurs an estimated risk of greater than 10 in a million include T-2E (naphthalene's risk is 28 in a million), P-N (xylene's risk is 25 in a million), and P-W (toluene's risk is 46 in a million). Thus, the soil sampling locations for which chemical concentrations suggest the greatest estimated carcinogenic risk appear to be P-N, T-1W, P-W and RP-1B (Table 4-1).

#### 4.2 GROUNDWATER

The presence of toxic chemicals in groundwater presents a more imminent risk to human health than does their presence in soil. Thus, any groundwater sample with a mean risk of greater than 1 in a million may indicate the need for continued monitoring at a minimum. At the conclusion of the risk assessment, it was noted that the maximum carcinogenic risk estimated for groundwater at the Mary Wheeler #1-E site is no greater than that estimated for drinking water from several large cities. The concentration of indeno(1,2,3-c,d)pyrene in sample EPNG-4 incurs the maximum estimated risk of 15 in a million. Methylene chloride, which has been suggested to have been introduced by the analytical laboratory, incurs an estimated risk of .009 in a million at a concentration of 2.1 ppb in water sample EPNG-1 (Table 4-2). The maximum estimated risk for a water

					98th
Sample	]	Conc.	Mean	Median	Percentile
<u>I.D</u> .	Constituent	(ppb)	<u> </u>	R	<b>R98</b>
EPNG-1	Methylene chloride	2.1	9.34E-09	1.85E-09	6.76E-08
	Xylene	0.34	4.84E-10	1.35E-10	3.30E-09
EPNG-2A	Benzene	8.1	9.98E-09	3.24E-09	6.51E-08
	Ethylbenzene	37.4	1.89E-08	5.09E-09	1.30E-07
	Xylene	192.4	2.74E-07	7.62E-08	1.87E-06
EPNG-2B	Benzene	1.6	1.97E-09	6.40E-10	1.29E-08
	Ethylbenzene	1.2	6.06E-10	1.63E-10	4.17E-09
	Xylene	12.1	<u>1.72E-08</u>	4.79E-09	1.18E-07
EPNG-3	Benzene	0.4	4.93E-10	1.60E-10	3.21E-09
	Ethylbenzene	4.3	2.17E-09	5.85E-10	1.49E-08
1	Xylene	13.54	1.62E-07	4.50E-08	1.10E-06
	Toluene	0.33	4.41E-10	1.19E-10	3.03E-09
EPNG-4	Benzene	0.95	1.17E-09	3.80E-10	7.63E-09
	Benzo(b)fluoranthene	2	5.50E-07	7.44E-08	4.06E-06
	Indeno(1,2,3-cd)pyrene	6	1.50E-05	2.23E-07	7.37E-05
S-5	Benzene	0.24	2.96E-10	9.60E-11	1.93E-09
	Toluene	0.32	4.28E-10	1.15E-10	2.94E-09

 Table 4-2. Risk Factors for Individual Chemicals in Water Samples Collected at the

 Manana - Mary Wheeler #1-E Well Site.

Note: This table revised 4/17/90.

sample was incurred by sample EPNG-2A. For this sample, the concentration of xylene yields an estimated cancer risk of .27 in a million, (Table 4-2). Xylene in samples EPNG-2B and EPNG-3 is present at concentrations which incur estimated risks of .017 in a million and 0.162 in a million, respectively.

The mean estimated cumulative cancer risk for the mixtures of chemicals in the groundwater samples (see Table 4-3) ranged from a low of .0007 in a million for sample S-5 to the high of 15 in a million for the mixture of chemicals in sample EPNG-4. Except for sample EPNG-4, all groundwater samples had an estimated risk of less than one in ten million. Crouch et al. (1983) suggest that a person exposed to a risk of 1 in a million per year has a life expectancy reduced by only one day. In addition, their review of drinking water quality from cities in the United States indicates that the mean annual cancer risk was greater than one in a million for all 25 cities investigated (Crouch et al., 1983).

Only one groundwater sample incurred an estimated risk which was appreciably greater than 1 in a million. For each of the other groundwater samples, the estimated carcinogenic risk was less than one in ten million. These data indicate that only one well at the Manana - Mary Wheeler #1-E well site may present a threat to human health if the plume of contamination reaches wells used for the production of drinking water. In addition, drinking water from 5 of 25 cities reviewed by Crouch et al. (1983) had estimated cancer risks of greater than 100 in a million.

Mary Wheeler #1-E Well Site.							
Well		Risk (x 10^6)*					
<u>I.D.</u>	Mean	Median	98th Percentile				
EPNG-1	0.01	0.002	0.07				
EPNG-2A	0.04	0.02	0.2				
EPNG-2B	0.02	0.008	0.13				
EPNG-3	0.02	0.008	0.14				
EPNG-4	15	0.6	87				
OCD-1							
S-1							
S-5	0.0007	0.0003	0.004				

Table 4-3. Estimated Cumulative Cancer Risk of Drinking Groundwater from Selected Wells at the Manana -Mary Wheeler #1-E Well Site

* Carcinogenic risk estimated using procedures of Crouch et al. (1983) based on a 70-kg human consuming 2 liters of water per day for 70 years.

Note: This table revised 4/17/90.



CDS LABORATORIES P 0 BOX 2605 EURANGO, CO 81302

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El Paso Natural Gas P 0 Box 4990 Farmington, NM 87499

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(303) 247-4220 January 30, 1990 METHOD 8240

CDS ID:	2035
SAMFLE ID:	F90013 Flora Vista, NM Water Well FV-1
DATE TAKEN:	01/10/90
TIME TAKEN:	1050
DATE REC'D:	01/11/90
COC:	ïfes

Volatile Organics	Results ppm	Detection Limit
Acetone	< .100	.100
Acrolein	< .010	.010
Acrylonitrile	< .010	.010
Benzene	< .005	.005
Bromaform	(.005	.005
Bromomethane	( .010	.010
Carbon tetrachloride	< .005	.005
Chlorobenzene	( .005	.005
Chloroethane	< .010	.010
Chloroform	< .005	.005
Chloromethane	.010	.010
cis-1,3-Dichloropropene	( .005	.005
Dibromochloromethane	< .005	.005
Dichlorobromomethane	< .005	.005
Dichlorodifluoromethane	(.010	.010
Ethyl Benzene	( .005	.005
Meinylene chloride	( .005	.005
Tetrachloroethylene	(.005	.005

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1.15

Volatile Organics (Cont.)	Results ppm	Detection Limit
Toluene	( .005	.005
trans-1,2-Dichloroethylene	< .005	.005
trans-1,3-Dichloropropene	< .005	.005
frichloroethylene	< .005	.005
Trichlorofluoromethane	< .010	.010
Vinyl chloride	< .010	.010
1,1,1-Trichloroethane	< .005	.005

Dr. Joe Bowden, Director Approved by:___

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	Sample Collected By AEN	BLASLEY IN	AMAN NOULLE	Phone 5	5-599-216	52		
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	Kjeldahl Nitrogen	Magnesium	Silica	Bromo	form			
	Nitrate	Manganese	Sulfate	Bromo	dichloromethane			
	Nitrite	Mercury	Sulfite	Carbon	Tetrachioride			
-	Oit & Grease	Molybdenum	Surfactants-MBAS	Chlorot				
-	Organic Carbon	Nickel			methane			
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-	Phosphorus, Total	Salanium	BHC Isomers		chloroethene (D			
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+-	Barium	Carbon Dioxide	2.4.5-TP-Silvex		Trichioroethane			
$\vdash$	Beryllium	Chloride	2.4.5-T		Trichloroethane			
	Boron	Color			Trichioroethylen	(TCE)		
	Cadmium	Fluoride	GROUP J		methanes			
	Calcium	Iodide	Sulfides	тох				
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P. O. BOX 1492 EL PASO, TEXAS 79978 PHONE: 915-541-2600

May 16, 1990

Mr. David G. Boyer Environmental Bureau Chief New Mexico Oil Conservation Division P.O. Box 2088 State Land Office Building Santa Fe, New Mexico 87504

SUBJECT: Site Investigation/Remediation for the Manana-Mary Wheeler #1E Gas Well Site at Flora Vista, New Mexico

Dear Mr. Boyer:

Enclosed for your review are two copies of EPNG's responses to your March 30, 1990 comments on the Site Investigation/Remedial Action Plan. We are looking forward to discussing these and any other pertinent issues tomorrow in Santa Fe.

incerely yours, Kenneth E( Beagley, Manager North Region Compliance Engineering

STATE OF NEW MEXICO



ENERGY, MINERALS AND NATURAL RESOURCES DEPARTMENT

**OIL CONSERVATION DIVISION** 

GARREY CARRUTHERS

March 30, 1989 200

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87504 (505) 827-5800

File copy

CERTIFIED MAIL RETURN RECEIPT NO. P-918-402-154

Mr. Kenneth E. Beasley, Manager North Region Compliance Engineering EL PASO NATURAL GAS COMPANY P. O. Box 1492 El Paso, Texas 79978

RE: Site Investigation Report/Remedial Action Plan for the Manana-Mary Wheeler #1E Gas Well Site, Flora Vista, New Mexico

Dear Mr. Beasley:

Attached are my comments on the above report. After review by you and your consultant, K.W. Brown and Associates, I would like to schedule a conference call or meeting in mid to late April to discuss my comments, your reply, and a schedule to commence work in late spring/early summer. The goal should be to accomplish physical removal of soils and complete cleanup by the end of the summer.

Two other issues, not mentioned previously in correspondence, but which I believe are critical to future protection of ground water, are the placement of proper well seals and locks on Flora Vista's water wells, and replacement of Manana's produced water tank. The issue of the unused but open, unsealed water wells has been mentioned numerous times, including as recommendations in my 1986 reports. Wells S-1, S-4, and S-5 and any other unused production wells should be sealed but with locking caps so that water level and sampling can be conducted as necessary. Exploratory wells SX-1 and SX-2 should have their caps examined and welded shut, if not already done so.

Secondly, Manana's fiber-glass tank sits partially below grade and has a buckled side. It should be replaced by an above ground metal stock tank, probably at the time of excavation. Without these additional precautions the cleanup could be undone in the event of vandalism or accident and charges could be made that EPNG was negligent in cleanup. OCD will contact Manana regarding the produced water tank, but EPNG may wish to assume responsibility for sealing unused wells to avoid any further future liability.

Along with my comments, I have also enclosed a copy of a map showing the Flora Vista numbering of water wells. Please send a copy of EPNG's December's Flora Vista analyses; I ended up with results of some other EPNG sampling.

Mr. Kenneth E. Beasl March 30, 1990 Page -2-

On a personal note, with the rehiring of Bill Olson, we're finally digging ourselves out of the remedial action hole we were in (Pardon the expression!). We have at least a dozen projects needing attention similar to this one. My apologies for taking so long to get these comments out.

5 . . .

Sincerely,

une

David G. Boyer, Hydrogeologist Environmental Bureau Chief

DGB/sl

Enclosures

cc: W. J. LeMay, Director OCD OCD Aztec Office J. Eichelmann, Burlington Northern S. Johnson, KW Brown & Associates

#### COMMENTS SITE INVESTIGATION/REMED ACTION PLAN AT MANANA MARY WHEELER #1E GAS WELL FLORA VISTA, NEW MEXICO

#### Prepared by David G. Boyer, Hydrogeologist March, 1990

#### I. Comments on Site Investigation Report

1. p.41, Table 3-4. The values for Benzo(b) fluoranthene and Indeno (1, 2, 3-cd) pyrene should be shown as 2 and 6 ug/l (instead of 0.002 and 0.006 ug/l). The correct values for WQCC standards for Ethylbenzene, Toluene and Xylenes (total) are 750, 750 and 620 mg/l, respectively.

2. p.45. A discussion of oil and grease concentrations found in soil samples taken from trenches 6 and 7 is presented. Additional investigation is necessary to verify the actual presence of a "slug" of oil in this area, especially if analytical techniques may have contributed to false positive readings (ref. "A Comparison of Methods of Measuring Total Petroleum Hydrocarbons in Soil," 1989 NWWA/API Proceedings of the Conference on Petroleum Hydrocarbons and Organic Chemicals in Ground Water) present a plan to perform the necessary verification.

3. p. 48. Please provide a copy of the referenced 1983 paper on risk assessment by Crouch ("The Risk of Drinking Water").

4. p.48, Comment on Section 4.0, Risk Assessment. As stated in this section, risk assessment techniques are used to estimate chronic toxic potential from contaminants dissolved in ground water. In addition to dissolved contaminants we have an oil phase that imparts visual and hydrocarbon odor characteristics to the water. The assessment technique would seem more appropriate if used after the cleanup to evaluate the risk of any remaining dissolved constituents.

5. p.52, Table 4.2. The contaminant concentrations listed in Table 4.2 are 1000 times greater than actually observed (eg. Benzene concentration in EPNG-2A is 8.1 ppb not 8,100 ppb). I suspect these are transcription errors. Please verify that the risk calculations shown are for the actual concentrations and not those shown in the table.

6. p. 57. In New Mexico, buried reserve pits commonly contain waste debris from drilling operations. It would not surprise me to find concrete fragments and scrap pipe in the pit.

- II. Remedial Action Plan
  - A. Section 6.1.1. Contaminated Soil

1. p.59. Describe further the procedure for visual inspection verification of removal of contaminated soil, especially since contamination extends to sands and gravels beneath the water table. Is temporary dewatering proposed as part of the soil removal process?

2. p.59. I noval of hydrocarbon stained s should extend to the vicinity of well S-1 since staining was noted in that location during the 1987 excavation (See the discussion on page 2 of Dr. Blair's 1987 report).

3. p.59. What procedures are proposed to determine the extent and the necessity for soil removal under lease structures (eg. oil/water separator, produced water tank, meter house).

4. p.59. Describe the type and source of the material that will be deposited in the excavation to replace the contaminated soil.

5. p.60. If the contaminated soil is stored on site prior to offsite removal, it must be stored such that oil and water drainage is intercepted prior to discharge to the bare ground. Likewise, saturated soil cannot be loaded directly into trucks such that oily water will discharge onsite or enroute to the disposal location.

B. Section 6.1.2. Contaminated Groundwater

1. p.60. Explain the impacts to the area adjacent to well S-5. Is the reference to the oil and grease analyses or the low levels of dissolved hydrocarbons found during the 1989 site investigation sampling?

2. p.60. Additional sampling of S-5 is necessary to verify if benzene and toluene are present at trace levels as shown in the 1989 site sampling. (1986 sampling also detected these at about the same levels).

3. p.60. During excavation, water will be made turbid with the disturbance and residual oil may be freed from the soil and float on top of the water. Equipment should be available on site during excavation to skim and remove floating oil.

C. Section 6.3. Monitoring Soil

1. p.61. Soil. Explain how representative samples are to be collected from the bottom of the excavation if the excavation is not dewatered.

2. p.61. Groundwater. The December, 1989 EID/OCD sampling of well S-1 detected acetone. Pre-cleanup sampling should verify the presence or absence of that contaminant.

3. p.62. In addition to replacement monitor wells, I would like to discuss digging a temporary trench just downgradient of the southwestern-most area of excavation to serve as an observation trench. It would provide visual verification (absence of product or sheens) that no floating hydrocarbons escaped excavation. More importantly, with ground water flow at 2 to 3 feet per day, it would allow natural volitization of any remaining dissolved hydrocarbons. With a minimum of agitation and circulation the trench could act as a final treatment "air stripper" and aeration system. It could be kept in use for a short period of time (90-120 days) after excavation and soil replacement has been completed to allow flushing of the replacement soil. 4. p.62. Sigest a schedule for short and the grant pumping and testing of well S-1 after completion of the excavation.

5. p.62. Provide the number and location of the proposed monitoring wells.

6. p.62. Analytical Parameters. In addition to the proposed sampling constituents, analyses of water should include major cations and anions. Soil analyses should include chromium and those PAH's detected and shown in Table 3.5 (p. 43).

7. p.63. Sampling Frequency. Monitor wells shall be removed, or properly plugged and abandoned at the conclusion of their use. This will be no later than at the time of decommissioning of the gas well, or an earlier time as approved by OCD after consultation with other involved parties (i.e. Flora Vista Water Users, EID, EPNG, Manana).

D. Additional Comments

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Provide a proposed timetable for the start and completion of the remedial action, including the time expected to be required for the excavation phase.

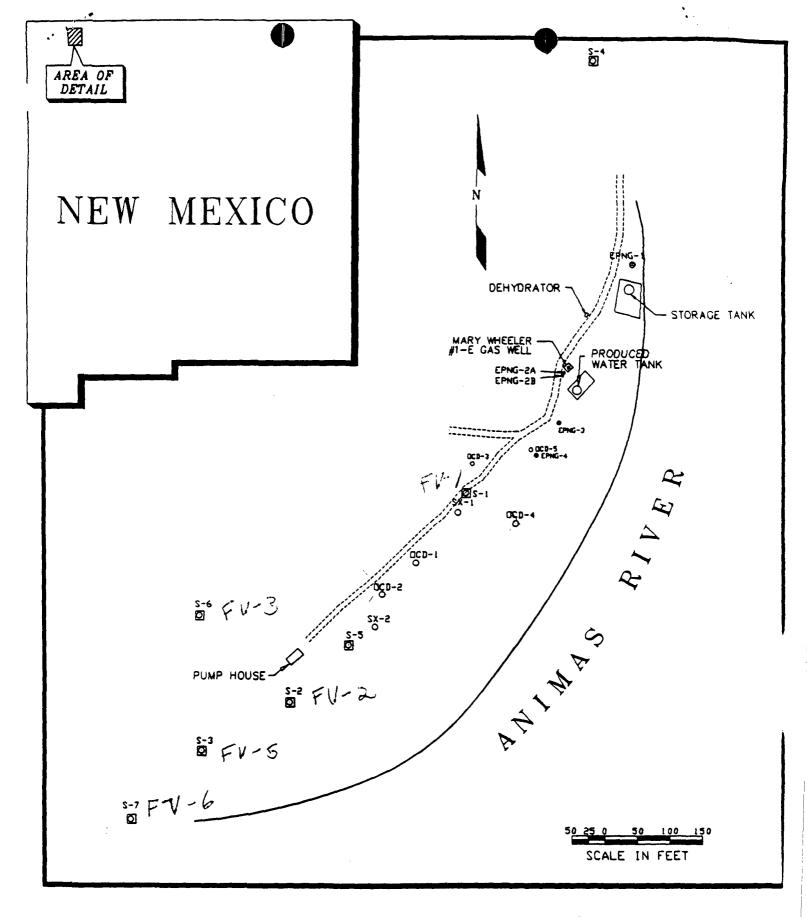
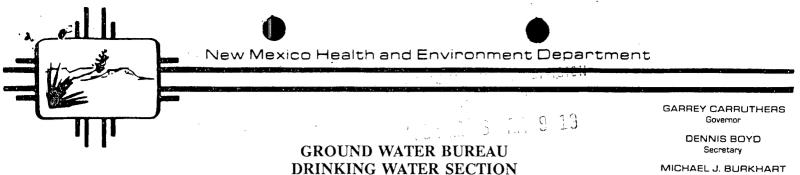


FIGURE 2-1. MAP SHOWING LOCATION OF THE MAÑANA - MARY WHEELER #1-E WELL SITE.



MICHAEL J. BURKHART Deputy Secretary

RICHARD MITZELFELT Director

March 2, 1990

Mr. Bert Barnes, President Flora Vista Water Users Association Box 171 Flora Vista, NM 87415

#### **SUMMARY**

The sheen observed on the water surface of the Flora Vista public water supply wells in December, 1989 was not caused by petroleum contamination. However, the entire well field remains highly vulnerable to documented contamination which remains in the soil near the El Paso Natural Gas facility adjacent to the well field. At least once a year each of the public water supply wells should be sampled for volatile organic compounds as a precautionary measure until the required site cleanup is complete.

Dear Mr. Barnes;

This letter is to transmit the results of the sampling that was performed by EID and OCD personnel on the Flora Vista WUA wells on December 27 & 28, 1989 and to summarize EID's conclusions and recommendations. EID and OCD personnel investigated the reported contamination of the well field at the request of the Flora Vista WUA.

A visible iridescent sheen was observed on the surface of wells 2, 5, and 6. All wells had been shut off by the WUA on December 21, 1989 when the sheen was observed during a routine pump replacement at well No. 5. There was some speculation that the sheen may have been caused by contamination from a nearby natural gas well owned by El Paso Natural Gas. This well and related facilities have been shown to be the source of contamination which required the Association to abandon well No. 1 in 1983.

Samples were taken from well Nos. 1, 2, 5, and 6. All wells except No. 1 were sampled on December 27, 1989 under static conditions with the well pumps off and again on December 28, 1989 after purging the wells by pumping them for at least one hour. All pumped waters during this test were discharged to waste on site rather than pumping to distribution.

The samples were analyzed by the State Scientific Laboratory for:

Page 2

Volatile organic compounds Base/Neutral Extractibles Heavy Metals Secondary Standards

Copies of the laboratory results are attached. The results indicate that no detectible organics were present in wells 2, 5 & 6 after they were purged. Before purging, well No. 2 contained 8.8 ug/l of methylene chloride. There is no drinking water MCL for methylene chloride but the WQCC regulations set a groundwater MCL value of 100 ug/l. Methylene chloride is occasionally found to be a laboratory contaminant.

Well No. 1, which has not been in service since petroleum contamination was observed in 1983, did contain a very high concentration, 8,800 ug/l, of acetone. We have no ready explanation for this result which the SLD confirmed by use of mass spectroscopy. Acetone is not commonly associated with petroleum contamination. It is a common laboratory solvent and could possibly be a lab contaminant. Well No. 1 did not have a proper seal. The casing was covered with a loose piece of sheet steel and a rock.

Heavy metal concentrations were all below detection limits. Secondary (aesthetic) standards for iron and manganese were exceeded in well Nos. 2 & 6. The Water Users Association is aware of the problems with iron and manganese and uses a sequestering agent to control the resulting taste and staining problems.

#### CONCLUSIONS

1. The observed sheen on wells No. 2, 5 & 6 was not due to petroleum contamination. It may have been due to iron bacteria or some other source.

2. No violations of the Maximum Contaminant Levels for chemicals listed in the Regulations Governing Water Supplies was noted for wells No. 2, 5 & 6. Well No. 1 contained a high concentration of acetone.

3. The Flora Vista well field is obviously very vulnerable to contamination from the gas well and related facilities as well-as the existing contamination in soil-around these facilities.

#### RECOMMENDATIONS

1. The remedial cleanup of the EPNG gas well site which has been required by OCD should take place as soon as possible to remove contamination resulting from operations at that site.

2. Flora Vista WUA should continue to refrain from using well No. 1 until remedial cleanup is complete and EID and OCD determine that no significant threat of contamination remains. A lockable cap should be provided for the well head to

Page 3

prevent accidental or deliberate contamination of well No. 1. The cap must be able to be removed for sampling, though.

3. Well No. 1 should be re-sampled to confirm or deny the apparent acetone contamination.

4. Since the entire well field is highly vulnerable to contamination from the EPNG site, VOC monitoring frequencies should be increased beyond the minimum requirements for compliance with the regulations. EID strongly advises that all the wells be sampled for volatile organic compounds at least once a year. Mr. Oscar Simpson of our Albuquerque office, at 841-6570, can assist with coordination of sample submittals with the state Scientific Laboratory Division.

stuart P. Castle

Stuart P. Castle, Chief Groundwater Bureau

#### SPC/NW

cc: Jon Thompson, Deputy Director Dan Vigil, Acting Dist. I Manager Dave Tomko, EID Farmington Stuart Castle, GWB Chief Oscar Simpson, DWS Rep. Richard Cheney (w/encl.) Kenneth Beasley, El Paso Natural Gas Co. (w/encl) David Boyer, OCD (w/encl.)

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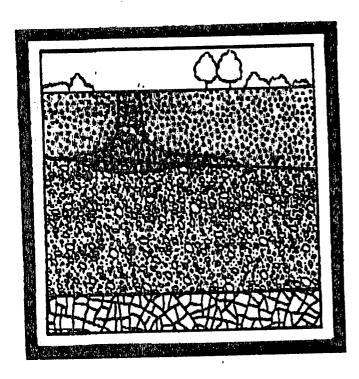
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	From: BOBBY SPEAKE	
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	Time: 16:45 C3T	
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	Project Number: 63712	
	Number of Pages: 1 cover sheet + pages of text	
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	Fax Operator: EDEN SPEAKE	
	K.W. Brown & Associates, Inc. 6 Graham Road College Station, TX 77840	
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# THE PROCEEDINGS OF THE

# Petroleum Hydrocarbons and Organic Chemicals in Ground Water: Prevention, Detection and Restoration



# A Conference and Exposition

# November 15–17, 1989 The Westin Galleria Houston, Texas

Presented by The Association of Ground Water Scientists and Engineers. division of NWWA; and The American Petroleum institute

#### A COMPARISON OF METHODS

#### FOR MEASURING TOTAL PETROLEUM HYDROCARBONE IN SOIL

by Nan Thomey, David Bratberg Applied Earth Sciences, Inc. Houston, Texas and Cathe Kaliss Star Enterprise Irving, Texas

#### ABSTRACT

Two analytical methods are used to measure total petroleum hydrocarbon concentration in soil samples. One method is a modification of EPA Method 418.1 which utilises a freon extraction followed by infrared spectroscopy. The second method uses the same extraction method, but is followed by gas chromatography/flame ionization detector analysis.

Soil samples collected during investigations of petroleum product releases from underground storage tanks were submitted to the laboratory and analysed by both methods. Specific controls were specified for the analyses to eliminate sampling bias from the results.

The data obtained by the two methods were compared and different results were observed. In certain soil matrices, a positive interference was documented in the results from Method 418.1. This positive interference could lead to unwarranted remediation activities if the results are used as cleanup guidelines. The results of the study and potential explanation for the differences are presented.



#### INTRODUCTION

During the course of subsurface investigations of petroleum product releases from underground storage tanks, soil samples may be collected in order to document the veritcal and horizontal extent of subsurface contamination. In addition, soil samples are collected during remediation activities to measure the effectiveness of the technique utilized and to determine the proper disposal location for excavated material.

Current regulations and technical guidelines in many states require that these soil samples be enalyzed for the common gasoline constituents benzene, toluene, ethylbenzene and total xylenes (BTEX). These analyses are performed using EPA Method 8020 as described in SW-846, Test Methods for Evaluating Solid Wasta-Physical/Chemical Methods. This gas chromatography technique utilizing a photoionization detector. In addition, many states require measurement of total petroleum hydrocarbons (TPH) to total petroleum hydrocarbons (TPH) to determine the presence of non-gasoline hydrocarbons such as diesel or kerosens, and extremely weathered gasoline products. Currently, there is not an EPA approved method. TPH analysis in soil. The method which has been for commonly used and referenced is EPA Method 418.1, described in Methods for Chemical Analysis of Water and Wastes. This method is for measurement of TPH by infrared spectroscopy (IR) in water and wastewater. The extraction procedure must be modified for soil analysis. The state of California, among others, specified a gas chromatography (GC) with flame ionization detection (FID) to determine TPH concentrations in both soil and water. This method is designed to provide product identification as well as quantification. The some called "California Method" is not approved by the state of This is because a headspace injection technique is Texas. used for gasoline analysis. The state of Texas does not accept this injection method because the EPA considers this technique suitable for screening purposes only.

assessments, During environmental anomolies Were noticed in the 418.1 TPH data. The data obtained were inconsistent with known site information. One type of situation involved soil samples which were collected from service station sites which had never dispensed any products except gasoline. Analytical results indicated that the samples contained no measurable BTEX constituents and the soils exhibited no hydrocarbon odors. However, TPH results indicated the presence of 100-200 mg/kg of petroleum hydrocarbons. These results were inconsistent with the composition of gasoline. A second type of situation involved samples which were collected in conjunction with remodeling activities at sites with no suspected or known

product releases. Soil borings were performed and several intervals were sampled and analyzed for sach boring. BTEX concentrations were below the analytical detection limits. The TPH results ranged from 100-250 mg/kg for all of the samples which were collected.

Based upon this information, a potential problem was suspected with the 418.1 TPH analyses. The BTEX chromatograms were examined and in most cases, there was no evidence of the presence of petroleum hydrocarbons. A positive interference in the IR method was suspected. The IR methodology was focused upon because the method used for soil analyses is actually a technique developed for wator analyses. The method did not appear to be directly applicable to the analysis of soils. Through discussions with the Texas Water Commission and other states, a method for TPH analysis by Gas Chromatograph (GC) utilizing a flame ionization detector (FID) was proposed and was approved by the State for use in the underground storage tank program.

To satisfy our own curiosity, we decided to conduct a method comparison study. We submitted a number of soil samples for TPH analysis. The samples were analyzed by IR and GC/FID. The data obtained by the two methods are presented and compared below.

"The state of Texas has approved the use of a GC/FID analysis for TPH. The method approved specifies the extraction solvent, type of column, and injection technique to be used.

#### DESCRIPTION OF NETHODS

Approximately 60 samples were submitted for analysis during the course of the study. All soils were classified by the same hydrogeologist to assure consistency.

Two methods were used for the analysis of the samples. One method was the EPA modified 418.1 procedure for infrared spectroscopy. The second method was a GC/FID capillary column analysis, based upon ASTM 3328. This is the GC method which is approved by the State of Texas. ATSM 3328 is a procedure for the comparison of waterborne oils. There is no mechanism in this method for quantification.

Both methods utilize the same extraction procedure. A known portion of the sample was dried with anhydrous sodium sulfate, and ground to form a free-flowing powder. The dried samples were then extracted by sonication. The solvent which was used was fluorocarbon-113. The extract was treated with 60-200 mesh silica gel. The silica gel treatment is to remove polar hydrocarbons which include

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### animal fats and other non-petroleum hydrocarbons.

The treated extract was then analysed by Infrared spectroscopy and GC/FID. Both analyses were performed on the same extract. This eliminated any bias from the sampling and extraction processes. The IR analysis measures carbon-hydrogen bonds. The GC/FID analysis utilizes flame ionisation to measure hydrocarbons. The IR analysis cannot identify specific hydrocarbons. GC/FID is a common analytical technique which is frequently used to identify and quantify specific hydrocarbons.

In the IR method, TPH is calculated by comparing the IR absorbance of the samples at about 2950  $cm^{-1}$ , to the response of a reference oil standard. The reference oil standard utilized in 418.1 is composed of n-hexadecane (C16), isoctane (C8), and chlorobenzene (C6). In the GC/FID method, the total area response of the sample is compared to the total area response of a specific fuel or mixture of fuels such as gasoline or diesel. These standards are in the range of approximately C-5 to C-22. Copies of standard chromatograms for gasoline and diesel are shown in Figure 1. A known 418.1 reference oil standard was analyzed by GC and the result calculated based upon the GC standard. A known GC gasoline standard was analyzed by IR and the result calculated the total area to based of utilizing different standards.

#### RESULTS.

The analytical results are presented by soil type in Table 1. Significantly different results by the two methods were observed. The IR results were higher than the GC/FID results for certain types of soil. The predominant samples axhibiting this effect were the soils whose major constituents were weathered limestone, silt, or clay.

All of the samples composed predominantly of weathered limestone had IR results which were higher than the GC/FID results. Some of these differences were quite significant. For example, on one sample which was submitted, the results were 760 and 67 mg/kg by IR and GC/FID, respectively. In a second case, the results were 3,100 and 1,000 mg/kg by IR. and GC/FID, respectively.

### TABLE 1

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### Summary of Results TPH Method Comparison

Soil type	TPH IR	TPC/GC
Weathered limestone, sandy clay	760	67
	31	ND
	3,100	1,000
Sand	130	230
sand	120	170
Sand	150	150
Sand	100	130
Sand, medium	ND	ND
Sand, medium	ND	ND
Sand, medium	ND	ND
Sand, some clay	ND	ND
Sand, medium	ND	ND
Sand, medium	ND	ND
Sand, clay mixture	ND	ND
Silty sand	30	ND
Silty sand	ND	ND
Silty sand	ND.	ND
Clayey sand	ND	ND
Silty clay	16	ND
Silty clay	ND	ND
Silty clay	ND	ND ND
Silty clay	ND	ND
Silty clay, weathered limestone	ND	ND "
Silty clay, silty clay	366	<25
Sandy clay, dry	ND	םא
Clay, brown, dry	ND	ND
Clay, yellow brown, dry	'ND	ND
Clay, weathered linestone	ND	ND
Clay, yellow brown, dense, dry	250	ND
Clay, dark gray, dense, stiff	ND	ND
Clay, dark gray, stiff	120	ND
Clay, dark gray, stiff	ND	ND
Clay, gray, stiff	160	ND
Clay, some silt, dense	ND	ND

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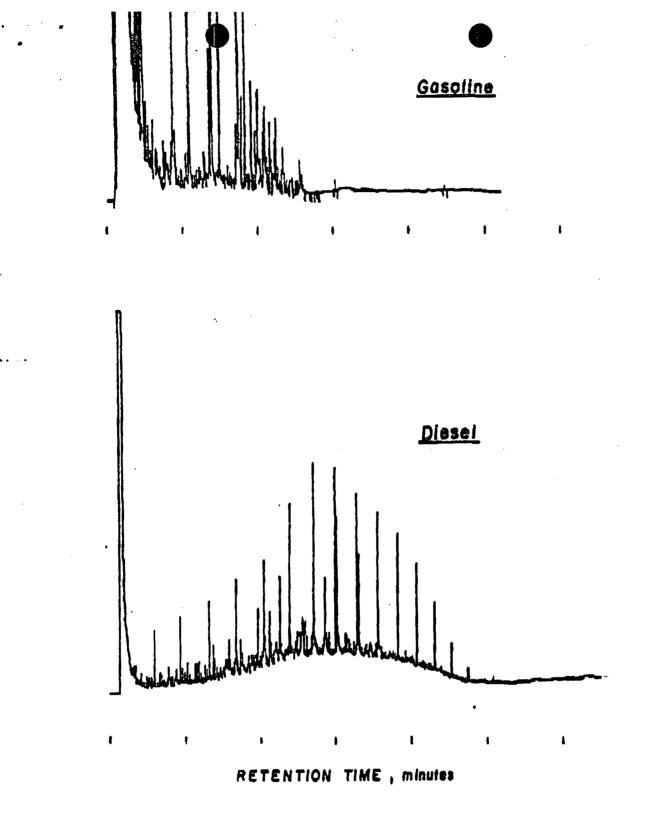
## TABLE 1 (Continued)

## Summary of Results TPH Method Comparison

### SOIL TYPE

COLA LIFA	<u>TPH IR</u>	TPH/GC
Clay, light brown, manganese concr Clay, dark grav		
Clay, dark gray		ND
Clay, light brown	30	ND
Clay, black	ND	ND
Clay, dark grav. some grave	ND	ND
	323	87
Clay, light brown	ND	63
Clay, dark brown, stiff	ND	ND
	ND	ND
Clayey silt		
Clayey silt	796	70
Clayey silt	סא	ND
Clayey silt	ND	ND
	ND	ND
Clayey silt, silty clay		
Silt, weathered limestone, wet	ND	ND
Silt, gray, loose	ND	ND
Silt, medium sand	90	ND
	ND	
		ND
Sandy, silt, wat		
	106	ND
		01.U
Cl vyey gravel		
	9,700	350
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#### CONCLUSION



Based upon the results of this study, it is apparent that EPA Method 418.1 is not an appropriate technique for measuring TPH concentration in certain types of soils. These types of soils can be categorized as weathered limestons, clays, and silts. One of the more frequent arguments for using this method is that it is a relatively inexpensive test. We believe this study demonstrates how thousands of dollars can be spent on unnecessary remediation and disposal costs because of actions taken on the basis of a laboratory analysis costing less than fifty dollars. It is our opinion that the GC/FID technique provides a more adequate representation of the degree of hydrocarbon contamination in soil.

If EPA modified Method 418.1 is used for TPH analysis in soils, positive results should be confirmed through GC/ FID analysis. Otherwise, the parameter of TPH should not be used to establish cleanup guidelines or to assign waste classifications.



El Paso

Natural Gas Company



P. O. BOX 1492 EL PASO, TEXAS 79978 PHONE: 915-541-2600

August 28, 1989

Mr. David G. Boyer Environmental Bureau Chief New Mexico Oil Conservation Division 310 Old Santa Fe Trail Suite 206 Santa Fe, New Mexico 87504

Subject: Investigation Report of the Manana - Mary Wheeler Site

Dear David:

Enclosed are two copies of the report by K.W. Brown and Associates on the site near Flora Vista. After you have had a chance to review it, we will be happy to get together with you and discuss the remediation. Henry Van or I will be happy to answer any questions you might have on the report in the interim.

RECEIVED

OCT - 2 1989

OIL CONSERVATION DIV. SANTA FE Sincerely yours,

Kenneth E) Beasley Manager of Compliance Engineering, North Region STATE OF NEW MEXICO

#### ENERGY, MINERALS AND NATURAL RESOURCES DEPARTMENT

**OIL CONSERVATION DIVISION** 

GARREY CARRUTHERS

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87504 (505) 827-5800

May 17, 1989

Mr. Kenneth Beasley Manager, North Region Compliance Engineering EL PASO NATURAL GAS COMPANY P. O. Box 1492 El Paso, Texas 79978

RE: Flora Vista - Manana Mary Wheeler #1E Investigation

Dear Mr. Beasley:

The Oil Conservation Division (OCD) has received the revised investigation plan submitted to us with your April 10, 1989, letter. Comments were made by telephone, and changes were incorporated in the copy provided to us by the Farmington EPNG Office. The final copy of Table 2 was telefaxed to us on May 8, 1989.

EPNG's revised investigation plan, with the agreed-to changes, is hereby approved. I understand that a contractor will be chosen shortly and that Task 1, Review of Existing Information, should start within 30-days. Please notify us when the contractor is selected and of the anticipated starting date.

If you need further information, please contact me at (505) 827-5812. Please note that we have resumed use of our old P. O. box number, and correspondence should be directed to that address.

Sincerely,

David G. Boyer, Hydrogéologist Environmental Bureau Chief

DGB/sl

CC: OCD Aztec Office



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P. O. BOX 1492 EL PASO, TEXAS 79978 PHONE: 915-541-2600

May 8, 1989

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OIL CONSERVATION DIV. SANTA FE

Mr. David G. Boyer Environmental Bureau Chief Energy and Minerals Department Oil Conservation Division 310 Old Santa Fe Trail Santa Fe, New Mexico 87504

Subject: Flora Vista - Manana Mary Wheeler #1E Investigation

Dear Mr. Boyer:

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Enclosed is the revised investigation plan for the Flora Vista site. I have incorporated your final comments which were received telephonically. As discussed today, this proposal now addresses all of our mutual concerns and we will begin implementation. Thank you again for your input. We will keep you posted on planned activities.

Sincerely Yours,

Kenneth E. Beasley Manager, North Region Compliance Engineering bc: S.D. Aragon W.H. Healy, Jr. (w/o att) D.W. Hill (w/o att) A.N. Pundari H.A. Shaffer (w/o att) L.R. Tarver (w/o att) H. Van file 24322

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A PROPOSAL

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FOR

# INVESTIGATING THE MANANA - MARY WHEELER #1E

# WELL SITE

NEAR FLORA VISTA, NEW MEXICO

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4.0	HYDROLOGIC INVESTIGATION
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## INTRODUCTION

El Paso Natural Gas Company (EPNG) has prepared this proposal for a soil/groundwater investigation of the Manana - Mary Wheeler #1E gas well site near Flora Vista. New Mexico. The proposal describes tasks involved in the site investigation including a review of existing information concerning the site, further characterization and topographic survey of the area, and interpretation of the collective observations and results. These efforts will culminate in a risk assessment of possible contamination from the inactive pit, and recommendations for treatment or removal of the contamination, if deemed necessary. The objective is to allow the continued use of existing water wells adjacent to the site.

#### 2.0 **REVIEW OF EXISTING INFORMATION**

EPNG will collect and review existing information on conditions at the site, specifically on the area between the inactive pit and nearby water wells. This review will include information in NMOCD files such as field reports, aquifer tests, water quality data and topographic details. EPNG will review this information to determine the preexisting conditions at the site for comparison with findings of the EPNG field investigation. From this comparison, conclusions will be made concerning the likelihood of migration and/or degradation of contaminants from the pit. This information will be used to aid in designing the hydrologic investigation.

#### 3.0 AREA TOPOGRAPHIC SURVEY

A topographic survey of the area will be conducted to provide relative elevations of points of interest at the site. The survey will provide a grid, with points of known elevations, for siting the location of pits for soil sampling. The data will also be used in the description the contaminant plume, if any, as well as the of piezometric surface of the immediate area. Data gener-ated during this task will be used to generate base maps which illustrate the locations of all relevant features and will establish baseline elevations for all aspects of the investigation.

#### HYDROLOGIC INVESTIGATION 4.0

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hydrogeologic investigation will be performed to Α define the hydrologic setting and aquifer characteristics, such as groundwater flow direction and veloci-Addity, transmissivity, and storage coefficient. tionally, the behavior of free oil and/or dissolved hydrocarbons, if present, will be evaluated in light of the physical characteristics of the saturated alluvial sediments. The extent of the investigation will encompass the area immediately surrounding the well site (including the reserve pit as well as other pits), and the area down gradient from the well which could potentially be affected by past operations.

The intent of the hydrologic investigation will be to illustrate, to the fullest degree possible, the lateral and vertical extent to which the site has been affected. To assess the magnitude of migration, organic constituents associated with hydrocarbons will be measured (e.g., benzene, toluene, xylene, and oil and grease) as will inorganic constituents (e.g., metals) and indicator parameters such as pH and electrical conductivity (EC).

To achieve the objectives of the hydrologic investigation, piezometers will be installed to determine the horizontal and vertical gradients, trenches will be dug to visually assess the dimensions of the plume (if any) using stained soil/sediments as evidence, and pump tests will be performed using the local water wells and piezometers to define aquifer characteristics.

#### <u>Piezometers</u>

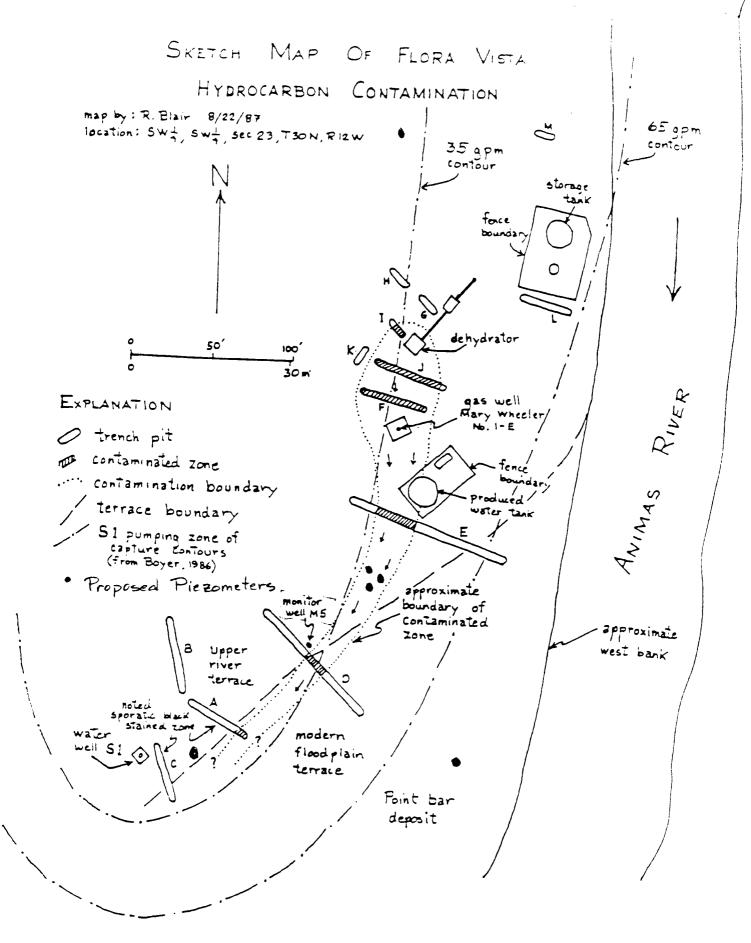
Approximately four sites will be selected for the installation of new piezometers, one of which will be installed upgradient to establish background concentrations for parameters measured at downgradient locations. Figure 1 shows possible piezometer locations superimposed on a site sketch. Of the remaining piezometer locations. one will be selected for a nested piezometer to determine the vertical gradient. Each piezometer will be inusing construction details, which are stalled tailored to the site, to allow representative groundwater samples to be collected. Figure 2 shows a typical piezometer installation guide. Before installation of piezometers, EPNG will sample piezometers previously installed by NMOCD after determining that they are suitable and cleaning them of sediment. It is expected that these piezometers will also assist in mapping the water table. Piezometers will be installed with a rotary rig if possible. A last resort will be by excavating with a backhoe to the desired depth and setting the casing.

In addition to defining the vertical gradient, the piezometer nest will aid in the determination of the vertical extent of constituents in the groundwater. In conjunction with the installation of the piezometers, soil samples may be collected for laboratory analysis.

#### Trenches

To define the extent and shape of affected soil/sediments, it is proposed that trenches be dug perpendicular to the direction of groundwater flow. The purpose of the

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	ONITORING WELL REPORT
WELL NO:	DRILLED BY:
	LOGGED BY:
SCREENED INTERVAL	: INSPECTED BY :
	TOP OF CASING ELEVATION:
	TOP OF RISER PIPE ELEVATION:
	CASING TYPE: CASING I.D.: PLATFORM ELEVATION:
WATERISSERVER	GROUND SURFACE ELEVATION:
	DEPTH OF CASING BELOW GROUND:
	BOREHOLE DIAMETER:
	RISER PIPE TYPE: RISER PIPE I.D.:
	BACKFILL TYPE:
	DEPTH AT TOP OF SEAL:
	SEAL TYPE: DEPTH AT BASE OF SEAL / TOP OF SAND PACK:
	DEPTH AT TOP OF SCREENED SECTION:
-	SCREEN TYPE:
	SCREEN OPENINGS:
	SAND PACK TYPE:
	DEPTH AT BASE OF SCREENED SECTION:
	DEPTH AT BASE OF PLUG:
	BOREHOLE DEPTH:

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trenches will be to allow visual inspection of the shallow soil profile for evidence of migrating hydrocarbons. Additionally, soil samples can be collected for the profile at points where there is evidence of affected soils. Initially the trenches would be dug at 100-foot intervals to define the lateral extent of the plume. If it is determined that the 100-foot spacing is too coarse, then additional trenches will be dug as needed to define the extent and magnitude of the plume.

In addition to excavating trenches in downgradient locations, one trench or pit will be situated in an upgradient position to allow the collection of background soils samples. Where hydrocarbons are suspected, water sampling of the trenches will be conducted.

## Pump Tests

In order to evaluate the potential impact oil field constituents could have on the local water wells, it will be necessary to determine the aquifer characteristics. To this end, if there is a need to expand on data from previous pump tests, it is proposed that the nearest water well, S-1, be pumped and the piezometers (and other wells) be monitored. At this time it is believed that only the S-5 well will need to be pumped to monitor for contaminants. This would be a short duration test, possibly 15 to 30 minutes. The results of the pump tests, along with the soils data, will numerically define the potential for migration of the oil field constituents prior to cleanup activities as well as subsequent to remedial efforts.

#### 5.0 COLLECTION AND ANALYSIS OF SOIL/GROUNDWATER SAMPLES

Soil samples will be collected during the excavation of the trenches described in Section 2. Sample locations within the trenches will be based on visual evidence and the need to define the lateral as well as the vertical extent of any plume which may be detected. Collection of will be from the face of the trenches the samples using stainless steel utensils. Samples will be collected from the former pit areas, including the reserve pit, to determine the potential for leaching and migration of wastes. All soil samples will be placed in glass jars and preserved in the field using ice to maintain the temperature of the samples at or near  $4^{\circ}$  C.

Groundwater samples will be collected for each of the piezometers installed during the investigation, from the piezometers currently present at the site, and from the local water wells. Sample collection will be accomplished using either dedicated PVC bailers, a Teflon bailer(s), dedicated pump, or a bladder sample pump. If dedicated equipment is not employed, then EPA-approved

decontamination procedures will be implemented between samples. Once collected, all groundwater samples will be filtered using 0.45 micron membranes (for metals only), placed in appropriate containers, preserved as needed, and stored in the field on ice at or near  $4^\circ$  C.

Both field analysis and laboratory analysis will be performed as samples are collected. Analysis of groundwater and soil samples will be conducted according to methods and procedures presented in SW-846, Third Edition (EPA, 1986) or an equivalent approved procedure. Parameters selected for groundwater analysis, as it pertains to groundwater standards, will be selected from Section 3-301 of the WQCC Regulations. The selection of specific parameters from Section 3-301 will be based on discussions between OCD and EPNG. Soil and groundwater samples will generally be analyzed for the parameters listed in Table 1. Soil and groundwater samples collected for screening purposes may be analyzed for selected constituents on the list.

Indicators	Organics	Inor	ganics
pH EC TDS Oil and Grease	Benzene Toluene Xylenes Ethylbenzene Chlorinated, Ali- phatic hydrocarbons Glycol PNA's	Barium Calcium Magnesium Sodium Potassium Carbonate/ Bicarbonate	Chlorides Sulfur Compounds Selenium Cadmium Chromium

Table 1. Analytical Parameters for Soil and Groundwater

# 6.0 DATA INTERPRETATION AND RISK ASSESSMENT

The information gathered from previous investigations and the hydrologic investigation will be collectively reviewed to evaluate the existence, spatial distribution, and possible effects of contamination from the inactive pit and surrounding area. Utilizing these data, a risk assessment will be conducted to evaluate possible contamination to the nearby water wells and surrounding environment. The assessment will be based upon a comparison of the collective data to establish health and environmental criteria and potential for future migration.

### 7.0 REMEDIAL ACTION PLAN

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The purpose of remedial action is to leave the site such that the existing Flora Vista water wells can be operated without exceeding current WQCC standards. The plan will contain details on the methodology to demonstrate that current USEPA drinking water standards are not violated or that no undesirable odors exist due to dissolved, emulsified and free-floating petroleum or other organic and inorganic contaminants. Further, the plan will demonstrate that the vadose zone does not contain materials such as drilling muds, inorganic salts, heavy metals or hydrocarbons in quantities sufficient to recontaminate ground water as the result of seasonal rises in water table, drainage, recharge or similar events. The remediation will consider all or portions of the reserve pit area depending on the results of the site assessment. Table 2 is a time line showing the activities leading to the completion of the remedial action plan. A proposed schedule for completion of remedial activities will be submitted with the plan.

Given the relatively small size of the inactive dehydration pit, it is suggested that the contaminated soil from the pit be removed as part of remediation. Since the pit is considered to be the major source of potential contamination at the site removal of the source is considered a cost-effective measure. In the extreme case where risk assessment suggests a hazard to water quality, total removal of all affected soils downgradient from the gas well area is proposed as a worst case remediation plan. The removal of this material will prevent future leaching of constituents to the groundwater.

It is anticipated that removal can be accomplished by excavation with a backhoe and a dump truck. The proper disposal method and disposal site will be defined prior to initiation of this task. Once the affected soils/sediments have been excavated, the area will be recontoured using clean borrow soil.

Weeks 1-2     Weeks 3-4     Weeks 5-8     Weeks 8-11     Weeks 11-13       Review of Existing Information Task 2     Beview of Existing Information     Weeks 5-8     Weeks 8-11     Weeks 11-13       Review of Task 2     Initiation & Completion of Task 3-5     Sample Analysis**     Meeks 5     Berlow       Task 2     Initiation & Completion of Task 3-5     Sample Analysis**     Meeks 6     Meeks 11-13       Task 3-5     Sample Analysis**     Data Inter- pretation     Preport Report     Preport       Task 6     Data Inter- pretation, Task 6     Data Inter- pretation, Risk Assessment, Report     Parendial Action	Table 2. Tim	e-line for Co	ompletion of	Table 2. Time-line for Completion of Activities at the Manana Well Site*	he Manana	Well Site*.
Initiation & Initiation & Completion of Field Work <i>Tasks 3–5</i> Sample Analysis** <i>Task 5</i> & Begin Draft Report Preparation <i>Task 6</i> Data Inter- pretation, Risk Assessment, & Finalize Draft Report <i>Task 6</i> Data Inter- pretation,	Weeks 1–2	Weeks 3–4	Weeks 5–8	Weeks 8-11	Weeks 11–13	Weeks 13–16
Sample Sample Analysis** <i>Task 5</i> & Begin Draft Report <i>Preparation</i> <i>Task 6</i> Data Inter- pretation, Risk Assessment, & Finalize Draft Report <i>Task 6</i>	Review of Existing Information Task 2	-				
Sample Analysis** <i>Task 5</i> & Begin Draft Report Preparation <i>Task 6</i> Data Inter- pretation, Risk Assessment, & Finalize Draft Report <i>Task 6</i>		Initiation & Completion of Field Work				
Data Inter- pretation, Risk Assessment, & Finalize Draft Report <i>Task 6</i>		Tasks 3–5	Sample			
Data Inter- pretation, Risk Assessment, & Finalize Draft Report <i>Task 6</i>			Analysis** Task 5			
Data Inter pretation, Risk Assessment, & Finalize Draft Report <i>Task 6</i>			& Begin Draft Report			
Data Inter- pretation, Risk Assessment, & Finalize Draft Report <i>Task 6</i>			Preparation Task 6			
				Data Inter- pretation,		
				Risk Assessment, & Finalize Draft		
				Report Task 6		
Plan Task 7					Remedial Action	
Task 7					Plan	
					Task 7	
						Final Report
						Preparation

 Task numbers correlate with main heading numbers in proposal
 * A turn-around time of three weeks will be requested from the lab, actual analysis time may require more than 3 weeks



P. O. BOX 1492 EL PASO, TEXAS 79978 PHONE: 915-541-2600

April 10, 1989

Mr. David G. Boyer Environmental Bureau Chief Energy and Minerals Department Oil Conservation Division 310 Old Santa Fe Trail Santa Fe, New Mexico 87504

Subject: Flora Vista - Manana Mary Wheeler #1E Investigation

Dear Mr. Boyer:

Enclosed is the revised investigation plan for the Flora Vista site. I have incorporated comments from your February 14, 1989 memorandum, our February 16 meeting and your June 15, 1988 letter. We at El Paso Natural Gas feel that this proposal now addresses all of your concerns and are eager to begin implementation. In order to perform the investigation and complete any necessary remediation while weather is in our favor we will need to proceed as soon as possible. Thank you for your input and consideration in this manner.

Sincerely Yours,

Kenneth E. Beasley Manager, North Region Compliance Engineering

Noft (4/10/85)

A PROPOSAL

FOR

INVESTIGATING THE MANANA - MARY WHEELER #1E

WELL SITE

NEAR FLORA VISTA, NEW MEXICO

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5.0	COLLECTION AND ANALYSIS OF SOIL/GROUNDWATER SAMPLES 5
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# 1.0 INTRODUCTION

El Paso Natural Gas Company (EPNG) has prepared this proposal for a soil/groundwater investigation of Manana - Mary Wheeler #1E gas well site near Flora V New Mexico. The proposal describes tasks involved in a soil/groundwater investigation of the Vista, the investigation including a review of existing inforaite mation concerning the site, further characterization and topographic survey of the area, and interpretation of the collective observations and results. These efforts culminate in a risk assessment of possible contamiwill from the inactive pit, and recommendations for nation the contamination, if deemed or removal of treatment The objective is to allow the continued use of necessary. existing water wells adjacent to the site.

#### 2.0 REVIEW OF EXISTING INFORMATION

EPNG will collect and review existing information on conditions at the site, specifically on the area between the inactive pit and nearby water wells. This review will include information in NMOCD files such as field reports, aquifer tests, water quality data and topographic details. EPNG will review this information to determine the preexisting conditions at the site for comparison with findings of the EPNG field investigation. From this comparison, will be made concerning the likelihood of conclusions migration and/or degradation of contaminants from the pit. This information will be used to aid in designing the hydrologic investigation.

#### 3.0 AREA TOPOGRAPHIC SURVEY

A topographic survey of the area will be conducted to provide relative elevations of points of interest at the site. The survey will provide a grid, with points of known elevations, for siting the location of pits for soil sampling. The data will also be used in the description of if any, as well as the the contaminant plume, piezometric surface of the immediate area. Data generated during this task will be used to generate base maps which illustrate the locations of all relevant features and will establish baseline elevations for all aspects of the investigation.

#### 4.0 HYDROLOGIC INVESTIGATION

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hydrogeologic investigation will be performed to the hydrologic setting and aquifer characterisdefine tics, such as groundwater flow direction and velocity, transmissivity, and storage coefficient. Addithe behavior of free oil and/or dissolved tionally, if present, will be evaluated in light of hydrocarbons, the physical characteristics of the saturated alluvial sediments. The extent of the investigation will

encompass the area immediately surrounding the well site (including the reserve pit as well as other pits), and the area down gradient from the well which could potentially be affected by past operations.

The intent of the hydrologic investigation will be to illustrate, to the fullest degree possible, the lateral and vertical extent to which the site has been affected. To assess the magnitude of migration, organic constituents associated with hydrocarbons will be measured (e.g., benzene, toluene, xylene, and oil and grease) as will inorganic constituents (e.g., metals) and indicator parameters such as pH and electrical conductivity (EC).

To achieve the objectives of the hydrologic investigation, piezometers will be installed to determine the horizontal and vertical gradients, trenches will be dug to visually assess the dimensions of the plume (if any) using stained soil/sediments as evidence, and pump tests will be performed using the local water wells and piezometers to define aquifer characteristics.

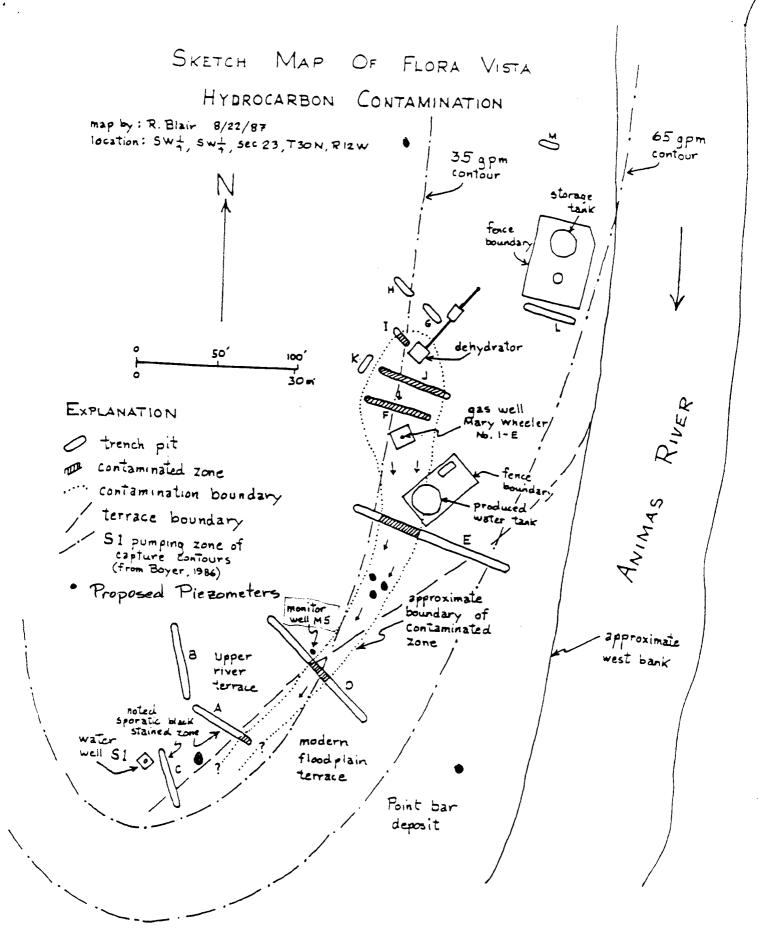
#### <u>Piezometers</u>

Approximately four sites will be selected for the installation of new piezometers, one of which will be installed upgradient to establish background concentrations for parameters measured at downgradient locations. Figure 1 shows possible piezometer locations superimposed on a site sketch. Of the remaining piezometer locations, one will be selected for a nested piezometer to deterlocations. mine the vertical gradient. Each piezometer will be installed using construction details, which are taito the site, to allow representative groundlored water samples to be collected. Figure 2 shows a typical piezometer installation guide. Before installation of piezometers, EPNG will sample piezometers previously installed by NMOCD after determining that they are suitable and cleaning them of sediment. It is expected that these piezometers will also assist in mapping the water table. Piezometers will be installed with a rotary rig if possible. A last resort will be by excavating with a backhoe to the desired depth and setting the casing.

In addition to defining the vertical gradient, the piezometer nest will aid in the determination of the vertical extent of constituents in the groundwater. In conjunction with the installation of the piezometers, soil samples may be collected for laboratory analysis.

#### **Trenches**

To define the extent and shape of affected soil/sediments, it is proposed that trenches be dug perpendicular to the direction of groundwater flow. The purpose of the



	MONIT	FORING WELL REPORT
PROJECT:		<b>V</b>
LOCATION:		
	<u></u>	
WELL NO:		DRILLED BY:
DATE COMPLETED	:	LOGGED BY:
SCREENED INTERV	AL:	
<u></u>		TOP OF CASING ELEVATION:
	-	TOP OF RISER PIPE ELEVATION:
		CASING TYPE:
		CASING L.D.:
		PLATFORM ELEVATION
VERTIFIER AND	THE REAL PROPERTY	GROUND SURFACE ELEVATION:
N		SURFACE SEAL:
V		DEPTH OF CASING BELOW GROUND:
		BOREHOLE DIAMETER
		RISER PIPE I.D.:
	◄	BACKFILL TYPE:
		DEPTH AT TOP OF SEAL:
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		DEPTH AT TOP OF SCREENED SECTION:
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	7	
		DEPTH AT BASE OF SCREENED SECTION:
	<b>−∢</b> †	DEPTH AT BASE OF PLUG:
L		BOREHOLE DEPTH:
		Figure 2

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trenches will be to allow visual inspection ΩÍ the profile for evidence of migrating shallow soil Additionally, soil samples can be collected hydrocarbons. for the profile at points where there is evidence affected soils. Initially the trenches would be dug of at 100-foot intervals to define the lateral extent of the plume. If it is determined that the 100-foot spacing then additional trenches will be dug as is too coarse, needed to define the extent and magnitude of the plume.

In addition to excavating trenches in downgradient locations, one trench or pit will be situated in an upgradient position to allow the collection of background soils samples. Where hydrocarbons are suspected, water sampling of the trenches will be conducted.

#### Pump Tests

In order to evaluate the potential impact oil field constituents could have on the local water wells, it will be necessary to determine the aquifer characteristics. this end, it is proposed that the nearest water То well be pumped and the piezometers (and other wells) he monitored. At this time it is believed that only the S-5 well will need to be pumped to monitor for contaminants. The results of the pump test, along with the soils data, will numerically define the potential for migration of the oil field constituents prior to cleanup activities as well as subsequent to remedial efforts.

## 5.0 COLLECTION AND ANALYSIS OF SOIL/GROUNDWATER SAMPLES

Soil samples will be collected during the excavation of the trenches described in Section 2. Sample locations within the trenches will be based on visual evidence and the need to define the lateral as well as the vertical extent of any plume which may be detected. Collection of the samples will be from the face of the trenches using stainless steel utensils. Samples will be collected from the former pit areas, including the reserve pit, to determine the potential for leaching and migration of wastes. All soil samples will be placed in glass jars and preserved in the field using ice to maintain the temperature of the samples at or near  $4^{\circ}$  C.

samples will be collected for each Groundwater of the piezometers installed during the investigation, from the piezometers currently present at the site, and from the local water wells. Sample collection will be accomplished using either dedicated PVC bailers, a Teflon dedicated pump, or a bladder sample pump. bailer(s), Ιf dedicated equipment is not employed, then EPA approved decontamination procedures will be implemented between samples. Constituents to be analyzed for are expected to

include some metals, PNA's, glycol, and both aliphatic and chlorinated hydrocarbons. Once collected, all groundwater samples will be filtered using 0.45 micron membranes (for metals only), placed in appropriate containers, preserved as needed, and stored in the field on ice at or near  $4^{\circ}$  C.

Both field analysis and laboratory analysis will be performed as samples are collected. Analysis of groundwater and soil samples will be conducted according to methods and procedures presented in SW-846, Third (EPA, 1986) or an equivalent approved proce-Edition dure. Parameters selected for groundwater analysis, as it pertains to the public water supply, will be selected from Section 3-301 of the WQCC Regulations. The selection of specific parameters from Section 3-301 will be on discussions between OCD and EPNG. based Soil samples and some groundwater samples collected for screening purposes will be analyzed for some, if not all of the parameters listed in Table 1.

Table 1. Analytical Parameters for Screening Purposes.

Indicatora	Organica	Inorgan	nics
рН	Benzene	Barium	Chlaride
EC TDS	Toluene Xylenes	Calcium Magnesium	Sulfidea Selenium
Oil & Grease		Sodium	Cadmium
		Carbonate	Chromium
		(and Bicarbona	ate)

#### 6.0

#### DATA INTERPRETATION AND RIEK ASSESSMENT

information gathered from previous investigations The and the hydrologic investigation will be collectively reviewed to evaluate the existence, spatial distribution, and possible effects of contamination from the inactive pit and surrounding area. Utilizing these data, a risk assessment will be conducted to evaluate possible contamination to the nearby water wells and surrounding environment. The assessment will be based upon a comparison of the collective data to establish health and environmental criteria and potential for future migration.

# 7.0 REMEDIAL ACTION PLAN

The purpose of remedial action is to leave the site such that the existing Flora Vista water wells can be operated without exceeding current WOCC standards. The plan will contain details on the methodology to demonstrate that current USEPA drinking water standards are not violated or undesirable odors exist due to dissolved, emulsified and free-floating petroleum or other organic and inorganic contaminants. Further, the plan will demonstrate that the vadose zone does not contain materials such as drilling muds, inorganic salts, heavy metals or hydrocarbons in quantities sufficient to recontaminate ground water as the result of seasonal rises in water table, drainage, recharge or similar events. The remediation will consider all or portions of the reserve pit area depending on the results of the site assessment. Table 2 is a time line showing the activities leading to the completion of the remedial action plan.

Given the relatively small size of the inactive dehydration pit, it is suggested that the contaminated soil from the pit be removed. Since the pit is considered to be the major source of potential contamination at the site removal of the source is considered a cost-effective measure. In the extreme case where risk assessment suggests a hazard to water quality, total removal of all affected soils downgradient from the gas well area is proposed as a worst case remediation plan. The removal of this material will prevent future leaching of constituents to the groundwater.

It is anticipated that removal can be accomplished by excavation with a backhoe and a dump truck. The proper disposal method and disposal site will be defined prior to initiation of this task. Once the affected soils/sediments have been excavated, the area will be recontoured using clean borrow soil.

Weeks 3-4	Weeks 5-11	Weeks 12-16	Weeks 16-20	Weeks 21–24
Initiation & Completion of Field Work				
185KS 3-5	Sample Analysis Task 5	Data Inter-		
		pretation, Risk Assessment, & Draft Report Preparation <i>Task 6</i>		
			Remedial Action Plan Task 7	
				Final Report Preparation

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STATE OF NEW MEXICO



ENERGY, MINERALS AND NATURAL RESOURCES DEPARTMENT

**OIL CONSERVATION DIVISION** 

GARREY CARRUTHERS GOVERNOR

February 14, 1989

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87504 (505) 827-5800

## MEMORANDUM

## TO: FLORA VISTA FILE

FROM: DAVID G. BOYER, Hydrogeologist

SUBJECT: EPNG MANANA WELL SITE INVESTIGATION PROPOSAL

The following are my comments on the above proposal dated January 27, 1989.

### Section 2.0. Review of Existing Information

OCD files contain much significant information on the site including two reports of work, results of an aquifer test, water quality data, topographic survey, etc. These should be thoroughly reviewed by EPNG and consultants prior to the investigation to determine which information is adequate and which needs additional work or updating.

### Section 4.0. Hydrologic Investigation

<u>Piezometers</u> - Before installation of new piezometers, use of existing OCD piezometers should be considered. However, OCD wells will need to be cleaned of sediment before use. If new wells are used, construction procedures must be submitted to OCD prior to construction.

#### Trenches

Trenching is the method preferred by OCD for defining the extent and magnitude of oil contamination water sampling of the trenches should be conducted where hydrocarbons are suspected.

#### Pump Tests

OCD believes that only well S-5 need be pumped to monitor for contaminants as it has not previously been pumped and tested.

### Section 5.0. Collection and Analysis of Soil/Ground Water Samples

Analysis of the soil samples shall include analyses of the materials in the various pits, including the reserve pit, to determined the potential for leaching and migration of the wastes.

Some water samples should include analyses for PNA's, glycol, aliphatic and chlorinated hydrocarbons.

MEMORANDUM February 14, 1989 Page 2

# 6.0. Data Interpretation and Risk Assessment

No comments.

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# 7.0. Remedial Action Plan

More than just the inactive dehydration pit may need to be removed depending on test results. Remedial action should conform with and satisfy the criteria listed in my memo of June 15, 1988 30.07

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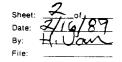
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ENGINEERING CALCULATION

KE. Beady, H. Van y EPNG D. Boyer (OC) • Somple the exceling OCD. (D. Boyer Konster Willber on site when sompling on site when sangeling). · TIEZOMETERS gais/mo 6.4 - One up - Two deningradiant 10.12 10.42 11.93 - Use the other OCD Prezzmetons 13.85 15.50 Include location and construction 17.46 19.39 grace-lunes C2 9.64 9.67 · ned a location may FIRENCHING Seeme OK as indicated on proposal. · PUMPTESTS - as Menno Feb. 14,59. Section 50 - Chlorinated Kychracalens) ( because solvents might have been resed) We (CPNG) will include analysis of the seil samples Aliphatic HC, Hydrocondon (lick at the Lee Acree for parameters, Sulfate, Sulfite, Arsenic, Jead Scientific reasoning that no leading will





FLORAVISTA

acour to accure non-leachability i) Country мw gals/mol 16.04 C1 6.4 30.07 10.12 C2 2) Chem! analysis 10.42 C3 58.12 58.12 11.93 3) Kemoval 13.85 13.71 86.18 15.50 In other words : 86.18 17,46 100.21 C7 a) Deliniate risense pit 114.23 C8 19.39 28.05 C2: 9.64 42.08 9.67 C3 b) Find out what is in there c) Remane d) Determine that no puller will continue the exist · When remained the contaminated sul pump the G.W. In degrees the E.W. level to allow remard of all 32.00 3.37 28.0 CO 4.19 44.0 the contaminated sail. The 64.06 SO2 5.50 H₂S 5.17 4.16 28.01 3.38 water can be simped to the swampy arra. · The Demestic water Well will be placed back in operation or at least have The capability. EProc'-will include a schedule of Remedial Action Tasko in the revised proposal.



P. O. BOX 1492 EL PASO, TEXAS 79978 PHONE: 915-541-2600

January 27, 1989

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Mr. David Boyer, Chief Environmental Bureau Energy and Mineral Department New Mexico Oil Conservation Division 310 Old Santa Fe Trail, 206 Sante Fe, New Mexico 87504

Reference: Flora Vista

Dear Mr. Boyer:

Enclosed for your review and consideration is our proposal for investigating the Manana Well Site near Flora Vista, New Mexico.

At your convenience, after you review this proposal, we would like to set up a meeting to discuss questions you may have and the project schedule.

If you have questions, please contact me at 915/541-2146 or Henry Van at 915/541-2832.

Very truly yours,

K. E. Beasley, P.E. Manager of Compliance Engineering North Region

KEB:cds

Enclosures

cc: J. F. Eichelmann III (w/ encl)
H. Van (w/ encl)



A PROPOSAL

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FOR

INVESTIGATING THE MANANA WELL SITE

NEAR FLORA VISTA, NEW MEXICO

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 Table 1. Analytical Parameters for Screening Purposes.
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#### 1.0 INTRODUCTION

El Paso Natural Gas Company (EPNG) has prepared this proposal for soil/groundwater investigation of the Manana well site near а The proposal includes a description of Flora Vista, New Mexico. investigation. The proposed tasks involved in the site a review of existing information investigation includes concerning the site, further characterization and topographic of the collective survey of the area, and interpretation observations and results. These efforts will culminate in a risk assessment of possible contamination from the inactive pit, and or removal of the recommendations for treatment contamination, if deemed necessary.

#### 2.0 REVIEW OF EXISTING INFORMATION

EPNG will collect and review existing information on conditions at site, specifically on the area between the inactive pit and the nearby water wells. EPNG will review this information to the pre-existing conditions at the site for comparison determine with findings of the EPNG field investigation. From this comparison, conclusions will be made concerning the likelihood of migration and/or degradation of contaminants from the pit. This information will be used to aid in designing the hydrologic investigation.

#### 3.0 AREA TOPOGRAPHIC SURVEY

A topographic survey of the area will be conducted to provide relative elevations of points of interest at the site. The survey will provide a grid, with points of known elevations, for siting the location of pits for soil sampling. The data will also be used in the description of the contaminant plume, if any, as well as the piezometric surface of the immediate area. Data generated during this task will be used to generate base maps which illustrate the locations of all relevant features and will establish baseline elevations for all aspects of the investigation.

#### 4.0 HYDROLOGIC INVESTIGATION

A hydrogeologic investigation will be performed define to the hydrologic setting and aquifer characteristics, such a groundwater flow direction and velocity, transmissivity, and coefficient. Additionally, the behavior of free oil storage if present, will be evaluated in and/or dissolved hydrocarbons, light of the physical characteristics of the saturated alluvial The extent of the investigation will sediments. encompass the area immediately surrounding the well site (including all pits), as well as the area down gradient from the well which could potentially be affected by past well operations.

The intent of the hydrologic investigation will be to illustrate, to the fullest degree possible, the lateral and vertical extent to

which the site has been affected. To assess the magnitude of migration, organic constituents associated with hydrocarbons will be measured (e.g., benzene, toluene, xylene, and oil and grease) as will inorganic constituents (e.g., metals) and indicators parameters such as pH and electrical conductivity (EC).

To achieve the objectives of the hydrologic investigation, piezometers will be installed to determine the horizontal and vertical gradients, trenches will be dug to visually assess the dimensions of the plume (if any) using stained soil/sediments as evidence, and pump tests will be performed using the local water wells and piezometers to define aquifer characteristics.

#### Piezometers

sites will be selected for the installation of new Four piezometers, one of which will serve to establish background concentrations for parameters measured at downgradient locations. Of the three remaining piezometer locations, one will be selected for a nested piezometer to determine the vertical gradient. Each piezometer will be installed using construction are tailored to the site, details, which to allow representative groundwater samples to be collected.

In addition to defining the vertical gradient, the piezometer nest will aid in the determination of the vertical extent of constituents in the groundwater. In conjunction with the installation of the piezometers, soil samples may be collected for laboratory analysis.

#### Trenches

To define the extent and shape of affected soil/sediments, it is proposed that trenches be dug perpendicular to the direction of groundwater flow. The purpose of the trenches will be to allow inspection of the shallow soil profile for evidence visual of migrating hydrocarbons. Additionally, soil samples can be collected for the profile at points where there is evidence of affected soils. Initially the trenches would be dug at 100-foot intervals to define the lateral extent of the plume. If it is determined that the 100-foot spacing is too coarse, then additional trenches will be dug as needed to define the extent and magnitude of the plume.

In addition to excavating trenches in downgradient locations, one trench or pit will be situated in an upgradient position to allow the collection of background soils samples.

#### Pump Tests

In order to evaluate the potential impact oil field constituents could have on the local water wells, it will be necessary to determine the aquifer characteristics. To this end, it is proposed that the nearest water well be pumped and the piezometers (and other wells) be monitored. The results of the pump test, along with the soils data, will numerically define the potential for migration of the oil field constituents prior to cleanup activities as well as subsequent to remedial efforts.

### 5.0 COLLECTION AND ANALYSIS OF SOIL/GROUNDWATER SAMPLES

Soil samples will be collected during the excavation of the trenches described in Section 2. Sample locations within the trenches will be based on visual evidence and the need to define the lateral as well as the vertical extent of any plume which may be detected. Collection of the samples will be from the face of the trenches using stainless steel utensils. All soil samples will be placed in glass jars and preserved in the field using ice to maintain the temperature of the samples at or near  $4^{\circ}$  C.

Groundwater will be collected for each of the samples piezometers installed during the investigation, from the piezometers currently present at the site, and from the local water wells. Sample collection will be accomplished using either dedicated PVC bailers, a Teflon bailer(s), dedicated pump, or a bladder sample pump. If dedicated equipment is not employed, then EPA approved decontamination procedures will be implemented between samples. Once collected, all groundwater samples will be filtered appropriate using 0.45 micron (for metals only), placed in containers, preserved with appropriate acids as needed, and stored in the field on ice at or near 4° C.

Both field analysis and laboratory analysis will be performed as samples are collected. Analysis of groundwater and soil samples will be conducted according to methods and procedures presented (EPA, in SW-846, Third Edition 1986) or an equivalent Parameters selected groundwater approved procedure. for analysis, as it pertains to the public water supply, will be selected from Section 3-301 of the WQCC Regulations. The selection of specific parameters from Section 3-301 will be based on discussions between OCD and EPNG. Soil samples and some groundwater samples collected for screening purposes will be analyzed for some, if not all of the parameters listed in Table 1.

Table 1.	Analytical	Parameters	for	Screening	Purposes.	
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Indicators	Organics	Inorganics		
 рН	Benzene	Barium	Chloride	
EC	Toluene	Calcium	Sulfides	
TDS	Xylenes	Magnesium	Selenium	
Oil & Grease	-	Sodium	Cadmium	
		Carbonate (and Bicarbona	Chromium ate)	

### 6.0 DATA INTERPRETATION AND RISK ASSESSMENT

The information gathered from previous investigations and the hydrologic investigation will be collectively reviewed to evaluate the existence, spatial distribution, and possible effects of contamination from the inactive pit and surrounding area. Utilizing these data, a risk assessment will be conducted to evaluate possible contamination to the nearby water wells and surrounding environment. The assessment will be based upon a comparison of the collective data to establish health and environmental criteria and potential for future migration.

#### 7.0 REMEDIAL ACTION PLAN

Given the relatively small size of the inactive pit, it is suggested that the contaminated soil from the pit be removed. The considered to be the major source of potential pit is contamination at the site; therefore, removal of the source is considered a cost-effective measure. Furthermore, in the extreme case where risk assessment suggests a hazardous to water quality, total removal of all tainted soils downgradient for the well area is proposed as a worst case remediation plan. The removal of this material will prevent future leaching of constituents to the groundwater. It is anticipated that removal can be accomplished by excavation with a backhoe and a dump truck. The proper disposal method and disposal site will be defined prior to initiation of this task. Once the affected soils/sediments been excavated, the area will be recontoured using clean have borrow soil.

ZIA REPORT 500 OAK STREET, N.E., P.O. BOX 27457 ALBUQUERQUE, NEW MEXICO 87125 (505) 242-5544 102 1 9 1938 1VLH8, 1988 CONS SAMI.

Mr. David G. Bover P.O. Box 2088 Santa Fe, New Mexico 87501

> Re: Flora Vista vs. Manana Gas, Inc., et al. San Juan County District Court Cause No. CV 86-00154-4

Dear Mr. Boyer:

Your deposition in the above-captioned cause that was taken on June 23, 1988, has been transcribed and is available for you to read and sign at our office. The office is open from 8:00 a.m. until 5:00 p.m., Monday through Friday. Please call before you come in.

If your deposition has not been signed by you within 30 days from today's date, the unsigned original transcript of your deposition will be forwarded to Mr. Byron Caton's office at that time.

Yours very truly,

**ZIA REPORTERS** 

Ouse Louise R. Williams

cc: Robert G. Stovall

265-2030 Rick Lougee



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ENERGY, MINERALS AND NATURAL RESOURCES DEPARTMENT

OIL CONSERVATION DIVISION

GARREY CARRUTHERS

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87504 (505) 827-5800

June 15, 1988

### MEMORANDUM

- TO: INTERESTED PARTIES
- FROM: DAVID. G. BOYER, Hydrogeologist Environmental Bureau Chief

REQUIREMENTS FOR FLORA VISTA CONTAMINATION INVESTIGA-SUBJECT: TION AND REMEDIAL ACTION PLAN

Attached are the technical requirements I believe necessary to perform a complete hydrologic investigation of the Flora Vista contamination site, and the clean-up criteria required to be met at completion of the remedial action plan. Also attached is a copy of the El Paso Natural Gas Company letter of December 23, 1987, that provides some suggestions for further study of the contamination. I will be happy to discuss the technical requirements, and provide clarification and additional information if needed.

# REQUIREMENTS FOR FLORA VISTA CONTAMINATION INVESTIGATION AND REMEDIAL ACTION PLAN

The Oil Conservation Division (OCD) believes that the most methodical approach for cleanup at the Flora Vista site is to conduct a hydrogeologic investigation and prepare a remedial action plan. A hydrologic investigation report must be prepared for OCD review and approval that includes the items detailed below. The report can also incorporate those items suggested for investigation in the El Paso Natural Gas Company (EPNG) letter of Concurrent with submittal of the hydrologic December 23, 1987. investigation report, is the submittal of a reclamation plan that provides for the removal of dissolved, emulsified and free-floating product from the aquifer and soils; and for the removal of actual and potential organic and inorganic ground water contaminants from site locations including the reserve and other pit locations. Actual reclamation will begin after OCD approval of the report and reclamation plan. All on-site locations will need to be examined to determine the totality of actual or potential contamination. Due to the proximity of the public water supply well, a successful reclamation action must include all possible contaminant sources. Α timetable for completion of the investigation and agency reviews is suggested.

- A. The hydrologic investigation report should provide the following:
  - 1.a. Definition of the horizontal and vertical extent and magnitude of petroleum and other fluid contamination both on and off-site in water and soil. Constituent investigation must include oil, glycol, emulsified and dissolved oil, salts, heavy metals, and chlorinated hydrocarbon contaminants. The investigation area must include the dehydration pit, other pit locations (i.e. tank drain, blowdown, etc.) the area between the contaminated water supply well and the Mary Wheeler #1E site, and the area between the contaminated water well and the other supply wells. It also will be necessary to pump and test well S-5 for possible contaminants.
  - 1.b. Definition of the horizontal and vertical extent of the drilling mud pit (aka. "reserve" or "slush" pit) and a determination of the type and concentration of drilling mud, inorganic salts, heavy metals, and hydrocarbons in pit sediments, and determination of the type and amount of any other material deposited in the pit prior to closure.
  - 2. Information on the rate and direction of oil and dissolved constituent migration;
  - 3. Monitoring well or excavated pit data including field observations of odors, soil discoloration, reports of petroleum encountered, and results of all chemical or physical analyses of water, soil or petroleum. If monitoring wells are installed, construction procedures must be submitted to OCD prior to construction;
  - 4. Water table elevation map(s);

- Map(s) showing locations of all known past pits, of trenches or pits excavated for the current investigation, and locations of samples taken for soil, water or petroleum analyses;
- 6. Map(s) showing petroleum contamination locations and thickness, and concentration of dissolved constituents;
- 7. A discussion of the potential for contamination of the other public water supply wells; and
- 8. Other technical information necessary or requested by OCD to determine the extent or magnitude of contamination.

An evaluation of existing information as outlined in the EPNG letter of December 23 would be useful in determining circumstances relating to the loss of petroleum and other fluids on site. However, it can not substitute for the detailed geohydrologic investigation required prior to remedial action commencement. Likewise, though pits could be useful, trenching as performed in August 1987 would seem to be the most expeditious procedure to use to locate the path the oil has followed to the well, and possibly beyond.

- B. Coincident with submittal of the hydrologic investigation report (see schedule below), a reclamation proposal for site remedial action shall be submitted that will include specific detail on technology or methodology to meet the following clean-up criteria:
  - 1. Dissolved, emulsified and free-floating petroleum, and other organic and inorganic water contaminants, must be removed from the ground water such that:
    - a) the water quality standards of Section 3-103 of the NM Water Quality Control Commission Regulations (as amended through December 24, 1987) are met; and
    - b) the USEPA drinking water standards for public water supplies in effect as of the date of this agreement are met; and
    - c) undesirable odors attributable to loss of petroleum or other fluids from the site are not present in ground water.
  - 2. The unsaturated (vadose) zone in the vicinity of the contamination shall not contain drilling muds, inorganic salts, hydrocarbons in quantities heavy metals or sufficient to recontaminate ground water. Contamination shall be deemed to occur if State of Federal numerical standards are exceeded, or through continued presence or reappearance of oil or grease, or undesirable odors in the water supply. Such recontamination shall not be allowed to occur as a result of seasonal rises in the water table, drainage, recharge events, or in any other manner. Soils contaminated with drilling muds, salts, heavy metals or hydrocarbons, and needing to be removed to prevent continued or

future contamination of ground water, shall be excavated and disposed of in locations approved in advance by OCD. Clean fill material may need to be provided to replace contaminated soils. Pumping of the affected water supply well for some length of time may be necessary to verify successful cleanup.

The remedial action plan should include an estimated schedule for completion of the various remedial action tasks.

C. The following schedule is proposed for investigation and remedial action:

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Within 120 days after parties agree to the study and remedial action requirements – submittal of hydrogeologic investigation report and proposed reclamation plan.

Within 30 days of OCD receipt of report and plan - OCD completes review and provides comments.

Within 15 days of receipt of OCD comments - response to OCD comments.

Within 15 days of receipt of response - OCD decision on approval of reclamation plan.

Within 15 days of OCD approval - begin remedial action.



P. O. BOX 4990 FARMINGTON, NEW MEXICO 87499 PHONE: 505-325-2841

December 23, 1987

Mr. David G. Boyer Hydrogeologist/Environmental Bureau Chief Energy and Minerals Department New Mexico Oil Conservation Division P.O. Box 2088 Santa Fe, New Mexico 87501-2088

Subject: Flora Vista Contamination Study

Dear Mr. Boyer:

On September 17, 1987 El Paso Natural Gas representatives met with your staff and held an informal discussion on the recent events at the Manana-Mary Wheeler #lE site. You requested additional information on the dehydrator operation at that time and we provided a chronology to you on October 28. At the meeting you also offered some suggestions on investigation and remediation at the site. Those recommendations were duly noted reviewed in the interim.

After some independent study as well as consideration of your suggestions we have put together some steps which we feel appropriate entities could implement to satisfy all concerned that the Flora Vista water supply is adequately protected from any possible contamination at the well site. Your views on the viability of this plan would be appreciated. The plan is, of course, conceptual and some details would need to be worked out later. However, we believe that the objectives and basic steps required to achieve them are spelled out clearly enough to allow NMOCD to evaluate them.

First, a detailed evaluation of existing information on site conditions would be required. This evaluation would include a review of all data and reports which have been generated in the years since problems at the Flora Vista Water Users well field were first identified. This review would help determine preexisting conditions which can be compared with field investigations to be conducted later.



Mr. David G. Boyer December 23, 1987 Page 2

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Several pits would be necessary to locate and define a contaminant plume should one exist. Additional piezometers might also be required at this time. Analyses for samples collected from pits and all piezometers should include benzene, toluene and xylene although other constituents could conceivably be identified during the investigation.

A risk assessment should then be conducted to evaluate the existence and possible effects of contamination at the well site. This assessment would be based on field information, previously collected data, and a comparison with established health and environmental criteria.

Finally, the plan should include excavation of contaminated soil from the pit area. This material should be removed from the area to an acceptable site.

In summary, the steps outlined above would allow concerned parties to identify and delineate any contamination in the area of the Manana-Mary Wheeler #1E well site and the Flora Vista Water Users well field. In addition, this type of plan would provide guidance for any necessary remediation and the subsequent assurance that the Flora Vista water supply will not be adversely affected by any contamination at the above-mentioned gas well location. El Paso is interested in any comments that you might have on the plan and thanks you in advance for your views.

Sincerely Yours,

Kenneth E. Beasley III Compliance Engineer

Mail to : Bryon Calon 333 East Man Forel Sarmington, NM 37401 Richord CHESTER F Miller, II Richard Louge .406 Aurport Road Farmington, NM \$7401 P.D. Srawer 887 Azbrignergue, NM 87'32 faterogener stuart castle EIS Water Supply Runnels Blog. Sont 6/15/28 Santa S-e bennis McQuillan Els Grand water Runnels Blog. Sent Ken Beakley Sant R Henry, Van 6/17/38 Bill, Olson Els Ground Wale Runnels Bldg

ANALYSIS OF HYDROLOGIC AND ENVIRONMENTAL EFFECTS OF DRILLING MUD PITS AND PRODUCED WATER IMPOUNDMENTS

VOLUME I:

Executive Summary and Report

For

THE AMERICAN PETROLEUM INSTITUTE

Reviewed Some in preparation ne 15, 1988, 4.J.8n mo Memorandumand June 23, 1988 Jop Tion

Prepared by DAMES & MOORE HOUSTON, TEXAS

October 1982

#### ACKNOWLEDGEMENTS

The authors express sincere appreciation to the American Petroleum Institute and to the Task Force on Mud Pits and Produced Water Impoundments for their cooperation, effort, lively discussion, and support in the successful completion of this study. The American Petroleum Institute and Dames & Moore Consultants in Environmental and Earth Sciences acknowledge Hall Environmental Services for invaluable assistance in data management and analysis, and IHI-KEMRON and SPECTRIX CORPORATION for chemical analysis of the multitude of water, soil, and vegetation samples performed for this study. Special thanks are given to G. Henderson, J. Woodrow, C. Wills, V. Keys, T. Westbrook, J. Slikas, A. Laramee, J. Salinas, E. Aguilar, C. Johnson, T. Mellini, and D. Bartlett for their efforts in sampling and installation of monitoring wells. Appreciation is given to C. Wills and Dr. N. Fotheringham for their dilligence and effort in the implementation of data analysis and G. Henderson for interpretation. We would also like to thank R. Orosco, S. Edwards, and G. A. Smalley for their guidance and assistance on the program. Special thanks to S. Jackson and J. Knowles of Triad Drafting Services for technical illustration, and R. Fisher, D. Szuba, J. Campbell, E. Cook, and others for word processing and patience. Deep appreciation is expressed to the personnel at all sites visited for their cooperation, help and guidance in the study, and to the owner companies and regional offices for information and assistance throughout.

The committee members of the Task Force on the Mud Pits and Produced Water Impoundments Study are:

J. W. Collins, Cities Service Company, Chairman
J. Greer, API staff
Dr. G. H. Holliday, Shell Oil Company
A. B. "Bud" Parker, ARCO Oil & Gas Company
H. R. Moseley, ERCO Petroleum Services
R. Abbott, CONOCO, Inc.

# 1.0 EXECUTIVE SUMMARY

# 1.1 BACKGROUND

In the proposed hazardous waste management regulations of December 18, 1978, the Environmental Protection Agency (EPA) classified oil and gas drilling muds and oil production brines (produced water) as "special wastes" (43 FR 58949-58968). As the result of a Congressional amendment, these wastes were exempted from the Resource Conservation and Recovery Act (RCRA) in the May 19, 1980 regulations. EPA was mandated by Congress to conduct studies to evaluate the effects of onshore disposal of drilling muds and produced waters because the effects of these wastes on human health and the environment were not well documented. The American Petroleum Institute (API) sponsored the independent study reported herein to provide data on the effects of these oil and gas field waste management and disposal activities.

# 1.2 PURPOSE OF THE API STUDY

The purpose of the API Study was to investigate the leaching potential of possible hazardous constituents from selected oil and gas field drilling mud reserve pits and emergency produced water impoundments. The objectives were:

- 1. To evaluate whether constituents leach in sufficient quantity to present a significant hazard to human or the environment; and
- 2. To investigate the possible transport pathways of leached constituents.

#### 1.3 PARAMETERS STUDIED

Parameters selected for chemical analysis in ground water, subsurface and surface soils and/or vegetation at these sites included arsenic (As), barium Ba), cadmium (Cd), hexavalent chromium (hCr), total chromium (tCr), lead (Pb), mercury (Hg), zinc (Zn), sodium (Na), chloride (Cl), bromide (Br) (one site only), total alkalinity (Alk), specific conductivity (SC), pH and temperature (T°C) (ground water only). The seven heavy metals As, Ba, Cd, tCr, Pb, Hg

and Zn were chosen because 1) they can occur as trace contaminants of clays or barite, both of which are major components of many drilling fluids; and 2) current regulations have set standards for acceptable concentration limits of these elements in drinking water and/or ground waters downgradient of disposal facilities. The other parameters--Na, Cl, Alk, SC, pH and TC--were measured primarily to provide supplemental information on ground water quality and chemical characteristics of water and sediments.

# 1.4 TYPES OF FACILITIES INVESTIGATED

# 1.4.1 Drilling Mud Reserve Pits

Mud pits are usually in operation only during the drilling of an oil or gas well, a period of time ranging from a few weeks to about a year, depending upon the depth of the well. Mud pits are used for onsite emergency reserve drilling mud storage during drilling and, at the completion of the well, to contain waste drilling fluids, cuttings, and site runoff. The type of drilling mud used (i.e., the "mud system") is generally a function of the depth to, and the intervening lithology of, the strata between the surface and the oil or gas producing formation. The majority of the wells drilled in the United States are advanced to depths of less than 5,000 feet and therefore require relatively small amounts of drilling fluids. For this study, deeper wells were selected because of the larger, more complex mud systems involved. Four of the six mud systems contained chromium lignosulfonate additives.

# 1.4.2 Produced Water Impoundments

Produced water impoundments are usually in operation for relatively long periods of time (10 to 30 years). The impoundments investigated in this study are used to contain produced water and small amounts of oils collected during emergency conditions. A given produced water impoundment generally services several wells. Produced water is directed to such impoundments when an emergency situation occurs (such as failure in the injection equipment or disposal well resulting in overfilling of the produced water storage tank). Produced water impoundments are constructed either from natural clay sediments--or are artifically lined--to provide very low permeability.

# 1.5 SITES STUDIED

The eight sites selected for the American Petroleum Institute Mud Pit Produced Water Impoundment Study (API Study) represented current industry practices for operation, maintenance and/or closure of mud pits and produced water impoundments. The eight sites are from three major hydrogeologic regions, each with different potentials for leachate migration and detection. These regions were:

- Northern, semi-arid sites on consolidated sedimentary facies. The three sites located in this region were two mud pit sites in the Wyoming Overthrust Belt region and one mud pit site in the Williston Basin of North Dakota.
- Southern, transitional climate sites on semi-consolidated sedimentary facies. The two sites in this region were located in Northeast Texas (a mud pit site) and in Southwestern Arkansas (a produced water impoundment site) on opposite sides of the Sabine Uplift.
- 3. Southern, coastal sites on unconsolidated sedimentary facies. Three sites were located in this region, one mud pit and one produced water impoundment site, both in Southern Louisiana, plus one mud pit site in the middle of the Gulf Coastal bend in Southern Texas.

The sites in Region 1 have high leaching potential based on subsurface lithogic conditions but low potential based on annual precipitation rates and patterns. The sites in Region 2 have high leaching potential based on both subsurface lithology and annual rainfall. Sites in Region 3 had either low or high leaching potential rates based on lithology (depending upon presence or absence of heavy clay layers near-surface) and high potential leaching rate based on annual rainfall. These three regions are considered representative of the majority of the regions where oil and gas drilling and producing operations occur.

# 1.6 CONTAMINATION PATHWAYS

In the planning phase of the study, three generic pathways were identified: a) leaching and subsurface movement of constituents downward and

laterally through subsurface soils to ground water via infiltration, percolation and ground water flow; b) surface movement to downslope soils and/or downstream drainage basins via erosion and runoff; and, c) in situ uptake by vegetation on or adjacent to disposal areas and subsequent incorporation into the general food web via herbivores.

# 1.7 RESULTS AND CONCLUSIONS

Results of chemical analyses of ground water, subsurface soil and surface soil, and vegetation samples at the sites indicate the most motile ions found to leach from both mud pits and produced water impoundments were Na and Cl. Levels of Cl return to levels below Secondary Drinking Water Standards within a matter of several hundred feet at most facilities, and do not constitute either a health hazard or a water potability problem for public and industrial ground water uses. Probable migration rates of heavy metals from both mud pits and produced water impoundments to ground water appear to be very slow and the observed concentrations measured in ground waters at the sites are low and do not indicate a hazard to human health. Subsurface soils, surface soil and vegetation samples at all sites showed elevated levels of heavy metals, Na and Cl in pit and/or downgradient locations. However, apparent rates of migration are slow, based on the observation that "contaminated" subsurface layers, when they exist, are found in relatively narrow, shallow bands close to the point of origin. Moreover, although the study indicates elevated levels of some heavy metals in plants growing on some reclaimed mud pits and bordering some impoundments, the small size of the area of affected plants does not represent a significant portion of the available forage resources of the areas studied, and is unlikely to constitute a significant environmental or health hazard.

# 5.0 CONCLUSIONS AND RECOMMENDATIONS

The observations made and the data collected during the API Study led to a number of conclusions regarding both the effects of mud pit and production impoundments and the efficacy of the methods used to evaluate such effects for these and other disposal facilities.

These conclusions, along with recommendations for changes in certain methods, are summarized in this section. In general, results of chemical analyses of ground water, subsurface soil and surface soil, and vegetation samples at the sites indicate the most motile ions found to leach from both mud pits and produced water impoundments were Na and Cl. Levels of Cl in ground water return to levels below Secondary Drinking Water Standards within a matter of several hundred feet at most facilities, and do not constitute either a health hazard or a water potability problem for public and industrial ground water uses. Probable migration rates of heavy metals from both mud pits and produced water impoundments to ground water appear to be very slow and the observed concentrations measured in ground waters at the sites do not indicate a hazard to human health.

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Specific conclusions and recommendations include the following:

- None of the extract mg/l concentrations for any samples at any of the sites exceed RCRA guidelines of 100 X PDWS. Therefore, samples of mud pit mixtures and soils near produced water impoundments analyzed in the API Study are not hazardous waste mixtures.
- 2. The data presented here, in conjunction with the reported inconsistency and unpredictability of the extraction methods as routinely run by reputable analytical laboratories, indicate a more reliable method for setting performance standards to regulate solid waste facilities under RCRA must be developed.
- 3. Location rankings indicate elevated uptake of heavy metals by plants may occur on soils with elevated heavy metal concentrations, and that heavy metals in drilling mud mixtures may be available for plant uptake. However, due to the small area of affected plants, the impact of disposal-derived heavy metal concentrations on the forage resources of the areas studied is insignificant and should not constitute a threat to human health or the environment.
- 4. Even though parameter concentrations in downgradient wells tend to be higher than those in upgradient or background control wells, the mean concentrations in downgradient wells for most heavy metals at most sites are below Primary and Secondary DWS limits. Some metals did exceed PDWS in downgradient and/or control wells during the period of study, particularly As, Ba and Cd. Mean Cl concentrations exceeded SDWS at all the study sites except DND and VLA. With the exception of KTX, all Cl exceedances were confined to downgradient wells.
- 5. All sites show significant differences in direction of ground water flow with season. In several sites, the direction change appears to be reflected in the relative concentrations of certain parameters,

especially Cl. These data indicate that changes in flow direction must be taken into account when monitoring wells are placed onsite, and must be incorporated into data analysis and interpretation.

- 6. Overall mean concentrations in ground water for most parameters are low at all sites. However, high levels of three or four metals are apparent at each site, and high Cl levels are apparent at four sites.
- 7. Comparison of the distribution of the number and magnitude of exceedances of DWS in up- and downgradient wells indicate that the percent exceedance is generally not very different among wells for most sites. However, the magnitude by which DWS limits are exceeded is almost always less in control wells than elsewhere.
- 8. Statistical tests indicate well location affects the concentrations of many of the measured parameters, particularly Ba, Na, Cl and Alk. Season affects most parameters at most sites, particularly concentrations of Ba, Cd, Hg, Zn and Cl. Treatment did not significantly affect most parameters at most sites, indicating that, in general, filtering ground water samples does not affect concentrations of most measured parameters.
- 9. With one exception (DND), rankings of mean concentrations indicate ground water samples collected from downgradient or pit area locations are higher for most parameters than are samples collected from control well locations.
- 10. Four extract methods were used for chemical analyses of soil samples. Data for these extracts were converted to mg/kg oven dry weight concentrations and were ranked to examine relative extract "strength" for each parameter. In general, concentrations in EPA extracts are higher than those in TDWR extracts at all sites except DND, where the rankings are reversed. At all sites (except DND), Cl concentrations

are higher in TDWR extracts than in EPA extracts, but Na is reversed, being always higher in EPA extracts than in TDWR extracts.

- 11. Concentrations found in different extracts for the same soil samples were regressed for each parameter. The results of the regression analyses indicate that the concentrations found in one extract had very little, if any, relationship to concentrations in other extracts, except for Na and Cl in TDWR versus EPA extracts.
- 12. Levels of heavy metals, Na and Cl tend to be elevated in subsurface soil locations in or near pits or impoundments at all sites. However, as evidenced by ground water data, most parameters do not appear to migrate any appreciable distance away from these disposal and storage facilities. The only parameters which showed definite evidence of vertical migration downward through subsurface soils were the mobile ions Na and Cl. High SC was also characteristic of contaminated zones.
- 13. No statistically significant relationship between soil and vegetation concentrations was evident for most of the measured parameters at the study sites. One of the reasons for the apparent lack of correlation between measured concentrations in soils and vegetation was the evident variability in extraction "success". As a result of these extraction problems, all surface soil and vegetation results should be evaluated only in relation to each other at each site and not in relation to absolute levels determined at other sites or in other studies.
- 14. The QA program results for soils and vegetation extracts indicate that all extraction methods investigated for the API Study had severe problems with both recovery and replication reliability. Given these results, all data presented in this report for soils and vegetation should be viewed as indicative of <u>relative concentrations</u> of the various ions within each site, but are suitable neither to rigorous

statistical treatment nor to in-depth visual inspection and comparison of <u>exact values</u> between sites. It should be emphasized that these reservations are applicable only to soils and vegetation extraction results and are not relevant to ground water results. STATE OF NEW MEXICO COUNTY OF SAN JUAN IN THE DISTRICT COURT

FLORA VISTA WATER USERS ASSOCIATION,

Plaintiff,

vs.

MANANA GAS, INC.,

CV 86-00154-4

Defendant/Third Party Plaintiff,

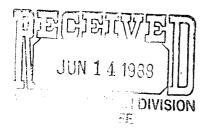
vs.

EL PASO NATURAL GAS COMPANY,

Third Party Defendant.

# NOTICE OF DEPOSITION

NOTICE IS HEREBY GIVEN that Defendant/Third Party Plaintiff, Manana Gas, Inc., will take the deposition of David Boyer, upon oral interrogatories, before a Certified Court Reporter and Notary Public from Zia Reporting Service, at the hour of 10:00 o'clock a.m. on Thursday, June 23, 1988, at the offices of Zia Reporting Service, 500 Oak Street, N.E., Albuquerque, New Mexico.

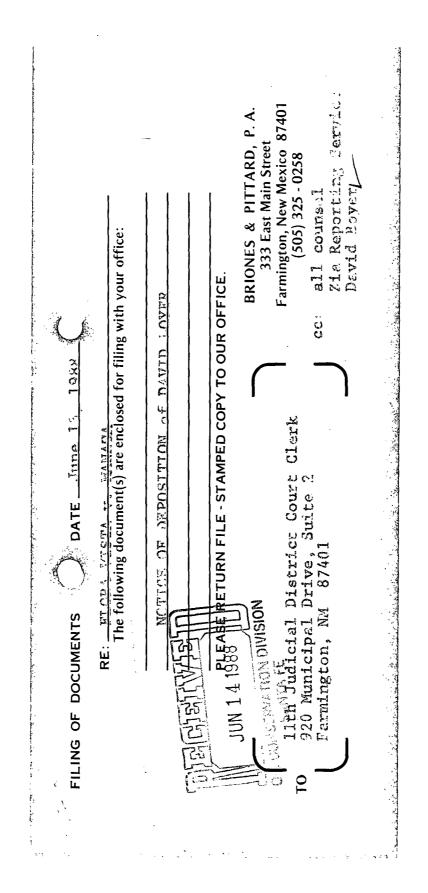


BRIONES, ODENWALD & CATON, P.A. 333 East Main Street Farmington, NM 87401 (505) 325-0258

Byron Caton

Attorney for Defendant/ Third Party Plaintiff

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I hereby certify that a copy of the foregoing was mailed to opposing counsel of record this 13th day of June, 1988.

- B Byron Caton

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P. O. BOX 4990 FARMINGTON, NEW MEXICO 87499 PHONE: 505-325-2841

December 23, 1987

Mr. David G. Boyer Hydrogeologist/Environmental Bureau Chief Energy and Minerals Department New Mexico Oil Conservation Division P.O. Box 2088 Santa Fe, New Mexico 87501-2088

Subject: Flora Vista Contamination Study

Dear Mr. Boyer:

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Sincerely Yours,

Kenneth E. Beasley III Compliance Engineer ENGINEERING CALCULATION

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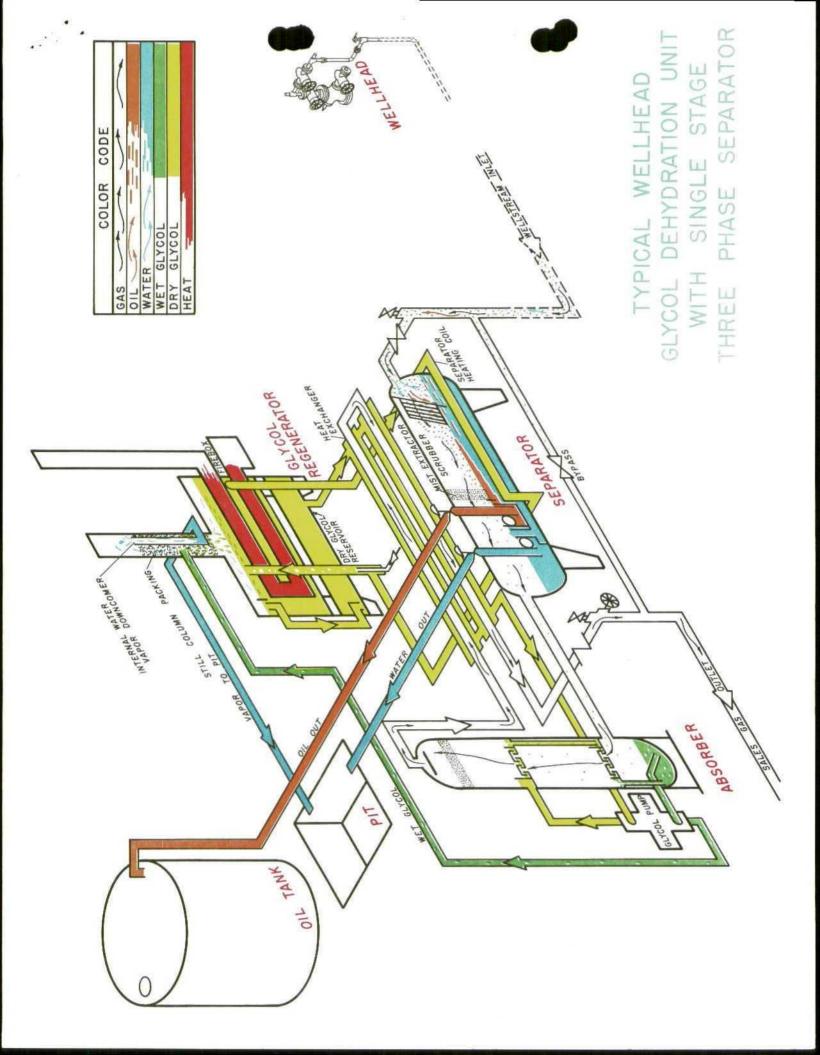
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P. O. BOX 4990 FARMINGTON, NEW MEXICO 87499 PHONE: 505-325-2841

October 28, 1987

Mr. David G. Boyer Hydrogeologist/Environmental Bureau Chief Energy and Minerals Department New Mexico Oil Conservation Division P.O. Box 2088 Santa Fe, New Mexico 87501-2088

Natural Gas Company

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NGAD-Beyonne, N. J.	
ENCAD	<b>B</b>

Subject: Flora Vista Contamination Study

On Wednesday September 17, 1987 in a meeting with your staff we briefly discussed the Manana Mary Wheeler #lE location and dehydrator operation. At that time a need for further information on the dehydrator was identified and you requested a chronology of operation and modifications to the unit. Written records are essentially non-existent and some of the individuals involved in the well's operation during the time in question are no longer available for comment. The following summarizes the dehydrator operation as best as can be recollected by location personnel.

The Manana-Mary Wheeler #1E was first delivered on July 16, 1980. El Paso Natural Gas installed a four million BS&B two phase dehydrator with the El Paso Natural Gas number 2303 at the location. The vapor line from the still column, the glycol storage vent and the liquid phase dump all were routed to an unlined pit immediately south of the dehydrator. (See Sketch #1) No records relating to the operation of the unit or condition of the associated pit are available from first delivery until 1983.

In the summer of 1983 the l" liquid dump line was re-routed and connected to the 2" pipeline leading to the producer's tank. The top was removed from a 55 gallon drum and the drum placed in the pit to catch any liquid which might drip from the vapor line from the still column on the dehydrator's glycol regenerator. (See Sketch #2) This action was taken since the Flora Vista Water Users had expressed concern about contamination of their water wells adjacent to the location. The pit was dry at this time with no sign of liquids in the pit from the liquid dump, glycol storage vent or still column vapor line. Operating personnel checked this pit regularly from the summer of 1983 to December 1986. No glycol leaks or fluids in the pit were ever in evidence during that time. In addition, El Paso installed a sensing element in 1983 that would shut in the well should slugs of liquid enter the dehydrator and cause an excessive signal to be transmitted to the liquid dump valve. (See Sketch #2) This shut-off device must be reset manually to allow the well to begin producing again once it has tripped. In the years since its installation the valve has been found in the shut position on various occasions.

In December of 1986 operating personnel turned the dehydrator off, disconnected the vapor line from the still column and the vent on the glycol storage and plugged all vents. There was no evidence of discharges to the unlined pit at this time. This action was taken as part of El Paso's program to insure compliance with NMOCD regulations prohibiting discharge to certain unlined pits in the vulnerable groundwater area.

Shutting down the dehydrator was actually unnecessary since discharge from the still column was being collected in the 55 gallon drum installed for that purpose. For this reason, the dehydrator was returned to service in early spring of 1987. The vent lines were re-installed and routed to the 55 gallon drum.

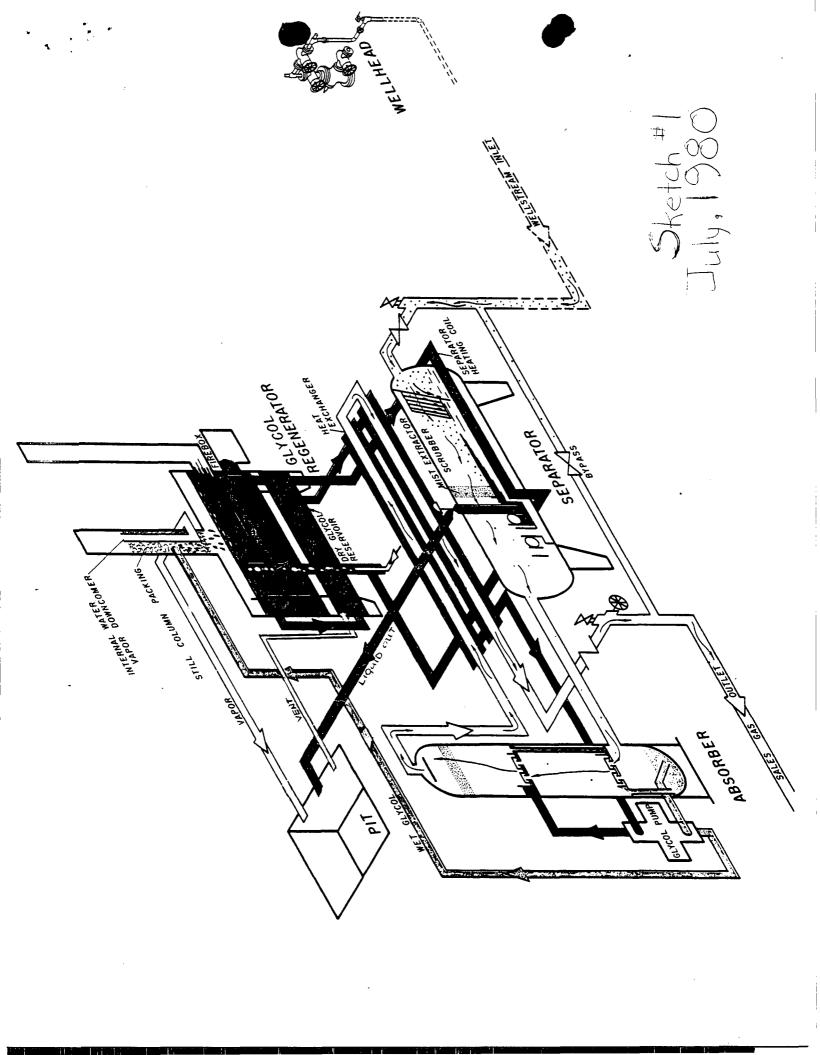
In June of 1987 the pit immediately south of the dehydrator was closed since its use was no longer required. The dehydrator was turned off and remains out of service.

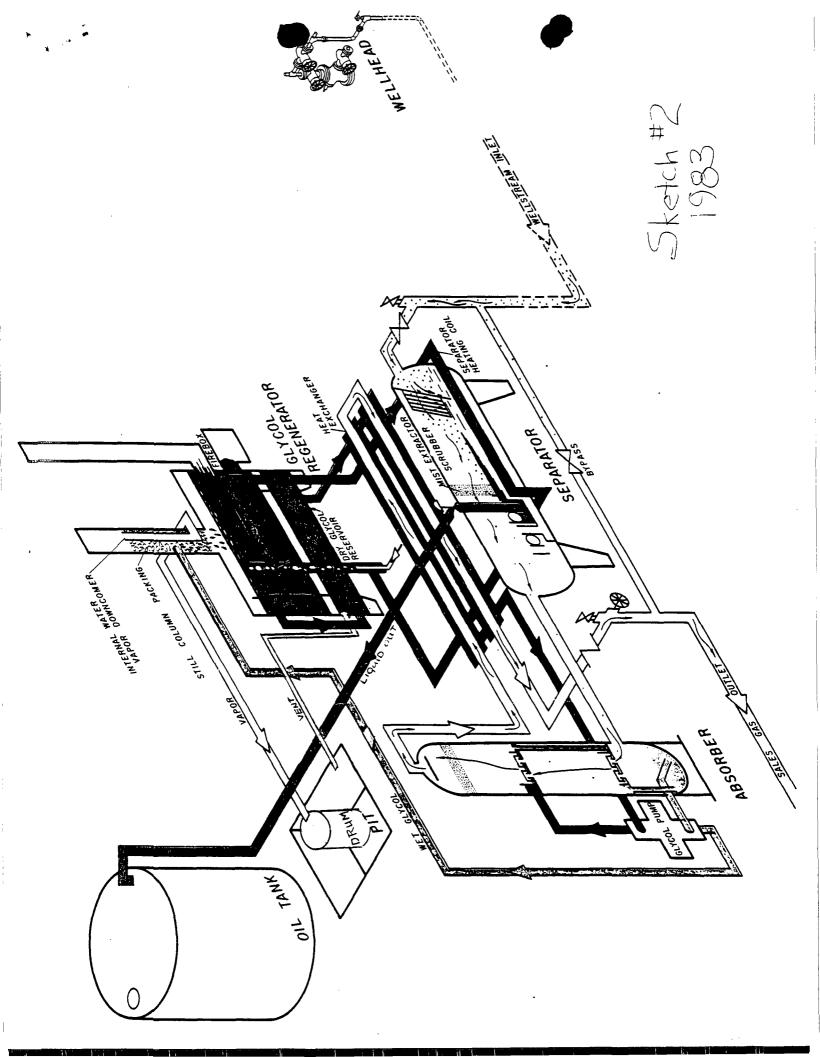
The above is the most complete summary available of dehydrator operations and modifications from 1980 to present. The absence of written records precludes greater detail but it is believed that this chronology will nevertheless provide some insight into the operation of the dehydration unit and related site conditions. El Paso has no direct knowledge of the operation of any primary separation equipment.

EL Paso is evaluating possible remedial measures to be taken at the site. NMOCD will be kept apprised of activities in this area. Please feel free to contact me for further questions.

Sincerely yours,

Kenneth E. Beasley III Compliance Engineer





STATE OF NEW MEXICO

ENERGY, MINERALS AND NATURAL RESOURCES DEPARTMENT

**OIL CONSERVATION DIVISION** 

GARREY CARRUTHERS

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87504 (505) 827-5800

September 8, 1987

CERTIFIED MAIL RETURN RECEIPT REQUESTED

Dr. Henry Van Environmental Affairs Department El Paso Natural Gas Company P.O. Box 1492 El Paso, TX 79978

RE: Flora Vista Contamination Study

Dear Mr. Van:

As you know, since March, 1985, the OCD has been investigating the source of oil contamination at the Flora Vista Municipal Well contaminated in 1983. We have maintained that the source of the contamination was from one or more of the production facilities at the Manana Gas Mary Wheeler #1-E gas well located southeast of Flora Vista in Unit M (SW/4, SW/4) of Section 23, Township 30 North, Range 12 West. Such facilities include well separator, separator tank, condensate tank, condensate water tank, the now-closed reserve pit and the El Paso Natural Gas dehydrator/separator and now-closed dehydrator pit.

On April 22, 1986 an aquifer (pumping) test was conducted at the water well to determine hydrologic characteristics, and to ascertain whether oil was still present in the subsurface adjacent to the well. Oil was drawn into the well five hours after the test commenced, and continued to be captured for the duration of the test (48 hours). On Monday, August 17, 1987, Manana Gas and the Flora Vista Water Users Association began a joint investigation of the site using a backhoe to dig a series of trenches, beginning at the water well and continuing up gradient until the source(s) were located. Trenching completed on August 17 indicated that the major source of contamination was at the site of the now-closed EPNG dehydrator pit. Further trenching on August 18 confirmed this. Kenneth Beasley, compliance engineer for El Paso Natural Gas in Farmington was notified of the situation at approximately 6 p.m. on August 17 and El Paso staff were on site for trenching on August 18.

Pollution of fresh surface or ground waters is prohibited by the New Mexico Oil and Gas Act, the New Mexico Water Quality Act, and the New Mexico Public Nuisance statutes. The New Mexico Oil Conservation Division is authorized to administer and enforce regulations for the former two statutes, and enforce the public nuisance statutes. Accordingly, this Division requests your cooperation in further defining the extent of contamination, its cause, and in developing a remedial action plan to contain, remove, and otherwise mitigate and abate the contamination at this location. At a meeting previously scheduled for 9:00 a.m., Wednesday, September 16, 1987, the OCD will initiate preliminary discussion with EPNG that hopefully will lead to voluntary remedial action by EPNG, either singularly or with Manana, to alleviate the problem. For this meeting, it would be helpful if EPNG could provide any diagrams of the dehydrator/separator unit originally at the site, any modifications subsequently made, and copies of any spill reports made to this agency or other regulatory agencies. OCD will provide suggestions for additional investigation, and possible regulatory mechanisms to define and facilitate remedial actions.

As this will be a preliminary meeting on this matter, and because other unrelated matters involving EPNG will be discussed, I do not intend to involve other parties or agencies in this meeting. However, I expect that other meetings that will include all interested parties will be scheduled as action on this issue develops. If you have any questions, please contact me at (505) 827-5812.

Sincerel KRA

David G. Boyer Hydrogeologist/ Environmental Bureau Chief

cc: Kenneth E. Beasley, EPNG Farmington OCD - Aztec Jeff Taylor, OCD





P. O. BOX 2079 • FARMINGTON, NM 87499 • (505) 327-3303 CLOVIS, NM • (505) 763-4255

September 3, 1987

Mr. Bert Barns Flora Vista Water Users' Association 113 East Ute Farmington, NM 87401

Re: Well Polution

# Dear Bert:

Transmitted herewith is a copy of a report prepared by Dr. R.W. Blair, Jr. regarding the trenches and the geology in the vicinity of the Flora Vista Water Users Association water wells and the Manana Oil & Gas Company gas well, Mary Wheeler #1E. The dates of observation noted should be August 17, 18, and 19, rather than August 18 and 19. Dr. Blair is submitting a corrected page number one for this report. I am, however, transmitting copies of the report for your review. I believe that they are self explanatory. If you have questions regarding the report, please feel free to contact me or Dr. Blair at your convenience.

Sincerely yours,

BREWER ASSOCIATES, INCORPORATED Richard P. Cheney, P E., L.S. President RPC:tm L1Ø87/F-331 OIL CONSERVATION DIVISION SANTA FE Enclosure cc: Mr.Rick Lougee 

SEP 1 1 1987

George T. Walters, P.L.S.

# FLORA VISTA CONTAMINATION STUDY SUMMARY OF TRENCHING ACTIVITIES 8/18 TO 8/19 1987

by R. W. Blair, Jr. consulting geologist

# SITUATION

Systematic trenching with a backhoe was commenced on August 18th and 19th, 1987 between the Flora Vista Water Users Association's water well S1 and Manana Oil and Gas well, Mary Wheeler no. 1-E. The objective was to trace and isolate the source of hydrocarbon contamination to the S1 water well. Fresent during trenching on the 18th were Ed Hartman, President of Manana Gas, Inc.; David Boyer, hydrologist with New Mexico Oil Conservation Division; R. W. Blair, Jr., consulting geologist representing Brewer Associates; Ray Penrod, representing the Flora Vista Water Users Association and the backhoe operator. Fresent during trenching on the 19th were Ed Hartman, R. W. Blair, Jr., Ray Penrod, Frank Chavez, district supervisor, N.M. Oil Conservation Division, Ken Beasley and six other El Paso Natural Gas Company employees, and the backhoe operator.

# TECHNIQUES AND METHODS

1. Thirteen trenches, seven to eight feet deep were dug in all, six (A thru F) on the 18th and seven (G thru M) on the 19th (see attached map). Trench location on the first day was determined by David Boyer. On the second day Mr. Ed Hartman, and myself (Blair) determined trench location.

2. All trenches were documented as to location, orientation and hydrocarbon contamination.

3. The trenches were dug in such a manner that: a) they intersected the existing water table, b) berms were placed every 8 to 10 feet at bottom of trench to minimize groundwater mixing, c) they were generally oriented perpendicular to the direction of groundwater flow direction, and d) they were positioned to maximize information concerning location and source of contamination.

4. Mr. Ed Hartman with my assistance obtained water samples from all contaminated trenches and a few of the uncontaminated trenches which are to be analyzed for hydrocarbons. David Boyer took water samples only from the six trenches excavated on August 18th. These are to be analyzed for specific organic pollutants such as benzene and toluene.

5. A map was constructed using standard pace and compass techniques of the trenches, well sites, and associated gas

# SEP 1 1 1987

production structures (see attached map).

# DISCUSSION

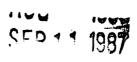
The water and gas well are located upon a point bar deposit (sand and gravel deposited from a meandering river). Two flat depositional surfaces are evident (see attached map). The lower surface represents the modern floodplain and is found adjacent to the Animas River and the upper surface or terrace (18 to 24 inches higher) is an older floodplain. The point bar deposit consists of sand and gravel with cobbles commonly 12 inches in diameter. The upper 12 to 18 inches of both surfaces consists of a sandy loam soil.

The water table is measured in the trenches at 5 to 6 feet upper terrace surface. Uncontaminated water is below the slightly muddy with some natural organic foam or scum present on the surface. Contaminated water displays an iridescent oil film along with a strong hydrocarbon smell. The contaminated zones are recognized in the gravel deposits from a black stain which coats the sand and cobble surfaces and from strong noxious organic odors emanating from the stained horizon. The contaminated zone is found to coinside with the water table flux zone, because the hydrocarbons tend to float and be carried along at the top of the water table surface. Thus, only rarely is the contaminated zone found within the upper five feet. The smelly black organic stain is noted in the contaminated trenches and is marked on the attached map with cross hatching. The contaminated zones vary in thickness from a few inches to several feet (the thickest zone is found just south of the dehydrator).

In trenches A and C a thin discontinuous black stained zone is noted, but it is not associated with any hydrocarbon odor. This zone may have been contaminated during the producing periods of the water well SI, because the cone of depression would have captured the contaminated flow documented to the west of the well. Upon cessation of pumping, fresh, clean ground water probably flushed and leached out the volatiles, but still left an insoluble organic black stain.

As noted from the map, the contaminated zone appears to be confined to a narrow band and is shaped like a giant comma, trending south. Ground water flow is to the south, as recorded by David Boyer (1986); thus, the source of hydrocarbon contamination is found in the vicinity of the dehydrator and specifically from an area immediately south of the dehydrator (old dehydrator pit). If liquid hydrocarbon waste was dumped into the gravels south of the dehydrator near the surface and above the water table, they would spread laterally in all directions; and therefore, could conceivably contaminate an area immediately north of the dehydrator as indicated in trench I. Once the waste fed into the water table it would be carried in the direction of ground water flow. The slight westward trend of the contamin-





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ation plume as it trends south may be due to natural flow along a highly permeable zone which corresponds closely with the upper and lower terrace boundary or it may reflect ground water capture due to the pumping of S1 and its associated cone of depression.

# CONCLUSIONS

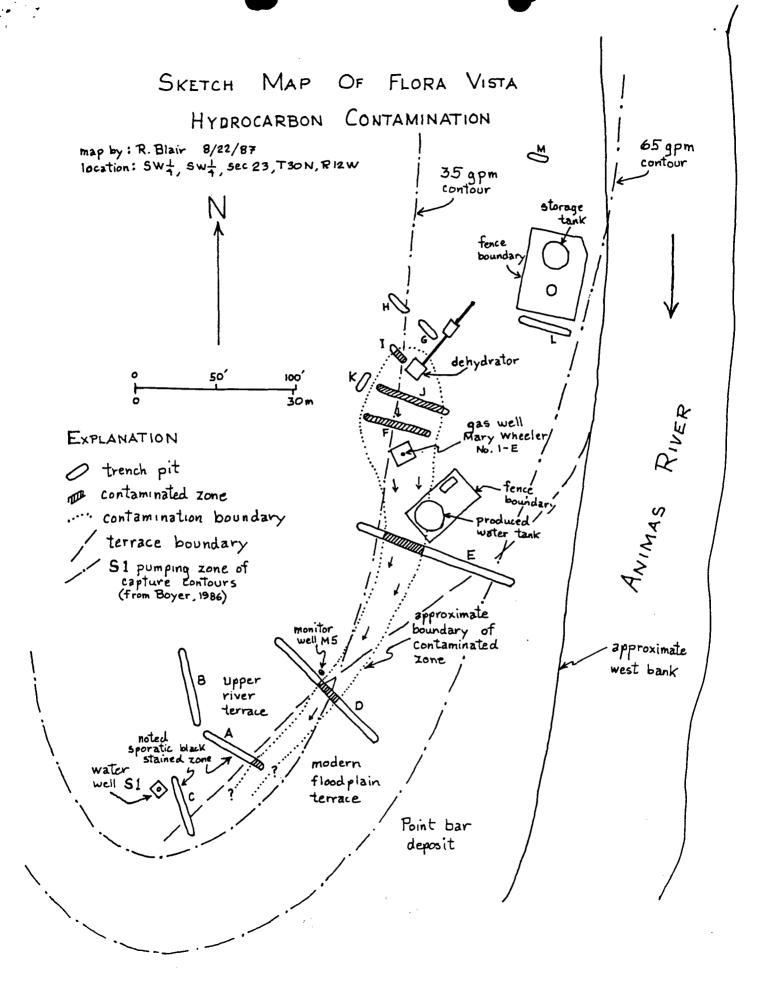
1. Contamination of the S1 water well was due to capture of the hydrocarbon plume identified from trenching and shown on the attached map.

2. The source of the hydrocarbon plume appears to be a circular zone located immediately south of the present location of a dehydrator owned by El Paso Natural Gas Company.

#### REFERENCES

Boyer, David G., 1986, Final report on Flora Vista Contamination Study, October 1986: Sante Fe, N.M., Environmental Bureau, New Mexico Oil Conservation Division, 55p.

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# OIL CONSERVATION COMMISSION

STATE OF NEW MEXICO 1000 RIO BRAZOS RD. - AZTEC 87410 LAND COMMISSIONER PHIL R. LUCERO



STATE GEOLOGIST EMERY C. ARNOLD

November 3, 1976

Monsanto Company 1330 Midland National Bank Tower Midland, Texas 79701

Re: Monsanto Company, NWP #1 K-23-30N-12W San Juan County

Gentlemen:

The attached Inspection Ticket shows that the subject well is venting gas at the wellhead. The gas surges to where a bull plug has been removed, which may indicate it to be coming through water.

Waste of gas cannot be tolerated.

Initiate action to repair or to plug this well.

If there are questions, please contact us.

Yours very truly,

Original Signed by A. R. Kendrick

A. R. Kendrick Supervisor District #3

Enclosures: Memo # 3-70, Inspection Ticket

xc: OCC, Santa Fe

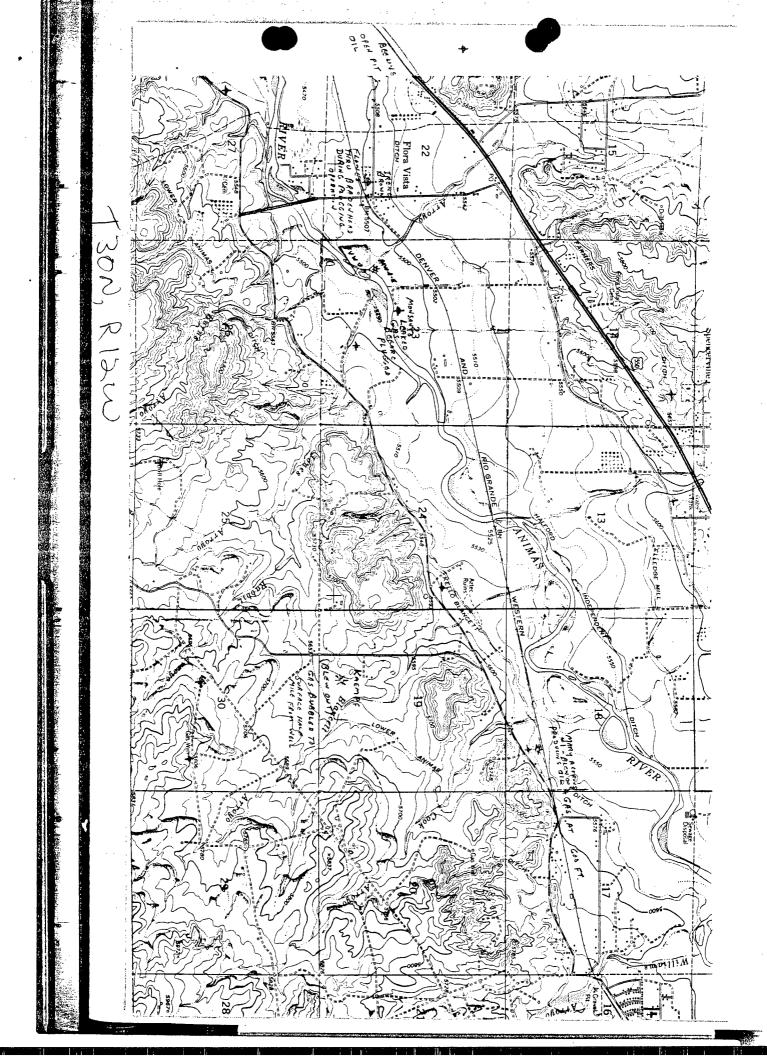
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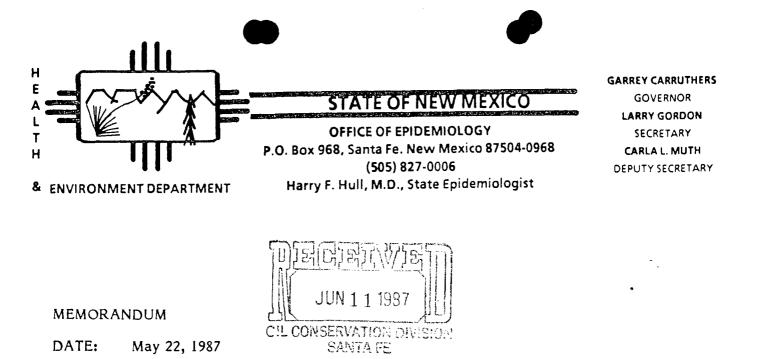
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DIRECTOR JOE D. RAMEY

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TO: Dennis McQuillan, Groundwater/Hazardous Wastes Bureau

FROM: Millicent Eidson, Environmental Epidemiologist m &

SUBJECT: USE OF FLORA VISTA COMMUNITY WATER SUPPLY WELL

I have been informed that the Flora Vista Community water supply well has been contaminated with petroleum, as evidenced by odor and a visible film on the water surface. It is the recommendation of the Office of Epidemiology that the well not be used, since we believe its use could pose a public health risk.

As discussed in Vol. 4 of <u>Drinking Water and Health</u>, National Academy Press, 1982, crude or refined oils may contain as many as one million different compounds. Thus it is difficult to assess toxicity precisely because of the number of compounds involved and the expense of laboratory sampling. However, the four major groups of compounds all have toxic effects. Intoxication with alkanes can result in transient depression of the central nervous system, or polyneuropathy with chronic exposure. Alkenes have weak anesthetic properties. The naphthenes act as general anesthetics and have depressant effects on the central nervous system. The aromatic hydrocarbons have many toxic effects including myelotoxicity, mutagenicity, and cancer.

cc: Stuart Castle, Water Supply Program

COUNTY OF SAN JUAN

# IN THE DISTRICT COURT

FLORA VISTA WATER USERS ASSOCIATION,

Plaintiff,

vs.

No. CV 86-00154-4

MANANA GAS, INC., and EL PASO NATURAL GAS COMPANY,

Defendants.

## NOTICE TO TAKE DEPOSITION

TO: MANANA GAS INC. -and-BYRON CATON, its Attorney 333 East Main Street Farmington, New Mexico 87401

NOTICE IS HEREBY GIVEN that Plaintiff will take the deposition of DAVID BOYER, upon oral interrogatories, in order to generate testimony for trial in lieu of live testimony, before a Certified Court Reporter and Notary Public from Santa Fe Deposition Service, at the hour of 10:00 o'clock a.m. on Wednesday, June 3, 1987, at the offices of Santa Fe Deposition Services, 1437 Paseo De Peralta, Santa Fe, New Mexico 87501. PLEASE BE THEN AND THERE PRESENT.

RICHARD L. LOUGEE Attorney at Law 108 North Orchard, Suite 201 Farmington, New Mexico 87401 (505) 327-5281

ENCLOSURE FOR YOUR INFORAJN Re: Flora Vista Water V. Manana Gas Our Your We enclose the following to keep you informed of the progress of this matter. Notice to Take Deposition
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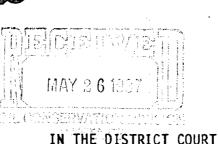
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COUNTY OF SAN JUAN

IN THE DESTRICT COURT

### FLORA VISTA WATER USERS ASSOCIATION,

Plaintiff,

vs.

No. CV 86-00154-4

MANANA GAS, INC., and EL PASO NATURAL GAS COMPANY.

Defendants.

### NOTICE TO TAKE DEPOSITION

TO: FLORA VISTA WATER USERS ASSOCIATION -and-RICHARD L. LOUGEE, its Attorney 108 North Orchard, Suite 201 Farmington, New Mexico 87401

NOTICE IS HEREBY GIVEN that Defendant Manana Gas, Inc., will take the deposition of <u>DAWIDEBOMER</u>, upon oral interrogatories, before a Certified Court Reporter and Notary Public from Santa Fe Deposition Service, at the hour of <u>NOTOCOM</u> <u>CONCLOCK amon Wednesday</u>, <u>June 3</u>, <u>1987</u>, at the offices of Santa Fe Deposition Services, 1437 Paseo De Peralta, Santa Fe, New Mexico 87501.

PLEASE BE THEN AND THERE PRESENT.

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BRIONES, ODENWALD & CATON, P.A. 333 East Main Street Farmington, New Mexico 87401 (505) 325-0258

200

Byron Caton Attorney for Defendant Manana Gas, Inc. State Bar Number 542

## IN THE ELEVENTH JUDICIAL DISTRICT COURT OF SAN JUAN COUNTY

CAUSE NO. CV 86-00154-4

#### FLORA VISTA WATER USERS ASSOCIATION,

Plaintiff,

vs.

MANANA GAS, INC., and EL PASO NATURAL GAS COMPANY,

Defendants.

#### SUBPOENA

 TO: DAVID BOYER
 525 Camino De Los Marquez Santa Fe, New Mexico

Greetings:

You and each of you are commanded to appear before a Certified Shorthand Reporter and Notary Public for the taking of your deposition on <u>Wednesday</u>, <u>May 27</u>, <u>198 7</u>, at 10:00 o'clock <u>a.m.</u>, at <u>Santa Fe Deposition Services</u>, located at 1437 Paseo De Peralta , <u>Santa Fe</u>, <u>New Mexico</u>.

And this do ye under the penalty of the law.

In witness whereof I have hereunto set my hand as clerk of said Court, and affixed the Seal of said District Court, on this <u>19th</u> day of May, 1987.

nagan Clerk of the District Court

COURT SEAL

By_____ Deputy Clerk

BRIONES, ODENWALD & CATON , P.A., Attorneys for <u>Defendant Manana Gas</u>, Inc. 333 East Main Street Farmington, New Mexico 87401 Telephone: (505) 325-0258

COUNTY OF SAN JUAN

IN THE DISTRICT COURT

### FLORA VISTA WATER USERS ASSOCIATION,

#### Plaintiff,

vs.

No. CV 86-00154-4

MANANA GAS, INC., and EL PASO NATURAL GAS COMPANY.

Defendants.

#### NOTICE TO TAKE DEPOSITION

TO: FLORA VISTA WATER USERS ASSOCIATION -and-RICHARD L. LOUGEE, its Attorney 108 North Orchard, Suite 201 Farmington, New Mexico 87401

NOTICE IS HEREBY GIVEN that Defendant Manana Gas, Inc., will take the deposition of DAVID BOYER, upon oral interrogatories, before a Certified Court Reporter and Notary Public from Santa Fe Deposition Service, at the hour of 10:00 o'clock a.m. on Wednesday, May 27, 1987, at the offices of Santa Fe Deposition Services, 1437 Paseo De Peralta, Santa Fe, New Mexico 87501.

PLEASE BE THEN AND THERE PRESENT.

BRIONES, ODENWALD & CATON, P.A. 333 East Main Street Farmington, New Mexico 87401 (505) 325-0258

Byron Cator

Attorney for Defendant Manana Gas, Inc. State Bar Number 542

I HEREBY CERTIFY THAT A TRUE AND CORRECT COPY OF THE FOREGOING WAS MAILED TO OPPOSING COUNSEL OF 19th DAY OF RECORD THIS _ May

Drew Bates Letter of 4/27/87 D what Kind of "Solubility" tested Son? D what coursed oper erolection in samples B-3, C-1, K & M; What Kind of test was used Sul Rate Analysis - what does it show? Why would "Stushing" remove sulfates This year and not previous yearly? In other works, why would there be a Change Since The pit has been board for 7 years? K, 10to ST/Day; actual more ment a (5) fotless. Sulfater not a problem, oil is Why M & D have hydrescorbond & not Text. SOUSNOM Bason indung mide





April 27, 1987

Flore Vista Water Users Association P. O. Box 171 Flore Vista, New Mexico 87415

Attention: Bert Barns

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The following are summarized lab results and recommendations from drilling 25+ pilot holes around Mary Wheeler AlE's gas well.

1) S#1, #2, #3 were samples taken at different depths 15' SW of Supply Well #4. These samples showed no solubility and feel that the hydrocarbon level is undetectable. S#3 sample was used as a base for this particular area. Samples = B-3, C-1, K, and M showed moderate to vigorous gas evolution for 20-30 seconds. All of these samples are in the reserve pit area and the gas released was not broken down except in sample M. Sample M gave off H₂S (Hydrogen Sulfide gas) rotten egg scent. The reasoning for the H₂S evolution for this sample over the other samples is due to the high concentration of hydrocarbons. No solubility tests have been performed on the second set of samples, but feel that the only possible samples that might have H₂S evolution is sample 0, see Fig. 7 in relation to sample M.

2) During the sulfate analysis of the first set of samples A-M, the sulfate count was entremely high as compared to samples N-V of the second set. The explanation for this, even though samples M and N are only 7 linear feet apart, the samples were taken 3 weeks spart. March, 1987, was an extremely cold month and the water table was lower than during the middle of April, 1987. Due to the permeability of the river rock and its high porosity of 25%, the system was probably flushed with the increase of runoff. Due to the high concentrations of sulfates recorded that a source for bacteria to thrive on in and around this area was present during the lessor flow periods fall-winter and then during the vast amount of runoff in the first part of April it had flushed the system. According to Boyer's report, that ground movement in this area is up to 100'/day, so to flush this area from the gas well to the supply well would take less than 3 days. These results can show the process of contamination from the reserve pit area to supply well \$1 into the water system. A corrective method in treating high sulfates is by using chlorine in the water system. Since the sulfate content has been reduced from 14,000 ppm down to less than 1000 ppm in three weeks during the first part of run-off these flushings might have brought an excessive amount of sulfates into the system which it was unable to handle.

Flora Vista Water Users Association April 27, 1987 Page 2

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3) Hydrocarbon test results were very noticeable in sample M from 5-7' depth ranging from 200-1000 ppm and also in sample 0 @ 7' ranging from 100-800 ppm. The high level of hydrocarbon detected from Richard Cheney's report did not exceed 23 ppm. Normal detection level is .01. Sample detection for representative sample No. A-M was less than 5 ppm and most of those second set of samples were less than .1 ppm. The level of hydrocarbons detected in samples M &O 100-1000+ ppm of  $C_6-C_{22}$  Shies very strongly away from the possibility of being dead animal decomposition. Most recent animal decomposition usually produces  $CO_2$  and Methane, something similar at a fertilizing plant. From the gas, peaks on the chromatograph chart and comparing to other oilfield hydrocarbon source areas showed similar results, especially in the heavy carbon chain levels  $C_7 - C_{10}$ . During high runoff times and the permeability level of the ground water beds that a good implementation of a charcoal filter will aid in absorbing the hydrocarbons out.

4) Bentonite tests were high for sample K which from visual sample description can be assumed to be drilling mud residual.

In summary, there is a drastic change in sulfate content between samples A-M and samples N-V resulting from the increased water runoff. Sulfates can be treated out of the water with chlorine, but if hydrocarbons are present then the possibility of chlorinated hydrocarbons can be formed. Chlorinated hydrocarbons if ingested are usually detoxified in the liver and can be classified as a carcinagen. A solution to both of these problems would be by implementing both a charcoal filter and utilizing ozone to purify the system. There are probably some State guidelines on this type of system. The samples that were taken are still in cold storage and additional tests can be taken.

Sincerely,

Inhew O. Beles

Drew Bates Engineering and Operations Manager

Date:	3-2	7-87
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Sudder Law ADDART S. M.

Mary Wheeler 1-E

Time Location 7:00 AM S4 @ SW of S4 Øi @ 2' #2 @ 4' #3 @ 6' #4 @ 8' 8:00 AM A Location Southeast corner toward river edge reserve pit 11 #1 C 2' #2 @ 5' #3 @ 7' water table 8:30 AM B Location 20' East of oil line in reserve pit 111 11 @ 3' Ø2 € 4' CMT Ø3 @ 5' #4 @ 7' frac sand 9:10 AM C Location SE 24' NE of oil line 1111 #1 @ 3' \$2 @ 4" 10:00 AM Location @ 85' East of wellhead edge of Reserve pit 10:45 AM D Location @ SW limit @ 2.5' 11:00 AM Location @ 11.5' East of B" East edge 1111 E @ #1 @ 3' E #2 @ 5'-7' 5' mud 7' water table 11:10 AM F Location @ 12.5' SE of separator pit 2' soil very unconsolidated river rock very permeable #1 € 3' #2 @ 5'-7' frac sand @ 6' water table @ 7' 12:00 PM G Location 8' South of separator pit south post corner #1 sample @ 2' Unable to go below 3' check for hydrocarbon 12:30 PM H Location 4' due South of Separator pit #1 @ 4* #2 @ 5' too rocky too many bits Shaley and very wet - little intermediate sand

Date: 3-27-87

# Mary Wheeler 1-E

Time	Location
12:44 PM	I Location 12' SW corner separator pit I #1 @ 3' I #2 @ 4' very unconsolidated and permeable unable to go below 4-1/2'
1:15 PM	J Location @ 38' SE of separator pit J #1 @ 4' J #2 @ 9' frac sand water table @ 7.5'
1:45 PM	K 45' SE of pit on surface Mud with hydrocarbons bogged down
2:00 PM	L 55' SE of separator pit #1 @ 3-5' #2 @ 7' water table
2:30 PM	M 75' South of separator pit @ 5-7' Threw drive line

# 4-16-87 @ 7:30 AM

# SAMPLING #

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N	@ 7' SE of M able to get down 5' sample taken @ 2' and 5' West edge of reservoir pit last of visible signs.
0	<pre>@ 20' East of M @ 2' @ 6' @ 7' still in reserve pit Black scent of drilling mud and possible hydrocarbons definitely not soil. 22' East of monitor well 7' definite hydrocarbons sample descriptions. Water table @ 7.5' Found combination lock @ 6'</pre>
P	<pre>@ 12' NE of monitor well @ 6.5' scent of hydrocarbons Black samples</pre>
Q	<pre>@ 17.5' East of P 24' NE of monitor well representative sample @ 6-7' no hydrocarbon clear sand check hydrocarbons no contamination.</pre>
R	<pre>@ 7' South of monitor well dark contaminated sample water table up to 3' in depth</pre>
S	@ 3', 5' water table 3'6" 8' West of drainage creek check sulfate and hydrocarbons.
T	@ 20' NE of S, @ 7' 3' water table Sulfates - Hydrocarbons
υ	@ 20' East of S, total 10' depth samples out
V	@ 25' SW of S, total 6' deep very permeable sulfates and hydrocarbons.

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1 1 1

Tech. Inc. 333 E. Main St. Farmington, NM 87401 (505)327-3311

No. is a

7 April 1987

Laboratory Analysis Report

alysis requested by: Mr. D. Bates Mr. Drew Bates Farmington NM 87401 Solubility test in 15% HC1 Sample #_ Analysis Result A#2 . .... -Solubility -Moderate gas evolution for 20_sec. A#3 Solubility Moderate gas evolution for 20 sec. B#3 Vigorous gas evolution for 30 sec. Solubility C# Solubility Vigorous gas evolution 30 sec. К Solubility Vigorous gas evolution 30 sec. Solubility Vigorous gas evolution for 30 sec. M М Hydrogen sulfide gas evolved. Very slight gas evolution. Solubility S#1 Very slight gas evolution. Solubility 3#2 Very slight gas evolution. 5#3 Solubility Analysis date: 5 April 1987 Sample Received: 28 March 1987 Analyst: H. P. Hamlow

Inc. Main St. mington, NM 87401 5)327-3311

7 April 1987

Laboratory Analysis Report

. . . . .

sis requested by: Mr. D. Bates Mr. Drew Bates Farmington_NM 87401 Analysis on Core Samples Sample #_____Analysis Result B#3_____Sulfate 24450

14 41 12 F. C

	B#3	Sulfate	2445Ø ppm	
	B#4	Sulfate	3620 ppm	
	C#1 7	Sulfate	52800 ppm	
	F#1	Sulfate	1810 ppm	
	F#2	Sulfate	450 ppm	
	G#1	Hydrocarbons	none or < 5 ppm	
	J	Hydrocarbons	none or < 5 ppm	
1	J#2	Hydrocarbons	none or < 5 ppm	
	an the second of the second	-Sulfate	3160	
	K	Hydrocarbons	none or < 5 ppm	
		Sulfate	24700 ppm	
	L#2	Hydrocarbons	none or < 5 ppm	
		Sulfate	7280 ppm	
	К	Hydrocarbons	none or < 5 ppm	
		Sulfate	24700 ppm	
	L#2	Hydrocarbons	none or < 5 ppm	
		Sulfate	7280 ppm	
	М	Eydrenarbonn	2:0:3-1(4:0:0) p.y.::	
			۶ مار در ۲۰۰۵ م ۲۰۰۵ مار در ۲۰۰۵ مار در ۲۰۰	

Tech. Inc. 333 E. Main St. Farmington, NM 87401 (505)327-3311

21 April 1987

# Laboratory Analysis Report

FloraVista Water Users 109 N. Orchard Farmington NM 87401

Analysis of dirt core samples.

Analysis requested by Mr. Drew Bates.

Sample #	Analysis	Result
N	Hydrocarbons	none or less than .1 ppm.
0@6'	Hydrocarbons	none or less than Ø.1 ppm
0 @ 7'	Hydrocarbons	1 <i>00-800</i> ppm.
P @ 6.5'	Hydrocarbons	none or less than Ø.1 ppm
ଭ	Hydrocarbons	none or less than Ø.1 ppm
R	Hydrocarbons	none or less than Ø.1 ppm
υ	Hydrocarbons	none or less than Ø.1 ppm
V	Hydrocarbons	none or less than Ø.1 ppm

Analysis date: 20 April 1987

Sample Received:16 April 1987

Analyst:H. P. Hamlow

ab # 2509-949

CLIPPO SPRO

NTBK# 1-81-AT

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Tech. Inc. 333 E. Main St. Farmington, NM 87401 (505)327-3311

21 April 1987

Laboratory Analysis Report

Flora Vista Water Users 108 N. Orchard Farmington NM 87401

Analysis of dirt core samples.

Analysis requested by Drew Bates.

Sam <b>ple #</b>	Analysis	Result
N	Sulfate	760 ppm
0 8 6'	Sulfate	108 ppm
0 0 7'	Sulfate	162 ppm
P@ 6.5'	Sulfate	128 ppm
Q	Sulfate	172 ppm
R	Sulfate	84 ppm
υ	Sulfate	172 ppm
V	Sulfate	240 ppm
Analysis date: 21	April 1987	

Sample Received:16 April 1987

Analyst: H. P. Hamlow

Jab # 2509-949

Server Methods

UNTBK# 2-81

SUS LABORATORIES

A DIVISION OF CASA DEL SOL, INC. 1474 MAIN AVENUE #131 POST OFFICE BOX 2805 DURANGO, COLORADO \$1301

(303) 247-4220

CESUSX

L.A.B. NG 23 83

Date: 8-1-83

To: Flora Vista Water User Attn. Richard Cheyney 909 W. Apache Farmington, N.M. 87401

> Sample ID 2456 GALLONS JUG 2359 OLD PIT 2454 BOST OR FIT 2455 South OF FIT



Normal detection Limit 0.01

1 .....

Analytical results
 0.70 mg Total Hydrocarbons
 7.50 mg Total Hydrocarbons
 8.10 mg Total Hydrocarbons
 4.00 mg Total Hydrocarbons

CDS Lab ID Ø

Sample Description:

SO AL CAMPLE

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CONCERT	TINTION BASE	ets on p	el		
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24:6	1.20	Mail	a.v.		•
2455	1.21	11.12	tr	• •	•••
2454	23.14	Mill	, <b>r</b>	. <b></b>	••

Analyte

C₆-C₂₂

C6-C22

 $C_{6} - C_{22}$ 

C6-C22

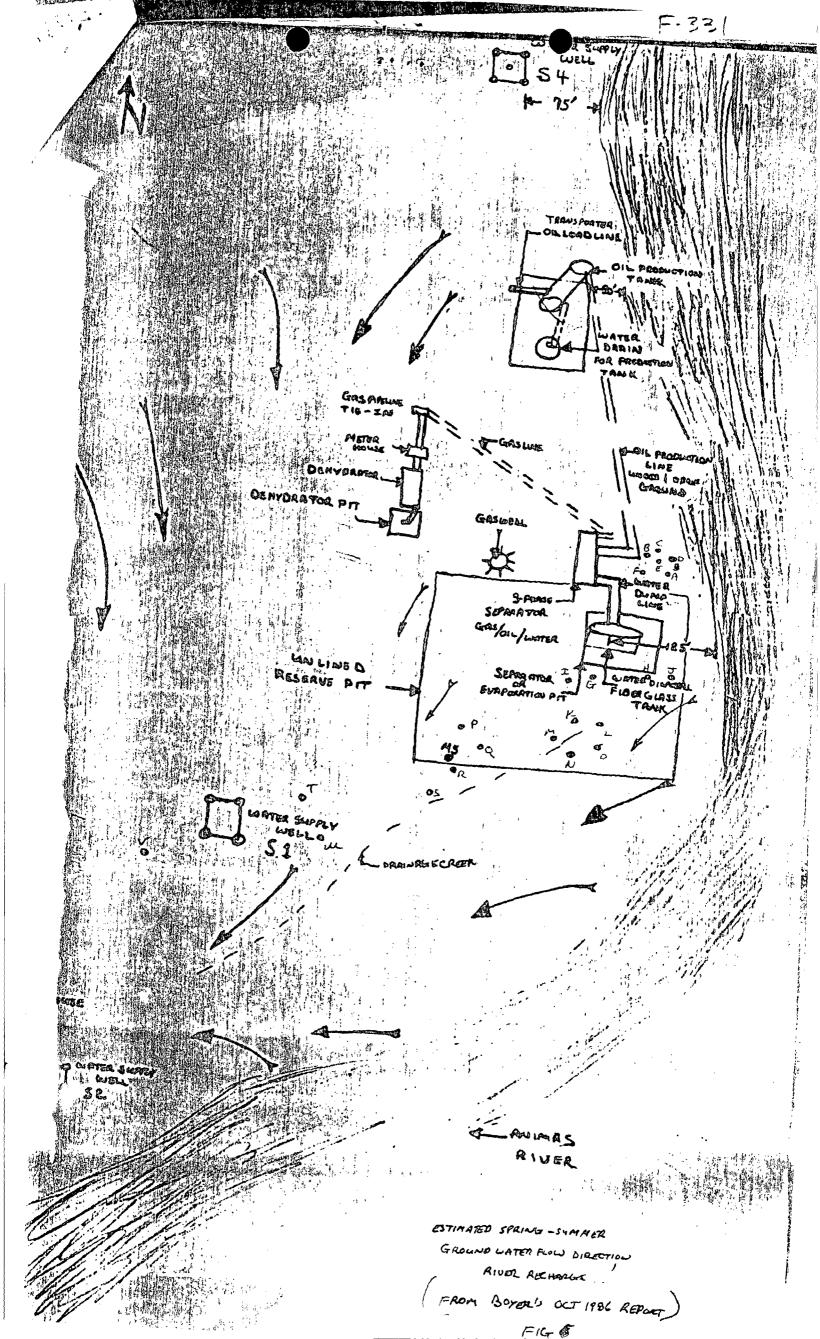
SER AMMENDED REPORT 8/34/83

THIS LABORATORY REPORT MAY NOT BE PUBLISH ED OR USED FOR ADVERTISING OR IN CONNECTION WITH ADVERTISING OF ANY KING WITHOUT PRIOR WRITTEN PERMISSION FROM COS LABORATORIES RESULTS ARE BASED ON ANALYSIS MADE AT THE TIME GAMPLES ARE RECEIVED AT LABORATORY.

Dr. Joe Bowden

Director

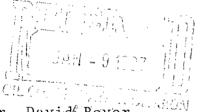
Members of: American association for the advancement of science American society of biological chemists American society for affiliation Sigma XI



January 7, 1987 Re: Flora Vista Water v. Manana Gas et al

ENCLOSED PLEASE FIND:

Notice to Take Deposition of David Boyer



Mr. David Boyer New Mexico Energy and Minerals Dept. Oil Conservation Division P.O. Box 2088 Santa Fe, New Mexico 87501

Richard L. Lougee Attorney at Law 108 N. Orchard Suite 201 Farmington, New Mexico 87401 505-327-5281

COUNTY OF SAN JUAN

1

i.

IN THE DISTRICT COURT

FLORA VISTA WATER USERS ASSOCIATION,

Plaintiff,

vs.

No. CV 86-00154-4

MANANA GAS, INC., and EL PASO NATURAL GAS COMPANY,

Defendants.

### NOTICE TO TAKE DEPOSITIONS

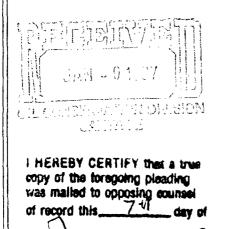
TO: DAVID BOYER

PLEASE TAKE NOTICE that the deposition of David Boyer will be taken at the State Land Office Building, 210 Old Santa Fe Trail, Room 206, Santa Fe, New Mexico on the 15th day of January, 1987 at 1:30 p.m. before a certified court reporter. PLEASE BE THEN AND THERE PRESENT.

hel and & Longe

RICHARD L. LOUGEE Attorney for Plaintiff 108 N. Orchard, Ste. 201 Farmington, NM 87401 (505) 327-5281

Canceller



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ENERGY AND MINERALS DEPARTMENT OIL CONSERVATION DIVISION

TONEY ANAYA GOVERNOR

November 18, 1986

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87501-2088 (505) 827-5800

CERTIFIED MAIL RETURN RECEIPT REQUESTED

Mr. Ed Hartman Manana Gas, Inc. P. O. Box 36990 Albuquerque, N.M. 87176

Dear Mr. Hartman:

The New Mexico Oil Conservation Division (OCD) has completed a report on the activities through May, 1986, of this Division and the Environmental Improvement Division (EID) regarding the contamination investigation of Flora Vista Water Well No. 1 (S1). This well, which was contaminated by oil and grease in early 1983, has been out of service since that date. OCD activities in 1985 included installation of five monitoring wells, sampling of water quality from these wells and other ground water at this location, and measurement of water levels to determine ground water direction and rate of flow. EID supplied material and staff to assist in well installation. Activities in 1986, all by the OCD, included additional sampling, water level measurements, an aquifer test to determine aquifer characteristics, and report preparation.

The OCD report is enclosed with this letter, but its major conclusions are summarized below:

- 1. Water Well No. 1 (S1) remains contaminated by oil. Oil remains in sediments immediately adjacent to the well, and any production pumping will cause continued movement of contaminants into the well.
- 2. Because of dilution and volatilization of dissolved hydrocarbon constituents in the wellbore, OCD samples did not show levels exceeding health standards. However, if only for esthetic reasons, oil is not acceptable in drinking water systems.
- 3. Zone of capture calculations based on aquifer test results show that pumping of the well has drawn and will continue to draw water and associated contaminants from the area of the Manana, Inc., Mary Wheeler No. 1E gas well.
- 4. The source of the oil is past activities at the Mary Wheeler No. 1E gas well. This determination is based on scientific information that is detailed in the report. The exact mechanism

of discharge (e.g., leaky produced water tank, oil storage tank drain pit, reserve or blowdown pit, EPNG dehydrator upsets, or other discharges) could not be established since OCD does not have records of upsets, equipment malfunctions or reserve pit discharges. The contamination may be a combination of one or more of these mechanisms.

- 5. Capture curves for well S5 show that regular pumping of that well would draw contaminants from the area of S1.
- 6. Information from the monitor wells show that natural ground water flow is generally from north to south. Ground water from the area of the gas well facility will not intercept any supply wells under non-pumping conditions. However, pumping cones of depression and capture zones will determine the distance and direction from which contaminants are drawn. These have not been determined for the currently used supply wells, and the threat of contamination (if any) cannot be judged.

This report was prepared using public funds, and will be available to any party upon request. All supporting documentation, calculations, sample analyses, etc., remain on file at the OCD office in Santa Fe. With completion of data analysis, and transmittal of the report, the OCD has essentially completed its study of the contamination.

The Oil Conservation Division is charged with regulating oil and gas production discharges so as to protect fresh water. Prior to 1985, only limited oil and gas rules applicable to fresh water protection were in effect in the San Juan Basin. The OCD is researching this matter and no decisions have been made regarding future OCD actions, legal or otherwise, that may be taken to remediate the problem at the site. Before proceeding in any course of action, we will be in contact with you to discuss various options.

In the meantime, I am available to discuss the report and provide suggestions as to further investigation and remedial action that you may wish to pursue. If you have any questions, you may contact me at 827-5812.

Sincerely, wind ARoang

DAVID G. BOYER // // Hydrogeologist/Environmental Bureau Chief

DGB:dp

cc: Paul Biderman, Secretary, EMD R. L. Stamets, Director, OCD Jeff Taylor, OCD-Legal NM OCD, Aztec District Office

# ENERGY AND MINERALS DEPARTMENT

**DIL CONSERVATION DIVISION** 



TONEY ANAYA GOVERNOR

November 18, 1986

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87501-2088 (505) 827-5800

CERTIFIED MAIL RETURN RECEIPT REQUESTED

Dr. Howard Reiquam Director, Environmental Affairs El Paso Natural Gas Company P. O. Box 1492 El Paso, Texas 79978

Dear Dr. Reiquam:

The New Mexico Oil Conservation Division (OCD) has completed a report on the activities through May, 1986, of this Division and the Environmental Improvement Division (EID) regarding the contamination investigation of Flora Vista Water Well No. 1 (S1). This well, which was contaminated by oil and grease in early 1983, has been out of service since that date. OCD activities in 1985 included installation of five monitoring wells, sampling of water quality from these wells and other ground water at this location, and measurement of water levels to determine ground water direction and rate of flow. EID supplied material and staff to assist in well installation. Activities in 1986, all by the OCD, included additional sampling, water level measurements, an aquifer test to determine aquifer characteristics, and report preparation.

The OCD report is enclosed with this letter, but its major conclusions are summarized below:

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- 2. Because of dilution and volatilization of dissolved hydrocarbon constituents in the wellbore, OCD samples did not show levels exceeding health standards. However, if only for esthetic reasons, oil is not acceptable in drinking water systems.
- 3. Zone of capture calculations based on aquifer test results show that pumping of the well has drawn and will continue to draw water and associated contaminants from the area of the Manana, Inc., Mary Wheeler No. 1E gas well.
- 4. The source of the oil is past activities at the Mary Wheeler No. 1E gas well. This determination is based on scientific information that is detailed in the report. The exact mechanism

of discharge (e.g., leaky produced water tank, oil storage tank drain pit, reserve or blowdown pit, EPNG dehydrator upsets, or other discharges) could not be established since OCD does not have records of upsets, equipment malfunctions or reserve pit discharges. The contamination may be a combination of one or more of these mechanisms.

- 5. Capture curves for well S5 show that regular pumping of that well would draw contaminants from the area of S1.
- 6. Information from the monitor wells show that natural ground water flow is generally from north to south. Ground water from the area of the gas well facility will not intercept any supply wells under non-pumping conditions. However, pumping cones of depression and capture zones will determine the distance and direction from which contaminants are drawn. These have not been determined for the currently used supply wells, and the threat of contamination (if any) cannot be judged.

This report was prepared using public funds, and will be available to any party upon request. All supporting documentation, calculations, sample analyses, etc., remain on file at the OCD office in Santa Fe. With completion of data analysis, and transmittal of the report, the OCD has essentially completed its study of the contamination.

The Oil Conservation Division is charged with regulating oil and gas production discharges so as to protect fresh water. Prior to 1985, only limited oil and gas rules applicable to fresh water protection were in effect in the San Juan Basin. The OCD is researching this matter and no decisions have been made regarding future OCD actions, legal or otherwise, that may be taken to remediate the problem at the site. Before proceeding in any course of action, we will be in contact with you to discuss various options.

In the meantime, I am available to discuss the report and provide suggestions as to further investigation and remedial action that you may wish to pursue. If you have any questions, you may contact me at 827-5812.

Sincerely,

Jane & Borgh

DAVID G. BOYER Hydrogeologist/Environmental Bureau Chief

DGB:dp

cc: Paul Biderman, Secretary, EMD R. L. Stamets, Director, OCD Jeff Taylor, OCD-Legal NM OCD, Aztec District Office



ENERGY AND MINERALS DEPARTMENT

OIL CONSERVATION DIVISION

TONEY ANAYA GOVERNOR

November 17, 1986

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87501-2088 (505) 827-5800

MEMORANDUM

TO: ADDRESSEES

FROM: DAVID BOYER, HYDROGEOLOGIST, OIL CONSERVATION DIVISION

SUBJECT: CORRECTED PAGE 3, FINAL REPORT ON FLORA VISTA CONTAMINATION STUDY

The enclosed sheet replaces pages 3 and 4 of the above report. Page 3 is corrected to show that a study progress report was prepared in January, 1986. Please insert the corrected page in your copy of the report.

DB:dp

Enc.

- 2) Samples were taken from the water supply and monitor wells for organic analyses in March, June, August, September, and October, 1985. The results are shown in Tables 1 and 2. Results of samples from the fiberglass separator tank at the Mary Wheeler No. 1E gas well are shown in Table 3. Inorganic analyses were run on the water supply wells (including the previously contaminated well) and the Animas River in March, and the produced water from the gas well in October.
- 3) As expected, several monitor wells filled with sand and were cleaned twice using compressed air from two different compressors. The use of these compressors apparently affected the water quality as discussed below.
- 4) In September all monitor and supply wells were surveyed by Brewer and Associates for both location and elevation. Several well elevations were resurveyed in October. Blueprints from recent aerial photos were received in late December. Water levels were measured in September and October, 1985.

A summary of work performed in 1986 follows:

- 1) Water levels were measured in January, February, April and May. These are shown in Figures 3 to 6.
- 2) Samples from the monitoring wells for water quality analysis were taken in January, and April. The contaminated well was sampled in April and May. The water system was sampled for organics in January, April and May. The new fiberglass tank at the Mary Wheeler 1E oil storage tank (used to collect water drained from that tank) was sampled in April and May. The 55-gallon drum at the El Paso Natural Gas dehydrator unit was sampled in January and February. The results of organic analyses of these samples are shown in Tables 1 to 4. Inorganic analyses were run on samples from the contaminated water well in April, and from the dehydrator barrel in February. Discussion on the results is presented in the section on water quality.
- 3) A 72-hour aquifer test was performed on the contaminated well between April 21 and 25. The test consisted of water level measurements from the pumped well and monitor wells on April 21, 48 hours of pumping April 22-24, and 24 hours of recovery April 24-25. Approximately 5 hours into the test, oil was drawn into the well. The results of this test are discussed in detail in the hydrogeology and water quality sections.
- 4) A second, short duration test was performed in May to better characterize the volume and nature of the oil. These results are also presented below.
- 5) A study progress report was prepared in January.

-3-

#### Hydrogeology

The valley of the Animas River contains alluvium consisting mainly of sand and gravel which is outwash material from Pleistocene glaciers in the San Juan Mountains in Colorado. In the vicinity of the Flora Vista wells this alluvium is about 22 to 25 feet thick. Examination of the aerial photograph blueprints provided by Brewer & Associates shows old river channels and meanders in the flood plain. Finer grained silts and clays can be expected to have been deposited in low velocity areas such as point bars and areas of overbank flooding. However, the area where the monitor wells were drilled was found to be a zone of very coarse sand and gravel with some rocks exceeding a foot in diameter.

The presence of a coarse sand and gravel zone usually indicates high ground water permeability. Examination of the pumping level estimates provided in the 1982 EID community water system environmental survey, together with a 1982 Brewer and Associates infiltration gallery feasibility study shows an aquifer permeability of about 750 gallons per day per square foot, or 100 feet per day. This was confirmed by the aquifer test conducted in April. The value is at the lower end of the range for clean sand and gravel mixtures, but still allows for rapid ground water movement.

In September and October, 1985, ground water levels were measured in those monitoring wells where fluid levels were present. The results were used to calculate the direction of ground water flow and the hydraulic gradient. The gradient values were 0.0080 and 0.0081, respectively, or about 43 feet per mile. These values are intermediate between the average river gradient at Flora Vista of 0.004 and the topographic gradient of 0.014 perpendicular to the river at the well field location. The 1986 water levels, measured in January, February, April, and May, showed gradients of 0.0080, 0.0086 and 0.0079, and 0.0071 respectively. The average of the six measured values is 0.0080 or about 42 feet per mile.

The 1985 ground water flow directions are shown on Figures 1 through 2. The direction in September is slightly east of south. By late October, the direction had changed to nearly 20° east of south and continued that direction through January and February (Figures 3 and 4). The April and May measurements (Figure 5 and 6) again show the direction of flow as slightly east of south.

Some reasons for these observed changes in the ground water flow direction may be postulated based on surface and ground water interaction in the area. When river flows are generally low, as in the fall and winter, water stored in the permeable alluvial material in the immediate vicinity of the river during times of spring and summer high flows is discharged back into the river. Additional ground water discharge to the river comes from sources to the northwest of the well field including ground water recharged to the alluvium from the Flora Vista Arroyo, the Halford Independent Ditch, the Farmers Ditch,



ENERGY AND MINERALS DEPARTMENT 525 Camino de los Marquez Santa Fe, New Mexico 87501

STATE OF NEW MEXICO

TONEY ANAYA GOVERNOR

October 14, 1986

MEMORANDUM

TO: DICK STAMETS AND DAVE BOYER, OCD

FROM: PAUL L. BIDERMAN, SECRETARY

SUBJECT: FLORA VISTA WELL INVESTIGATION

Dave, thanks to you and the staff for your persistence in finishing the investigation of the Flora Vista well contamination. While you weren't able to answer every question, you have certainly succeeded in shedding light on some of the crucial issues needed to proceed further.

I would like both of you to let me know your thinking at this time over whether the state should sue or leave the matter to any private litigation that may develop. Please brief me soon on the available options so we can decide what further steps if any are appropriate. Again, thanks for all the work thus far in this difficult problem.

PLB:rm

OFFICE OF THE SECRETARY (505) 827-5950

MINING & MINERALS DIVISION

(505) 827-5970

ADMINISTRATIVE SERVICES DIVISION CONSERVATION & MANAGEMENT DIVISION (505) 827-5925

OIL CONSERVATION DIVISION (505) 827-5800 Land Office Building, P.O. Box 2088, Senta Fe, New Mexico 87501

(505) 827-5860

RESOURCE & DEVELOPMENT DIVISION (505) 827-5900



# ENERGY AND MINERALS DEPARTMENT

OIL CONSERVATION DIVISION

GOVERNOR

October 9, 1986

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87501-2088 (505) 827-5800

Mr. Bert Barnes, President Flora Vista Water Users Assoc. P. O. Box 171 Flora Vista, New Mexico 87415

Mr. Richard P. Cheney, Vice Pres. Lawrence A. Brewer & Assoc. Inc. P. O. Box 2079 Farmington, New Mexico 87499

Gentlemen:

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The New Mexico Oil Conservation Division (OCD) has completed a report on the activities through May, 1986, of this Division and the Environmental Improvement Division (EID) regarding the contamination investigation of Flora Vista Water Well No. 1 (S1). This well, which was contaminated by oil and grease in early 1983, has been out of service since that date. OCD activities in 1985 included installation of five monitoring wells, sampling of water quality from these wells and other ground water at this location, and measurement of water levels to determine ground water direction and rate of flow. EID supplied material and staff to assist in well installation. Activities in 1986, all by the OCD, included additional sampling, water level measurements, an aquifer test to determine aquifer characteristics, and report preparation.

The CCD report is enclosed with this letter, but its major conclusions and recommendations for further study are summarized below:

- 1. Water Well No. 1 (S1) remains contaminated by oil. Oil remains in sediments immediately adjacent to the well, and any production pumping will cause continued movement of contaminants into the well.
- 2. Because of dilution and volatilization of dissolved hydrocarbon constituents in the wellbore, OCD samples did not show levels exceeding health standards. However, if only for esthetic reasons, oil is not acceptable in drinking water systems.
- 3. Zone of capture calculations based on aquifer test results show that pumping of the well has drawn and will continue to draw water and associated contaminants from the area of the Manana, Inc., Mary Wheeler No. 1E gas well.

- 4. The source of the oil is past activities at the Mary Wheeler No. 1E gas well. The exact mechanism of discharge (e.g., leaky produced water tank, oil storage tank drain pit, reserve or blowdown pit, EPNG dehydrator upsets, or other discharges) could not be established since OCD does not have records of upsets, equipment malfunctions or reserve pit discharges. The contamination may be a combination of one or more of these mechanisms. This determination is based on scientific information that is detailed in the report.
- 5. Capture curves for well S5 show that regular pumping of that well would draw contaminants from the area of S1. Short test pumping of well S5 should be conducted to determine contamination near that well. It should then be used as a monitor well.
- 6. Information from the monitor wells show that natural ground water flow is generally from north to south. Ground water from the area of the gas well facility will not intercept any supply wells under non-pumping conditions. However, pumping cones of depression and capture zones will determine the distance and direction from which contaminants are drawn. These have not been determined for the currently used supply wells, and the threat of contamination (if any) cannot be judged.

This study has not received peer review by other qualified hydrogeologists. While I stand by all results and conclusions in the report, it might be to your benefit to have another professional hydrogeologist review it prior to litigation. All supporting documentation, calculations, sample analyses, etc., remain on file at the OCD office in Santa Fe.

This report was prepared using public funds, and will be available to any party upon request. This includes examination of the supporting material described above. With completion of data analysis, and transmittal of the report, the OCD has essentially completed its study of the contamination. We will be available to answer questions and provide limited assistance, but esentially, this report completes our investigative work at the site. The OCD still has the option of litigation, but no decisions have been made to proceed in this area.

This study has involved considerable OCD time and resources, much more than originally anticipated. Besides being useful in determining current contamination and its source, the study demonstrated both what can be accomplished with knowledgeable professional staff, and the difficulties in providing timely completion of the study given the myriad of other duties and responsibilities of the single staff hydrogeologist. The study was undertaken during a transition period when the OCD was attempting to demonstrate that it could handle the responsibility for protecting ground water that may be threatened by the industry it regulates. The record of accomplishment the past two years proves that OCD can be effective in this area.

Page 2

Page 3

For the future, consideration will be given to OCD undertaking enforcement action directly after making a sufficient determination of the cause of contamination and the likely responsible party. The Environmental Improvement Division has taken the lead in this area by negotiating "Settlement Agreements" with those persons who have contaminated ground water. These can be used before or after lawsuits are filed, but expensive and lengthy litigation is avoided. EID settlements have provided for cleanup and ground water restoration among other things. They are commonly used for handling leaks from underground gasoline storage tanks. At this time, no decision has been made by OCD on using this approach for the Flora Vista contamination incident.

Again, I remain available to discuss the report, and my apologies at not completing it sconer. You may contact me at 827-5812.

Sincerely,

DAVID G. BOYER ' / Hydrogeologist/Environmental Bureau Chief

DGB:dp

cc: Paul Biderman, Secretary, EMD
R. L. Stamets, Director, OCD
NM OCD, Aztec District Office
NM EID Water Supply Section
NM EID Ground Water/Hazardous Waste Bureau
NM EID, Farmington Field Office

STATE OF NEW MEXICO



ENERGY AND MINERALS DEPARTMENT

OIL CONSERVATION DIVISION

TONEY ANAYA GOVERNOR September 23, 1986

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87501-2088 (505) 827-5800

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Mr. Richard P. Cheney Vice President Lawrence A. Brewer & Assoc., Inc. P. O. Box 2079 Farmington, New Mexico 87401

Dear Mr. Cheney:

This letter is in reference to our phone conversation this morning concerning the El Paso Natural Gas Company dehydrator unit at the Flora Vista Mary Wheeler No. 1-E gas well, Unit M, Section 23, Township 30 North, Range 12 West. The actual measured amount of steam vapor condensation liquids from the dehydration unit is unknown since the OCD has no records of the discharges. However, a rough estimate might be one 55-gallon barrel per 30 to 60 days. This volume would be dependent on the periods the well is operated and on the efficiency of the unit. These fluids contain elevated levels of aromatic hydrocarbons, but very little oil.

Enclosed is a 1984 sketch from Frank Chavez, OCD-Aztec District Supervisor, of the sources of produced water (and other fluids) at a single well site. The figure is general and not specific to the Flora Vista site. According to Mr. Chavez, at Flora Vista, the water drain line from the dehydrator separator previously went to the dehydrator pit or the 55-gallon drum. If the Manana primary separator was not working properly, oil, water and gas would be caught at the dehydrator and oil and water dumped to the drum or pit. Since the gas well produces at least several 42-gallon barrels of water per day, the 55-gallon drum would certainly overflow to the pit if the Manana equipment malfunctioned. Mr. Chavez says that at the present time the piping has been rerouted so that upsets go to the oil storage tank instead of the drum or pit. OCD files do not contain any records of the dates or volumes of any upsets.

Since I am not directly familiar with either the specifics of operation of dehydrator units, or the physical arrangement of equipment and piping at the Flora Vista dehydrator, I suggest that you contact Mr. Chavez or other persons knowledgeable in gas well production operations.

Sincerely,

DAVID G. BOYER Environmental Bureau Chief/Hydrogeologist

DGB:dp

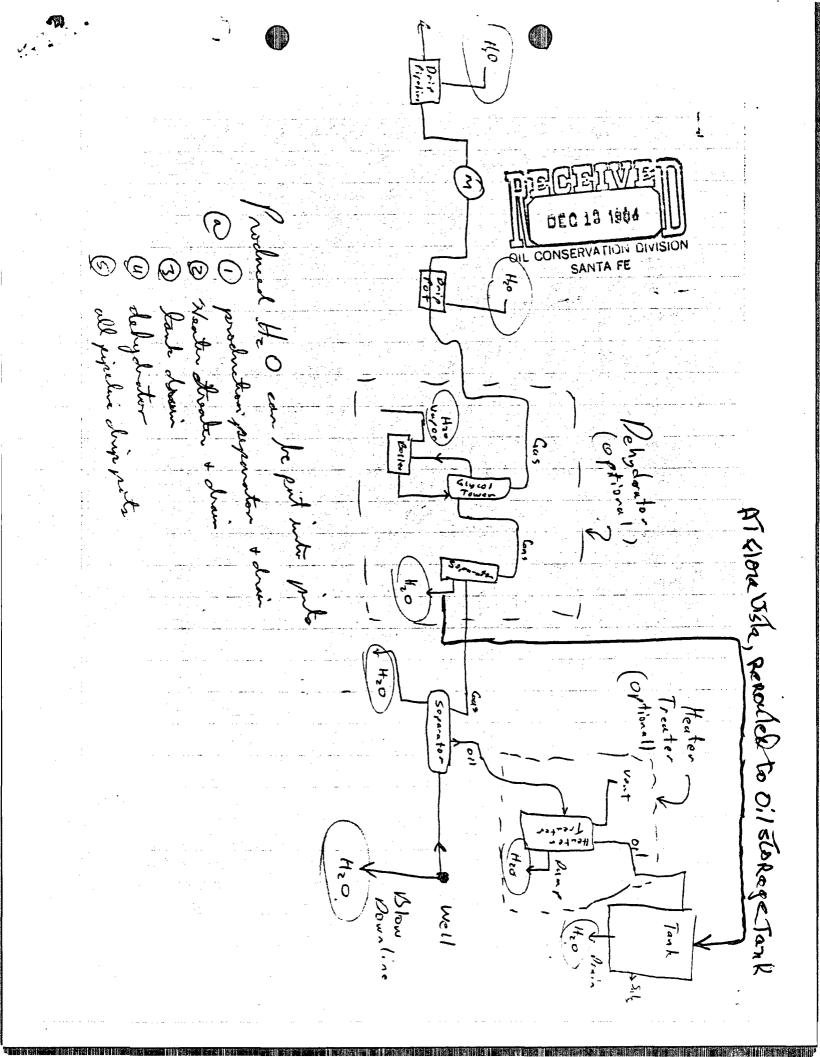
Enclosure

cc: R. L. Stamets Frank Chavez

Memo FRANK T. CHAVEZ District Supervisor To Dave Doyen This is a shetch of how H20 I can be collected E discharged at atleast 7 places from one well site

**Oil Conservation** 

Aztec, New Mexico



Note to file - EPNG- Mary Wheeler Calendation Assume one "drip" of '2 ml every 15 soc (Let & then a leaky Sauce ) Then : 1<u>ml</u> × <u>bosec</u> × 1440min × <u>k</u> × 0.976gt Get =0.689p2 30 sec Min Day 1000ml & 4975 of 20 gold / 30 days, 40 gal /60 days More if sosterdrip, rein in barel. Less if well off, dehydrator efficient. By comparision - Worst case seen by me: A. Dehydrator at Leeper 601 Comm #1 (Celler Hill at Rever) on 5/2/85 Collected 250 ml drip in bottle in 22 Sect on 232 gallous / day A FRough 9/23/867

STATE OF NEW MEXICO



ENERGY AND MINERALS DEPARTMENT

OIL CONSERVATION DIVISION

TONEY ANAYA GOVERNOR

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September 4, 1986

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87501-2088 (505) 827-5800

Mr. Richard P. Cheney Vice President Lawrence A. Brewer & Associates, Inc. P. O. Box 2079 Farmington, N.M. 87401

Dear Mr. Cheney:

This letter is in reference to the final report on the Flora Vista well contamination which I had said would be finished by this week. It isn't, and I apologize and take responsibility for the failure to complete it within the agreed upon time. I don't believe in making commitments I can't keep; its unprofessional. However, in this case, I have badly overextended myself. Your report is second to another one (which was due August 14) nearing completion. I estimate I am four weeks away (maximum) from completing yours. This will include my working evenings and weekends (which my supervisor disapproves of) until completion.

The main problem is that I am the only hydrogeologist the Oil Conservation Division has to perform groundwater-related permitting and contamination investigation statewide. I have a chemical engineer and petroleum geologist that assist in hydrologic matters to the extent their expertise allows. I am also responsible for administrative duties as supervisor and Environmental Bureau Chief. The attached material summarizes these responsibilities and the workload. (This excerpt is from an EPA grant request; a six-month EPA funded technical position was approved starting this month.)

While this letter does nothing to advance your report, I hope it helps to put the matter in perspective. Again, my apologies to you, and my personal commitment for a finished report within 30 days.

Sincerely,

DAVID G. BOYER Hydrogeologist/Environmental Bureau Chief

Attachment

DGB:dp

cc: R. L. Stamets Paul Biderman

Section 1. GENERAL INFORMATION	٠
(A) Owner of well Flora Vista Water Users Association Owner's Well No.	
Street or Post Office Address PQ, Box 171 City and State Florg Visto, N.M. 874-15	
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(B) Dulling Contractor Bob SAVAGE License No. WD-847	
Address P.O. Box 2434 FARMINGTON, NM. 87401	
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Tom Barker

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FOR USE OF STATE ENGINEER ONLY

Date Received

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Section 7. REMARKS AND ADDITIONAL INFORMATION

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The undersigned hereby certifies that; to the best of his knowledge and belief, the foregoing is a true and correct record of the above described hole.

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MUSTRUCTIONS: This form should be executed in triplicate, preferably typewritten, and submitted to the appropriate district office , the State Engineer. All sections, opt Section 5, shall be answered as completely and accurately as preschie when any well is , the State Engineer. All sections, opt Section 5, shall be answered as completely and accurately as preschie when any well is , the State Engineer. All sections, opt Section 5, shall be answered as completely and accurately as preschie when any well is , the State Engineer. All sections, opt Section 5, shall be answered as completely and a securately as preschie when any well is , the State Engineer. All sections, opt Section 5, shall be answered as a plugging record as a plugging record as a plugging record as a section of the sections of the sections of the section of

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## STATE ENGINEER OFFICE

# WELL RECORD

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Revised June 1972

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#### Revised June 1972

# STATE ENGINEER OFFICE

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# WELL RECORD

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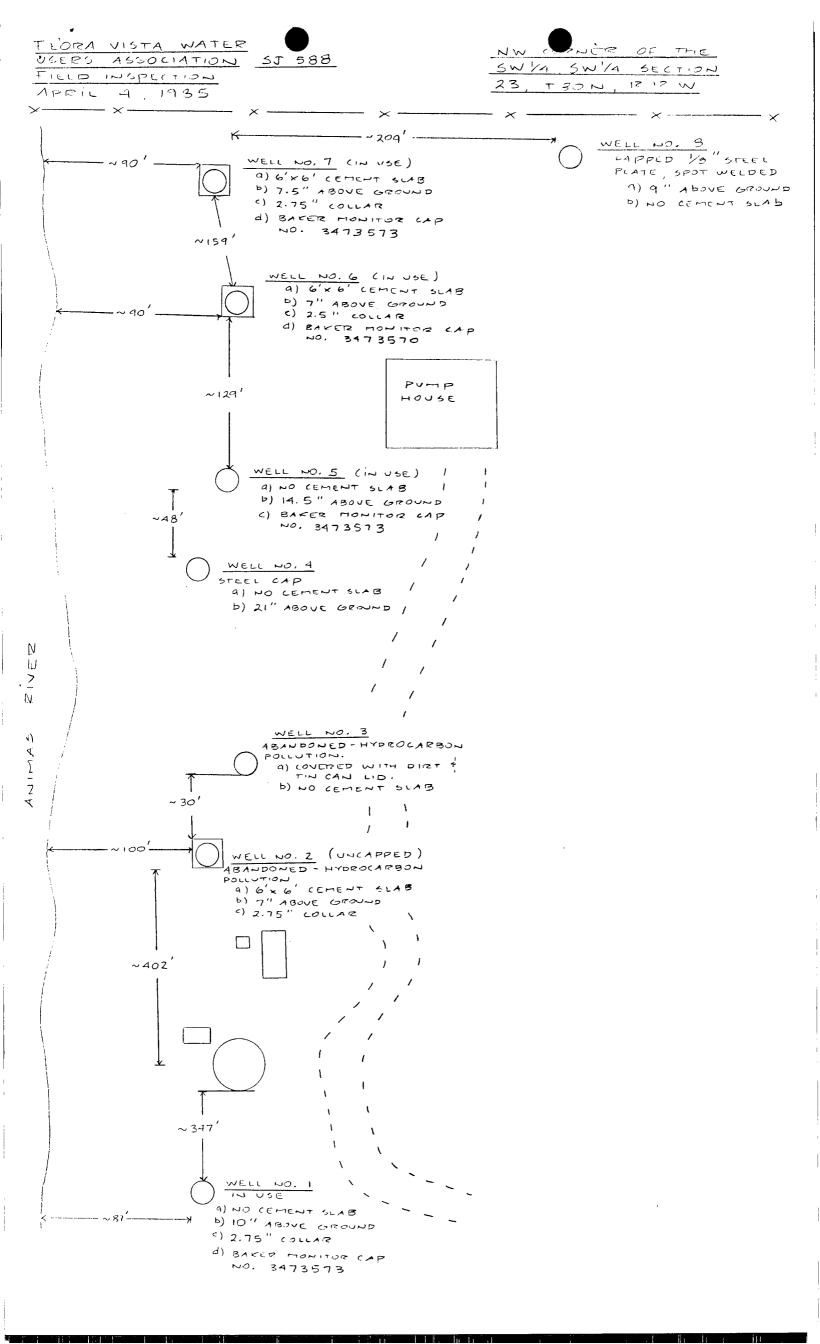
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ACTIONS: This form should be cuted in triplicate, preferably ty pewritten, and multicle to the appropriate district office affect fragmentates, as possible of the suswored as completely as accurately as possible of the suswored as completely as possible of the suswored as completely as possible of the suswored as completely as possible of the suswored by the fraction of the suswored by the fraction of the suswored as completely as possible of the suswored by the fraction of the suswored by the suswored by



50 YEARS



STATE OF NEW MEXICO ENERGY AND MINERALS DEPARTMENT OIL CONSERVATION DIVISION



GOVERNOR

April 9, 1986

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE. NEW MEXICO 87501 (505) 827-5800

Mr. Richard P. Cheney Vice President Lawrence A. Brewer & Associates, Inc. P. O. Box 2079 Farmington, N.M. 87401

Dear Mr. Cheney:

As mentioned previously by letter and in person, the Oil Conservation Division wishes to conduct an aquifer pumping test on the Flora Vista Well No. 1 that was previously contaminated. We have tentatively scheduled the entire week of April 21-25 for work in Farmington, mainly at the Flora Vista site. Before final arrangements are made, I need to know if all the equipment needed to conduct the test is available. The following are our anticipated requirements:

- 1. Installation of a pump in Well No. 1 with a capacity near that of the original pump, or with a capacity close to the rate to be pumped if the well is to be placed back in service. The pump should be capable of non-stop pumping at its capacity for at least 72 hours. A valve or other method to "fine-tune" the discharge rate should be installed. A pipe or tube to allow access to the pumped well for drawdown and recovery measurements also needs to be provided. Installation and testing should be completed no later than Friday, April 18, to allow 48 to 72 hours of water level stabilization prior to the test.
- 2. An accurate method for measuring discharge is needed. My preference would be a calibrated in-line flow meter close to the pump. Any valve should not interfere with flow measurement accuracy.
- 3. A tap should be available close to the pump discharge to collect water samples for analysis.
- 4. Discharge should be routed a good distance away from the area, either by pipe or hose. Discharge should not be into the river or wherever it could impact the monitor or supply wells. An on-site inspection prior to start of the test would be the best way to determine the discharge location.

Mr. Richard Cheney April 9, 1986 Page 2

Our current plans are for pumping continuously beginning no later than 8 a.m., Tuesday, April 20. Pumping is to be continued for 72 hours followed by collecting 24 to 36 hours of recovery data. This agency will arrange for State staff to be on site for the entire pumping sequence. We will also provide interpretation of the test results.

To complete our planning, we need to know as quickly as possible whether this equipment will be ready by the dates listed. Please contact either myself at 827-5812 or Jami Bailey at 827-5884 regarding these preparations.

Sincerely,

DAVID G. BOYER

Environmental Bureau Chief

DGB:dp

cc: Bert Barnes, Flora Vista Water Users Assoc.



IN THE ELEVENTH JUDICIAL DISTRICT COURT STATE OF NEW MEXICO, COUNTY OF SAN JUAN

No. CV 86-00154-4

FLORA VISTA WATER USERS ASSOCIATION,

Plaintiff,

v.

MANANA GAS, INC., and EL PASO NATURAL GAS COMPANY,

Defendants.

### $\underline{A} \underline{N} \underline{S} \underline{W} \underline{E} \underline{R}$

COMES NOW the defendant, El Paso Natural Gas Company, and for its answer to plaintiff's complaint states:

#### FIRST DEFENSE

Plaintiff's complaint fails to state a claim upon which relief may be granted.

#### SECOND DEFENSE

1. This defendant is without information sufficient upon which to form a belief as to the truth of the allegations of paragraphs 1, 2, and 5 of plaintiff's complaint, and the same are therefore denied.

This defendant denies the allegations of paragraphs 3,
 9, 10, 11, and 12 and all of the separate prayers for relief
 numbered 1 through 4 of plaintiff's complaint.

3. In answer to paragraph 4 of plaintiff's complaint, this defendant admits that the complaint alleges acts which occurred in San Juan County, New Mexico, but denies that any actionable conduct occurred therein and denies all other allegations of paragraph 4.

4. In answer to paragraph 6 of plaintiff's complaint, this defendant admits there is a well located at the place indicated in paragraph 6 of plaintiff's complaint but denies it is solely a gas well and denies all other allegations of paragraph 6.

5. In answer to paragraph 7 of plaintiff's complaint, this defendant admits that El Paso Natural Gas Company has received gas from the Mary Wheeler No. 1-E well but denies all other allegations of paragraph 7.

#### FIRST ALTERNATIVE AFFIRMATIVE DEFENSE

Plaintiff's claims are barred by its own comparative negligence, which bars or should ratably reduce the amount of plaintiff's claims.

#### SECOND ALTERNATIVE AFFIRMATIVE DEFENSE

Plaintiff's claims are barred by the comparative negligence or independent negligence of third persons or of co-defendants to this lawsuit, which should bar or, in the alternative, reduce in a pro rata fashion plaintiff's recoverable damages.

-2-

#### THIRD ALTERNATIVE AFFIRMATIVE DEFENSE

The damages about which plaintiff complains were all the proximate result of an act of God.

### FOURTH ALTERNATIVE AFFIRMATIVE DEFENSE

The damages about which plaintiff complains were the proximate result of an independent, intervening cause other than the conduct of this answering defendant.

FIFTH ALTERNATIVE AFFIRMATIVE DEFENSE

Plaintiff's claims are barred by estoppel. Plaintiff and its predecessors have allowed this defendant and other defendants to proceed with well operations and reclamation actions to their detriment, and plaintiff is therefore estopped from complaining of any actions and conduct on the part of this or any other defendant.

### SIXTH ALTERNATIVE AFFIRMATIVE DEFENSE

Plaintiff's claims are barred by laches.

SEVENTH ALTERNATIVE AFFIRMATIVE DEFENSE

Plaintiff's claims or some of the events which should have given rise to claims, if any, are barred by the appropriate statutes of limitations.

#### EIGHTH ALTERNATIVE AFFIRMATIVE DEFENSE

Plaintiff's claims are barred by plaintiff's own failure to mitigate the alleged damages.

WHEREFORE, this answering defendant prays that plaintiff's complaint and all claims and causes of action therein contained be dismissed with prejudice and that this defendant be awarded its costs and such other and further relief as to the court seems proper.

> ELDON J. MITRISON Assistant General Counsel El Paso Natural Gas Company P. O. Box 1492 El Paso, Texas 79978 915/541-2600

CIVEROLO, HANSEN & WOLF, P. A. Attorneys for Defendant El Paso Natural Gas Company P. O. Drawer 887 Albuquerque, New Mexico 87103 505/842-8255

B

THIS WILL CERTIFY that a true copy of the foregoing pleading was mailed to opposing counsel of record this 18th day of March, 1986.

Wolf С.

RICHARD C. CIVEROLO C. LEROY HANSEN WAYNE C. WOLF WILLIAM P. GRALOW LAWRENCE H. HILL KATHLEEN D. LEBECK DENNIS E. JONTZ CARL J. BUTKUS W. R. LOGAN JAMES J. WIDLAND CYNTHIA A. FRY CIVEROLO, HANSEN & WOLF, P.A.

COUNSELORS AND ATTORNEYS AT LAW OLD BANK BUILDING 219 CENTRAL AVENUE, N.W., SUITE 400 P. O. DRAWER 887 ALBUQUERQUE, NEW MEXICO 87103-0887 TELEPHONE (505) 842-8255

> 1. 1.

TERRY R. GUEBERT ROBERTO C. ARMIJO MARY E. LEBECK PAUL L. CIVEROLO ROSS L. CROWN R. GALEN REIMER JULIA R. WALL CLINTON W. THUTE JAY F. STEIN ELLEN M. KELLY MICHAEL H. SMITH ANTHONY J. D. CONTRI

March 12, 1986

Mr. David G. Boyer Environmental Bureau Chief NEW MEXICO OIL CONSERVATION DIVISION P.O. Box 2088 Santa Fe, New Mexico 87504-2088

RE: Flora Vista Water Users Association v. Manana Gas, Inc. and El Paso Natural Gas Co.; No. CV 86-00154-4

Dear David:

Pursuant to our telephone conversation of today, please be advised that this firm represents El Paso Natural Gas Company in the above-referenced case. We wish to review all data, technical reports, or other information of this type that are public documents pertaining to your "Report to the Flora Vista Water Users Association." We plan on doing so on Monday, April 7, 1986, at 9:30 a.m. Either Wayne Wolf of this office, myself, or both will meet you at that time.

Enclosed please find a copy of the Complaint filed in this action for your reference. This is the most recent pleading that we have.

Sincerely,

CIVEROLO, HANSEN & WOLF, P.A.

ay F. Stein

JFS/da Encl. cc: Wayne Wolf

RECEIVED MAR 6 1986

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STATE OF NEW MEXICO

End DOUNTY OF SAN JUAN

### IN THE DISTRICT COURT

FLORA VISTA WATER USERS ASSOCIATION,

Plaintiff,

vs.

No. CV 86-10154-4

MANANA GAS INC, and EL PASO NATURAL GAS COMPANY,

Defendants.

#### COMPLAINT TO RECOVER DAMAGES FOR PROPERTY DAMAGES

COMES NOW, the Plaintiff, by and through its attorney, Richard L. Lougee, and for its cause of action against the Defendants herein alleges and states as follows:

1. That Plaintiff is, and at all times material hereto was, a New Mexico corporation formed in accordance with the provisions of Sec. 73-5-2 et seq. NMSA 1978.

2. That the Defendant Manana Gas Inc. is, and at all times material hereto was, a New Mexico corporation formed under the laws of that state.

3. That the Defendant El Paso Natural Gas Company is, and at all times material hereto was, on information and belief, a Texas corporation formed under the laws of that state. 4. That the events which give rise to this Complaint occurred in the County of San Juan, State of New Mexico.

5. That the Plaintiff operates, and at all times material hereto has operated, an approved community water system for the Flora Vista area located approximately halfway between Farmington and Aztec on U. S. Highway 550.

6. (That in January of 1980, a gas well owned and operated by the Defendant Manana Gas Inc. was drilled in unit M (SW/4 SW/4) of Section 24, Township 29 North, Range 11 West in the vicinity of Plaintiff's water system as aforedescribed.

7. That the gas well aforedescribed in paragraph 6 above, more particularly the Mary Wheeler No. 1-E, was placed in service in July 1980, with the natural gas being received by the Defendant El Paso Natural Gas Company via a pipeline at the site.

8. That on February 24, 1983, Plaintiff discovered that one of its water wells was contaminated with oil and grease and said well was taken out of service.

9. That Plaintiff has since learned that its entire site is contaminated and cannot be used for the production of domestic water.

10. That, on information and belief, the cause of the aforedescribed contamination is a leak in an unlined slush pit owned and operated by the Defendant Manana Gas Inc. at the Mary

2

Wheeler No. 1-E site.

11. That, on information and belief, a further cause of the aforedescribed contamination is an improperly maintained gas dehydrator owned and operated by the Defendant El Paso Natural Gas Co. Inc. at the aforedescribed site.

12. That as a result of the Defendants' negligence as aforedescribed in failing to care for and maintain their equipment, the Plaintiff has suffered and will continue to suffer damages in an amount to be proven at trial.

WHEREFORE, Plaintiff prays judgment against the Defendants herein:

1. For compensatory damages in an amount to be proven at trial.

2. For consequential damages in an amount to be proven at trial.

3. For its attorney's fees and costs

4. For such other and further relief as the Court deems proper.

3

RICHARD L. LOUGEE Attorney for Plaintiff 108 N. Orchard, Ste. 201 Farmington, NM 87401 (505) 327-5281

#### VERIFICATION

STATE OF NEW MEXICO) : ss COUNTY OF SAN JUAN )

BERT BARNS, after first being duly sworn upon his oath states: That he is the President of Flora Vista Water Users Association, the Plaintiff in the above entitled cause; that he has read the foregoing Complaint and believes the contents contained therein to be true and correct.

BERT BARNS

On this 13 th day of <u>february</u>, 1976, before me personally appeared BERT BARNS to me known to be the person described in and who executed the foregoing document, and acknowledged that he executed the same as his free act and deed.

Vature D. Euro

My Commission Expires:

Jan 17, 1986

ELEVENTH JUDICIAL DISTRICT COURT COUNTY OF SAN JUAN STATE OF NEW MEXICO DISTRIC COURT SAN JUAN COUNTY.

EB 14 8 43 AM '85

Jela m Donagan

FLORA VISTA WATER USERS ASSOCIATION,

Plaintiff ,

vs.

MANANA GAS INC. and EL PASO NATURAL GAS COMPANY, No. CV 86-00154-4

RECEIVED - HAND DELIVERED TO EMIL AT 6545 PRAIRIE AL AT 7:30 PM SUNDAY M-3/86

Defendants.

#### SUMMONS

TO: Defendants

Defendant(s), Greeting:

You are hereby directed to serve a pleading or motion in response to the Complaint within 30 days after service of the Summons, and file the same, all as provided by law.

You are notified that, unless you so serve and file a responsive pleading or motion, the Plaintiff(s) will apply to the Court for the relief demanded in the Complaint.

Attorney for Plaintiff: RICHARD L. LOUGEE 108 N. Orchard, Ste. 201, Farmington, NM 87401

WITNESS the Honorable James L. Brown, District Judge of said Court of the State of New Mexico and Seal of the District Court of said county, this  $\underline{14^{++}}$  day of February, 1986.

EULA M. DUNAGAN DISTRICT COURT CLERK

by Cecilia Hansen Deputy

(Seal)

NOTE:

This summons does not require you to see, telephone or write to the District Judge of the Court at this time.

It does require you or your attorney to file your legal defense to this case in writing with the Clerk of the District Court within 30 days after the summons is legally served on you. If you do not do this, the party suing may get a Court Judgment by default against you.





# 🗧 BÐ HARTMAN

President Mañana Gas, Inc. Realtor Certified Public Accountant P.O. Box 36990 Albuquerque, New Mexico 87176 TELE: (505) 884-4863 (O) (505) 884-4945 (H)



# FEB 14 8 42 AH '86

STATE OF NEW MEXICO

EL > ACOUNTY OF SAN JUAN

IN THE DISTRICT COURT

FLORA VISTA WATER USERS ASSOCIATION,

Plaintiff,

vs.

No. CV 86-00154-4

MANANA GAS INC, and EL PASO NATURAL GAS COMPANY,

Defendants.

#### COMPLAINT TO RECOVER DAMAGES FOR PROPERTY DAMAGES

COMES NOW, the Plaintiff, by and through its attorney, Richard L. Lougee, and for its cause of action against the Defendants herein alleges and states as follows:

1. That Plaintiff is, and at all times material hereto was, a New Mexico corporation formed in accordance with the provisions of Sec. 73-5-2 et seq. NMSA 1978.

2. That the Defendant Manana Gas Inc. is, and at all times material hereto was, a New Mexico corporation formed under the laws of that state.

3. That the Defendant El Paso Natural Gas Company is, and at all times material hereto was, on information and belief, a Texas corporation formed under the laws of that state. 4. That the events which give rise to this Complaint occurred in the County of San Juan, State of New Mexico.

5. That the Plaintiff operates, and at all times material hereto has operated, an approved community water system for the Flora Vista area located approximately halfway between Farmington and Aztec on U. S. Highway 550.

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7. That the gas well aforedescribed in paragraph 6 above, more particularly the Mary Wheeler No. 1-E, was placed in service in July 1980, with the natural gas being received by the Defendant El Paso Natural Gas Company via a pipeline at the site.

8. That on February 24, 1983, Plaintiff discovered that one of its water wells was contaminated with oil and grease and said well was taken out of service.

9. That Plaintiff has since learned that its entire site is contaminated and cannot be used for the production of domestic water.

10. That, on information and belief, the cause of the aforedescribed contamination is a leak in an unlined slush pit owned and operated by the Defendant Manana Gas Inc. at the Mary Wheeler No. 1-E site.

11. That, on information and belief, a further cause of the aforedescribed contamination is an improperly maintained gas dehydrator owned and operated by the Defendant El Paso Natural Gas Co. Inc. at the aforedescribed site.

12. That as a result of the Defendants' negligence as aforedescribed in failing to care for and maintain their equipment, the Plaintiff has suffered and will continue to suffer damages in an amount to be proven at trial.

WHEREFORE, Plaintiff prays judgment against the Defendants herein:

1. For compensatory damages in an amount to be proven at trial.

2. For consequential damages in an amount to be proven at trial.

3. For its attorney's fees and costs

4. For such other and further relief as the Court deems proper.

RICHARD L. LOUGEE Attorney for Plaintiff 108 N. Orchard, Ste. 201 Farmington, NM 87401 (505) 327-5281

#### VERIFICATION

STATE OF NEW MEXICO) : ss COUNTY OF SAN JUAN )

BERT BARNS, after first being duly sworn upon his oath states: That he is the President of Flora Vista Water Users Association, the Plaintiff in the above entitled cause; that he has read the foregoing Complaint and believes the contents contained therein to be true and correct.

BERT BARNS

On this 13 th day of <u>February</u>, 1986, before me personally appeared BERT BARNS to me known to be the person described in and who executed the foregoing document, and acknowledged that he executed the same as his free act and deed.

Brown Notary Public

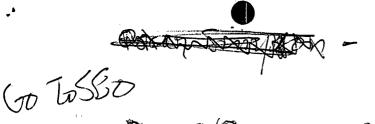
My Commission Expires:

San 12, 1986



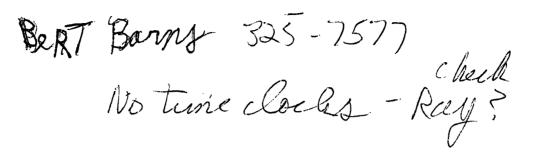
MEMORANDUM OF MEETING OR CONVERSATION

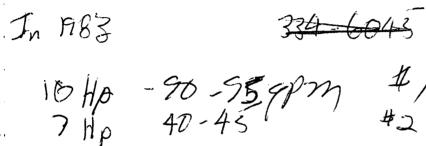
Time Date Telephone X Personal 1:20pm 13/86 ( in Flora Vista Pron Sie Originating Party Other Parties PenRod-FLODA Vista Dave Boyer-Ocb Kay ASS DC Subject The search For more specific info. on attime of well contamination 1983 pumping that delivered 90-95 gpm, #2 had 7 Hp @ 40-45 gpm capsity & located behind pump house. Well #) service in late 80 os carly 1981 Loci nol ( ລຸ) remember seeing an oil place in well-only gassy smell. Ray does not secall exactly when ma suldoes not think it was at same time as The well went bad - maybe a & year later Kay saw & green Silm in a pit by the debudratos during & ity of Ayter provides some water for system-eventoda File search - FVWA Records: lan 83-2,955,3 gal; Feb 83-3,714,016 gal; 1983 Total-46, 893, 442 gal. Aztec records (estimated due to broken mely during part of period, 1/15/83-1,154,000 qu/: 1/15-2/15/83-1,116,000; 2/15-3/15/83-24, 1983 1,156,000 gel (bate I leak ~ Feb Distribution Flore Vista File Signed (PS. Pender said #) well on more Than # 2 - Notime clocks)



March 1, 156,000

4/





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We corregol No oil Phase in well - Gatty Smell Rey says pit replaced 1 yrapter well contaminated (not at sometime es well vent bad.) Savonen silms by Leckin Sunface Casing P

FUWA City MAzkec 2,955,288 Jon 1,154,000 3,714,716 Feb 1,116,000 6,669,304 0:86 4,3,99,304 2270,000 Total Aug 50 59 Days 74,564gp2 38775 107 569 113039 38067 300 <u>36613</u> 70957 74457214 >75,800 71,000gpX 75,000 gpRx thay 140 min = 52 gpm 66900 water from wells 34 90 Errom Azter Soz 1983 use: 309, 49,672 Snow wells on \$5, 84794 9pl ~ 85,000 gpl Avgi 7090 From #1 if pumped equally 7090 + Since #1 on more

V 5 Ft below land water wells 27 At deep ; screened lower 8 ? ft. some wells are pumped ~ 12 Ft drawdown at 75 gpm continuously, some internet-Tendy newer water wells : Well #2 100 gpm

50 YEARS



TONEY ANAYA

GOVERNOR

STATE OF NEW MEXICO ENERGY AND MINERALS DEPARTMENT OIL CONSERVATION DIVISION



POST OFFICE BOX 2088

STATE LAND OFFICE BUILDING SANTA FE. NEW MEXICO 87501 (505) 827-5800

February 7, 1986

MEMORANDUM

TO: ADDRESSEES

FROM: DAVID BOYER, HYDROGEOLOGIST

SUBJECT: CORRECTION FOR GAS WELL LOCATION

The location of the Mary Wheeler Well No. 1-E gas well was incorrectly given on page 2 of the report entitled "Progress Report on Flora Vista Contamination Study, January 1986", which was recently sent to you. The correct location is Unit M of Section 23, Township 30 North, Range 12 West. A corrected page 2 is enclosed as a replacement page. I apologize for any inconvenience this error may have caused.

Enc.

average delivery in 1983 of about 100,000 gpd. The system was placed in service in 1981 with two wells each with pump capacities of 60-70 gallons per minute (gpm).

In January, 1980, a gas well owned by Manana Gas, Inc. of Albuquerque was drilled in unit M (SW/4 SW/4) of Section 23, Township 30 North, Range 12 West. The well, Mary Wheeler No. 1-E, was placed in service in July, 1980, with the natural gas being received by El Paso Natural Gas via a pipeline at the site.

Manana facilities at the site include the well, an oil-water separator, a fiberglass tank (capacity approximately 120 barrels) holding produced water and some oil from the separator, an oil tank for storing oil produced with the gas and a tank drain pit for discharging water separated from the oil (now replaced with a small fiberglass tank). A reserve (mud pit) and a blowdown pit were both likely present at one time but have been covered over. The original fiberglass tank was replaced with a second identical tank in early 1983 due to discovery of a leak. El Paso Natural Gas facilities include a gas dehydrator, a dehydrator pit with a 55 gallon drum serving as a collector, and a gas meter house.

The entire site occupies an area of approximately 220 x 75 feet and is located northeast of water supply well S1. Distances from the water well to the fiberglass produced water tank, gas well, and dehydrator pit are 235 feet, 255 feet, and 285 feet, respectively. Figure 1 shows the relationship of the various oil and gas facilities to the water wells.

-2-





STATE OF NEW MEXICO ENERGY AND MINERALS DEPARTMENT OIL CONSERVATION DIVISION



GOVERNOR

February 5, 1986

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE. NEW MEXICO 87501 (505) 827-5800

Mr. Ed Hartman Manana Gas, Inc. P.O. Box 36990 Albuquerque, NM 87176

Dear Mr. Hartman:

At the request of the Flora Vista Water Users Association and their consultants, Brewer & Associates, the Oil Conservation Division (OCD) has performed ground water quality sampling and made hydrologic measurements in the vicinity of a community water supply well (S1) that was contaminated with oil and grease in February 1983. A possible source of this contamination may have been produced water and/or other fluids that could have been released to the subsurface from the Mary Wheeler No. 1E gas well (M-23-30N-12W) and/or attendant field processing equipment at the site. A progress report of the investigation with the results available to date is being provided to you and to El Paso Natural Gas Company. The OCD plans to make additional measurements and collect further data through the fall of this year.

Two samplings of fluids from Manana's produced water tank were taken one each in September and October, as part of the study. On both occasions, I understand that the OCD Aztec District Office notified your local representative as to the day and approximate time of day of the sampling. However, no Manana representative was present for either sampling.

If you have any questions regarding this report, or planned future work, please contact me at the above address, or by telephone at 827-5812.

Sincerely,

Environmental Bureau Chief

DGB/dp

cc: R. L. Stamets, OCD Director Frank Chavez, OCD Aztec District Office





STATE OF NEW MEXICO ENERGY AND MINERALS DEPARTMENT OIL CONSERVATION DIVISION



GOVERNOR

February 5, 1986

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87501 (505) 827-5800

Mr. Howard Reiquam, Ph.D., Director Environmental Affairs Dept. El Paso Natural Gas Co. P.O. Box 1492 El Paso, Texas 79978

Dear Mr. Reiquam:

At the request of the Flora Vista Water Users Association and their consultants, Brewer & Associates, the Oil Conservation Division (OCD) has performed ground water quality sampling and made hydrologic measurements in the vicinity of a community water supply well (S1) that was contaminated with oil and grease in February 1983. A possible source of this contamination may have been produced water and/or other fluids that could have been released to the subsurface from the Mary Wheeler No. 1E gas well (M-23-30N-12W) and/or attendant field processing equipment at the site. A progress report of the investigation with the results available to date is being provided to you and to Manana, Inc. The OCD plans to make additional measurements and collect further data through the fall of this year.

If you have any questions regarding this report, or planned future work, please contact me at the above address, or by telephone at 827-5812.

Sincerely,

DAVID G. BOYER Environmental Bureau Chief

DGB/dp

cc: R. L. Stamets, OCD Director Frank Chavez, OCD Aztec District Office





TONEY ANAYA

GOVERNOR

STATE OF NEW MEXICO ENERGY AND MINERALS DEPARTMENT OIL CONSERVATION DIVISION



POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87501 (505) 827-5800

January 31, 1986

Mr. Bert Barnes, President Flora Vista Water Users Association P.O. Box 171 Flora Vista, NM 87415 Mr. Richard P. Cheney, Vice President Lawrence A. Brewer & Associates, Inc. P.O. Box 2079 Farmington, NM 87401

Gentlemen:

The New Mexico Oil Conservation Division (OCD) has completed a report on the activities through January, 1986, of this Division and the Environmental Improvement Division (EID) regarding the contamination investigation of Flora Vista Water Well No. 1 (S1). This well, which was contaminated by oil and grease in early 1983, has been out of service since that date. OCD activities in 1985 included installation of five monitoring wells, sampling of water quality from these wells and other ground water at this location, and measurement of water levels to determine ground water direction and rate of flow. EID supplied material and staff to assist in well installation.

The OCD report is enclosed with this letter but its major conclusions and recommendations for further study are summarized below:

- No verifiable contamination was detected in 1985 in either the unused water supply wells or the monitor wells except for low level contamination detected in samples taken within 24 hours of cleaning the well with an air compressor. The latest sampling for which results are available (October 1985) did not detect hydrocarbons either in the monitor wells or in a composite of the water wells currently supplying the system.
- 2) Methane at concentrations 1200 times ambient levels was detected in the monitor well closest to the gas well in August. The source is likely the decay of

shallow buried organic material. The gas well itself is not a likely source of methane since it has 227 feet of surface casing cemented back to the surface.

- 3) Ground water movement in the vicinity of the monitor wells in fall 1985 - winter 1986 was towards the river and away from the currently used water supply wells. The flow has a seepage velocity range of 3 to 4 feet per day.
- 4) Based on the available information, the produced water tank at the Manana Mary Wheeler 1E gas well, the gas well itself, and the dehydrator pit are all likely to have been within the zone of influence ("cone of depression") of the pumping S1 well at the time the water well was contaminated in February 1983. Actual pumping rates and pumping cycle information at the time of contamination would better define the extent of pumping well influence.
- 5) The estimate of travel time for unretarded soluble contaminants to have moved from the vicinity of the gas well to the pumping water well is approximately 100 days.
- 6) The rate of ground water movement is such that a single plume of contaminated produced water originating in the vicinity of the gas well in 1983 has now moved beyond the water well.
- 7) Because of the passage of time, water pumped from well S1 would not show contamination unless a zone of residual oil saturation is present at or near the produced water tank or other facilities.
- 8) To determine the presence and concentrations of any residual oil between the site of the leaky pit and well S1, exploration digging with the backhoe is recommended, followed by sampling. If oil is found, capture of soluble constituents is again a possibility and well No. 1 (S1) may again evidence contamination if pumped continually.
- 9) The OCD intends to measure water levels and sample water quality of the monitor wells and other available wells through at least the fall of 1986.
- 10) Well S5 should be capped to prevent introduction of contaminants. However, both wells S1 and S5 should have caps that allow for access for periodic water level measurements, water quality sampling, and pumping if necessary.



Letter to Bert Barnes and Richard P. Cheney January 31, 1986

-3-

- 11) An aquifer test using well S1 as the pumped well should be performed for at least 72 hours at a rate of 60 gpm or at the rate the well would be pumped if put back in service. This test would determine accurate aquifer parameters and detect any contamination in the immediate vicinity of the well.
- 12) Since well S5 is at a distance greater than 500 feet from the gas well and out of the direct path of ground water flow, it is unlikely that pumping S5 will cause capture of any remaining contaminants from the gas well. To test this assumption, additional flow calculations should be made before placing back in service.
- 13) Sampling of individual pumping water supply wells for purgeable aromatic hydrocarbons should be performed on a regular basis. For convenience, a sampling schedule identical to that required for total trihalomethanes is initially suggested.

The OCD would like to schedule a 72-hour aquifer test in March using well S1. Since we do not have a pump or a water flow measuring device, any assistance the Association could provide would be appreciated. A flow device could either be a calibrated in-line meter, orifice weir, or other accurate device. The pump should have a valve to control discharge so that a constant rate is maintained. Also, a pipe or hose will be needed to divert the water away from the monitor wells to prevent recharge. If these items can be obtained, the test can be performed.

The attached report and the proposed work represent a substantial commitment of time and effort by the three-person staff of the OCD Environmental Bureau. I hope that the information we have provided, and that which we will provide over the next eleven months, will be useful in any action you take to resolve the matter.

-4-Letter to Bert Barnes and Richard P. Cheney January 31, 1986

If you have any questions regarding this letter or the report, please contact me at 827-5812.

Sincerely,

A. Boyc DAVID G. BOYER,

Hydrogeologist Environmental Bureau Chief

DGB/dp

Enc.

cc: Paul Biderman, Secretary EMD R. L. Stamets, Director OCD NM OCD, Aztec District Office NM EID Water Supply Section NM EID Ground Water/Hazardous Waste Bureau NM EID, Farmington Field Office



December 26, 1979

POST OFFICE BOX 2588 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87501 (505) 827-2434

GOVERNOR LARRY KEHOE SECRETARY

BRUCE KING

Manana Gas, Inc. P. O. Box 145 Farmington, New Mexico 87401

Attention: Ed Hartman

Administrative Order NSL-1125

Gentlemen:

Reference is made to your application for a nonstandard location for your Mary Wheeler Well No. 1 E to be located 892 feet from the South line and 624 feet from the West line of Section 23, Township 30 North, Range 12 West, NMPM, Basin Dakota Pool, San Juan County, New Mexico.

By authority granted me under the provisions of Rule 3 of Order No. R-1670-V, the above-described unorthodox location is hereby approved.

Sincerely,

JOE D. RAMEY, Director

JDR/RLS/dr

cc: Oil Conservation Division - Aztec
Oil & Gas Engineering Committee - Hobbs

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				CERTIFICATION I hereby certify that the Information con- tained herein is true and complete to the best of my knowledge and belief. Ed. Hurtman Name Pres. Position Manana Gas, Inc. Company 11/29/79 Date
			The second secon	I hereby certily that the well location shown on this plat was plotted from field nates of actual surveys made by me or under my supervision, and that the same is true and correct to the best of my knowledge and belief. <u>November 29, 1979</u> pite Surveyed <u>handba</u> <u>mema</u> cylatered Protestional Engineer ndfor Lind Euryeya N.M. #1161
	Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Dakot Producing Form Dakot renge dedicate one lease is of yalty). c lease of dif unitization, un No If ans po;' list the over essary.)	Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc. Inc.	MEXICO P. O. HOX 2008 S GEPARIMENT INTA FE, NEW MI.XI All distances must be from the outer bound Leave Inc. Mary W Leave 23 30N 12W of well: It from the South time and 624 Producing Formation Pool Dakota Basin D. renge dedicated to the subject well by colored per one leave is dedicated to the well, outline each a yalty). I lanswer is "yes," type of consolidation . No If answer is the well until all interests have H	All distances must be from the outer boundaries of the Section All distances must be from the outer boundaries of the Section Levice Levice Levice Levice Theme 23 JON 12W San theme 23 JON 12W San theme 24 the time 10 Dakota Basin Dakota renge dedicated to the subject well by colored pencil or hachare or one lense is dedicated to the well, outline each and identify the ow 29 JON 16 answer is "yes," type of consolidationCOMMUNITIES 10 If answer is "yes," type of consolidation

	DISTRIBUTION		INSERVATION COMMISSION	Form C-104
	FILE	REQUEST	FOR ALLOWABLE AND	Supersedes Old C-104 and C-110 Effective 1-1-65
	LAND OFFICE	AUTHORIZATION TO TRA	NSPORT OIL AND NATURAL GA	s
	TRANSPORTER OIL			30-045-24092
	GAS /		h. K. S. Len	1
1.	PRORATION OFFICE			• <u>; </u> ] 
	Manana Gas, Ind			<u>I'' (</u>
		ngton, New Mexico 87	401 (505) 325-3066	ALMEN .
	Reason(s) for filing (Check proper box) New Well		Other (Please explain)	
	Recompletion	Change in Transporter of: OII Dry Gas	and t	End.
	Change in Ownership	Casinghead Gas Conden	sate	
	If change of ownership give name and address of previous owner			
11.	DESCRIPTION OF WELL AND I	EASE	·	
	Lease Name Mary Wheeler	Well No. Pool Name, Including Fo 1 E Basin-Dak		tree Fee
	Location M QQ	2 South	624	West
			e and 624	
	Line of Section 23 Tow	nship 30 N Range	12 W _{, NMPM} , San .	Juan County
111.	DESIGNATION OF TRANSPORT	ER OF OIL AND NATURAL GA	S Address (Give address to which approve	d cruy of this form is to be sent.
	The Permian Corport	ation	Box 1183, Houston, '	Texas 77001
	Name of Authorized Transporter of Cas El Paso Natural Gas		Address (Give address to which approve Box 1492, El Paso, '	
	if well produces oil or liquide,	Unit Sec. Twp. Pge.	is gas actually connected? When	
	give location of tanks. If this production is commingled wit	M 23 30 N' 12 W		
IV.	COMPLETION DATA			Plug Back Same Res'v. Diff. Res'v.
	Designate Type of Completio	n – (X) X	X	
	Date Spudded 1/28/80	Date Compl. Ready to Prod. 3/8/80	Total Depth 6488	Р.В.Т.D. 6441
	Elevations (DF, RKB, RT, GR, etc.)	Name of Producing Formation	Top Oil/Gas Pay	Tubing Depth 6198
	5495 KB Perforations 6197-6202, 62	Dakota 10-12, 6240-44, 6246	6197	Depth Casing Shoe
	6315-21, 6344-46,	6362-65, 6410-12.	CEMENTING RECORD	6484
	HOLE SIZE	CASING & TUBING SIZE	DEPTH SET	SACKS CEMENT
	$12\frac{1}{4}$ 7 7/8	<u> </u>	<u> </u>	<u> </u>
	<i></i>	2_3/8	6198	
v.	TEST DATA AND REQUEST FO	DR ALLOWABLE (Test must be a)	fter recovery of total volume of load oil a	nd must be equal to or exceed top allow-
	OIL WELL Date First New Oil Bun To Tanks	able for this de	pth or be for full 24 hours) Producing Method (Flow, pump, sas lift	eic.)
	Length of Test	Tubing Pressure	Casing Pressure	
	Faudiu of Test	Tubing Pressure		MILLINLD
	Actual Prod. During Test	Oil-Bbla.	Water - Bbls.	°MAR 111980
	0.4.0 WD.5.	<u> </u>		OIL CON. COM
	GAS WELL Actual Prod. Test-MCF/D	Length of Test	Bbls. Condensate/MMCF	DiST. 3 Grauly of Condeneator
	812 MCF, 829 AOF Testing Maibod (pitot, back pr.)	3 Hours Tubing Fressure (Shut-in)	Casing Pressure (Shut-14)	Choke Size
	1 point	1794	1816	3/4"
VI.	CERTIFICATE OF COMPLIANO	CE	RATE RATE	LIGN CONNESION
	I hereby certify that the rules and r Commission have been compiled y	ith and that the information given	APPROVED	, 19
	above is true and complete to the	best of my knowledge and belief.	BY	3
		<u>г</u>	TITLE <u>SUPERVISOR DISTRICT</u>	
	<u>ak Kenda</u>	cli	. If this is a request for allows	ble for a newly drilled or deepened led by a tabulation of the deviation
	Vice Preside	ent	tests taken on the well in accord	lance with RULE 111. t be filled out completely for allow-
	2/10/	le) CO	sble on new and recompleted wel	lu.
	(De	ie)	well name or number, or transporte	III, and VI for changes of owner, n or other such change of condition.
			1) Sanarata Korma C.104 milet	DE TIER DE BECH DON' (8 MILITON)
	• •		· · · · · · · · · · · · · · · · · · ·	

	t	1-1-1-									
SANTA FE	1		NEV	WEXICO	OIL COM	SERVATIO	N COM	MISSION		· · · · · · · · · · · · · · · · · · ·	Type of Legise
ALE .	/					OMPLETIC			ID LOG	ate [	
AND OFFICE	2								5.50	ne On 7	& Gas Lease No.
		<u> </u>								7777	777777777777777777777777
TRERATOR		JJ								////	
INTYPE OF WELL		·		·····					7. Un	11 Aurec	206ent 1/9062
1.		oir [	GA:	· F						,	
STYPE OF COMPLE	TION	WELL	we		DRY	QTHER_			8.10	rn. or L	ease Name
NEW XO		DEEPEN			ESVH.	OTHER			M	arv	Wheeler
Name of Operator										11 110.	
Manana	a Gas,	Inc.								1	
Address of Crerator									10. 1	ield and	Fool, or Wildcet
Box 145	5, Far	mingt	on, New	/ Mexic	:0 874	401 32	<u>5-30</u>	66	Bas	in D	Dakota
Location of Well										1111	
м		89	2		South	h	6	24		////	
ST LETTERM	LOCAT	ED	FEET	FROM THE _	bouci		7777	2 - FI	ET FROM		MHHHH
West LINE OF	2	3	א <u>א</u>	12	W		////	(IIII)	San San	-	
Date Spudded	16. Date	T.D. Head	ched 17. Da	te Compl. (h	leady to 1	'rod.) 18.1	Elevatio	$\Delta M$	KB, RT, GR, etc.	/ 19. E	llev. Cashinghead
1/28/80 5. Total Depth 6488	2/	′9/80	3	/8/80			549	5 KB		5	5483
, Total Depth	1	21, Plug B	lack T.D.	2.2.	If Multipl	e Compl., Ho	w 2	3. Intervals	Hotary Tool:	¹ S	, Cable Tools
					Many	the star		Drilled i	y → 0-6488		1 1
Producing Interval(	s), of this	completion	i – Top, Bott	om, Name				,			, Was Directional Survey Made
6107 641										]	
6197 64]		akota									No
					-		· · · .	1		27, Wa	s Well Cored
ES-Ind, C	-Cal	iper				ort all string:					No
CASING SIZE	WEIGI	HT LB./F	·	TH SET	1	E SIZE	5 501 111		ING RECORD	<b>.</b>	AMOUNT PULLED
8 5/8		24		27	121			390			none
4 ¹ / ₅		10.5		.84		7/8		1372			none
					1						
·		LINE	ER RECORD	<del></del>	·			D	TUBING	RECO	RD
SIZE	TOP	, 	BOTTOM	SACKSC	EMENT	SCREEN		SIZE	DEPTH S		PACKER SET
								2 3/8	6198		none
, Perforation Record	(Interval )		mbarl			32.		SUOT ED			5575 FTC
197-6202, 6				6246-4	8,	h	INTER		ACTURE, CEMEN		D MATERIAL USED
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410-12.	,		, +	,							0 lbs sand
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		1.0				UCTION	~ <u></u>				
te First Production		· ·				ing = Size an	id type 1	ump)	1		(Prod. or Shut-in)
ne of Test	Hours Te		oint ba I Choke Size	CK pre		011 - Bbl.		as - MCF	Water - Eb	shut	Gas-Gil Ratio
3/8/80	3		3/4	Test F		011 – 1) <i>0</i> 1.		AICI			
JV8/80 Jw Tubing Press.	Casing F	ressure	Calculated	24- CII - H	351.	Gus-h		Wate	r - Bbl.	Tolis	API (Corr.)
.I. 1794	Б.I.		Hour Hate	->			MCF	1	51		and the second
Disposition of Gus			vented, etc.)	<u> </u>			AOF		A SAL WIELE	and Hy	
									A Participation	1	a80
List of Attachments					•				MAR	1 2 1	-or4.
. Thereby compy that								·*	C	ON. C	<u> </u>
, I hereby certify that	the inform	notion show	en on both si	les of this f	orm is tru	e and complet	te to the	e best of m	y know howe and	kelzer.	3
1:1	2 ~	1 0	- 1		•		• •			-	1 las
SIGNED	4	14-	ic/(		TLE VI	ce Pres	side	nt	DATE		11/20

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# NEW MEXICO OIL CONSERVATION COMMISSION MULTIPOINT A ONE POINT BACK PRESSURE TECHTOR GAS WELL

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Typ	n Geot	] Initial			Annual			Sp	ecial	i ent Lon 3/8,		<u> </u>		
Con	'l'ariy				Conau	ction				L			· · · · · · · · · · · · · · · · · · ·	, 
	Manana	a Gas,	, Ir	nc.			aso Na	tural	L Ga	s Com	bany			· · · · · · · · · · · · · · · · · · ·
.00	, Basin				Form	ikot	a					Unit 1	·	
Cor	pletton Date			Total Depth			Flug Back			Elevation		1	or Lease 17 a	
	2/29/8			6484				41		5483	СН		ry Whee	eler
. sq	. Size 4.500	10.5	5	°4.052	6484		Perforance From	6197	Te	6412	2	Well N	1 E	
1-2	. 5120	W1.		d	Set At		Perfetaile					Unit		Twp. Elge
	2.375	4	7	<u> <u></u> <u></u> <u>995</u> <u></u> <u></u></u>	6198	3	From	Ope Posi.or		nded			23-30N-	-12W
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	Tubono			â					12	9		Ne	w Mexic	20
•	L	н.		Gq	% CO 2		% N 2	1	% H2S		plovel		r Run	Japs
				0.730										
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EL PASO NATURAL GAS COMPANY POST OFFICE BOX 990 FARMINGTON, NEW MEXICO

#### NOTICE OF GAS CONNECTION

## DATE July 23, 1980

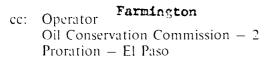
### THIS IS TO NOTIFY THE OIL CONSERVATION COMMISSION THAT CONNECTION FOR PURCHASE OF

GAS FROM	anana Gas, Inc. Operator	<u>Mary_Mheele</u> Well	
93-151-01	72501-51	М	23-30-12
Meter Code	Site Code	Well Unit	S-T-R
Basin Dako			ural Gas Company
	Pool	Name of	Purchaser
WAS MADE ON	July 15, 1980	FIRST DELIVERY	July 16, 1930
	Date		Date
829			

AOF ____

812 CHOKE





El Paso Natural Cas Company Purchaser Original Signed by Charles A. Thomas Representative

Assistant Chief Dispatcher Title



File

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CONTINUED BASIN DA			AS) MAR			11005		AUC	SEPT		NOV	01.6	YR-PROD M	PAGE 102 P ACCUM.
WELL S T R	JAN	FEB	MAK 53333211:	APRIL	MA Y + + + + + + + + + + + + + + + + + + +	JUNE	J(ILY #############		3671 ********	*******	********		18-PK()/ 8	r RUCUN. Freisiari
IC 229N14W GAS UTL WAT	214 2	265 8	661 45	456 20	308 10	221 10	186 5	119	41 2	6 5	5	NUNE 5	2483 117	1334267 7410
TWIN MOUNDS 102530N14W GAS	1400	1699 20	17 70 21 5	1665	705 11 20	1942	2070	1927	1280	886	974 13	902	17220	1157870
E 1H2530N14W GAS Dil WAT	5	5	5	5	20	20	20	20	20	7712 108	8512 119	2929 20 10	170 19153 247 10	19153
COMPANY TOTAL UIL GAS MAT	297 52663 124	45682 133	390 59479 207	57910 152	53448 157	54814 172	420 56068 217	552 56368 217	79776 166	80618 175	743 79328 158	82071 173	758225 2051	27410345
CURTIS J. LITTLE Federal Com R. 201223013W GAS	5186	4573	5087	******* 3956	******* 5470	*******	******** 5516	******** 3909	4026 34	4 2 36 37	4326 26	2973	49258	68491
FEDERAL 2 11 201128N13W GAS 01L	66 Last	41 PROD. DA	50 TE 12/78	46	69	23	41	30	34	37	20	21	484 U	779 390289 28145
COMPANY TOTAL OIL	5186	4573	5087	3958	5498	23	5518	30 3909	4026	4236	4326	2973	49258	453780
LIVELY EXPLORATION LIVELY 112720N BW GAS		659	830	******	******** 587	******* 517	********	********	****** 616	816	635	837	8384	230075
2E2128N BW GAS	1074 12 4475	659 23 8	830 19 4313	634 15 8	587 20 4117	517 13 7 3963	675 12 4081	504 12 8 4134	616 20 7 4106	816 16 4139	635 9 4361	16 8 3954	8384 187 93 50462	230075 3293 650625
01L NAT 3N 929N 8W GAS	- 31 6787	4566 20 7454	4313 22 7801	4253 18 7015	40 6550	6351	6818	6097	6323	4139 50 9 6460	50 6401	6801	50462 280 104 80858	660626 5781 927732
WAT	1155	6	1369	1013	1170	6	6	971	5	6	6	6	71	329
4D1429N BW GAS Dil Wat	1155	1735 5	1369	5	5	966 4	1061 4	5	1000	100Î 5	1054 5	1167 5	13662 3732	250756 21
5P 129N 9W GAS Gil Wat				226	358	598	396	486	470	43Ī 4	366 5	401 5	14	58535 610
6G3029N 8W GAS 01L WAT	2395	2160	2517	2496	2574 5	2428	2553	2587 5	2418	2423	219Ž 5	2326 5	29069 60	389912
Y 7E3530N 8W GAS WAT BN1229N 8W GAS	2026 778	2112 875	21 4 9 10 3 8	21 70 859	2056 871	1065 813	1080 934	167 796	294 763	2086 799	685	778	15205 56 9981	254789 166716
01L WAT 90 329N 8% GAS 10F1729N 8% GAS	3020 3120	1743 2795	1777 2968	1055 2812	724 3222	850 3024	970 3131	667 3082	6 16 3056	625 3082	544 2972	1092 3084	68 13683 36348	44 175625 391553
01L WAT 1111628N 8W GAS	10 2830	10 2522	10	2519	10 2509	2270	9 1687	2524	10 2080	3102 25	10	10 2203	117 28752	320 323548
DIL WAT 12L362BN BW GAS	22 5 1388	16 5 860	2395 35 1767	1224	- 38 5 1401	15 740	874	24 5 1092	1125	25 5 861	2111 17 5 1388	18 5 1122	238 58 13842	4763 180941
	10 3206	10 3137	10 10 2876	10	10	79	3404	2325	2931	3472	10	10	23 108 35001	1 840 392137
UIL WAT	14	26	21	2777	2821 14 5	2463 15 5	105	20	20	25	2865	2724	218	2078
16L2029N 9W GAS 01L 17P 126N 8W GAS	1033 5 2144	570 1 1911	764 1 2136	941 6 2048	913 2006	840 1 1918	1014 2 2321	766 1873	904 1,334	947 1866	645 1871	817 1652	10154 26 23080	135352 544 185741
DIL Wat 18M 126N 8W GAS	41 1909	46 1096	57 2011	36 1427	53 1817	34 5 1750	45 1917	41 6 1702 19	45 3 1547 27	40 2054	52 1799	37 1961	528 67 20990	5198 203576
DIL WAT 1901226N BW GAS	- 19 4787	11	3407	2	27 6 3166	23 4340	4269	19 6 4165	27 5 4178	3329	3825	22	229	3381 539476
UIL WAT	62 15	4214 57 15	55	4387 50 18	14	45 16	55	15	52	31	15	15	46651 595 185	6639
DIL	2793	2554 29 5	27 22	2620 26 5	2707 28 5	2449	2445	2431	2533	2502	2811	2744 28 5	31311 348 58	306862
2113127N 7W GAS Dil Wat	2809 29 5	2106	1949 18 5	1818 20 5	2045 23	1901	1836	2287	2062 23 5	2060 23 5	1607 21 5	1800 9 5	24280 221 58	234079 2793
25L2930N 8W GAS WAT 26N 729N 8W GAS	1262	1384 5 672	1237 5 768	1200 5 813	1292	1405 4 849	1344 5 891	1090 906	1022	1274 972	971 759	1372	14853 60 10112	117804 81912
27P1828N 8W GAS	3335	3515	3074	3032	3094	27 32	2341 19	2228	3134 19	3174 8	3122	3316	36097	343855
30M2030N 9W ĞAŠ Dil Lively Com	2784	2455	3104	3146	2432	2325	2881	23 1947	2710	24 2874	2503	2893	32055	226393
14K3630N BW GAS WAT 15H3230N BW GAS WAT	1630 1842	1423 1544 5	1623 1755 5	1442 1602 5	1584 5 1696	1405 1647	1439 1662	1363	1463 1584 5	1433 1669	1118 1443 5	1480 5 1740 5	17403 58 19831	206365
WAI COMPANY TOTAL OIL GAS WAT	> 283 59670 135	5 54062 135	> 56350 135	5 53529 137	52581 52581 135	9 49610 121	208 52024 119	242 47837 131	278 48991 122	289 53451 135	2 323 48048 133	2 185 49643 133	5'8 3095 625796 1571	46939 7252010
LYNCO DIL CORPORAT DICK HUNT FEDERAL 111230N14N GAS			133	******	*******	******	*******	******	*****					
2H 130N14W GAS	87 964	1992 1892	1821 1425	1856 1540	1292 165 370	1763 571	1610 2183	1792 2011	1463 191		1774 1330	972 508	16422 165 12985	394351 11726 234038
OIL SCHUMACHER 3E1830NIOW GAS	LAST	PROD. DA	TE 08/79		60								80 S	590 540645
COMPANY TOTAL DIL GAS	1051	3884	3246	3396	1662	2334	3793	3803	1654		3104	1480	29407	12316 1169034
M AND M PRODUCTION Mexico federal a 111024n 6w gas Cil	AND CPE 624	RATION 609	903	******* 817	******** 448	******* 352	******** 367	326	****** 391	111	374	307	5629	251223 1567
MANANA GAS INCORPO Annie B Inijjonizy gas	RATED 2618	2355	2620	******	******** 21.47	******** 2419	********* 2126	2149	22 <u>18</u>	1915	1913	1805	75057	134301
BETTY HARTMAN	2979	6	2520 15	1672	2147	10	2120 2784	2760	13	8	11	13	25857 117	124381 604
DIL MARY WHEELER	30	2765 31	2739 19	2207	2552	2678	23	19	2451 20	2620	2426	2565 20	31526 260	131922 1213
1D2330Ñ12₩ GAS DIL ■ E 1M2330N12₩ GAS DIL	6779 90	6584 74	6536 81	4468	5016 67	5664 83	4769 72 5327 240	5159 83 8372 284	5388 90 6504 228	5370 79 7200 228	5478 77 6729 236	5534 77 6336 184	66745 950 40468 1400	324785 4780 40468
COMPANY TOTAL DIL GAS	12376	111 11704	11795	106 8347	91 9715	109 10761	345 15006	397 18440	351 16561	331 17105	351 16546	294 16240	2727	1400 7997 621457
MARATHON DIL COMPA JICARILLA APACHE	NY	8		******	*********	********	********	******	******	44.20		50/1	50/20	
8P2726N 5W GAS	3958 43 15	5216 38 14	5[19 23 14	4915 33 14	4629 20 15	3214 14 14	2463 15 15	3937	2591	4438 12 10	4911 26 11	5041 20	50432 275 155	960344 8780
942826N 5W GAS Dil Wat	22490 115 2	16958	83	19529 113 84	21110 117 87	13667 103 72	11552 50 58	19496 173 2	19198 117 72	16562 133 70	19625 190 70	14025 83 74	211904 1360 595	4561929 48276
10B2726N 5W GAS Dil Wat	6293 38 22	4741 48 21	5935 45 22	5664 30 17	6142 150 18	<del>\$4</del> 56 18	2934 17 14	7029 47 13	4736 37 12	5245 23 11	6478 50 13	4800 40	64453 525 192	1351080 12827
11H2826N 5W GAS Dil Dil	5262 60 30	6022 63 28	5373 52 28	5689 40 27	2536 73 28	3991 43 19	4682 32 17	4209 38 18	5535 57	4899 35 17	3742	4632 53 15	56572 609 261	1339506 21569
1243326N 5W GÂS Dil Wat	5640 37 44	5996 44 33	6457 41 35	6058 41 34	5719 47 35	4006 38 30	4943 23 24	4002 23 22	4956 41 21	6101 40 18	6523 40 20	6129 41 21	66530 456 337	1430964 13648
13M3326N 5W GAS 01L WAT	3471 23 30	3603 20 28	3146 14 23	2878 18 23	2715 12 24	2887 13 23	4336 20 17	2969 14 14	3513 23 13	4503 26 13	2693 19 13	3082 20 13	39796 222 234	994610 11411
	20	2.9	23	2,	47	63	••	- 7	e 4	••		• •	F14	

CONTINUES BASIN DAD	KOTA (P	ROPATED G	A S }											PAGE 111
WELL S T P	JAN	FF8		APRIL	44Y ********		JULY	AUG	SEPT		NOV	05C	YP-PR()) 4	P ACCUM.
2512930N BH GAS	1287	1036	11.	1324	1511	1000	1344	948	1626		1195	1276	14935	132739
26N 720N BW GAS WAT	667	726	686	610	665	642	731	656	666	650	827	65 ရှိ	8171	90095
2791828N 88 GAS	2557	3093	2639	2869	7755 1	7558	2156	3050	2959	2876	2791	2597	32300	376755
WAT 30M2030N 9W GAS AIL WAT	2833	871	458	1745	2366	2426	2341	2336	2219	2510	2694	3077 5	25866 5	252259
LIVELY COM 14K3630N BH GAS	1 3 2 0	612	312	192	538	817	636	945	951	1078	1550	1741	10699	217054
15H3230N 9W GAS WAT	1429	1202	1262	1420	1651	1437	1497	1645	1466	1515	1340	1408	17373	295023
COMPANY TOTAL OIL	49633	40190 129	39132	60383	100388	s1373	75888	86509	79833	77838	78035	99346	7594 868045 845	8133833
MEATED FEBERAL JON				******	*******	*******	*******	*******	******					
111024N 6W GAS 010	581	523	527	519	288	384	439	397	262	338	817		5075	256238 1567
MANANA GAS INCOPPOR	RATED		1030	*******	*******	*******	********	*******	******		1205	1202	10412	1/2201
INI330N12W GAS BETTY PAPTMAN	1919 14	1672	1939 14	1696	1733	1460	1554	1572	1529	1667	1285	1393	19417	143791 695
1P1430N12W GAS	2388 16	2315 25	2619 30	2419 21	2318 24	2205	2342 17	2104 17	2595 19	2592 16	1958 28	2638 19	28493	160315
MARY WHFELER 102330N12W GAS	5818 69	4937	5816	5309 56	4347	4738	4818 94	4598 31	4073	5594 63	4818	5342 55	60208 826	384994 5636
F 1M2330N12W GAS	6054 205	5237 142	5955 102	56 28 1 4 5	5036	5288 157	5377	5170 123	4874 127	5613	5169 127	5231 140	64632 1628	105100 3028
COMPANY TOTAL BIL	16199	14888	16328	15653	13232	1 3681	14841	13442	13881	15488	1 3238	14684	172723	-19388
MARATHON OIL COMPAN JICARILLA APACHE 892726N 5W GAS		3790	6300	******	*******	********* 1950	********* 3796	********* 3871	******* 3650	4047	3704	46.85	45013	1005357
DIL WAT	4774 27 9	3780	4300	3507	2729	19	3790	6	3	4067 17 5	14	4685 31 6	146	8926
9A2826N 5W GAS Dil Wat	21349 152 74	19158 127	15983 114 62	4583 70 40	11117	366		24811 141 62	15379 143 60	10419 127 50	10199 107 54	23248 135 62	156671 1199 562	4719530 49475
1082726N 5W GAS	64 78	5322 30	5629 28	2075	3718	128	273	8685 4 <u>8</u>	5810 40	5528	4374 27 10	7571	55591	1406671 13190
WAT 11M2826N 5W GAS	5288	10 3464	5670	3111	2066	3229	6189	6000	4910	6605	5780	4921	57133 556	1396639
DIL WAT 1243326N 5W GAS	58 15 5769 55	42 12 4024	58 16 6491	15 6405	14 4985	3302	47 12 7698	67 12 6584	46 10 5519	67 12 3027	73 10 8129	7519	145 69451 447	22125 1500415
	16	37 14	46 16	14	37	330Ž 20	47	6584 52 13	11	6	8129 28 12	21 15	146	14095
13M3326N 5W GAS Dil UAT	2797	2905 15	3697 32 13	2489	2702	2421	3792 24	2977	2962 14 8	3555	4560 38	4665 49 9	39422 279 119	11690
14M3426N 5W GAS DIL	6149 68	457î 49	5888 62	2820 20	4963 55	2867 27	5530 40	6596 230	5374 58	5233 60 12	4565 50	5903 52	60459 770	1894725 21627
WAT 1613426N SW GAS Cil	4874 48	13 3269 31	3094 31	29 23 20 12	3368 26	2233 24 5	3538	2542	2679 18	3619	4199	6693 6693 49 12	43003 384	732605
OHIO GOVT WAT	13	12	13		10		ī0	24 12	11	íż	52		134	
2P1528N11W GAS DIL DHIO A GOVT	417	273	311	183	247	210	231	229	208	243	213	233	2998	4525241 38430
212328N11¥ GAS DIL WAT	15738 68 6	12019	10330 37 6	15756 28 6	11579 56 5	13084 53	10124 46 4	8201 40	9447 32 4	13229 72 4	9333 48 4	13708 63 4	142547 593 47	7786091 59134
OHIO C GOVT 252628N11W GAS DIL	2167	2469	254B 20	2364	2303	1597	936	1857	1220	2060	1612	1641	22774	5727696
JP2629NI 1W GAS	1i 5952	7927	79 13 79 16	13	5633 49	4971	4210	5832	5 4889 50	7659	4620 17	70 12	112 112 73730 527	44 32 890
	30	80	55	7050 53	49 7	22	19	40 11		60 11	11	53	527 107 76591	42026
E 3M2628N11W GAS DIL WAT						5370 286	12285 109 24	12969 145 24	11906 89 20	12295 130 21	10335 80 19	11432 97 16	936	76591 936
COMPANY TOTAL OIL GAS WAT	81752 192	691 81 161	71857 183	53265	55410 127	41728 34	58802 104	91053 178	73951	77529 161	71604 166	991 92	845313 1799	36237474
		ROPERTIES		******	*******	********	*******	*******	******	101	100		•	
IA1324N 7W GAS OLL WAT	4502 21 5	4417	4325 16 5	3479 14 5	4930 19 5	4801 11 5	4077 18 5	5382 20 5	42 80 20 5	3122	1160 5		44 474 163 55	2007309 8821
MILLER B 5N1224N 7W GAS DIL WAT	5645 3 5	4557 4	1429 5	3489 5	5829 5	5877 5	5768 5	6040 50 5	5926 5	4837 34 5	3270 5 5	5099 11 5	57765 197 55	2335647 8071
COMPANY TOTAL DIL	10124	8975	5754 10	6966 10	10759	10678	9815 10	11422	10205	7959 10	4430 10	5099	102239 110	16897 4342951
APACHE P. MCHUGH		-		******	*******	*******	*******	*******	******			-	••••	
APACHE 101826N 3W GAS	2602	2128	1520	1201	1788	320	913	1706	1605	535	1859	1849	18026	453124
211826N 3W GAS	1820	1184 13 PROD. DA	1180	1171	936 13	701	939 5	1035	1302	776	834	1603	13481	351700
301926N 3W GÁŠ Dil 411926N 3W GÁŠ	LAST	PRUD, DA	TE 12/74									145	S 145	265286 3171 103629
SEITZEN 3W GAS	1985	2369	1341	1056	1872	579	869				1675		14567	103629 1329 439003
641726N 3W GAS OIL	2886	2369 35 3486 99	1341 32 848 55	1056 15 3337 76	1872 29 2684	1457 40	1127 502			14	1675 50 249 7	2821 38 6337 181	256 22425 673	9236 532595 29870
702026N 3W GAS	1985 28 2886 109 841 11	1137	897	840	81 606 16	254	502 13			,	554	719	6347	268923
882026N 3W GAS DIL BOLIN HARDIE	LAST	PROD. PR	10R TO 6.	/73									ť	880 539
103429N 8W GAS	528	414	530	449	681	679	718	640	488	569	694	414	6804	470730 5240
COLKET 1C1525N11W GAS	11124	8453	10019	8721	7108	2436		10487	10473	10168	7500	7774	94332 670	1102131
E 1K1525N11W GAS	14968	8453 62 9210 75	10883 67	9623 69	6446	2041		2249	7507	10168 83 6543 48	6112	4779 37	80 361 636	1021
ERIN STAYS COM 1A 225NLIW GAS E 1C 225NLIW GAS	6438 91	71 80	6528		6044	800		2964	13262	9366	8033	9231 133 17023	76533	488034
	41	14	13141 505	6687 83 19395 401	11772	1572		02	1.24	136 64	23635 410	17023	86538 1831	6923 86538 1831
FLORANCE 1E26 28N BW GAS	2639 24	1957	3202	22 54	2288	2819 21	2566 21	2740	2875 26	2646 28	2728	1984 18	30598 305	685962 7863
HARDIE 102629N BM GAS	29	2027	33 2026	1838	1783	1596	2152	26 1738	1416	28 1476	1589	1397	20809	615707
E 1K2629N BW GAS	,	-	_								5967	3607	9574 153	1370 9574 153
202929N 88 GAS	188	289 2	256	176	207	210	135	120	148	162	322	652 T	2865	428740 2734 34515
E 2M2829N BW GAS	2256		1476	1744	1638	1305	2176	2999	1652	1559	16569 210 1389	17946 147 1327	34515 357 20639	34515 357 601533
E 462429N 8W GAS	24.70	1616	4 4 ( 0	1246	1039	1303	2110	2/77	8547	17197	1384	11861	52329 48	545
502328N 8W GAS	2592 20	2057 21	2684 28	2287 74	2394	2689 23	2661 16	2594 18	2585 6	2534	2530	24465	30053 235	52329 48 827130 8592
	4. J		2.0					2.0	-	•		-		-

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	CONTINUED BASEN DA	KOTA (PS	RORATED G	AS)	,										PAGE 121
	WELL S T R	JAN	FEB	MAR	APRIL	*******	JUNE	JULY	AUG	SEPT	001	NOV	DEC	YR-PR00	AP ACCUM.
	21L3127N TH GAS	2093	1880	2170	1833	1823	2361 17 10	1890 20	523 3		348	760	3037 33	18718 109	276029 3066
	E 2103127N 7H GAS	10 9229 298	10 6144 105	10 7002 157	10 6076 120	10 5562 92	5153 115	5234 93	1292		759 17	1182	8370	74 56003 1243	70827 1820
	25L2930N 8W GAS	1269	1135 10	157 30 978 10	30 1160 10	20 1004 15	1104 15	1188	1072	1134	679	10	1272	12001	144740
	26N 729N 8W GAS	868	657	805 10	697 10	631 15	770	20 782 10	20 763 10	20 742 20	699 20	666 15	594	137 8674 140	98759
	27P1828N BW GAS	2823	2255 30	2763 3 30	2171 20 10	2725 14 15	2934	2619 11 15	2417	2747	2101		2267	27822 141 160	404577 1886
	BOMZOJON 98 GAS	3113	2738	2988	2463	2666	2515	2670	2771	2662	1956		2248	28790	281049
	LIVELY COM 14K3630N BW GAS	5 1843	15 1399	15 1394	20 1223	20 1307	20 1354	20	20 1121	20	15 961		10 1089	180	231387
	15H3230N BE GAS	1358	1268	1415 15	1223 10 1293 15	1210	1403	1310 1299	1266	1332 15 1499	927	891	985	14333 110 14814 110	299837
	COMPANY TOTAL OIL	94423	15 500	91243	101733		10 746	10	10 64949	10 310	152	174	705		91 84 762
	WAI	447	77345	91173 770	101733 845	105330	76228 647	82041 650	64949 326	79492 330	66097 308	41292 178	109605 323	989708 6049	9109764
	MEXICO FEDERAL A III024N 6W GAS	AND OPE 409	ERATION 410	410	******* 357	******** 321	******* 258	******** 242	* <b>*****</b> * 296	*******	301	332	307	3818	260116
			.10	110		521	250	2.72	270		501	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	501	5010	1567
	MANANA GAS INCORPO ANNIE B 1N1330N12W GAS	1615	1110	1217	******* 1026 2	******** 1632	1339 2	1304 2	******** 1690 1	1519	1045	183	1252	14932 41	158723 736
	BETTY HARTMAN 191430N12W GAS	2149 17	6 2078	6 1963 19	2 1464 11	2364 30	2 2204 11	2 1940	1 2093 19	2	3 1762	3 394	7 3001 25		183656
	MARY WHEELER 102330N12W GAS	17 5229	14 4196	19 4373		30 4797	11 1461	6	19	1909 18 851	3307	3 6236		23341 177 38762	423756
•	E 1M2330N12W GAS	5027 128	46 4203 100	4446 86	1775 25 3237 110	76 4677 121	1197	426 25 473 48		752	56 3541 142	78 6702 130	6111 77 5309 152	39584 1022	6123 144684
	COMPANY TOTAL DIL					121	2		20	3					- 4050 12544
		14028	11589	11999	7502	13490	6201	4143	3783	5031	205 9675	13515	261 15673	1757	910319
	MABATHON OIL COMPA JICARILLA APACHE 8P2726N 5W GAS	4877 24	2697 20	3038 31 5	3638 24 5	3349	2359	1641	3241		1495	3834	4058	34226	1039583
	E 8K2726N 5W GAS	-6	-4	5	-3	4	23	62	3241 21 2	16464	16934	18 3 18453	17848	69699	69699 736
	942826N 5W GAS	19571 153	6853	10521 102	8449	10580	7533	5714	10620	436 11 6224 33	63 11	144	21719	736 47 107784	4826314
	DIL WAT 1082726N 5W GAS	153 68 7419	57 30 26 39	68	8449 25 42 3768	123	80 34 2739	33 23 2727	37 30 4480	33	13 2601	5265	105 52 3862	748 414 43391	50223 1450062
	OIL WAT E 10C2726N 5W GAS	82 11	20	5164 27 10	3768 18 11	2727 33 10	137	2727 2727 33 10	37		3	10659	20 5 12834	348 78 38061	13538
											14568 130 31	67 23	53	270	38061 270
	11M2826N 5W GAS UIL WAT	5740 55 12	3506 38 6	2238 20 11	1283 23 11	807 15 6	959 8 5	981 12 6	2962 45 3		2192	4128 98	5575 89 12	30371 406 85	1427010 22531
	E 11C2826N 5W GÂS DIL					-	-	_	-		33625 895	21514	11492	66631 1214	66631 1214
	12A3326N 5W GÂS DIL	7252	3942 12	3525 21 15	2435 21 14	3733 31 15	2442	1172	2593 18		3469	5316	5580	41459 292 113	1541874 14387
	E 12J3326N SW GAS	12	4	15	14	15	4	6	•	31047 694 21	34651	33790 461	27438 202	126926 1718	126926 1718
	13M3326N 5W GAS	4092	2683 14 5	3283 28 9	2418 21 9	2109	1427	1816	2625 35 3	150	1729	16 5426 61	4411 37	32166 253	1066198 11943
	E 13E3326N 5W GAS	147	-5	-9	-9	34	6	Ŧ	ĨĴ	9791 295 13	2 15990 405	10561	9617	45959 922	45959 922
	HAT 14M3426N SW GAS	7320 52	4701	1697	4723	3290	3346	3422 23	4867 43	13	4330	9151 9151	21 5500	52347	1947073
	ULL WAT 1613426N 5W GAS	5621 5621	43 10 4481	1697 27 13 3968 39 15	4723 45 13 4137	3290 79 12 3658	3346 23 10 2713	6	43 3 3631 31		610 610	124 11 4101	55 6 4990	537 101 39796	22004 772401 6257
	DIL WAT E 16A3426N SW GAS	55 15	26 13	39 15	6 14	40 14	9	1886 12 6	31		3	57	25529	353 120 69796	6257 69796
	OIL WAT										19235 645	25032 849 11	133	1627	1627
	2P1528N11W GAS OHIO A GOVT	226	244	273	238	250	237	250	219	192	214	218	259	2820	4528061 38430
	21232 8N11W GAS OIL WAT	13911 71	7597 54 3	8382 18	10434	7069 31	9825 56	1705	531			3072 24 2	13227 72 6	75753 369 30	7861834 59503
	OHIO C GOVT ZE2628N11W GAS OIL	4 1623	1127	3 1387	4 1623	4 112g	3 1481	1 11 2 0	860	774	902	2 1230	661	12000	5741604 58418
	E 242628N11W GAS	1623 13 13	8 12 5188 27	5 5345	6497 28	4208	3534	11	14623	1325	1267	1725	8713 62	40 425	40425 190
	01L WAT 3P2628N11W GAS	7723	4	5345 20 3 6276		18 3 497	6604	497				5 2741	62 863	13908 98 40425 190 31 44316 179	190 4477206 42225
	DIL WAT	7723 26 10 11309	5792 18 10 10289	15 11 8360	37	1	3	3009	4657	1649 12 1000		3827	11110	179 60 71031	
	E 3M2628N11W GAS DIL WAT	118	83 13	40 14	7018 37 14 7018 37 14	1033 55 2	6286 15 8	3009	6768 47 7	20 22 33 1		32	1 i óğ	565	147622 1501
	COMPANY TOTAL BIL	96782	61739	63 393	63679 167	44139 137	51 485 113	27137	49516 78	69837	153808	170044	195286	1046865	37282338
			123 DPERTIES	177	167	137 ********	113	88 ******	78 ********	62 ******	166	161	224	1683	
	MILLER A 1A1324N 7W GAS UIL WAT			58	1409 22 5	2877 92 5	3490 18 5	3498	3674 23 5		2360	3245 21 5	3668 21 5	24 279 204 35	2031588 9025
	MILLIFR R	5472	5282 12	4260	5 5307	5 5843	5 6413	5459 2	5 5474	5215	5 37 52		5 4448	62189	
	5N1224N TH GAS DIL WAT		-12			2		ź				5264 72 5	5	92 10	2397831 8163
	COMPANY TOTAL OIL	5472	5282	4318	67Îğ	8725	9903	8957	91 ² 3	5215	6112	8509	81 15 10	86468	4423418
	JEROME P. MCHUGH				*******	******	*******	*******	, *******	*****	,	10	10		
	101826N 3H GAS	2600	1561	1447	952	1184	600	934	1252	972	1068	1140	1607	15317	468441 2481
	E 1A1826N 3W GAS	4103 102	4556 93	2721 92	3116 96	4127 111	4107 124	4272 132	2488 59	9	5749 197	4682 143	2480	42401 1222	42401 1222
	2L1826N 3W GAS	2733 24	1066	1072	1043	852 10	489	662 5	944		1380	999	1197	12437	364137 5896
	BD1926N 3W GAS		PROD. DAT								-		39	39 S	265286 3171 1102
	E 3H1926N 3H GAS UIL 4L1926N 3H GAS	67 203	18 3 100	148	13	26	10	10 10	27	45 4	488 86	275 62	172	1102 264 464	1102 264 104093
	SE1726N 3W GAS	2588 31	1361	1285	1231	1191	808	1007	1576		614	1746	663	14070	453073
	MAT	31	29	35	33 464 40	22 910 187	11 778 108	4 945 119	7 639 71		2 1949 291	8 1161 167	917 917	187 16 7763	9423 1763 1023
	E 5G1726N 3W GAS DIL WAT				40	187	108	119	71		291	167	23	7763 1023 23	1023
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		K ******	JAN BBZZZZA	F E8		APRIL	HAY	JUNE	JULY	AUG	SEPT		NOV	DEC	YR-PROD M	P ACCUN.
	2E2128N 8	W GAS	3552 35 30	21 76	2410 22	29 10 2 2	2691 30	1798	22 79 26	2443	1320	1725	3759	2251 23	29 31 4 251	761 090 6293
	E 202128N 8	WAT BW GAS	2102 67	2176 32 30 1489	1351	1530	793	217	1238	1487	1100	920 19	1259	1267	14753	105067
	3N 929N 8	WÁT	15 8586	43 15 5571	10 6673	42 6 7183	19 2 608	2746	2B 212	1 1286	26 3 3503	5514	60 13	11 5170	382 71 47052	3865 1104771
		HAT	30	15	10	10	1			1	3	2	3	4	79	329
	E 3E 929N 8	UIL UIL	12599	8014 21 25	9782 24 25 432	13244	1376	1707	1047	2105	12245	6144	1413	4880	74556 132 110	400275 224
	401429N 8	UIL	30 1929	236	4 32	1185	1909	1395	1551	1352	1172	899	1191	1230	1448ĭ	289459 21
	5P 129N 9	W GAS	319	209	381	436	418	473	441	412	435	315	391	318	4548	71029 610
	6G3029N 8	WAT GAS	10 2311	10 356	10 919	12 1799	16 3307	2645	2340	2084	1573	2271	2099	2034	23738	464496
	E 713530N 8	NAT W GAS	6656	5	1224	4	•	4329 10	1087	2908	5239	33 79	272	1900	59 27496	5Z 79585
	Y 783530N 8	WAT GAS	10		10/80	• • •			2	2	6	+		1	370	254789
	BN1229N 8	UIL WAT	5 5	393	445	386	432	366	284	444	550	455	604	811	5684 24	188669
	90 329N 8	HAT	1408	286	28	1353	1271	958	1301	836	1296	1146	1043	1043	11969	210076
	10F1729N 8	W GAS ÚIL HAT	2467	402	30 29 , 2	2658 5	474	208	244	119	499Î 2	765 1	1285	3983	20685 23 38	477695 380
	E 10M1729N 8	hA T	6638	786	1121	15829 200	2263	187	343	1103	10417	3849	1606	13439	57581	136900
	1111628N 8	UIL WAT	2529 20 10	1999	1505	200 2542 33	474	428	567	3479	2548 18	70	900 2 1	3415	20457 115 37	394595 5229
	E 1181628N 8	W GAS DIL	7643	4233 68	1372	3857	1135	136	1295 36 25	9528 106	7913	3393 54 17	2321	6676 80	48902 713	101912
	12L3628N 8	W GAS DIL	832 10	20	593 11	13 846 14	941 15	214	934	1433	38 1004	829	1109	893 10	174 9628 84	217062
	1341828N 8	WAT W GAS	30 2863	57	379	2510	18 1948	13823	2186	1805	1783	2296 2296	1674	1720	20 60 3 70	465402 2301
	16L2029N 9	MAT W GAS	630	284	713	16 457	386	501	434	369	332	305	158	148	4715	158912 579
	17P 126N 8	DIL WAT	1546			1		2049	1973		1385	1869	1284		18	579 246088
		WAT	26	1711 35 879	17 29 28 5	1426	2080 25 3	83	18	1463	15	16	4	1615 29 8	20130 233 49	6400
	E 17H 126N 8	W GAS	671 18	879	702 42	589	891 43	871 55	868 16	736	705	703	481	605 15	8701 309	28096 2160
	18M 126N 8	W GAS	1038	518 11	335	4115	197	512	1643 14	2789	2102	1261	194	2030 14	12731	251378 3964
	E 18D 126N 8	WAT GAS	2749	2 666 14	2 282	1338 1338	286	1318	4652	4680	4353 23	26 49	264	4338	27575	118436
	19C1226N 8	WAT	4347	2388	4529	3750	2254	495	9 1451 15	4412	4622	19 2131	1166	60 3 4645	36190	652561 9773
	E 19N1226N B	UIL NAT W GAS	67 60 1605	33 70 2129 45	4529 57 75	21	26 11 389	6 65	15 16 504 11	4412 29 25 2730	16 14 1895	14 24 1443	18 22 243	53 35 2323	355 370 17562	
		UIL WAT	17	5	2436 32 5	1859 12 7	5		3	2730	17	24	î	35	234	41748 631
	20P1226N 8		2564 31	1833	1770	769	1080	257	1664	2474 28 2	2189 22 12	2426	1285	2806	21117	371912 6226
	E 2081226N 8	GAS	4130	3248 34	2683	1649	707	38	1737 38	4792	3545	2752	2301 22	3455	31037	75542
	2113127N 71	GAS	2792 26	15 1782 15	10 465	1801	823	473 18	248	252	42 3660 38	2418 21	2289	2298 27	120 19301 199	295330 3265
1	E 2103127N 7	GAS	26 10 5292 117	3407	747	19 10 2375	868	12	1221	1528 24	5415 137	4598	3977	3258	53 32810	103637
	2512930N 81		25 1376 10	62 20 563	244	70 15 1186	18 3 1314	913	911	29 896	36 808	86 16 625	85 16 687	87 424	697 138 9947	2517 154687
	26N 729N 81	HAT GAS	686	56 <u>3</u> 658	721 10	650	70 8	681	655	587	419	714	518	586	33 7583	106342
	27P1828N 81	H GAS DIL	3462	337	452	2614	2583	3603	3288	3225	24002	2935	273	2351	26741 100	431318
	30M2030N 91	WAT N GAS	15 3548	666	502	2188	2447	1331	2486	2 2984	18 3098	2434 2434	2705	2774	27163	308212
	LIVELY COM	WAT	30		2	3	4	4	2	2	4	4	4	2	61	-
	14K3630N 81 15H3230N 81	WAT	1687	357 447	249 478	1387	2192 2926	1706 2013		1228 3 1326	1512 9 1339	918 1361	1672	1283	14191	245578 314476
		WAT	1468			87 <u>i</u> 2	د						1271	1139	14639 12	
	COMPANY TOTAL	GAS	103414 422	492 48140 286	324 51466 232	84 6 98 404	43823	34330	42858 146	71008	97853 318	66513 195	44939	88203 139	777245	98\$8723
ļ	MAND M_PRODU				2.52	******	*******	*******	*******	*******	******			127	2017	
1	M AND M PRODU MEXICO FEDERA 111024N 61	AL A GAS OIL	411	308	1759 180	358	404	814	698	553	727	283	200	296	6811 180	266927 1747
	MANANA GAS II		ATED		100	*******	*******	*******	******	*******	*****				100	••••
	ANNIE B IN1330N121	GAS	768 11		187	444	1191	1266	1205	1317	967	829 2	1325	1047	10546	169 <u>269</u> 792
ł	BETTY HARTMAN 1P1430N12	N	1851		281	986	3299	1942 17	17	2361 16	2048 10	20.69 17	2052	2303 17	19192 120	202848
1	MARY HHEELER 102330N121	GAS	10 3299	1985	5449	2900	2384	1390	699	588 17	818	817	16 3193			454123
	E 1M2330N121	011	3299 75 3913 108	34 3767 106	5449 93 4377 128	2662 46	49 3461 85	14 1309 23	75	1006	834 37	729 14	2177 143	6845 118 6527 171	30 367 515 30 837 941	6638 175521 4991
	COMPANY TOTAL		9831		128	6192	10333	5987	1945	5272	4689	4444	8747	16722	90942	1061129
				5752	10294	699Z	10335	5907	1979	5272 *******	4667	****	8747	16722	90942	1001761
	MARATHON GIL JICARILLA API 892726N 51	AČHE GAS	3889	611			396Z	458	895	4822 34	106			3223 31	17957	1057540 9155
	E 8K2726N 51	WAT	17553 70	16179 1	6214	9262	i	10375	1	13107	574	2198 10		31	85462	9155 155161 1203
	9A2826N 5	WAT	11	19466 1	63 11 3340	9262 63 5	8 9367	10375 28 5	8	10 6073	-	10 4646		1 7750	467 53 88294	1203 4714608
		DIL HAT	22644 103 2752	19466 1 246 50	93		9367 8 22 3355					{		12758 27 22	484	50707
I	E 902826N 5K	GAS DIL LAT	2752 264	246 50 22976 248 21 837	214 19	14415 209 14	33	279	21045 310 12	15145	13031 142 10 219	2356		5118 83	118458 1789 113	
	1082726N 5	GAS	264 10 4177 43	837	17	17	291 30	3105	12	15145 199 12 6745 53	219	-	9489 108	6566 36	300	1484114
1	E 10C2726N 5	WAT	14600 97 21	14515 1	4793		3	4		14121	539		16807	13440 31 5	86815	126876
	11M2826N 5	I GAS	3407	133 20 235	17		2237	2820 29		6192	428		28 10	~ <del>5</del>	289 78 15319 191	1442329 22722
	E 11C2826N 5	DIL WAT GAS	66 12 16933 176 10	7539	5734		19 3 2293	29 3		77 10 23208	564	14340			191 28 80611	
		WAT	176	5	100	27 10	2302 2302			23208 315 12	564 20 2	158		315-	889	147242
	12A3326N 51	GAS OIL HAT	5912 63 12	1112			2302		210	8302 69 13	6589 89 10	1837		3658 37 5	29922 267 47	1571796 14654
(	E 12J3326N 5W	GÂS DIL WAT	26110	19622 2 217 12	1973 171 12		-		2322	33791 212 10	24130 202 10	15965		10181	154096	281022 2965
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			FEB	MAR	APRIL	MAY	JUNE	YJUL *******	AUG	*******				********	
	13A1828N BW GAS DIL WAT	1691	1586	1465	1520	1550	1574	1645	1497	1854	2041	1268	2093	19885 120	485287 2421
	1612029N 9W GAS UIL WAT	503	1104	660	485 1	54 8 20 2	695	596 12 1	452	512	545	623 2	672	7395 33 18	166307
	17Р 126N 8₩ GAS 0(L мат	1376	1567	1333	1293	1415	1336	1349 34 2	1382		1102	1300 25 4	1096	15960 330 28	262048 6730
	E 17H 126N 8W GAS	604 29	557 27	619 35	575	559	575	594 10 15	657 11	608 13	619 21	469 8 3	604 17 3	7040 241 37	35136 2401
	18H 126N 8H GAS	1093	1003	1469	1353	1412	1295	1450 36	1255	1363	1279	1269	1251	15553 238 30	266931 4202
	E 180 126N 8W GAS	3524	2363	3475	3293	3031 54	2676	2662	2793	2512	2601 21	14636	2117 30	45683 238	164119 1392
	1901226N 8W GAS	3459	328Å	2852	3822	2624	2767	3195 49	3131 32	2892 31	2257	1825	2423	34535	687096 10182
	E 19112268 88 645	1524 19	1323	1738	1560	799	688	603	7 17	731	661 18	743	7 7 7 7	12090	53838 828
	20 P1 22 6N BW GAS	151	1293	2070	2315	1855 34	1874 32	1908 30	1853	1764	1844	1207	1953	21450 326	393362 6552
	E 20B1226N BW GAS DIL	3 10 Z	2493	3105	2596 29	2462	2535	2516	2509	2464 21	2426	2251	2565	31024 321	106566
	2113127N 7W GAS 011	1961	2233	7686 15	1624 13	1626 16	1964 12	1882	1799 15	1940 19	1891 18	1756	1474	27836 252	323166 3517
	E 2103127N 7W GAS	290	2891	2985	2802	2598	2543	2483	22222	2163	2273	1922	2135	29930 646	133567 3163
	2512930N BW GAS	64	895	103	95 o	889	872	761	859	787	890	834	671	10087	154774
	26N 729N 8W ĜÂŜ Mat 27P1828N 8W ĜAŜ	595 2 2674	550 2345	636 5 2126	640 2273	658 2112	636 2379	665 2110	677 3 2069	681 2142	644 3 2518	566 3 1997	529 2223	7477 33 26968	113819 458286
	30M2030N 9W GAS	2449	2642	2354	2340	2626	2389	2553	2530	Z 300	2783	2579	2508	33 36 30053	458286 2019 338265
	UIL MAT	4	6	6	2548	3	2	2	2	2	4	3	3	39	338265
	LIVELY COM 14K3630N 8W GAS	1010	1 530	1451	1326	1 34 7	1439	1276	1238	1240	1411	1364	1435	16067	261645
	15H3230N 88 GAS	1158	1398	1132	1044	1125	1155	1039	1071	1051	1059	994	932	13158	327634
	COMPANY TOTAL OIL GAS WAT	72185 120	97 21 7	64232 108	61323	464 65661 118	49408	49351 102	57907 81	56519 84	63557 131	79588 125	90694 161	808240 1298	10698963
	LOBO PRODUCTION MESA TWIN MOUNDS 103030N14W GAS 2L3030N14W GAS COMPANY TOTAL GAS				******	******	******	** ** * * * * *	*******	******				U S	
	M AND M PRODUCTION MEXICO_FEDERAL_A_				******	******	******	*******	******	******					
	111024N 6W GAS 01L	193	560	602	479	567	225	292	539	408	360	257	256	4738	271665 1747
	MANANA GAS INCORPOR ANNIE B 1N1330N12W GAS	1189	1025	869	******** 76 7 2	******** 500	******** 356	******** 798	******** 1059	****** 969	1077	1202	824	10637	179906
	BETTY HARTMAN 1P1430N12W GAS	2596	2032	2108	2 1745	1588	1451	1800	2034	1812	2166 11	1596	2018	22946 145	225794
	MARY WHE EL ER 102330N12W GAS	5281	4028	- 16 294	8 2759	2624	547	14 2030	11 5761	10 4480	11 5287	28 3239	6 4304		1900 494757
•	E 1M2330N12W GAS	6733	4216 141	884	4235	3333	834	3135	103 5859 110	3788	4085 140	3190	3452	40634 676 43744 1130	7314 219265 6121
	COMPANY TOTAL OIL	333	212	4155	194	111 8045	3188	170	14713	11049	251 12615	198	10598	1979 117961	1119722
	MARATHON DIL COMPAN BULACK	n			******	*******	*****	*******	******	******					
	1H1 62 7N 11W GAS	1 39 6 20	938 3	2015	2014	1741 24	1774	1638	758	1530	1950	1470	681 3	17905 110 20	1297356 10538
	E 191627N11W GAS OLL	2328	2050	4479	3735	3417	3271	2767	2414	2692	2988	3191	3403 10	36735 198 26	213271
	2N1 52 7N 11W GAS	1210	1141	1270	1384	1258 12	1389 10	1437	1290	1242	1460	1226	1435	15742 84 24	987728 8124
	R 3G2127N11W GAS	8422	5826	8266	8872	4719 73	4713	2344		F 0 3	427	419		43162	313113
	4A2827N11W GAS UL WAT	558	433	342	773	636	611 15	629 7 5	446 5 4	583 5 4	62 / 5 4	618	552	6808 67 25	1123480 11881
	EVENSEN BAT 201927N10W GAS	2487	3218 20	3610	3663	2934	3275	2712 22	2847 14 17	2941 17	3176	3318	3217 12 17	37458 180 107	1360454 17102
	FRONTIER AZTEC A 10 827N11W GAS GIL	2887	2762	3377	3157	3260	2872	22 2593	1572	3400	17 2698	2880	3908	35366 112 132	1893620
	FRONTIER AZTEC B								18	29	26	28	31		
	112827N11W GAS	688	391 13	140 36	146	48	84	100 8	129 3	283 5	598 7	701 7	990 2 7	4298 52 37	1529893 12430
	JICARILLA APACHE 8P2726N 5W UAS ULL	5729 15	+ 86+ 55	4290 34 8	2961								2	17844	1075384 9261
	E 8K2726N 5W GAS	13202	13933	12931	9385	1898	2273				13849	749	1	68225	223336
	WAT 9A2826N 5W GAS 01L	25887 120	12473	11215 33 22	6	1	20494	103				5368 20		75540 238 125	4990148 50945
	₩ĂŤ E 902826N 5₩ GAS ULL	25887 120 56 26401 159	27 12490	22		3032	18 16546 135	58			15635	816	5624 38	80602 361	199060 2156
	ША́Т 1082726N 5# GAS DIL	50 13 35	7	4496 40		1	1443				7158	258	682	19094 169	14007
	E 1002726N 5W GAS	13709		40 3 5684 18			-			754	20405	831	1	41383	168259
	11 M2 82 6N 5W CA3	62	1023	10127	5737 81	1487	3277			•	ŝ		4315 51 2	153 19 25966 299	1468295
	E 11C2826N 5W GAS		-2	7425	2704	5025 20	Ťí				22099	5350 15	ź	42603 483	189845
	010 NAT 12A3326N 5W GAS	11095	2721	68	1097	Ĩ					8197	1064		24174	1595970
	E 12J3326N SH GAS	65 13 30627 145	32 3655 25	11678 28	1						31298	1157		21 84415 469	365437
	UIL WAT 13M3326N 5W GAS	6218	446 <u>1</u>	3678	2114	931	511				2529	3258	211	23917	1114796
	E 13E3326N SW GAS	38 960 7	6976	34 3473 23	2114 14 3	2 10 2 35	831				3 8942 123	316	,	109 21 32247 350	12258 126656 1774
	01 L MAT 14M3426N 5W GAS	132 10 11374	37 10 10463	2447		35 1 590	1682				123				1774 2009785 22537
	UL	72	10463 55 5	28		2								27556 160 14	22331

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<b>.</b>	-	183377 16920 0	5786 217 210	197 171 0	4 JAQUEZ 4 549- 0 0	582 522 582	625 5 10 0	652 620	COMPANY 654 0	EL DPMENT 21 764 16 10	8ETA DEV 460 18 0	101	0 6AS1	(PRURATEE #000K-GAS- 01L #AT	N DAKUTA K2430N12W	BASIN 1X
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Ø 1351		454123	30367	8 <b>.</b> 9	Y WHEELER	BI8 MAR 81	19 - 14 19 - 19	669	ORAT	AS INCOR 00 238	MANANA G 5449 29	1985	0 6451	L PRORATE 000K GAS	N DAKDTA D2330N12W	BASI
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012 Manana Tank at Flora Viste



OLD Manana Tank at Flora Uista 10/85



SLORe Viste HamperTert 4/21-25/36



SLove Viste Aquife, Jert 4/51-55/86



Store Viste Aquiper Text 4/31-35/86



Flora Vista Hquife, Test 4/51-25/86



51020 Visto Haufe, Tes. 4/51-55/86



SLOVA Vista Hquife, Test 4/51-25/86



SLove Vista Mquipe, Tert 4/21-35/86



SLOVA Vista Myjule, Test 4/51-55/36



ELONA Vista Aquify Text 4/51-25/86



5102a Vista Aquife, Test 4/31-35/86



Slove Visle Rejule, Test 4/31-35/86



\$1000 Visla Nguifer Test 4/51-25/86



SLova Vista Aquipes Test 4/21-25/26



Flore Viste 8/15/955 550



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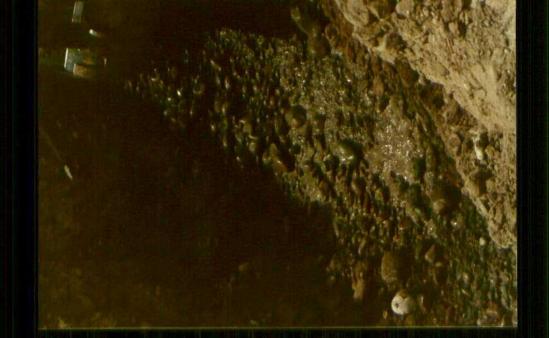
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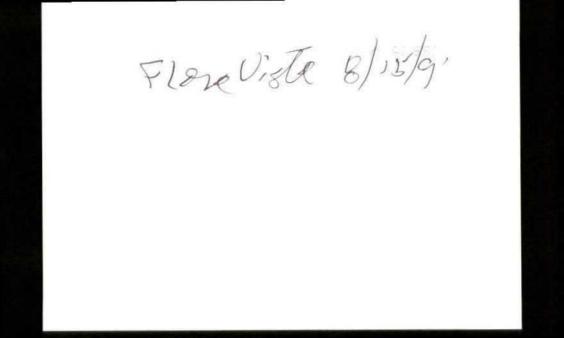


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Plone Vister 3/15/9,



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N-S trench from lank & Flore Vista / 62 Paro June 13, 89 Teerdh T-5



- 524 NTE S in Grend 7-5 Flore Vista/ & Paro June 13 7



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Flora Vita / 8 Paro Jo han markes boundar mp to T-June 13 5,07 5 ad 9-1 far ma

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Flore Vita / 2 Para MW 5 (00) backgroa June 16, 89 e we · h MW# 4 (m 20



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Bell Hole, note sul of pope and milliolened lammated clay area a Flare Unto/ & Paro Reserve Pet Center June 16/87 Ward wall



-9 2. stert from "I south " tark of to south " Flora Unto / El Paros no 16,89 Sunt X



FLORA Visla Eontamination Site Tromaly "A," East END 3/17/87 Negalivet Pholo by ASB



Site Trench "A" B/17/07 Negative #1 Photo by Ry 13



Flora Vista Contamination Site Trench "A" 8/17/87 Nogatile#



FLORGe Vista Contomination Site Trench, "A" 8/17/87 Negativet



FLOR Vista Contamination +0) Site, Eastend, Oren, L'S" 8/17/87 Negatine \$24 Photo by 2008



Store Visla Contamination Site Eastend, trench "b" 8/17/87 23)



Flora Vista contemination Sete Looking east, French "S" 8/17/87 Negatine #



FLORE Vista Con Jamination Site Soil snowThen h"S" 8/17/87 Negative # Photo by Book



Flore Viste Contamination (C) Site Trench "B" B/17/87 Photo by 28%-Negativie #20



Flora Visle Contamination site Trench" S' looling usi 8/17/60 Negotine 4 Phote by 2.98



FLORe Viste contamination (1) Site Trienc 4" " Cooking West 8/17/87 Negatine #18 Photoby Q.S.K



FLORe Vista contamenation Site Trench" D" centes 8/17/87 Photoby Rogtz

(7)



FLOR Vista Contemination Site Trench "D' Cente, looking cart. 8/17/87 Negoline # Photoby 283 (16)



Floro Vista Contamination Site Digiting Terench "F" 5/17/89 Negatine \$ Photo by ASE



SLore Vista Contamination Site belydra Ton 8/17/87 Negotine# Photo by Dig



FLora Vista Contamina), on Site, Digging Tiserch" &" 8/15/87 Degotine # B



RION Vista Contemination site Trench E" 3/17/87

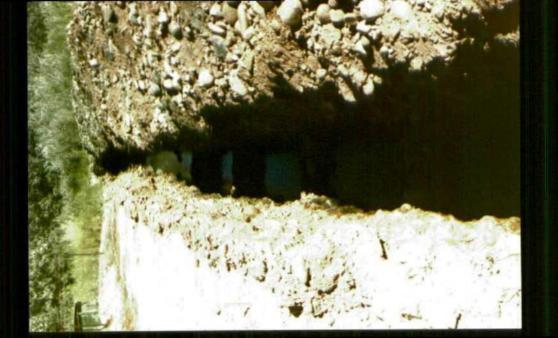
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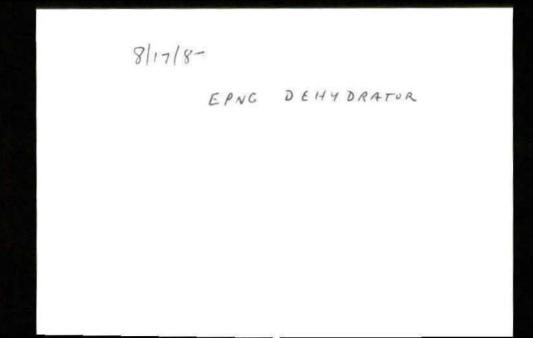
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FLOSA Visto Contamination Sile Trench E' Looking last 8/15/87 Negalive H 3







# 8/17/87

### EPNG DEHYDRATOR

### NOTE FRESH DRIP



# 8/17/87

### EPNG DEHYDRATOR

### NoTE FRESH DRIP

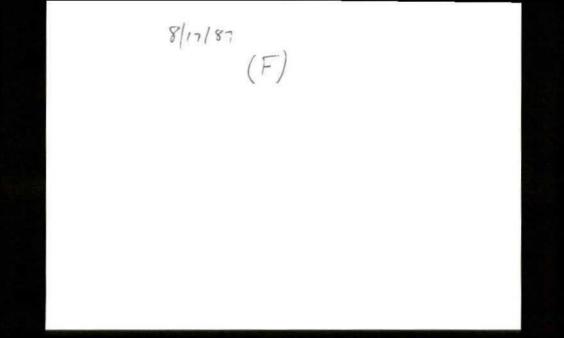
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8/17/87

### EPNG DEHYDRATOR







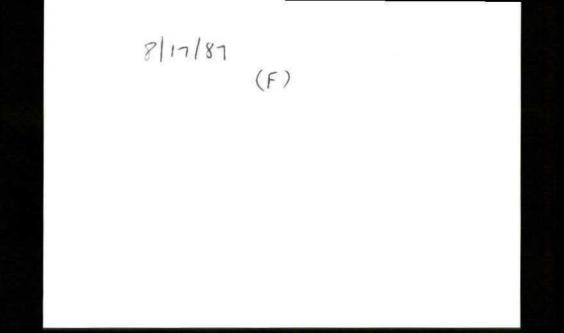
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8/17/87

(F)







22 8/17/87 (F)



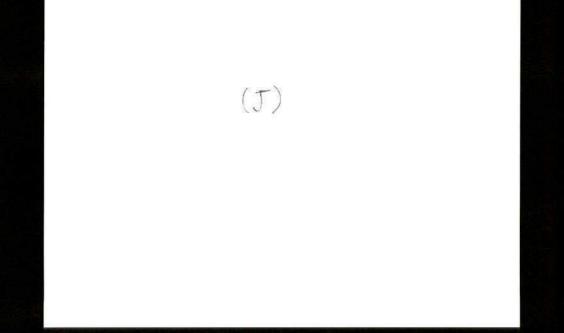
## (F) AFTER IT WAS FILLED IN

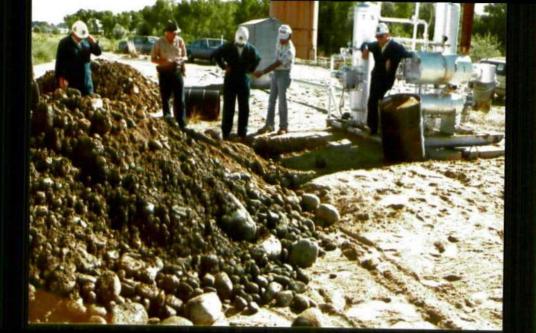


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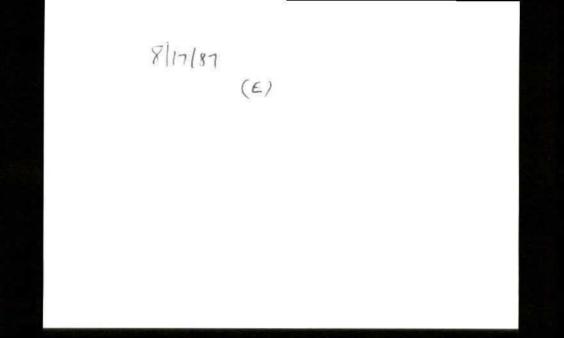


8 8/17/87 (E)

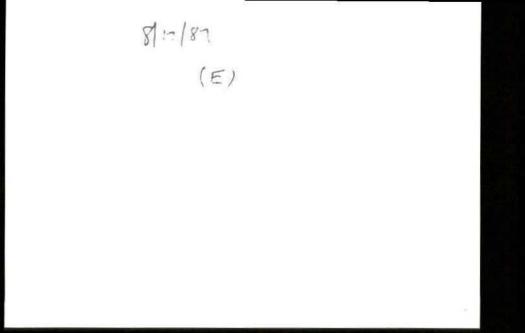


18 8/17/87 (E)











## 8/17/87 (F) AFTER IT WAS FILLED IN

1. 1.



## **ENGINEERS · ARCHITECTS · SURVEYORS**

P. O. BOX 2079 • FARMINGTON, NM 87499 • (505) 327-3303 400 PILE ST., SUITE 300 • CLOVIS, NM 88101 • (505) 763-4255 DURANGO, COLORADO • (303) 247-5766

December 26, 1985

Mr. David Boyer New Mexico Oil Conservation Division Post Office Box 2088 Santa Fe, New Mexico 87504-2088

RE: FLORA VISTA WATER USERS ASSOCIATION

Dear Mr. Boyer:

Please find enclosed one set of prints of the Flora Vista Well and Test Well Locations, including aerial photographs. We have also included a copy of this letter for the file.

Sincerely yours,

BREWER ASSOCIATES, INCORPORATED

George T. Walters, P.L.S. Vice President

GTW:gv F-331

Enclosures

cc: File

George T. Walters, P.L.S. Robert A. Echols, Jr., P.E.



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**ENGINEERS · ARCHITECTS · SURVEYORS** 

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November 8, 1985

Mr. Dave Boyer State Of New Mexico Oil Conservation Division PO Box 2088 Santa Fe, New Mexico 87504

RE: FLORA VISTA WATER USERS ASSOCIATION ELEVATIONS FOR TEST WELL MONITORS

Dear Mr. Boyer:

We are submitting herewith two (2) copies each of the elevations which we have taken to the nearest 1/100th of a foot for the above referenced project. In addition, we are also enclosing two (2) copies each of the area outlining that portion of the Crouch Mesa County Road Extension within the Lee Acres Landfill area.

Very truly yours,

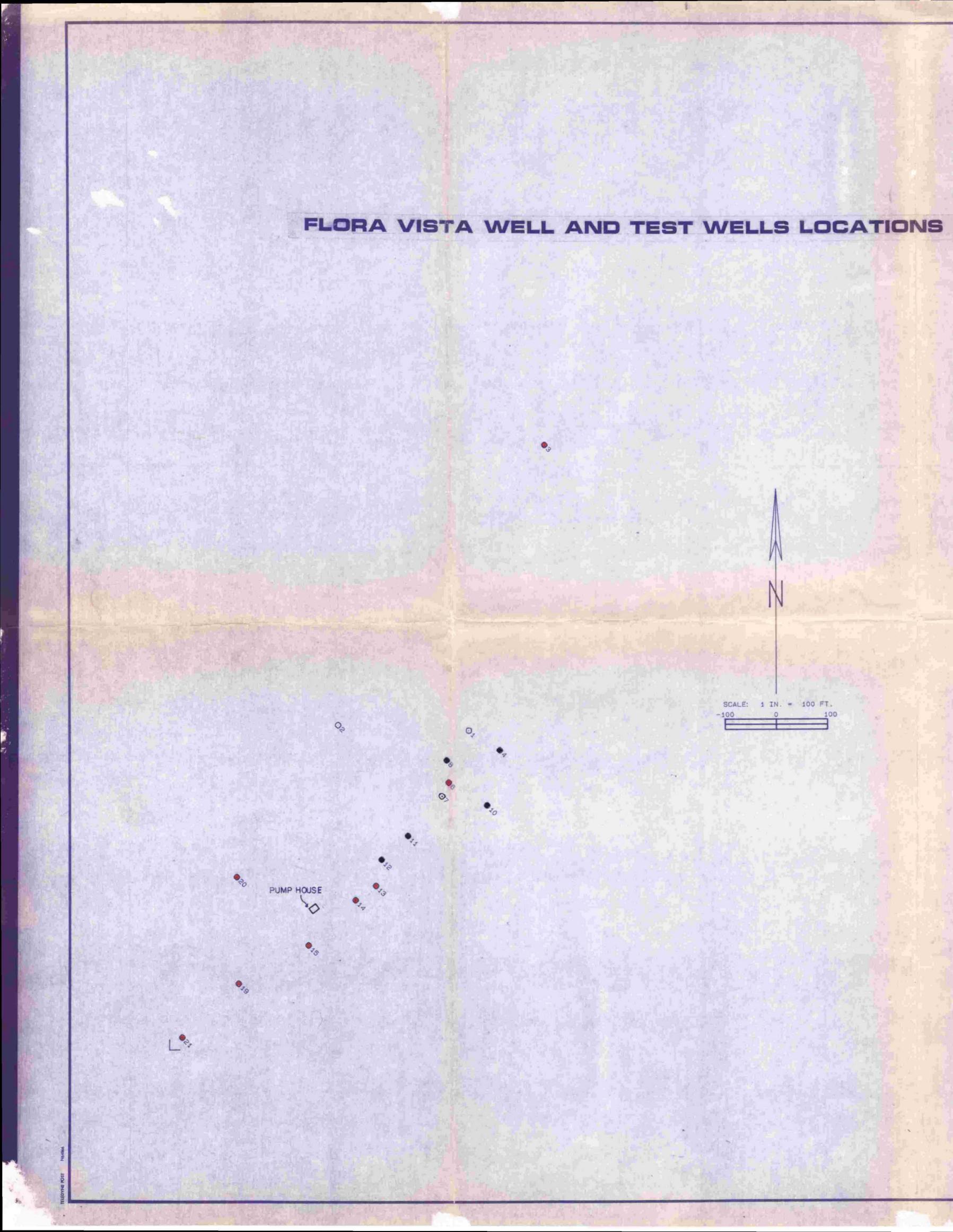
BREWER ASSOCIATES, INCORPORATED

Robert A. Echols, Jr., P.E. Engineering Division Director

RAE:rr F-331

Enclosures

Lawrence A. Brewer, P.E., L.S. George E. Bennett, Jr., A.I.A. George T. Walters, P.L.S. Robert A. Echols, Jr., P.E.



SCALE: 1 IN. = 100 FT.

100

-100

RUN BY	: JDW 9/13/85	
	LORA VISTA WELL'S'I	TEST LOCATION
POINT	NORTH	EAST
1	10000	10000
5	10010.7201674	9744.2245555
3	10558.0800542	10148.890608
15 4	9961.833901	10059.8338515
5	9939.4073038	10083.6995464
6	9898.6519976	9959.7176712
×17	9872.0911083	9946,3881401
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9	9738.7881714	9774.5036874
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16	9653.0379898	9706.7613464
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18	9606.4530754	9583.2667245
19	9506.4186671	9549.2187282
20	9714.4714054	9546.2960701
(21)	9401.6061461	9438.5830708
22	9535.9786329	9542.4122932
	000.0700020	0046.4166006
50	9180.7791972	12105.5650402
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	SET IRON ROD	ATRIVER

BM, 100', ON SLAB AT WELL 6

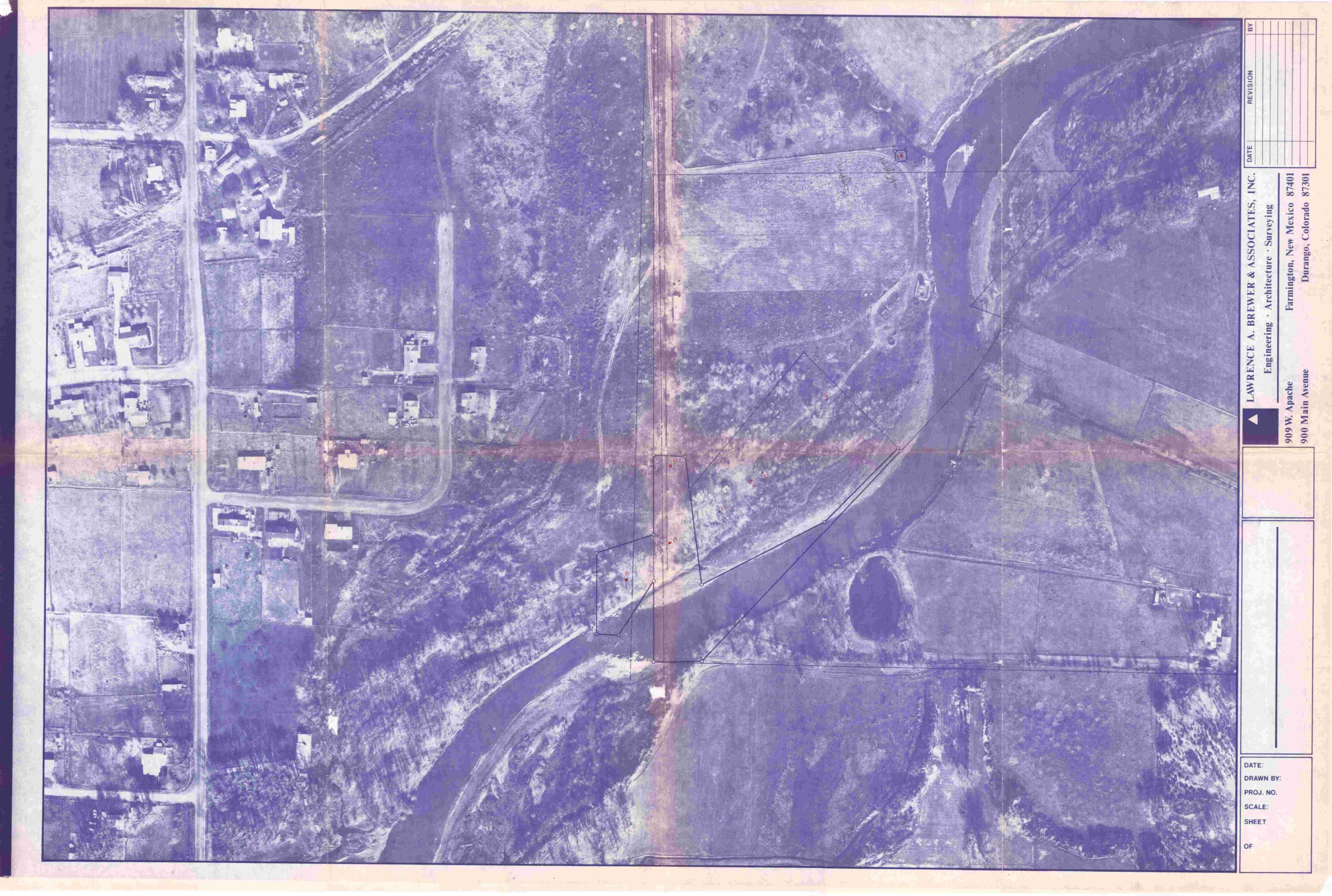
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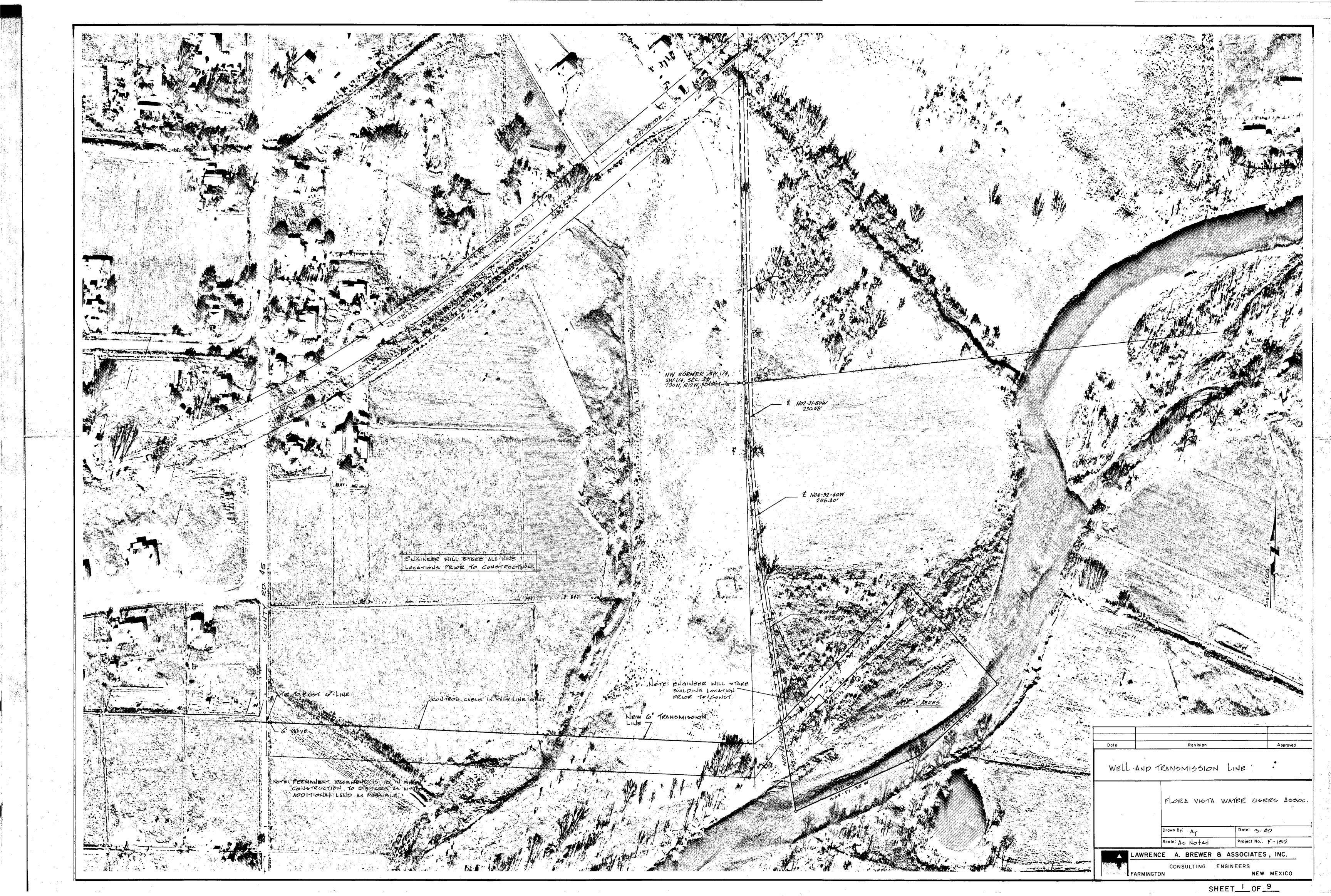
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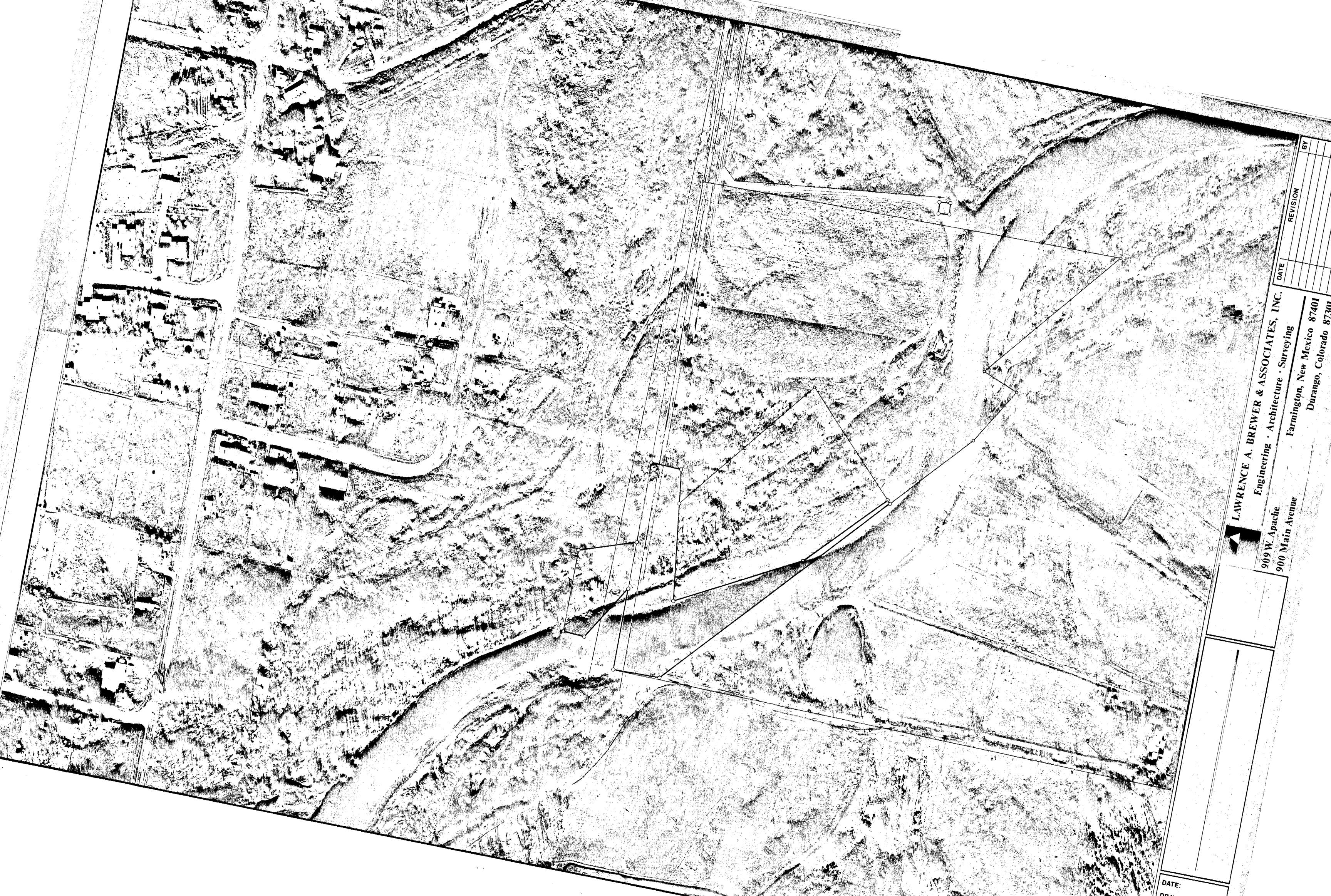
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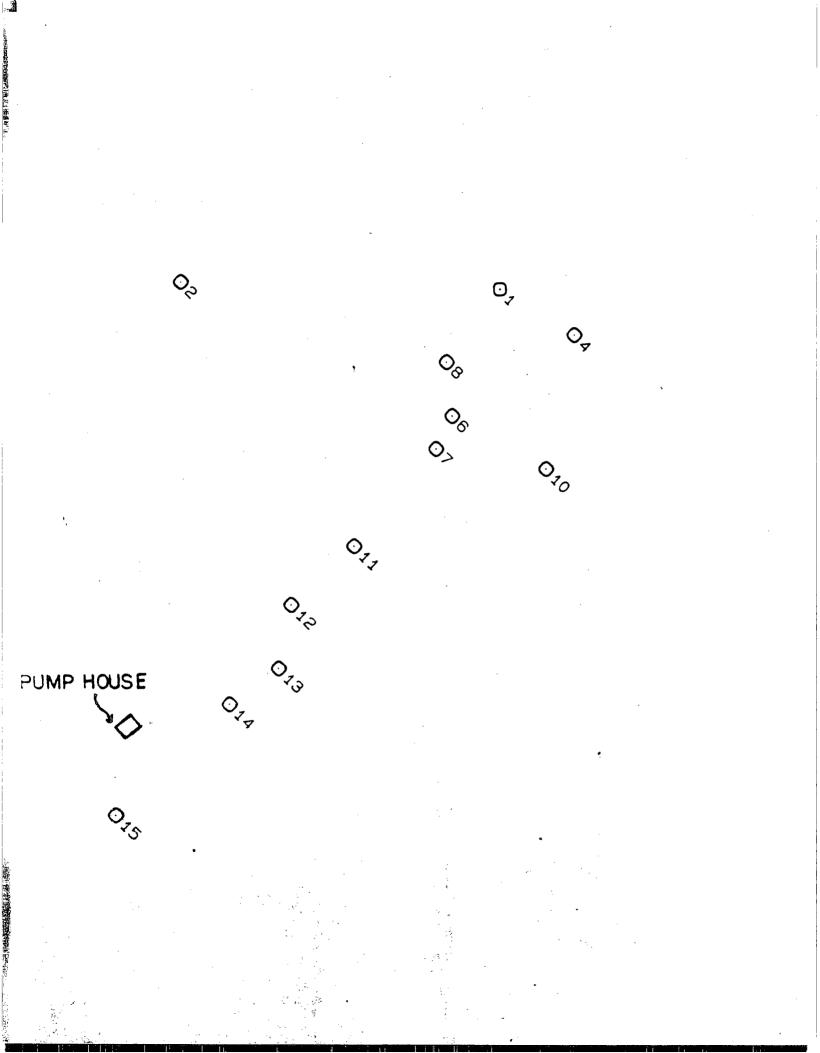
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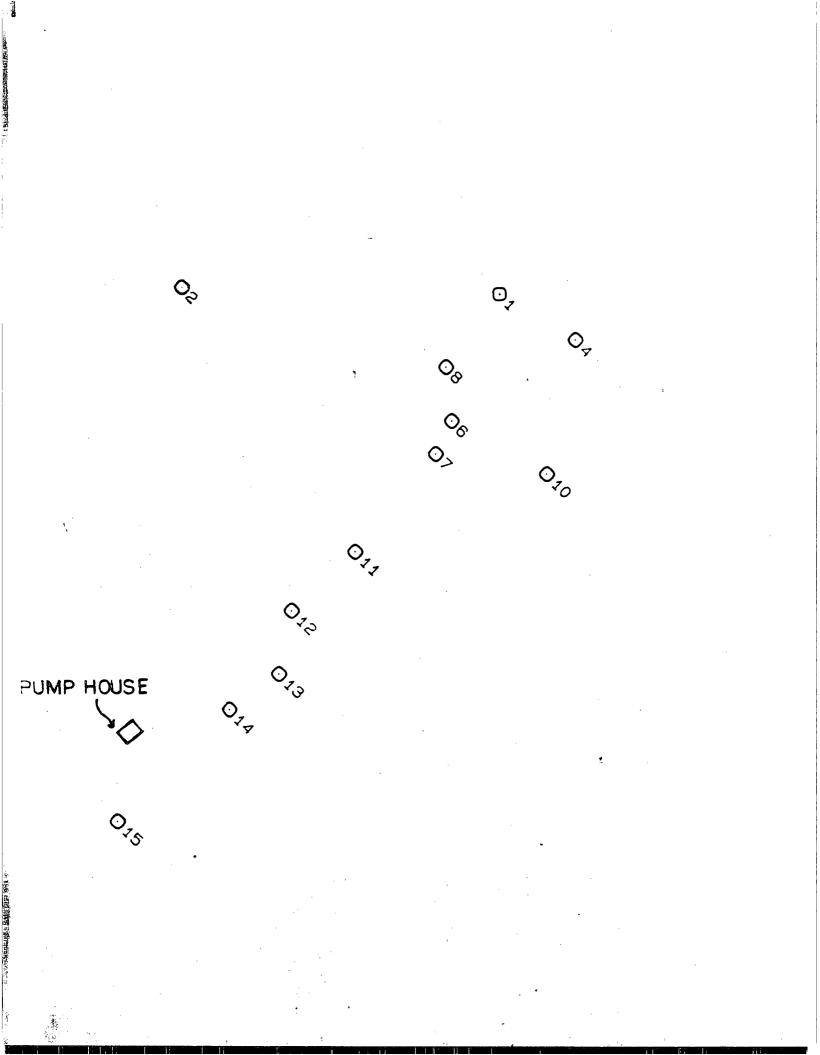
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N BY		ASSOCIATES, INC.			
13 / /: J. 331  "= I(	ELDDA VIETA NEW MEXICO	ENGINEERS · ARCHITECTS · SURVEYORS			
D.W		909 W. APACHE FARMINGTON, NEW MEXICO 87401 CLOVIS, NEW MEXICO DURANGO, COLORADO			4

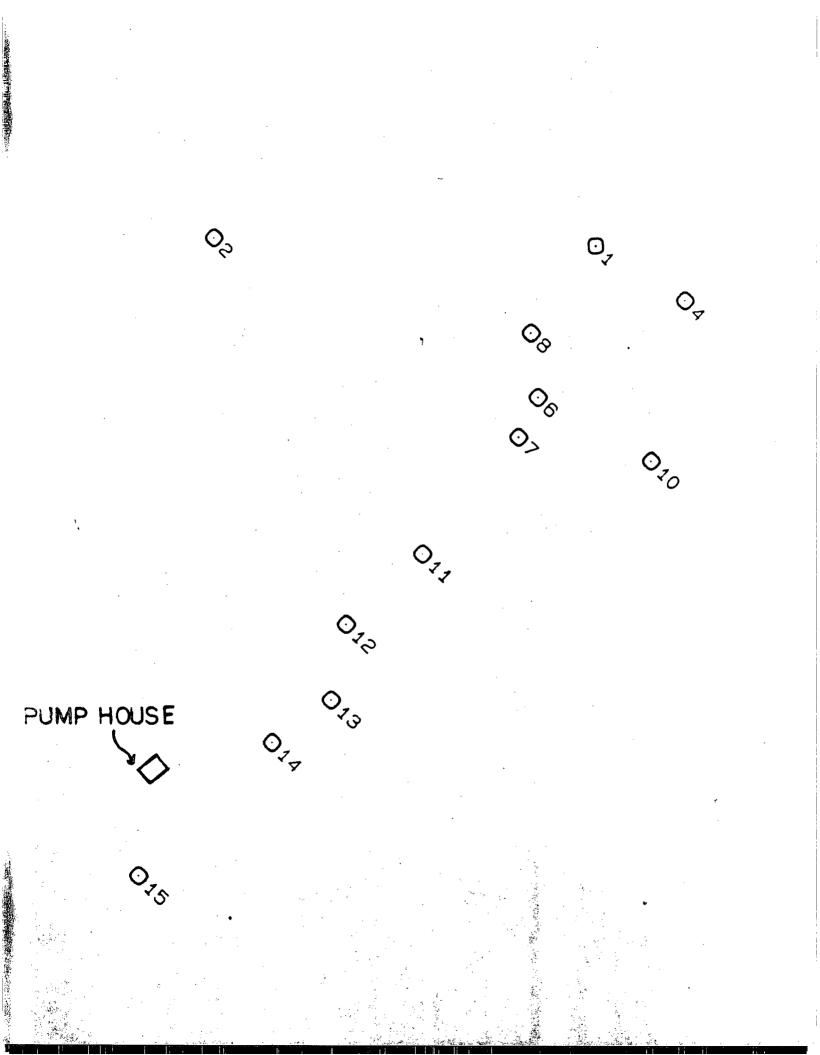


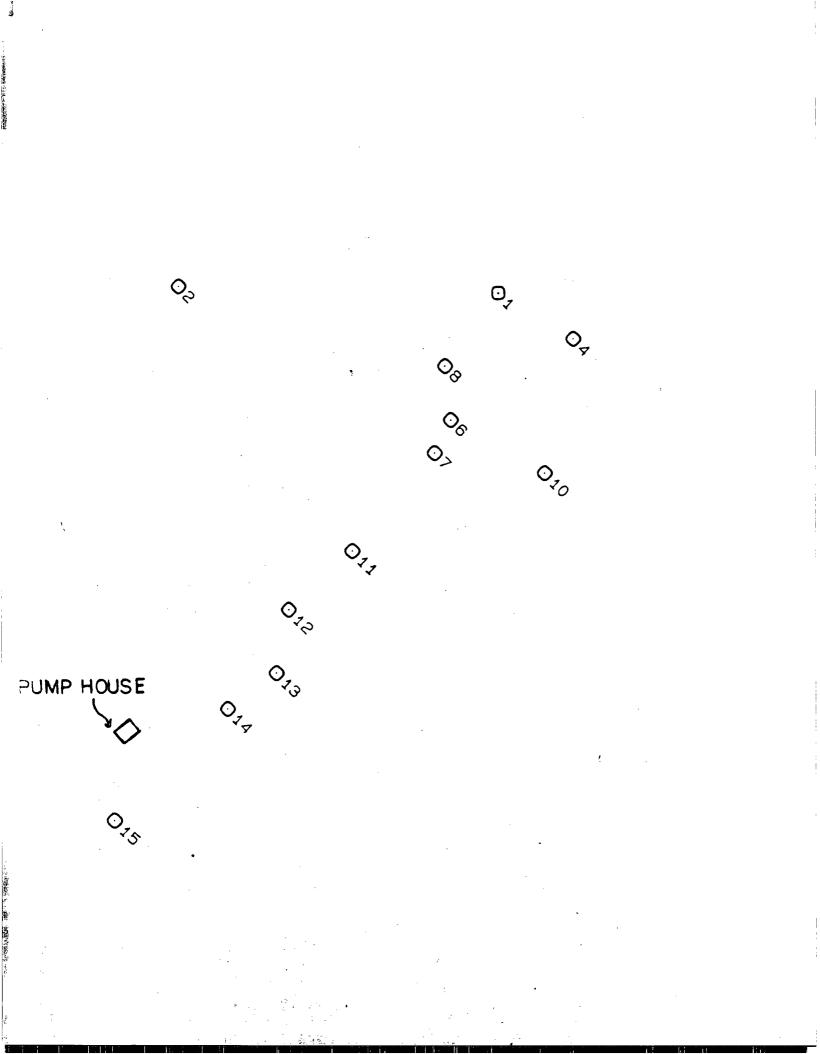


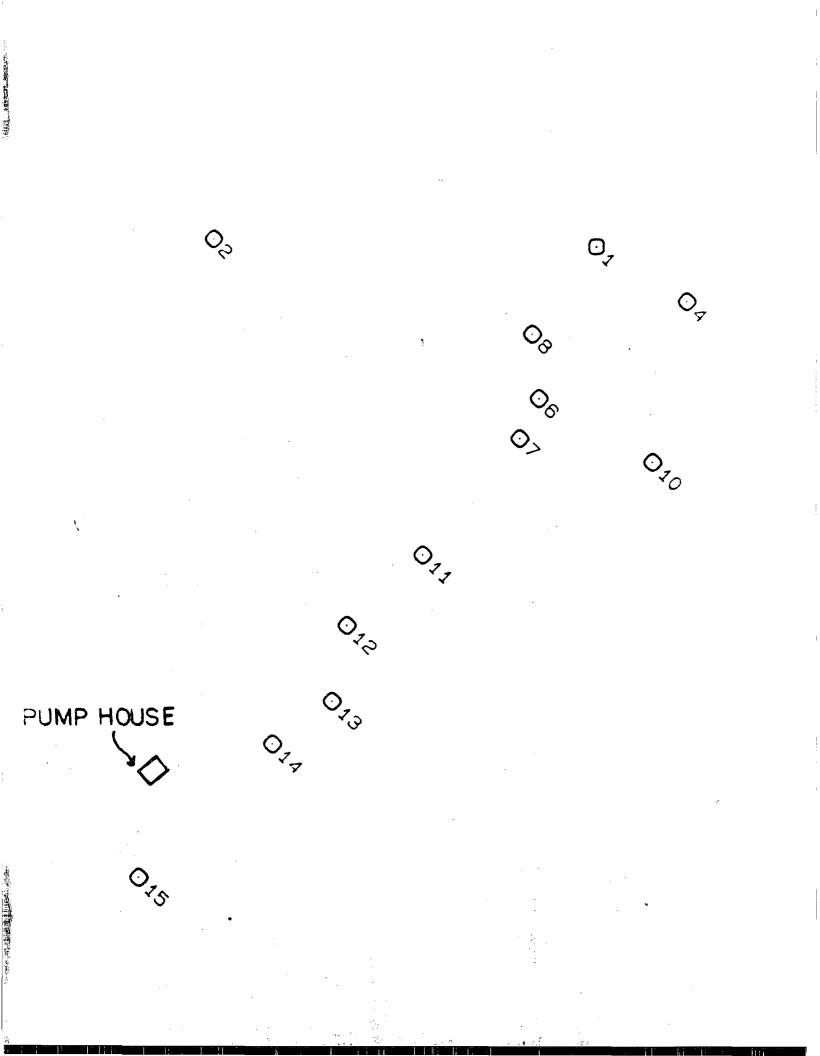


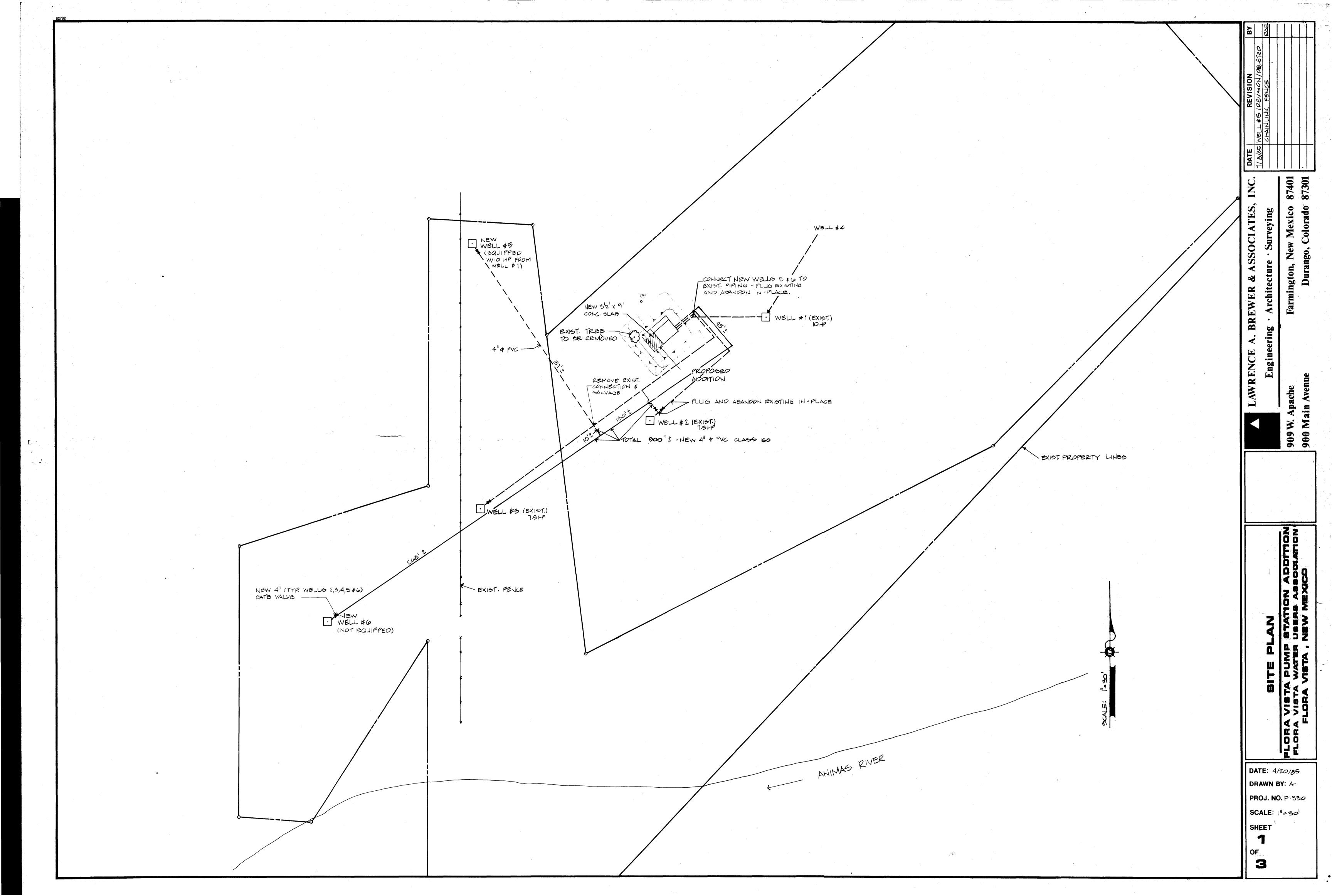


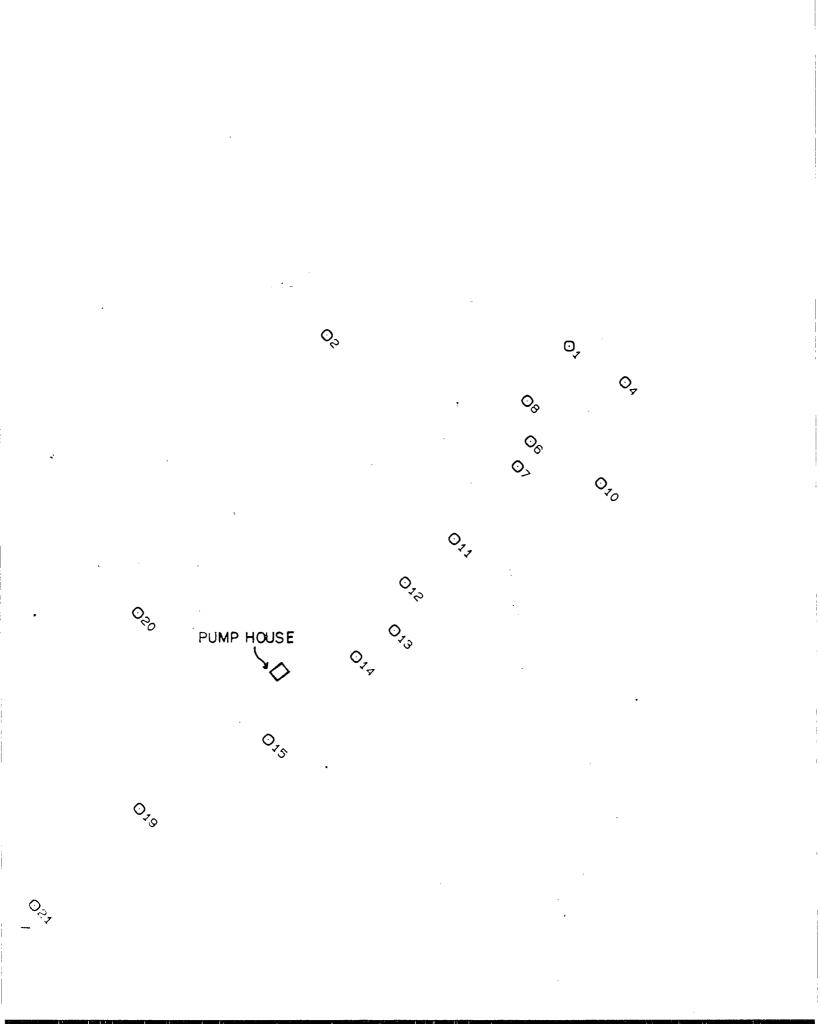












# FLORA VISTA WELL AND TEST WELLS LOCATIONS

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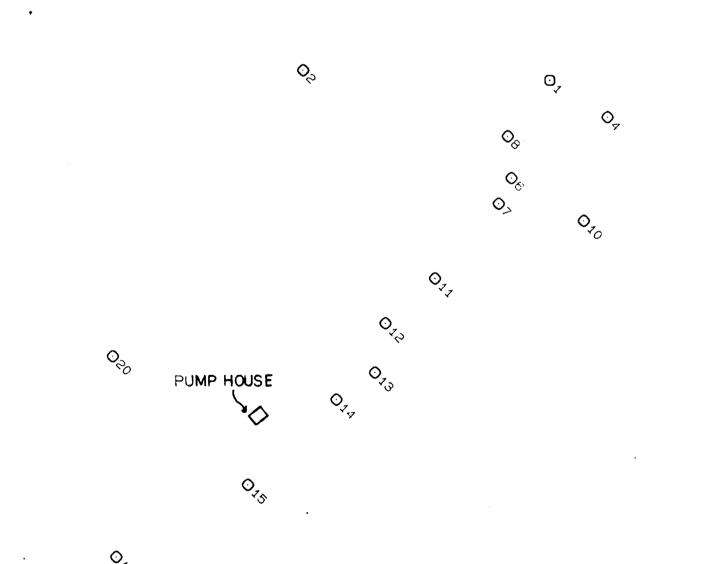
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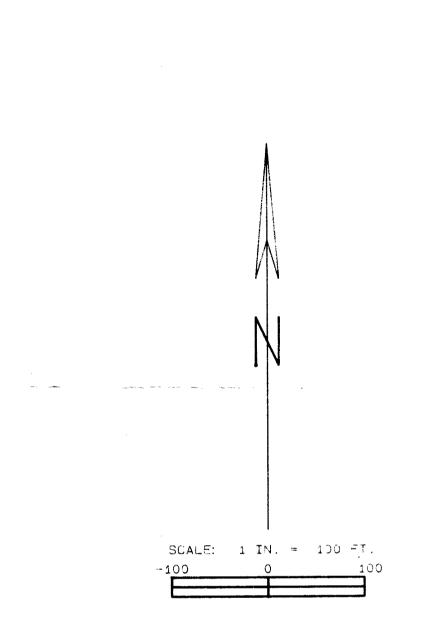
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	JDW 9/13/85 .ORA VISTA WELL'S TES NORTH	ST LOCATION EAST
1	10000	10000
2	10010.7201674	9744.2245555
3	10558.0800542	10148.890608
4	9961.833901	10059.8338515
5	9939.4073038	10083.6995464
6	9898.6519976	9959.7176712
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20	9714.4714054	9546.2960701
21	9401.6061461	9438.5830708
22	9535.9786329	9542.4122932
50	9180.7791972	12105.5650402
	SET IRON ROD AT	RIVER

BM, 100, ON SLAB AT WELL 6

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					PROJ. F	
					SCALE: SHEET	1"=100'
					Revise	R
			•			128/61

Manana Gas, Inc. P.O. BOX 36990 ALBUQUERQUE, NEW MEXICO 87176

ED HARTMAN, PRES.

TELE: (505) 884-4863 (505) 8<del>84-0814</del>

December 7, 1985

New Mexico Conservation Division 1000 Rio Brazos Rd. Aztec, N.M. 87410 Oil Conservation Division P.O. Box 2088 Santa Fe, N.M. 87501

Gentlemen:

Enclosed are the following Produced Water Pit Registration forms:

Name of Well Sullivan # 1 Aunt Maggie # 1 Aunt Maggie # 3 Phyllis Hartman # 1 Betty Hartman # 1 Annie B # 1Charlie # 1 Bobbie Herrera # 1 (P.C.) Bobbie Herrera # 1 (Fruit) Mary Ackroyd # 1 Mary Ackroyd # 2 Finch # 1 Clarence # 1 Gale # 1_R Mary Wheeler # 1 Mary Wheeler # 1-E Gigi # 1

Location J-25-29N-11W I_25_29N_11W J-25-29N-11W N-24-29N-11W P-14-30N-12W N-13-30N-12W I-7-30N-11W K-4-30N-11W  $K_4_3 ON_1 W$ J-18-30N-11W J-18-30N-11W A_22_29N_11W C-24-29N-11W M-13-29N-11W  $D_{-23} - 30N_{-1}2W$ M-23-30N-2W 12.00 M-24-29N-11W

Please advise if you need any further information in this matter.

Very truly yours Ed Hartman EMH/nh Encl. DEC 10.005 DEC 10.005 DIST. 3 STATE OF NEW MEXICO Energy and Minerals Department

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OIL CONSERVATION DIVISION P. O. Box 2088 Santa Fe, New Mexico 87501 (505) 827-5800 AZTTC DISTRICT OFFICE 1000 Rio Brazos Road Aztec, New Mexico 87410 (505) 334-6178

M E B B IVE

DEC 1 0 1985

OIL CON. DIV.

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PRODUCED WATER PIT RECISTRATION FORM (Instructions on Back)

Owner/Operator: Manana Gas, Inc., P.O. Box 36990, Albuquerque, N.M. 87176 (List information only for pits operated by you at a lease or at other locations) Owner/Operator: Well and Lease, or Facility Name: <u>Mary Wheeler #1-E</u> Location: Unit M Sec. 23-30N-12W (A) (B) (C)(D) (E) Pit Maximum Daily Pit Type Depth to Sample of Discharge to Discharge to Each Ground Water Each Pit Pit TDS (in mg/l) Sample Date or conductivity & temperature Primary Pit/

 Produced Water Pit
 5 bpd*
 Fiberglass
 7 ft. (est.)
 37,050
 9/25/85

 Separator
 6 Deep
 12 Diameter

 Ancillary Pit(s)

* phonecall from Ed Hartman, pres. 12/30/85 ASB Est. by Trucker Volume



MEMORANDUM OF MEETING OR CONVERSATION

Time 9:20AM Date Telephone 12/30/85 Personal Originating Party Other Parties Mañana Gas-Ed Vave Boyer HaRImar Subject ischarge volume Discussion The pit registration form for the above well how no volume Hartman says was 5661 on pumper 74 even Volumes Mary Whe have Tank in noc nce Now cracker no Te pump ~ lena Ę, replac 4 DA Conclusions or Agreements Distribution Flora Vista Sile P:T Fert Repistration Sile Signed





P. O. BOX 2079 • FARMINGTON, NM 87499 • (505) 327-3303 400 PILE ST., SUITE 300 • CLOVIS, NM 88101 • (505) 763-4255 DURANGO, COLORADO • (303) 247-5766

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November 8, 1985

Mr. Dave Boyer State Of New Mexico Oil Conservation Division PO Box 2088 Santa Fe, New Mexico 87504

RE: FLORA VISTA WATER USERS ASSOCIATION ELEVATIONS FOR TEST WELL MONITORS

Dear Mr. Boyer:

We are submitting herewith two (2) copies each of the elevations which we have taken to the nearest 1/100th of a foot for the above referenced project. In addition, we are also enclosing two (2) copies each of the area outlining that portion of the Crouch Mesa County Road Extension within the Lee Acres Landfill area.

Very truly yours,

BREWER ASSOCIATES, INCORPORATED

Robert A. Echols, Jr., P.E. Engineering Division Director

RAE:rr F-331

Enclosures

Lawrence A. Brewer, P.E., L.S. George E. Bennett, Jr., A.I.A. George T. Walters, P.L.S. Robert A. Echols, Jr., P.E.



MEMORANDUM OF MEETING OR CONVERSATION

Time 983D Date Telephone 9/9/85 Personal Originating Party Other Parties Dave Boyer-octo sonta se Charles Gholson - at HZJec TESTING OF FLORA VISCA Manana Gas Well Subject Discussion Head Space Test For FORA Vista Monitor Well #5 Shows 1256 ppm methane - Charlie previously examined manana reiondo and some dros in production; = Bradenhead Ter 10 - Charle a bealt w es mechanical integriling - Location of in swampy area long in area of al may mean source is shallow ander splan of organic Conclusions or Agreements Distribution Flora Visto File Signed

50 YEARS



TONEY ANAYA

GOVERNOR

STATE OF NEW MEXICO ENERGY AND MINERALS DEPARTMENT OIL CONSERVATION DIVISION



POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87501 (505) 827-5800

## MEMORANDUM

TO R. L. STAMETS, DIRECTOR OCD

FROM D. G. BOYER, GEOLOGIST, ENVIRONMENTAL BUREAU

DATE AUGUST 26, 1985

SUBJECT: MEETING WITH RICHARD CHENEY (BREWER AND ASSOCIATES) AND BERT BARNES (PRESIDENT, FLORA VISTA WATER USERS ASSOCIATION).

On Friday, August 23, I met with Cheney and Barnes to discuss results of OCD's investigation of the Flora Vista water well contamination, and to give them some suggestions as to where I think future investigation emphasis ought to be placed.

I described to them the work we have already accomplished (monitoring wells installed, and multiple samples taken from them and on-line water system), and results of our analysis (trace levels of toluene at 13 and 5 ppb in two monitor wells between the contaminated and supply well and pumping wells, and toluene at 6 ppb in the contaminated well itself). The levels of toluene detected are far below both the existing WQCC standard (15,000 ppb) and the proposed lower standard (250 ppb). The results are the only 1985 resuts that show any contamination whatsoever in any of the monitoring sites, or the contaminated well. One sample, taken in August 1983, by a former OCD employee and analyzed by a laboratory no longer in business, showed "oil and grease" hydrocarbon contamination, and no trace of benzene or toluene contamination at the detection levels available (10 ppb). Until July, 1985 the out-of-service contaminated well had no sanitary cover and only a bucket with a rock covered the ten inch diameter well. Therefore, there is no assurance that the contamination we are seeing now was not introduced during the past 2 1/2 years that the pump has been removed from the well.

This fact would surely be brought out in any future litigation.

I told them the USGS was interested in looking at the site for an aquifier test if the nearby pumping wells do not cause too much interference. The wells will need to be surveyed, and water levels obtained before the USGS makes a decision. Surveying will be performed by Cheney's firm in the next few weeks under my direction. Cheney and I agreed that heavy pumping of the contaminated well for at least several days is needed to determine the current status of water quality, and whether any hydrocarbon contamination would be captured if the well was put back in service.

Both Cheney and Barnes were upset by the lack of OCD (and EID) action in 1983. They estimate that the Water Users Association has spent approximately \$150,000 on new wells, land aquisitions, etc., and they are looking to recover damages from the responsible parties. They do have some evidence of their own indicating the source of the contamination (see Cheney letter of March 2, 1983, attached). Cheney talked about involving the State agencies (OCD/EID) in any lawsuit since it appears they did not perform their responsibilities in a timely manner. I told him I would have our attorney call their lawyer (Rick Lougee, 327-5281) in Farmington to discuss the issue. Cheney also mentioned possible legislation to require that all groundwater quality responsibilities be placed under one agency. My answer was that OCD/EID lines of authority are now better defined and that OCD agency response is now better. Though that may be the case now, Cheney said that the law may have to be modified to avoid future problems of this type.

As a result of this meeting, I plan to have Jeff Taylor call their attorney, and I intend to return to Farmington in the next several weeks to assist in well surveying, and continue development of the monitor wells (they still have sand and mud in some of them).

cc: Jeff Taylor

STATE OF NEW MEXICO MEMORANDUM OF MEETING OR CONVERSATION OIL CONSERVATION DIVISION Time Date X Telephone 8/5/85 Personal 10:30AM Originating Party Other Parties Dave RERT BARNES CC President Boyer 181a Water Users Assoc Vista Subject Sication of Einding No1 m D. ussion Barnet NATI m llaat We we an and Cen am. WENDUA Olhon M P SIN The nee 0 (i) Conclusions or Agreements 0 tion. E 50 tà (`h 011 Signed <u>Distribution</u> FLO 20 think Leyal Countel JESS Teylon Stemets - Dehey Ton Dick

State of New Mexico Environmental improvement division DISTRICT I FIELD OFFICE/724 WEST ANIMAS FARMINGTON, NEW MEXICO 87401 (505)327-9851

July 3, 1985

TONEY ANATA GOVERNOR

DENISE D. FORT DIRECTOR

Mr. Bert Barnes, President Flora Vista Water Users Association P.O. Box 171 Flora Vista, New Mexico 87415

Dear Mr. Barnes:

On June 20, 1985 the annual environmental survey of the Flora Vista Water Supply System was conducted. The survey is attached for your review and file.

The system supplies water to approximately 1,600 consumers. Currently the system is purchasing water from the City of Aztec and is pumping water from 2 wells. The pump for the chlorination feed line is broken, therefore much of the water is not being disinfected. Since the chlorinator is not working, Well #3 is not being used because it has sulfur contamination and there is no chlorine to remove the sulfur. Two new wells are drilled and will be put in use when the pump house renovations are done.

Following is a list of deficiencies noted during the inspection:

- There is no schedule for preventative maintenance and no records are kept of the maintenance which has been done nor who is doing it. Most manufacturers provide information with their product for scheduling preventative maintenance. Based on the number of complaints this office receives about sediment in the water, it seems that your system has quite a number of problems that a regular maintenance schedule could help to cut down. Historically, many problems that plague water systems could have been avoided if the system had had a preventative maintenance program.
- There is no method of keeping track of the equipment in the system whether it is in use or in the spare parts inventory.

Mr. Bert Barnes Page 2 July 3, 1985

- There is no product information available should it be needed in case of an emergency.
- The pump house expansion is anticipated this summer. When done, the fan switch for the chlorination room should be located outside the pump house. A signal light indicating fan operation should also be provided outside the entrance.
- The garden hose attached to the pipes in the pump house needs a backflow prevention device on it.
- The New Mexico Occupational Health and Safety Bureau strongly recommends that 26° Baume (56%) ammonia be provided at the pump house to check for chlorine gas leaks. The ammonia solution must not be stored in the room with the chlorine cylinders. It is also recommended that a self-contained breathing apparatus or gas mask system with oxygen tanks be provided should a chlorine leak occur. However, a written contingency plan outlining the responsibilities of the System and the Flora Vista Fire Department, and agreed upon by both in the case of a leak or similar type emergency would be acceptable by the Division. Further information can be obtained from Sam Rogers, Technical Services Section, Occupational Health and Safety Bureau. To ensure that the needs of the system and its employees are met, I recommend that you utilize the Occupational Health and Safety Bureau's Onsite Consultation Services. A pamphlet explaining the service is enclosed.

I would like to thank Ray Penrod and Rex Rosenbaugh for their time and help during the survey. If I can be of any assistance, please contact me at 327-9851.

Sincerely, Carol M Maller

Carol Miller, Environmental Associate

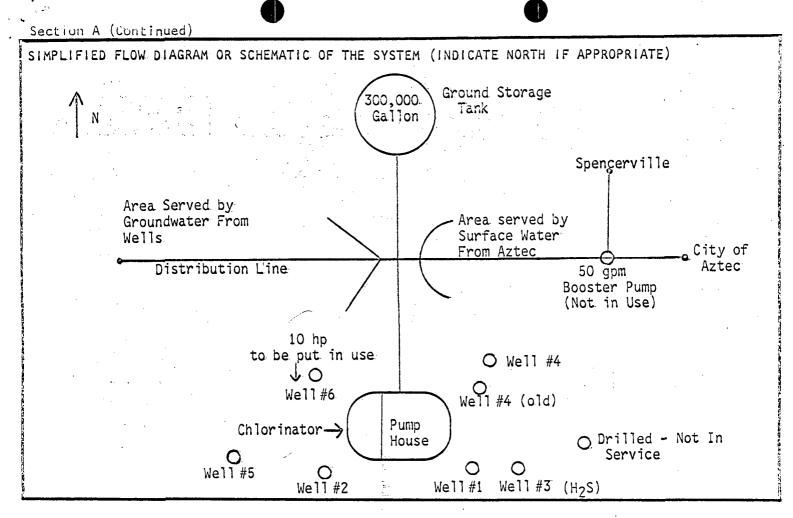
Enclosure CM:lm cc:Water Supply Section File

N-Av - Not A Est Estim	vailable ated			Inst	pection Form	n
Section A - GE	NERAL INFORMATIC	H ENVI	CONMENT	ton t	Inspecti	on
			vironment Departm nprovement Divisio		Date:	06-20-85
	WATER SUPPLY SYSTEM NAME F		COUNTY:	San Juan		
System Address						
OWNER:	<u>P.0.</u>	Box 171, Flora N OWNER ADDRESS			<u></u>	PHONE
	ter Users Assoc.	1	(in diriere	int than above,	1	334-6045
Population	1#	# Meters	Max. Syste	m. Production	Average Sy	stem Produc-
Served	Connections		A	GPD	No Records	Kept tion GPD
1600	580	580	Poten.	Actual 🗌		
System Source		oution Only		Well(s) X	# of Wells	2
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	Qualifying Infor phosphate chemic		er to keon	chloning from	woonting wi	+ h
	on and manganese				reacting-wi	
	ouse late summer					
- <u></u>	- control panel		h-ninos		·	
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pump house.			<u>IIIIca alla m</u>		U_USE_MILII-	<del></del>
Wells #1 & 2	are in use			<u></u>		
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	minated with nat					
	e chlorination r					
	d towall at all t					
contained bre	athing apparatus			Certifica-		Certification
	el - Name/Classi	ification	tion Requ	lired	Obtained	<u></u>
Ray Penrod			-0-	······	-0-	
Rex Rosenba	ugh - Assistant		_0		-0-	
					· · · · · · · · · · · · · · · · · · ·	
						د و منها کر از محمد میک میزد و میک ۲۰ وارد و میک می ورد. منبع و منطقه از محمد میک میزد و میک ۲۰ وارد و میک می و
	· · · ·				Busin	ness Phone:
Information fu	rnished by:					•
Ray Penrod / 1	lanager					4-6045
Ray Penrod / 1 Survey Perform	lanager				Busir	•

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### Section B - Source Information

WELL Identification	Well Depth	Pump Depth	Pump Capacity GPM	Well Drawdown feet	Pump Type	Static Water Level	Age of Pump	Date Well Drilled
Well #1	25'	18'	45	5'-6'	Submers.	≃ 8'	5 yrs.	1930
Well #2	25'	20 <u>1</u> '	45	5'-6'	Submers.	≃ 8 ¹	5 yrs.	1980
Well #3	26'	21 '	90	10'	Submers.	= 8'	<u>1</u> yr.	3/84
Well #4	25'	N/A	20	N/A	Submers.	≃ _{8'}	<u>1_vr</u> .	1984
Well #5	25'	20'	90		Submers.	≈ 8'	to be install	d_1985
Well #6	25'	201	90		Submers.	≈ 8'	H.	1985
Old Well #4 (Cas Old Well #4 (Contaminated	25'	21'	45	15'	Submers.	<i>≃</i> 8'	2 yrs.	8/84
	umber ( each) (	Total Capacity	Remarks	, Deficier	ncies, and	Recommen	dat ions	1
Artesian Wells								
Springs								
Infiltration Galleries								

EID - 042 Revised 11/82

Page 2

Section C - Gravity Storage and Pressure Tank Reservoirs

0000	14112/0	corago ana	105501						
Gravity Stora Reservoir Ind		Storage Capacity (gallons)	Date Built	Exterior Condition		Cathodic Protection	Tank Openings Secured	Overflow Screened	Vent Screen
#1		300,000	1979	Good	Steel	No	Yes	No	yes
			· · ·						
Pressure Tank	I.D.	Volume (gallons)	Age	Exterior Condition	Additional	Informatio	n		
s. 1 . :									
							· · · · ·	<u></u>	
Remarks, Defi	ciencies	or Recommen	dations	5.					
Section D - W	1								
Booster Pump	Total Nu of Stati			ficien <u>ci</u> es ine centrif	or Recommer	idations.			
Stations	1	1 00 30	t in Us		agai pamp				
Type of Disin Facilities	fection	Chlorine G	as	Remar Purch	ks, etc. Ga ased water	s chlorinat	ion provi 1 at Aztec	ded at we	11s.
System Pressure	Max. 90 Min. 55	1	Remarks limits	s, etc. The on the amo	re may be p unt of wate	ressure pro r purchased	blems in from Azt	summer du ec.	e to
Pipe Materials in system		L Ga	arden ho	ose on line	luding cros s in pump b contaminatio	s-connectionouse needs	ons observ a vacuum	ved) breaker t	:0
Type of Distr	L(	<del>-Cement-A</del> \$be	stos) ravity			- whe	en wells e in use ^{ot}	:h 🗍	
	Next S		, 		······		e în use	· · ·	
Contaminant Micro-	ple Da	te Frequen	<u>cy  </u>		Rema	rks			
biological		>							
Turbidity									
Organics	7/87	<u>Ν/Λ</u> 3 yrs.					<u></u>		
<u>Inorganic</u> Radiological	3/86	$\frac{3 \text{ yrs.}}{4 \text{ yrs.}}$	{						
Secondary	1			<u>_</u>					
Section E - (	NENERAL OF	PERATION AN	D MAINT	TENANCE (O	е м)				
- Those this syst					Yes		No [X]		1
Are preventa						Yes	,	No X	
The importance indicate observed	ce of Cros erved cro	ss Connecti ch por such	on Cont ons abo	rol was di ove (Pipe M	scussed wit aterials in	h the Opera System)	for Yes	No No	
E10 - 042 Rev	vised 11/	82		<u></u>				P	age 3
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## Section F - SURFACE WATER SUPPLY INFORMATION

Source (check one)	Remarks or Deficiencies	N/A
Lake		Â
Stream		· .

## Section G - WATER TREATMENT PLANT INFORMATION

Unit Operations	Remarks or Deficiencies
Plant Intake Structure	
Plant Location (Siting)	
Pretreatment, Raw Water Storage or Presettling Reservoirs	
Coagulation - Sedimentation	n de la constante de la consta
Chemical Addition	
Filtration	
Other Treatment (Ion Ex- change, Softening,Reverse Osmosis, etc.)	N/A

Additional Comments or Remarks

Page 4

-ELD - OH2 Bouisad 11/82

 Jason Kellahin W. Thomas Kellahin Karen Aubrey KELLAHIN and KELLAHIN Attorney: at Law El Patio - 117 North Guadalupe Post Office Box 2265 Santa Fe, New Mexico 87504-2265

Telephone 982-4285 Area Code 505

May 2, 1985

Jeff Taylor, Esq. Oil Conservation Division P. O. Box 2088 Santa Fe, New Mexico 87504

"Hand Delivered"

Re: NMOCC Case 8224

Dear Mr. Taylor:

On behalf of Tenneco Oil Company, I am requesting that the OCD Staff provide to us the following post hearing documents concerning its ground water study of the Flora Vista site:

- 1. All field notes and data;
- Schematic of site, with all monitoring wells or pit locations, including the direction of gradient and survey points;
- 3. All chemical analysis reports from all laboratories and for any and all samples taken;
- 4. Copies of all correspondence, documents, notes, and data concerning the Flora Vista site, including but not limited to, the Manana Mary Wheeler No. 1 well from the date of first reported contamination, and of any Flora Vista well.

Very truly yours,

Original signed by W. THOMAS KELLAHIN

W. Thomas Kellahin

WTK:ca

cc: Mr. Richard L. Stamets Oil Conservation Division P. O. Box 2088 Santa Fe, New Mexico 87504

### KELLAHIN and KELLAHIN

Jeff Taylor, Esq. May 2, 1985 Page 2

cc: Jennifer Pruitt, Esq. Environmental Improvement Division P. O. Box 968 Santa Fe, New Mexico 87501

William F. Carr, Esq. Attorney at Law P. O. Box 2208 Santa Fe, New Mexico 87501

Perry Pearce, Esq. Montgomery Law Firm P. O. Box 2307 Santa Fe, New Mexico 87501

Millard F. Carr, Esq. Tenneco Oil Company P. O. Box 3249 Englewood, Colorado 80155

Mr. Marty Buys Tenneco Oil Company P. O. Box 3249 Englewood, Colorado 80155

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STATE OF NEW MEXICO
SCIENTIFIC LABORATORY DIVISION NUMERATING 19 1535 Concamino de Salud NE, Albuquerque, New Mexico 88 (505) 841-2500
REPORT TO: DAVID G BOYER S.L.D. NO.: OR- 657-F.B FILE CONSERVATION DIVISION DIVISION DIV DATE REC. : 7/03/85
P.O. BOX 2088 PHONE 827-5812
SANTA FE. NM 87501 USER CODE: 82235
CONTAINERS WHICH ACCOMPANY THIS FORM ARE COLLECTIVELY REFERED TO AS SAMPLE.
SUBMITTER: NM DIC CONSERVATION CODE
LOCATION: FLORA VISTA CODE TOWNSHIP RANGE SECTION TRACTS
SOURCE: MON. WELL   CODE
COLLECTED: 6/28/85 BY EARP/RAILEY CODE 8 30628/03560
SAMPLE TYPE: WATER SOIL OTHER CODE
NEAREST CITY: FLORA VISTA CODE
pH=; Conductivity=umho/cm at ^O C; Chlorine Risidual= Dissolved Oxygen=mg/l; Alkalinity=; Flow Rate=
Sampling Location, Methods and Remarks (i.e. odors, etc.) BRILED MONITOR WELL
I certify that the statements in this block accurately reflect the results
of my field analyses, observations and activities. O. Bala
Method of shipment to the Laboratory Kond Carned
This form accompanies
Containers are marked as follows to indicate preservation (circle):
NP: No preservation; sample stored at room temperature.
P-Ice Sample stored in an ice bath (not frozen).
$P-Na_2S_2O_3$ : Sample preserved with $Na_2S_2O_3$ to remove chlorine residual.
I (we) certify that this sample was transferred from
to at (location) on
(date & time) and that the statements in this block are correct
Evidentiary Seals: Not Sealed Intact: Yes No
Signatures
(we) certify that this sample was transferred from
to at (location) on
(date & time) and that the statements in this block are correct
Evidentiary Seals: Not Sealed Intact: Yes No
Signatures

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with	n the	e statements in this b	.ock. Reviewers	signa	ture	: KMenerhe	en l
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LABORTORY DRG 266 Atb REPORT TO: David G. Boye New Mexico Oil Conservation Division LAB NUMBER 3-22-85 P. 0. Box 2088 875 **85-0266** -C Santa Fe, NM SLD Users Code No. 82235 COLLECTIVELY REFERRED TO AS "SAMPLE". CONTAINERS WHICH THIS FORM ACCOMPANIES ARE CERTIFICATE OF FIELD PERSONNEL Sample Type: Water 🖾 Soil 🔲 Other Municipal sy tem Water Supply and/or Code No. Flora Vista Mon Well 2 ば on Juan City & County Flore Late Collected (date & time) 0930 By (name) umho/cm at — °C; Chlorine Residual= pH= 🔄 ; Conductivity= Dissolved Oxygen= 🧹 mg/l; Alkalinity= ; Flow Rate= Sampling Location, Methods & Remarks (i.e. odors etc.) Sample Grom jut lug priorts Mon Well Intallation, temple had unidentified orlor. Backhoe digging pit had hydroculic Shind leach I certify that the statements in this block accurately reflect the results of my field analyses, observations and activities. Signed A and FRey FRey Free and concur I certify that I witnessed these field analyses, observations and activities and concur with the statements in this block. Signed Method of Shipment to Laboratory Hand Carrie THIS FORM ACCOMPANIES Q septum vials with teflon-lined discs identified as: specimen _____; duplicate _____; triplicate _____; blank(s) _____; and _____amber glass jug(s) with teflon-lined cap(s) identified as and ______other container(s) (describe) _______identified as ______identified as _______identified as ______identified as _______identified as _______identified as ______identified as ______identified other container(s) (describe) identified as Containers are marked as follows to indicate preservation (circle): NP: No preservation; sample stored at room temperature ( $\sim 20$  °C). P-ICE) Sample stored in an ice bath. P-Na₂0₂S₂: Sample preserved with 3 mg  $Na_2O_2S_2/40$  ml and stored at room temperature. CERTIFICATE(S) OF SAMPLE RECEIPT I (we) certify that this sample was transferred from ______ to _____at (location)_____ on (date & time) _____ and that the statements in this block are correct. Disposition of Sample______. Seal(s) Intact: Yes 🗖 No 🔲 📜 Signature(s) I (we) certify that this sample was transferred from _____ to _____ at (location) on (date & time) and that the statements in this block are correct. Disposition of Sample______. Seal(s) Intact: Yes 🗖 No 🔲 🛛 Signature(s)

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STATE OF NEW MEXICO
SCIENTIFIC LABORATORY DIVISION TETETTIV 700 Camino de Salud NE, Albuquerque; New Mexico 87106
ENVIRONMENT
REPORT TO: $DAVID G_{+} BCYFR$ S.L.D. No.: $OR - \frac{653 - 14 - 16}{653 - 14 - 16}$
CONSERVINEW MEXICO OIL CONSERVATION DIV DATE REC. : 7/05/85
PHONE 847-5872
SANTA FE, NM 87501 CONTAINERS WHICH ACCOMPANY THIS FORM ARE COLLECTIVELY REFERED TO AS SAMPLE.
SUBMITTER: NM OIC CONSERVATION OV CODE
LOCATION: FORA VISTA CODE TOWNSHIP RANGE SECTION TRACTS
SOURCE: MON WELL 2 CODE AQUIFER DEPTH
COLLECTED: 6/28/85 BYEARP/BAILEY CODE 850628101019
SAMPLE TYPE: WATER SOIL OTHER CODE
NEAREST CITY: FLORA VISTA CODE
pH=; Conductivity=umho/cm at°C; Chlorine Risidual=
Dissolved Oxygen=mg/l; Alkalinity=; Flow Rate=
Sampling Location, Methods and Remarks (i.e. odors, etc.) BAILED MONITOR WELL Z
NATE WELL Z
I certify that the statements in this block accurately reflect the results
of my field analyses, observations and activities.
Method of shipment to the Laboratory fond carried
This form accompanies
Containers are marked as follows to indicate preservation (circle):
NP: No preservation; sample stored at room temperature.
P-Ice Sample stored in an ice bath (not frozen).
$P-Na_2S_2O_3$ : Sample preserved with $Na_2S_2O_3$ to remove chlorine residual.
I (we) certify that this sample was transferred from
to at (location) on
(date & time) and that the statements in this block are correct
Evidentiary Seals: Not Sealed Intact: Yes No
Signatures
(we) certify that this sample was transferred from
to at (location) on on
(date & time)and that the statements in this block are correct Evidentiary Seals: Not Sealed Intact: Yes No
Signatures

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ANALYSES REQUESTED LAB: No.: ORG- 653							
REQUIRED. WHENEVER POSSIBLE LIST SPECIFIC COMPOUNDS SUSPECTED OR REQUIRED.	<b>)</b>						
PURGEABLE SCREENS SCREENS							
ALIPHATIC: HYDROCARBON. SCREEN       ALIPHATIC. HYDROCARBONS         X       X         AROMATIC       HYDROCARBON. SCREEN         X       X         HALOGENATED       HYDROCARBON. SCREEN         CHLORINATED       HYDROCARBON. SCREEN         CHLOROPHENOXY       ACID         HALOGENATED       HYDROCARBON. SCREEN         CHLOROPHENOXY       ACID         HALOGENATED       HYDROCARBON. SCREEN	· • •						
ORGANOPHOSPHATE PESTICIDES ORGANOPHOSPHATE PESTICIDES POLYCHLORINATED BIPHENYLS ( POLYNUCLEAR AROMATIC HYDROC TRIAZINE HERBICIDES							
SPECIFIC COMPOUNDS SPECIFIC COMPOUNDS							
REMARKS :	······································						
ANALYTICAL RESULTS							
COMPOUND EPPBI, COMPOUND EPPB halogeneted purge non delector Kenzene NiD. Tolsene 5 ethylbenzene NiD.	]						
REMARKS: Come Rive Corre Deat Mois also	m						
detected that may be acetone							
Seal(s) Intact: Yes NO Seal(s) broken by: date: I certify that I followed standard laboratory procedures on handling and analysis of this sample unless otherwise noted and that the statements in this block and the analytical data on this page accurately reflect the analytical results for this sample. Date(s) of analysis: 9 millings. Analyst's signature: I certify that I have reviewed and concur with the analytical results for this sample and with the statements in this block. Reviewers signature: May May May May May May May May May May							

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LABORTORY ORF 261 A+B REPORT TO: David G. Boye New Mexico Oil Conservation Division LAB NUMBER 3-22-85 P. 0. Box 2088 85-0267 -Santa Fe, NM 8750 Code No. 82235 ALL CONTAINERS WHICH THIS FORM ACCOMPANIES ARE COLLECTIVELY REFERRED TO AS "SAMPLE". CERTIFICATE OF FIELD PERSONNEL Sample Type: Water 🖾 Soil 🗖 Other Water Supply and/or Code No. Flore Vista Munerrial, Mon. 10200#2 City & County FLOR Vista, San Juran Collected (date & time) 1030, 3/20 85 By (name) Rough pH=____; Conductivity=____umho/cm at____°C; Chlorine Residual=__ Sampling Location, Methods & Remarks (i.e. odors etc.) 1.1.2 Sample Snomintdug prior to mon well installation. ) orlard noted. Backhoe leakinghydraulie slu I certify that the statements in this block accurately reflect the results of my field analyses, observations and activities. Signed Roy I certify that I witnessed these field analyses, observations and activities and concur with the statements in this block. Signed Method of Shipment to Laboratory Hang Corried THIS FORM ACCOMPANIES septum vials with teflon-lined discs identified as: specimen_____; duplicate_____; triplicate____; blank(s)____ and____amber glass jug(s) with teflon-lined cap(s) identified as_____ and____other container(s) (describe)_____identified as______ Containers are marked as follows to indicate preservation (circle): NP: No preservation; sample stored at room temperature (~20°C). P-ICE: > Sample stored in an ice bath.  $P-Na_2O_2S_2$ : Sample preserved with 3 mg  $Na_2O_2S_2/40$  ml and stored at room temperature. CERTIFICATE(S) OF SAMPLE RECEIPT I (we) certify that this sample was transferred from _____ to _____ at (location)_____ on (date & time) ______ and that the statements in this block are correct. Disposition of Sample______. Seal(s) Intact: Yes 🗖 No 🔲 🔒 Signature(s)_____ I (we) certify that this sample was transferred from to _____at (location)_____ on (date & time) and that the statements in this block are correct. Disposition of Sample______. Seal(s) Intact: Yes 🗖 No  $\Box$  . Signature(s)

ANALYSES REQUESTED. PLEASE CHECK THE APPROPRIATE BOXES BELOW TO INDICATE THE TYPE OF ANALYTICAL SCREENS REQUIRED. WHENEVER POSSIBLE LIST SPECIFIC COMPOUNDS SUSPECTED OR REQUIRED.							
QUAL IT AT IVE	QUANTITATIVE	PURGEAI SCREEI		QUAL ITAT IVE	QUANT ITAT IVE	EXTRACTAB SCREENS	
		ALIPHATIC HYDROCARBON SCREEN				ALIPHATIC HYDROCARBONS	
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	HALOGENATED HYDROCARBON SCREEN			· .	CHLOROPHENOXY ACID HERBICIDES		
		GAS CHROMATOGRAPH/MASS SPECTROMETER			1	HYDROCARBON FUEL SCREEN	
				1	ORGANOPHOSPHATE PESTICIDES		
<b></b>				1	POLYCHLORINATED BIPHENYLS (PCB's) POLYNUCLEAR AROMATIC HYDROCARBONS TRIAZINE HERBICIDES		
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Sea	l(s)	Intact: YesNO_X.	Seal(s) broken	h by:		da	te <u>:</u>
I certify that I followed standard laboratory procedures on handling and analysis of this							
sample unless otherwise noted and that the statements in this block and the analytical data							
on this page accurately reflect the analytical results for this sample. Date(s) of analysis: 3-25-85 . Analyst's signature: A final sample.							
I certify that I have reviewed and concur with the analytical desults for this sample and							
with the statements in this block. Reviewers signature: Mayon here							
							/

STATE OF NEW MEXICO
700 Camino de Salud NE, Albuquerque, New Mexico 87106
* ENVIRONMENT 19 1935 (505) 841-2500
REPORT TO:DAVID G BOYER S.L.D. NO.: OR- 654-A
THE SNEW MEXICO OIL CONSERVATION DIV BALL REC.
P.O. BOX 2088 PHONE 827-5812
CONTAINERS WHICH ACCOMPANY THIS FORM ARE COLLECTIVELY REFERED TO AS SAMPLE.
SUBMITTER: NM OIL CONSERVATION DIV CODE
LOCATION: FLORA VISTA CODE TOWNSHIP RANGE SECTION TRACTS
SOURCE: MON. WELL 3 CODE CODE
COLLECTED: 6 / 28 / 85 BY EARP/BAILEY CODE QUICE 20093000
SAMPLE TYPE: WATER) SOIL OTHER CODE
NEAREST CITY: FLORA VISTA CODE
pH=; Conductivity=umho/cm atOC; Chlorine Risidual=
Dissolved Oxygen=mg/l; Alkalinity=; Flow Rate=
Sampling Location, Methods and Remarks (i.e. odors, etc.) BAILED MONITOR WELL
I certify that the statements in this block accurately reflect the results
of my field analyses, observations and activities Bailer
Method of shipment to the Laboratory Kand Carried
This form accompanies /_Septum Vials,Glass Jugs,
Containers are marked as follows to indicate preservation (circle):
NP: No preservation; sample stored at room temperature.
P-Ice) Sample stored in an ice bath (not frozen).
P-Na ₂ S ₂ O ₃ : Sample preserved with Na ₂ S ₂ O ₃ to remove chlorine residual.
I (we) certify that this sample was transferred from
to at (location) on
(date & time) and that the statements in this block are correct
Evidentiary Seals: Not Sealed Intact: Yes No
Signatures
(we) certify that this sample was transferred from
to at (location) on
(date & time) and that the statements in this block are correct
Evidentiary Seals: Not Sealed Intact: Yes No
Signatures

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	YSES REQUESTED E CHECK THE APPROPRIATE RED. WHENEVER" POSSIBLE			ATE	AB. No.: ORG- THE TYPE OF ANALYTICA SUSPECTED OR REQUIRED	L SCREENS
QUANTITATIVE	PURGEA	· · ·	QUAL ITAT IVE	QUANT ITAT IVE	EXTRACTAE SCREEN	• •
	ALIPHATIC HYDROCARBO	N SCREEN			ALIPHATIC HYDROCAR	BONS
XX	AROMATIC HYDROCARBON				CHLORINATED HYDROC	ARBON PESTICIDES
RTY	HALOGENATED HYDROCAR				CHLOROPHENOXY ACID	HERBICIDES
	GAS CHROMATOGRAPH/MA	SS SPECTROMETER			HYDROCARBON FUEL S	
			ŕ		ORGANOPHOSPHATE PE	
		· · · ·			POLYCHLORINATED BI	
					POLYNUCLEAR AROMAT	IC HYDROCARBONS
	·				TRIAZINE HERBICIDE	S
	SPECIFIC COMP	POUNDS			SPECIFIC COMP	OUNDS
						-
REMARK	 'S :			I	<u> </u>	
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0-	Xylene	Trace prh</td <td>-</td> <td></td> <td></td> <td>11000</td>	-			11000
			<b>*</b> I	DEŢE	CTION LIMIT	prog. 1
REM	ARKS: All samples	- Should Re	RU	kn	utted in chuts	ticate.
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ono	at almh and	d two a	FO		20 006	
I cert sample on this	) Intact: Yes NOX ify that I followed star unless otherwise noted s page accurately refrec ) of analysis: Turkur	and that the state	n by <u>:</u> proce atemen 1 resu	dure ts 1	da s on handling and and n this block and the for this_sample.	
I cert:	) of analysis <u>: / fullys</u> ify that I have tevieved he statements in this bl	and concur with	the	anal	ytical, results for th	us sample and
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LABORTORY DRG 265 ATB REPORT TO: David G. Boye New Mexico Oil Conservation Division LAB NUMBER 3-22-85 P. O. Box 2088 87501 85-0265 C Santa Fe, NM users Code No. 82235 ALL CONTAINERS WHICH THIS FORM ACCOMPANIES ARE COLLECTIVELY REFERRED TO AS "SAMPLE". CERTIFICATE OF FIELD PERSONNEL Sample Type: Water 🖾 Soil 🗖 Other Water Supply and/or Code No. Flora Visto Municipal MonWell #3 City & County FLORA Vista, Son Juren Collected (date & time) 1205 PM/3/20/2(By (name) Boyg / Reca pH=____; Conductivity=_____umho/cm at_____C; Chlorine Residual=__ ; Flow Rate=____ Dissolved Oxygen=/ mg/1; Alkalinity= Sampling Location, Methods & Remarks (i.e. odors etc.) Sample story Pit duy prior to mon. well installation No overs noted. Backhoe leaking hydraulic sheid I certify that the statements in this block accurately reflect the results of my field analyses, observations and activities. Signed Activities and Activities and concur I certify that I witnessed these field analyses, observations and activities and concur with the statements in this block. Signed Method of Shipment to Laboratory Hong concern THIS FORM ACCOMPANIES septum vials with teflon-lined discs identified as: specimen ______; duplicate _____; triplicate _____; blank(s)_____ and _____amber glass jug(s) with teflon-lined cap(s) identified as ______ and _____other container(s) (describe) ______identified as ______ Containers are marked as follows to indicate preservation (circle): NP: No preservation; sample stored at room temperature (~20°C). P-ICE: Sample stored in an ice bath.  $P-Na_2O_3S_2$ : Sample preserved with 3 mg  $Na_2O_3S_2/40$  ml and stored at room temperature. CERTIFICATE(S) OF SAMPLE RECEIPT I (we) certify that this sample was transferred from ______ to _____ at (Tocation)_____ on (date & time)_____ and that the statements in this block are correct. Disposition of Sample______. Seal(s) Intact: Yes 🗖 No 🔲 . Signature(s)_____ _____ I (we) certify that this sample was transferred from _______ to ______ at (location)______ on (date & time)______ and that the statements in this block are correct. Disposition of Sample______. Seal(s) Intact: Yes 🔲 No 🗍 . Signature(s)

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	SE.	SES REQUESTED CHECK THE APPROPRIATE			ATE	AB. No.: ORG- THE TYPE OF ANALYTICA SUSPECTED OR REQUIRED	L SCREENS
QUAL ITAT IVE	PURGEABLE SCREENS						
×	X	ALIPHATIC HYDROCARBON AROMATIC HYDROCARBON HALOGENATED HYDROCARB	SCREEN			ALIPHATIC HYDROCAR CHLORINATED HYDROC CHLOROPHENOXY ACID	ARBON PESTICIDES
		GAS CHROMATOGRAPH/MASS SPECTROMETER				HYDROCARBON FUEL S ORGANOPHOSPHATE PE POLYCHLORINATED BI POLYNUCLEAR AROMAT TRIAZINE HERBICIDE	CREEN STICIDES PHENYLS (PCB's) IC HYDROCARBONS
		SPECIFIC COMP	POUNDS			SPECIFIC COMP	OUNDS
		Benzens etc.					
REMAR	RKS			<u></u>	<u> </u>	·····	
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cron	2	IPOUND the purgeaflex nated purgeafles	[PPB] nore letic ted how letic ted	COMPOUND [PPB]		[PPB]	
			·····				
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		/*/ 0					
I cert sample on the Date(s I cert	tif e u is s) tif	Cf Intact: Yes NO X y that I followed star nless otherwise noted page accurately reflect of analysis: $3 - 25 - 5$ y that I have reviewed statements in this bl	and that the sta the analytical Analysi and concur with	n by <u>:</u> proce atemen l resu t's si n the	edure nts i ilts ignat anal	da es on handling and and in this block and the for this sample ture:	analytical data

LLIL STATE OF NEW MEXICO
SCIENTIFIC LABORATORY DIVISION ENVIRONMENT AUG 19 1933700 Camino de Salud NE, Albuquerque; New Mexico 87100 (505) 841-2500
REPORT TOUS OR- 655- A-B
NEW MEXICO OIL CONSERVATION DIV. DATE REC. : 7/03/85
P.O. BOX 2088 PHONE 827-5812
SANTA FE, NM 87501 USER CODE: 8 2 2 3 5
CONTAINERS WHICH ACCOMPANY THIS FORM ARE COLLECTIVELY REFERED TO AS SAMPLE.
SUBMITTER: NM OIL CONSERVATION DIN CODE
LOCATION: I-LORA VISTA CODE TOWNSHIP RANGE SECTION TRACTS
SOURCE: MON. WELL 4 CODE AQUIFER DEPTH
COLLECTED: 6/28/85 BY EARP/BAILEY CODE: 8 5 06 28 08 00 00 00
SAMPLE TYPE: WATER SOIL OTHER CODE
NEAREST CITY: FLORA VISTA CODE
pH=; Conductivity=umho/cm atC; Chlorine Risidual=
Dissolved Oxygen= mg/l; Alkalinity= ; Flow Rate=;
Sampling Location, Methods and Remarks (i.e. odors, etc.) Bailed man ^{(TO} R well
I certify that the statements in this block accurately reflect the results
of my field analyses, observations and activities.
Method of shipment to the Laboratory fond declivered
This form accompanies Septum Vials, Glass Jugs,
Containers are marked as follows to indicate preservation (circle):
NP: No preservation; sample stored at room temperature.
P-Ice Sample stored in an ice bath (not frozen).
$P-Na_2S_2O_3$ : Sample preserved with $Na_2S_2O_3$ to remove chlorine residual.
I (we) certify that this sample was transferred from
to at (location) on
(date & time) and that the statements in this block are correct
Evidentiary Seals: Not Sealed Intact: Yes No
Signatures
(we) certify that this sample was transferred from
to on on (date & time) and that the statements in this block are correct
Evidentiary Seals: Not Sealed Intact: Yes No
Signatures

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A	NAL'	YSES REQUESTED			Ľ	AB. No. : ORG-	655
PLEASE CHECK THE APPROPRIATE BOXES BELOW TO INDICATE THE REQUIRED. WHENEVER POSSIBLE LIST SPECIFIC COMPOUNDS SUSP							
<u>RE</u>	MOTK	ED. WHENEVER POSSIBLE	LIST SPECIFIC C		- [2]	SUSPECIED: OR: REQUIRED	
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rt JAU	QUANT	SCREE	NS	quAL T	QUANT TTAT	SCREEN	5
		ALIPHATIC HYDROCARBC		-	4. 1	ALIPHATIC HYDROCAR	
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$\underline{X}_{-}$	X	GAS CHROMATOGRAPH/MA		2.1	с ^а .	CHLOROPHENOXY ACID HYDROCARBON FUEL S	
		GAS CHROMATOGRAPH/TH	55. SPECIROIEIER		<u></u>	ORGANOPHOSPHATE PE	
						POLYCHLORINATED BI	
					1	POLYNUCLEAR AROMAT	
						TRIAZINE HERBICIDE	
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		nless otherwise noted					analytical data
on i Date	cnis a(s)	page accurately refle of analysis: 7 July	ST THE ANALYTICAL	resu	lts anst	for this sample.	
I Ce	ertif	y that I have reviewe	d and concur with	the:	anal	ytical results for th	nis sample and
with	n the	e statements in this b	Lock. Reviewers	signa	ture	: K. Menerhen	×
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		and a second					

LABORATORY ORG 264 Att REPORT TO: David G. Boye New Mexico Oil Conservation Division LAB NUMBER 3-22-85 P. 0. Box 2088 87 85 0264 C Santa Fe, NM SLD Users Code No. 82235 ALL CONTAINERS WHICH THIS FORM ACCOMPANIES ARE COLLECTIVELY REFERRED TO AS "SAMPLE". CERTIFICATE OF FIELD PERSONNEL Sample Type: Water Soil 🔲 Other Water Supply and/or Code No. Slova Vista Municipul, Mon Well #4 City & County FLOR Vite Jum By (name)_ ROYM Baca Collected (date & time) 1400 3/20/85 ____umho/cm at_____ °C; Chlorine'Residual= pH= ; Conductivity= ____; Flow Rate= Dissolved Oxygen= mg/l; Alkalinity= Sampling Location, Methods & Remarks (i.e. odors etc.) Sample grom Pit due mois to mon well installation No odory noted. Backhoe leaking hydrauli Fleeid I certify that the statements in this block accurately reflect the results of my field analyses, observations and activities. Signed Analyses, observations and activities and concur I certify that I witnessed these field analyses, observations and activities and concur with the statements in this block. Signed Method of Shipment to Laboratory THIS FORM ACCOMPANIES septum vials with teflon-lined discs identified as: specimen ; duplicate ; triplicate ; blank(s) and ___________; duplicate ; triplicate ; blank(s)________ and __________; duplicate ; triplicate ; blank(s)________ and _________; duplicate ; triplicate ; blank(s)_________; and ________; duplicate ; triplicate ; blank(s)________; and ________; duplicate ; triplicate ; blank(s)_______; and _______; duplicate ; triplicate ; blank(s)______; and _______; duplicate ; triplicate ; blank(s)______; and _______; duplicate ; triplicate ; blank(s)______; and ______; duplicate ; triplicate ; blank(s)_____; and ______; duplicate ; triplicate ; blank(s)_____; duplicate ; blank(s)____; duplicate ; blank(s)_____; duplicate ; blank(s)____; duplicate ; blank(s)___; duplicate ; blank(s)____; duplicate ; blank(s)___; duplicate ; blank(s)__; duplicate ; blank(s)___; duplicate ; blank(s)___; duplicate ; blank(s)__; duplicate ; blank(s)___; duplicate ; blank(s)__; duplicate ; blank(s)_; duplicate NP: No preservation; sample stored at room temperature ( $\sim 20^{\circ}$ C). P-ICE) Sample stored in an ice bath. P-Na20252: Sample preserved with 3 mg  $Na_2O_3S_2/40$  ml and stored at room temperature. CERTIFICATE(S) OF SAMPLE RECEIPT I (we) certify that this sample was transferred from to _____ at (location) on (date & time)_____ and that the statements in this block are correct. Disposition of Sample______. Seal(s) Intact: Yes 🗖 No 🔲 🐪 Signature(s) I (we) certify that this sample was transferred from _____ to _____ at (location)_____ on _____ and that the statements in this block are correct. (date & time) Disposition of Sample______. Seal(s) Intact: Yes 🗖 No 🗖 . Signature(s)

PL	EASE	YSES REQUESTED CHECK THE APPROPRIATE ED. WHENEVER POSSIBLE			ALE .		L-SCREENS -
QUALITATIVE	QUANTITATIVE			QUAL ITATIVE	QUANT ITAT IVE	EXTRACTAE	BLE
		ALIPHATIC HYDROCARBO	N SCREEN			ALIPHATIC HYDROCAR	BONS
	<u>\</u>	AROMATIC HYDROCARBON				CHLORINATED HYDROC	ألا والمناكدة فسيتقصص ويتبعن متبالي وجواها اليعاقب والمتحد
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<b></b>						ORGANOPHOSPHATE PE	
		<u></u>					
				-		POLYCHLORINATED BI	
						POLYNUCLEAR AROMAT	IC HYDROCARBONS
			·			TRIAZINE HERBICIDE	S
		SPECIFIC COMP	POUNDS			SPECIFIC COMP	POUNDS
		Sensena etc.					
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F	REMA	MPOUND whic pungeables anatal pungeables IRKS: M. pungeables CI	EPPBI rendetated in u Ues detected ERTIFICATE OF AN		OMI	POUND	[PPB]
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STATE OF NEW MEXICO 85-1656 B
ENVIRONMENT SIDE SCIENTIFIC LABORATORY DIVISION SCIENTIFIC LABORATORY DIVISION (505) 841-2500
REPORT TO: SANTA DAVID G. BOYFR S.L.D. NO.: OR- 656 - A
NEW MEXICO OF CONSERVATION DIV DATE REC. : 7/03/85
P.O. BOX 2088 PHONE 827-58/2
SANTA EF NM 97501 USER CODE: 82235
CONTAINERS WHICH ACCOMPANY THIS FORM ARE COLLECTIVELY REFERED TO AS SAMPLE.
SUBMITTER: NM DIL CONSERVATION DIV CODE
LOCATION: FLORA VISTA CODE TOWNSHIP RANGE SECTION TRACTS
SOURCE: MON. WELLS CODE AQUIFER DEPTH
COLLECTED: 6/28/85 BY EARCH CODE 85062808/0000
SAMPLE TYPE: WATER SOIL OTHER CODE
NEAREST CITY: FLORA VISTA CODE
pH=; Conductivity=umho/cm at°C; Chlorine Risidual=
Dissolved Oxygen=mg/l; Alkalinity=; Flow Rate=
Sampling Location, Methods and Remarks (i.e. odors, etc.) Barceo monitor, well
I certify that the statements in this block accurately reflect the results
of my field analyses, observations and activities.
Method of shipment to the Laboratory 400 camed
This form accompanies / Septum Vials,Glass Jugs,
Containers are marked as follows to indicate preservation (circle):
NP: No preservation; sample stored at room temperature.
P-Ice Sample stored in an ice bath (not frozen).
P-Na $_2$ S $_2$ O $_3$ : Sample preserved with Na $_2$ S $_2$ O $_3$ to remove chlorine residual.
· · · · · · · · · · · · · · · · · · ·
I (we) certify that this sample was transferred from
to at (location) on
(date & time)and that the statements in this block are correct
Evidentiary Seals: Not Sealed Intact: Yes No
Signatures
(we) certify that this sample was transferred from
to at (location) on
(date & time)and that the statements in this block are correct
Evidentiary Seals: Not Sealed Intact: Yes No
Signatures

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XX	HALOGENATED HYDROCAR	·			CHLOROPHENOXY ACII	
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	unless otherwise noted page accurately refle					analytical data
	of analysis: 7 July					
I certi	fy that I have reviewe	and concur with	the:	anal	ytical results for th	nis sample and
with the	e statements in this bi	lock. Reviewers	signa	ture	: 12 Meyerhen	<u></u>
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LABORATORY DRG 262 AYB REPORT TO: David G. Boye New Mexico Oil Conservation Division LAB: NUMBER 3-22-85 P. O. Box 2088 875c 85 0262 c Santa Fe, NM esers Code No. 82235 ALL CONTAINERS WHICH THIS FORM ACCOMPANIES ARE COLLECTIVELY REFERRED TO AS "SAMPLE". CERTIFICATE OF FIELD PERSONNEL Sample Type: 🛛 Water 🔯 🖉 Soil 🔲 Other Water Supply and/or Code No. SLove Dista Municipal, Mon Well #5 City & County PLona Vista Sen Junan Collected (date & time) 17-30, 3/20/85 By (name) Roya/ pH=____; Conductivity=_____umho/cm at_____°C; Chlorine Residual= ; Flow Rate= Dissolved Oxygen= mg/T; Alkalinity= Sampling Location, Methods & Remarks (i.e. odors etc.) Sample From Pitdue priorto mon well installation Backhoe leaking hydraulie skuid. I certify that the statements in this block accurately reflect the results of my field analyses, observations and activities. Signed Accurately reflect the results of my field I certify that I witnessed these field analyses, observations and activities and concur with the statements in this block. Signed Method of Shipment to Laboratory THIS FORM ACCOMPANIES 2 septum vials with teflon-lined discs identified as: specimen  $\lambda_{e}$ ; duplicate  $\chi$ ; triplicate; blank(s) and ______ amber glass jug(s) with teflon-lined cap(s) identified as ______ and _____ other container(s) (describe) _______ identified as ______ Containers are marked as follows to indicate preservation (circle): NP: No preservation; sample stored at room temperature ( $\sim 20^{\circ}$ C). P-TCE:) Sample stored in an ice bath. P-Na20352: Sample preserved with 3 mg  $Na_2O_2S_2/40$  ml and stored at room temperature. CERTIFICATE(S) OF SAMPLE RECEIPT I (we) certify that this sample was transferred from ______ to _____ at (location)_____ on (date & time)_____ and that the statements in this block are correct. Disposition of Sample______. Seal(s) Intact: Yes 🗖 No 🗖 Signature(s) I (we) certify that this sample was transferred from _______ to _____ at (location)_____ _____ On (date & time) ______ and that the statements in this block are correct. Disposition of Sample______. Seal(s) Intact: Yes 🔲 No 🔲 . Signature(s)

PL		YSES REQUESTED CHECK THE APPROPRIATE ED. WHENEVER POSSIBLE			ATE ]	AB. CRG-	L SCREENS	
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I co sam on Dato I co	CERTIFICATE OF ANALYTICAL PERSONNEL Seal(s) Intact: Yes NON. Seal(s) broken by: date: I certify that I followed standard laboratory procedures on handling and analysis of this sample unless otherwise noted and that the statements in this block and the analytical data on this page accurately reflect the analytical results for this sample. Date(s) of analysis: 3-25-85 Analyst's signature: K I certify that I have reviewed and concur with the analytical results for this sample and with the statements in this block. Reviewers signature:							
						V	/	

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LABOBATORY ORG 275 A REPORT TO: David G. Boye New Mexico Oil Conservation Division LAB NUMBER 3-22-85 87501 85-0275 B P. 0. Box 2088 Santa Fe, NM SLD Users Code No. 82235 ALL CONTAINERS WHICH THIS FORM ACCOMPANIES ARE COLLECTIVELY REFERRED TO AS "SAMPLE". CERTIFICATE OF FIELD PERSONNEL Sample Type: Water 🔲 Soil 🗹 Other Water Supply and/or Code No. Soil Sample From wil at Mon Well # 5 City & County FLova Vista Son Juan Collected (date & time) 1435, 3/20/85 By (name) 13049/2 pH=____; Conductivity=_____umho/cm at ____°C; Chlorine Residual=___ Dissolved Oxygen= _____mg/1; Alkalinity=____ ; Flow Rate=____ Sampling Location, Methods & Remarks (i.e. odors etc.) Sample of black oily layer 39-49" deep. Mason Tor copped with Foil I certify that the statements in this block accurately reflect the results of my field analyses, observations and activities. Signed Thomas Thomas I certify that I witnessed these field analyses, observations and activities and concur with the statements in this block. Signed PLLP 2. Saca Method of Shipment to Laboratory annel THIS FORM ACCOMPANIES septum vials with teflon-lined discs identified as: specimen____; duplicate____; triplicate____; blank(s)____ and____amber glass jug(s) with teflon-lined cap(s) identified as_____ and *I* other container(s) (describe) Mahon ian identified as Containers are marked as follows to indicate preservation (circle): No preservation; sample stored at room temperature (~20°C). NR: P-ICE: Sample stored in an ice bath. Compared OFSzlP-Na₂O₃S₂: Sample preserved with 3 mg Na₂O₃S₂/40 ml and stored at room temperature. CERTIFICATE(S) OF SAMPLE RECEIPT I (we) certify that this sample was transferred from _____ ___ to _____ at (location)_____ on (date & time)______ and that the statements in this block are correct. Disposition of Sample______. Seal(s) Intact: Yes 🔲 No 🔲 . I (we) certify that this sample was transferred from ______ to _____ at (location) on (date & time)_____ and that the statements in this block are correct. Disposition of Sample______. Seal(s) Intact: Yes 🔲 No 🔲 . Signature(s)

	YSES REQUESTED CHECK THE APPROPRIATE ED. WHENEVER POSSIBLE			ATE 7	AB. ORG-	L SCREENS		
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* DETECTION LIMIT         REMARKS: N, extremely alphitic dutate         CERTIFICATE OF ANALYTICAL PERSONNEL         Seal(s) Intact: Yes NO Seal(s) broken by: date:         I certify that I followed standard laboratory procedures on handling and analysis of this sample unless otherwise noted and that the statements in this block and the analytical data on this page accurately reflect the analytical results for this sample.         Date(s) of analysis: 4/n/s       . Analyst's signature: 1. Mally         I certify that I have reviewed and concur with the analytical results for this sample and with the statements in this block. Reviewers signature: A Magnum								

STATE OF NEW MEXICO
ENVIRONMENT ENVIRONMENT TILG 19 1985 Camino de Salud NE, Albuquerque, New Mexico 87106 (505) 841-2500
REPORT TO:BOYFR S.L.D. NO.: OR-1057-H.B.
P.0. BOX 2088 PHONE 827-5812
SANTA FE, NM 87501 USER CODE: 8 2 3 5
CONTAINERS WHICH ACCOMPANY THIS FORM ARE COLLECTIVELY REFERED TO AS SAMPLE.
SUBMITTER: NM OIL CONSERVATION ON CODE
LOCATION: FLORA VISTA CODE TOWNSHIP RANGE SECTION TRACTS
SOURCE: MUNICIPAL WELL #1 CODE AQUIFER DEPTIN
COLLECTED: 6/28/85 BY BOYER BAILERCODE 8506281/3028
SAMPLE TYPE: WATER SOIL OTHER CODE
NEAREST CITY: FLORA VISTA CODE
pH=; Conductivity=umho/cm at°C; Chlorine Risidual=
Dissolved Oxygen=mg/l; Alkalinity=; Flow Rate=;
Sampling Location, Methods and Remarks (i.e. odors, etc.) BRICED OPEN WELL-NO SANITARY SEAL 10"ID STEEL CASING.
I certify that the statements in this block accurately reflect the results
of my field analyses, observations and activities.
Method of shipment to the Laboratory Hand cameb 7
This form accompaniesSeptum Vials,Glass Jugs,
Containers are marked as follows to indicate preservation (circle):
NP: No preservation; sample stored at room temperature.
P-Ice Sample stored in an ice bath (not frozen).
$P-Na_2S_2O_3$ : Sample preserved with $Na_2S_2O_3$ to remove chlorine residual.
I (we) certify that this sample was transferred from
to at (location) on
(date & time) and that the statements in this block are correct
Evidentiary Seals: Not Sealed Intact: Yes No
Signatures
(we) certify that this sample was transferred from
to at (location) on
(date & time) and that the statements in this block are correct
Evidentiary Seals: Not Sealed Intact: Yes No
Signatures

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Date(s)	of analysis: Phile	285. Knalyst	:'s si	gnat	ure: Attur	- (Bharus
L certify	y that I have reviewed	1 and concur with	1 the	anal	ytical results for a	this sample and
with the	Statements in this hi	OCK ROVIONOTO	gim-	ture	· · · · · · · · · · · · · · · · · · ·	
with the	statements in this bl	LOCK. Reviewers	signa	ture	: T. Meyerhen	

LABORTORY ORF 263 AtB REPORT TO: David G. Boye New Mexico Oil Conservation Division LAB NUMBER C. S. S. S. P. O. Box 2088 3-22-85 85-0263-6 Santa Fe, NM SLD Users Code No. 82235 CONTAINERS WHICH THIS FORM ACCOMPANIES ARE COLLECTIVELY REFERRED TO AS "SAMPLE". CERTIFICATE OF FIELD PERSONNEL Sample Type: Water Soil 🔲 🛛 Other Water Supply and/or Code No. FLore Viste Municator Supply Well #1 City & County Slova Vista, San, Juan Collected (date & time) <u>1310, 3/20/85</u> By (name) <u>7</u> pH=___; Conductivity=____umho/cm at____°C; Chlorine Residual=__ ____; Flow Rate= Dissolved Oxygen= mg/l; Alkalinity= Sampling Location, Methods & Remarks (i.e. odors etc.) Sample at start of pumping, no odors presrousle contaminated with hydrocarbout, discharge at gasoline I certify that the statements in this block accurately reflect the results of my field analyses, observations and activities. Signed Activities and concur I certify that I witnessed these field analyses, observations and activities and concur with the statements in this block. Signed the statements in this block. Method of Shipment to Laboratory specimen_____; duplicate_____; triplicate_____; blank(s)_____ and_____amber glass jug(s) with teflon-lined cap(s) identified as______ and_____other container(s) (describe)______identified as______ Containers are marked as follows to indicate preservation (circle): NP: No preservation; sample stored at room temperature (~20°C). P-ICE:) Sample stored in an ice bath.  $P-Na_2O_2S_2$ : Sample preserved with 3 mg  $Na_2O_2S_2/40$  ml and stored at room temperature. CERTIFICATE(S) OF SAMPLE RECEIPT I (we) certify that this sample was transferred from _____ to _____at (location) on (date & time)______ and that the statements in this block are correct. Disposition of Sample______. Seal(s) Intact: Yes 🗖 No 🗌 🛛 Signature(s) ------_____ I (we) certify that this sample was transferred from ______ to ______ at (location)_____ on (date & time)______ and that the statements in this block are correct. Disposition of Sample______. Seal(s) Intact: Yes 🗖 No 🔲 🔒 Signature(s)

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		HALOGENATED HYDROCAR				CHLOROPHENOXY ACID	HERBICIDES
		GAS CHROMATOGRAPH/MAS	SS SPECTROMETER			HYDROCARBON FUEL S	
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l co with	ertif n the	y that I have reviewed statements in this bl	l and concur with Lock. Reviewers	h the signa	anal ature	lyticat results for the	his sample and
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LABORATORY DRG 26/AHL REPORT TO: David G. Boye Conservation Division New Mexico Oi LAB NUMBER 3-22-85 P. 0. Box 2088 855 0 /2/5 Santa Fe, NM 8750 users Code No. 82235 COLLECTIVELY REFERRED TO AS "SAMPLE". ALL CONTAINERS WHICH THIS FORM ACCOMPANIES ARE CERTIFICATE OF FIELD PERSONNEL Sample Type: Water 🛛 Soil 🔲 Other upply Well I Water Supply and/or Code No. FLora Vista Municip City & County FLORA VISTO, Son Juan Collected (date & time) 1538, 3/20/ By (name) pH=___; Conductivity= 600 umho/cm at 12.5°C; Chlorine Residual= -___; Flow Rate= 7961 Dissolved Oxygen= ___ mg/l; Alkalinity=___ Sampling Location, Methods & Remarks (i.e. odors etc.) Sample prionts affation of pumping previous ontopinated with hydrocard I certify that the statements in this block accurately reflect the results of my field analyses, observations and activities. Signed Arth Koy A 1 certify that I witnessed these field analyses, with the statements in this block. Signed Method of Shipment to Laboratory Hone carried THIS FORM ACCOMPANIES 2 septum vials with teflon-lined discs identified as: x ; duplicate ; triplicate ; blank(s) specimen and _____amber glass jug(s) with teflon-lined cap(s) identified as and other container(s) (describe) identified as Containers are marked as follows to indicate preservation (circle): AHP: No preservation; sample stored at room temperature ( $\sim 20$  °C). P-ICE Sample stored in an ice bath. P-Na,0,S,: Sample preserved with 3 mg  $Na_2O_2S_2/40$  ml and stored at room temperature. CERTIFICATE(S) OF SAMPLE RECEIPT I (we) certify that this sample was transferred from to _____at (location) on (date & time)_____ and that the statements in this block are correct. Disposition of Sample______. Seal(s) Intact: Yes 🗖 No 🔲 🔒 Signature(s) I (we) certify that this sample was transferred from to _____at (location) on and that the statements in this block are correct. (date & time) Disposition of Sample______. Seal(s) Intact: Yes 🗖 No 🔲 Signature(s)

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LABORATORY DRG 259 AXB REPORT TO: David G. Boye New Mexico Oil Conservation Division LAB. NUMBER 3-22-85 P. 0. Box 2088 85-0259-02 875 Santa Fe, NM users Code No. 82235 ALL CONTAINERS WHICH THIS FORM ACCOMPANIES ARE COLLECTIVELY REFERRED TO AS "SAMPLE". CERTIFICATE OF FIELD PERSONNEL Sample Type: Water 🖾 Soil 🗖 Other Water Supply and/or Code No. 51 Die Viete Mun Supply Composite #273 City & County FLORA Vieta San Juan Cty Collected (date & time) 1626 3/20/85 By (name)____ Rough pH=  $\checkmark$ ; Conductivity= 670 umho/cm at M °C; Chlorine Residual= -Sampling Location, Methods & Remarks (i.e. odors etc.) Pumphouse spigot, They wells off, No oclors I certify that the statements in this block accurately reflect the results of my field analyses, observations and activities. Signed I certify that I witnessed these field analyses, observations and activities and concur with the statements in this block. Signed Method of Shipment to Laboratory Hand corrich THIS FORM ACCOMPANIES 2 septum vials with teflon-lined discs identified as: specimen X; duplicate X; triplicate ; blank(s) and _____amber glass jug(s) with teflon-lined cap(s) identified as and ______other container(s) (describe)_______identifi identified as Containers are marked as follows to indicate preservation (circle): NP_: No preservation; sample stored at room temperature ( $\sim 20$  °C). P-ICE? Sample stored in an ice bath.  $P-Na_2O_3S_2$ : Sample preserved with 3 mg  $Na_2O_3S_2/40$  ml and stored at room temperature. CERTIFICATE(S) OF SAMPLE RECEIPT I (we) certify that this sample was transferred from to _____ at (location) on (date & time) · · · · and that the statements in this block are correct. Disposition of Sample______. Seal(s) Intact: Yes 🗖 No 🗔 🛛 Signature(s) I (we) certify that this sample was transferred from ______ to ______ at (location)_____ on (date & time) and that the statements in this block are correct. Disposition of Sample______. Seal(s) Intact: Yes 🗖 No 🗖 . Signature(s)

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LABOR DRY DRG 268 AYB David G. Boye REPORT TO: New Mexico Oil Conservation Division LAB NUMBER 3-22-85 P. 0. Box 2088 8750 85 0268 C Santa Fe, NM Storoser's Code No. 82235 ALL CONTAINERS WHICH THIS FORM ACCOMPANIES ARE COLLECTIVELY REFERRED TO AS "SAMPLE". CERTIFICATE OF FIELD PERSONNEL Sample Type: Water 🖾 Soil 🔲 Other Water Supply and/or Code No. Slow Visto Mun, Supply Woll \$5 Son Juan Ctel City & County FLOR Vista Collected (date & time) 1555 3/20/85 By (name) Royn // °C; Chlorine Residual= pH= -; Conductivity= G/D umho/cm at Sampling Location, Methods & Remarks (i.e. odors etc.) Pumphouse spinot, other well, off. no odorf This well nearest previously contam well I certify that the statements in this block accurately reflect the results of my field analyses, observations and activities. Signed <u>A and Z Roy</u> I certify that I witnessed these field analyses, observations and activities and concur with the statements in this block. Signed **Fig. F. Bacq** Method of Shipment to Laboratory THIS FORM ACCOMPANIES 2 septum vials with teflon-lined discs identified as: specimen_____; duplicate_____; triplicate_____; blank(s)____; and _____amber glass jug(s) with teflon-lined cap(s) identified as and _____other container(s) (describe) ______ identifi identified as Containers are marked as follows to indicate preservation (circle): NP: No preservation; sample stored at room temperature (~20°C). P-ICE) Sample stored in an ice bath. P-Na203S2: Sample preserved with 3 mg  $Na_2O_2S_2/40$  ml and stored at room temperature. CERTIFICATE(S) OF SAMPLE RECEIPT I (we) certify that this sample was transferred from _______ to _____ at (location) on (date & time) _____ and that the statements in this block are correct. Disposition of Sample . Seal(s) Intact: Yes 🔲 🛛 No 🗖 . Signature(s) _____ I (we) certify that this sample was transferred from ______ to _____ at (location)__ on (date & time) ______ and that the statements in this block are correct. Disposition of Sample______. Seal(s) Intact: Yes 🗖 No 🔲 . Signature(s)

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with th	e statements in this bl	ock. Reviewers	signa	iture	: K Mener has	re sambre aug
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REPORT TO:	David G. D	Soyer	LABORATOR	DRG	213 1
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	Nau Mexico Oil Nau Mexico Oil P.O. Bax 208 Squita Fe, M-	8 820		`	2-85
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City & County	<b>y</b>		• •		
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analyses, obs I certify the	at the statements servations and act at I witnessed the tements in this bl	ivities. Signed se field analyses			-
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ALYTICAL RESULTS fro NA: No acid added. ALYTICAL RESULTS fro NF, NA Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other:	(Non-filtered Other- <i>specify</i> : <b>STAPLES</b>	)0.45 μn : Units Date analy: μmho5/9	A Calcium (00915) Calcium (00915) Calcium (00930) Calcium (00930) Colum (00930) Colum (00930) Colum (00930) Colum (00930) Colum (00940) Columination (00940) Columinat	$ \begin{array}{c}                                     $	Units mg/l mg/l mg/l mg/l mg/l	4/15 5.90 4/15 5.90 3/28 3/28 3/28 4/15 4/15 4/15 4/15 5/6
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ALYTICAL RESULTS fro NF, NA Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Other: Other: Other: Other: NF, NA Nitrate-N + , Nitrate-N	(Non-filtered Other- <i>specify</i> : <b>STAPLES</b>	)0.45 μn : mhoS/9 mg/l5/2g	Teed       NA       P         A       Calcium (00915)       Agnesium (00925)         A       Sodium (00930)       Potassium (00935)         Bicarbonate (00440)       Chloride (00940)         Chloride (00945)       Total filterable residu (dissolved) (70300)         Other:       Clour 42         F, A-H2 SO4       Nitrate-N + , Nitrate	$ \begin{array}{c}                                     $	Units mg/l mg/l mg/l mg/l mg/l mg/l	4/15 4/15 5.90 3/28 3/28 3/28 3/28 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 5/6 5/6 5/6 5/6 5/6 5/6 5/6 5/
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ubmitted.	(Non-filtered Other- <i>specify</i> : <b>STAPLES</b>	)0.45 μn : 	Teed       NA         X       Calcium (00915)         X       Calcium (00915)         X       Calcium (00925)         X       Sodium (00930)         Y       Potassium (00935)         X       Bicarbonate (00440)         X       Chloride (00940)         X       Sulfate (00945)         X       Total filterable residu         (dissolved) (70300)       Other:         F, A-H2 SO4       Image: Solved (00631)         X       Nitrate-N + , Nitrate         dissolved (00631)       Ammonia-N dissolv         (00608)       Image: Solved (00631)	/03 9.4 29.9 0.39 0.39 0.148.3 1/7.0 183.4 365 003 005 005 005 005 005 005 005 005 00	Units mg/l mg/l mg/l mg/l mg/l mg/l	4/15 4/15 5.90 3/28 3/28 3/28 3/28 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 5/6 5/6 5/6 5/6 5/6 5/6 5/6 5/
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ubmitted       A         Image: Notation of the state o	(Non-filtered Other- <i>specify</i> : <b>STAPLES</b>	) 0.45 μm : <u>Units Date analy:</u> mmho <u>S/9</u> mg/l mg/l mg/l mg/l	Total Filter         A         Calcium (00915)         A         Calcium (00915)         A         Calcium (00915)         A         Calcium (00935)         Sodium (00935)         Bicarbonate (00440)         Chloride (00940)         Sulfate (00945)         Total filterable residu (dissolved) (70300)         Other:         F, A-H ₂ SO ₄ Nitrate-N + , Nitrate- dissolved (00631)         Ammonia-N dissolv (00608)         Total Kjeldahl-N (	/03 9.4 29.9 0.39 0.39 0.148.3 1/7.0 183.4 365 003 005 005 005 005 005 005 005 005 00	Units mg/l mg/l mg/l mg/l mg/l	4/15 4/15 5.90 3/28 3/28 3/28 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 5/6 5/6 5/6 5/6 5/6 5/6 5/6 5/
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ALYTICAL RESULTS fro NF, NA Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Othe	Other-specify:       SS9       7:58	) 0.45 μm : <u>Units Date analy:</u> mmho <u>S/9</u> mg/l mg/l mg/l mg/l	Total Filter         A         Calcium (00915)         A         Calcium (00915)         A         Calcium (00915)         A         Calcium (00935)         Sodium (00935)         Bicarbonate (00440)         Chloride (00940)         Sulfate (00945)         Total filterable residu (dissolved) (70300)         Other:         F, A-H ₂ SO ₄ Nitrate-N + , Nitrate- dissolved (00631)         Ammonia-N dissolv (00608)         Total Kjeldahl-N (	/03 9.4 29.9 0.39 0.39 0.148.3 1/7.0 183.4 365 003 005 005 005 005 005 005 005 005 00	Units mg/lmg/l mg/lmg/l mg/l mg/l mg/l mg/l	4/15 5.90 7/15 5.90 3/28 3/28 3/28 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15
ALYTICAL RESULTS fro   ALYTICAL RESULTS fro   NF, NA   Conductivity (Corrected)   25°C (00095)   Total non-filterable   residue (suspended)   (00530)   Other:	Other-specify:       SS9       7:58	) 0.45 μm : <u>Units Date analy:</u> mmho <u>S/9</u> mg/l mg/l mg/l mg/l	Teed       NA         X       Calcium (00915)         X       Calcium (00915)         X       Calcium (00915)         X       Calcium (00925)         X       Sodium (00935)         X       Bicarbonate (00440)         X       Chloride (00945)         X       Total filterable residu (dissolved) (70300)         X       Other:         F, A-H2 SO.       Nitrate-N +, Nitrate- dissolved (00631)         Ammonia-N dissolv (00608)       Total Kjeldahl-N (         Other:       Other:	/03 9.4 29.9 0.39 0.39 0.39 0.39 0.39 0.39 0.35 77.0 78.3.4 365 76 76 7.55 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.0 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0	Units mg/l _ mg/l _	4/15 5.90 7/15 5.90 3/28 7/28 7/6 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 5/6 4/15 5/6 4/15 5/6 4/15 5/6 5/6 5/6 5/6 5/6 5/6 5/6 5/
ALYTICAL RESULTS fro NF, NA Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other	Other-specify:       SS9       7:58	) 0.45 μm : <u>Units Date analy:</u> mmho <u>S/9</u> mg/l mg/l mg/l mg/l	Teed       NA         X       Calcium (00915)         X       Calcium (00915)         X       Calcium (00915)         X       Calcium (00925)         X       Sodium (00935)         X       Bicarbonate (00440)         X       Chloride (00945)         X       Total filterable residu (dissolved) (70300)         X       Other:         F, A-H2 SO.       Nitrate-N +, Nitrate- dissolved (00631)         Ammonia-N dissolv (00608)       Total Kjeldahl-N (         Other:       Other:	/03 9.4 29.9 0.39 0.39 0.39 0.39 0.39 0.39 0.35 77.0 78.3.4 365 76 76 7.55 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.0 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0	Units mg/l _ mg/l _ mg/l _ mg/l _ mg/l _ mg/l _ mg/l _ mg/l _ mg/l _	4/15 5.90 7/15 5.90 3/28 3/28 3/28 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15
ALYTICAL RESULTS fro NF, NA Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other: Other	Other-specify:       SS9       7:58	) 0.45 μm : <u>Units Date analy:</u> mmho <u>S/9</u> mg/l mg/l mg/l mg/l	Teed       NA         X       Calcium (00915)         X       Calcium (00915)         X       Calcium (00915)         X       Calcium (00925)         X       Sodium (00935)         X       Bicarbonate (00440)         X       Chloride (00945)         X       Total filterable residu (dissolved) (70300)         X       Other:         F, A-H2 SO.       Nitrate-N +, Nitrate- dissolved (00631)         Ammonia-N dissolv (00608)       Total Kjeldahl-N (         Other:       Other:	/03 9.4 29.9 0.39 0.39 0.39 0.39 0.39 0.39 0.35 77.0 78.3.4 365 76 76 7.55 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.0 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0	Units mg/l _ mg/l _	4/15 5.90 7/15 5.90 3/28 7/28 7/6 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 5/6 4/15 5/6 4/15 5/6 4/15 5/6 5/6 5/6 5/6 5/6 5/6 5/6 5/
Submitted       A         IALYTICAL RESULTS fro         IALYTICAL RESULTS fro         IALYTICAL RESULTS fro         NF, NA         Conductivity (Corrected)         25°C (00095)         Total non-filterable         residue (suspended)         (00530)         Other:	Other-specify:       SS9       7:58	) 0.45 μm : <u>Units Date analy:</u> mmho <u>S/9</u> mg/l mg/l mg/l mg/l	Teed       NA         X       Calcium (00915)         X       Calcium (00915)         X       Calcium (00915)         X       Calcium (00925)         X       Sodium (00935)         X       Bicarbonate (00440)         X       Chloride (00945)         X       Total filterable residu (dissolved) (70300)         X       Other:         F, A-H2 SO.       Nitrate-N +, Nitrate- dissolved (00631)         Ammonia-N dissolv (00608)       Total Kjeldahl-N (         Other:       Other:	/03 9.4 29.9 0.39 0.39 0.39 0.39 0.39 0.39 0.35 77.0 78.3.4 365 76 76 7.55 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.0 7.0 7.55 7.0 7.55 7.0 7.55 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0 7.0	Units mg/l _ mg/l _	4/15 5.90 7/15 5.90 3/28 7/28 7/6 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 4/15 5/6 4/15 5/6 4/15 5/6 4/15 5/6 5/6 5/6 5/6 5/6 5/6 5/6 5/

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NALYTICAL RESULTS fro         NF, NA         Conductivity (Corrected)         25 °C (00095)         Total non-filterable         residue (suspended)         (00530)         Other:         Other:         Other:         Other:         Nitrate-N +, Nitrate-N         total (00630)         Ammonia-N total (00610)	Other-specify: m SAMPLES Units Date an µmho mg/l	alyzed F, NA Calcium (00915) Calcium (00925) Sodium (00930) Potassium (00935) Bicarbonate (00440) Chloride (00940) Sulfate (00945) Total filterable residue- (dissolved) (70300) Other:	mg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/l	
NALYTICAL RESULTS fro         NF, NA         Conductivity (Corrected)         25°C (00095)         Total non-filterable         residue (suspended)         (00530)         Other:         Other:         Other:         Other:         Nitrate-N +, Nitrate-N         total (00630)         Ammonia-N total (00610)	Other-specify: m SAMPLES Units Date an 	aiyzed         F, NA                Calcium (00915)             Sodium (00925)             Sodium (00930)             Potassium (00935)             Bicarbonate (00440)             Chloride (00940)             Chloride (00945)             Chloride (00631)             Chloride (00631)             Ammonia-N dissolved	mg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/l	4/5
NALYTICAL RESULTS fro         NF, NA         Conductivity (Corrected)         25°C (00095)         Total non-filterable         residue (suspended)         (00530)         Other:         Other:         Other:         Other:         Other:         Nitrate-N +, Nitrate-N         total (00630)         Ammonia-N total (00610)         Total Kjeldahl-N         (         Chemical oxygen	Other-specify:           m SAMPLES           Units Date an	aiyzed         F, NA <ul></ul>	mg/img/img/img/img/img/img/img/img/img/img/img/img/img/img/img/img/i	4/5 4/22
NALYTICAL RESULTS fro         NF, NA         Conductivity (Corrected)         25°C (00095)         Total non-filterable         residue (suspended)         (00530)         Other:	Other-specify: m SAMPLES Units Date an 	alyzed       F, NA <ul> <li>Calcium (00915)</li> <li>Magnesium (00925)</li> <li>Sodium (00930)</li> <li>Potassium (00935)</li> <li>Bicarbonate (00440)</li> <li>Chloride (00940).</li> <li>Sulfate (00945)</li> <li>Total filterable residue (dissolved) (70300)</li> <li>Other:</li> <li>F, A-H₂ SO₄.</li> <li>Nitrate-N +, Nitrate-N dissolved (00631)</li> <li>Ammonia-N dissolved: (00608)</li> <li>Total Kjeldahl-N ()</li> <li>Nitrate-N + ()</li> </ul>	mg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/l	4/5
NALYTICAL RESULTS fro         NF, NA         Conductivity (Corrected)         25°C (00095)         Total non-filterable         residue (suspended)         (00530)         Other:	Other-specify:           m SAMPLES           Units Date an	aiyzed         F, NA <ul></ul>	mg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/l	4/5 4/22
NALYTICAL RESULTS fro         NF, NA         Conductivity (Corrected)         25°C (00095)         Total non-filterable         residue (suspended)         (00530)         Other:	Other-specify:           m SAMPLES           Units Date an           //mho	alyzed       F, NA <ul> <li>Calcium (00915)</li> <li>Magnesium (00925)</li> <li>Sodium (00930)</li> <li>Potassium (00935)</li> <li>Bicarbonate (00440)</li> <li>Chloride (00940).</li> <li>Sulfate (00945)</li> <li>Total filterable residue (dissolved) (70300)</li> <li>Other:</li> <li>F, A-H₂ SO₄.</li> <li>Nitrate-N +, Nitrate-N dissolved (00631)</li> <li>Ammonia-N dissolved: (00608)</li> <li>Total Kjeldahl-N ()</li> <li>Nitrate-N + ()</li> </ul>	mg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/l	4/5 4/22 5/10
NALYTICAL RESULTS fro         NF, NA         Conductivity (Corrected)         25°C (00095)         Total non-filterable         residue (suspended)         (00530)         Other:         Chemical (00630)         Ammonia-N total (00610)         Total Kjeldahl-N         (         Chemical oxygen         demand (00340)         Total organic carbon         (         Other:         Other:	Other-specify:           m SAMPLES           Units Date an           //mho	alyzed       F, NA <ul> <li>Calcium (00915)</li> <li>Magnesium (00925)</li> <li>Sodium (00930)</li> <li>Potassium (00935)</li> <li>Bicarbonate (00440)</li> <li>Chloride (00940).</li> <li>Sulfate (00945)</li> <li>Total filterable residue (dissolved) (70300)</li> <li>Other:</li> <li>F, A-H₂ SO₄.</li> <li>Nitrate-N +, Nitrate-N dissolved (00631)</li> <li>Ammonia-N dissolved (00608)</li> <li>Total Kjeldahl-N ()</li> <li>Other:</li> </ul>	<u> </u>	4/5 4/22 5/10
NALYTICAL RESULTS fro         NF, NA         Conductivity (Corrected)         25°C (00095)         Total non-filterable         residue (suspended)         (00530)         Other:	Other-specify:           m SAMPLES           Units Date an           //mho	alyzed       F, NA <ul> <li>Calcium (00915)</li> <li>Magnesium (00925)</li> <li>Sodium (00930)</li> <li>Potassium (00935)</li> <li>Bicarbonate (00440)</li> <li>Chloride (00940).</li> <li>Sulfate (00945)</li> <li>Total filterable residue (dissolved) (70300)</li> <li>Other:</li> <li>F, A-H₂ SO₄.</li> <li>Nitrate-N +, Nitrate-N dissolved (00631)</li> <li>Ammonia-N dissolved (00608)</li> <li>Total Kjeldahl-N ()</li> <li>Other:</li> </ul>	mg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/l	4/5 4/5 4/22 5/10
NALYTICAL RESULTS fro         NF, NA         Conductivity (Corrected)         25°C (00095)         Total non-filterable         residue (suspended)         (00530)         Other:	Other-specify:           m SAMPLES           Units Date an           //mho	alyzed       F, NA <ul> <li>Calcium (00915)</li> <li>Magnesium (00925)</li> <li>Sodium (00930)</li> <li>Potassium (00935)</li> <li>Bicarbonate (00440)</li> <li>Chloride (00940).</li> <li>Sulfate (00945)</li> <li>Total filterable residue (dissolved) (70300)</li> <li>Other:</li> <li>F, A-H₂ SO₄.</li> <li>Nitrate-N +, Nitrate-N dissolved (00631)</li> <li>Ammonia-N dissolved (00608)</li> <li>Total Kjeldahl-N ()</li> <li>Other:</li> </ul>	mg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/l	4/5 4/22 5/10
NALYTICAL RESULTS fro         NF, NA         Conductivity (Corrected)         25°C (00095)         Total non-filterable         residue (suspended)         (00530)         Other:         Other:         Other:         Other:         Other:         Nitrate-N +, Nitrate-N         total (00630)         Ammonia-N total (00610)         Total Kjeldahl-N         (         Chemicat oxygen         demand (00340)         Total organic carbon         (	Other-specify:           m SAMPLES           Units Date an           //mho	alyzed       F, NA <ul> <li>Calcium (00915)</li> <li>Magnesium (00925)</li> <li>Sodium (00930)</li> <li>Potassium (00935)</li> <li>Bicarbonate (00440)</li> <li>Chloride (00940).</li> <li>Sulfate (00945)</li> <li>Total filterable residue (dissolved) (70300)</li> <li>Other:</li> <li>F, A-H₂ SO₄.</li> <li>Nitrate-N +, Nitrate-N dissolved (00631)</li> <li>Ammonia-N dissolved (00608)</li> <li>Total Kjeldahl-N ()</li> <li>Other:</li> </ul>	mg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/l	4/5 4/5 4/22 5/10
NALYTICAL RESULTS fro NF, NA Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Other: Other: Other: Other: Other: Nitrate-N +, Nitrate-N total (00630) Ammonia-N total (00610) Total Kjeldahi-N () Chemical oxygen demand (00340) Total organic carbon () Other: Other: Dother: Dother: Nitrate-N + Nitrate-N total (00630) Nitrate-N + Nitrate-N total (00630) Nitrate-N + Nitrate-N total (00630) Chemical oxygen demand (00340) Total organic carbon () Other: Dother: Dother: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note: Note:	Other-specify:           m SAMPLES           Units Date an           //mho	alyzed       F, NA <ul> <li>Calcium (00915)</li> <li>Magnesium (00925)</li> <li>Sodium (00930)</li> <li>Potassium (00935)</li> <li>Bicarbonate (00440)</li> <li>Chloride (00940)</li> <li>Sulfate (00945)</li> <li>Total filterable residue (dissolved) (70300)</li> <li>Other:</li> <li>F, A-H₂ SO₄</li> <li>Mitrate-N +, Nitrate-N dissolved (00631)</li> <li>Ammonia-N dissolved: (00608)</li> <li>Total Kjeldahl-N</li></ul>	mg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/l	4/5 4/5 4/22 5/10

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ND NM OIL CO AL State Lar	ENTAL BUREAU INSERVATION DIV nd Office BTdg, NM 87501 Boyer	ISION PO Box 208	38:			
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Bailed X Pump Dipped: D Tap:	Water level	<u></u>	Discharge Tatum 2.2	37 sec	Sample type	b
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eld comments Sam	ple from	1 Gad	Primp out	af 0,	pen we	K _
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	I Other- <i>specify:</i>	Inits Date analyz			Units	Date analyzed
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ALYTICAL RESULTS fm NF: NA Conductivity (Corrected) 25°C (00095)	I Other- <i>specify:</i> om SAMPLES	inits Date analyz	embrane filter ed F, NA Calcium (00915) Magnesium (00925) Sodium (00930)		Units: mg/l mg/l mg/l	
ALYTICAL RESULTS from NF, NA-	I Other- <i>specify:</i> om SAMPLES	Inits Date analyz	embrane filter ed F, NA Calcium (00915) Magnesium (00925)		Units: mg/l mg/l	
ALYTICAL RESULTS from NF, NA Conductivity (Corrected) 25 °C (00095) Total non-filterable	I Other- <i>specify:</i>	Inits Date analyz mho ng/l	Embrane filter           ed         F; NA            Calcium (00915)            Magnesium (00925)            Sodium (00930)            Potassium (00935)            Bicarbonate (00440)            Chloride (00940)		Units mg/l mg/l mg/l mg/l mg/l	
A Submitted:  NA: No acid added:  IALYTICAL RESULTS fm NF; NA Conductivity (Corrected): 25°C (00095) Total non-filterable residue (suspended): (00530) Other: TCAPSCAN Other-AC	Dother-specify:	Inits Date analyz	embrane filter ed F, NA Calcium (00915) Calcium (00935) Sodium (00930) Potassium (00935) Bicarbonate (00440) Chloride (00946) Sulfate (00945) Total filterable residue		Units: mg/l mg/l mg/l mg/l	
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ALYTICAL RESULTS from NA: No acid added: ALYTICAL RESULTS from NF, NA- Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Other: Other: Other: AH ₂ SO ₄	Dother-specify:	Inits Date analyz mho ng/l	embrane filter ed F; NA Calcium (00915) Magnesium (00925) Sodium (00930) Potassium (00935) Bicarbonate (00440) Chloride (00940) Sulfate (00945) Total filterable residue (dissolved) (70300) Other:		Units mg/l mg/l mg/l mg/l mg/l	
ALYTICAL RESULTS from NA: No acid added: ALYTICAL RESULTS from NF, NA- Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Other: Other: Other: AH ₂ SO ₄	1 Other- <i>specify:</i>	Inits Date analyz mho ng/l	embrane filter         ed       F; NA: <ul> <li>Calcium (00915)</li> <li>Magnesium (00925)</li> <li>Sodium (00930)</li> <li>Potassium (00935)</li> <li>Bicarbonate (00440)</li> <li>Chloride (00940)</li> <li>Sulfate (00945)</li> <li>Total filterable residue (dissolved) (70300)</li> <li>Other:</li> <li>F; A-H₂ SO₄.</li> </ul>		Units mg/l mg/l mg/l mg/l mg/l	
Image: Second state       Image: Second state         Image: NA: No acid added:       Image: Second state         Image: No a	1 Other- <i>specify:</i> om SAMPLES	Inits Date analyz	Embrane filter           Ed         F; NA: <ul> <li>Calcium (00915)</li> <li>Magnesium (00925)</li> <li>Sodium (00930)</li> <li>Potassium (00935)</li> <li>Bicarbonate (00440)</li> <li>Chloride (00940)</li> <li>Sulfate (00945)</li> <li>Total filterable residue (dissolved) (70300)</li> <li>Other:</li> <li>F; A-H₂ SO₄:</li> <li>Nitrate-N +, Nitrate-N dissolved (00631)</li> </ul>		Units mg/l mg/l mg/l mg/l mg/l	
Alternative and a second added: Alternative added:	I Other- <i>specify:</i> om SAMPLES U µµ ∠.005 √ ∠.005	Inits Date analyz mho ng/l g/l ng/l	Embrane filter           ed         F; NA: <ul> <li>Calcium (00915)</li> <li>Magnesium (00925)</li> <li>Sodium (00930)</li> <li>Potassium (00935)</li> <li>Bicarbonate (00440)</li> <li>Chloride (00940)</li> <li>Sulfate (00945)</li> <li>Total filterable residue (dissolved) (70300)</li> <li>Other:</li> <li>F; A-H₂ SO₄:</li> <li>Nitrate-N +, Nitrate-N dissolved (00631)</li> <li>Ammonia-N dissolve</li> </ul>		Units mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l	
J. NA: No acid added:         IALYTICAL RESULTS fm         NF: NA-         Conductivity (Corrected)         25°C (00095)         Total non-filterable         residue (suspended)         (00530)         Other:         Other:         A.H.SO.         Nitrate-N +, Nitrate-N         total (00630)         Ammonia-N total (00610)         Total Kjekdahl-N         (         )         Chemical oxygen	I Other- <i>specify:</i> om SAMPLES U µµ ∠.005 ↓.005	Inits Date analyz mho ng/l g/l ng/l ng/l	Embrane filter           Ed         F; NA: <ul> <li>Calcium (00915)</li> <li>Magnesium (00925)</li> <li>Sodium (00930)</li> <li>Potassium (00935)</li> <li>Bicarbonate (00440)</li> <li>Chloride (00940)</li> <li>Sulfate (00945)</li> <li>Total filterable residue (dissolved) (70300)</li> <li>Other:</li> <li>F; A-H₂ SO₄:</li> <li>Nitrate-N +, Nitrate-N dissolved (00631)</li> </ul>		Units mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l	
Armonia-N total (00610)  Armonia-N total (00630)  Armonia-N total (00610)  Chemical oxygen demand (00340)	I Other- <i>specify:</i> om SAMPLES U µµ ∠.005 √ ∠.005 √ 	Inits Date analyz mho ng/l g/l ng/l ng/l ng/l	Embrane filter           Ed         F; NA: <ul> <li>Calcium (00915)</li> <li>Magnesium (00925)</li> <li>Sodium (00930)</li> <li>Potassium (00935)</li> <li>Bicarbonate (00440)</li> <li>Chloride (00940)</li> <li>Sulfate (00945)</li> <li>Total filterable residue (dissolved) (70300)</li> <li>Other:</li> <li>F; A-H₂ SO₄:</li> <li>Nitrate-N +, Nitrate-N dissolved (00631)</li> <li>Ammonia-N dissolve (00608)</li> <li>Ammonia-N dissolve</li> <li>(00608)</li> <li>Ammonia-N dissolve</li> <li>(00608)</li> </ul>		Units mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l	
Armonia-N total (00610)  F. A-H ₂ SO ₄ Nitrate-N ⁺ , Nitrate-N. total (00630)  Cher: J Constant (00610)  Cherical oxygen demand (00340)  Total organic carbon  ()	I Other- <i>specify:</i> om SAMPLES U µµ ∠.005 √ ∠.005 √ 	Inits Date analyz	embrane filter         ed       F; NA: <ul> <li>Calcium (00915)</li> <li>Magnesium (00925)</li> <li>Sodium (00930)</li> <li>Potassium (00930)</li> <li>Potassium (00930)</li> <li>Bicarbonate (00440)</li> <li>Chloride (00940)</li> <li>Sulfate (00945)</li> <li>Total filterable residue (dissolved) (70300)</li> <li>Other:</li> <li>F; A-H₂ SO₄.</li> </ul> <li>Fi Artha SO₄.</li> <li>Initrate-N +, Nitrate-I dissolved (00631)</li> <li>Ammonia-N dissolve (00608)</li> <li>Total Kjeldahl-N</li>	d	Units           mg/l	Date analyzed
Alternative and a second and a second	I Other- <i>specify:</i> om SAMPLES U µµ ∠.005 √ ∠.005 √ 	Inits Date analyz mho ng/l g/l ng/l ng/l ng/l	embrane filter         ed       F; NA: <ul> <li>Calcium (00915)</li> <li>Magnesium (00925)</li> <li>Sodium (00930)</li> <li>Potassium (00930)</li> <li>Potassium (00930)</li> <li>Bicarbonate (00440)</li> <li>Chloride (00940)</li> <li>Sulfate (00945)</li> <li>Total filterable residue (dissolved) (70300)</li> <li>Other:</li> <li>F; A-H₂ SO₄.</li> <li>Nitrate-N +, Nitrate-f dissolved (00631)</li> <li>Ammonia-N dissolve (00608)</li> <li>Total Kjeldahl-N ()</li> <li>N ()</li> </ul>	d Date Re	Units           mg/l           mg/l	Date analyzed
Submitted:         NA: No acid added:         IALYTICAL RESULTS fm         NF: NA-         Conductivity (Corrected)         25°C (00095)         25°C (00095)         Total non-filterable         residue (suspended)         (00530)         Other:         Other:         A: Other:         Cother:         Cher:         Cher: <tr< td=""><td>I Other-<i>specify:</i> om SAMPLES U µµ ∠.005 √ ∠.005 √ </td><td>Inits Date analyz mho ng/l g/l ng/l ng/l ng/l</td><td>embrane filter         ed       F; NA:            <ul> <li>Calcium (00915)</li> <li>Magnesium (00925)</li> <li>Sodium (00930)</li> <li>Potassium (00930)</li> <li>Potassium (00930)</li> <li>Bicarbonate (00440)</li> <li>Chloride (00940)</li> <li>Sulfate (00945)</li> <li>Total filterable residue (dissolved) (70300)</li> <li>Other:</li> <li>F; A-H₂ SO₄.</li> </ul> <li>Fi Artha SO₄.</li> <li>Initrate-N +, Nitrate-I dissolved (00631)</li> <li>Ammonia-N dissolve (00608)</li> <li>Total Kjeldahl-N</li></td><td>d Date Re</td><td>Units           mg/l           mg/l</td><td>Date analyzed</td></tr<>	I Other- <i>specify:</i> om SAMPLES U µµ ∠.005 √ ∠.005 √ 	Inits Date analyz mho ng/l g/l ng/l ng/l ng/l	embrane filter         ed       F; NA: <ul> <li>Calcium (00915)</li> <li>Magnesium (00925)</li> <li>Sodium (00930)</li> <li>Potassium (00930)</li> <li>Potassium (00930)</li> <li>Bicarbonate (00440)</li> <li>Chloride (00940)</li> <li>Sulfate (00945)</li> <li>Total filterable residue (dissolved) (70300)</li> <li>Other:</li> <li>F; A-H₂ SO₄.</li> </ul> <li>Fi Artha SO₄.</li> <li>Initrate-N +, Nitrate-I dissolved (00631)</li> <li>Ammonia-N dissolve (00608)</li> <li>Total Kjeldahl-N</li>	d Date Re	Units           mg/l           mg/l	Date analyzed
Submitted:         In NA: No acid added:         NA: No acid added:         NF: NA         Conductivity (Corrected):         25 °C (00095)         Total non-filterable:         residue (suspended):         (00530)         Other:         Other:         Cother:	I Other- <i>specify:</i> om SAMPLES U µµ ∠.005 √ ∠.005 √ 	Inits Date analyz mho ng/l g/l ng/l ng/l ng/l	embrane filter         ed       F; NA: <ul> <li>Calcium (00915)</li> <li>Magnesium (00925)</li> <li>Sodium (00930)</li> <li>Potassium (00930)</li> <li>Potassium (00930)</li> <li>Bicarbonate (00440)</li> <li>Chloride (00940)</li> <li>Sulfate (00945)</li> <li>Total filterable residue (dissolved) (70300)</li> <li>Other:</li> <li>F; A-H₂ SO₄.</li> </ul> <li>Fi Artha SO₄.</li> <li>Initrate-N +, Nitrate-I dissolved (00631)</li> <li>Ammonia-N dissolve (00608)</li> <li>Total Kjeldahl-N</li>	d Date Re	Units           mg/l           mg/l	Date analyzed

ICAP ·SCREEN

Lab Number: HM 486Date Submitted: 3/22/84By: Boyer

Sample Code: <u>Florn</u>	Vista Water Supply
Date Reported:	5/31/35
By: J. Bakky	

Determination

Concentration (µg/m1)

· · ·			( )0
Aluminum		· · · · · · · · · · · · · · · · · · ·	2.10
Barium		<del></del>	2.10
Beryllium		+	4.10
Boron			4.10
Cadmium			2,10
Calcium		110.	•
Chromium		·	4.10
Cobalt	· .		4.10
Copper	÷.,		<.10
Iron			2.10
Lead	1. 	- <u></u>	L.10
Magnesium		9.1	
Manganese		.47	·····
Molybdenum			4.10
Nickel		· .	2.10
Silicon	*	5,3	
Silver	۲ <u>ـ</u>		<.10
Strontium		2.7	
Tin			<,10
Vanadium			4.10
Yttrium	· · · · ·		4.10
Zinc			4.10

SCIENTIFI 700 Camin	co Health and Environment Departm C LABORATORY DIVISION o de Salud NE ue, NM 87106 — (505)	lent	GENERAL WATER an TROGEN	CHEMISTRY ANALYSIS
DATE RECEIVED 3 23 8	5 NO. W - 1211 USER	59300 59600	^{ЖХ} отнев: 82235	
Collection DATE	SITE INFORM-► ATION	and the second	ofite Well	+ #2 \$ 3
allected by - Person/Agency		site description FLARA	isto Wate	Werd Assoc
END NM OIL NAL State L	MENTAL BUREAU CONSERVATION DIVISION and Office Bldg, PO E e, NM 87501 Boyer		Station/	pigot mkille volue with velle Pfo
AMPLING CONDITION	S E	AUULUIS	Well code	VUA
Bailed Pump Dipped Tap	Water level	Discharge	1009pm Sample	type grala
oH (00400)	Conductivity (Uncorrected)	Water Temp. (0001 µmho	10) Conduc	tivity at 25 °C (00094) μmho
	posite of we	lf #2 \$3 n	ear and to	, 5W of
mimp	house			
AMPLE FIELD TREAT	AENT — Check proper boxes	<b>}</b> `	·····	
A NA: No acid added	from SAMPLES	ite analyzed <b>NE</b> NA NE		Units Date analyzed
Conductivity (Corrected) 25°C (00095)		ったの X Calcium (009	/ / / -	mg/I/15
<ul> <li>☐ Total non-filterable residue (suspended) (00530)</li> <li>▲ Other:</li> <li>P H</li> <li>☐ Other:</li> </ul>	mg/lmg/l	Magnesium (           Magnesium (           Sodium (0093           Potassium (00           Bicarbonate (           Solitate (0094           Total filterable (dissolved) (7	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$
17, All260 NA	· · · · · · · · · · · · · · · · · · ·		urice 0,49	<u> </u>
<ul> <li>Nitrate-N +; Nitrate-N total (00630)</li> <li>Ammonia-N total (00610)</li> <li>Total Kjeldahl-N</li> </ul>	mg/l mg/l	F, A-H₂ SO₄       □ Nitrate-N+, N       dissolved (00       □ Ammonia-N c	631)	mg/l
( ) Chemical oxygen demand (00340)	mg/l mg/l	(00608)	-N	mg/l
Total organic carbon ( ) ] Other:	mg/l	Cther:		
☐ Other: ☐ Other: -		Analyst	Date Reported	5 Reviewed Dy
aboratory remarks		<u>_</u> <u>_</u>		

AMPLING CONDITIONS Bailed Pump Wa Dipped Tap	AVATION DIVISION fice Bidg, PO Box 87501 Ater level ater level	Example       Wellet       #2 \$ \$         rescription       FLDAR Visto       Water for Under MSSO         2088       Station/ well code       Station/ well code         Discharge       IDDEPARM       Sample type         Water Temp. (00010)       11 °C       Conductivity at 25°C (00094)         Thomas And to SWO       Mater Temp. (00010)       11 °C         F #2 \$ 3 NEar RnD to SWO       SWO         ered in field with       A: 2 mi H2SO4/L added         Si µmembrane filter       A: 2 mi H2SO4/L added
Dilection DATE Dilection DATE Dilection TIME ENVIRONMENTAL ENVIRONMENTAL NM OIL CONSER NAL State Land Of Santa Fe, NM Attn:David_Boyen AMPLING CONDITIONS Bailed Pump Wa Dipped □ Tap H (00400) Co ield comments Comments Comments Submitted NA: No acid added □ Othe NALYTICAL RESULTS from SA NF, NA Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Cother: Other:	SITE INFORM- ATION BUREAU VATION DIVISION Frice Bldg, PO Box 87501 Atter level Atter level Collection site de Domune State State Collection site de Collection site de Collection site de Domune State State Collection site de State State Collection site de State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State State St	Perception FLAR Visto Wellet #2 43 Perception FLAR Visto Water Wellet #2 43 Provide Wellet Wellet #2 43 Provide Wellet Wellet Wellet #2 43 Provide Wellet Wellet Wellet Wellet Wellet Wellet Provide Wellet Wellet Wellet Wellet Wellet Provide Wellet Wellet Wellet Wellet Wellet Wellet Provide Wellet Wellet Wellet Wellet Wellet Provide Wellet Wellet Wellet Wellet Provide Wellet Wellet Wellet Wellet Provide Wellet Wellet Provide Wellet Wellet Provide Wellet Wellet Provide Wellet Wellet Wellet Provide Wellet W
Illected by - Person/Agency         ENVIRONMENTAL         ND       NM OIL CONSER         MAL       State Land Of         PORT       Santa Fe, NM         Attn:       David Boyer         MPLING CONDITIONS       Image: Condition of the second	ater level	FLOW Notice       From Spreot introl         2088       From Spreot introl         Station/       From Spreot introl         Velicode       Owner         Owner       FVWMA         Discharge       IODAPAM         Sample type       Sample type         mho       11 °C         Conductivity at 25°C (00094)       µmh         J # J J A Mean Read to SW of       µmh         F # J J A Mean Read to SW of       A: 2 mi H ₂ SO ₄ /L added         analyzed       F. NA       Units
ENVIRONMENTAL ND NM OIL CONSER AL State Land Of Santa Fe, NM Attn:David_Boyer MPLING CONDITIONS BailedPump Wa Dipped Tap H (00400) Co eld comments Company MMPLE FIELD TREATMENT No. of samples NF: UMPLE FIELD TREATMENT No. of samples NF: Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Other:	AVATION DIVISION fice Bidg, PO Box 87501 Ater level ater level	2088 2088 Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge Discharge
ND       NM OIL CONSER         IAL       State Land Of         PORT       Santa Fe, NM         Attn:       David Boyer         MPLING CONDITIONS       Value         Bailed       Pump       Wa         Dipped       Tap       Wa         H (00400)       Co       Co         eld comments       Company       Max         MPLE FIELD TREATMENT       NF:       O         No. of samples       NF:       O         submitted       NF:       O         Conductivity (Corrected)       25°C (00095)       O         Total non-filterable       residue (suspended)       O         Other:       O       Conductivity (Corrected)       Conductivity (Corrected)         Other:       O       O       O	AVATION DIVISION fice Bidg, PO Box 87501 Ater level ater level	Station/         Station/         well code         Owner         Discharge         N 100000         Sample type         Synch         Mater Temp. (00010)         11         °C         Conductivity at 25°C (00094)         µmh         %         ####################################
AMPLING CONDITIONS         Bailed       Pump         Dipped       Tap         H (00400)       Co         ield comments       Commonwealth         MMPLE FIELD TREATMENT       NF:         No. of samples       NF:         submitted       NF:         NA: No acid added       Other         Social conductivity (Corrected)       25°C (00095)         Total non-filterable       residue (suspended)         (00530)       Other:         Other:       Other:	ater level	well code         Owner       FUWUA         Discharge       I DOPAM         Sample type       Stable         mho       II       °C         Conductivity at 25°C (00094)       µmh         Image: Superior Conductivity at 25°C (00094)       µmh         Superior Conductivity at 25°C (00094)       µmh         Image: Superior Conductivity at 25°C (00094)       µmh         Image: Superior Conductivity at 25°C (00094)<
Bailed Pump Wa Dipped Tap Wa No of samples Submitted NR: No acid added Othe NALYTICAL RESULTS from SA NF, NA Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Other: Other:	inductivity (Uncorrected)	well code         Owner       FUWUA         Discharge       I DOPAM         Sample type       Stable         mho       II       °C         Conductivity at 25°C (00094)       µmh         Image: Superior Conductivity at 25°C (00094)       µmh         Superior Conductivity at 25°C (00094)       µmh         Image: Superior Conductivity at 25°C (00094)       µmh         Image: Superior Conductivity at 25°C (00094)<
Bailed       Pump       Wa         Dipped       Tap       Co         H (00400)       Co         ield comments       Company         MPLE FIELD TREATMENT       NF:         No. of samples       NF:         submitted       NF:         NA: No acid added       Other         Store (00095)       Conductivity (Corrected)         Total non-filterable       residue (suspended)         (00530)       Cother:         Other:       Conduct:	inductivity (Uncorrected)	Discharge       I DDPPM       Sample type         Water Temp. (00010)       11       °C       Conductivity at 25°C (00094)         μmh       11       °C       Conductivity at 25°C (00094)         μmh       11       °C       Δ         μmh       μmh       μmh         μ       43       μmh         μ       43       μmh         μ       μ       μmh         μ       μ       μmh         μ       μ       μ         μ       μ       μ         μ       μ       μ         μ       μ       μ         μ       μ       μ         μ       μ       μ         μ       μ       μ         μ       μ       μ         μ       μ       μ         μ       μ       μ         μ       μ       μ         μ       μ       μ         μ       μ       μ         μ       μ       μ         μ       μ       μ         μ       μ       μ         μ       μ       μ         μ       μ
Dipped Tap H (00400) Co ield comments Comments Commandue MMPLE FIELD TREATMENT AMPLE FIELD TREATMENT Mo. of samples submitted NA: No acid added Othe NALYTICAL RESULTS from SA NF. NA Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Cother: Other:	inductivity (Uncorrected)	N IDDPPM     Spale       mho     Vater Temp. (00010)     Conductivity at 25°C (00094)       μmh     11 °C     μmh       μ     43 nCar and to 500 of     μmh       μ     43 nCar and to 500 of     μmh       ered in field with     A: 2 ml H ₂ SO ₄ /L added       s μmembrane filter     A: 2 ml H ₂ SO ₄ /L added
ield comments Company AMPLE FIELD TREATMENT No. of samples submitted NA: No acid added  Othe NALYTICAL RESULTS from SA NF. NA Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Gother: Other:	La FL Wella La FL Wella Check proper boxes Whole sample- (Non-filtered) F: Filte 0.45 er-specify: AMPLES	mho     1/     °C     μmh <i>μ</i> = <i>μ</i> = <i>μ μ</i> = <i>μ μ μ</i> = <i>μ μ μ</i> ered in field with S μmembrane filter <i>μ</i> = <i>μ μ</i> Mathematical States and the states
AMPLE FIELD TREATMENT	Check proper boxes - Check proper boxes Whole sample- (Non-filtered) F: Filte 0.45 er-specify: AMPLES	ered in field with 5 μmembrane filter A: 2 ml H ₂ SO ₄ /L added nalyzed F. NA Units Date analyzed
No. of samples submitted       Image: NF: ()         Submitted       Image: NF: ()         Image: NA: No acid added       Image: Other         Image: Other:       Image: Other         Image: Other:       Image: Other	Whole sample (Non-filtered) F: Filte 0.45 er-specify:	nalyzed F, NA Units Date analyzed
No. of samples submitted INF: ( NA: No acid added I Other NALYTICAL RESULTS from SA NF, NA Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Other: Other:	Whole sample (Non-filtered) F: Filte 0.45 er-specify:	nalyzed F, NA Units Date analyzed
25°C (00095)		
residue (suspended) (00530) Other: Other:	µmho	□ Calcium (00915) mg/l □ Magnesium (00925) mg/l
Other:	mg/l	Sodium (00930)         mg/l           Potassium (00935)         mg/l           Bicarbonate (00440)         mg/l           Chloride (00240)         mg/l
	······································	□ Chloride (00940) mg/l
		(dissolved) (70300) mg/l
F, A-H2SO4		
Nitrate-N+, Nitrate-N		F, A-H ₂ SO ₄
total (00630)	mg/l	Nitrate-N +, Nitrate-N
Ammonia-N total (00610)	mg/l	dissolved (00631)0.2.8 mg/I4/5
	mg/l	(00608) mg/l/22
Chemical oxygen	//	X Total Kieldahl-N
demand (00340)	mg/l	() 0.77 mg/l 5/10
( )	mg/l	Other:
Other:		Analyst Date Reported Reviewed by
LD-oratory - Emarks		5 10 85 (Dem
	i	
LD 726 (12/84) DISTRIBUTION:		

Albuquerqu	5 NO. HM-48		o 🗆 59600 🖄 o	THER: 82	235	
lection DATE	SITE		and Compati	Te U	ellet #2 9	\$3
lection TIME	INFORM- ATION					
lected by - Person/Agency	- 1	Collection site descriptio	FLOR Viet	a W	stor Uper	<u>(1350)</u>
Bay	<u>~</u>	] [		1		· Ale il
ND NM OIL AL State L	e, NM 87501	DIVISION dg, PO Box 208	8	  	no notife	weth BO
				Station/		
				well code		
MPLING CONDITION	IS:	•		Owner F	VWUN	
Bailed Pump Dipped Dap	Water level		Discharge ~ 10.	DAber	Sample type	ab
H (00400)	Conductivity (U	Incorrected)	Water Temp. (00010)	/ // °C	Conductivity at 25°C (	(00094) µmh
eld comments			4742 0			1
Com	pople If	- wellt	#2 \$ 3 ner	<u>7 RNE</u>	B SUD	5
pinto	house				······	
MPLE FIELD TREAT!	AENT Check on	oper boyes				
No. of complete						
	wnole sampl	le 🖌 Eitered in	i field with 🐂 🔨 🔥 👝			
submitted	NF: Whole sampl (Non-filtered)		i field with: A: 2		tabled 5m) t	10.3
· · · · · · · · · · · · · · · · · · ·	(Non-filtered)	) <b>7 Γ΄ 0.45 μ</b> me		-ml H2302/	tabled 5m) t	103
submitted [.	Other-specify:	) <b>7 Γ΄ 0.45 μ</b> me		-mi 112304/	tabled 5n) f	110.3
submitted [.	Other-specify:	ο 7 0.45 μme	embrane filter: 🏠 🏝 📼	-mi H ₂ SO ₄ /		ANO 3
SUDMITTED L	Other-specify:	) <b>7²Γ΄ 0.45 μme</b>	embrane filter: MA: -			
Submitted [] NA: No acid added: NALYTICAL RESULTS NF, NA	Other-specify:	) <b>7²Γ΄ 0.45 μme</b>	d F, NA Calcium (00915)		Units [ mg/l mg/l	
Submitted	Other-specify:	Units Date analyze	Importance filter         Importance filter           Ind         F, NA           Ind         Calcium (00915)           Ind         Magnesium (00925)           Ind         Sodium (00930)		Units [ mg/l mg/l mg/l	Date analyzed
Submitted	Other-specify:	) Υ Γ 0.45 μme Units Date analyze μmho	Importance filter         Importance filter           Importance filter         Importance filter <td>· · · · · · · · · · · · · · · · · · ·</td> <td>Units [  mg/l  mg/l  mg/l</td> <td>Date analyzed</td>	· · · · · · · · · · · · · · · · · · ·	Units [ mg/l mg/l mg/l	Date analyzed
Submitted	Other-specify:	Units Date analyze	Imbrane filter         Image: Calcium (00915)           Image: Calcium (00915)         Image: Calcium (00925)           Image: Calcium (00930)         Image: Calcium (00930)           Image: Calcium (00930)         Image: Calcium (00935)           Image: Calcium (00935)         Image: Calcium (00935)           Image: Calcium (00940)         Image: Calcium (00940)	· · · · · · · · · · · · · · · · · · ·	Units [ mg/l mg/l mg/l mg/l	Date analyzed
Submitted	INF: (Non-filtered) Other-specify: from SAMPLES 4.005	) Υ Γ 0.45 μme Units Date analyze μmho	Imbrane filter         Image: Calcium (00915)           Image: Calcium (00915)         Image: Calcium (00925)           Image: Calcium (00930)         Image: Calcium (00930)           Image: Calcium (00930)         Image: Calc	· · · · · · · · · · · · · · · · · · ·	Units [ mg/l mg/l mg/l mg/l	Date analyzed
ALYTICAL RESULTS NF, NA Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Other TCA P SCA N Other TCA P SCA N	INF: (Non-filtered) Other-specify: from SAMPLES	) Υ Γ 0.45 μme Units Date analyze μmho	mbrane filter       A: -         Id       F, NA         Image: Calcium (00915)         Image: Magnesium (00925)         Image: Sodium (00930)         Image: Potassium (00940)         Image: Potassium (00945)         Image: Potassium (00945)         Image: Potassium (00945)         Image: Potassium (00945)         Image: Potassium (000940)         Image	· · · · · · · · · · · · · · · · · · ·	Units [ mg/l mg/l mg/l mg/l mg/l	Date analyzed
ALYTICAL RESULTS ALYTICAL RESULTS NF, NA Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Other LCAPSCAN Other: AS Other: SC	INF: (Non-filtered) Other-specify: from SAMPLES 4.005	) Υ Γ 0.45 μme Units Date analyze μmho	Imbrane filter         Image: Calcium (00915)           Image: Calcium (00915)         Image: Calcium (00925)           Image: Calcium (00930)         Image: Calcium (00930)           Image: Calcium (00930)         Image: Calc	· · · · · · · · · · · · · · · · · · ·	Units [ mg/l mg/l mg/l mg/l mg/l	Date analyzed
Submitted NA: No acid added ALYTICAL RESULTS NF, NA Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Other TCAPSCAN Other: SE F. A-H₂SO.	INF: (Non-filtered) Other-specify: from SAMPLES 4.005	) Υ Γ 0.45 μme Units Date analyze μmho	mbrane filter       A: -         Image: A: -       A: - <tr td="">       A: -</tr>	· · · · · · · · · · · · · · · · · · ·	Units [ mg/l mg/l mg/l mg/l mg/l	Date analyzed
Submitted NA: No acid added ALYTICAL RESULTS NF, NA Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Other TCAPSCAN Other: SE F. A-H₂SO.	INF: (Non-filtered) Other-specify: from SAMPLES 4.005	) Υ Γ 0.45 μme Units Date analyze μmho	mbrane filter       A: -         id       F, NA         id       Calcium (00915)         id       Magnesium (00925)         id       Sodium (00930)         id       Potassium (00935)         id       Bicarbonate (00440)         id       Chloride (00945)         id       Total filterable residue (dissolved) (70300)         id       Other:         id       F, A-H2 SO4		Units [ mg/l mg/l mg/l mg/l mg/l	Date analyzed
Submitted	<pre> NF: (Non-filtered) Other-specify: from SAMPLES</pre>	Units Date analyze	mbrane filter       A: -         Image: A: -       A: - <tr td="">       A: -</tr>		Units [ mg/l mg/l mg/l mg/l mg/l	Date analyzed
submitted NA: No acid added NA: No acid added NE: No NF, NA Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Other Ica PSCAN Other: SC F. A-H ₂ SO. Nitrate-N + Nitrate-N total (00630) Ammonia-N total (00610)	NF: (Non-filtered) ○ Other-specify: from SAMPLES 4.005 4.005	0.45 μme	mbrane filter       A: -         d       F, NA	e	Units         I           mg/l	Date analyzed
submitted NA: No acid added NA: No acid added NALYTICAL RESULTS NF, NA Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Other TGAPS(AN Other: AS Other: SC F, A-H ₂ SO ₄ Nitrate-N + , Nitrate-N total (00630) Ammonia-N total (00610) Total Kjeldahl-N ( ) Chemical oxygen	NF: (Non-filtered) ○ Other-specify: from SAMPLES 4.005 4.005	) <b>Δ.F</b> 0.45 μme	mbrane filter       A: -         ind       F, NA         ind       Calcium (00915)         ind       Magnesium (00925)         ind       Sodium (00930)         ind       Potassium (00935)         ind       Bicarbonate (00440)         ind       Chloride (00945)         ind       Total filterable residue (dissolved) (70300)         ind       Other:         F, A-H2 SO4         ind       Nitrate-N + , Nitrate- dissolved (00631)         ind       Ammonia-N dissolved (00608)	e	Units [ mg/l mg/l mg/l mg/l mg/l mg/l	Date analyzed
submitted NA: No acid added NA: No acid added NA: No acid added NA: No acid added NE, NA Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Other ICAPSCAN Other: SC F. A-H ₂ SO. Nitrate-N + Nitrate-N total (00630) Ammonia-N total (00610) Total Kjeldahl-N ( ) Chemical oxygen demand (00340)	NF: (Non-filtered) ○ Other-specify: from SAMPLES 4.005 4.005	0.45 μme	mbrane filter       A: -         d       F, NA	e	Units         I           mg/l	
Submitted	NF: (Non-filtered) □ Other-specify: from SAMPLES 4.005 4.005	0.45 μme	mbrane filter       A: -         ind       F, NA         ind       Calcium (00915)         ind       Magnesium (00925)         ind       Sodium (00930)         ind       Potassium (00935)         ind       Bicarbonate (00440)         ind       Chloride (00945)         ind       Total filterable residue (dissolved) (70300)         ind       Other:         F, A-H2 SO4         ind       Nitrate-N + , Nitrate- dissolved (00631)         ind       Ammonia-N dissolved (00608)	e	Units         I           mg/l	Date analyzed
submitted NA: No acid added: NA: No acid added: NA: No acid added: NF, NA Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Other CAPSCAN Other: SC F. A-H ₂ SO ₄ Nitrate-N + Nitrate-N total (00630) Ammonia-N total (00610) Total Kjeldahl-N ( ) Chemical oxygen demand (00340) Total organic carbon ( )	INF: (Non-filtered) Other-specify: from SAMPLES 4.005 4.005	0.45 μme	mbrane filter       A: -         ind       F, NA         ind       Calcium (00915)         ind       Magnesium (00925)         ind       Sodium (00930)         ind       Potassium (00935)         ind       Bicarbonate (00440)         ind       Chloride (00945)         ind       Total filterable residue (dissolved) (70300)         ind       Other:         ind       F, A-H2 SO4         ind       Nitrate-N +, Nitrate- dissolved (00631)         ind       Ammonia-N dissolved (00608)         ind       Total Kjeldahl-N (         ind       Other:	ed	Units         I           mg/l	Date analyzed
submitted NA: No acid added NALYTICAL RESULTS NF, NA Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Other CAPSCAN Other CAPSCAN Other SC F.A-H ₂ SO ₄ Nitrate-N + Nitrate-N total (00630) Ammonia-N total (00610) Total Kjeldahl-N ( ) Chemical oxygen demand (00340) Total organic carbon ( )	NF: (Non-filtered) □ Other-specify: from SAMPLES 4.005 4.005	0.45 μme	mbrane filter       A: -         ind       F. NA         ind       Calcium (00915)         ind       Magnesium (00925)         ind       Sodium (00930)         ind       Potassium (00935)         ind       Bicarbonate (00440)         ind       Chloride (00945)         ind       Total filterable residue (dissolved) (70300)         ind       Other:         F. A-H2 SO4         ind       Nitrate-N + , Nitrate- dissolved (00631)         ind       Ammonia-N dissolve (00608)         ind       Total Kjeldahl-N (         ind       N	e	Units         I           mg/l	Date analyzed
submitted NA: No acid added: NA: No acid added: NALYTICAL RESULTS NF, NA Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Othen LCAP SCAN Other: AS Other: AS Other: AS Total (00630) Ammonia-N total (00610) Total Kjeldahl-N ( ) Chemical oxygen demand (00340) Total organic carbon ( ) Other: Other: Other:	NF: (Non-filtered) □ Other-specify: from SAMPLES 4.005 4.005	0.45 μme	mbrane filter       A: -         ind       F, NA         ind       Calcium (00915)         ind       Magnesium (00925)         ind       Sodium (00930)         ind       Potassium (00935)         ind       Bicarbonate (00440)         ind       Chloride (00945)         ind       Total filterable residue (dissolved) (70300)         ind       Other:         ind       F, A-H2 SO4         ind       Nitrate-N +, Nitrate- dissolved (00631)         ind       Ammonia-N dissolved (00608)         ind       Total Kjeldahl-N (         ind       Other:	e	Units         I           mg/l	Date analyzed
submitted NA: No acid added: NA: No acid added: NA: No acid added: NA: No acid added: NE, NA Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Othen IcaP Scan Other: AS Other: AS Other: AS Total (00630) Ammonia-N total (00610) Total Kjeldahl-N ( ) Chemical oxygen demand (00340) Total organic carbon ( ) Other:	NF: (Non-filtered) □ Other-specify: from SAMPLES 4.005 4.005	Units Date analyze     Units Date analyze     μmho     mg/l     mg/l	mbrane filter       A: -         ind       F, NA         ind       Calcium (00915)         ind       Magnesium (00925)         ind       Sodium (00930)         ind       Potassium (00935)         ind       Bicarbonate (00440)         ind       Chloride (00945)         ind       Total filterable residue (dissolved) (70300)         ind       Other:         ind       Nitrate-N +, Nitrate- dissolved (00631)         ind       Ammonia-N dissolved (00608)         ind       Total Kjeldahl-N ()         ind       Other:	e	Units         I           mg/l	Date analyzed
submitted NA: No acid added: NA: No acid added: NALYTICAL RESULTS NF, NA Conductivity (Corrected) 25°C (00095) Total non-filterable residue (suspended) (00530) Othen LCAP SCAN Other: AS Other: AS Other: AS Total (00630) Ammonia-N total (00610) Total Kjeldahl-N ( ) Chemical oxygen demand (00340) Total organic carbon ( ) Other: Other: Other:	□ NF: (Non-filtered)         □ Other-specify:         from SAMPLES	Units Date analyze     Units Date analyze     μmho     mg/l     mg/l	mbrane filter       A: -         Imbrane filter       Aignesium (00915)         Imbrane filter       Bicarbonate (00440)         Imbrane filter       Bicarbonate (00440)         Imbrane filter       Sulfate (00945)         Imbrane filter       Total filterable residue (dissolved) (70300)         Imbrane filter       Other:         Imbrane filter       F, A-H2 SO4         Imbrane filter       Armonia-N dissolve (00608)         Imbrane filter       Analyst	e	Units         I           mg/l	Date analyzed
Submitted	□ NF: (Non-filtered)         □ Other-specify:         from SAMPLES	Units Date analyze     Units Date analyze     μmho     mg/l     mg/l	mbrane filter       A: -         Imbrane filter       Aignesium (00915)         Imbrane filter       Bicarbonate (00440)         Imbrane filter       Bicarbonate (00440)         Imbrane filter       Sulfate (00945)         Imbrane filter       Total filterable residue (dissolved) (70300)         Imbrane filter       Other:         Imbrane filter       F, A-H2 SO4         Imbrane filter       Armonia-N dissolve (00608)         Imbrane filter       Analyst	e	Units         I           mg/l	Date analyzed

ICAP ·SCREEN

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HM 487 Lab Number:____ Sample Code: Flora Vista Was Composite Wells # Date Submitted: 3/22/85 Date Reported: 5/31/85 By: Doger By: Determination Concentration (µg/m1) 2.10 Aluminum <.10 Barium <.10 Beryllium <. 10 Boron 4.10 Cadmium 111. Calcium 4.10 Chromium <.10 Cobalt 2,10 Copper Ē <.10 Iron -... <.10 Lead 9.4 Magnesium 35 Manganese 4.10 Molybdenum 4.10 Nickel 5,4 Silicon 4.10Silver 2.5 Strontium <.10 Tin <.10 Vanadium 5 <.10 Yttrium <.10 Zinc

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Alected by - Person Agency ND NAL Sta: PORT San		AB <u>O: W(' 12 09</u> SITE INFORM- ATION OC &	Sample location	Lora Vista h	Vates	235 Supply pigot	Well#5
ENV ENV ND NM NAL Sta PORT San	DIL CONS	CE	Collection site descriptio	" Sample Fis	om S	pigor	in well
ND: NM ( NAL Sta: PORT San	DIL CONS	TAL BUREAU				Mar Mi	the other
		SERVATION D Office Bld NM 87501	IVISION g, PO Box 208		Station/		
	TIONS				owner F	VWU	Å
Bailed XF	ump	Water level		Discharge	2-22	Sample type	mal.
	ap	Conductivity (11a		Water Temp: (00010)	171	Conductivity et	<u>57400</u>
H (00400)	-	Conductivity (Und	S/D µmho		/ •C	Conductivity at	μmh
ield comments	This	W wes	00 betu	veen conter	mine	tel we	, Qtt I
a	ndp	umpho	use			-R-W-1	¢
MPLE FIELD TR	EATMEN	T Check prov	Der boyes			······	······································
No. of samples		Mhole comple	Filtered in	i field with A: 2 m	ni H₂SO₄/	Ladded	
ANA: No acid ac	Ided 🗆 C	Other-specify:	: : - 3: 				
NALYTICAL RESI	JLTS from	SAMPLES					
NF; NA			Units Date analyze	<b>K</b> Calcium (00915)		Unit	115
25°C (00095)		642	_µmho	Magnesium (00925)		<u>5</u> mg/	11 - 7/15 6.1
Total non-filterable				<ul> <li>Sodium (00930)</li> <li>Potassium (00935)</li> </ul>	34.		
residue (suspended (00530)	)		mg/l	🐉 Bicarbonate (00440)		<u>'54.1 mg/</u>	11576
Cother: PH		7,49		Chloride (00940)		<u> /8,7        mg/</u> 26 • /        mg/	
] Other:			<u> </u>	Total filterable residue		<u>30</u> mg/	-df
Other:				- (dissolved) (70300)	23 M	onema/	11 _ 5/6
FAH2501 11	4			M Other: Flourid E	·C	7,52	4/8
Nitrate-N+, Nitrate	N			F, A-H ₂ SO ₄			
total (00630) Ammonia-N total (0			mg/l	Nitrate-N + , Nitrate-N			
] Ammonia-N total (0) ] Total Kjeldahl-N			mg/l	dissolved (00631)		mg/	
( )			mg/l	Ammonia-N dissolvèd (00608)		mg/	//
Chemical oxygen demand (00340)			_ mg/l	Total Kjeldahl-N.			//
Total organic carbor	۱ 	3,2	6/13	_ Other:	· <u> </u>	mg/	·
()				Analyst			viewed by
(    ) ] Other: ] Other:				I			
Other:				<u> </u>	17	3 85	Sem
				<u> </u>	<u> </u>	3 82	<u> </u>
Other:					<u> </u>	3 82	- Clan
Other:						3 82	<u></u>

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ATE	19 85	LAB W(-12/7	USER CODE	o 🗌 59600 [	XX OTHER: 82	2235	
	:	SITE INFORM- ► ATION	Sample location	lora Vist	The Wates	Supply	Well#5
1555			Collection site descriptio	"Sample	FRAM	n'n ann	in uppo
	Supp 10	OCK		ange	- ha	ine us	the Other
					-11.4	alto Alus	T JL
		NTAL BUREAU	10101			-	00
ND N	M OIL CU	NSERVATION DIV	PO Box 208	8			
	anta Fe,		, 10 DOX 200	•		· · · · · · · · · · · · · · · · · · ·	
	David Bo						
Aut		<i>ريا چا</i>	*******************		Station/		
					weil code		
MPLING CON	DITIONS				Owner P	FULULY	4
	🛠 Pump 🗆 Tap	Water level		Discharge	o ppm	Sample type	rab
H (00400)	_	Conductivity (Unco	rrected) / Δ μmho	Water Temp. (00010)	11 °C	Conductivity at 25	^e C (00094) μmh
eld comments	Thi	1 Wwel	O betu	vern Con	Temina	tel wel	Q#J
	andf	ump hou	use				
MPLE FIELD	TREATMEN	T — Check prope	r boxes			<u></u>	
and the second	and the second s			and the second		· · · · · · · · · · · · · · · · · · ·	
No. of samples submitted		IF: Whole sample (Non-filtered)	F: Filtered in 0.45 µme	field with mbrane filter	: 2 ml H ₂ SO ₄	/Ladded	
					: 2 ml H ₂ SO ₄	/Ladded	<u></u>
submitted	d addect	(Non-filtered) Other- <i>specify:</i>	<b>Χ</b>		: 2 ml H ₂ SO ₄	/Ladded	······
NA: No acid	d addect	Cther- <i>specify:</i>	<b>Χ</b>	mbrane filter X A	: 2 ml H ₂ SO ₄ .	/L added Units	Date analyzed
Submitted NA: No acid NALYTICAL RI NF, NA Conductivity (Co	d added: ESULTS fro	Cther- <i>specify:</i>	X Γ: 0.45 μme	mbrane filter X A		······	Date analyzed
NA: No acio	d added: ESULTS fro	C (Non-filtered) Other- <i>specify:</i>	X Γ: 0.45 μme	d F, NA Calcium (00915 Magnesium (00	5)  0925)	Units mg/l mg/l	Date analyzed
NA: No acid NA: No acid NA: No acid NF, NA Conductivity (Cd 25°C (00095) Total non-filteral	d addect ESULTS fro orrected)	C (Non-filtered) Other- <i>specify:</i>	X Γ: 0.45 μme	d F, NA Calcium (00915 Magnesium (00 Sodium (00930	5) D925)	Units mg/l mg/l mg/l	Date analyzed
NA: No acid ALYTICAL R NF, NA Conductivity (Cd 25°C (00095) Total non-filteral residue (susper	d addect ESULTS fro orrected)	(Non-filtered) Other- <i>specify:</i> m SAMPLES	Units Date analyze	d F, NA Calcium (00915 Magnesium (00 Sodium (00930 Potassium (009 Bicarbonate (00	5) 0925) 1) 035) 0440)	Units mg/l mg/l	Date analyzed
IALYTICAL RI NF, NA Conductivity (Co 25°C (00095) Total non-filteral residue (susper (00530)	d addect ESULTS fro orrected)	(Non-filtered) Other- <i>specify:</i> m SAMPLES	X Γ: 0.45 μme	d F, NA Calcium (00915 Calcium (00915 Magnesium (00 Sodium (00930 Potassium (009 Bicarbonate (00 Chloride (00244	5) D925) D35) 0440) O)	Units mg/I mg/I mg/I mg/I mg/I	Date analyzed
A Submitted NA: No acid IALYTICAL RI NF, NA Conductivity (Cd 25°C (00095) Total non-filteral residue (susper (00530) Cilter:	d addect ESULTS fro orrected)	(Non-filtered) Other- <i>specify:</i> m SAMPLES	Units Date analyze	d F, NA Calcium (00915 Magnesium (00 Sodium (00930 Potassium (009 Bicarbonate (00 Chloride (00344) Sulfate (00944)	5) 0925) 1) 035) 0440) 0)	Units mg/i mg/i mg/i mg/i mg/i	Date analyzed
ALYTICAL RI NF, NA Conductivity (Cd 25°C (00095) Total non-filteral residue (susper (00530) Cilter: Other:	d addect ESULTS fro orrected)	(Non-filtered) Other- <i>specify:</i> m SAMPLES	Units Date analyze	d F, NA Calcium (00915 Calcium (00915 Magnesium (00 Sodium (00930 Potassium (009 Bicarbonate (00 Chloride (00244	5) 2925) 1) 235) 0440) 0) sidue	Units mg/I mg/I mg/I mg/I mg/I	Date analyzed
ubmitted <b>ALYTICAL Ri</b> <b>NF, NA</b> Conductivity (Cr 25°C (00095) Total non-filteral residue (susper (00530) Cther: Other: Other:	d addect ESULTS fro orrected)	(Non-filtered) Other- <i>specify:</i> m SAMPLES	Units Date analyze	d F, NA Calcium (00915 Magnesium (00 Sodium (00930 Potassium (009 Bicarbonate (00 Chloride (00344) Sulfate (00944) Total filterable ra	5) 2925) 1) 235) 0440) 0) sidue	Units mg/l mg/l mg/l mg/l mg/l mg/l	Date analyzed
A Submitted NA: No acid IALYTICAL RI NF, NA Conductivity (Cd 25°C (00095) Total non-filteral residue (susper (00530) Cilier: Other: Other: F, A-H ₂ SO ₄	d added: ESULTS fro orrected) ble nded)	(Non-filtered) Other- <i>specify:</i> m SAMPLES	Units Date analyze	d F, NA Calcium (00915 Magnesium (00 Sodium (00930 Potassium (009 Bicarbonate (00 Chloride (00344) Sulfate (00944) Total filterable ra (dissolved) (703	5) 2925) 1) 235) 0440) 0) sidue	Units mg/l mg/l mg/l mg/l mg/l mg/l	Date analyzed
A Submitted NA: No acid IALYTICAL RI NF, NA Conductivity (Cd 25°C (00095) Total non-filteral residue (susper (00530) Cilier: Other: Other: F, A-H ₂ SO ₄	d added: ESULTS fro orrected) ble nded)	(Non-filtered) Other- <i>specify</i> : m SAMPLES	Units Date analyze	d F, NA Calcium (00915 Magnesium (00 Sodium (00930 Potassium (009 Bicarbonate (00 Chloride (00344) Sulfate (00944) Chloride (00344) Sulfate (00944) Other:	5) 0925) 1) 135) 0440) 0) ssidue 3GO)	Units mg/l mg/l mg/l mg/l mg/l mg/l	Date analyzed
Armonia-N tota	d added: ESULTS fro orrected) ble nded) - - - - - - - - - - - - -	(Non-filtered) Other- <i>specify:</i> m SAMPLES	للسنة 0.45 μme <u>Units Date analyze</u> μmho mg/l	d F, NA Calcium (00915 Magnesium (00 Sodium (00930 Potassium (009 Bicarbonate (00 Chloride (00344) Sulfate (00944) Total filterable ra (dissolved) (703	5) D925) 035) 0440) 0, ssidue 3GO) trate-N	Units mg/l mg/l mg/l mg/l mg/l mg/l	Date analyzed
Submitted NA: No acid NA: No acid NF, NA Conductivity (Cd 25°C (00095) Total non-filteral residue (susper (00530) Cither: Other: Other: F, A-H ₂ SO. Nitrate-N + Nitt total (00630) Ammonia-N total	d added: ESULTS fro orrected) ble nded) - - - - - - - - - - - - -	(Non-filtered) Other- <i>specify:</i> m SAMPLES		d F, NA Calcium (00915 Magnesium (00 Sodium (00930 Potassium (009 Bicarbonate (00 Chloride (00945) Sulfate (00945) Total filterable ra (dissolved) (703 Other: Nitrate-N +, Nit dissolved (0063	5) 0925) 1) 135) 0440) 00 9 9 9 9 9 9 9 10 10 10 10 10 10 10 10 10 10	Units mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l	4/5
Submitted NA: No acid NA: No acid NF, NA Conductivity (C 25°C (00095) Total non-filteral residue (susper (00530) Total non-filteral residue (susper (00530) Collier: Other: Other: F, A-H ₂ SO ₄ Nitrate-N +, Nit total (00630) Ammonia-N tota Total Kjeldahl-N () Chemical oxyge	d added:	(Non-filtered) Other- <i>specify:</i> m SAMPLES		d F, NA Calcium (00915 Magnesium (00 Sodium (00930 Potassium (009 Bicarbonate (00 Chloride (00944) Sulfate (00944) Sulfate (00944) Total filterable ra (dissolved) (703 Other: Nitrate-N +, Nitt dissolved (0063 Ammonia-N dis (00608)	5) 0925) 035) 0440) 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001	Units mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l	4/-5 4/-22
Submitted NA: No acid NA: No acid NF, NA Conductivity (C 25°C (00095) Total non-filteral residue (susper (00530) Citter: Other: Other: Other: F, A-H ₂ SO ₄ Nitrate-N + Nit total (00630) Ammonia-N tota Total Kjeldahl-N () Chemical oxyge demand (00340)	d added: ESULTS fro orrected) ble nded) - - - - - - - - - - - - -	(Non-filtered) Other- <i>specify:</i> m SAMPLES		d F, NA Calcium (00915 Magnesium (00 Sodium (00930 Potassium (009 Bicarbonate (00 Chloride (00945) Sulfate (00945) Total filterable ra (dissolved) (703 Other: Nitrate-N +, Nit dissolved (0063	5) 0925) 035) 0440) 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001 001	Units mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l	4/5
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Submitted  NA: No acid NA: No acid NF, NA  Conductivity (Cd 25°C (00095)  Total non-filteral residue (susper (00530) Citier: Other: Other: Cother:  F, A-H ₂ SO ₄ Nitrate-N + , Nit total (00630) Ammonia-N tota Total Kjeldahl-N ( ) Chemical oxyge demand (00340 Total organic ca ( ) Other:	d added:	(Non-filtered) Other- <i>specify:</i> m SAMPLES		mbrane filter         d         F, NA         Calcium (00915         Magnesium (00         Sodium (00930         Potassium (009         Bicarbonate (00         Chloride (00340)         Total filterable radio (0063)         Intrate-N + , Nitt         dissolved (0063)         Total Kjeldahl-N (         Other:         Other:	5) 0925) 1) 1335) 0440) 0) 9 1035) 0440) 0) 9 10440) 10 10 10 10 10 10 10 10 10 10	Units mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l Mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l mg/l	4/-5 4/-22 5/10
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	and	)ump bourse	un on o	ha annonces e	
	CO IVA F	whip house	<u></u>		
No. of samples submitted:		(Non-filtered)	Filtered in field with	A: <u>2 mi H₂SO₄/L 2</u> /	added 5ml HNO3
No. of samples submitted:		IF: Whole sample (Non-filtered)		At 2 mi H ₂ SO ₄ /L 2/	Units Date analyzed
No. of samples submitted: NA: No a IALYTICAL NF, NA Conductivity	Acid added RESULTS from (Corrected)	IF: Whole sample (Non-filtered) Other-specify: m SAMPLES Units Date	e analyzed F, NA	915)	Units Date analyzed
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No. of samples submitted: NA: No a NA: No a	Cid added RESULTS from (Corrected)	IF: Whole sample (Non-filtered) Other-specify: m SAMPLES Units Date // // // mg/l c.005 mg/l	e analyzed F. NA Calcium (009 Calcium (009 Calcium (009 Potassium (0 Bicarbonate Chloride (009 Sulfate (0094 Total filterable (dissolved) (7 Chher: F. A-H ₂ SO ₄	915)	Units         Date analyzed          mg/l
No. of samples submitted: NA: No a NA: No a	Cid added	IF: Whole sample (Non-filtered) CF: ( Other-specify: m SAMPLES Units Date mg/l <.005	e analyzed F, NA a analyzed F, NA Calcium (009 Calcium (009 Potassium (0 Bicarbonate Chloride (009 Sulfate (0094 Total filterable (dissolved) (7 Chher: F, A-H ₂ SO ₄	915) (00925) (30) (00440) 940) 45) 9 residue 70300) Nitrate-N	Units         Date analyzed           mg/l
No. of samples submitted: NA: No a NA: No a NA: No a NF, NA Conductivity 25°C (00095 Total non-filte residue (sus (00530) Other: Other: Other: Other: A-H ₂ SO ₄ Nitrate-N +, I total (00630) Ammonia-N Total Kjeldah	Cid added RESULTS from (Corrected) (Corrected) mables anded Nitrate-N total (00610)	IF:       Whole sample (Non-filtered)       IF:       IF:	e analyzed F. NA Calcium (009 Calcium (009 Calcium (009 Potassium (0 Bicarbonate Chloride (009 Sulfate (0094 Total filterable (dissolved) (7 Chher: F. A-H ₂ SO ₄	915)       930)       00925)       00935)       (00440)       940)       45)       9 residue       70300)       Nitrate-N       0631)	Units         Date analyzed          mg/l
No. of samples submitted: NA: No a NALYTICAL NF, NA Conductivity 25°C (00095 Total non-filte residue (sus (00530) Other: Other: Other: A-H ₂ SO. Nitrate-N +, 1 total (00630) Ammonia-N (1 Total Kjeldah ( ) Chemical oxy	Icid added I	IF:       Whole sample (Non-filtered)       IF:       IF:	e analyzed F, NA a analyzed F, NA Calcium (009 Calcium (009 Calcium (009 Potassium (0 Bicarbonate Chloride (009 Chloride (000 Chloride (000 Chlori	915)         (00925)         )30)         )935)         (00440)         940)         45)         9 residue         70300)         Nitrate-N         2631)	Units         Date analyzed           mg/l
No. of samples submitted: NA: No a NALYTICA! NF, NA Conductivity 25°C (00095 Total non-filte residue (sus (00530) Other: Other: Other: A-H ₂ SO ₄ Nitrate-N + , i total (00630) Ammonia-N i Total Kjeldah ( ) Chemical oxy demand (003	Acid added RESULTS from (Corrected) (Corrected) (Corrected) Anded AN Anded AN AN AN AN AN AN AN AN AN AN	IF:       Whole sample (Non-filtered)       IF:       IF:	e analyzed F, NA a analyzed F, NA Calcium (009 Calcium (009 Calcium (009 Potassium (0 Bicarbonate Chloride (009 Chloride (000 Chloride (000 Chlori	915)         (00925)         )30)         )935)         (00440)         940)         45)         9 residue         70300)         Nitrate-N         2631)	Units         Date analyzed           mg/l
No. of samples submitted: NA: No a NALYTICAL NF, NA Conductivity 25°C (00095 Total non-filte residue (sus (00530) Other: Other: Other: Cother: F, A-H ₂ SO. Nitrate-N + , i total (00630) Ammonia-N ( Total Kjeldah () Chemical oxy demand (003 Total organic ()	Acid added RESULTS from (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Cor	IF:       Whole sample (Non-filtered)       IF:       IF:	e analyzed F, NA a analyzed F, NA Calcium (009 Calcium (009 Calcium (009 Potassium (0 Bicarbonate Chloride (009 Chloride (000 Chloride (000 Chlori	915)         (00925)         )30)         )935)         (00440)         940)         45)         9 residue         70300)         Nitrate-N         2631)	Units         Date analyzed           mg/l
No. of samples submitted: NA: No a NALYTICAP NF, NA Conductivity 25°C (00095 Total non-filter residue (sus (00530) Other: Other: Cother: Cother: F, A-H ₂ SO ₄ Nitrate-N + , total (00630) Ammonia-N filter Total Kjeldah () Chemical oxy demand (003 Total organic () Other: () Cherical oxy demand (003 Total organic () Other: () Other: () () () () () () () () () ()	Acid added RESULTS from (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Corrected) (Cor	IF:       Whole sample (Non-filtered)       IF:       IF:	e analyzed F, NA a analyzed F, NA Calcium (009 Calcium (009 Calcium (009 Potassium (0 Bicarbonate Chloride (009 Chloride (000 Chloride (000 Chlori	915) (00925) (30) (00935) (00440) 940) 45) 9 residue 70300) Nitrate-N 0631) dissolved II-N Date Repor	Units         Date analyzed           mg/l
No. of samples submitted: NALYTICAL NF, NA Conductivity 25°C (00095 Total non-filter residue (sus (00530) Other: Other: Cother: F, A-H ₂ SO ₄ Nitrate-N+, f total (00630) Ammonia-N f Total Kjeldah () Chemical oxy demand (003 Total organic	Cid added	IF:       Whole sample (Non-filtered)       IF:       IF:	e analyzed F, NA a analyzed F, NA Calcium (009 Calcium	915) (00925) (30) (00935) (00440) 940) 45) 9 residue 70300) Nitrate-N 0631) dissolved II-N Date Repor	Units         Date analyzed           mg/l
No. of samples submitted: NA: No a NALYTICAL NF. NA Conductivity 25°C (00095 Total non-filter residue (sua (00530) Other: Other: Cother: Cother: Cother: F. A-H ₂ SO ₄ Nitrate-N + , f total (00630) Ammonia-N f Total Kjeldah () Chemical oxy demand (003 Total organic () Other: Other: Other: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother:	Cid added	IF:       Whole sample (Non-filtered)       IF:       IF:	e analyzed F, NA a analyzed F, NA Calcium (009 Calcium	915) (00925) (30) (00935) (00440) 940) 45) 9 residue 70300) Nitrate-N 0631) dissolved II-N Date Repor	Units         Date analyzed           mg/l
No. of samples submitted: NA: No a NALYTICAL NF. NA Conductivity 25°C (00095 Total non-filter residue (sua (00530) Other: Other: Cother: Cother: Cother: F. A-H ₂ SO ₄ Nitrate-N + , f total (00630) Ammonia-N f Total Kjeldah () Chemical oxy demand (003 Total organic () Other: Other: Other: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother: Cother:	Cid added	IF:       Whole sample (Non-filtered)       IF:       IF:	e analyzed F, NA a analyzed F, NA Calcium (009 Calcium	915) (00925) (30) (00935) (00440) 940) 45) 9 residue 70300) Nitrate-N 0631) dissolved II-N Date Repor	Units         Date analyzed           mg/l

ICAP SCREEN

Lab Number: HM 485 Date Submitted: 3/22/85 By: Boyer

Sample Code: Flora Vista Water Supply Well #5 Date Reported: 5/31/85 By: Rath

Determination

Concentration (µg/m1)

Aluminum			<.10
Barium			2.10
Beryllium			2.10
Boron			4.10
Cadmium			4.10
Calcium	· .	112.	•
Chromium			4.1D
Cobalt			4.10
Copper	÷ .		c.10
Iron	- 		4.10
Lead	6 		L.10
Magnesium		11.	
Manganese		. 39	
Molybdenum		+	4.10
Nickel			L.1D
Silicon +		5.9	
Silver			6.10
Strontium		2.8	
Tín			2:10
Vanadium			4.10
Yttrium			<.1D
Zinc			4.10

70	CIENTIFIC LAB O Camino de S buquerque; NN			, G		OGEN ANALY	SIS
ATE RECEIVED	22 8.5 1	B. WC -120		300 🗆 59600 🕅	OTHER: 822	235	
3 20 85	· · ·	SITE	Sample location	Animas Re	ries		
Dilection TIME 1635		ATION	Collection site descrip	is River di	eather		1.
billected by — Person/Agen	Boygy/	KIZ	Anima	s pire all	Terrey	opposi	Ne N
51					-FED	a infle	evell
IND: NM	OIL CONS	AL BUREAU	DIVISION		#3		
PORI	ate Land nta Fe, N		dg, PO Box 20				
	David Boy			• •			
Au.,					Station/		
					well code- Owner		
	Pump	Water level		Discharge		Sample type	4
	Тар					67.	
oH (00400)	$\smile$	Conductivity (U	HG umh	Water Temp. (00010)	•C	Conductivity at 25°	νC (00094) μmh
ield comments						<u> </u>	
AMPLE FIELD T	DEATMENT	Chook on					
	<b>LIENIMEN</b>		pher poxes				
No. of samples		Whole sample	e» 👝 📻 Filtered	in field with 🗖 🗛 🦿	2 ml HaSO ./I	added	
No. of samples submitted	Ş⊄ NF	(Non-filtered)	<b>Δ F:</b> 0.45 μr	in field with A: 2	2.ml H₂SO₄/I	added	
No. of samples	<u> </u>	(Non-filtered)	<b>Δ F:</b> 0.45 μr	nembrane filter	2.ml H₂SO₄/I	_ added	
No. of samples submitted	added:	(Non-filtered) )ther- <i>specify:</i>	⊡ <b>F</b> : 0.45 μr	nembrane filter	2.ml H₂SO₄/I		Date analyzed
No. of samples submitted NA: No acid NALYTICAL RE: NF, NA	added:	(Non-filtered) Other- <i>specify:</i>	<b>Δ F:</b> 0.45 μr	zed A NA NF	2.ml H₂SO₄/l	Units	Date analyzed
No. of samples submitted NA: No acid NALYTICAL RE: NF, NA	added:	(Non-filtered) )ther- <i>specify:</i>	⊡ <b>F</b> : 0.45 μr	zed A NA NF De Calcium (00915) Magnesium (00925		Units ,↓mg/l 중mg/l	4/15 4,0
No. of samples submitted NALYTICAL RES NF, NA Conductivity (Cor 25°C (00095)	added:  C SULTS from rected)	(Non-filtered) Other- <i>specify:</i>	Units Date analy	zed A NA N F A Calcium (00915) Magnesium (00925) Se Sodium (00930)		Units ,↓mg/I ≤mg/I .ろmg/I	4115
No. of samples submitted	added:  C SULTS from rected)	(Non-filtered) Other- <i>specify:</i>	Units Date analy μmho	zed A NA N F Calcium (00915) Calcium (00930) Sodium (00930) Potassium (00935) Bicarbonate (00440	i)i2. 	Units .4 mg/l \$ mg/l 3 mg/l 5 mg/l 62.0 mg/l	4/15 4/15 4,0 3/28 3/28 5/6
No. of samples submitted NA: No acid a NALYTICAL RE: NF, NA Conductivity (Cor 25°C (00095) Total non-filterable residue (suspend (00530)	added:  C SULTS from rected)	(Non-filtered) Other- <i>specify:</i>	Units Date analy	zed A NA NF Zed A NA NF A Calcium (00915) A Sodium (00930) A Potassium (00935) B Bicarbonate (00440 Chloride (00940)	i)(2 i)(2 25   ()	Units 14 mg/l 5 mg/l 5 mg/l 5 mg/l 62.0 mg/l 12.5 mg/l	4/15 4/15 4,0 3/28 3/28 5/6 4/15
No. of samples submitted	added:  C SULTS from rected)	(Non-filtered) Other- <i>specify:</i> SAMPLES	Units Date analy μmho	zed A NA N F Calcium (00915) Calcium (00915) Calcium (00930) Calcium (00930) Calcium (00930) Calcium (00930) Calcium (00930) Calcium (00930) Calcium (00930) Calcium (00930) Calcium (00940) Chloride (00940) Chloride (00945)	i) i)                                                                                                                                                                                                                                                                                                                                                                          _	Units -4 mg/l - 8 mg/l - 3 mg/l - 5 mg/l - 6 2.0 mg/l - 12.5 mg/l - 5 3 mg/l -	4/15 4/15 4,0 3/28 3/28 5/6
No. of samples submitted	added:  C SULTS from rected)	(Non-filtered) Other- <i>specify:</i> SAMPLES	Units Date analy μmho	zed A NA NF Zed A NA NF A Calcium (00915) A Ganesium (00925) A Sodium (00930) A Potassium (00935) B Bicarbonate (00440) A Sulfate (00940) A Sulfate (00945) A Sulfate (00945) A Sulfate (00945)	(i) (2) (i) (2) (j)	Units 14 mg/l 5 mg/l 5 mg/l 5 mg/l 62.0 mg/l 12.5 mg/l	4/15 4/15 4,0 3/28 3/28 5/6 4/15
No. of samples submitted NA: No. acid a NALYTICAL RE NF, NA Conductivity (Cor 25°C (00095) Total non-filterable residue (suspend (00530) Other: PH Other: Other:	added:  C SULTS from rected) e e e	(Non-filtered) Other- <i>specify:</i> SAMPLES	Units Date analy μmho	zed A NA NF Zed A NA NF A Calcium (00915) A Ganesium (00925) A Sodium (00930) A Potassium (00935) B Bicarbonate (00440) A Sulfate (00940) A Sulfate (00945) A Sulfate (00945) A Sulfate (00945)	(i) (2) (i) (2) (j)	Units 14 mg/l 5 mg/l 5 mg/l 5 mg/l 62.0 mg/l 12.5 mg/l 2.3 mg/l 69 724 mg/l 2.3	4/15 4/15 4/15 4/15 3/28 3/28 75/6 4/15 4/15 5/6 5/6 4/15
No. of samples submitted NA: No: acid a NALYTICAL RE: NF, NA Conductivity (Cor 25°C (00095) Total non-filterable residue (suspend (00530) Other: Other: Other: Other:	added:  C SULTS from rected) e ed)	(Non-filtered) Other- <i>specify:</i> SAMPLES	Units Date analy μmho	zed A NA NF Zed A NA NF Calcium (00915) Calcium (00930) Calcium (00930) Sodium (00930) Calcium (00930) Sodium (00930) Chloride (00940) Chloride (00940) Chloride (00945) Total filterable residu (dissolved) (70300) Chter: Rourid	(i) (2) (i) (2) (j)	Units 14 mg/l 3 mg/l 3 mg/l 3 mg/l 3 mg/l 4 5 mg/l 12.5 mg/l 12.5 mg/l 6 2 2 4 2 2 4 2 2 4 2 2 4 2 2 4 2 2 4 2 2 4 2 2 2 2 2 2 2 2 2 2 2 2 2	4/15 4/15 3/28 3/28 15/6 4/15 4/15 4/15 5/6
No. of samples submitted NALYTICAL RE: NF, NA Conductivity (Cor 25°C (00095) Total non-filterable residue (suspend (00530) Other: Other: Other:	added:  C SULTS from rected) e ed)	(Non-filtered) Other- <i>specify:</i> SAMPLES	Units Date analy μmho	zed A NA NE Zed A NA NE A: 4 A: 4 Zed A NA NE A: 4 Zed A NA NE	$ \begin{array}{c}                                     $	Units 14 mg/l 5 mg/l 5 mg/l 5 mg/l 62.0 mg/l 12.5 mg/l 2.3 mg/l 69 724 mg/l 2.3	4/15 4/15 4/15 4/15 3/28 3/28 75/6 4/15 4/15 5/6 5/6 4/15
No. of samples submitted NALYTICAL RE: NF, NA Conductivity (Cor 25°C (00095) Total non-filterable residue (suspend (00530) Other: Other: Other: Nitrate-N +, Nitrat total (00630)	added:  C SULTS from rected) e ed)	(Non-filtered) Other- <i>specify:</i> SAMPLES	Units Date analy μmho mg/l	A       A         Zed       A         NA       NF         A       Calcium (00915)         A       Magnesium (00925)         A       Sodium (00930)         A       Potassium (00935)         Bicarbonate (00440)       A         Chloride (00940)       A         Sulfate (00945)       A         Other:       Potasium (00930)         F, A-H2 SO4       A         Nitrate-N + , Nitrate       A	$ \begin{array}{c}                                     $	Units .4 mg/l 8 mg/l 3 mg/l 3 mg/l 5 mg/l 62.0 mg/l 12.5 mg/l 68 274 mg/l .3 7 .3 7	4/15 4/15 4/15 4/15 3/28 3/28 75/6 4/15 4/15 5/6 5/6 4/15
No. of samples submitted NALYTICAL RE NF, NA Conductivity (Cor 25°C (00095) Total non-filterable residue (suspend (00530) Other: Other: Other: Nitrate-N +, Nitrat total (00630) Ammonia-N total	added:  C SULTS from rected) e ed)	(Non-filtered) Other- <i>specify:</i> SAMPLES	Units Date analy mho mg/l mg/l	zed A NA NE Zed A NA NE A: 4 A: 4 Zed A NA NE A: 4 Zed A NA NE	i) i)   ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie ie	Units 14 mg/l 5 mg/l 5 mg/l 5 mg/l 62.0 mg/l 12.5 mg/l 2.3 mg/l 69 724 mg/l 2.3	4/15 4/15 4/15 4/15 3/28 3/28 75/6 4/15 4/15 5/6 5/6 4/15
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ENERGY AND MINERALS DEPARTMENT OIL CONSERVATION DIVISION

TONEY ANAYA GOVERNOR

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87501 (505) 827-5800

February 15, 1985

MEMORANDUM

TO: R. L. STAMETS, DIRECTOR, OCD

FROM: D. G. BOYER, GEOLOGIST, ENVIRONMENTAL BUREAU

SUBJECT: FLORA VISTA WATER SUPPLY SAMPLING

The OCD and EID are cooperating in an effort to delineate and define the current status of hydrocarbon contamination in the vicinity of the Flora Vista Water Users Association water supply wells. Staff of both agencies plan to be in the Farmington area March 6 to 8 to perform field investigations at this and other sites. Investigation at the Flora Vista site will include sampling of the contaminated well and on-line water supply wells, trenching in the vicinity of the suspected source(s) to determine current contamination status, and installation of monitoring wells as necessary. Testing of samples will include analyses for organic contaminants, heavy metals and general water chemistry.

Because of the nature of the subsurface material (coarse gravels, cobbles and boulders), a backhoe machine will be necessary to reach the water table. I understand the Water Users Association has such equipment available. Attempts today to contact Mr. Bert Barnes (President) to discuss the proposed schedule and backhoe availability have been unsuccessful. I will continue coordination of this effort and will keep you informed.

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## STATE OF NEW MEXICO ENERGY AND MINERALS DEPARTMENT OIL CONSERVATION DIVISION

TONEY ANAYA GOVERNOR January 25, 1985

POST OFFICE BOX 2088 STATE LAND OFFICE BUILDING SANTA FE, NEW MEXICO 87501 (505) 827-5800

Representative Don Silva, Chairman House Energy and Natural Resources Committee Executive Legislative Building Santa Fe, New Mexico 87503

Dear Chairman Silva:

This letter is in response to the request of the House Energy and Natural Resources Committee for information regarding hydrocarbon contamination of a public water-supply well in the community of Flora Vista in San Juan County. The January 23, 1985, Environmental Improvement Division (EID) letter to you summarizes the current situation. One of the public supply wells was contaminated in 1983 and taken out of service while other nearby water wells have shown no evidence of contamination based on EID monitoring. The suspected source of contamination was one or more pits at a nearby natural gas well that received unusable water and other fluids produced with the gas. While the primary pit at this location was lined, it is my understanding that the fiberglass tank developed a leak. The fiberglass tank has been replaced and a small steel tank has been placed in the second pit. The Oil Conservation Division (OCD) is cooperating with EID in planning further tests and sampling on the contaminated water well later this winter. To our knowledge, this is the only contamination case of this type in Northwest New Mexico.

The OCD has initiated steps to prevent such contamination from occurring in the future in the San Juan Basin. Last July the OCD organized a study committee composed of industry, government and citizen groups to investigate the location and extent of San Juan Basin aquifers vulnerable to contamination by disposal of produced water. The committee is to also prepare recommendations for protection of such aquifers. There has been a ban on improper surface disposal of such fluids in Southeast New Mexico since 1969. No such ban has been actively sought in the San Juan Basin before this time as the volumes of produced water there have been generally much smaller and the quality has been better. However, new concerns related to reports of dissolved hydrocarbons in the water exceeding health standards and the Flora Vista case have caused us to pursue such a ban there as well. Representative Don Silva, Chairman House Energy and Natural Resources Committee Page 2

An Oil Conservation Commission (OCC) hearing scheduled for February 20, will hear study committee recommendations to prohibit and/or limit such disposal. It is expected that the outcome of this process will be an OCC order prohibiting or controlling disposal of such fluids in a manner that will protect fresh water supplies used for drinking, industrial or agricultural uses.

If you wish further information on this matter, please do not hesitate to contact me.

Sincerely

R. L. STAMETS, Director

RLS/DB/dr

cc: Paul Biderman, Secretary Energy and Minerals Department Richard Holland, Deputy Director Environmental Improvement Division Members, House Natural Resources Committee



January 23, 1985

Representative Don Silva, Chairman House Energy and Natural Resources Committee Executive Legislative Building Santa Fe, New Mexico 87503

Dear Chairman Silva:

As you know, Flora Vista had to shut down one of its public water-supply wells in 1983 due to hydrocarbon contamination, the source of which appears to be nearby petroleum-production activities. The Oil Conservation Division (OCD) of the Energy and Minerals Department regulates such activities, including their environmental impacts, under the authority of the Oil and Gas Act and the Water Quality Act. The Environmental Improvement Division (EID), however, became involved with the problem for the following reasons:

- 1. EID regulates and monitors the qualilty of water served by such public systems;
- 2. EID has been conducting, with funds provided by the U.S. Environmental Protection Agency, a statewide investigation of organic water contaminants, including those attributable to the petroleum industry; and
- 3. EID explored the possibility of using the federal "Superfund" program to remedy the problem.

Based upon EID's monitoring, Flora Vista's remaining water wells have shown no evidence of contamination. In the summer of 1984, the EID attempted to drill monitoring wells in the area for the purposes of determining the exact source(s) of contamination and whether or not any of Flora Vista's remaining wells were in danger of contamination. EID's hollow-stem drill rig, however, was not capable of penetrating the large quartzite gravel and boulders; EID is exploring the feasibility of acquiring the appropriate drilling capabilities either by capital purchase or by contract services. Lastly, the EID has been advised by the U.S. Environmental Protection Agency that petroleum contamination problems such as Flora Vista are ineligible for "Superfund" cleanup actions. Don Silva January 23, 1985 Page 2

By copy of the this letter I am informing the OCD of the House Energy and Natural Resources Committee's interest in Flora Vista, and of your request that the OCD provide information to the committee regarding OCD's present and pending activities to address this problem.

Sincerely, Richard Holland Deputy Director, EID

RH/DM/dlr

cc: Richard Stamets, Director, Oil Conservation Division Members, House Natural Resources Committee





DENISE D. FORT DIRECTOR

ENVIRONMENTAL IMPROVEMENT DIVISION P.O. Box 968, Sente Fe, New Mexico 87504-0968 (505) 984-0020

STATE OF NEW MEXICO

December 7, 1984

Marty Buys Environmental Coordinator Tenneco Oil Co. P.O. Box 3249 Englewood, CO 80155

Dear Mr. Buys:

Douglas Earp, a member of my staff, attended the meeting of the Short Term Water Study Committee on November 29 and after having discussed with him the written recommendations presented and discussed at that meeting, I would like to offer the following comments for your consideration prior to preparation of a revised Recommendations document. The organization and numbering of my comments follow the format of the committee report.

A. Definitions

- 3. Blowdown Pit: Wording related to the length of time that liquid is allowed to stand in the pit might be deleted from this definition if changes of the type suggested subsequently in regard to the blowdown exemption are made.
- 6. Special Areas: This definition should more clearly specify that the current list of special areas is based on preliminary information and that any other areas (outside of the "vulnerable areas") in which ground water is subsequently found to be within 50 feet of the ground surface will automatically be classified as "special areas". The random scatter of areas currently included on the list suggests that the water table in surrounding sections might also be within 50 feet of the ground surface. Areas of vulnerable, high quality ground water should be protected whether or not they are presently being used as a source of water.
- 7. Vulnerable Aquifer: The term "floodplain area" might be defined more specifically (e.g. with respect to the 50 year or 100 year storm event or with respect to elevation above the river). Wording of the parenthetical exclusion following "unconfined aquifers" is too general and might provide an unacceptable loophole. An unconfined aquifer should only be exemptable if it can be reasonably demonstrated that the thickness, makeup, and hydrologic properties of the overlying geologic materials eliminate the possibility that contaminants will ever reach the ground water.

Mr. Marty Buys Page -2-December 7, 1984

#### B. Exemptions

- 1. Blowdown Pits: Discussion during the November 29 meeting demonstrated that the intention of this exemption is to eliminate pits which only contain liquid infrequently. I understand the need to apply limited resources to address the most serious problems first, but I have several reservations with this exemption.
  - First, exemption of pits "in which liquid is not allowed to stand for a period exceeding 96 hours" may inadvertently exempt pits having the greatest potential to contaminate ground water since pits located in highly permeable materials will drain faster than pits located in less permeable materials.
  - Secondly, it is erroneous to assume that contaminants will only migrate toward the water table as long as liquid is present in the pit (i.e. under a positive hydrostatic pressure). Liquid will continue to percolate downward by unsaturted flow as long as there is net potential gradient in the downward direction. A suction or tension (i.e. negative pressure) gradient of this type will be present until such time as the moisture potential equilibrates throughout the unsaturated soil profile (i.e. possibly for several weeks following each application of liquid to the pit). Contaminants can thereby be carried to the water table by surges of unsaturated flow resulting form a series of short term applications of liquid (blowdown or natural precipitation) to the overlying pit, as well as by saturated flow if liquid even occasionally remains in the pit for an extended period of time.

A more acceptable approach to the blowdown pit exemption would be to evaluate sites on the basis of a) the total volume of liquid discharged to a pit during a specified period of time; b) chemical characteristics of the blowdown liquid (regarding organic as well as inorganic constituents); and c) the attenuation capacity and hydraulic characteristics of the unsaturated zone (e.g. clay and organic matter content). An even better approach would be to eliminate the blowdown pit exemption entirely because of the uncertainties regarding chemical characteristics of the blowdown liquid and variabilities in frequency of blowdown at various wells.

3. Adequacy of the discharge volume and depth to ground water limitations included in the dehydrator pit exemption should be evaluated experimentally. Chemical characteristics of the liquid and chemical attenuation capacity and hydraulic characteristics of the soil should also be considered prior to granting an exemption.

#### C. Permits

- 1. Discharge Volume Permit: The adequacy of numerical standards included in this section should be verified prior to adoption.
- 2. Quality Permit: The operator should be required to demonstrate that concentrations of both organic (including aromatic hydrocarbons) and

Mr. Marty Buys Page -3-December 7, 1984

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inorganic compounds (TDS and Cl at a minimum) in liquid discharged to the pits will not cause degredation of the ground water. A discharge permitted solely on the basis inorganic chemical constituents is not acceptable.

- 3. Soil Characteristics Permit: The term "low permeabilities" should be quantified. Also, the capacity of the soil to retard movement of both inorganic and organic contaminants, based in part on clay and organic matter content, should be considered during permit evaluation.
- D. Conclusions: The statement that "no real evidence of contamination of these waters was found" based on a review of records of various agencies is misleading. At least one water supply well near Flora Vista (water quality data provided in table attached to this letter) has been abandoned due to phenolic and oil and grease contamination originating from the vicinity of a gas well. The lack of additional data may reflect past failures to look for ground water contamination related to oil and gas activities as much as it reflects the absence of contamination.

As a general comment, I would like to reiterate the EID position that the permit evaluation process must include consideration of organic constituents (particularly the benzene family) as well as inorganic constituents of liquid discharged to unlined pits.

Your consideration of these comments is appreciated. Please feel free to contact me or Douglas Earp at (505) 984-0020 if you have questions or if we can be of further assistance to your committee.

Sincerely.

Anthony Drypolcher Bureau Chief Ground Water/Hazardous Waste Bureau

AD/DE/ps

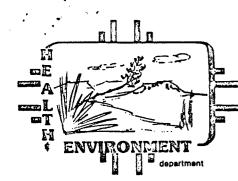
cc: Dave Boyer Chris Shuey

Sampling Station	Contaminated Well	Online
Sampling Date	8/83	6/19/84 (µ
Calcium		148./150.
Magnesium	20.75	9.6/11.
Sodium		41.4
Potassium		0.00
Bicarbonat <b>e</b>		312.7
Sulfate	200.	187.
Chloride	10.4	20.9
Nitrate-N	0.5	0.05
Ammonia-N		0.11
Aluminum	2.21	<0.10
Arseni <b>c</b>	1.56	
Barium	< 0.005	<0.10
Beryllium		< 0.10
Boron	< 0.004	<0.10
Cadmium	< 0.002	< 0.10
Chromium	< 0.005	< 0.10
Cobalt	< 0.003	<0.10
Copper	< 0.002	<0.10
Iron	0.15	< 0.10
Lead	< 0.001	<0.10
Manganes <b>e</b>	0.32	0.45
Mercury	0.63	-
Molybdenum	< 0.005	<0.10
Nickel	<0.10	<0.10
Selenium	< 0.002	-
Silicon		5.9
Silver		<0.10
Strontium		2.6
Tin		<0.10
Vanadium		<0.10
Yttrium		<0.10
Zinc		<0.10
TDS	55 <b>8</b> .	655.
pH (unit <b>s)</b>	7.3	7.3
COD	•••	4.0
TOC	125.	1.0
Aromati <b>c</b> Purgeabl <b>es</b>	<0.01	ND (Sampled later in 1984)
Phenols	0.40	
Oil & Greas <mark>e</mark>	32.8	
Collector	OCD	EID
Analyst	Ana Cor	SLD
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Table 23. Flora Vista Area 30N.12W.23 All units are mg/L unless otherwise indicated.

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State of New Mexico

ENVIRONMENTAL IMPROVEMENT DIVISION DISTRICT I FIELD OFFICE/724 WEST ANIMAS FARMINGTON, NEW MEXICO 87401

Steven Asher, Director

Telephone #(505)327-9851

May 23, 1984

Mr. Bert Barnes, President Flora Vista Water Users Association P.O. Box 171 Flora Vista, New Mexico 87415

Dear Mr. Barnes:

The annual environmental survey of the Flora Vista Water Supply System was conducted on April 24, 1984. The survey report is attached for your review and file.

The system supplies water to approximately 1,600 consumers. Presently, the system purchases some of its water supply from the City of Aztec and generates the rest of its water supply from wells. Wells number 1 and 3 remain out of service due to oil and grease contamination.

The association's contract with Aztec limits the purchase of water to 36 million gallons per year. This limitation may result in periods of low pressure during the summer when water demand is the highest.

Deficiencies noted during the inspection are listed below:

- 1. Maintenance records are not being kept. Proper maintenance records would help in identifying problem areas of the system.
- 2. Fan switch, in pump house chlorination room, should be provided on O outside of the pump house Private in the state of the pump house of the pump house
- O outside of the pump house. Relocating the fan switch outside the pump house would essentially eliminate the possibility of an operator being overcome by chlorine gas.
- 3. No screen on overflow pipe for 300,000 gallon tank. The screen would prevent the entrance into the tank's overflow system of small animals, insects or birds.

TONEY ANAYA

GOVERNOR

JOSEPH F. JOHNSON

DEPUTY SECRETARY

Virs'

Joseph Goldberg, Secretary

WATER SUPPLY REGULATION SECTION

Mr. Bert Barnes Page 2 May 23, 1984

I wish to thank Mr. Ray Penrod and Mr. Philip Cheney for their time and cooperation in conducting the survey. If I can be of any assistance to you, please contact me.

Sincerely,

H noeligner Dispherer Arcelious Stephens, Environmentalist III

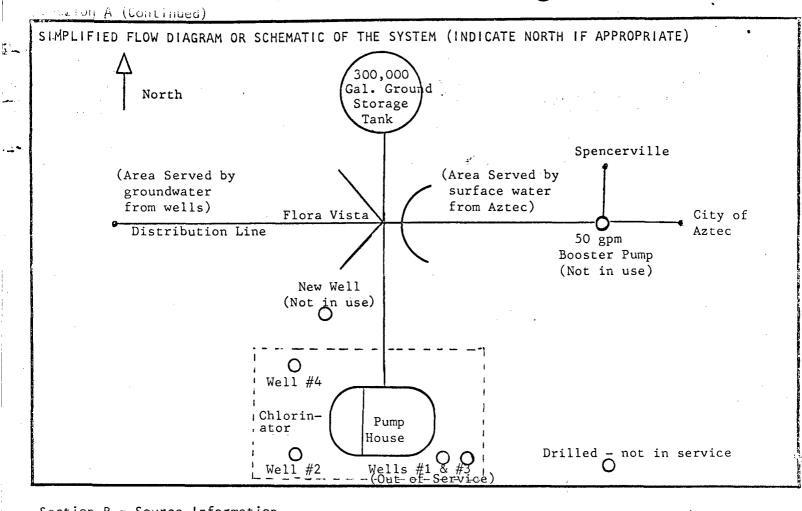
AS:lm cc:File Water Supply Section

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DF客GF客F(V/F客) JUN 221984 С<del>.,</del>

WATER SUPPLY REGULATION SECTION

N/A Not Applicable		Community Water Supply System
N-Av - Not Available Est Estimated	H ENVIRONMENT	Inspection Form
	h and Environment Department mental Improvement Division	Inspection i Date: 04-24-84
WSS CODE: WATER SUPPLY 100-24 SYSTEM NAME Flora Vis	sta Water Users	COUNTY: San Juan
System Address/LOCATION	Flora Vista, New Mexi	co 87415
	DRESS (if different the	
Population # # Meters Served Connections	Max. System Proc	duction Average System Produc- GPD tion
1,600 Est. 550 550	Poten. 🗌 Actu	ual03-20-84
System Source Distribution Only (check Approp. Boxes) Spring(s)	Well	(s) X # of Wells <u>2 in use</u> Surface
Additional or Qualifying Information:	·.	
Aquadene – a phosphate chemical added t	o water to keep chlori	ne from turning water black
from iron and manganese.	<u></u>	
Pump house to be expanded later this ye	ar	
Fan switch in pump house chlorination r		
Wells #1 and #3 contaminated with oil a system gets its water from the city of	NAMES OF TAXABLE PARTY. NO. OF TAXABLE PARTY.	
system gets its water from the city of	Aztec and werrs #2 and	#4. well #4 is a new well.
	· · · · · · · · · · · · · · · · · · ·	
••••••••••••••••••••••••••••••••••••		
<u> </u>		
System Personnel - Name/Classification	Level of Certif	ica- Level of Certification Obtained
Ray Penrod/Manager	None	None
Philip Cheney/Assistant	None	None
Information furnished by:		Busings bone:
Ray Penrod/Manager Survey Performed by: Arcelious Stephens		REGULATION SECTION Phone: 327-9851



#### Section B - Source Information

WELL Identification	Well Depth	Pump Depth	Pump Capacity GPM	Well Drawdown feet	Pump Type	Static Water Level	Age of Pump	Date Well Duilled
Well #1	25'	18'	60–70	5'-6'	Submers.	~ 8'	2 yrs.	1980
Well #2	25'	20½'	60-70	5'-6'	Submers.	<u>≃ 8'</u>	2 yrs.	1980
Well #3	251	N/A	N/A	N/A	Submers.	~ 8'	N/A	1980
Well #4	26 '	21'	45	15'	Submers.	≈ 81	New	03/83

Remarks, Defigiencies, and Recommendations

Source	Number (each)	Total Capacity	Remarks, Deficiencies, and Recommendations
Artesian Wells			<u>ທີ່ອີຊຕາສາ(ທາສາມ</u>
Springs			JUN 2 2 1984
Infiltration Galleries			WATER SUPPLY
			REGULATION SECTION .

EID - 042 Revised 11/82

THE CONTRACTOR OF A DESCRIPTION OF A DES	GEVITY	Stor	age a	rress	ure i	dink and	sciencer (5			•	
Gravity Stor Reservoir In				Date Buil		terior	Tank Material	Cathodic Protection	Tank Øpenings Secured	Overflow Screened	Vent Screen
#1	,000	1979	Go	od	Steel	No	Yes	No	Yes		
										/ 	
Pressure Tank	I.D.	Vol (ga	ume llons)	Age		erior dition	Additiona	l Informatio	on	· ·	
		<u> </u>							<u></u>		
Remarks, Defi	ciencies	or	Recomm	endatio	ns		l				
Section D - W	ATER DIS	TRIB	UTION								
Booster Pump Stations	Total N of Stat		Rare	ely, if	, Deficiencies or Recommendations if ever, used n-line centrifugal pump						
Type of Disin Facilities	fection		Chlorin	ne Gas ,		Remar Purch	ks, etc.Ga nased water	s chlorinat: disinfected	ion provid d by Azted	ded at wel c.	lls;
System Pressure	Max. 90 Min. 55		psi psi	Remar limit	ks, e ation	tc. May s on aπ	v be pressu nount of wa	re problems ter purchase	in summer ed from Az	r due to ztec.	
Pipe Materials In system	PVC 🕅		EL 🗖 er 🔀		valve	s insta		ss-connectic ery metered			
Type of Distr	ibution	Syst	em	Gravit	у 🕅		Pre	ssure $\mathbf{x}_{are}^{Whe}$	en wellßot e in use.	h []	
Contaminant	Next <u>ple D</u>		Sampl Frequ				Rem	arks			
Micro- biological	$\geq$	$\leq$	2								
Turbidity Organics			<u>N/A</u> N/A				<u></u>	l	णिञ्चतिः	8.171775	350
Inorganic	_07/84		3_yea	re							₹╢╢─
Radiological	03/86		4 yea					2		221984	Ш_
Secondary						الاح الذي عصر بيري			WATER	SUPPLY	
Section E -	GENERAL	OPER	ATION	AND MAI	NTENA	NCE (O	8 M)	- 		ON SECTION	)N
Does the sys							Yes 🗍		No		•
Are preventa			· · · · · · · · · · · · · · · · · · ·							i Harikatari Tixtari	
The importan Indicate obs	ce of Cr erved cr	055 055	Connec connec	tion Co tions a	ntrol bove	was di (Pipe M	scussed wi Materials i	th the Opera n System)	ator Yes		Feargent is a consistent
EID - 042 Re	vised 11	/82								Γ	age 3

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## Section F - SURFACE WATER SUPPLY INFORMATION

- ----

Source (check	one)	Remarks or Deficiencies
Lake		
Stream		

N/A

## Section G - WATER TREATMENT PLANT INFORMATION N/A

Unit Operations	Remarks or Deficiencies	
Plant Intake Structure		Countration and the second
Plant Location (Siting)		A STATISTICS AND A STAT
Pretreatment, Raw Water Storage or Presettling Reservoirs		3
Coagulation - Sedimentation		Contraction of the Designation
Chemical Addition		a stationer
Filtration		
Other Treatment (Ion Ex- change, Softening,Reverse Osmosis, etc.)		
Additional Comments or Rem	arks	r ar a caracter and a contract

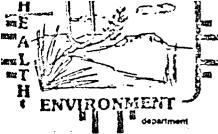
Page 4



Analytical and Environmental Services-

## (505) 345-8964 7300 Jefferson St. NE Albuquerque, NM 87109

	Ana	lytic	al and El	nviro:	nmental 3	ervi	Ces merenderand	in Distances of	<b> </b>	
AnaCo	· · · · · · · · · · · · · · · · · · ·									
		•								
Laboratorie		-						•	1000	
	ronmental nd Water		ovement Di	V1 S10	n		Date: 10 JV- 669	August	1983	4615
	Box 958	Secu	00				Page 1 of	2	00	DUCED
	a Fe, NM	8750	4-0968					-	PICC WAT	ER PIT
									AN	Alysis
Attentio	n: Oscar	Simps	50 N						10 FL	m; E.O.F ORV: 57.4
Analyte			Sample Ide	entifi	cation/Ana	ytic	al Results	5	A R	EA
	#1	(1-F)	#2	(2-E)	#3(deh)	(drat)	or) #4 (4	1-F)	#5 (9	5-F)
• <del>)</del>	<b>π1</b>	(1-6)	π2	(2-1)	• • •	rurau	U() #4 (-	r <b>/</b>	<i>"</i> <b>0</b> ( .	2 ° 1 <i>)</i>
As	1.56	ррт	3.18	ppm	23.64	ppm	3.13	nom	9.85	
Cd	<0.002	ppm	<0.002		0.28		<0.002		0.02	
Cr	<0.005		<0.005		1.28		<0.005	•	0.05	
РЬ	<0.001		<0.001		12.64		0.65		1.14	
Hg	0.63	ppn	0.12		5.80		0.04	•	3.18	
Se		ppm	<0.002	ppm	1.60	DDM	<0.002		0.15	
Ba		ppm	0.07	ppm	2.08	ppm	0.07		0.42	
Fe	0.15	ppm		ppm	19.60	ppm	0.51		<0.05	
Cu		ppm	<0.002	ppm	<0.002		<0.002		<0.002	
Mn	0.32	ppm		ppm	0.64		2.18		<0.005	
Mg	20.75	ppm	34.32	• •	5.32		34.32		34.32	
Zn	<0.004	ppm	<0.004	ppm	<0.004		<0.004		<0.004	
A1		ppm	4.38		<0.05		1.77	ppin	<0.05 16.17	
B		ppm	3.63	ppm	<0.004		<0.004		<0.003	
Co	<0.003	ppm	0.72	ppm	5.00	•			<0.005	
Mo Ni	<0.005 <0.01	p pm	<0.005 <0.01	ppm	<0.005 <0.01		<0.005 <0.01		<0.01	
CN	<0.001		19.58		0.058		0.082		42.30	
F		ppm		ppm	0.33			ppm	12.1	
NO3 as N		ppm	10.8			ppm		ppm		ppm
S04		mg/1	1,750		1,600			ppm		ppm
C1	10.4		1,164			ppm		ppm	1,874	•••
pH	7.3	P P	7.4	PP'''	6.9	<b>P</b> P	7.8	P.P.Y	8.3	F F · ·
TDS	558.0	ma/1	34,454.0	ma/1	NA*		884.0	mg/1	12,300.0	mq/1
Phenols	0.40			ppm	0.30	naa	0.60		0.35	
0il& Gre			37.1		NA*	с <i>т</i>	37.5		54.7	
Benzene	<0.01		0.18		13.0	naa	<0.01		0.05	
Toluene	<0.01		<0.01		13.4		<0.01		0.042	
			(some	• •	(72 % gly				(heavier	
			hydrocarl	oons)	rest hyd				hydroca	
			-		carbon s		ts)		present	
TOC	125.07	nqc	171.03	ppm	797.36	ppm	126.93	ppm	640.75	ppm



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GOVERNOR .

ROBERT MENEILL SECRETARY

ROBERT L. LOVATO, M.A.P.A. DEPUTY SECRETARY

> JOSEPH F. JOHNSON DEPUTY SECRETARY

August 23, 1983

Mr. George Madrid Lawerance A. Brewer & Associates P.O. Box 2079 Farmington, NM 87499

RE: Water Analysis of Floravista

Dear Mr. Madrid:

As per our phone conversation of August 18, 1983, I am forwarding you a copy of the water analysis results of the Floravista Well field area.

STATE OF HEW ME

ENVIRONMENTAL IMPROVEMENT DIVISION

P.O. Box 963, Santa Fe, New Mexico 87504-0968

(505) 984-0020

Steve Asher, Director

Sample #1 is from water well #1.

Sample #2 is from the oil-water separator next to the gas well.

Sample #3 is from the glychol dehydrator. 52 UPGE

Sample #4 is a sample of water that was in a previously dug open pit. The water level in the pit possibly represents the water table.

Sample #5 is not associated with the Floravista problem.

If you have any questions regarding this matter, please contact me at the above telephone number and addresses.

Sincerely,

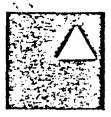
Auro Sompson II

Oscar A. Simpson Water Resource Specialist Ground Water Section

OAS:egr

Enclosure

#### EQUAL OPPORTUNITY EMPLOYER



# LAWRENCE A. BREWER & ASSOCIATES, INC.

Lawrence A. Brewer, P.E., L.S. Richard P. Cheney, P.E., L.S. George T. Walters, L.S. Robert A. Echols, Jr., P.E.

CONSULTING ENGINEERS

August 1, 1983

Mr. Oscar Simpson New Mexico Oil & Gas Commission P. O. Box 2088 Santa Fe, New Mexico 87501

RE: FLORA VISTA WATER USERS ASSOCIATION WATER WELLS

Dear Mr. Simpson:

In accordance with Mr. Richard Thurstonson's request, we are submitting herewith the following information regarding the above referenced project:

- 1. One copy each of the lab analysis of various water samples obtained at the Flora Vista well field.
- A copy of the driller's log of Well No. 3 (logs of Well No. 1 and No. 2 are similar to the driller's log of Well No. 3).
- 3. One copy each of all correspondence filed in our office since the wells were determined to be contaminated with hydrocarbons.

We hope this information will be helpful.

Sincerely yours,

LAWRENCE A. BREWER & ASSOCIATES, INC.

Dorge a. Madud

George A. Madrid, E.I.T.

GAM/tlc F-331

cc: Mr. Frank Chavez Mr. Richard Thurstonson

visite welle set of unalytes melytes, visite products,

CDS LABORATORIES

A DIVISION OF CASA DEL SOL, INC. 1474 MAIN AVENUE #131 POST OFFICE BOX 2605 DURANGO, COLORADO 81301

(303) 247-4220

F-331

ESUS

Date: 8-1-83 To: Flora Vista Water User Attn. Richard Cheyney 909 W. Apache Farmington, N.M. 87401 CDS Lab ID # Sample Description: _L.A.B. AUG 23 '83

* Analytical results Analyte Sample ID 2456 GALLON Jug 0.70 mg Total Hydrocarbons C₆-C₂₂ 2359 OLD PIT 7.50 mg Total Hydrocarbons  $C_{6} - C_{22}$ 2454 ENST OF PIT 8.10 mg Total Hydrocarbons  $C_{6} - C_{22}$ 2455 Seat TH OF P.T 4.00 mg Total Hydrocarbons  $C_{6} - C_{22}$ * TOTAL HYDROCARBONS IN

Normal detection Limit 0.01

CONCENTE	ATTON BASE	ED ON M	·g I		
2359	21.43	MylL	٥F	HLO	Rec'o
2456	1.99	Mg[L	n	15	••
2455	11.41	MIL	1	1 1	••
2454	23.14	mall	<b>, i</b>	• •	

SEE AMMENDED REPORT 8/24/83

Dr. Joe Bowden

Director

Members of: AMERICAN ASSOCIATION FOR THE ADVANCEMENT OF SCIENCE AMERICAN SOCIETY OF BIOLOGICAL CHEMISTS AMERICAN SCIENTIFIC AFFILIATION SIGMA XI

350 HL SAMPLE

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Date Received: 6/1/83 For: Richard Cheyney	Date Results Sent To:		Date Billed:	<u></u>
ForRichard Cheyney	To:		To:	
<u> </u>				
	·····	· · · · · · · · · · · · · · · · · · ·		- <u></u>
Sample Info: Whore				
Sample Info: Where	ntainer			
	eservative		Amount ş Paid	/ /
	cocivacive			//
FIELD INFORM	ATION/DATA	PI	HYSICAL PARAMETERS	
Sample Description:	(Gallon Jug)	Acidity		mg/L
Sample Suberlpelou.		Alkalinity		mg/L mg/L
Date Taken:	Time:	Color		
Date Received in Lab	:	Conductivity	v at 250	umhos/cm
Date Completed:		Dissolved O		mg/L
QA Check:		Hardness (Ca	aCO3)	mg/L
		рН	J	units
Alkalinity	mg/L	Specific Gra	avity	mg/L
Conductivity at 250	umhos/cm	Temperature		°Č
Dissolved Oxygen	mg/L	Total Combus		mg/L
pH	units	Total Disso		mg/L
Temperature	oc	Total Solids		mg/L
Flow		Total Susper		mg/L
		Turbidity (a	as FTU)	mg/L
Total Dissolved mg/L Aluminum Antimony Arsenic Barium Bēryllium Boron Cadmium Calcium Chromium: Total +3 Form +4 Form Cobalt Copper Iron Lead	COD Chloride Chlorine Demand Coliform Cyanide Fluoride MBAS Nitrogen: Ammonia Nitrate Nitrate Nitrate Total		Gross Alpha Gross Beta Radium 226 Radium 228 	pCi/L
Magnesium	Phenols			
Manganese	Phosphate			
Mercury Molybdenum	Silica Sulfate		·	
Nickel	Sulfide	<del></del>		
Phosphorus	Suilide			
Potassium	Oil & grease	7.1 mg/L		
Selenium	$C_6 - C_{22}$	2  mg/L		
Silver	-6 -22	- , <i>-</i>		
Sodium				
Thallium		1	THIS LABORATORY REPORT ED OR USED FOR ADVERTIS WITH ADVERTISING OF ANY	NG OR IN CONNECTION
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Vanadium	Sum Ka	~		
Zinc	Dr. Joe Bowden	Director	Members of: AMERICAN ASSOCIATION FOR THE	ADVANCEMENT OF SCIENCE

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AMERICAN SOCIETY OF BOLOGICAL CHEMISTS AMERICAN SOCIETY OF BOLOGICAL CHEMISTS AMERICAN SCIENTIFIC AFFILIATION SIGMA XI

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CD9 LABORATORIES	INTERNAL L	AB REPORT	E.M. 14 123 CDS_ID	# <u>2359</u>
Date Received: 5/25/83	Date Results Sent		Date Billed:	=-331)
For: Richard Cheyney	To:		To:	
				<u></u>
Attn. George Madrid Sample Info. Where				
		······································	Invoice #	
	tainer		Amount Ş	/
HowPres			Faid	///
FIELD INFORMA	TION/DATA	PH	HYSICAL PARAMETERS	
Sample Description: F	rom backhoe nit	Acidity		mg/L
dug 5/23/83 in old		Alkalinity	<del></del>	mg/L
Date Taken:	Time:	Color		
Date Received in Lab:		Conductivity	v at 250	umhos/cm
Date Completed:		Dissolved Ox		mg/L
QA Check:	······································	Hardness (Ca		mg/L
<pre></pre>	<u> </u>	pH		units
Alkalinity -	mg/L	Specific Gra	wity	mg/L
Conductivity at 25°	umhos/cm	Temperature		—  c ^{, -}
	mg/L	Total Combus	stables	mg/L
	units	Total. Dissol		mg/L
Temperature	°C	Total Solids		mg/L
Flow -		Total Susper		mg/L
		Turbidity (a		mg/L
TRACE METALS Total Dissolved mg/L Aluminum Antimony Arsenic	CHEMICAL PARAMET Bicarbonate BOD Carbonate Carbon Dioxide	ERS mg/L	RADIOMETRIC ANALY Gross Alpha Gross Beta Radium 226 Radium 228	
Barium	COD			
Bēryllium	Chloride			
Boron	Chlorine Demand	,		
Cadmium	Coliform		PESTICIDES	pCi/L
Calcium	Cyanide			
Chromium:	Fluoride			
Total	MBAS			
+3 Form	Nitrogen:			
+4 Form	Ammonia			
Cobalt	Nitrate		HERBICIDES	pCi/L
Copper	Nitrate/Nitri	.te		
Iron	Nitrite			
Lead	Total			
Magnesium	Phenols			
Manganese	Phosphate			- 4 -
Mercury	Silica		Oil & grease fro	om filtrate
Molybdenum	Sulfate	<u></u>	only.	-
Nickel	Sulfide		C6-C22 from extr	acts of
Phosphorus			filtrate	
Potassium	Oil & grease	<pre></pre>		
Selenium	C ₆ -C ₂₂	7.5 mg/L o	$FH_2O$	
Silver Sodium				
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Vanadium	Olisan Keny	la -		
Zinc	Dr. Joe Bowden	Director	Members of:	
	Jr. Joe bowden	DILECTOR	AMERICAN ASSOCIATION FOR THE AD AMERICAN SOCIETY OF BIOLOGICAL ( AMERICAN SCIENTIFIC AFFILIATION SIGMA XI	IVANCEMENT OF SCIENCIE DHEMISTS

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CDS LABORATORIES	INTERNAL L	AB REPORT	CDS I	ID#_2454
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Sample Info: Where			Invoice #	
When Con	tainer		Amount \$	/ / /
How Pres	servative		Paid	/ /
			WCTCAL DADANCEPDO	
FIELD INFORMA			HYSICAL PARAMETERS	_
Sample Description: E		Acidity		mg/L
Flora Vista Date Taken: Date Received in Lab:		Alkalinity	<u></u> _,,	mg/L
Date Taken:	Time:	Color		<u> </u>
Date Received in LaD:		Conductivity	7 at 250	umhos/cm
Date Completed:		Dissolved Ox	kygen	mg/L
QA Check:		Hardness (Ca	acu3)	mg/L
		pH		units
Alkalinity	mg/L	Specific Gra	avity	mg/L
Conductivity at 25°	umhos/cm	Temperature		°č
Dissolved Oxygen	mg/L	Total Combus		mg/L
рн	units o _C	Total Dissol	lved Solids	mg/L
Temperature	0 ^C	Total Solids		mg/L
Flow		Total Susper		mg/L
		Turbidity (a	as FTU)	mg/L
Total Dissolved mg/L Aluminum Antimony Arsenic Barium Beryllium Boron Cadmium Calcium Chromium: Total +3 Form +4 Form Cobalt Copper Iron Lead Magnesium Manganese Mercury Molybdenum Nickel	Bicarbonate BOD Carbonate Carbon Dioxide COD Chloride Chlorine Demand Coliform Cyanide Fluoride MBAS Nitrogen: Ammonia Nitrate Nitrate/Nitri Nitrite Total Phenols Phosphate Silica Sulfate Sulfide		Gross Alpha Gross Beta Radium 226 Radium 228 PESTICIDES HERBICIDES	pCi/L
Phosphorus				
Potassium	Oil & grease	40 mg/L		
Selenium	C ₆ -C ₂₂	23. mg/L		
Silver				
Sodium			THIS LABORATORY REPORT	MAY NOT BE PUBLISH
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Tin		A. i.	WRITTEN PERMISSION FRO RESULTS ARE BASED ON A	M CDS LABORATORIES
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Vanadium				
Zinc	Dr. Joe Bowden	Director	Members of:	ADVANCEMENT OF SCIENCE

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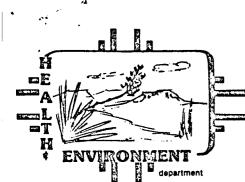
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ate Received: <u>6/1/8</u>	Date Results S	ent:	Date Billed:	
or:Richard Cheyne	ey To:		To:	
ample Info: Where			Invoice #	
When	Container		Amount \$	
How	Preservative		Paid	/
FIELD INF	ORMATION/DATA	P	HYSICAL PARAMETERS	
Sample Description	n:	r pitAcidity		mg/L
Flora Vista	Time:	Alkalinity		mg/L
Date Taken:	Time:	Color		
Date Received in I	Lau:	Conductivit		umhos/c
Date Completed:	· · · · · · · · · · · · · · · · · · ·	Dissolved O	xygen	mg/L
QA Check:		Hardness (C	aCO3)	mg/L
		pН		units
Alkalinity	mg/L	Specific Gr	avity	mg/L
Conductivity at 2	50 umhos/cm	Temperature		°C
Dissolved Oxygen	mg/L	Total Combu	stables	mg/L
pH	units	Total-Disso	lved Solids	mg/L
Temperature	oC	Total Solid		mg/L
Flow		Total Suspe	nded Solids	
	<u> </u>	Turbidity (		mg/L
Aluminum Antimony Arsenic Barium	BOD Carbonate Carbon Dioxi COD	de	Gross Beta Radium 226 Radium 228	
Beryllium	Chloride			
Boron	Chlorine Dem	and	55.077.075.50	
Cadmium	Coliform		PESTICIDES	pCi/L
Calcium	Cyanide			
Chromium:	Fluoride			
Total +3 Form	MBAS	<u></u>		
+3 Form	Nitrogen:			
Cobalt	Ammonia Nitrate		HEBRICIDES	pCi/L
Copper		tuito	HERBICIDES	per/r
Iron	Nitrate/Ni	LIILE	<b></b>	
Lead	Nitrite			
Magnesium	Total Phenols	·····	<del></del>	
Manganese			<del></del>	<u> </u>
Manganese	Phosphate Silica			
	Silica Sulfate			
	Surrace			
Molybdenum	C1 f: J_			
Molybdenum Nickel	Sulfide			
Molybdenum Nickel Phosphorus				
Molybdenum Nickel Phosphorus Potassium	0il & grease	< 3 mg/L		
Molybdenum Nickel Phosphorus Potassium Selenium	0il & grease	< 3 mg/L 11.4		
Molybdenum Nickel Phosphorus Potassium Selenium Silver		• -		
Molybdenum Nickel Phosphorus Potassium Selenium Silver Sodium	0il & grease	• -	THIS LABORATORY REPOR ED OR USED FOR ADVERT	ISING OR IN CONNECTIC
Molybdenum Nickel Phosphorus Potassium Selenium Silver Sodium Thallium	0il & grease	• -	ED OR USED FOR ADVERT	ISING OR IN CONNECTIO
Molybdenum Nickel Phosphorus Potassium Selenium Silver Sodium Thallium Tin	0il & grease	• -	ED OR USED FOR ADVERT WITH ADVERTISING OF AU WRITTEN PERMISSION FR RESULTS ARE BASED ON	ISING OR IN CONNECTIO VY KIND WITHOUT PRIO OM CDS LABORATORIE: ANALYSIS MADE AT TH
Molybdenum Nickel Phosphorus Potassium Selenium Silver Sodium Thallium	0il & grease	• -	ED OR USED FOR ADVERTI WITH ADVERTISING OF AN WRITTEN PERMISSION FR	ISING OR IN CONNECTIO VY KIND WITHOUT PRIO OM CDS LABORATORIE: ANALYSIS MADE AT TH

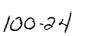
AMERICAN ASSOCIATION FOR THE ADVANCEMENT OF SCIEN AMERICAN SOCIETY OF BIOLOGICAL CHEMISTS AMERICAN SCIENTIFIC AFFILIATION SIGMA XI

•		E G		Revised June 1972
			STATE ENGINEER OFFICE	
			WELL RECORD	
		S.	ection 1. GENERAL INFORMATION	
				•
			Nater Users Association Ow	
City and Street or	State <u>Flor</u>	A Vista,	N.M. 87415	
		-	-1 Exploratory and is located in the:	
			% of Section 23 Township 30-N 1	Range 12 WN.M.P.M.
b. Tract	No	of Map No	of the	
			of the County.	
			feet, N.M. Coordinate System	Zone in Grant.
(B) Dialling (	'ontractor	BOB SA	VAGCLicense No.	WD-847
Address	PO. Box	2434	FARMINGTON, NM. 87401	
Dulling Began	12-7-	81_ Completed	12-9-81 Type wools CAble To	Size of hole 10 in.
Lievation of fa	nd surface or		at well is ft. Total de	pth of well 25 ft.
Completed wel	tus 🕅 sh	allow 🗀 artesi	an. Depth to water upon complet	tion of well ft.
n - 1		Section	2. PRINCIPAL WATER-BEARING STRATA	
Depth From	m Feet	Thickness in Feet	Description of Water-Bearing Formation	Estimated Yield (gallons per minute)
6	25	19	Boulders + SANd	/00
•				

			Section 3, RECORD	OF CASING		
I	 Pounds	Threads	Depth in Feet	Length	Type of Shoe	Perforations

a set a sub-			Section 6, LOG OF HOLD
Depthy	in Feet	Thickness in Feel	
l'gim	<u> </u>	in beet V	Color and Type of Manage countered
0	25.	25	Boulders & SAND
	<u> </u>		Doulders & JANd
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### STATE OF NEW MEMICO

ENVIRONMENTAL IMPROVEMENT DIVISION DISTRICT I FIELD OFFICE/724 WEST ANIMAS FARMINGTON, NEW MEXICO 87401 Russell F. Rhoades, MPH, Director TONEY ANAYA GOVERNOR

ROBERT McNEILL SECRETARY

ROBERT L. LOVATO, M.A.P A. DEPUTY SECRETARY

> JOSEPH F. JOHNSON DEPUTY SECRETARY

> > 5-4

Telephone #(505)327-9851

April 13, 1983

REGULATION SECTION

Mr. Bert Barnes, President (Flora Vista Water Users Association P.O. Box 171 Flora Vista, New Mexico 87415

Dear Mr. Barnes:

The annual environmental survey of the Flora Vista Water Supply System was conducted March 24, 1983. The survey report is attached for your review and file.

The system supplies water to approximately 1500 consumers. Since February, 1983, the system has been purchasing all of its water from the City of Aztec (WSS #098-24). Prior to February, roughly half of the system's water was from two of three wells drilled by the association in 1980. In February, one of the wells became contaminated with oil and grease, presumably from a newly drilled gas well 300 feet upstream. The wells were taken out of service at that time, and the affected lines of the system were flushed to remove the contamination. The wells will not be placed back into service unless the source of contamination is located and eliminated, and the contaminated aquifer cleaned up. According to Richard Cheney of Brewer and Associates, Inc., the outlook for future use of the wells is bleak.

The association's contract with Aztec limits the purchase of water to 36 million gallons per year. This limitation may result in periods of low pressure during the summer when water demand is the highest. Negotiations to purchase water from the City of Farmington are currently under way.

The only deficiency noted during the inspection was that records of maintenance performed are not being kept. Proper maintenance records would help in identifying problem areas of the system. Mr. Bert Barnes April 13, 1983 Page 2

I wish to thank Mr. Ray Penrod for his time and cooperation in conducting the survey. I would like to encourage the association to have Mr. Penrod attend the Northwest New Mexico Water and Wastewater Operator Short School scheduled for May 9-12, 1983, in Farmington. More information is available from this office. If I can be of any service to you, feel free to contact me.

Sincerely, David A. Tomko

David A. Tomko, Environmentalist III

57.4

DAT:1m cc: Water Supply Section File

Community Water Supply System

N/A	- 🕶	Not	Applicable
N-Av	-	Not	Available

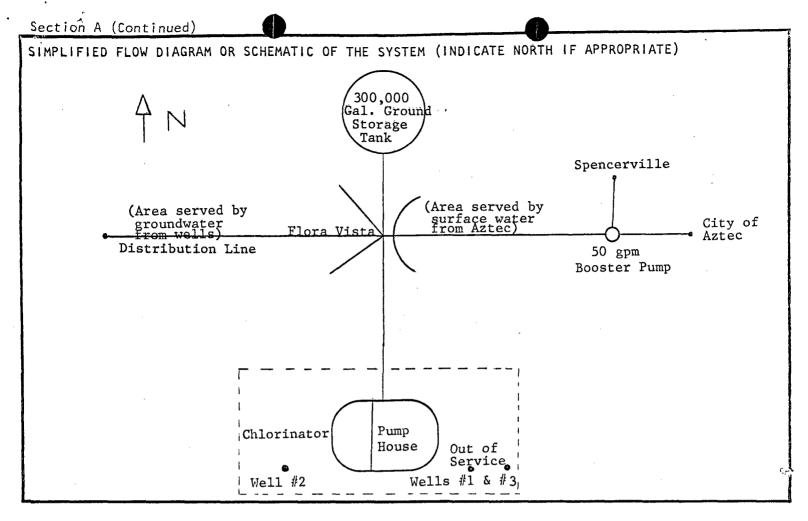
Est. - Estimated



Inspection Form

<b>.</b> . <b>.</b> -		• ENVIR						
Section A - G	ENERAL INFORMATIO	rironment Department provement Division	Inspection Date: 03-24-83					
WSS CODE:	WATER SUPPLY	COUNTY:						
100-24	SYSTEM NAME Flora Vista Water Users Association San Juan							
				Bail Stall				
System Addres		Box 171, Flora	Vista, New Mexico 87415					
OWNER:		فالتشريب والمستجمع والمتحاج ويتبر والمتحج ويتبارك والمستحد والمراجع والمراجع	(if different than above)	PHONE				
Flora Vista V	Vater Users Assoc			334-6045				
Population	#	# Meters	Max. System Production	Average System Produc-				
Served	Connections		<u>N-Av</u> GPD	tion				
1500 Est.	431	431	Poten. 🗌 Actual 🗌 '	<u>100,000</u> GPD (January 1983)				
System Source		ution Only X	Well(s)	# of Wells _3				
(check Approp			The second se	Not in use due				
	Spring(	s)Infiltr	ation Gallery 🔲 Surfa					
				<u>contamination</u>				
	Qualifying Infor							
<u>In 1981, the</u>	<u>water system pla</u>	ced into service	e 2 wells with a pump con	trol house and				
chlorination	facilities (see )	March 23, 1982	survey). In February, 198	3. the wells became				
contaminated	with oil and gre	ase, presumably	from a near by drilled ga	s well 300 feet upstream.				
The wells wer	e shut down at t	hat time, and th	ne affected lines of the s	<u>ystem were flushed. Due</u>				
to the contam	ination, the well	ls cannot be pla	aced back into service unl	ess the source of				
contamination	is verified and	eliminated and	the contaminated aquifer	cleaned up, which might				
	veral years, if e							
Currently, th	e systems only so	ource of water i	is purchased from the City	of Aztec and is limited				
to 36 million	gallons per year	. Negotiations	are currently under way	to purchase finished				
water from Fa	rmington, Purcha	asing water from	Farmington would require	the construction of				
	of pipeline to a	•	ems.					
······································								
			Level of Certifica	Level of Certification				
System Person	nel – Name/Classi	fication	tion Required	Obtained				
Ray Penrod /	Maintenance Super	rvísor	None Destate	TINODE				
			MAY 0 2179					
			WATER SUP REGULATION SE					
In Company to the C								
Information f	•	Donmad		Business Phone:				
<u> </u>		Penrod		334-6045				
Survey Perfor	-	Tomko, Environ	montolict III	Business Phone:				
	Daviu A.	TONKO, ENVIRON		327-9851				

EID - 042 Revised 11/82



#### Section B - Source Information

WELL Identification	Well Depth	Pump Depth	Pump Capacity GPM	Well Drawdown feet	Pump Type	Static Water Level	Age of Pump	Date Well Drilled
Well #1	25 '	18'	60-70	5'-6'	Submers.	= 8'	2 yrs.	1980
Well #2	25'	20 ¹ ₂ '	60-70	5'-6'	Submers.	≥ <u>8'</u>	2 yrs.	1980
Well_#3	25'	N/A	N/A	N/A	No Pump	<u>≃_8'</u>	N/A	1980

Remarks, Deficiencies, and Recommendations

All wells shut down due to oil and grease contamination. Contamination found in well #1. Well #2 has not been contaminated, but has been shut down to prevent pulling of contamination through the aquifer. Well #3 was to be used when system expanded and has never been placed into service.

Source	Number (each)	Total Capacity	Remarks, Deficiencies, and Recommendations
Artesian Wells			
Springs			:
Infiltration Galleries			

EID - 042 Revised 11/82

birs

Gravity Storage Reservoir Indent.	Storage Capacity (gallons)	Date Built	Exterior Condition	Tank Material	Cathodic Protection	Tank Dpenings Secured	Overflow Screened	Vent Screen
#1	300,000	1979	Good	Steel	No	Yes	Yes	Yes
Pressure Tank I.D.	Volume (gallons)		Exterior Condition	Additional Information				

Remarks, Deficiencies or Recommendations

The storage tank was drained when the oil and grease contamination occurred. The tank has not been placed back into service yet due to insufficient pressure from having entire system on Aztec water.

Section D - WATER DISTRIBUTION

Booster	Total Nu		r Remarks, Deficiencies or Recommendations					
Pump Stations	of Stati	j 1	Rarely, if ever, used. 50 gpm in-line centrifugal pump.					
Type of Disinfection Facilities Gas C			Chlorine		Remarks, etc. Gas chlorination provided at wells; ' Purchased water disinfected by Aztec			
System Pressure	Max. <u>90</u> psi Min. <u>55</u> psi		4	arks, et tations	c. May be pressure problems in summer due to on amount of water purchased from Aztec.			
Pipe Materials in system		STEEL [ Other [	_ Chec		es (including cross-connections observed) s installed at every metered connection. stos			
Type of Distribution System			Grav	Gravity X Pressure X When wells th				
Contaminant	Next S ple Da		mple equency		Remarks			
Micro- biological		1	2					
Turbidity		$\Box$	N/A					
Organics			N/A					
Inorganic	07/84	3	years	Only	applies if present wells will be placed back into			
Radiological	03/86	4	years	serv	ice. Otherwise the Aztec Water Supply System			
Secondary					· · · · · · · · · · · · · · · · · · ·			
Section E -	GENERAL O	PERATIO	ON AND M	AINTENAN	NCE (0 & M)			
Does the sys	tem keep	up-to-	date O&M	records	? Yes X (operation) No X (Maintenance)			
Are proventative maintenance activities routinely practiced? Yes 🔬 No 🗋								
					was discussed with the Operator Yes 🔀 No 🗌 (Pipe Materials in System)			

## Section F - SURFACE WATER SUPPLY INFORMATION

	N/A
Source (check one)	Remarks or Deficiencies
Lake	
Stream	

## Section G - WATER TREATMENT PLANT INFORMATION N/A

Unit Operations	. Remarks or Deficiencies	
Plant Intake Structure		
Plant Location (Siting)		
Pretreatment, Raw Water Storage or Presettling Reservoirs		8
Coagulation - Sedimentation		
Chemical Addition		
Filtration		
Other Treatment (Ion Ex- change, Softening,Reverse Osmosis, etc.)		
Additional Comments or Rem	arks	
		f

Page 4

MONTGOMERY & ANDREW PROFESSIONAL ASSOCIATION ATTORNEYS AND COUNSELORS AT LAW

J. O. Som (1883-1963) Frank Andrews (1914-1981)

A & Montgomery Secto D. Mountermany Frank Andrews # Victor & Ortona John E. Conwily Jack M. Morgan Jultray A. Brannen John B. Pound Harry R. Kilpartele. Humas W. Ohum William C. Madican Walter J. Malandras Bruce C. Huir Methad W. Brennan Hobort P 'Worcester John B. Draper

Name M. Anderson Hunterpla H. S.a. Kis, Jr H. Human Datey Janot McL. McKay Edward F. Mitchell III Carrie E. Parker Maaroon A. Sandors Mark F. Sheradan drught for tanned Phyllin A flore Wm Abin Wright Itini V. Greyeli Conduce M. Witt Wesley B. Howard, Jr Thurman W. Moore III Jack L. Former

REPLY TO FARMINGTON OFFICE

Mr. Ed Hartman Manana Gas, Inc. 610 Reilly Rieghts Farmington, NM 87401 March 4, 1983

WATER SUPPLY WATER SUPPLY FARMINGTON OFFICE Suile 325 First National Bank of Farmington One First Place Prest Office Ros 2700 Fuenthylion, New Measuri (17492-2700

Leverstanner (1005) 327 5074

ALBUDIN NOUP OFFICE Suite 325 500 Copper Avenue, N.W Pret Other Day 1300 Allespergen, New Marka 07.03-1309

Entry Lummer (505) 242 9677

SANTA FE OFFICE 375 Paseo du Poralta Pust Office Box 2307 Santa Fe, New Mexico 87504-2307

> Telephone (505) 982-3873 Telecopy (505) 982-4289

> > C~--

Re: Flora Vista Water Users Association, Inc.

Dear Mr. Hartman:

We have now received a more detailed report from the engineer concerning the contamination of the flora Vista Water Users Association's wells. As you will note in the engineer's report, a copy of which is enclosed for your reference, the problem appears at this time to be quite significant and potentially quite costly. Under these circumstances, I would once again recommend that you contact your liability insurance carrier, it you have not already done so, so that we can work toward a prompt resolution of the problem.

If you wish to allempt to cooperate in further investigation to minimize the effects of the contamination, please let me know at once. Since it was necessary to close down the wells and the effects are already apparent, your prompt attention will be appreciated.

Very truly yours,

ORIGINAL SIGNED BY R. THOMAS DAILEY R. Thomas Dailey

RID/pe toc. ce: Tlora Vista Water Duers Assa.



## LAWRENCE A. BREWER & ASSOCIATES, INC.

Lawrence A. Brewer, P.E., L.S. Richard P. Cheney, P.E., L.S. George T. Walters, L.S. Robert A. Echols, Jr., P.E.

31-15

CONSULTING ENGINEERS

March 2, 1983

Mr. Thomas R. Dailey, Attorney P.O. Box 2700 Farmington, New Mexico 87499

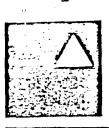
Re: Flora Vista Water Users Association Water Wells

Dear Tom:

As you are aware, the No. 1 and No. 3 wells for the above referenced system have been infiltrated with oil and grease contaminates and have been taken out of service. Samples taken directly from Well No. 1 indicate that the level of oil and grease in the water is 16 milligrams per liter. This seems to be sufficient to develop an oil film on the surface of the water and, in addition, has created a very objectionable odor. In an attempt to determine the source of the contamination, we have conducted an on-site visual survey of the property. As you are probably aware, a gas well has been drilled less than 300 feet upstream from the No. 1 and 3 water wells. We dug a backhoe pit between Well No. 1 and the gas well as far as possible below the groundwater table. As the water began to filter into the pit, a noticeable odor was detected and an oily film appeared on the surface of the water. At this time, it appears that the contamination is probably coming from the gas well. This could be from either one or two sources. It could be coming from an unlined slush pit or the well could have a leak in the casing that is producing the contamination. I have advised the Association to discontinue use of the well field because I believe that continued pumping could pull the contaminates throughout the entire water bearing strata. At this time, I am of the opinion that the entire site has been contaminated and can no longer be used for the production of domestic water. Of course, further investigation will be needed to confirm this opinion. If, in fact, the site has been lost as a water producing area, the monetary damage to the Association will be extensive. The wells cost approximately \$5,000.00 a piece to drill and equip. There Page 2. Mr. Thomas R. Dailey March 2, 1983

is also a well house with an electronic monitoring system installed adjacent to the well field that cost approximately \$31,000.00 to construct. There is also approximately \$12,000.00 worth of control cable and six-inch waterline that will be a total loss. In addition to these improvements, the Association paid \$20,000.00 for the land. The Association can purchase a limited amount of water from the City of Aztec. This supply is not sufficient to furnish the entire Association with water in the summer months. It also costs considerably more to purchase than it did to produce the same amount of water from the wells. All of the previously mentioned billing and equipment items were installed utilizing funds obtained from a Farmers Home Administration loan. This loan must be repaid. The loss of the facilities will create a severe economic strain on the Association. It is my opinion that the Farmers Home Administration, The New Mexico Environmental Improvment Division, and the Oil Conservation Division of the Energy and Minerals Department should be made aware of the problems. The Association will need assistance to determine the actual source and extent of pollution. If I can be of additional assistance, please feel free to contact me.

Sincerely yours, LAWRENCE/A. BREWER & ASSOCIATES, INC. Richard P. Chéney, Vice President F-152-D, F-152-E RPC:ps



# LAWRENCE A. BREWER & ASSOCIATES, INC.

Lawrence A. Brewer, P.E., L.S. Richard P. Cheney, P.E., L.S. George T. Walters, L.S. Robert A. Echols, Jr., P.E.

F 77-1525

CONSULTING ENGINEERS

March 2, 1983

Mr. Thomas R. Dailey, Attorney P.O. Box 2700 Farmington, New Mexico 87499

Re: Flora Vista Water Users Association Water Wells

Dear Tom:

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Sincerely yours, LAWRENCE/A. BREWER & ASSOCIATES, INC. Richard P. Chénéy, Vice President ₩-152-D, ₩-152-E RPC:ps

FLORA VISTA WATER USERS ASSOCIATION, INC. Post Office Box 171 Flora Vista, New Mexico 87415

March 1983

Last year we reported to you that we would be supplying most of our water needs from our own wells. This was accomplished but due to circumstances beyond our control we have discontinued pumping these wells. Copies of letters from our engineer and attorney are enclosed for your information.

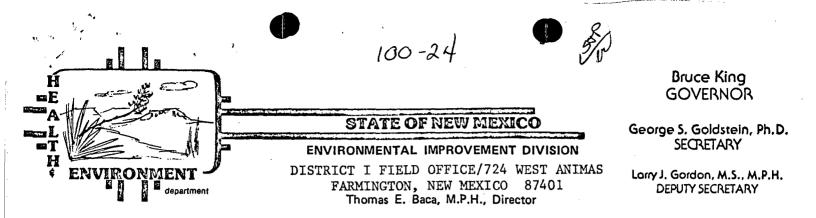
Due to the above and other problems in getting information from Farmers Home Administration, we are asking for another chance in our by-laws. We wish to change our annual meeting to the fourth Monday in April of each year. Hopefully this will provide ample time to prepare annual reports with varified information. Your support in this action will be appreciated.

We wish to thank Richard "Dick" Thurstonson for his services as director and president during the past three years. H. Kyne White has been a great service to our Association by his interest and loyal attendance at our board meetings and representing all of us at the County Rural Water Users Association meetings. Thanks to both Dick and Kyne for jobs well done. Both gentlemen have agreed to run for re-election for another term.

In accordnace with Article V, Section 3 of our Association by-laws this is to notify you of our annual meeting to be held at 7:00 PM at the new fire station, Flora Vista on Monday April 25. In the event you cannot attend this meeting, please exercise your right to vote by marking your ballot and returning it to us in the stamped envelope enclosed. PLEASE RETURN EALLOT AT YOUR EARLIEST CONVENIENCE.

BEST WISHES TO EACH OF YOU AND YOURS FOR THE BALANCE OF 1983.

BOARD OF DIRECTORS Flora Vists Water Users Association, Inc.



Telephone #(505)327-9851

April 16, 1982

Mr. Dick Thurstonson, President Flora Vista Water Users Association P.O. Box 171 Flora Vista, New Mexico 87415

Dear Mr. Thurstonson:

The annual environmental survey of the Flora Vista Water Supply System was conducted March 23, 1982, with the aid of Mr. Ray Penrod. Enclosed is a copy of the survey report.

The system is supplying water to approximately 1500 consumers. The water is obtained from two sources. Consumers roughly east of the Flora Vista Post Office receive water purchased from the City of Aztec (WSS #098-24). Approximately 1.5 million gallons are purchased monthly. The consumers west of the Post Office receive water pumped from two wells that is then chlorinated and either delivered to the consumers or stored in a 300,000 gallon ground storage tank. The well supply system has been in operation for approximately one year.

The following deficiencies were observed during the inspection:

- Employee safety devices are needed at the chlorine room. The switch to the ventilation fan should be mounted outside the door to the chlorine room to allow venting of any chlorine gas from the room prior to entry. The employees should be provided with a self-contained breathing apparatus that would allow safe entry into a chlorine atmosphere to repair a leaking chlorine cylinder.
- 2. The control panel for the pumps is not functioning properly. The pumps fail to restart after being shut off due to electric failures, lightning, etc. Also, the low/high level controls for the storage tank are not functioning.
- 3. The storage tank lid should be kept locked at all times to prevent unauthorized entry to the tank.

Mr. Dick Thurstonson April 16, 1982 Page 2

- 4. The storage tank overflow should be screened to prevent entry of insects and small animals that may contaminate the water.
- 5. Records of maintenance performed need to be kept.
- 6. A sample for radiological parameters was collected the day of the inspection. The results are forthcoming.

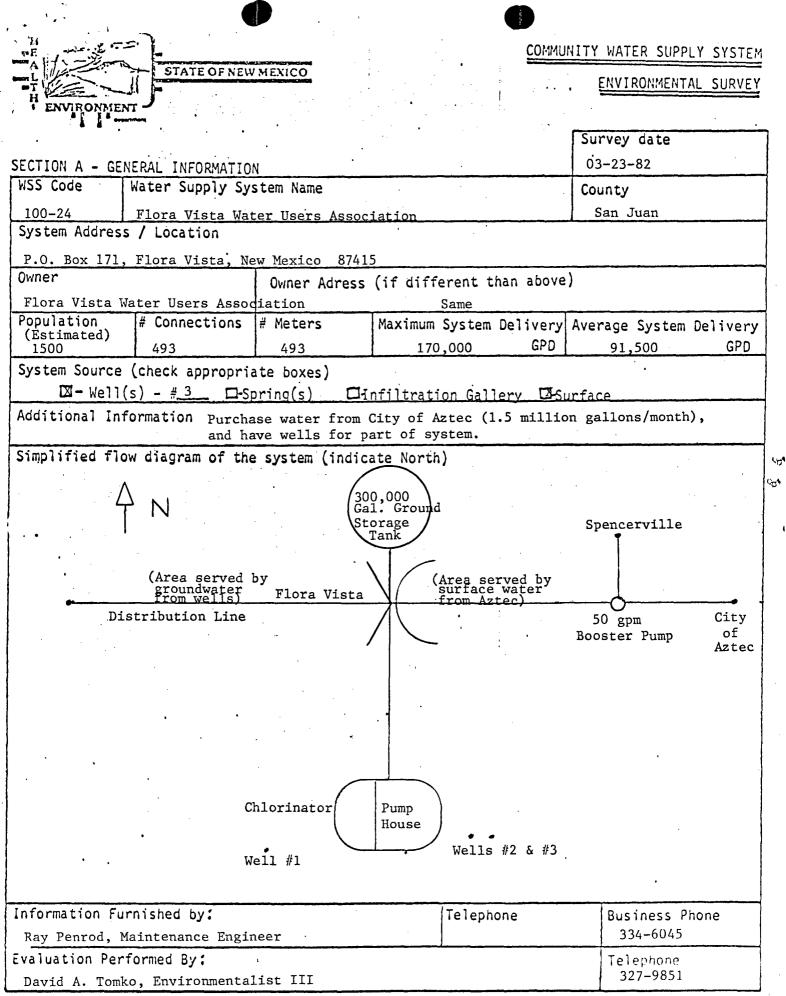
I wish to thank Mr. Ray Penrod for his time and cooperation in conducting the survey. I would like to encourage Mr. Penrod to attend the Northwest New Mexico Water and Wastewater Operators Short School scheduled to be held May 10 - 14, 1982 in Farmington. Information regarding the school can be obtained from this office or from Cas Ruybalid of Bloomfield Water and Wastewater. If I can be of service to you, feel free to contact me.

Sincerely,

David A. Torsko

David A. Tomko, Environmentalist III

Enclosure DAT:1m cc: File VWater Supply Section, EID



, , , , , , , , , , , , , , , , , , ,			. '					
Well Lieftification	Depth	Pump Depth	Capacity GPM	Well Drawdown	Pump Type	Static Water level	Meets Sanitary Ruguir, 7	Nge Of Well
#1	25'	20 ¹ 2'	60-70 gpm	5'-6'	Submersibl 7½ np	e ≃8'	Yes	1 yr.
_ #2	25'	18'	60-70 gpm	5'-6'	Submersibl 7½ hp	e ≃8'	Yes	1 yr.
#3	÷25'					~8'		1 yr.
Spring	Adequatel	y Protecte	ed? D-Yes			- <u></u> .	<u> </u>	
Remarks or De	ficiencies						······	
No Pump at W	lel1 ∦3		• •		1			
			,		r s			
				,		,		

SECTION C - FINISHED WATER MANAGEMENT

Finished Water Boostin		I Remarks or Deficiencies
or Reducing Stations	<b>F</b>	
		Rarely Used
Total Pumping	Amount	· ·
Capacity .	50 gpm	In-line centrifugal pump
<u>Disinfection</u> - Type	Gas	Safety measures needed; self-contained breathing apparatus and switch to chlorine room fan should be located outside of the chlorine room
Amount Used /24 hrs	10 lbs/day	-
Average Daily Residual	≃1.0 ppm	· .
Average Pressure	65 psi	
Cross-connection Control Program?	□ - Yes ⊠ - No	Check valves installed at every metered connection
Pipe Material Used	PVC - Cement Asbestos	2" - 6"
Type of Distribution System	Pressure and Gravity	
<u>Microbiological</u>	No. of Samples p 2	er month
Compliance Status	1 non-sampling in	1981
<u>Chemical Quality</u> Organic	Last Sample Date	
Inorganic	07-06-81 In Compliance	
Radiological	03-23-82 No results yet	Also see City of Aztec Water Supply System for analyses (098-24)
Secondary		

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Tank Identificatio	n Capacity	Age	Condition	Construct. Materiais	Cathodic Protection	Tank Locked?	Overflow Screened?
#1	300,000	<u>1 yr.</u>	Excellent	Steel	No	No	No
<u> </u>	•						
·	•			1			
Pressure Tanks	Capacity	Age	Condition	Operating	and Pressu	ire Range	•
Tank lid not locke SECTION E - SYSTEM		inspect	ion and over	rflow scree	n should be	reattach	ed.
Personnel Classifi	cation	Level	of Certific	a.Required	Level of C Current	Certificat y Obtaine	tion
Maintenance Engine	eer/Operator	None			None		
				: 			
							•. 
			•	,			

SECTION F - NOTATIONS AND RECOMMENDATIONS

 Adequate Operating & Maintenance Records? Adequate operating records; No maintenance records kept
 Adequate Operating & Maintenance Personnel? Yes
 Is Preventative Maintenance Practiced? Yes
 Adequate Disinfection? Yes
 Adequate Storage? Yes
 Adequate Water Quality Data? Yes - radiological samples to be analyzed

**RECOMMENDATIONS:** 

7. Control pgnd/not functioning properly. Pumps are not reset automatically. Also, float switch at storage tank not operating properly.

Flora Vista Water Users Association President: Dick Thurstonson 334-6031 (Home) 334-9060 (Office) Vice Fresident: Bert Barners 325-7577 Grif

SECTION & - SURFACE WATER SUPPLY INFORMATION REFER TO CITY OF AZTEC (098-24) FOR

	Source (check one)	TREATMENT OF SURFACE WATER Remarks or Deficiencies
<i>ſ</i> .	🗋 - Lake	
	🗍 • Stream	
•	Raw Water Storage Capacity	Watershed Pollutant Possible Sources

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# SECTION H - TREATMENT PLANT INFORMATION

Operations	Remarks or Deficiencies
Plant Intake	
Plant Location	
Presettling Reservoirs	
Coagulation-Sedimentat (chemicals used)	ion
• Filtration	
Disinfection	
Other Treatment	,:
Average Daily Turbidit	
Additional Comments or	Remarks •
•	

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# **RANNEY METHOD WESTERN CORPORATION**

WATER SUPPLY ENGINEERS AND CONTRACTORS

RANNEY

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**CONTRACTOR** 

P. O. BOX 6387 Kennewick, Washington 99336 Telephone: (509) 586-6947

RANNEY COLLECTORS SURFACE WATER INTAKES ARTIFICIAL RECHARGING CONSTRUCTION DEWATERING

October 28, 1981

Lawrence A. Brewer & Associates, Inc. P.O. Box 2079 Farmington, New Mexico 87401

Attn: Mr. Richard P. Cheney Vice President

Re: Ranney Collector Water Supply Flora Vista Water Users Association Flora Vista, New Mexico

Gentlemen:

This letter-report sets forth the results of a test drilling program conducted for the Flora Vista Water Users Association in order to determine the feasibility of developing a naturally-filtered water supply of at least 600 gpm by means of a Ranney Collector constructed adjacent to the Animas River.

During the course of this investigation, test drilling was carried out at three sites, designated as Sites A,B, and C and located as shown in Figure SW-99-01. A total of four test holes were drilled at these sites, the written logs of which are attached to this report.

#### Site A:

Test Holes 1 and 2 were located at Site A, which is on property belonging to the Water Users Association. Test Hole 1, located 40 feet from the Animas River and 300 feet downstream from the upstream property line, encountered sand, gravel, silt and clay from ground surface to a depth of 22 feet, where shale bedrock occurred. In general, the material encountered was tight, with cemented layers, and had to be drilled. The only loose material was a sand and gravel layer occurring at the shallow depth of 7 to 10 feet.

Preliminary test pumping on September 5, with the well perforated at a depth of 13 to 18 feet, showed a drawdown of 7.0 feet after 2 hours of pumping at a rate of 75 gpm.

Subsequent pumping on September 6 showed a drawdown of 6.4 feet after 2 hours of pumping at a rate of 90 gpm. The chemical analyses of water samples taken during this period are given as follows:

	~	Parts Per Mi	illion, except pl	H
Constituent	T.H. 1 9/5/81	T.H. 1 9/6/81	Association Wells 9/1/81	Animas River 9/1/81
Hardness	308		360	291
P. Alkalinity	0		0	0
M.O. Alkalinity	170		239	154
Iron	0.15	0.03	trace	trace
Manganese	trace	0.07	0.3	- 0
pH	7.8	7.8	7.8	8.5
Temperature( ^O F)	60	60	62	80

The chemical quality of the water from Test Hole 1 is similar to that of the Association's existing wells and does not contain excessive amounts of iron or manganese.

In order to further explore the water potential of this site, a 6-inch observation well (T.H.2) was drilled 50 feet downstream from Test Hole 1 and a 12-hour pumping test was conducted. Test Hole 1 was pumped from 10:00 a.m. until 10:00 p.m. on September 8, 1981. Water levels were recorded continuously in Test Hole 2 and the abandon Association well, located 130 feet inland from Test Hole 1, with the following results:

	D / 1		-	0-9	Öqpm	$\frac{1}{5}$
	Depti	n to Water (feet)		Q - /	Ogp	νęj –
Lange Time AT	T.H.2	Abandon A Assoc. Well	4.	n²/za	Remark	S
$\begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \begin{array}{c} \\ \end{array} \\ $	5.24 6.25	$\frac{4}{-1} \xrightarrow{\text{Assoc. Well}} 7.09$	05	Static 261,67	water	levels
8 1:00 p.m. /80	6.38 6.51	1,14 20,837.16 0. 127 13,897.19 0.	07 10	140.23 93.69		
6 2:00 p.m. 340 4,2 3:00 p.m. 300	6.65 6.73	1,A) 10,427.29 6, 149, 6,327.35 0,	2026	56.33		
+ 4:00 p.m. 360 3. + 5:00 p.m. 420 3 6:00 p.m. 420	6.86 6.93 6.99	1.62 6947.42 0.	337 40	46.94- 41.24- 35.21		
2,7 7:00 p.m. 540	7.05	1,61 4.63 7.52 0.	43 ,46	31.30		
ネス 9:00 p.m. 660 み,00 10:00 p.m. 726	7.137.18	1.99 279 7.58	47 51	23.47 23.47 End	l test	

The total drawdowns for the 12-hour test were 1.94 feet for Test Hole 2 and 0.51 feet for the abandon Association Well.

These drawdowns are considered to be excessive, indicating a low permeability in the order of magnitude of 750 gallons per day per square foot. More important, however, is the fact that water levels continued to decline throughout the test period and stabilization did not occur. This indicates that no recharge occurs from the Animas River. Apparently, the cemented sand and gravel zones occurring in the upper portion of the formation effectively seal off the Animas River and prevent recharge from the river.

With this lack of recharge or replenishment from the Animas River, it is concluded that a firm, dependable, naturallyfiltered water supply cannot be developed and therefore this site is not considered suitable for the construction of a Ranney Collector.

# Site B:

Test Hole 3 was located at Site B, which is on the Irene Brown property, just upstream from the Flora Vista Bridge. Test Hole 3, located 240 feet upstream from the bridge and 70 feet from the Animas River, showed the existence of sand, gravel, and silt from ground surface to a depth of 24 feet, where impermeable silty sand and clay was encountered. As was the case with Test Hole 1, the material encountered was tight, with cemented layers, and had to be drilled.

However, loose sand and gravel was encountered at depths of 17 to 24 feet. Initial attempts to pump the well with the perforations at a depth of 18 to 23 feet were unsuccessful. A second attempt to pump the well, with the perforations relocated at a depth of 15 to 20 feet, was unsuccessful due to excessive sand clogging the pump and hoses.

Again, it appears that the cemented sand and gravel zones, which extend to a depth of 17 feet, effectively seal off the Animas River and prevent recharge.

## Site C:

Test Hole 4 was located at Site C, which is on the Todd Hickman property about ½ mile upstream from Site A. Test Hole 4, located 30 feet from the Animas River, showed the existence of sand, gravel and silt from ground surface to a depth of 25 feet, where fine silty sand was encountered. As with the previous drilling, the material encountered was tight, with cemented layers, the only loose sand and gravel being a small zone occurring at a depth of 20 to 23 feet. Test pumping, with the well perforated at a depth of 18.5 to 23.5 feet, showed a drawdown of 10.4 feet at a pumping rate of 50 gpm without stabilization of the water levels.

In summary, the results of the test drilling program show that this entire stretch of the Animas River Valley, i.e., Sites A,B, and C, is underlain by alluvial deposits which consist of cemented sand and gravel in the upper portion and only small zones of loose, water-bearing sands and gravels near the base of the formation. These upper cemented sand and gravel zones effectively seal off the Animas River and prevent recharge to the aquifer.

Since river recharge or replenishment does not occur, it is concluded that a firm, dependable, naturally-filtered water supply cannot be developed and it is therefore recommended that no further consideration be given to the development of a Ranney Collector water supply within this stretch of the river valley.

We enclose our Invoice No. 99-01 in the amount of \$16,839.34 covering the total cost of this investigation. If you have any questions or need additional information, please let us know.

Very truly yours,

RANNEY METHOD WESTERN CORPORATION

Frederich C. Michele

Frederick C. Mikels, P.E. President & Chief Engineer

Enclosure

FCM/sk

750 gpl x 52 = 100 ST/Day = K F2 7.48 gel (Do Semilog plot) (Sept. 81 Test)  $\overline{\tau} = \frac{K}{n} (\underline{L}) = 1005T \quad 0.008 = 3.25T/Day$   $\overline{\tau} = \frac{K}{n} (\underline{L}) = 1005T \quad 0.008 = 3.25T/Day$ 116855/ yr. (Sept 20) (Oct 25)  $K = \frac{Q}{\pi(h_2^2 - h_1^2)} \frac{Q_1 n_2}{n_1} \frac{TH_2 \cdot n_2}{A \cdot n_1} = 5057$ Them Eg,  $aTT = 12h_7$ ,  $h_2 = 19 - s_2 = 19 - 1.54 = 17.06$  $h_1 = 19 - s_1 = 19 - 0.51 = 18.45$ Snom. Rinney 1007  $K = \frac{90}{\pi(251.04 - 341.88)} = \frac{90}{\pi(-50.84)}$ -D,96) (For Q=73qpm, K=646qpd/572  $K = T = \frac{Q}{b} + \frac{1}{2\pi b} \left( \frac{50}{5} - \frac{5}{2} \right) + \frac{1}{7} \left( \frac{1}{2\pi b} + \frac{90}{5} + \frac{140}{5} + \frac{140}{5} + \frac{1}{30} + \frac{1}{130} + \frac{1}{130}$ = 725 gpR/5(2



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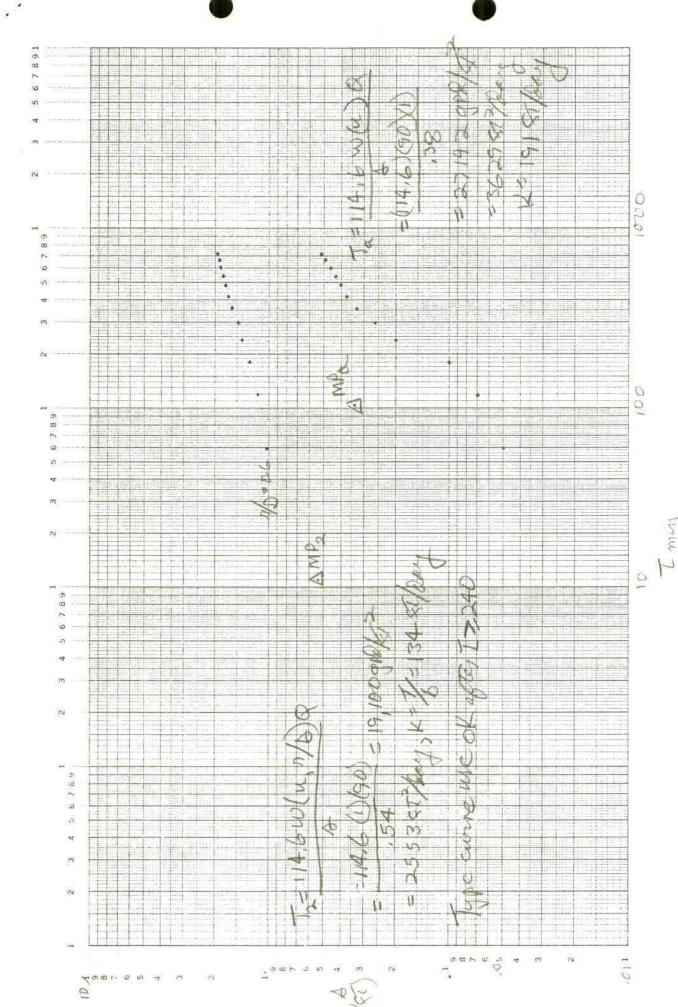
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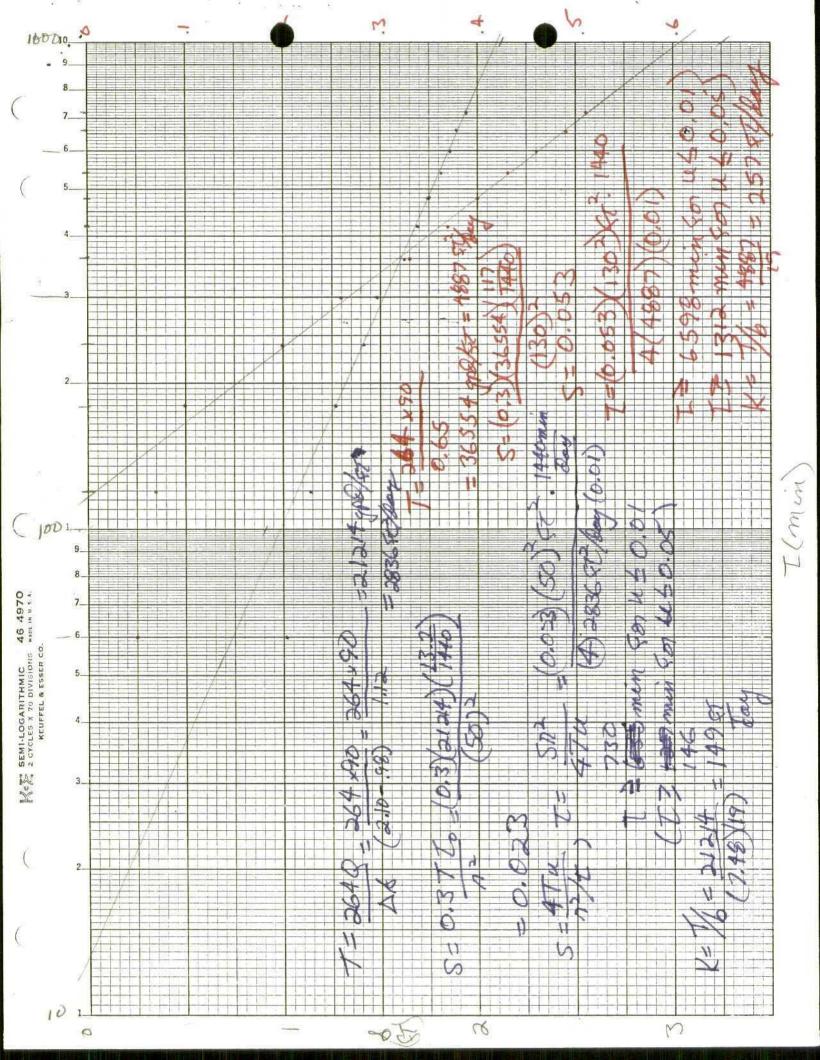
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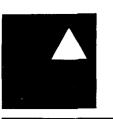
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# LAWRENCE A. BREWER & ASSOCIATES, INC.

CONSULTING ENGINEERS

May 24, 1982

Flora Vista Water Users Association P.O. Box 171 Flora Vista, New Mexico 87415

Southside Water Users Association Route 2 Box 266-C Aztec, New Mexico 87410

Re: CAC Project 81-29 Merger and Infiltration Gallery Feasibility Study

Gentlemen:

Transmitted herewith is the Feasibility Study for the above referenced CAC project. As you can see in the "Conclusions", we did not find a suitable location for an infiltration gallery. Without the ability to produce an acceptable quantity of domestic water, there appears to be no advantage to merging the two systems either for operational purposes or for production purposes. If you have questions regarding this report, please feel free to contact me at your convenience.

Sincerely yours,

LAWRENCE A. BREWER & ASSOCIATES, INC.

Richard P. Cheney, P.I Vice President

RPC:pmd F-152 S-170

**Enclosures** 

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## SECTION I GENERAL

## A. Purpose of Study:

The purpose of this study was to determine the feasibility of developing a radial infiltration gallery to produce domestic water to serve the Flora Vista and Southside Water Users Associations. These Associations are located in the Animas River Valley between the Cities of Farmington and Aztec, New Mexico. The Flora Vista Water Users Association serves the area north of the Animas River and the Southside Water Users Association serves the area south of the Animas River. The Flora Vista Association has 475 service taps at this time serving approximately 1,520 people. The Southside Water Users Association has approximately 180 service taps serving approximately 576 people. Both Associations obtain water from the City of Aztec. The Flora Vista Water Users Association has recently developed a well field consisting of three wells and a control station that services approximately 70% of the users. The balance of the users are still served by the City of Aztec. The Southside Water Users Association has installed a similar system; however, the ratio of users served by the Association's well is approximately 30% and the balance of the Southside users are served with water obtained from the City of Aztec. An additional purpose of the study was to determine the feasibility of merging the two domestic water systems into one system. It was anticipated that if the systems could produce a sufficient quantity of water to serve both systems that several economic advantages could be gained by combining the systems and producing water for their own consumption. Water produced by a radial well collector or infiltration gallery does not require any additional treatment and can be

-1-

distributed directly to the users with only the addition of the proper amounts of chlorine. It was further anticipated that by combining the systems, they could afford to hire a permanent full-time manager to supervise the operation and maintenance of all facilities.

# B. Area To Be Served:

The area proposed to be served by the development of a radial well collector consists primarily of those areas adjacent to the Animas River lying between the City Limits of Farmington and Aztec, New Mexico. A recent study prepared for San Juan County indicated that this service area had a population of approximately 5,500. The service area is outlined on the map attached to this report.

# C. Need For Facility:

Due to the rapid population increase in the San Juan Basin, all communities are finding it more difficult to meet the growing demand for domestic water. Rural water users associations who are totally dependent on cities such as Farmington, Aztec, and Bloomfield, find that their service is the first to be curtailed in times of peak demand. As indicated previously, the total population of the proposed service area is approximately 5,500. The present combined service population is approximately 2,100. This indicates that there is a present growth possibility of approximately 2.5 times the existing service. The additional demand would result from existing residents requesting service from the rural water users association. For these reasons, it would be advantageous for the rural systems to develop their own supplies either singly or in groups if feasible.

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Development of a firm, dependable domestic water supply owned and controlled by the Associations would tend to stabilize consumer costs throughout the area. Farmington and Aztec generally charge the rural water users association a rate that is equal to 1.5 times the domestic rate in the respective cities. Rural associations with few customers per mile of pipeline must charge higher rates for water than are available in other areas. It was anticipated that the development of an Association owned production system would allow the Associations to provide service to their customers at more economical rates. Both Flora Vista and Southside Water Users Associations own water rights in the Animas River and currently produce water from wells adjacent to the river. If a collection gallery of any type was constructed, water rights would have to be transferred to the gallery location. This would involve submitting an application to change point of diversion to the New Mexico State Engineer. Such an application is not appropriate at this time.

# D. Proposed Facilities:

The proposed facilities would consist of the construction of a radial well collector or an infiltration gallery with appurtenant pumping and chlorination facilities. These two types of systems were considered because they can be developed to such an extent that the only treatment required for production of the water is chlorination. Shallow wells in the area contain high amounts of iron and manganese and water produced from these type of wells require additional treatment. An infiltration gallery or radial collector well can develop water at a rate that will induce oxygen into the aquifer if there is an adequate hydraulic tie to the river. Oxygen induced into the

-3-

natural water supply will cause a bacterial growth that will fix iron and manganese in the aquifer so that it will not be present in product water. A radial well collector consist of a caisson sunk to a predetermined depth with horizontal pipes driven hydraulically parallel with the river bed. The length of hydraulically driven pipe is directly dependent upon the depth of the water bearing strata. If the strata has a shallow depth and a low permeability, then an infiltration gallery is constructed. An infiltration gallery consists of a caisson sunk to a predetermined depth with collector lines installed by conventional methods to an adequate length to develop the required amount of water. In both instances, the aquifer must have a strong hydraulic tie to the river so that water of an acceptable quality and quantity may be produced. Pumps, controls and chlorination facilities would be installed in an enclosed housing covering the top of the caisson in both instances. Attached is a partial copy of a report prepared by Serge Spirodonoff in 1963. This report effectively describes a radial well collector and infiltration gallery system. As can be seen from the Spirodonoff Report, an extensive geohydrological survey must be conducted in order to properly design the collector system. This geohydrological survey would be required and is applicable to any type of infiltration gallery system design. With this in mind, the Ranney Method Western Corporation was contracted to drill test wells and provide development testing for the collector system. The Ranney Report is attached as Appendix "B" to this report. The initial area selected for testing was property identified as Site "A" on a location map owned by the Flora Vista Water Users Association.

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The Association presently has three wells in operation at this site and it was felt that the alluvium in this area held possibilities for the development of a collector gallery. As can be seen in the report on Site "A", a cemented area of sand and gravel occurring in the upper portion of the alluvium formations effectively sealed off the Animas River and prevented recharge from the river making the site unsuitable for the development of a collector. Site "B" located approximately one-half mile downstream from Site "A" was selected because of some knowledge of the formation underlying the river at that point. Drill test performed for design of the footings of the Flora Vista Bridge indicated that a sand and gravel layer of appropriate thickness might be available near the location of the bridge. Site "B" is also near a six-inch water transmission line owned by the Flora Vista Water Users Association which could have been utilized for transmission of product water. The same cemented sand and gravel layer was encountered and test pumping was not successful at this site. An additional site identified as Site "C" on the location map was selected approximately one-half mile upstream from Site "A". Site "C" was selected after a review of the 1937, geohydrological survey conducted by the State Engineer that indicated that the river had been in this position since 1937. The drilling, however, indicated that the same cemented sand and gravel layers existed at this location as at the previous two locations. At this point, the drilling and research investigations had consumed all of the available funds for the project and it was determined that further exploration was not economically feasible at this time.

## E. Conclusions:

The Ranney Method Western Corporation Report specifically states, "In

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summary, the results of the test drilling program show that this entire stretch of the Animas River Valley, i.e. Sites "A", "B" and "C", is underlane by alluvial deposits which consist of cemented sand and gravel in the upper portion and only small zones of loose water bearing sand and gravel. near the base of the formation. These upper cemented sand and gravel zones effectively seal off the Animas River and prevent recharge to the aquifer. Since river recharge or replenishment does not occur, it is concluded that a firm, dependable naturally filtered water supply can not be developed and it is, therefore, recommended that no further consideration be given to the development of a Ranney Collector water supply within this stretch of the river valley."

The first paragraph on page three of the Ranney Report further states, "More important, however, is the fact that water levels continue to decline throughout the test period and stabilization did not occur. This indicates that no recharge occurs from the Animas River. Apparently, the cemented sand and gravel zones occurring in the upper portion of the formation effectively seal off the Animas River and prevent recharge from the river." Because there is no positive hydraulic tie to the river, it is the opinion of the Ranney Company and Lawrence A. Brewer & Associates, Inc. that the development of any type of infiltration gallery in this stretch of the river would be inappropriate. The feasibility of a merger of the two systems was dependent upon the development of a suitable water supply for both associations. Without the ability to produce domestic water of adequate quantities, there is no operational or economical advantage to combining the systems. At this time, it is recommended that the associations continue operations as they

-6-

have in the past and that they seek further funding for additional studies to determine if there are other areas adjacent to the river that might be suitable for the development of infiltration galleries or if there would be areas adjacent to the river suitable for the development of a system that would produce water by conventional means such as a standard filtration plant.

# F. Acknowledgements:

This Geohydrological Survey and Report was prepared utilizing funds provided by the New Mexico Department of Energy through the Community Assistance Council. Other funds were provided by the respective Associations.

* * * * * * *

-7-

# Design and Use of Radial Collector Wells

-Serge V. Spiridonoff-

A paper presented on Sep. 26, 1963, at the California Section Meeting, San Francisco, Calif., by Serge V. Spirodonoff, San. Eng. Assoc., State of California Dept. of Public Health, Bureau of San. Eng., Santa Rosa, Calif.

RADIAL collector wells are horizontal perforated conduits that intercept and collect ground water derived principally from surface water infiltration. Such supplies are usually found in the sand and gravel deposits underlying and hydraulically connected with surface sources, such as rivers, lakes, or oceans.

The radial collector well was developed and became practical in the 1930's and later was greatly improved. Since then, many municipalities throughout the world have successfully operated this type of infiltration gallery to obtain part of their water supply, among them Sacramento and Crescent City, Calif.

Radial collector wells provide an inexpensive and relatively simple method of obtaining water of a high quality for industrial and municipal use. The outstanding features of the radial collector well is:

1. The horizontal perforated collector pipe (the configuration and length of which vary) enable a large area of an aquifer to be exploited.

2. The removal of fine sand and gravel in the path of the projected collector pipe establishes an artificial aquifer of much higher permeability than the virgin soil. After construction, the collector pipe simply serves as a subdrain in a filter surrounded by a circle of coarse gravels several feet in diameter.

3. The unrestricted access and independent control of each collector pipe permit easy regulation of flow into a caisson and inspection and backwash of the collector pipe.

# Design Details

The radial collector well consists of the following:

A central, sectionally poured-inplace, reinforced-concrete, vertical shaft of a large diameter-13 ft or more-is sunk as a caisson to a predetermined depth. The average depth of the caisson below ground level is 70 ft, although the depth of one is more than 200 ft (the construction of which, incidentally, required lowering the caisson through 130 ft of wet ground). The wall thickness for a relatively shallow collector shaft is 18 in.; for a relatively deep one, 24 in. The bottom of the caisson is a heavy reinforcedconcrete slab, the thickness and the weight of which has no effect on caisson stability, owing to the interrelations of caisson size and weight.

Approximately 4 ft from the bottom of the caisson shaft are one or more tiers of horizontal perforated steel pipe, each pipe surrounded by a screen and connected to a valved port in the caisson wall. The number of ports for

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SERGE V. SPIRIDONOFF

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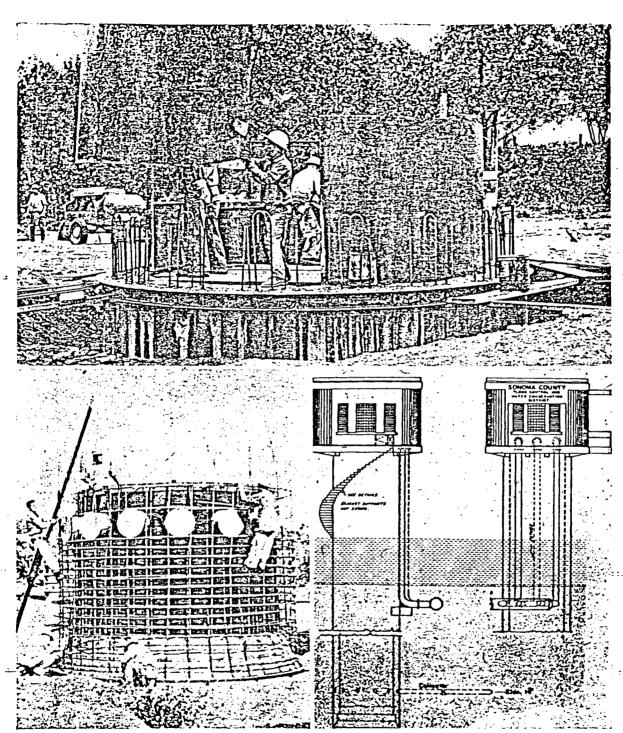


Fig. 1. Caisson Construction and General Structure of Radial Collector Wells

In photograph at top, steel inner forms for a section of reinforced-concrete caisson are being set. At lower left is shown the construction of another such section to be emplaced 60 ft below the river bed; on this section may be seen the port holes through which the collector pipe will be projected. At lower right is a diagram showing the general structural characteristics of radial collector wells. collection piping is usually more than that indicated by design requirements in order to anticipate future needs or unforeseen difficulties that may be encountered during the projection of the collector piping. The ports are usually no less than  $22\frac{1}{2}$  deg apart.

The collector piping usually has an \$-in. OD and a \$-in. wall thickness. The pipe is frequently made of a special copper alloy steel, but it may be of other acceptable material. The slots in the pipe, cut with a saw, are  $\frac{3}{3}$  in. wide, and  $1\frac{1}{2}$  in. long; the perforated area equals about 20 per cent of the total surface area. In general, 8-ft lengths of pipe are used.

The total length of collector pipe required depends upon the following variables, as determined in field tests: the quality of the water; minimum loss of head; and porosity and transmissibility of the formation. The lower the entrance velocity the better the water quality, the fewer the small particles that enter the collector, and the smaller the chance of incrustation.

The length of the individual collection pipe will be in the range of 110– 200 ft, the value depending on the porosity and transmissibility of the collection medium. Photographs of caisson construction and a general structural diagram at a radial collector well are shown in Fig. 1.

# **Construction Methods**

Inside the laterals, as they are driven outward during construction, a rubber packing, called the "sand line," is inserted in 2-in. or larger pipe and prevents water from entering the pipe at all points except through the slots of the driving head. This water flows at a very high velocity and carries all the finer material from the vicinity of the driving head into the caisson, developing a bed of gravel and coarse sand around the lateral as it is moved outward. The holes in the driving head are sized to allow the passage of sand and gravel up to a size larger than that of the smaller slots of the screen. After all but 1-2 ft of an 8-ft section of collection pipe has been driven out into the stratum, another 8-ft section is welded to the first. The extension of the lateral is continued until the driving head strikes an immovable object or the pipe has reached the design length. For each foot of collector pipe projected, 4-7 cuft of fine sand is usually removed.

Collector pipe is driven into the aquifer with two hydraulic jacks of 100-ton capacity each. These are placed against opposite points on the wall of the caisson shaft. The force required, greatest in the first few feet of penetration, diminishes with distance. At 100 ft, force for projection is approximately equal to the weight of the pipe projected plus the frictional resistance of the pipe moving against loose gravel. Hydraulic jack and sand line operation during projection of the collector pipe is shown in Fig. 2.

A difficulty in the installation of the collector pipe involves the selection of a proper relation between the rate of removal of particles from the stratum and the rate of projection of the pipe. If this relation is not correct, the pipe may easily be projected from the horizontal plane.

The ground water infiltrating the horizontal collector pipe flow into the central vertical shaft, where turbine pumps operated by automatic switches discharge the water into the distribution system. Chlorination is normally the only treatment needed. Deep well pumps, motor starters, and other equipment are usually located on top of the collector shaft and above high-water levels.

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The sand and gravel deposits underlying and hydraulically connected with a surface source, such as a river, lake, or sea, act as a filter medium. Such

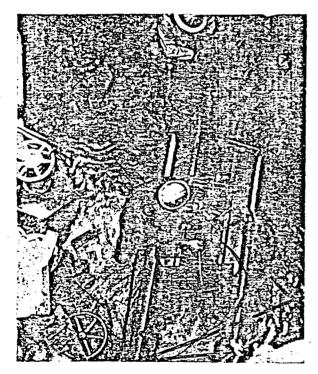


Fig. 2. Hydraulic Jack and Sand Line Operation During Projection of Collector Wells

In the photograph, water and fine material being removed from around the driving head of the collector pipe are discharged into the caisson. The quickopening valve controlled by the operator's right hand is open; his left hand is on an air valve (closed during flushing), which is connected to the sand line. A surging condition is created in the vicinity of the driving head by reciprocally opening and closing the two valves, thus reversing the direction of flow. The pipe being projected and the jack are clearly visible. deposits are among the most productive sources for water supplies.

Since the development of the radial collector in the early 1930's, more than 200 collector systems have been constructed. The median yield of a single collector is estimated at 6,000-7,000 gpm; however, collectors constructed along the Wabash River produce 10,000-15,000 gpm (Table 1). The collector constructed for Sacramento¹ averages 9 mgd for each 1,000 ft of collector pipe, and a collector constructed for the Sonoma Flood Control and Water Conservation District² is designed to provide 30 mgd for only 2,000 ft of collector pipe. Compared with the capacity of conventional types of infiltration galleries (1 mgd per 1,000 ft of collector pipe),⁸ the advantages of the radial collector appear to be very substantial.

The yield of the radial collector well may be estimated by the methods of computation discussed by Kazmann,⁴ or those of Mikels and Klaer,⁵ or those of Meinzer and Wenzel.⁶

In general, however, the potential yield of a collector depends upon the following five factors: the thickness of the water-bearing formation; the available drawdown; the effective radius of the collector; the permeability of the water-bearing formation; and the distance to the effective line source or effective line of infiltration.

#### Limitations

To be successful, a radial collector should be located in an aquifer of exceptional permeability and with an abundant surface source. In most areas, there is a sharp decline in the permeability of the aquifer and a deterioration in the quality of water in older aquifers at a distance from the river or stream channel. These

APPENDIX "B"

# RANNEY METHOD WESTERN CORPORATION

WATER SUPPLY ENGINEERS AND CONTRACTORS

RANNEY COLLECTORS

ARTIFICIAL RECHARGING

P. O. BOX 6387 Kennewick, Washington 99336 Telephone: (509) 586-6947

October 28, 1981

Lawrence A. Brewer & Associates, Inc. P.O. Box 2079 Farmington, New Mexico 87401

Attn: Mr. Richard P. Cheney Vice President

Re: Ranney Collector Water Supply Flora Vista Water Users Association Flora Vista, New Mexico

Gentlemen:

This letter-report sets forth the results of a test drilling program conducted for the Flora Vista Water Users Association in order to determine the feasibility of developing a naturally-filtered water supply of at least 600 gpm by means of a Ranney Collector constructed adjacent to the Animas River.

During the course of this investigation, test drilling was carried out at three sites, designated as Sites A,B, and C and located as shown in Figure SW-99-01. A total of four test holes were drilled at these sites, the written logs of which are attached to this report.

#### Site A:

Test Holes 1 and 2 were located at Site A, which is on property belonging to the Water Users Association. Test Hole 1, located 40 feet from the Animas River and 300 feet downstream from the upstream property line, encountered sand, gravel, silt and clay from ground surface to a depth of 22 feet, where shale bedrock occurred. In general, the material encountered was tight, with cemented layers, and had to be drilled. The only loose material was a sand and gravel layer occurring at the shallow depth of 7 to 10 feet.

Preliminary test pumping on September 5, with the well perforated at a depth of 13 to 18 feet, showed a drawdown of 7.0 feet after 2 hours of pumping at a rate of 75 gpm.

 $\begin{array}{cccc} \Theta & & & \\ \Theta & = & \\ A & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline & \\ \frac{R}{4} & = & \\ \hline & & \\ \frac{R}{4} & = & \\ \hline \\ \frac{R}{4} & = & \\ \\ \frac{R}{4} & = & \\$ 

ASSOCIATED WITH: RANNEY METHOD WESTERN OF CALIFORNIA, INC. The second state of the se

4, 5 3:00 p.m. dev) 6.73

6.86

6.93

6.99

7.05

7.10

7.13

7.18

4 4:00 p.m. 360

3, 45:00 p.m. 480 3 6:00 p.m. 480 - 00 n m. 540

2,7 7:00 p.m. 540

2.78:00 p.m. 602

2,2 9:00 p.m. 660

2,00 10:00 p.m. 754

Subsequent pumping on September 6 showed a drawdown of 6.4 feet after 2 hours of pumping at a rate of 90 gpm. The chemical analyses of water samples taken during this period are given as follows:

		Parts Per Mi	llion, except pH	<u>I</u>
Constituent	T.H. 1 9/5/81	T.H. 1 9/6/81	Association Wells 9/1/81	Animas River 9/1/81
Hardness P. Alkalinity	308		360 0	291 0
M.O. Alkalinity Iron Manganese	170 0.15 trace	0.03 0.07	239 trace 0.3	154 trace 0
pH Temperature( ^O F)	7.8 60	7.8 60	7.8 62	8.5 80

The chemical quality of the water from Test Hole 1 is similar to that of the Association's existing wells and does not contain excessive amounts of iron or manganese.

In order to further explore the water potential of this site, a 6-inch observation well (T.H.2) was drilled 50 feet downstream from Test Hole 1 and a 12-hour pumping test was conducted. Test Hole 1 was pumped from 10:00 a.m. until 10:00 p.m. on September 8, 1981. Water levels were recorded continuously in Test Hole 2 and the abandon Association well, located 130 feet inland from Test Hole 1, with the following results:

0,26

0.33

0.40

0,43

0,46

0,49

0,51

56.33

46.94

46.24

35,21 31.30 25,51 25,61

23,47 End test

	Depth to N	Water (feet)		
Days Time AT	<u>T.H.2</u>	Abandon Assoc. Well	12/2a	Remarks
$\frac{1}{42} - 10:00 \text{ a.m. } 0$	-	$\frac{1^{2}}{1.677.14}$		water levels
$\frac{24}{12:00}$ noon 120	6.38 1.14	20,837.16 6.0	7 140.83	
8 1:00 p.m. /80	6.51 127	13,897.19 0,10		
6 2:00 p.m. (44)	6.65 J,A)	10,427.29 6,2		

The total	drawdowns	for the	12-hour te	st were 1.94	feet for
Test Hole	2 and 0.5	l feet fo	or the aban	don Associat	ion Well.

1,49 6.3 7.35 1,62 694 7.42 1,69 5,96 7.46

1.75 5.2, 7.49

1,81 4.637.52

1,86 4,17 7.55

1,89 3,79 7.58

1,94 3,47 7.60

These drawdowns are considered to be excessive, indicating a low permeability in the order of magnitude of 750 gallons per day per square foot. More important, however, is the fact that water levels continued to decline throughout the test period and stabilization did not occur. This indicates that no recharge occurs from the Animas River. Apparently, the cemented sand and gravel zones occurring in the upper portion of the formation effectively seal off the Animas River and prevent recharge from the river.

With this lack of recharge or replenishment from the Animas River, it is concluded that a firm, dependable, naturallyfiltered water supply cannot be developed and therefore this site is not considered suitable for the construction of a Ranney Collector.

# Site B:

Test Hole 3 was located at Site B, which is on the Irene Brown property, just upstream from the Flora Vista Bridge. Test Hole 3, located 240 feet upstream from the bridge and 70 feet from the Animas River, showed the existence of sand, gravel, and silt from ground surface to a depth of 24 feet, where impermeable silty sand and clay was encountered. As was the case with Test Hole 1, the material encountered was tight, with cemented layers, and had to be drilled.

However, loose sand and gravel was encountered at depths of 17 to 24 feet. Initial attempts to pump the well with the perforations at a depth of 18 to 23 feet were unsuccessful. A second attempt to pump the well, with the perforations relocated at a depth of 15 to 20 feet, was unsuccessful due to excessive sand clogging the pump and hoses.

Again, it appears that the cemented sand and gravel zones, which extend to a depth of 17 feet, effectively seal off the Animas River and prevent recharge.

## Site C:

Test Hole 4 was located at Site C, which is on the Todd Hickman property about ½ mile upstream from Site A. Test Hole 4, located 30 feet from the Animas River, showed the existence of sand, gravel and silt from ground surface to a depth of 25 feet, where fine silty sand was encountered. As with the previous drilling, the material encountered was tight, with cemented layers, the only loose sand and gravel being a small zone occurring at a depth of 20 to 23 feet. Test pumping, with the well perforated at a depth of 18.5 to 23.5 feet, showed a drawdown of 10.4 feet at a pumping rate of 50 gpm without stabilization of the water levels.

In summary, the results of the test drilling program show that this entire stretch of the Animas River Valley, i.e., Sites A,B, and C, is underlain by alluvial deposits which consist of cemented sand and gravel in the upper portion and only small zones of loose, water-bearing sands and gravels near the base of the formation. These upper cemented sand and gravel zones effectively seal off the Animas River and prevent recharge to the aquifer.

Since river recharge or replenishment does not occur, it is concluded that a firm, dependable, naturally-filtered water supply cannot be developed and it is therefore recommended that no further consideration be given to the development of a Ranney Collector water supply within this stretch of the river valley.

We enclose our Invoice No. 99-01 in the amount of \$16,839.34 covering the total cost of this investigation. If you have any questions or need additional information, please let us know.

Very truly yours,

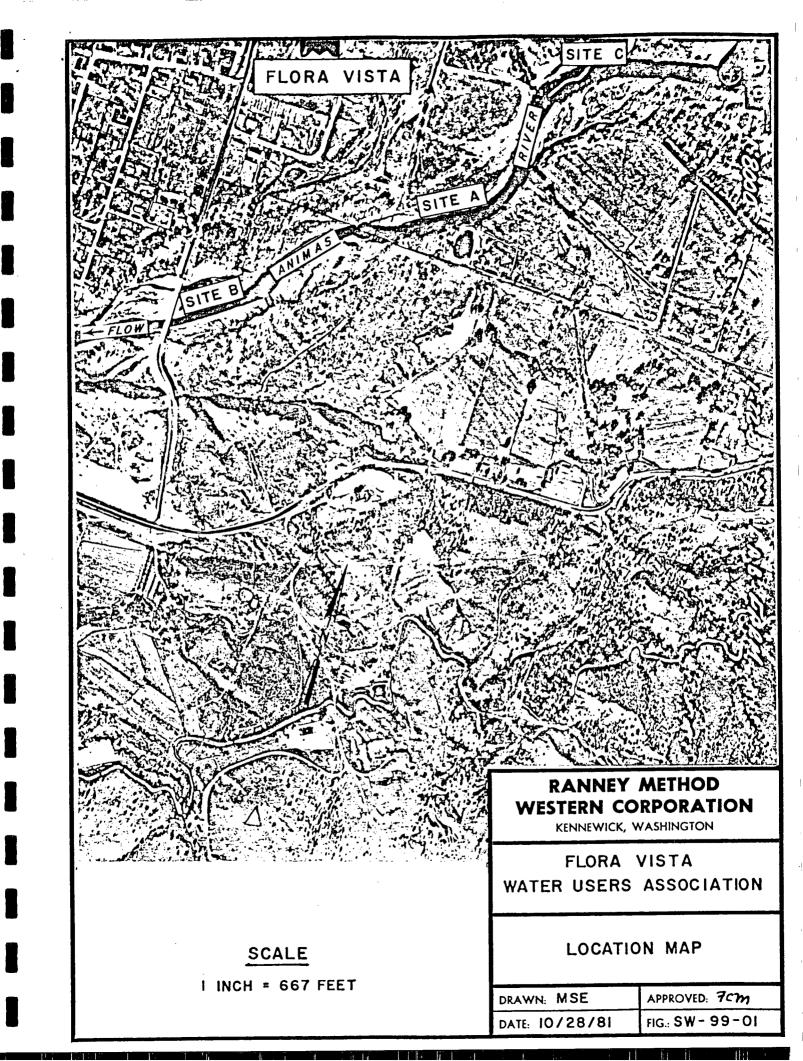
RANNEY METHOD WESTERN CORPORATION

Frederich C. Michele

Frederick C. Mikels, P.E. President & Chief Engineer

Enclosure

FCM/sk





# Ranney Method Western Corporation

Well No. T.H. 1 (Site A) Survey No. SW-99	
Owner <u>Flora Vista Water User's Association</u> Flora Vista, New Mexico	
WellLocation 40 ft. from Animas River, 300 ft. downstream from Assoc.'s propert line.	Ļγ
Diameter <u>8-inch</u> Depth <u>23 feet</u>	•
Total Casing 20 feet Perforations: Top 13 ft. Bottom 18 ft.	-
Elevations: Top of Casing 2 ft. above ground Ground	-
Static Water Level9-5-81 4.0 feet below ground surface.	
FROM TO DESCRIPTION OF MATERIAL	- -
0 2 Surface sand & gravel (to 6" +).	-
2 7 Red cemented sand & gravel (to 6"), hard, drilled, no wate	er.
7 10 Sand, small gravel, silty (few 6"), loose, bailed, fair wa	ate
10 13 Silty sand, small gravel, broken, snug, drilled, slow wate	er.
13 16 Cemented sand & gravel, silt, tight, drilled, slow water.	-
16 22 Red silt, sand & gravel, broken, tight, drilled, slow wate	er.
22 23 Greenish shale.	
<u>9-5-81: 7.0 foot of drawdown after 2 hours of pumping at</u>	
75 gpm. Well temperature-60°F., Total Hardness-308 ppm,	
M.O. Alkalinity-170 ppm, Iron-0.15 ppm, Manganese-trace.	-
<u>9-6-81: 6.4 feet of drawdown after 2 hours of pumping at</u>	
90 gpm. Well temperature-60°F., Iron-0.03 ppm,	
Manganese-0.07 ppm.	-
Driller Shorty Thompson Well Drilling Date Started9-2-81	
Report by <u>Gordon E. Hamm</u> Date Completed <u>9-5-81</u>	



# Ranney Method Western Corporation

Well No. <u>T</u>	.H. 2 (S	ite A)		Survey NoSW-99			
Owner Flor	e Vista	Water User's Associa	tion Flora				
Well Location	<u>40 ft. f</u>	rom Animas River, 50	ft. downstream	m from T.H. 1			
Diameter	6-inch		Depth 19.5 ft				
				13.5 ft. Bottom 19.5 ft.			
Elevations: To	op of Casing.	2 ft. above ground	Ground				
Static Water	Level9	-8-81 3.24 fee	t	below ground surface.			
FROM	то		DESCRIPTION OF MATE	RIAL			
0	3	Surface sand & grav	el (to 6" +).				
3	14	Cemented sand & gravel, silty, drilled, slow water.					
14	19.5	Brown sand, small gravel, silt, drilled, water.					
		Pocket of black silt at 18 feet.					
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DrillerSh	orty Thor	mpson Well Drilling	Date Started	9-5-81			
Report by	<u>Gordon E</u>	Hamm	Date Completed	9-7-81			
	_		t	A state of the			

RANNEY

# Ranney Method Western Corporation

Well No	H. 3 (Si	te B) Survey No. SW-99				
Owner Flora	a Vista	Water User's AssocationFlora Vista, New Mexico				
		own Property. 240 ft. upstream from bridge. 70 ft. from Animas River.				
Diameter 8-inch		inch Depth 25.5 feet				
Total Casing Pulled		lled Perforations: Top15_ft Bottom_20_ft				
Elevations: To	op of Casing.	Ground				
Static Water I	_evel	9-16-81 2.35 feet below ground surface.				
FROM	то	DESCRIPTION OF MATERIAL				
0	4	Silt, sand & gravel (to 6" +).				
4	6	Cemented sand, gravel & cobbles, drilled, no water.				
6	10	Red silty sand, pea gravel, cobbles, drilled, slow water.				
10	12	Brown sand, pea gravel (3" to 4"), heaves, bailed, fair water				
12	17	Cemented sand & gravel, broken, hard, drilled, no water.				
17	22	Grey sand, small gravel, heaves, bailed, fair water.				
22	24	Tan fine to coarse sand, very little gravel, bailed, fair wate				
24	25.5	Silty sand, clay, broken gravel, tight, drilled, no water,				
	· · · · · · · · · · · · · · · · · · ·	Oil streaks, bad smell.				
	Could not test pump well because of sand clogging					
	pump and hoses.					
Driller <u>Sho</u> a	cty Thom	pson Well Drilling Date Started 9-14-81				
Report by _Go	ordon E.	Hamm Date Completed9_17-81				
1						

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TANKA MARK
RANNEY

# Ranney Method Western Corporation

Well NoT	.H. 4 (S	ite C)	Survey No					
Owner Flor	a Vista	Water User's Associa	tion Flora	Vista, New Mexico				
Well Location	Todd Hi	ckman Property, 30 f	t. from Animas	River.				
Diameter	<u>    10-i</u>	nch	Depth25.5 ft					
Total Casing Pulled			Perforations: Top_1	3.5 ft. Bottom 23.5 ft.				
Elevations: T	op of Casing	2 ft. above ground	Ground					
Static Water	Level	9-22-81 6.2 feet	<u> </u>	below ground surface.				
FROM	то	DESCRIPTION OF MATERIAL						
0	44	Silty sandy soil.						
4	11	Red silty sand & gravel, (few 6"), snug, drilled, no water.						
11	20	Cemented sand & gravel, tight, drilled, no water.						
20	23	Brown sand & pea gravel (t0 5"), loose, bailed, fair water.						
23	25	Fine sand, broken gravel, very tight, drilled, slow water.						
25	25.5	25.5 Fine silty sand, oil streaks, drilled, no water.						
		· · · · · · · · · · · · · · · · · · ·	·					
	10.4 feet of drawdown for a pumping rate of 50 gpm.							
		Water level did not	stabilize.					
				<u> </u>				
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<del></del>	<u> </u>	<u> </u>						
Driller Shor	rty Thomp	oson Well Drilling	Date Started	9-19-81				
Report by	<u>Gordon E.</u>	. Hamm	Date Completed	9-23-81				
-				· · · · · · · · · · · · · · · · · · ·				

