

NEW MEXICO STATE LAND OFFICE
OFFICE OF THE STATE GEOLOGIST
SANTA FE, NEW MEXICO

MISCELLANEOUS NOTICES

Submit this notice in triplicate to the State Geologist or proper Oil and Gas Inspector at least five days before the work specified is to begin. A copy will be returned to the sender on which will be given the approval with any modifications considered advisable or the rejection by the State Geologist or Oil and Gas Inspector of the plan submitted. The plan as approved should be followed and work should not begin until approval is obtained.

Indicate nature of notice by checking below:

NOTICE OF INTENTION TO CHANGE PLANS	☐	NOTICE OF INTENTION TO PULL OR OTHERWISE ALTER CASING	☐
NOTICE OF INTENTION TO REPAIR WELL	☒		☐
NOTICE OF INTENTION TO DEEPEN WELL	☐		☐

Hobbs, New Mexico . August 12, 1933

PLACE

DATE

Mr. E. H. Wells State Geologist,

Santa Fe, N. Mex.

Following is a notice of intention to do certain work as described below at the

Stanolind Oil & Gas Company's McKinley Well No. 1 in NE-4
COMPANY OR OPERATOR LEASE
 of Sec. 8, T. 19 S, R. 30 E, N. M. P. M., Hobbs
 Oil Field, Lea County.

DETAILS OF PROPOSED PLAN OF WORK

To run Guiberson Spiral Tubing Packer on 3" A.P.I. 10 thd. Upset Tubing to run inside 6-5/8" 26# O.D. Casing, to shut off water from upper horizon of Oil Pay sand.

DUPLICATE

Approved AUG 19 1933, 19____
 except as follows:

[Signature]
NAME TITLE
 Address _____

Stanolind Oil & Gas Company
COMPANY OR OPERATOR
 By [Signature]
 Position Superintendent
 Send communications regarding well to
 Name Stanolind Oil & Gas Company
 Address Hobbs, New Mexico

THE UNIVERSITY OF CHICAGO
DIVISION OF THE PHYSICAL SCIENCES
DEPARTMENT OF CHEMISTRY
CHICAGO, ILLINOIS 60637

TO THE EDITOR:
I am writing to you to inform you of the results of our recent experiments. We have found that the reaction rate of the system is significantly affected by the concentration of the reactants. The data shows a clear trend where the rate increases as the concentration of the reactants increases. This is consistent with the expected behavior for a second-order reaction. We have also observed that the reaction is highly sensitive to temperature changes, with the rate doubling for every 10-degree increase in temperature. These findings are important for understanding the kinetics of this reaction and for developing more efficient catalytic systems. We will be publishing these results in the near future.

Sincerely,
Dr. John Doe

Dr. Jane Smith
Department of Chemistry
University of California, Berkeley

Enclosed for you are two copies of the manuscript of our paper. The first copy is for your personal use, and the second copy is for the library. We would appreciate it if you could let us know when you receive the manuscript. We are also enclosing a copy of the letter of transmittal from the publisher. Please let us know if you have any questions or need any further information.

Yours truly,
Dr. John Doe

cc: Dr. John Doe, 100 University of Chicago, Chicago, Illinois 60637
cc: Dr. Jane Smith, 100 University of California, Berkeley, California 94720

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University of Chicago

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