

NEW MEXICO OIL CONSERVATION COMMISSION

Santa Fe, New Mexico

MISCELLANEOUS REPORTS ON WELLS

ROBBS OFFICE

Submit this report in triplicate to the Oil Conservation Commission or its proper agent within ten days after the work specified is completed. It should be signed and sworn to before a notary public for reports on beginning drilling operations, results of shooting well, results of test of casing shut-off, result of plugging of well, and other important operations, even though the work was witnessed by an agent of the Commission. Reports on minor operations need not be signed and sworn to before a notary public. See additional instructions in the Rules and Regulations of the Commission.

Indicate nature of report by checking below:

REPORT ON BEGINNING DRILLING OPERATIONS		REPORT ON REPAIRING WELL	
REPORT ON RESULT OF SHOOTING OR CHEMICAL TREATMENT OF WELL		REPORT ON PULLING OR OTHERWISE ALTERING CASING	
REPORT ON RESULT OF TEST OF CASING SHUT-OFF		REPORT ON DEEPENING WELL	
REPORT ON RESULT OF PLUGGING OF WELL		Report on Running & Cementing Casing	

Lubbock, Texas

April 1, 1944

Place

Date

OIL CONSERVATION COMMISSION,
SANTA FE, NEW MEXICO.

Gentlemen:

Following is a report on the work done and the results obtained under the heading noted above at the _____
DeKalb Agricultural Association, Inc. Nancy G. Stovall Well No. 1 in the _____

Company or Operator _____ Lease _____
NE 1/4 of SW 1/4 of Sec. 8, T. 20-S, R. 38-E, N. M. P. M.,
Wildcat Field, Lea County.

The dates of this work were as follows: April 1, 1944

Notice of intention to do the work was ~~(max 102)~~ submitted on Form C-102 on _____ 19____
and approval of the proposed plan was ~~(max 102)~~ obtained. (Cross out incorrect words.)

DETAILED ACCOUNT OF WORK DONE AND RESULTS OBTAINED

Ran 2693.58' - 9-5/8" O. D. 36# Grade H-40 Casing. Set 2' below ground level and
14' below rotary. Set at 2711.58.

Cemented with 920 sacks cement. Pressure rose from 250# circulating pressure to
600# before plug was bumped. Mud weight 10.5# / gal. Cement weight 15.5# / gal.
Fill up 1400' or top of cement 1311'. Top anhydrite at 1510'.

Witnessed by _____ No witness _____
Name _____ Company _____ Title _____

Subscribed and sworn before me this _____

4 day of April, 1944

Celia Parker CCELIA PARKER
Notary Public

I hereby swear or affirm that the information given above
is true and correct.

Name E. G. McClure

Position Production Superintendent

Representing DeKalb Agricultural Assn., Inc.
Company or Operator

My commission expires June, 1945

Address 517 Lubbock National Bld. Lubbock, Texas

Remarks:

Roy Yarrhrough
Name _____
Title _____

THE UNIVERSITY OF CHICAGO
DEPARTMENT OF CHEMISTRY
RESEARCH REPORT NO. 1000

The following data were obtained from the study of the reaction of the various compounds with the various reagents. The results are given in the following table. The first column gives the name of the compound, the second column gives the name of the reagent, and the third column gives the result of the reaction. The results are given in the form of a percentage of the total amount of the compound which reacted with the reagent. The results are given in the form of a percentage of the total amount of the compound which reacted with the reagent.

Compound	Reagent	Result
1. 2,4-Dinitrophenol	1. 2,4-Dinitrophenol	100%
2. 2,4-Dinitrophenol	2. 2,4-Dinitrophenol	100%
3. 2,4-Dinitrophenol	3. 2,4-Dinitrophenol	100%
4. 2,4-Dinitrophenol	4. 2,4-Dinitrophenol	100%
5. 2,4-Dinitrophenol	5. 2,4-Dinitrophenol	100%
6. 2,4-Dinitrophenol	6. 2,4-Dinitrophenol	100%
7. 2,4-Dinitrophenol	7. 2,4-Dinitrophenol	100%
8. 2,4-Dinitrophenol	8. 2,4-Dinitrophenol	100%
9. 2,4-Dinitrophenol	9. 2,4-Dinitrophenol	100%
10. 2,4-Dinitrophenol	10. 2,4-Dinitrophenol	100%

The following data were obtained from the study of the reaction of the various compounds with the various reagents. The results are given in the following table. The first column gives the name of the compound, the second column gives the name of the reagent, and the third column gives the result of the reaction. The results are given in the form of a percentage of the total amount of the compound which reacted with the reagent. The results are given in the form of a percentage of the total amount of the compound which reacted with the reagent.