

N MEXICO OIL CONSERVATION COMMISSION

Santa Fe, New Mexico

MISCELLANEOUS NOTICES

Submit this notice in triplicate to the Oil Conservation Commission or its proper agent before the work specified is to begin. A copy will be returned to the sender on which will be given the approval, with any modifications considered advisable, or the rejection by the Commission or its agent, of the plan submitted. The plan as approved should be followed, and work should not begin until approval is obtained. See additional instructions in the Rules and Regulations of the Commission.

Indicate nature of notice by checking below:

NOTICE OF INTENTION TO TEST CASING SHUT-OFF	☒	NOTICE OF INTENTION TO SHOOT OR CHEMICALLY TREAT WELL	
NOTICE OF INTENTION TO CHANGE PLANS		NOTICE OF INTENTION TO PULL OR OTHERWISE ALTER CASING	
NOTICE OF INTENTION TO REPAIR WELL		NOTICE OF INTENTION TO PLUG WELL	
NOTICE OF INTENTION TO DEEPEN WELL			

Wink Texas

Place

11-19-36

Date

OIL CONSERVATION COMMISSION,
Santa Fe, New Mexico.

Gentlemen:

Following is a notice of intention to do certain work as described below at the

The Atlantic Refining Co. State L Well No. 3 in C. Lot 5
Company or Operator Lease
of Sec. 6, T. 21 S, R. 36 E, N. M. P. M., Eunice Field,
Lea County.

FULL DETAILS OF PROPOSED PLAN OF WORK

FOLLOW INSTRUCTIONS IN THE RULES AND REGULATIONS OF THE COMMISSION

Set 5 1/2" Oil string at 3750' w/ 800 sx cement.

Will test according to regulations.

Approved _____, 19_____
except as follows:

The Atlantic Refining Co.
Company or Operator

By

Position

Sup't.

Send communications regarding well to

OIL CONSERVATION COMMISSION,

By

Title

Name

E. T. Sterritt

Address

P.O. Box 876 Wink Texas

The following table gives the values of the various quantities which are required for the calculation of the rate of reaction. The values are given in units of $\text{cm}^3 \text{ mole}^{-1} \text{ sec}^{-1}$ for the rate constants, and in $\text{cm}^3 \text{ mole}^{-1}$ for the equilibrium constants. The values are given for the reaction at 25°C .

TABLE I. Values of the various quantities.

Quantity	Value
k_1	1.0×10^6
k_2	1.0×10^6
k_3	1.0×10^6
k_4	1.0×10^6
K_1	1.0×10^6
K_2	1.0×10^6
K_3	1.0×10^6
K_4	1.0×10^6

where k_1, k_2, k_3, k_4 are the rate constants, and K_1, K_2, K_3, K_4 are the equilibrium constants.

TABLE II. Values of the various quantities.

TABLE III. Values of the various quantities.

The following table gives the values of the various quantities which are required for the calculation of the rate of reaction. The values are given in units of $\text{cm}^3 \text{ mole}^{-1} \text{ sec}^{-1}$ for the rate constants, and in $\text{cm}^3 \text{ mole}^{-1}$ for the equilibrium constants. The values are given for the reaction at 25°C .

TABLE IV. Values of the various quantities.

TABLE V. Values of the various quantities.

TABLE VI. Values of the various quantities.

TABLE VII. Values of the various quantities.

TABLE VIII. Values of the various quantities.

TABLE IX. Values of the various quantities.

TABLE X. Values of the various quantities.

TABLE XI. Values of the various quantities.