

NEW MEXICO STATE LAND OFFICE
OFFICE OF THE STATE GEOLOGIST
SANTA FE, NEW MEXICO

MISCELLANEOUS NOTICES

Submit this notice in triplicate to the State Geologist or proper Oil and Gas Inspector at least five days before the work specified is to begin. A copy will be returned to the sender on which will be given the approval with any modifications considered advisable or the rejection by the State Geologist or Oil and Gas Inspector of the plan submitted. The plan as approved should be followed and work should not begin until approval is obtained.

Indicate nature of notice by checking below:

NOTICE OF INTENTION TO CHANGE PLANS	NOTICE OF INTENTION TO PULL OR OTHERWISE ALTER CASING
NOTICE OF INTENTION TO REPAIR WELL	
NOTICE OF INTENTION TO DEEPEN WELL	Notice of intention to treat with acid

Hobbs, N.M. January 25, 1935

PLACE

DATE

Mr. E.H. Wells State Geologist,
Santa Fe, N. Mex.

Following is a notice of intention to do certain work as described below at the _____

Tide Water Oil Co. O.L. Coleman Well No. 4 in NE 1/4
COMPANY OR OPERATOR LEASE
of Sec. 17, T. 21S, R. 36E, N. M. P. M., Eunice
Oil Field, Logan County.

DETAILS OF PROPOSED PLAN OF WORK

Gas has decreased until well would not flow through choke, treating to increase gas, will use 2000 gallons of Dowell XX acid.

RECEIVED AS P. K.
CP Miller

TRIPPLICATE

Approved 11/25/35, 19____
except as follows:

[Signature]

NAME TITLE
Address _____

Tide Water Oil Company
COMPANY OR OPERATOR
By _____
Position Production Superintendent
Send communications regarding well to
Name F. Schneider
Address Hobbs, New Mexico

THE UNIVERSITY OF CHICAGO
DIVISION OF THE PHYSICAL SCIENCES
DEPARTMENT OF CHEMISTRY

SECTION III. ANALYSIS

The analysis of the sample was carried out by the method of *gravimetric analysis*. The sample was weighed on a *gravimetric balance* and the weight was determined to ± 0.0001 g. The sample was then placed in a *gravimetric flask* and the flask was weighed on the same balance. The flask was then placed in a *gravimetric oven* and the sample was heated to a constant weight. The flask was then weighed again and the weight was determined to ± 0.0001 g. The difference between the two weights was the weight of the sample. The sample was then placed in a *gravimetric flask* and the flask was weighed on the same balance. The flask was then placed in a *gravimetric oven* and the sample was heated to a constant weight. The flask was then weighed again and the weight was determined to ± 0.0001 g. The difference between the two weights was the weight of the sample.

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