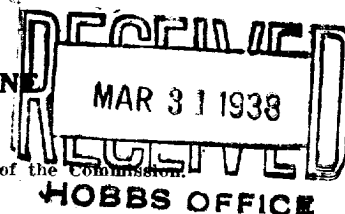


NEW MEXICO OIL CONSERVATION COMMISSION

Santa Fe, New Mexico

REQUEST FOR PERMISSION TO CONNECT WITH PIPE LINE



This request should be SUBMITTED IN TRIPLICATE. See instructions in the Rules and Regulations of the Commission.

Lunice, N.M.March 29, 1938

Place

Date

OIL CONSERVATION COMMISSION,
Santa Fe, New Mexico.

DUPLICATE

Gentlemen:

Permission is requested to connect Danglade-Clower Rosa Glier
 Wells No. 1 in SW NE of Sec. 33, T. 22, R. 37, N. M. P. M.,
Pemrose Field, Lea County, with the pipe line of the
Shell Pipeline Co. Houston, Texas.
 Status of land (State, Government or privately owned) Privately owned
 Location of tank battery 450 feet N of well
 Description of tanks 2 500 bbl. 16ft. Bolted National Tanks.
 Logs of the above wells were filed with the Oil Conservation Commission March 19, 1938, 19____
 All other requirements of the Commission have ~~have not~~ been complied with. (Cross out incorrect words.)
 Additional information:

Yours truly,

Permission is hereby granted to make pipe line connections
requested above.

OIL CONSERVATION COMMISSION,

By A. ANDREAS
 State Geologist
 Title Member Oil Conservation Commission
 Date March 31 1938

Danglade-Clower
 Owner or Operator
 By J. C. Clower
 Position One half owner
 Address Br. #380 Lunice, N.M.

1. The first part of the experiment is to determine the concentration of the solution. This is done by measuring the volume of the solution and the mass of the solute.

2. The second part of the experiment is to determine the molar mass of the solute. This is done by measuring the change in boiling point of the solution.

3. The third part of the experiment is to determine the van't Hoff factor of the solute. This is done by measuring the change in freezing point of the solution.

4. The fourth part of the experiment is to determine the degree of dissociation of the solute. This is done by measuring the change in osmotic pressure of the solution.

5. The fifth part of the experiment is to determine the molecular weight of the solute. This is done by measuring the change in viscosity of the solution.

6. The sixth part of the experiment is to determine the molecular weight of the solute. This is done by measuring the change in refractive index of the solution.

7. The seventh part of the experiment is to determine the molecular weight of the solute. This is done by measuring the change in density of the solution.

8. The eighth part of the experiment is to determine the molecular weight of the solute. This is done by measuring the change in surface tension of the solution.

9. The ninth part of the experiment is to determine the molecular weight of the solute. This is done by measuring the change in electrical conductivity of the solution.

10. The tenth part of the experiment is to determine the molecular weight of the solute. This is done by measuring the change in optical activity of the solution.

11. The eleventh part of the experiment is to determine the molecular weight of the solute. This is done by measuring the change in magnetic susceptibility of the solution.